THE IDENTIFICATION OF PARA-AMINOBENZOYL POLYGLUTAMATES IN EXTRACTS OF E. COLI

bу

ROBERT ALLAN HENRY FURNESS

The University of Manitoba

A Thesis

submitted to

The Faculty of Graduate Studies and Research

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in partial fulfilment

of the requirements for the degree of

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ABSTRACT

A glutamate-containing component of coenzyme A-disulfide prepared from *E. coli* has been isolated and shown to be a co-purifying contaminant. Physical and chemical techniques were used to characterize the contaminant as a mixture of p-aminobenzoyl poly(\gamma-L-glutamates) containing 5,6,7 and 8 glutamyl residues with the hexa- and heptaglutamyl forms predominating. Similar distributions of these molecules were present in extracts of both *E. coli* strains B23 and K12 grown either in minimal or enriched medium. A preliminary examination of the cellular pteroyl polyglutamate pool, from which the p-aminobenzoyl polyglutamates are derived, has shown a similar distribution. These results have cast some doubt on the reported size distributions of pteroyl polyglutamates in *E. coli*.

TO BARBARA

and

TO MY PARENTS

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ABBREVIATIONS

absorbance

 $A_{275/260}$ ratio of absorbance at 275 nm to that at 260 nm

ADP adenosine-5'-diphosphate

AMP adenosine-5'-monophosphate

ATP adenosine-5'-triphosphate

°C degree Celsius

5-CH₃H₄pteglu N⁵-methyl-5,6,7,8-tetrahydropteroylglutamate

5,10-CH-H₄pteglu N^5,N^{10} -methenyl-5,6,7,8-tetrahydropteroylglutamate

5,10-CH₂H₄pteglu N⁵, N ¹⁰-methylene-5,6,7,8-tetrahydropteroylglutamate

5-CHO-H₄pteglu N⁵-formy1-5,6,7,8-tetrahydropteroylglutamate

10-CHO-H₄pteglu N¹⁰-formy1-5,6,7,8-tetrahydropteroylglutamate

cm centimeter

CoA coenzyme A

(CoA)₂ oxidized coenzyme A-disulfide

CoASSG coenzyme A-glutathione mixed disulfide

DEAE diethylamino-ethyl

dATP deoxyadenosine-5'-triphosphate

FPGS folylpolyglutamate synthetase

g gram

glu glutamic acid

 $glu(\gamma-glu)_{n-1}$ poly(γ -glutamic acid) of n residues

gsh glutathione

h hours

H₂pteglu_n

7,8-dihydropteroylpoly(γ -glutamate)

H4pteglun

5,6,7,8-tetrahydropteroylpoly(γ-glutamate)

HC1

hydrochloric acid

Km

Michaelis constant

LiBH4

lithium borohydride

M

molar

min

minute

m1

milliliter

mo1

mole

NAD+

oxidized nicotinamide adenine dinucleotide

nm

nanometer

pABA

para-aminobenzoic acid

pABAg1u

N-(p-aminobenzoy1)-glutamic acid

 $\mathtt{pABAg1u}_{\mathbf{n}}$

p-aminobenzoylpoly(γ-glutamate) of "n" glutamyl

residues

pteglu_n

pteroylpoly(γ -glutamate) of "n" glutamyl residues

PO4 3-

inorganic phosphate ion

Rf

migration relative to the solvent front

RNA

ribonucleic acid

SDS

sodiumdodecylsulfate

TEAB

triethylamine-bicarbonate buffer

Tris

tris-(hydroxymethyl)aminomethane

UTP

uridine-5'-triphosphate

UV

ultraviolet

Vmax

maximum enzymic reaction velocity at saturating

substrate concentration

xg

times gravity

Zn

zinc metal

HISTORY

HISTORY

Introduction

During the study of two CoA-containing disulfide compounds found in E. coli it became apparent that a co-purifying contaminant existed which contained several glutamyl residues. Several such polyglutamate compounds have been found in nature. Ubiquitous are the pteroyl poly-γ-glutamates, folic acid and its conjugates, and also one of its catabolites, pABA polyglutamate. The polyglutamate portion of $pteglu_n$, which is of variable length (n = 1-7 usually) appears to be of importance in the binding of the folate cofactors to certain enzymes. One exception to this rule is the H2-pteglu6 found stoichiometrically in bacteriophage T4 tail plate. This molecule is apparently necessary as a structural component for the proper assembly of phage tail plates. The presence of pABA glu_n (in yeast) was first reported in 1946, and it was later established as a breakdown product of pteglun.

Encapsulated strains of certain $Bacillus\ spp.$ (notably $B.\ subtilis$, $B.\ licheniformis$ and $B.\ anthracis$) have as part of their capsule a very large homopolymer of glutamic acid. Like the pteglu_n and pABA glu_n the molecule contains γ -amide links but unlike them, it contains D-glu instead of the naturally predominant form L-glu. It is also much larger, being polymerized to the degree of up to several thousand residues as compared to the short oligopeptides of pteglu_n.

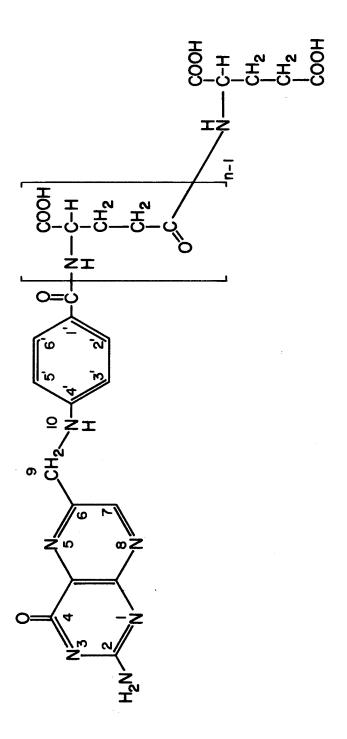
Pteroylpolyglutamates

1) Occurrence

The structure of the pteroyl polyglutamates, shown in Figure 1, was elucidated by Angier $et\ al\ (1946)$. There has been some confusion over nomenclature as folic acid is composed of the pteridine-pABA portion of the molecule and one glutamyl residue and is hence pteroylglutamate. This is the reason for the use of the terminology pteroylpolyglutamates rather than "folyl polyglutamates" or "folic acid conjugates" as was used earlier. The term conjugates referred to the fact that the growth factor folic acid was conjugated to a chain of several glutamyl residues.

The first isolation of pteglu_n was by Pfiffner et al (1946), who isolated the pteroyl heptaglutamate from yeast. Chemical analysis along with microbiological assay of the folates gave the ratio of seven γ-glutamyl residues per pteridine. The presence of pteglu_n was established in several organisms, but studies were complicated by several factors. The presence of folate compounds was determined by microbiological assay as with L. casei, S. faecalis and P. cerevisiae, organisms incapable of de novo synthesis of folic acid but able to utilize different forms of the pteroyl monoglutamate (i.e. pyrazine oxidation state and N⁵⁻, N¹⁰⁻ substituents), pteglu₂ or pteglu₃ for growth (Schertel et al 1965; Johns and Bertino 1965). It is not possible with bioassay to distinguish all forms of these folates without modification; for example 5-CH₃H₄ pteglu and pteglu₃ yield the same results. Pteroyl polyglutamates were assayed by cleavage to pteglu₂ by

Figure 1. The structure of pteglu $_n$. The molecule consists of a pteridine ring (1-8) joined to a p-aminobenzoyl (1'-6',10) residue. This is linked by an amide bond to an oligopeptide of (γ -L-glutamic acid) of length n. Co-enzyme activity is determined by oxidation state at C⁵ - C⁸, the one-carbon constituent carried at N⁵ or N¹⁰ and also the number of glutamyl residues present.



a γ-glutamyl carboxypeptidase ("conjugase") contained in chicken pancreas (Bird et al 1945, 1946). Chromatographic properties of natural folate derivatives do not allow easy separation of pteglun by size or charge, probably due to interaction of the pteridine moiety with chromatography matrices (Shin et αl 1972). Also, chemical determinations of glutamate chain length require isolation of the peptide in pure form. For these reasons determinations of the distributions of $pteglu_n$ chain lengths in different organisms was not practical. Corynebacterium excreted a pteglu3 into its growth medium (Hutchings et al 1948) while Streptococcus pneumoniae contained small amounts of pteglu and pteglu2, as well as large amounts of pteglu₃ and higher conjugates (Sirotnak et αl 1963). The presence of pteglu, was also established in tissues from rat (Bird et al1965), chicken liver (Noronha and Silverman 1962) and human whole blood (Herbert et αl 1962). Schertel et αl (1965) identified pteglu₂ and pteglu3 in yeast, as well as a polyglutamate form they assumed to be pteglu7 on the basis of the work of Pfiffner $et \alpha l$ (1946).

Characterization of glutamyl chain length was simplified by cleavage of the C^9-N^{10} bond, either oxidatively using alkaline permanganate (Houlihan and Scott 1972) or reductively with Zn/HCl (Baugh et~al~1974). Cleavage removes the chromatographically troublesome pteridine portion, leaving pABA glu_n, which can be separated more easily on the basis of glutamyl chain length. The results are not entirely quantitative, as discussed at length by Maruyama et~al~(1978), Baugh et~al~(1979) and Lewis and Rowe (1979), but are much clearer

than those previously obtained. As can be seen from Table 1, the distribution of pteglu_n chain length varies from organism to organism, but the polyglutamate group is apparent in all organisms studied and one chain length predominates in each.

There is still much confusion in the literature over the cellular distribution and even the predominant form of pteglu_n in some organisms. Much of this is a result of inconsistency and inadequacy of techniques, as discussed above. This is true especially in mammalian systems, where most of the folate pool exists as 5-CH₃H₄ pteglu_n , due to the failure of both oxidative and reductive cleavage to hydrolyse this form (Lewis and Rowe 1979).

In bacterial systems, however, some of the variation appears to be due to culture conditions. There is, for example L. casei, for which Baugh et al (1974) reported pteglu₄ as the predominant folate, but other authors found mostly pteglu₈ (Buehring et al 1974; Brody et al 1979; Brown et al 1974). Bassett et al (1976b) attributed this shift in chain length to exogenous folate availability as the only significant difference between the work of Baugh et al (1974) and the other groups was the former's use of almost twice the medium folate concentration. This they confirmed.

Another instance of a shift in $pteglu_n$ size profile is in $E.\ coli$ B upon infection with bacteriophage T4D, which contains $H_2pteglu_6$ as a component of its tail plate (Kozloff and Lute 1965). The $H_2pteglu_6$ is a stoichiometric component of the wedge-shaped structures comprising

Size distributions of pteroyl polyglutamates in various organisms.

TABLE I

			Glutamy1	Residu	es in pt	eglu _n (?	of To	tal)			
Organism	1	2	3	4	5	6	7	8	9	>9	Reference
Strep. faecalis ATCC 8043	16.5	8.5	20	54.5							Baugh <i>et al</i> (1974)
ATCC 8043 methotrexate resist	1	5.7	6.9	81.3	5.1						Baugh <i>et al</i> (1974)
8043		1	10	86	2.6						Brody et al (1979)
8043			4	61	23						Buehring et al (1974)
Lactobacillus casei ATCC 7469	3.2	0	9.3	59.4	23	5.1					Baugh et αl (1974)
								100			Brody <i>et al</i> (1979)
						8	14	42	19		Buehring et al (1974)
Corynebacterium spp.				>95							Shane (1980a)
Clostridium acidi-urici			>95								Curthoys et al (1972)
Escherichia coli B			>90 ~20	~20	~50						Wood et al (1968) Kozloff & Lute (1973) Powers and Snell (1976
K12	13	14.5	36	12	16	8	1.5				Bassett et al (1976b)
NCIB 8109	13	18	30	13	20	5	1				Bassett et al (1976b)
Saccharomyces cerevisiae						16	70	12			Bassett et al (1976a)
Neurospora crassa	<u>-</u>				14.5	80	0	5.6			Chan & Cossins (1980)
Rat liver					>95	· · · · · · · · · · · · · · · · · · ·					Shin <i>et al</i> (1972) Brody <i>et al</i> (1979)
Bacteriophage T ₄			,			100					Kozloff & Lute (1965)

the outer portion of the hexagonal baseplate and is found in close association with the apically located phage gene 11 product (Kozloff et al 1975, 1979). Kozloff's group has further shown that the Hopteglu6 is required as a structural component involved in attachment of the long tail fibers at or near the apices (Kozloff et alThis was accomplished using the inhibition of in vitro tail fiber attachment with oligo (γ-glutamy1) chains, γ-glutamy1 carboxypeptidase and antisera directed against a polyglutamate hapten. proposed that the presence of the seven anionic charges allows strong tail fiber attachment yet allows the necessary rotation freedom. This does not explain, however, the absolute requirement for H2pteglu6. Kozloff and Lute (1973) noted a shift from predominantly pteglu $_3$ to This shift was chloramphenicol-sensitive and apparently due to a late phage gene product. T4D gene 28 -infected cells accumulate larger conjugates (pteglug-pteglu₁₂) but little pteglu₆. Viable phage are not formed. It appears that pteglu6 is formed from a larger conjugate by cleavage due to a phage gene product. The successful complementation of T4D gene 28 -infected bacterial extracts with the addition of pteglu₆ confirmed this (Kozloff et αl 1973). The mechanism of the shift to larger conjugates is still unclear (Nakamura and Kozloff 1978).

2) Synthesis

The first report of a folylpolyglutamate synthetase (FPGS) was by Griffin and Brown (1964), who described such an activity in E. coli.

Subsequent work (Brown $et\ al\ 1974$) described folate synthesis in the succession H_2 pte, H_2 pteglu, H_4 pteglu and H_4 pteglu₂ and reported the extraction of an ATP-dependent FPGS catalyzing the formation of $10\ CHO-H_4$ pteglu₂ only (Masurekar and Brown 1975). This led to the postulation of two enzymes for polyglutamate synthesis.

Sakami et al (1973) described two FPGS activities in Neurospora crassa, one specific for the conversion of H_4 pteglu to H_4 pteglu₂ and one specific for elongation of H_4 pteglu₂ to H_4 pteglu₃. These activities could be chemically fractionated and were genetically characterized. Two methionine auxotrophs, met-6 and mac each were found to be lacking one of these activities (Chan and Cossins 1980).

The most extensive work has been that of Shane (1980a, 1980b, 1980c) on FPGS from Corynebacterium spp. He characterized the cellular folate content as to glutamyl chain length, one-carbon substituent and oxidation-reduction state (Shane 1980a) and isolated an FPGS enzyme to greater than 95% purity. This enzyme (Shane 1980b) has a molecular weight of 53 000 by SDS gel electrophoresis a pH optimum of 9.5 or greater, an absolute requirement for L-glutamate and for a monovalent cation (K⁺, 200 mM). The activity was Mg ATP-dependent, but dATP or even UTP would suffice to an extent. Folate requirement was variable but monoglutamates were much more effective substrates than higher forms, with H4pteglu the most effective (Km = 2 μ M). Despite the much lower affinity of his enzyme for di- and triglutamates, Shane showed that there was no appreciable difference in activities between the purified enzyme and crude extract, implying that only one and not

two enzymes were present. He speculated that formation of longer glutamyl chains had different optimum conditions. In support of the one-enzyme hypothesis he referred to "preliminary studies" indicating the utilization of $5,10\text{CH}_2\text{H}_4\text{pteglu}_{2-4}$, but this reference was unsupported by data.

Kinetic studies (Shane 1980c) indicate the reaction proceeds by an Ordered Ter Ter mechanism with substrates joining the enzyme complex in the order MgATP, $H_4pteglu$, glutamate, followed by release of products in the order ADP, $H_4pteglu_2$, PO_4^{3-} . Two speculative mechanisms consistent with this sequence are presented, both involving nucleophilic attack of the free glutamy1 α -amine, which would not be ionized at the pH for optimum activity (pH 9.5 or greater). One mechanism proposes nucleophilic attack on the mixed anhydride of a pteroyl (γ -glutamy1-phosphate) intermediate (Figure 2a), while the other proposes a concerted nucleophilic attack of the glutamy1 α -amine on a γ -carboxyl carbonium ion of the pteglu coupled with a similar attack of one of the γ -carboxyl oxygens on the γ -phosphorus of ATP (Figure 2b). Both mechanisms are plausible but the evidence at present (Shane 1980c) is at best indirect.

There have been several studies of mammalian FPGS. Gawthorne and Smith (1973) showed H4pteglu, 5-CHO-H4pteglu and 5-CH3H4pteglu to be substrates for enzyme extracts of sheep liver. Taylor and Hanna (1977) showed much lower activity for 5-CH3H4pteglu using enzyme from Chinese hamster ovary (CHO) cells. Spronk (1973) and McGuire et al (1980) found rat liver FPGS to be inactive with 5-CH3H4pteglu, as were 10-CHO-H4pteglu and H2pteglu. Studies with non-mammalian enzymes (Sakami et al 1973,

Figure 2. Possible mechanisms of chain elongation by folylpoly-glutamate synthetase.

Two mechanisms of glutamate addition to the free γ -carboxyl of pteglunhave been proposed by Shane (1980c). One invokes a nucleophilic attack on an active intermediate (a) while the other involves a concerted nucleophilic attack without prior formation of an active form of pteglu (b).

Shane 1980b) also reported relative inactivity with 5-CH $_3$ H $_4$ pteglu. These data are not necessarily in conflict with the fact that 5-CH $_3$ H $_4$ pteglu $_n$ are present as a large proportion of mammalian folate pools (Lewis and Rowe 1979) as they may only indicate that conversion to 5-CH $_3$ H $_4$ pteglu $_n$ takes place after polyglutamation.

McBurney and Whitmore (1974) isolated several auxotrophic mutants of CHO cells showing defects in intracellular folate metabolism. these were AUX B1 (a glycine, adenosine and thymidine auxotroph) and AUX B3 (a glycine and adenosine auxotroph). Not only did these lines have reduced folate concentrations but AUX B1 contained only monoglutamates while AUX B3 also contained some di- and triglutamates as opposed to the pentaglutamate of the parent line. The authors decided on genetic bases that these mutants had defects in one single locus. Using extracts of these lines Taylor and Hanna (1977) demonstrated that AUX B1 and AUX B3 lacked the parental folylpolyglutamate synthetase activity. Interestingly, they also showed that the number of glutamyl residues added to the substrate is dependent upon the H4pteglu concentration. At low levels (1 µM) mostly H4pteglu3 and some H4pteglu4 are formed while at higher levels (5 μM) mostly $H_4pteglu_2$ is made. very high levels (100 µM) only H4pteglu2 results. Taylor and Hanna (1979) further studied FPGS in extracts of revertant lines of AUX B1 and found increased heat sensitivity over wild-type in FPGS activity. This increased lability was apparent when either $\mathtt{H_4pteg1u_2}$ or $\mathtt{H_4pteg1u_4}$ synthesis was monitored. This led the authors to conclude that both diglutamate and tetraglutamate activities were on one enzyme.

McGuire et al (1980) have partially purified the rate liver FPGS

which shows similarities to the Corynebacterium enzyme (Shane 1980b). It requires a monovalent cation, of which K (20 mM) is optimal, has a high pH optimum (8.4), exhibits an absolute requirement for Lglutamate and a less stringent requirement for MgATP that can be alleviated by other nucleotide triphosphates. In contrast, the substrate specifity for folates is broader in that Hopteglu and 5,10CH-H4pteglu are utilized much more easily (compared to the primary substrate H4pteglu) than is the case with the Corynebacterium enzyme. Also, the rat liver enzyme catalyses in vitro the formation of pteglu4 in large amounts and pteglu5 (the predominant form in rat liver Table I) in an almost equal quantity. As with Taylor and Hanna's (1977) CHO extract the longer products are formed at low H_4 pteglu concentrations while at higher concentrations mostly pteglu2 is formed. It was found that pteglu5 was not effective as a substrate and also was a strong inhibitor of the enzyme. This led McGuire et alto propose that, on binding of pteglu5 to the enzyme, the length of the glutamyl chain places the terminal γ -carboxyl out of the active site. This suggests that, at least in rat liver, the predominant length of $pteglu_n$ is controlled solely by the size of the active site of FPGS. The binding of the end product $pteglu_5$ to the active site also indicates that feedback inhibition of polyglutamation may be one form of regulation of intracellular folate levels.

3) Significance

The polyglutamate conjugates of folic acid (with the exception of bacteriophage T4 as previously mentioned) appear to be the active coenzyme forms of the vitamin. Kisliuk and Gaumont (1974) showed that H₄pteglu₃ and H₄pteglu₆ were much better substrates for *L. casei* thymidylate synthetase than was H4pteglu. Polyglutamate forms of folate analogs were strong inhibitors of the enzyme, while $pABAg1u_6$ and $glu(\gamma-glu)_5$ were not bound by the enzyme, indicating that the $poly(\gamma-glu)$ portion of the molecule enhanced binding but did not determine it. Coward et αl (1974) demonstrated that dihydrofolate reductase from various mammalian sources has a higher affinity for H₂pteglu₃ H₂pteglu₅ or H₂pteglu₇ than for H₂pteglu. Powers and Snell (1976) found that E. coli ketopantoate hydroxymethyl transferase (involved in the synthesis of pantothenic acid) had much lower Km values for 5,10-CH₂H₄pteglu₄ and 5,10-CH₂H₄pteglu₅ than for other glutamyl chain lengths of the coenzyme. Baggott and Krumdieck (1979) obtained similar results with chicken liver 10-formyltetrahydrofolate:5'-phosphoribosy1-5-amino-4-imidazole-carboxamide formyl transferase, an enzyme of purine synthesis. The enzyme had a specificity (as defined by the ratio $^{Vmax}/_{Km}$) for the tetraglutamyl form 250-fold of that for the monoglutamyl form of the coenzyme. In their estimation this would make the tetraglutamate the only active form at the intracellular folate concentration found (1-2 μ M).

The work of McBurney and Whitmore (1974) and that of Taylor and

Hanna (1977, 1979) provides both genetic and biochemical evidence that the polyglutamyl forms of folate are necessary for the biosynthesis of glycine, purines and pyrimidines in mammalian cells. Moreover, the ability of AUX B3 FPGS to make $H_4pteglu_3$ but not $H_4pteglu_4$ (Taylor and Hanna 1979) indicates the requirement of the larger oligo(γ -glutamyl) coenzymes.

The work of McBurney and Whitmore (1974) still leaves open the possibility that polyglutamation of folic acid is a means of ensuring cellular retention of the vitamin as the AUX B1 and AUX B3 FPGS mutants contain 10% and 40% of the assayable folic acid of the parental This is supported by the observation (Herbert et al 1962) that serum contains only pteroyl monoglutamate, indicating it may be only a transport form of the vitamin as suggested by Rabinowitz (1960). However, in light of the enzyme kinetic studies cited above, this must be regarded as an auxiliary mechanism in cellular folate metabolism. As noted in both the bacterial (Bassett et αl 1976; Powers and Snell 1976) and mammalian (McGuire et αl 1979; Taylor and Hanna 1977) systems, the degree of polyglutamation is in inverse proportion to available folate levels. This relationship may be a manifestation of metabolic efficiency. When folate is limiting, both retention and reactivity are enhanced at the expense of ATP and glutamic acid. With excess folate, enhancement of activity and retention are not paramount, and ATP and glutamate are conserved. Such controls are quite feasible in the control of so important a metabolite.

Para-aminobenzoy1po1yglutamates

At about the same time as the isolation of $pteglu_n$ Ratner et al (1946) isolated a pABAglu_n from autolysed yeast. This was identified by the spectral characteristics of the pABA portion, especially the disappearance of the absorption at 273 nm on exposure to acid (pH 1) and by the presence of glutamic acid. The absorption drop is due to ionization of the arylamine, which indicates the peptide link to glutamate is through the pABA carboxyl. This was confirmed by diazotization of the free aryl amine, as opposed to the unreactiveness of blocked amines (e.g. N-acetyl pABA). Chemical analysis yielded a ratio of 10 or 11 glutamates per pABA, all being the L-isomer (bioassay). On the basis of a difference between estimates of glutamic acid nitrogen and α-amino nitrogen the presence of one dicarboxylic amino acid other than glutamic acid was postulated but not determined. The structural analogy between the pABA peptide and folic acid was noted by the authors, but no confirmation was attempted.

Stokstad et al (1947) noted that pABA glu was a product of UV photolysis of pteglu. Enzymic degradation of pteglu to pABAglu was demonstrated by Keresztesy and Silverman (1953). Futterman and Silverman (1957) showed the reaction to be dependent upon NAD and ATP and thought it to be non-enzymic oxidation of an unstable reduction product, probably H_4 pteglu.