F<sub>420</sub>H<sub>2</sub> dehydrogenation activity and electron transport in the methanoarchaea *Methanospirillum hungatei* GP1, *Methanosarcina barkeri* Fusaro, and *Methanosphaera stadtmanae* 

by

Patrick Chong

A thesis submitted in partial fulfillment of the requirements for the degree of

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### THE UNIVERSITY OF MANITOBA

#### **FACULTY OF GRADUATE STUDIES**

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A Thesis/Practicum submitted to the Faculty of Graduate Studies of The University of

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#### **DOCTOR OF PHILOSOPHY**

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### Abstract

The  $F_{420}H_2$  dehydrogenase (Fpo complex) of *Methanosarcina mazei* Göl catalyzes the  $F_{420}H_2$ -dependent reduction of methanophenazine, which is used to reduce the heterodisulfide complex (CoM-S-S-CoB) *via* the heterodisulfide reductase (Hdr). This activity is believed to be unique to methylotrophic members of the *Methanosarcinaceae*, but a similar activity, designated  $F_{420}H_2$  dehydrogenation, was observed in non-methylotrophic methanoarchaea, independent of growth substrate or phylogeny. A requirement for phenazine-dependent  $F_{420}H_2$  dehydrogenation activity is not readily apparent during electron transport in methanoarchaea that possess both the  $F_{420}$ -reducing hydrogenase ( $F_{420}$  H<sub>2</sub>ase) and  $F_{420}$ -nonreducing hydrogenase (MV H<sub>2</sub>ase); MV H<sub>2</sub>ase, along with Hdr, catalyzes the H<sub>2</sub>-dependent reduction of CoM-S-S-CoB. In the absence of  $F_{420}$  H<sub>2</sub>ase or MV H<sub>2</sub>ase, could there be a physiological function for the observed  $F_{420}H_2$  dehydrogenation activity? Several methanoarchaea were chosen for this study.

Methanospirillum hungatei GP1 has an  $F_{420}$  H<sub>2</sub>ase, but lacks MV H<sub>2</sub>ase; in the absence of the latter enzyme, there is no clear link between H<sub>2</sub> and CoM-S-S-CoB reduction in this methanoarchaeon. Soluble and membrane-bound  $F_{420}$ H<sub>2</sub> dehydrogenation activities were purified separately to apparent homogeneity and associated with the  $F_{420}$  H<sub>2</sub>ase; no other sources of  $F_{420}$ H<sub>2</sub> dehydrogenation activity was detected. Analyses of the genome of Msp. hungatei JF1 show that neither a distinct  $F_{420}$ H<sub>2</sub> dehydrogenase nor an MV H<sub>2</sub>ase is encoded by the genome, consistent with biochemical analyses. Genomic analyses did reveal the presence of genes encoding for a non-catalytic MV H<sub>2</sub>ase subunit that may provide an electronic link between the  $F_{420}$  H<sub>2</sub>ase and Hdr, providing a possible means for  $F_{420}$ H<sub>2</sub>-dependent CoM-S-S-CoB reduction in Msp. hungatei GP1.

Methanosarcina barkeri Fusaro, a methylotrophic member of the Methanosarcinaceae, was expected to be a model for the study of the  $F_{420}H_2$  dehydrogenase. Biochemical and/or genomic analyses indicate the presence of  $F_{420}$  H<sub>2</sub>ase, MV H<sub>2</sub>ase, and  $F_{420}H_2$  dehydrogenase in this methanoarchaeon. Kinetic analyses of the membrane-bound phenazine-dependent  $F_{420}H_2$  dehydrogenase activities of Ms. barkeri Fusaro and Ml. tindarius produced unexpectedly different activity profiles.

Purification to apparent homogeneity of the soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activities resulted in the isolation of the  $F_{420}$  H<sub>2</sub>ase; no other source of  $F_{420}H_2$  dehydrogenation activity was detected in our studies. While the phenazine-dependent  $F_{420}H_2$ -oxidation activity was not associated with a distinct  $F_{420}H_2$  dehydrogenase, the observed activity shares some similarities with the  $F_{420}H_2$  dehydrogenase activity of *Ms. mazei* Gö1. In the presence of  $F_{420}$  H<sub>2</sub>ase and MV H<sub>2</sub>ase, it is possible that, under the growth conditions tested, a distinct  $F_{420}H_2$  dehydrogenase may not be needed for CoM-S-S-CoB reduction.

Methanosphaera stadtmanae possesses an MV  $H_2$ ase, but an  $F_{420}$   $H_2$ ase activity has not been reported.  $O_2$ -labile, phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was observed at much lower levels relative to other methanoarchaea. The  $F_{420}H_2$  dehydrogenation activity was enriched, but could not be purified to homogeneity. This activity appears to be associated with two proteins:  $F_{420}H_2$  oxidase and  $F_{420}$   $H_2$ ase.

Analysis of the cell-free extract of Msph. stadtmanae indicated very low levels of  $F_{420}$ -reducing activity; this activity was partially characterized, and possesses distinctive qualities relative to other methanoarchaea. The genome of Msph. stadtmanae contains gene sequences encoding for subunits of a putative  $F_{420}$   $H_2$ ase; BLAST analyses of these sequences indicate similarity with the corresponding enzyme from Methanothermobacter thermoautotrophicus  $\Delta H$ , a  $H_2/CO_2$  metabolizing member of the Methanobacteriales. The finding of an  $F_{420}$   $H_2$ ase activity provides a better understanding of electron transport in Msph. stadtmanae.

 $F_{420}H_2$  oxidase activity ( $F_{420}H_2$ -dependent reduction of  $O_2$ ) was also detected in cell-free extract.  $F_{420}H_2$  oxidase and phenazine-dependent  $F_{420}H_2$  dehydrogenation activities declined rapidly when stored under aerobic conditions, indicating possible coassociation of the two activities. Further studies are required to resolve the  $F_{420}$   $H_2$  as and  $F_{420}H_2$  dehydrogenation/ $F_{420}H_2$  oxidase activities of *Msph. stadtmanae*.

While much has been elucidated of the electron transport pathways in the methanoarchaea, our studies show that variations in these pathways do occur in individual methanoarchaea. This is not unexpected as the methanoarchaea are a taxonomically diverse group of microbes, unified by the capacity to couple the production of methane with energy conservation.

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## List of Abbreviations

ADH	
CH₃OH	Alcohol dehydrogenase
CoB	
CODH/ACS	Coenzyme B, 7-mercaptoheptanoylthreoninephosphate
CoM	Carbon monoxide dehydrogenase/Acetyl-CoA synthase
CoM-S-S-CoB	Coenzyme M, 2-mercaptoethanesulfonic acid
DEAE	
DNase	Diethylaminoethyl
DPI	Dlemylaminoethyl
DSZM	Deoxyribonuclease Diphenyleneiodonium chloride
Feh	Deutche Sammlung von Zellkulturen und Mikroooranismen
LJOIL,	Energy congoming land
FDTA	Dithiothreitol
F <sub>120</sub>	Ethylenediaminetetraacetic acid
420 • • • • • • • • • • • • • • • • • • •	2-hydroxy 5 doggffania
$F\Delta D$	reduced 8-hydroxy-5-deazaflavin
FA FA	Flavin adenine dinucleotide
FA.	Faviii adenine dinucleotide
1 Gred	Padyand farmed and
Fno	$F_{420}H_2$ :NADP <sup>+</sup> oxidoreductase
Frh	$F_{420}H_2$ :phenazine oxidoreductase
1 111	Figure ducing by decomposition
1 40	Fuell-caring and and
1141711 1	Totachardennestlana
Hdr	Tetrahydrometnanopterin Tetrahydrosarcinopterin
HFPFS	Heterodisulfide reductase
kDa	(N-[2-Hydroxyethyl]piperazine-N'-[2-ethanesulfonic acid]
	leila dalta
Mc	Methanobacterium
MCAC	
M	MetnanococcusMetnanococcusMetnanococcus
MP	
Ms	
Msp	
Msph	Methanospirillum
Mst	MethanospiritiumMethanosphaera
<i>Mtb</i>	Methanosphaera Methanosaeta
MTZ	Methanosaeta Methanothermobacter
MV	Metronidazole
Mvh	Methyl viologenMethyl viologen-reducing hydrogenase
NAD <sup>+</sup>	Nicotinamide adenine dinucleotide
NADH	Reduced nicotinamide adenine dinucleotide
NADP <sup>+</sup>	
NADPH	Reduced nicotinamide adenine dinucleotide phosphate
	medianiae adenne dinucleotide phosphate

PAGE	Polyacrylamide gel electrophoresis
RNase	oryaci ylainide ger electrophoresis
CDC DACE	Ribonuclease
SDS FAGE	Sodium dodecyl sulfate nolyacrylamide gel electrophorosis
1115	Tris(hydroxymethyl)aminomothena
Tris/Cl	Tris(hydroxymethyl)aminomethane Chloride

# 1. Methanogenesis and electron transport in the methanoarchaea

### 1.1. The Production of Methane

Methane (CH<sub>4</sub>, MW = 16.0426 g/mol), the chief constituent of natural gas, is the simplest and most abundant of the hydrocarbons (Stafford *et al.* 1978). There are three natural sources of methane: (1) in the various natural gases originating from the physical and chemical breakdown of prehistoric plant material; (2) from biogenic sources such as flooded rice paddies and wetlands, where methane is produced *via* the decomposition of plants by anaerobic microorganisms; (3) from the gastrointestinal tracts of ruminant animals such as cattle or sheep, where methane is produced through the concerted activity of symbiotic anaerobic microorganisms who produce or consume H<sub>2</sub> (Meynell 1978, Bodelier *et al.* 2000). Methane is also produced in various man-made constructs, including sewage digesters, landfills, food plant waste fermenters, and oil wells (Garcia 1998). Anaerobic decomposition of organic matter results in the production of a gas containing 50-75% methane, 25-45% carbon dioxide, and small amounts of hydrogen, nitrogen, and hydrogen sulfide (Price and Cheremisinoff 1981).

The concentration of atmospheric methane has steadily increased over the last 300 years from 0.7 parts per million (p.p.m.) to 1.7 p.p.m., almost in line with the increase in human population growth (Pearce 1989, Thauer 1998). It has been estimated that 10<sup>9</sup> tons of methane are produced per year by methane-producing microorganisms. As global warming increases, methane is also released from the polar ice caps and the frozen tundra, where methane is trapped in the ice (Pearce 1989). The majority of the combustible gas (two-thirds) diffuses into aerobic zones where it is oxidized by

methanotrophic bacteria to CO<sub>2</sub>, with the remaining amount buried as methane deposits or released to the atmosphere where it is photochemically converted to carbon dioxide (CO<sub>2</sub>) (Anthony 1986, Ermler *et al.* 1997, Thauer 1998).

Global warming is due to the greenhouse effect, the phenomena where infrared energy emitted by the Earth's surface is absorbed by gases in the atmosphere and redirected back to the Earth (Dickinson and Cicerone 1986, Lorius *et al.* 1990). These gases include H<sub>2</sub>O vapor, CH<sub>4</sub>, CO<sub>2</sub>, ozone (O<sub>3</sub>), nitrous oxide (N<sub>2</sub>O), and various chlorofluorocarbons (Wang *et al.* 1976, Dickinson and Cicerone 1986). One molecule of methane can have a 20- to 25-fold higher effect on climate compared to CO<sub>2</sub>, making methane a highly potent greenhouse gas (Kilgore 1993, Bodelier *et al.* 2000). The contribution of methane to global warming has enhanced interest in the microbial processes that contribute to the global methane cycle (Conrad *et al.* 2000).

Oxidation of methane by methanotrophic microorganisms in oxic zones is a process that helps to regulate the levels of methane emanating from freshwater and marine environments. Within anaerobic marine sediments, significant quantities of methane that (> 80%) can be removed by a consortium of microorganisms that use methane as a source of electrons to reduce SO<sub>4</sub> to H<sub>2</sub>S (Boetius *et al.* 2000, Orphan *et al.* 2001, Sowers and Ferry 2002). This involves species of anaerobic methanotrophic (ANME) *Archaea* that convert methane to CO<sub>2</sub> and H<sub>2</sub>, and H<sub>2</sub>-scavenging sulfate-reducing bacteria, likely species of *Desulfobacteraceae* or *Desulfobulbaceae* (Hoehler *et al.* 1994, Orphan *et al.* 2001, Lloyd *et al.* 2006). A novel microbial sink for methane was recently reported by Raghoebarsing *et al.* (2006) in freshwater sediments, where methane oxidation is coupled to nitrate or nitrite reduction to N<sub>2</sub>. While the exact mechanism is

largely unknown, it is believed that the reaction involves a microbial consortium, with an archaeon (ANME) oxidizing the methane, and a novel bacterium that catalyzes the denitrification step. This process occurs in anaerobic sediments containing low levels of oxidizable substrates (other than methane) and high levels of nitrate.

A potential source of biofuel, methane can be siphoned from landfills and used for the following: fire kilns, furnaces, and boilers; internal combustion engines or turbines to generate electricity; or it may be cleaned and processed to a higher grade fuel (Richards 1989). Many commercial products are indirectly synthesized using methane. Carbon black, a substance formed *via* the incomplete combustion of methane, is used to make commercial items such as rubber tires, tubes, converyor belts, electronics components, and black pigment for printing cartridges.

# 1.2. The microbial basis for the production of methane - the methanoarchaea

The first report on the isolation of an organism from river mud that could oxidize H<sub>2</sub> and reduce one carbon compounds (such as CO<sub>2</sub>, CO, and formate) to methane was made by Stephenson and Strickland in 1933; this landmark paper marked the beginning of the modern era for the study of methanogenesis (Thauer 1998). While the isolated culture was not pure, it was significant to note that the culture did not yield methane from compounds containing two or more carbon atoms (Wolfe 1993).

The development of the Hungate technique, for cultivation of anaerobic microorganisms, allowed for isolation of these fastidious microorganisms for whole cell physiological studies in the 1950s, followed by cell extract studies in the 1960s (Hungate 1950, Wolfe 1993). In the ensuing years, many of the key enzymes, cofactors, and genes

of methanogenesis would be elucidated. In the 1970s, a pressurized anaerobic atmosphere was used to grow these strictly anaerobic microorganisms (Balch and Wolfe 1976). Another vital discovery was the recognition that the methane-producing "bacteria" were microorganisms belonging to the domain *Archaea* (Woese 1990).

### 1.2.1. The Archaea

In 1977, Carl Woese and coworkers exploited the partial sequences of the 16S-ribosomal RNA (16S rRNA) to monitor different phylogenetic types of microorganisms (Woese 1981). Prior to 1977 there had been a long held view that all organisms could be divided into two Super-Kingdoms, the prokaryotes and the eukaryotes (Danson 1988). Analysis of the 16S rRNA distinguished a group of microorganisms from both the eukaryotes and the prokaryotes, and in 1979 allowed them their own branch on the tree of life as a Super-Kingdom (Danson 1988, Cowan 1992). These organisms were christened archaebacteria (later renamed *Archaea*) and include diverse groups of microorganisms that thrive in unusual habitats with extremes of temperature, salinity and pH (Woese *et al.* 1990, Gray 1996, Schäfer *et al.* 1999).

The term *Archaea* represents a once held belief that these organisms originated from life forms preceding the division of the Eukaryal and Eubacterial domains; however, based on the sequence of universally present proteins, the *Archaea* have been placed on the branch leading to the Eukarya (Schäfer *et al.* 1999). *Archaea* are also distinguished from the Eukarya and Eubacteria by unique metabolic activities, suggesting uncommon structures of proteins and cofactors (Ferry 1997). Phylogenetically, most current isolates can be placed into two major kingdoms of *Archaea*, the *Crenarchaeota* 

and the *Euryarchaeota*; a third kingdom, the *Korarchaeota* and possibly a fourth, the *Nanoarchaeota* have been proposed (Woese *et al.* 1990, Aravalli *et al.* 1998, Huber *et al.* 2002).

The kingdom *Euryarchaeota* includes a diverse group of microbes, known collectively as the methanoarchaea, that couple the generation of energy with the production of methane; thus far, all organisms capable of energy-coupled methane generation are found within this single group (Woese *et al.* 1990).

### 1.2.2. Enter the methanoarchaea

Over two billion years ago, methane-producing microorganisms thrived in all types of habitat, but once O<sub>2</sub> saturated the oceans, these anaerobic microbes were forced to seek refuge in environments outside the reach of O<sub>2</sub> (Pearce 1989). The microbial degradation of organic material in an anaerobic environment forces the microorganisms to use molecules other than O<sub>2</sub> as electron acceptors (Price and Cheresinoff 1981). The methanoarchaea comprise the largest and most phylogenetically diverse members of the *Archaea*, and are ubiquitous in anaerobic environments, residing in diverse habitats, including aquatic sediments, bogs, marshes, tundra, from the wetwood of infected trees, in the rumen of animals, the mammalian large intestine, human dental plaque, guts of insects such as termites, in marine picoplanktons, and in deep-sea, hyperthermophilic geothermal vents (Zeikus and Henning 1975, Baross and Deming 1983, Belay *et al.* 1988, Cowan 1992, Aravalli *et al.* 1998, Leadbetter and Breznak 1996). Collectively, methanoarchaea are found in environments with wide ranges in salinity, temperature, and pH (Zinder 1993).

Methanoarchaea require an oxidation/reduction potential in the growth medium more negative than -0.3 V, where the O<sub>2</sub> concentration is approximately 10<sup>-56</sup> M (Zinder 1993). Pure methanogenic cultures have been observed surviving aerobic conditions for no more than 30 hours, with methane production decreasing after 200 minutes exposure to air (Kiener and Leisinger 1983, Fetzer *et al.*1993). Certain methanoarchaea, including *Methanothermobacter* (*Mtb.*) *thermoautotrophicus* and *Methanocaldococcus* (*Mc.*) *jannaschii*, form biofilms which may aid in limiting contact with O<sub>2</sub> (LePaglia and Hartzell 1997). Some methanoarchaea also possess enzymes to detoxify O<sub>2</sub> and reactive O<sub>2</sub> species, such as superoxide dismutase, catalase, or F<sub>420</sub>H<sub>2</sub> oxidase (Takao *et al.* 1991, Brioukhanov *et al.* 2000, Shima *et al.* 2001, Seedorf *et al.* 2004).

In natural ecosystems, the indigenous microflora can also develop a protective effect on the methanoarchaea against O<sub>2</sub> toxicity (Wagner *et al.* 1999). As a result, methanoarchaea can reside in communities under O<sub>2</sub>-limited conditions (i.e. microaerophilic), where O<sub>2</sub> is consumed by facultative microbes, and their metabolic byproducts are scavenged by the methanoarchaea for their own metabolic processes (Schink 1997). These communities make methanoarchaea a valuable resource in the biological treatment of wastewater pollution, where they coexist with O<sub>2</sub>-consuming bacteria (Kato *et al.* 1993).

The methanoarchaea derive their metabolic energy from the conversion of a restricted number of substrates to methane (Cowan 1992, Ferry 1999). The metabolic capacity to couple energy conservation with methane production, known as methanogenesis, is unique to the methanoarchaea (Zeikus 1977, Balch *et al.* 1979). Traces amounts of methane, which is not linked to energy conservation, are produced by

some sulfate-reducing bacteria well as from the sulfate-reducing archaeon *Archaeoglobus* fulgidus, during growth on lactate and sulfate (Beeder et al. 1994, Klenk et al. 1997, Shcherbakova and Vaĭnshteĭn 2000).

Methanogenesis represents the essential terminal step in the anaerobic breakdown of organic material under light-, nitrate-, and sulfate-limited conditions (Garcia *et al.* 2000, Sowers and Ferry 2006). Microbial partnerships are required for the complete breakdown of complex carbon substrates, such as cellulose, proteins, fatty acids, alcohols other than methanol (CH<sub>3</sub>OH), etc, to CO<sub>2</sub> and methane. The mineralization of organic matter typically follows four steps: hydrolysis; fermentation; acetogenesis/hydrogenesis (syntrophic); and methanogenesis (Pankhania and Robinson 1984). Various species of hydrolytic, fermentative, syntrophic, and acetogenic bacteria break down the complex carbon molecules to H<sub>2</sub>, acetate, CO<sub>2</sub>, and formate, which are then metabolized by the methanoarchaea as methanogenic substrate (McInerney *et al.* 1979, Mah 1982, Daniels 1984). The methane generated is oxidized to CO<sub>2</sub> by other microbial processes, and made available for photosynthetic fixation, completing the carbon cycle (Sowers and Ferry 2006).

The major energy-yielding step is associated with the reduction of a methyl (CH<sub>3</sub>-) moiety to methane ( $\Delta G^{\circ}$ '= -40 kJ/mol), with the electrons for the reductive step obtained from the oxidation of H<sub>2</sub> or formate (80% of species), CH<sub>3</sub>OH, methylamines or dimethylsulfide (under marine conditions, high sulfate) (26% of species), acetate (11% of species), and in a few instances ethanol or propanol (Rouvière and Wolfe 1988, Garcia 1998).

Interspecies  $H_2$  transfer between bacteria and methanoarchaea is a physiologically important process in microbial communities, maintaining low partial pressures of  $H_2$  (Mah 1982).  $H_2$  produced by bacterial metabolic processes is removed by the methanoarchaea and used as a source of electrons. Partnerships of  $H_2$ -producers with an  $H_2$  consumer allows for the complete breakdown of organic polymers (Zinder 1993, Sowers 1995). In natural habitats, partnerships with methanoarchaea can only take place in the absence of sulfate, as sulfate-metabolizing bacteria, such as the *Desulfovibrio* spp., have a higher affinity ( $Km = 2 \mu M$ ) and lower threshold for  $H_2$ , allowing these microbes to use  $H_2$  at low partial pressures more readily than their methanogenic competitors ( $Km = 4-8 \mu M$ ) (Kristjansson *et al.* 1982, Zinder 1993).

### 1.2.3. Phylogeny of the methanoarchaea

Presently, the methanoarchaea are divided into five distinct orders:

Methanobacteriales, Methanococcales, Methanomicrobiales, Methanopyrales, and

Methansarcinales (Thauer 1998). Recently, it has been suggested, on the basis of studies of the gene sequences encoding the enzymes from the methanogenic pathway (H<sub>2</sub>:CO<sub>2</sub>) and coenzyme F<sub>420</sub> synthesis, that the methanoarchaea should be separated into two independently evolving classes: Class I, encompassing the Methanobacteriales,

Methanococcales, and Methanopyrales, and Class II which contains the

Methanomicrobiales and Methanosarcinales (Bapteste et al. 2005).

The *Methanosarcinales* also have a number of non-methanogenic relatives, such as the *Halobacteriales*, *Thermoplasmatales*, and *Archaeoglobales*, *Archaea* which may be considered to be "degenerated methanoarchaea that have lost most of the proteins

involved in methanogenesis..." (Bapteste *et al.* 2005). Members of the *Archaeoglobales*, which are hyperthermophilic *Archaea* that contain many of the proteins required for reverse-methanogenesis (production of CO<sub>2</sub>), are hypothesized to be the missing link between the methanoarchaea and the sulfur-metabolizing *Archaea* (Bapteste *et al.* 2005).

The *Methanosarcinales* have evolved more extensively than the other methanogenic orders and show more variation in environmental niche and phenotypic characteristics; this is exemplified by the recently published genome of *Methanosarcina mazei* Gö1 which contains a significant number of bacterial genes, indicating that numerous lateral gene transfers may have played a significant role in the evolution of *Ms. mazei* Gö1, as well as for other members of the *Methanosarcinales* (Deppenmeier 2002). Members of the order *Methanosarcinales* are the most metabolically versatile of the methanoarchaea, as demonstrated by their collective ability to metabolize CO<sub>2</sub>, acetate, CH<sub>3</sub>OH, and methylamines (Rouvière *et al.* 1992). This is also evident by the presence of enzymes and structures involved in electron transport in members of the *Methanosarcinales* that are apparently not found in the *Methanomicrobiales* or in the methanoarchaea comprising Class I, which possess a single pathway for methanogenesis and are typically limited to the use only one or two methanogenic substrates (Deppenmeier *et al.* 2002, Galagan *et al.* 2002, Maeder *et al.* 2006).

# 1.3. An overview of methanogenesis and electron transport in the methanoarchaea

The process of methanogenesis is comprised of a series of oxidation/reduction reactions using H<sub>2</sub>, formate, CH<sub>3</sub>OH, or acetate as the source of carbon and/or electrons

for the production of CH<sub>4</sub> and ultimately to energy conservation. The metabolic pathways of methanogenesis involve a number of unusual cofactors, many of which are also found in ANME, cousins of the methanoarchaea that reside in methane-rich deep sea marine sediments (Deppenemier *et al.* 1996, Hallam *et al.* 2004).

Barker (1956) speculated that the production of methane from CO<sub>2</sub>, CH<sub>3</sub>OH, or acetate metabolism occurred *via* single-carbon (C<sub>1</sub>) reduction reactions partially carried out on pteridine and tetrapyrrole conenzyme carriers (Zeikus *et al.* 1985). In the ensuing years, these carriers – methanofuran, tetrahydromethanopterin, component B, and coenzyme M - were isolated and characterized. In the following section, the three main methanogenic pathways will be discussed: the reduction of CO<sub>2</sub> with H<sub>2</sub> (hydrogenotrophic); the reduction and/or disproportionation of CH<sub>3</sub>OH or methylamines (methylotrophic); and the fermentation of acetate (aceticlastic).

### 1.3.1. The hydrogenotrophic pathway

The conservation of hydrogenotrophic machinery in the vast majority of methanoarchaea suggests that this pathway may have been the ancestral form of methanogenesis (Bapteste *et al.* 2005). Approximately 80% of known methanoarchaea are hydrogenotrophic (Garcia 1998). H<sub>2</sub> is a major fermentation product of many species of anaerobic bacteria, fungi, and protozoa in many natural habitats, providing a steady source of electrons for methanogenesis (Zinder 1993).

The reduction of CO<sub>2</sub> to CH<sub>4</sub> consumes four molecules of H<sub>2</sub> (Fox *et al.* 1987):  $CO_2 + 4H_2 \rightarrow CH_4 + 2H_2O, \Delta G^{\circ \circ} = -131 \text{ kJ/mol CH}_4$ 

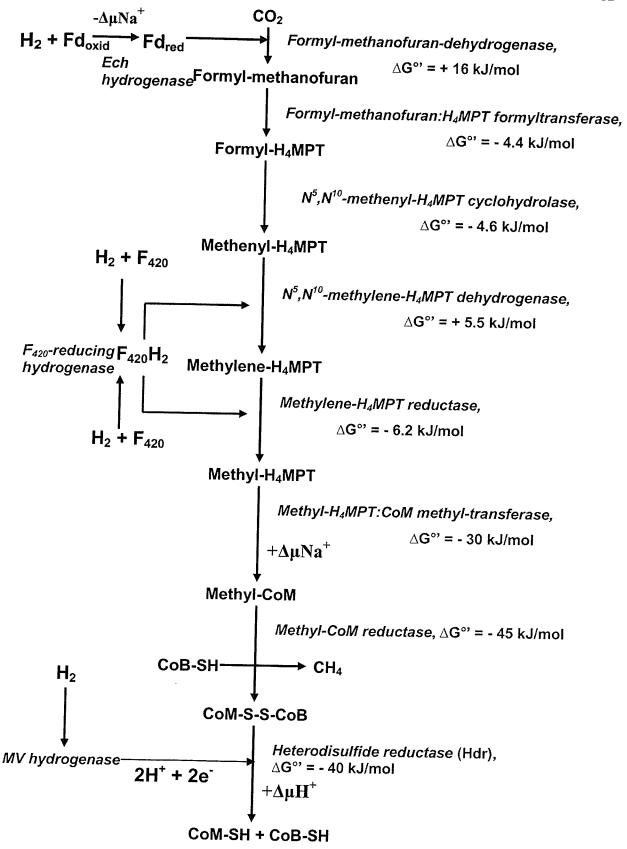
Formate (HCOOH) is another common microbial fermentation product used by some methanoarchaea as a carbon and electron source for methanogenesis; formate is converted to CO<sub>2</sub> and 2H<sup>+</sup> + 2e<sup>-</sup> via formate dehydrogenase (FDH) (Sparling and Daniels 1986, Ferry 1990, Zinder 1993):

$$HCOO^- \rightarrow CO_2 + H^+ + 2e^-$$

FDH has been purified from a number of methanoarchaea; studies with *Methanococcus (Mc.) thermolithotrophicus* indicate that synthesis of FDH may be regulated by the presence of H<sub>2</sub> (Sparling and Daniels 1990).

An overview of the hydrogenotrophic pathway is depicted in Figure 1.1. The first step in this pathway begins with the reduction of CO<sub>2</sub> to *N*-formyl-methanofuran (ΔG°'= +16 kJ/mol), catalyzed by the **formyl-methanofuran dehydrogenase** (Thauer *et al.* 1993). Formyl-methanofuran dehydrogenase has been purified from a number of methanoarchaea, including *Methanothermobacter* (*Mtb.*) *marburgensis*, *Methanobacterium* (*Mb.*) *formicicum*, and *Methanosarcina* (*Ms*) *barkeri* strain Fusaro (Ferry 1999). The enzyme from *Ms. barkeri* consists of five subunits (FmdEACDB), with a molybdopterin guanine dinucleotide, using CO<sub>2</sub> as substrate and forming N-carboxymethanofuran as an intermediate (Karrasch *et al.* 1990, Vorholt and Thauer 1997, Bartoschek *et al.* 2000).

Mtb. thermoautotrophicus possesses two isoforms of formyl-methanofuran dehydrogenase, containing either molybdenum or tungsten (Hochheimer et al. 1995).



**Figure 1.1**. Pathway of methanogenesis from  $H_2$  and  $CO_2$  in *Mtb. thermoautotrophicus*  $\Delta H$  and other hydrogenotrophic methanoarchaea (Deppenmeier *et al.* 1996, Shima *et al.* 2002).

The genes encoding the tungsten-dependent enzyme are transcribed when tungsten or molybdenum is present in the growth medium, whereas the molybdenum-dependent enzyme is only transcribed when molybdate is present (Hochheimer *et al.* 1996). The tungsten-containing isoform is comprised of four subunits (FwdABCD), while the molybdenum variant is a three subunit enzyme (FmdABC) (Hochheimer *et al.* 1995). Interestingly, the FmdA subunit is apparently identical to the FwdA subunit, and it has been proposed that the A-subunit for both variants is encoded by the same gene, in the *fwd* operon encoding the tungsten-dependent enzyme (Hochheimer *et al.* 1996).

Originally dubbed CDR (carbon dioxide reduction factor), methanofuran (Figure 1.2) was renamed for the uncommon presence of a furan ring (Leigh and Wolfe 1983, Leigh *et al.* 1984).

Figure 1.2. Structure of methanofuran

The requirement for methanofuran during methanogenesis on  $CO_2$  was initially reported by Romesser and Wolfe (1982), and established by Leigh *et al.* (1985) as a formyl-carrier. The fixation of  $CO_2$  to methanofuran (Figure 1.3) is an endergonic reaction ( $\Delta G^{\circ}$ ' = +16 kj/mol) that requires reducing equivalents produced by the membrane-bound Ech hydrogenase, with an energy conserving sodium gradient (2-4 Na<sup>+</sup>) driving the reaction (Kaesler and Schönheit 1989a, 1989b, de Poorter *et al.* 2006). The reducing equivalents are in the form of reduced ferredoxin (FD-H<sub>2</sub>) generated *via* an

 $H_2$ -oxidizing Ech hydrogenase that is driven by consumption of a Na<sup>+</sup> gradient (- $\Delta\mu$ Na<sup>+</sup>) (Figure 1.1) (Deppenmeier 2002, Meuer *et al.* 2002, Hedderich 2004).

$$R \rightarrow NH_3 + CO_2 + H_2 \rightarrow R \rightarrow NH_3 + H_2O + H_2$$

Figure 1.3. Reduction of CO<sub>2</sub> to formyl-methanofuran

The formyl group is then transferred to tetrahydromethanopterin ( $H_4MPT$ ) to form  $N^5$ -formyl-  $H_4MPT$  via formylmethanofuran:  $H_4MPT$  formyltransferase (Figure 1.4) ( $\Delta G^{\circ}$ '=-4.4 kJ/mol) (Brietung and Thauer 1990, Thauer et al. 1993).

Figure 1.4. Formation of N<sup>5</sup>-formyl-H<sub>4</sub>MPT from formyl-MFR and H<sub>4</sub>MPT

Formylmethanofuran: H<sub>4</sub>MPT formyltransferase has been purified from Mtb. thermoautotrophicus, Ms. barkeri, Methanopyrus (M.) kandleri, as well as the sulfate-reducing archaeon A. fulgidus. It is a monomer or tetramer of a polypeptide of apparent molecular mass 32-41 kDa (Thauer et al. 1993). H<sub>4</sub>MPT (Figure 1.5), which was discovered from deproteinated extracts of Mtb. thermoautotrophicus, acts as a  $C_1$  carrier at various oxidation states, existing in a derivatized form with a bound formyl, methenyl, methylene, or methyl group linked at either or both  $N^5$  or  $N^{10}$  positions of the pterin group (Keltjens et al. 1983, Escalante-Semerena et al. 1984a, 1984b).

Figure 1.5. Structure of tetrahydromethanopterin (H<sub>4</sub>MPT)

H<sub>4</sub>MPT resembles tetrahydrofolate (H<sub>4</sub>F), a  $C_1$  carrier in humans and bacteria although the two compounds are not functionally equivalent (Edward and Maden 2000). H<sub>4</sub>MPT and H<sub>4</sub>F both consist of a pterin ring linked to an arylamine with the  $C_1$  moiety binding at  $N^5$  and  $N^{10}$ . H<sub>4</sub>F contains glutamate residue(s) linked to the benzene ring, while H<sub>4</sub>MPT consists of a ribitol residue linked to ribose-5-phosphate, followed by hydroxyglutarate (Edward and Maden 2000). H<sub>4</sub>MPT is also a carbon carrier in the biosynthesis of cell carbon, providing units for amino acids and purines (DiMarco 1990). Methanoarchaea of the order *Methanosarcinales* contain tetrahydrosarcinopterin (H<sub>4</sub>SPT), which possesses an additional glutamate residue attached to the hydroxyglutarate *via* an amide linkage to the  $\alpha$ -carboxylic acid group (van Beelen *et al.* 1984).

The next three reactions cover the reduction of  $N^5$ -formyl-  $H_4MPT$  to  $N^5$ -methyl-  $H_4MPT$ .  $N^5$ ,  $N^{10}$ -methenyl- $H_4MPT$  cyclohydrolase catalyzes the conversion of  $N^5$ -formyl-  $H_4MPT$  to  $N^5$ ,  $N^{10}$ -methenyl- $H_4MPT$  ( $\Delta G^{\circ}$ '=-4.6 kJ/mol) (Figure 1.6) (Thauer *et al.*1993). The respective enzymes from *Mtb. thermoautotrophicus*, *M. kandleri*, and *Ms. barkeri* contain two identical subunits of 37-41 kDa with no identifiable prosthetic groups (Ferry 1999).

$$H^{+} + H_{2}N + H_{2}N + H_{2}O + H_$$

**Figure 1.6**. Conversion of  $N^5$ -formyl-  $H_4MPT$  to  $N^5, N^{10}$ -methenyl- $H_4MPT$ 

 $N^5$ , $N^{10}$ -methylene-  $H_4$ MPT dehydrogenase catalyzes the reversible reduction of  $N^5$ , $N^{10}$ -methenyl- $H_4$ MPT to  $N^5$ , $N^{10}$ -methylene-  $H_4$ MPT, using reducing equivalents provided by reduced coenzyme  $F_{420}$  ( $F_{420}H_2$ ) ( $\Delta G^{\circ}$ '=-5.5 kJ/mol) (Figure 1.7) or  $H_2$  (von Bünau *et al.* 1991, Thauer *et al.* 1993). Coenzyme  $F_{420}$  (8-hydroxy-5-deazaflavin) is the major electron carrier found in all methanoarchaea, although the amount varies in each methanoarchaeon. It is an obligate hydride donor/acceptor that is essential during methanogenesis in the hydrogenotrophic as well as methylotrophic pathways, and also for biosynthetic reactions. During growth on  $H_2$ : $CO_2$ ,  $F_{420}H_2$  is regenerated by the  $F_{420}$ -reducing hydrogenase (Thauer *et al.* 1993).

$$F_{420}H_2 + H_2N + H_3N + F_{420} + H_4N + H_2N + H_3N + H_4N + H_5N + H_5N$$

**Figure 1.7**. Reduction of  $N^5$ ,  $N^{10}$ -methenyl- $H_4$ MPT, with electrons from reduced coenzyme  $F_{420}$ , to  $N^5$ ,  $N^{10}$ -methylene-  $H_4$ MPT

The F<sub>420</sub>-dependent enzyme has been purified from *Ms. barkeri*, *Mtb. thermoautrophicus*, and *M. kandleri* (Mukhopadhyay and Daniels 1988, te

Brommelstroet *et al.* 1991a, te Brommelstroet *et al.* 1991b, Klein *et al.* 1993). The

enzyme is found as either a hexamer or an octamer, consisting of identical subunits (30-36 kDa) (Ferry 1999).

The H<sub>2</sub>-dependent variant has been described for *Mtb. marburgensis*, *M. kandleri*, and *Mc. thermolithotrophicus* (Zirngibl *et al.* 1990, Ma *et al.* 1991, Hartmann *et al.* 1996). The native enzyme consists of only one type of subunit, and is the only known hydrogenase that does not contain iron sulfur (FeS) clusters (Zirngibl *et al.* 1992, Albracht 1994). The presence of a tightly bound cofactor containing Fe was recently reported for this enzyme (Buurman *et al.* 2000, Lyon *et al.* 2004).

Studies with Mtb. marburgensis indicate that under conditions with non-limited  $H_2$ , the  $H_2$ -dependent enzyme is transcribed, whereas the  $F_{420}$ -dependent version is produced when  $H_2$  is limited (Nölling  $et\ al.\ 1995$ ). Under Ni-limited conditions, the  $H_2$ -and  $F_{420}$ -dependent  $N^5, N^{10}$ -methylene-  $H_4MPT$  dehydrogenases together catalyze the reduction of  $F_{420}$  with  $H_2$ , as the Ni-dependent  $F_{420}$ -reducing hydrogenase activity is essentially zero (Afting  $et\ al.\ 1998$ ).

 $F_{420}H_2$ -dependent **methylene-H<sub>4</sub>MPT reductase** catalyzes the reversible reduction of  $N^5$ ,  $N^{10}$ -methylene-H<sub>4</sub>MPT to  $N^5$ -methyl-H<sub>4</sub>MPT ( $\Delta G^{\circ}$ '=-17.2 kJ/mol)

$$F_{420}H_2 + H_2N + H_3N + F_{420}$$

**Figure 1.8**. Reduction of  $N^5$ ,  $N^{10}$ -methylene-H<sub>4</sub>MPT to  $N^5$ -methyl-H<sub>4</sub>MPT.

(Figure 1.8) (Thauer et al. 1993).

The reductase has been purified from Mtb. thermoautotrophicus, Mtb. marburgensis and Ms. barkeri (Ma and Thauer 1990, Ma and Thauer 1990b, te Brommelstroet et al. 1990). The enzymes contain one subunit (35-38 kDa) without a prosthetic group, and directly transfer a hydride ion from  $F_{420}H_2$  to  $N^5$ ,  $N^{10}$ -methylene- $H_4MPT$  (Vaupel and Thauer 1995, Ferry 1999).

The next major step involves transfer of the methyl group from N<sup>5</sup>-methyl-H<sub>4</sub>MPT to coenzyme M (2-mercaptoethanesulfonic acid, CoM-SH), via methyl-H<sub>4</sub>MPT:CoM methyltransferase (ΔG°'=-29.7 kJ/mol) (Figure 1.9) (Thauer et al. 1993).

Figure 1.9. Transfer of the methyl group from H<sub>4</sub>MPT to CoM-SH.

CoM-SH, the simplest coenzyme known (to date), functions as a methyl carrier during methanogenesis (Taylor and Wolfe 1974, DiMarco 1990). Cell extracts from *Methanobacterium* M.o.H. (*Methanobacterium bryantii*) were separated *via* dialysis by McBride and Wolfe (1971), who discovered a factor that was required for methanogenesis from CH<sub>3</sub>-Cbl; as this factor was necessary for methyl transfer, it was named coenzyme M. Studies by Taylor and Wolfe (1974) distinguished CoM-SH as a novel coenzyme. CH<sub>4</sub> is produced from CH<sub>3</sub>-S-CoM, the terminal intermediate of

methanogenesis from all substrates (McBride and Wolfe 1971). In addition, CoM-SH also acts as an electron donor for the fumarate reductase reaction in *Mtb*. *thermoautotrophicus* (Bobik and Wolfe 1989). Previously believed to be restricted exclusively to the methanogenic *Archaea*, CoM-SH also plays an important role in the carboxylation of epoxypropane in the Gram-negative bacterium *Xanthobacter* strain Py2, where it acts as a C<sub>3</sub> carrier (Balch *et al.* 1979, Allen *et al.* 1999). Several analogs of CoM have been synthesized, including 2-bromoethanesulfonic acid which acts as a methyl-CoM reductase inhibitor (Gunsalus *et al.* 1978).

Methyl-H<sub>4</sub>MPT:CoM methyltransferase, characterized in *Mtb. marburgensis* and *Ms. mazei* Gö1, is an integral membrane-bound complex that contains a corrinoid cofactor (5'-hydroxybenzimidazolyl cobamide) (Gärtner *et al.* 1993, Becher *et al.* 1992b). Transfer of CH<sub>3</sub> to CoM-SH involves two steps. In the first step, the reduced corrinoid cofactor (Co(I)) accepts the methyl group from methyl-H<sub>4</sub>MPT; in the second reaction the methyl group is transferred to CoM-SH, a step that is dependent on the presence of Na<sup>+</sup> (Gärtner *et al.* 1994, Weiss *et al.* 1994). The latter step results in the generation of a transmembrane electrochemical Na<sup>+</sup> gradient (ΔμNa<sup>+</sup>); studies with the purified methyl-transferase reconstituted into liposomes catalyzed an electrogenic Na<sup>+</sup> transport with a stoichiometric ratio of 1.7 mol Na<sup>+</sup> per mol methyl-H<sub>4</sub>MPT demethylated (Schäfer *et al.* 1999). Inhibition of methyl-H<sub>4</sub>MPT:CoM methyltransferase activity results in the inhibition of Na<sup>+</sup> transport, indicating that methyl-H<sub>4</sub>MPT:CoM methyltransferase is a primary Na<sup>+</sup> pump (Becher *et al.* 1992a). The ΔμNa<sup>+</sup> generated is used to drive the reduction of CO<sub>2</sub> to formyl-MF (Shima *et al.* 2002).

ATP synthases uses protonmotive force across the membrane to drive the synthesis of ATP from ADP and Pi, but some organisms also synthesize ATP using sodium-motive force (Junge *et al.* 1997). There has been some experimental evidence suggesting that the Na<sup>+</sup> gradient generated may be used to produce ATP in *Mc. voltae*, *Mtb. thermoautotrophicus* ΔH and *Ms. mazei* Gö1 (Dybas and Konisky 1992, Becher and Müller 1994, Šmigán *et al.* 1994), such that ATP production in methanoarchaea may be mediated by transmembrane chemiosmotic gradients of proton and/or sodium ions (Chen and Konisky 1993)

Extensive studies have been conducted on the methyl-H<sub>4</sub>MPT:CoM methyltransferase from *Mtb. marburgensis*, which contains eight non-identical subunits (34 (MtrH), 28 (MtrE), 24 (MtrC), 23 (MtrA), 21 (MtrD), 13 (MtrG), 12.5 (MtrB), and 12 (MtrF) kDa) encoded by *mtrEDCBAFGH* (Harms *et al.*1995). The *Mtr* operon is located between the methyl-CoM reductase operon (*mcr*) and a downstream open reading frame that is predicted to encode a Na<sup>+</sup>/K<sup>+</sup> exchanger (Ferry 1999). The 23 kDa subunit MtrA contains a corrinoid and is believed to be partially associated with the cytoplasmic face of the cell membrane, with the corrinoid-binding site protruding into the cytoplasm. Na<sup>+</sup> translocation is believed to be coupled to conformational changes in MtrA upon methylation, leading to the methyl group being released and transferred to CoM-SH, forming CH<sub>3</sub>-S-COM (Harms *et al.* 1995). The genes encoding the eight subunits from the methyl-H<sub>4</sub>MPT:CoM methyltransferase from *Ms. mazei* Göl have also been cloned and sequenced (Lienard and Gottschalk 1998).

CH<sub>3</sub>-S-CoM is reduced to produce methane, requiring the activity of **methyl-**CoM reductase to catalyze the reduction of the methyl group with electrons from 7-

mercaptoheptanoyl-O-phospho-L-threonine (HTP-SH, **CoB-SH**) (Ellermann *et al.* 1987, 1988, Thauer *et al.* 1993). In the process, free methane is released and the heterodisulfide complex CoM-S-S-CoB is formed ( $\Delta G^{\circ}$ ' = -45 Kj/mol) (Figure 1.10).

Figure 1.10. Reduction of CH<sub>3</sub>-S-CoM with CoB-SH and formation of CH<sub>4</sub> and CoM-S-S-CoB

The methyl-CoM reductase has been purified from a number of methanoarchaea, but has been studied in greatest detail from Mtb. thermoautotrophicus and Mtb. marburgensis (Ferry 1999). This enzyme consists of 3 subunits,  $\alpha$ , (65 ± 3 kDa),  $\beta$  (46 ±2 kDa), and  $\gamma$  (35 ± 5 kDa), in a  $\alpha_2\beta_2\gamma_2$  arrangement (300 kDa) (Rospert et al. 1990). Methyl-CoM reductase contains two molecules of coenzyme  $F_{430}$ , a yellow nickel porphinoid that is tightly bound to the enzyme (Ellefson et al. 1982).

Two isoforms of the reductase (MCR I and MCR II) were purified from *Mtb. thermoautotrophicus* and *Mtb. marburgensis*, which differ only by a 5 kDa difference in the γ-subunit: there are also differences in the N-terminus amino acid sequences of the subunits in the respective isoforms (Rospert *et al.* 1990). MCR II is predominantly found when cells are grown in an excess of H<sub>2</sub> and CO<sub>2</sub>, while MCR I is

found in greater amounts when the gas supply becomes limited (Rospert et al. 1990, Bonacker et al. 1993).

The major energy conserving step in methanogenesis has long been associated with the reductive cleavage of CoM-S-S-CoB formed during the release of methane from methyl-CoM. CoM-S-S-CoB is an important intermediate since it is a substrate for the ultimate energy conservation reaction step catalyzed by the **heterodisulfide reductase** (Hdr) (Shima *et al.* 2002). It had been speculated that the reduction of methyl-CoM with CoB-SH was the site of energy conservation ( $\Delta G^{\circ}$ '=-45 kJ/mol), but this has since been abandoned (Blaut *et al.* 1987, Thauer *et al.* 1993).

Hdr catalyzes the reductive cleavage of CoM-S-S-CoB back to the free cofactors CoM-SH and CoB-SH ( $\Delta G^{\circ}$ )=-40 kJ/mol) (Thauer *et al.* 1993). During growth on H<sub>2</sub>/CO<sub>2</sub>, the source of reducing equivalents is provided by H<sub>2</sub>, *via* the F<sub>420</sub>-nonreducing hydrogenase (*aka* methyl viologen (MV)-reducing hydrogenase) (Hedderich and Thauer 1988). Studies with *Mtb. thermoautotrophicus* have demonstrated that the free energy change ranges from -55 to -80 kJ/mol, depending on the dissolved hydrogen partial pressure (de Poorter *et al.* 2003). Hdr is part of a membrane-associated H<sub>2</sub>:hetero-disulfide oxidoreductase complex that is found in most hydrogenotrophic methanoarchaea, as well as methanoarchaea capable of growth on CH<sub>3</sub>OH or acetate (Heiden *et al.* 1993, Deppenmeier *et al.* 1991, Simianu *et al.* 1998). Electrons extracted by the F<sub>420</sub>-nonreducing hydrogenase are transported to Hdr and then to CoM-S-S-CoB, restoring pools of CoM-SH and CoB-SH (Hedderich *et al.* 1990). The reduction of CoM-S-S-CoB by the H<sub>2</sub>:hetero-disulfide oxidoreductase complex generates an energy conserving transmembrane electrochemical proton gradient ( $\Delta \tilde{u}H^+$ ); studies with

Ms. barkeri demonstrate that the transmembrane proton gradient leads to ATP synthesis from ADP + Pi, via the  $A_1A_0$  ATP synthase (Müller  $et\ al.\ 1999$ , Deppenmeier 2002).

Hdr from *Mtb. marburgensis* was isolated as part of a complex consisting of six subunits: 80, 51, 41, 36, 21, and 17 kDa: the 80, 36, and 21 kDa subunits comprise Hdr, while the 51, 41, and 17 kDa subunits form an F<sub>420</sub>-nonreducing hydrogenase (Mvh) (Hedderich *et al.* 1990, Setzke *et al.* 1994). The H<sub>2</sub>:heterodisulfide oxidoreductase complex was found to be loosely bound to the cell membrane, and the bulk of the activity was isolated from the soluble fraction (Hedderich *et al.* 1994, Setzke *et al.* 1994, Deppenmeier 2004). Hedderich *et al.* (1994) described the *Hdr* operon, which contains 3 genes; *HdrA*, *HdrB*, and *HdrC* encode the 72.2, 33.5, and 21.8 kDa subunits, respectively. HdrA contains a flavin binding site, while the HdrC subunit contains FeS clusters (Hedderich *et al.* 1994, Stojanowic *et al.* 2003). HdrB, which contains the active site for CoM-S-S-CoB reduction, is believed to be the membrane anchor for the complex, although transmembrane helices typically observed in integral membrane proteins are not part of the secondary structure of HdrB (Fricke *et al.* 2006).

The  $H_2$ :heterodisulfide oxidoreductase complex of Ms. barkeri also consists of an  $F_{420}$ -nonreducing hydrogenase and Hdr; unlike the complex isolated from Mtb. thermoautotrophicus, the  $H_2$ :heterodisulfide oxidoreductase complex of Ms. barkeri was found to be tightly associated to the cell membrane (Heiden  $et\ al$ . 1993). This multienzyme complex includes the electron carrier cytochrome b, which wasn't found in the Mtb. thermoautotrophicus complex (Heiden  $et\ al$ . 1993). Hdr from Ms. barkeri is a heterodimer consisting of 46 and 23 kDa subunits; the 23 kDa subunit contains cytochrome B, and the 46 kDa subunit was predicted to be the active site for CoM-S-S-

CoB reduction (Heiden et al. 1994). Künkel et al. (1997) reported the discovery of the MbHdrDE operon encoding the Hdr in Ms. barkeri, where MbHdrD encodes the 46 kDa subunit, and MbHdrE encodes the 23 kDa cytochrome subunit. MbHdrD was found to be equivalent to the MtHdrBC subunits, indicating that the flavin component isolated from Mtb. thermautotrophicus is not essential for CoM-S-S-CoB reduction (Künkel et al. 1997).

Studies with *Ms. mazei* Gö1 indicate that methanophenazine may be an intermediate between the F<sub>420</sub>-nonreducing hydrogenase and Hdr; this scenario is not observed in methanoarchaea outside of the *Methanosarcinaceae* (Abken *et al.* 1998, Ide *et al.* 1999).

### 1.3.2. The methylotrophic pathway

Approximately 26% of methanoarchaea are capable of using the methylotrophic pathway (Garcia 1998). In marine, hypersaline, and other high sulfate environments, methylotrophic metabolism predominates over H<sub>2</sub> and acetate utilization, since sulfate-reducers are able to utilize H<sub>2</sub> more readily (Ollivier *et al.* 1994). Methyl amines are formed through the degradation of osmoregulatory amines, such as choline, creatine, and betaine, or the bacterial reduction of trimethyl oxide, a common metabolite and excretory product of marine animals (Neil *et al.* 1978, Hippe *et al.* 1979, Strøm *et al.* 1979, Oremland *et al.*1982). CH<sub>3</sub>OH is a substrate used by methylotrophic methanoarchaea, though it is not a major precursor for CH<sub>4</sub> in anaerobic environments; CH<sub>3</sub>OH is produced *via* bacterial degradation of lignin and pectin (Donnelly *et al.* 1980, Oremland *et al.*1982).

Methanogenesis with CH<sub>3</sub>OH occurs via the following pathways:

$$4\text{CH}_3\text{OH} \rightarrow 3\text{CH}_4 + \text{CO}_2 + 2\text{ H}_2\text{O}, \Delta G^{\circ}$$
'=-105 kJ/mol CH<sub>4</sub> (CH<sub>3</sub>OH disproportionation)

or

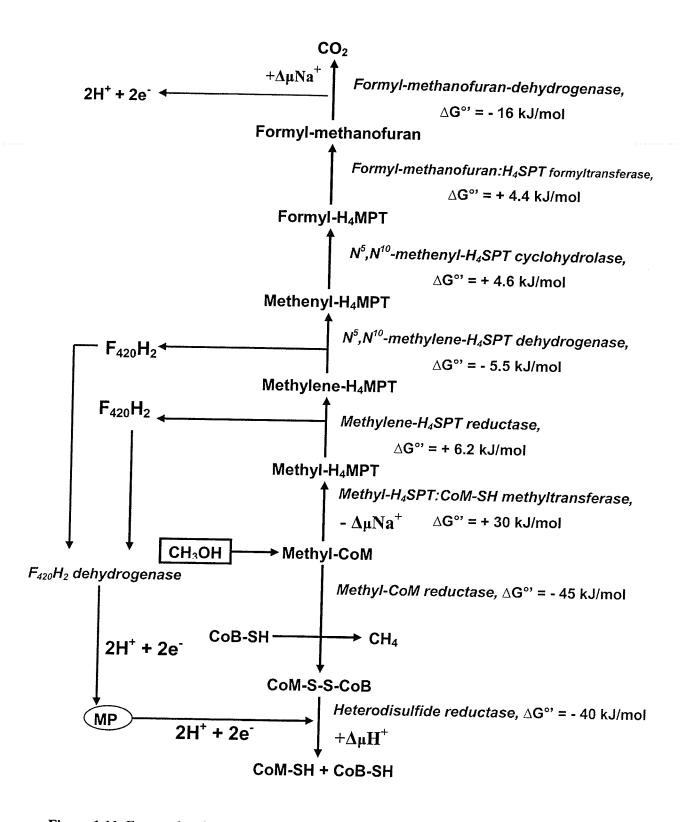
CH<sub>3</sub>OH + H<sub>2</sub> 
$$\rightarrow$$
 CH<sub>4</sub> + H<sub>2</sub>O,  $\Delta$ G°'=-113 kJ/mol (reduction of CH<sub>3</sub>OH with H<sub>2</sub>)

Methylamines are also metabolized in a similar manner to CH<sub>3</sub>OH disproportionation (Hippe *et al.* 1979, Keltjens and Vogels 1993):

$$4\text{CH}_3\text{NH}_3\text{Cl} + 2\text{H}_2\text{O} \rightarrow 3\text{CH}_4 + \text{CO}_2 + 4\text{NH}_4\text{Cl}, \Delta G^{\circ}$$
'=-76.7 kJ/mol 
$$2(\text{CH}_3)_2\text{NH}_2\text{Cl} + 2\text{H}_2\text{O} \rightarrow 3\text{CH}_4 + \text{CO}_2 + 2\text{NH}_4\text{Cl}, \Delta G^{\circ}$$
'=-74.8 kJ/mol 
$$4(\text{CH}_3)_3\text{NHCl} + 6\text{H}_2\text{O} \rightarrow 9\text{CH}_4 + 3\text{CO}_2 + 4\text{NH}_4\text{Cl}, \Delta G^{\circ}$$
'=-75.8 kJ/mol

CH<sub>3</sub>OH is transformed into methane in a series of oxidation/reduction reactions by which the initial steps differ from the metabolism of CO<sub>2</sub> and H<sub>2</sub> (Zeikus *et al.* 1985). The process of CH<sub>3</sub>OH disproportionation can be subdivided into two separate parts, one oxidizing and one reducing. In the oxidizing part of the pathway, one mole of CH<sub>3</sub>OH is oxidized to CO<sub>2</sub> in order to provide reducing equivalents that are used in the reducing part of the methylotrophic pathway to reduce the remaining 3 moles of CH<sub>3</sub>OH to methane. An overview of CH<sub>3</sub>OH disproportionation is given in Figure 1.11.

Throughout the CH<sub>3</sub>OH oxidation process, the enzymes of the hydrogenotrophic pathway are used, but in the reverse direction. The metabolism of CH<sub>3</sub>OH is initiated



**Figure 1.11**. Expected pathway of methanogenesis in *Methanosarcina* spp. growing on methanol (Keltjens and Vogels 1993, Deppemeier *et al.*1996, Deppenmeier 2004).

by two methyl transferases, as described below (van der Meijden et al. 1983, Zeikus et al. 1985).

The methanol :coenzyme M methyltransferase system (MtaABC), comprise a two enzyme system, MT1 (MtaBC) and MT2 (MtaA). The first enzyme in this system is the methanol:corrinoid methyltransferase (MT1), which autocatalyzes the transfer of the CH<sub>3</sub>OH methyl group to its corrinoid cofactor (Keltjens and Vogels 1993, Sauer *et al.* 1997). MT1 consists of two subunits, MtaB (49 kDa) and the corrinoid-containing MtaC (24 kDa) (Sauer *et al.* 1997). MtaB and MtaC form a tight complex, while MtaA is purified separately (Sauer and Thauer 1998).

The second enzyme, **methyl-corrinoid:CoM methyl transferase** (MT2), catalyzes the transfer of the methyl group from MT1 to coenzyme M, forming methyl-S-CoM (Keltjens and Vogels 1993, Ferry 1999). MT2 consists of only one 36 kDa subunit (MtaA) which does not contain any prosthetic groups. The methyl group of methyl-S-CoM is transferred to tetrahydrosarcinopterin (H<sub>4</sub>SPT), a step leading either to the oxidation of the C<sub>1</sub>-bound intermediate to CO<sub>2</sub> and generation of 2 molecules of F<sub>420</sub>H<sub>2</sub>, or reduction to methane with electrons donated by CoB-SH (Ferry 1999). Two isoforms of MT2 have been reported, MtaA and MtbA (Grahame 1989, Yeliseev *et al.* 1993). MtaA predominates when *Ms. barkeri* is grown on CH<sub>3</sub>OH, while MtaB is produced when *Ms. barkeri* is grown on H<sub>2</sub>/CO<sub>2</sub> or trimethylamine (Harms and Thauer 1996).

As in the hydrogenotrophic pathway,  $CH_3$ -CoM is reduced by the  $CH_3$ -CoM reductase using CoB-SH as electron donor. Free  $CH_4$  is released and CoM-S-S-CoB is formed. In methylotrophic methanoarchaea such as  $Ms.\ mazei$  Gö1, a membrane-bound  $F_{420}H_2$ :heterodisulfide oxidoreductase complex, consisting of the enzymes  $F_{420}H_2$ 

dehydrogenase (Fpo complex) and Hdr, catalyzes the  $F_{420}H_2$ -dependent reduction of CoM-S-S-CoB to regenerate CoM-SH and CoB-SH (Deppenmeier 2002, 2004).  $F_{420}H_2$  is generated *via* methyl-group oxidation (Deppenmeier *et al.* 1996). Methanophenazine (MP) is believed to be the intermediate linking electron transfer between the enzymes  $F_{420}H_2$  dehydrogenase and Hdr (Abken *et al.* 1998, Bäumer *et al.* 1998, Deppenmeier *et al.* 1999). Reduction of methanophenazine and CoM-S-S-CoB generates  $\Delta \tilde{u}H^+$  that is used to drive the synthesis of ATP (Deppenmeier 2002, 2004).

Msph. stadtmanae, Msph. cuniculi, and Methanomicrococcus blatticola can also produce methane via the reduction of CH<sub>3</sub>OH with H<sub>2</sub> (M. blatticola can also reduce methyl-amines), requiring only the terminal reductive steps of the hydrogenotrophic pathway following methyl transfer to CoM-SH (Miller and Wolin 1985, Biavati et al. 1988, Sprenger et al. 2000). Unlike the CH<sub>3</sub>OH -metabolizing methanoarchaea from the Methanosarcinaceae, cytochromes are not found in Msph. stadtmanae, Msph. cuniculi, or M. blatticola (Fricke et al. 2006). M. blatticola cannot reduce CO<sub>2</sub> to methyl-CoM, nor can it or produce CO<sub>2</sub> from CH<sub>3</sub>OH (Sprenger et al. 2005); it has not been reported whether the undetected components of the pathway are present only in quantities sufficient for cellular biosynthesis.

The recently elucidated genome of *Msph. stadtmanae* indicates that all of the coding sequences for the enzymes required for CO<sub>2</sub> reduction to CH<sub>3</sub>-CoM and CH<sub>3</sub>OH oxidation to CO<sub>2</sub> are present, with the exception of the genes for the biosynthesis of molybdopterin, an essential component of the formyl-methanofuran dehydrogenase (Fricke *et al.* 2006); indeed, Lin and Sparling (1995, 1998) showed the *Msph. stadtmanae* were able to produce limited amounts of methane using H<sub>2</sub> and serine, which indicates a

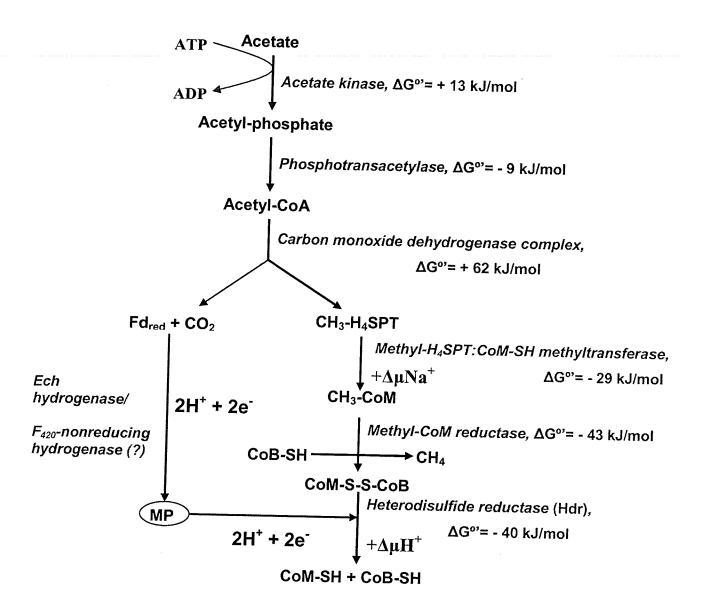
requirement for H<sub>4</sub>MPT and methylene-H<sub>4</sub>MPT reductase. Low activities of the CO<sub>2</sub> reduction/ CH<sub>3</sub>OH oxidation enzymes have been detected in cell extracts, consistent with a requirement of this pathway for biosynthesis (Schwörer and Thauer 1991, van de Wijngaard *et al.* 1991, Fricke *et al.* 2006).

### 1.3.3. The aceticlastic pathway

Under natural physiological conditions, acetate is the most important methanogenic substrate as it is the ultimate endproduct of many fermentative pathways (Gottschalk 1979, Zinder 1993). Only about 11% of the methanoarchaea are capable of metabolizing acetate, but this substrate, produced by homoacetogenic microorganisms, accounts for the majority (~ two/thirds) of the methane produced under natural and manmade conditions (Jetten *et al.* 1992, Ferry 1993, Garcia 1998). During aceticlastic growth, acetate is both a carbon and electron source for methanogenesis. The methyl group is reduced to methane, while the electrons generated from the oxidation of the CO moiety are used for the reduction of CoM-S-S-CoB.

The breakdown of acetate by aceticlastic methanoarchaea is essential in the maintenance of pH homeostasis in anaerobic environments, preventing the acidification of the environment (Jetten et al.1992, Zinder 1993). The fermentation of acetate to methane by Methanosarcina spp. proceeds via the following pathway (Figure 1.12), and is coupled to the smallest change of free energy of all methanogenic substrates (Schäfer et al. 1999):

$$\text{CH}_3\text{COO}^- + \text{H}^+ \rightarrow \text{CH}_4 + \text{CO}_2$$
,  $\Delta \text{G}^{\circ}$ '=-36 kJ/mol CH<sub>4</sub>



**Figure 1.12**. Pathway of methanogenesis from acetate for members of the *Methanosarcinaceae* (Ferry 1993, Deppenmeier *et al.* 1996, Deppenmeier 2004).

Approximately 60-70% of the methane produced in nature results from the aceticlastic pathway by *Methanosarcina* and *Methanosaeta* spp. (formerly known as *Methanothrix*) (Patel 1984). *Methanosarcina* spp. have a higher growth rate relative to *Methanosaeta*, but *Methanosaeta* spp. have a higher affinity for acetate and can be found in environments with lower concentrations of acetate (Jetten *et al.* 1992).

Methanogenesis from acetate by *Ms. barkeri* begins with its activation to acetyl-CoA via acetate kinase and phosphotransacetylase prior to cleavage by the CO dehydrogenase/acetyl-CoA synthase complex (Ferry 1993, Ferry 1999):

 $\text{CH}_3\text{COO}^- + \text{ATP} + \text{CoA} \rightarrow \text{CH}_3\text{CO-S-CoA} + \text{ADP} + \text{Pi}$ ,  $\Delta G^{\circ}$ '= +4.0 kJ/mol

In *Methanosaeta* species, **acetate thiokinase** (*aka* acetyl-CoA synthetase) performs a similar step (Teh and Zinder 1992, Kemner and Zeikus 1994):

 $\text{CH}_3\text{COO}$  + CoA + ATP  $\rightarrow$   $\text{CH}_3\text{CO-S-CoA}$  + AMP + PPi ,  $\Delta G^{\circ}$  = -6.0 kJ/mol

The central enzyme in the pathway of methanogenesis from acetate is **carbon** monoxide dehydrogenase (CODH) (Ferry 1993, Ferry 1999). CODH is actually part of a dual function system that also includes the **acetyl-CoA synthase** (ACS), forming the CODH/ACS complex. The primary function of this complex is the formation or cleavage of the C-C and C-S bonds of acetyl-CoA, but is commonly referred to as CODH because the enzyme complex also oxidizes CO to CO<sub>2</sub> (Ferry 1993). Acetyl-CoA is cleaved into methyl- and carbonyl- groups *via* CODH.

There have been extensive studies of the CODH/ACS complex from Ms. thermophila (Simianu et al. 1998). The subunits of the CODH/ACS complex are encoded by the cdh operon, transcription of which is regulated in response to the growth substrate (Simianu et al. 1998). The complex consists of five subunits ( $\alpha$  (CdhA),  $\beta$ 

(CdhC), γ (CdhE), δ (CdhD), ε (CdhB)) resolved into two enzyme components (Ferry 1999). This includes a 200 kDa CO-oxidizing Ni/Fe-S component containing the α- and ε-subunits, and a 100 kDa corrinoid/iron-sulfur component containing the δ-and γ-subunits (Ferry 1999). The β-subunit is unstable and has not yet been characterized (Ferry 1999). There are three metal clusters, A, B, and C, found within the nickel/iron-sulfur (Ni/Fe-S) component: Cluster A is believed to be responsible for cleavage of acetyl-CoA, while cluster C is the proposed site for the oxidation of CO, with electrons transferred to and from cluster C via cluster B (Ferry 1999). Likewise, the CODH/ACS complex of Ms. barkeri also consists of a five subunit complex with biochemical properties that are identical to that of Ms. thermophila. The CODH/ACS complex from Methanosaeta soehngenii and Methanosarcina frisia have similar properties as Ms. thermophila, but lack cluster A (Ferry 1999).

A mechanism for acetyl-CoA cleavage has been proposed by Ferry (1999). In this scheme, acetyl-CoA binds to the CdhC subunit, where the C-S bond is cleaved, followed by transfer of the acetyl-group to the nickel atom of cluster A in CdhA. C-C bond cleavage then occurs at cluster A and the methyl group is then transferred to the CdhE subunit. Finally, the CO-group is released to cluster C of CdhA where it is oxidized to CO<sub>2</sub>.

Following cleavage of acetyl-CoA by the CODH/ACS complex, the carbonyl group is oxidized by CODH to produce CO<sub>2</sub> and reducing equivalents, which are transferred to CoM-S-S-CoB *via* the electron carrier ferredoxin (Fd). Coenzyme F<sub>420</sub> is not an electron carrier for CODH, but may be involved during the oxidation of the methyl group of acetate to CO<sub>2</sub> to provide electrons for reductive biosynthesis (Ferry 1999).

Meanwhile, the methyl group is transferred to  $H_4SPT$ , and then to CoM-SH, generating energy conserving  $\Delta\mu Na^+$  (Ferry 1993, 1999). The final steps are similar to those described for the hydrogenotrophic and methylotrophic pathways, specifically the production of methane, formation of CoM-S-S-CoB and regeneration of CoM-SH and CoB-SH. As with the other pathways, reduction of CoM-S-S-CoB leads to the generation of an energy conserving  $\Delta\mu H^+$ .

The CO:heterodisulfide oxidoreductase system, studied in *Methanosarcina* spp., executes the reduction of CoM-S-S-CoB *via* reduced ferredoxin (Fd<sub>red</sub>), and a membrane-bound Fd<sub>red</sub> oxidizing complex, including methanophenazine, that transfers the electrons to Hdr (Deppenmeier *et al.* 1996, Schäfer *et al.* 1999, Deppenmeier 2004). The Ech hydrogenase produces H<sub>2</sub> from Fd<sub>red</sub>, while the F<sub>420</sub>-nonreducing hydrogenase and Hdr catalyze CoM-S-S-CoB reduction; it has also been suggested that the Ech alone might catalyze the oxidation of Fd<sub>red</sub> and transfer of electrons to Hdr without the F<sub>420</sub>-nonreducing hydrogenase (Kemner and Zeikus 1994b, Deppenmeier 2002). Methanophenazine (MP) is expected to be the electronic link between the hydrogenase(s) and the Hdr (Deppenmeier 2002, 2004). This scenario is unlikely in *Methanosaeta concilii*, which contains low levels of hydrogenase activities (Sowers *et al.* 1984, Patel 1984, Deppenmeier *et al.* 1996).

Methanosarcina acetivorans lacks the Ech hydrogenase, but instead contains a multisubunit membrane bound protein complex, the Ms. acetivorans Rhodobacter nitrogen fixation complex (Ma-Rnf) (Li et al. 2006). Ma-Rnf is believed to be an electron transferring complex, interacting with reduced ferredoxin and transferring the electrons to methanophenazine. As is the case for other RNF proteins, this complex is

also hypothesized as being an ion pump, translocating either protons or sodium ions through the membrane (Li et al. 2006).

Under physiological conditions, where the environmental concentrations of acetate are low (0.5-5 mM), the catabolic reactions yield very small amounts of energy, often not enough to form 1 ATP from ADP; the process itself also requires an investment of 1 ATP to activate acetate prior to degradation (Jetten *et al.* 1992). Additional means to generate energy must be present in aceticlastic methanoarchaea.

One prospect is the energy conserved *via* cleavage of the pyrophosphate (PPi) bond. PPi is formed by *Methanosaeta* spp. when ATP is hydrolysed to AMP and PPi, during activation of acetate to acetyl-CoA (Kohler and Zehnder 1984), and also during various biosynthetic processes, including DNA/RNA synthesis, amino acid activation, polysaccharide synthesis, and fatty acyl-CoA formation (Maeshima 2000). Jetten *et al.* (1992) reported the presence of membrane-bound pyrophosphatase in *Methanosaeta* similar to proton translocating pyrophosphatases found in plant vacuoles and phototrophic bacteria, possibly linking the hydrolysis of PPi to energy conservation.

Bäumer *et al.* (2002) reported the presence of genes encoding two separate proton-translocating pyrophosphatases in *Ms. mazei* Gö1. Using inverted membrane vesicles from CH<sub>3</sub>OH -grown cells, it was shown that hydrolysis of 1 mol of PPi was coupled to translocation of 1 mol of protons across the membrane.

# 1.4. Elements of electron transport in the methanoarchaea - Electron carriers

A number of electron carriers are present in methanoarchaea, ranging from cytochromes and methanophenazine, which are found exclusively in members of the

*Methanosarcinaceae* (Garcia *et al.* 2000), to electron carriers such as coenzyme  $F_{420}$  or NADP<sup>+</sup> which are found in all methanoarchaea (Schäefer *et al.* 1999). In the following sections, the electron carriers Coenzyme  $F_{420}$ , cytochrome, ferredoxin, and methanophenazine will be discussed.

#### 1.4.1. Coenzyme $F_{420}$

8-hydroxy-5-deazaflavin (coenzyme F<sub>420</sub>, Figure 1.13) was originally isolated from methanoarchaea, but similar compounds have also been found in sulfate-reducing *Archaea*, halophilic *Archaea*, various *Streptomyces* species, *Mycobacterium tuberculosis*, *Nocardia aurantia*, and some eukaryotes (Daniels *et al.* 1984, Lin and White 1986, Shima *et al.* 2002).

F<sub>420</sub> was first isolated from *Methanobacterium* strain MoH by Cheeseman *et al.* (1972), and later identified as the N-(N-L-lactyl-γ-L-glutamyl)-L-glutamic acid phosphodiester of 7,6-didemethyl-8-hydroxy-5-deazariboflavin 5'phosphate by Eirich *et al.* (1978), who characterized coenzyme F<sub>420</sub> as an intensely yellow-blue-green fluorescent compound present in all of the methanoarchaea tested (Lin and White 1986, White and Zhou 1993). While not exclusive to the methanoarchaea, F<sub>420</sub> acts as an important indicator as a molecular marker for the preliminary screening and identification of the presence of methanoarchaea because of its (generally) high cellular abundance (up to 100 mg F<sub>420</sub>/kg of cells) and its intense fluorescence (Eirich *et al.* 1979, Fox *et al.* 1987, Peck and Archer 1989, Zhou and White 1993). The reduced form does not exhibit fluorescence (DiMarco 1990). Heine-Dobbernack *et al.* (1988) demonstrated that F<sub>420</sub>

could be used as a parameter to estimate the biomass of methanoarchaea, using Mb. bryantii and Ms. barkeri as test subjects.

 $F_{420}$  is the major redox carrier in methanoarchaea (DiMarco 1990).  $F_{420}$  is functionally equivalent to NAD; it is an obligate hydride (H<sup>-</sup>) acceptor/donor and its redox potential is ~ -350 mV (-320 mV for NAD) (Jacobson and Walsh 1984).

Figure 1.13. Oxidized and reduced forms of coenzyme  $F_{\rm 420}$ 

Coenzyme  $F_{420}$  also acts as a sensor for the presence of  $O_2$ . Spectral analysis by Schönheit *et al.* (1981) observed that the majority of  $F_{420}$  (>70%) was lost after a culture of *Mtb. thermoautotrophicus* was exposed to air for 50 minutes.  $F_{420}$  is converted to factor 390 ( $F_{390}$ ) upon exposure to oxygen: AMP or GMP couples to the 8-OH group of the deazaflavin to form derivatives which show an absorbance maximum at 390 nm (Kiener *et al.* 1988, van de Wijngaard and van der Drift 1991). Studies by Gloss and Hausinger (1988) indicated that  $O_2$  was the only trigger for  $F_{390}$  formation, and that the addition of heavy metals or methanogenesis inhibitors did not lead to  $F_{390}$  synthesis. Accumulation of  $F_{390}$  leads to a stoppage of cell metabolic activities, since the active substrate  $F_{420}$  would not be accessible as substrate for the various methanogenic enzymes (Hausinger *et al.* 1985). Once anaerobic conditions are restored,  $F_{390}$  is then converted

back to  $F_{420}$  by the  $F_{390}$  hydrolase (Vermeij *et al.* 1985, Gloss and Hausinger 1988, Kengen *et al.* 1991).

In methylotrophic methanoarchaea, reduced F<sub>420</sub> (F<sub>420</sub>H<sub>2</sub>) plays a significant role in the F<sub>420</sub>H<sub>2</sub>:heterodisulfide oxidoreductase system, using the reversed CO<sub>2</sub> reduction pathway to oxidize one mol of CH<sub>3</sub>OH (or methylamine) to provide F<sub>420</sub>H<sub>2</sub> for the reduction of CH<sub>3</sub>OH /methylamine to methane (Keltjens and Vogels 1993, Deppemeier *et al.* 1996). In hydrogenotrophic methanoarchaea, methanogenesis from CO<sub>2</sub> and H<sub>2</sub> involves two F<sub>420</sub>H<sub>2</sub>-dependent reduction reactions: N<sup>5</sup>,N<sup>10</sup>-methenyl-H<sub>4</sub>MPT to N<sup>5</sup>,N<sup>10</sup>methylene-H<sub>4</sub>MPT, and N<sup>5</sup>,N10-methylene-H<sub>4</sub>MPT to N<sup>5</sup>-methyl-H<sub>4</sub>MPT (Shima *et al.* 2002). Hydrogenotrophic methanoarchaea reduce coenzyme F<sub>420</sub> via the H<sub>2</sub>-dependent F<sub>420</sub>-reducing hydrogenase (Thauer *et al.* 1993). Under Ni<sup>+</sup>-limiting conditions some methanoarchaea may use the tandem activities of the F<sub>420</sub>H<sub>2</sub>- and H<sub>2</sub>-dependent N<sup>5</sup>,N<sup>10</sup>-methylene-H<sub>4</sub>MPT dehydrogenases to produce F<sub>420</sub>H<sub>2</sub>, when the F<sub>420</sub>-reducing hydrogenase is inoperative (Ferry 1999).

Coenzyme  $F_{420}$  is also used to generate NADPH for reductive biosynthetic metabolism via the  $F_{420}H_2$ :NADPH oxidoreductase (Tzeng et~al.~1975, Kell et~al.~1980, Yamazaki et~al.~1980, Elias et~al.~2000).  $F_{420}$  appears to be essential for photorepair (reduction of thymine dimers) of UV-damaged DNA in Mtb.~thermoautotrophicus (Kiener et~al.~1985).  $F_{420}H_2$  is also used to detoxify  $O_2$  by way of the recently reported  $F_{420}H_2$  oxidase from Methanobrevibacter~arboriphilus, which catalyzes the oxidation of two  $F_{420}H_2$  with one  $O_2$ , to two  $F_{420}$  and two  $H_2O$  (Seedorf et~al.~2004).

Methanoarchaea that use ethanol or isopropanol to reduce  $CO_2$  to methane  $\emph{via}$   $NADP^+$ -dependent alcohol dehydrogenase also possess an  $F_{420}$ -dependent  $NADP^+$ 

reductase that is able to oxidize NADPH +  $H^+$  to provide electrons for the reduction of  $F_{420}$  (Berk and Thauer 1997):

2 NADPH + H<sup>+</sup> + F<sub>420</sub> 
$$\rightarrow$$
 2 NADP<sup>+</sup> + 2 F<sub>420</sub>H<sub>2</sub>

NADPH-dependent F<sub>420</sub>-reduction has been observed in *Methanogenium* organophilum and *Methanobacterium palustre*, both of whom possess the NADP-dependent alchohol dehydrogenase (Bleicher and Winter 1991, Berk and Thauer 1997).

 $F_{420}$  can be reduced from a number of sources ( $H_2$ , NADPH, etc), and can be used as a source of reductant to for many compounds (CoM-S-S-CoB,  $O_2$ , NADP<sup>+</sup>, etc), making it a versatile, but not universal electron carrier.

#### 1.4.2. Cytochrome

Cytochromes are proteins with iron-containing porphyrin ring (heme) prosthetic groups attached to them. A variety of cytochromes of the *b*- and *c*-type have been identified in methylotrophic members of the family *Methanosarcinaceae*: the presence of this particular electron carrier is restricted solely to members of this family (Kuhn *et al.* 1983, Kamlage and Blaut 1992, Müller *et al.* 1993, Deppenmeier *et al.* 1996). Heiden *et al.* (1993, 1994) reported that cytochrome *b* was purified as part of the H<sub>2</sub>:heterodisulfide oxidoreductase complex of *Ms. barkeri* (cytochrome *b2*, HdrE). Cytochrome *b* was reduced by H<sub>2</sub> (*via* F<sub>420</sub>-nonreducing hydrogenase) and oxidized by CoM-S-S-CoB *via* Hdr, although the reaction rate was too quick to be resolved (Heiden *et al.* 1993). Experiments with *Ms. mazei* Gö1, which contains two *b*-type (midpoint potentials: -135 and -240 mV) and two *c*-type (midpoint potentials: -140 and -230 mV) cytochromes,

have illustrated that cytochromes are reduced by  $F_{420}H_2$  and also oxidized by CoM-S-S-CoB rapidly (Kamlage and Blaut 1992, Schäfer *et al.* 1999).

The two operons encoding F<sub>420</sub>-nonreducing hydrogenase (*vhoGAC/vhtGAC*) also contain genes encoding b-type cytochromes (*vhoC/vhtC*); expression studies have concluded that the F<sub>420</sub>-nonreducing hydrogenase and cytochrome *b1* participate as part of the H<sub>2</sub>:heterodisulfide oxidoreductase system; in contrast, the F<sub>420</sub>-nonreducing hydrogenases from methanoarchaea outside of the *Methanosarcinaceae* do not encode genes encoding for cytochrome(Deppenmeier *et al.* 1996), suggesting other means for electron transport.

#### 1.4.3. Ferredoxin

Electron-transporting iron-sulfur proteins called ferredoxins, as well as polyferredoxin (12[Fe-4S]), have been isolated from members representing all the methanogenic orders. Ferredoxins generally mediate electron transfer between enzymatic systems (Moura et al. 1982). Ferredoxin is required for methanogenesis when acetate is used as substrate, where reduced ferredoxin acts an electron acceptor for the Ech hydrogenase, and is used for the reduction of CoM-S-S-CoB (Deppenmeier 2002). In Ms. barkeri Fusaro, reduced ferredoxin (formed by the Ech hydrogenase from H<sub>2</sub> by sodium-motive force) is also a source of electrons for CO<sub>2</sub> fixation to formyl-methanofuran in the initial CO<sub>2</sub> reduction stage, during growth on H<sub>2</sub> and CO<sub>2</sub> (Meuer et al. 2002, Deppenmeier 2002).

Reeve et al. (1989) reported that polyferredoxin was encoded as part of the mvh (mvhB) operon of Mtb. thermoautotrophicus  $\Delta H$ ; this gene has also been conserved

Methanothermus fervidus (Steigerwald et al. 1990). Polyferredoxin contains low electron accepting activity, so its function as the physiological electron acceptor of the methylviologen-reducing hydrogenase is questionable (Schäfer et al. 1999). mvhB is not encoded by Msph. stadtmanae, suggesting that the polyferredoxin may not be essential for activity of the F<sub>420</sub>-nonreducing hydrogenase (Fricke et al. 2006).

### 1.4.4. Methanophenazine

Phenazines, discovered in the late nineteenth century, are naturally occurring pigments formed by bacteria, such as the *Pseudomonas* and *Streptomyces* species, which are excreted into the medium (Ingram and Blackwood year, Turner and Messenger 1986). Many of the phenazine compounds are secondary metabolites which have antibiotic activity, and participate in redox reactions under *in vitro* conditions (Turner and Messenger 1986). Non-natural phenazines are synthesized as analogues of the natural bacterial product and tested as bactericides, fungicides, herbicides, and anti-tumour agents (Turner and Messenger 1986). There are approximately 100 different naturally occurring phenazines, of which all but one are of bacterial origin (Beifuss and Tietze 2005).

A low molecular weight redox active ( $-165 \pm 6 \text{ mV}$ ) compound identified as methanophenazine was purified from the membranes of CH<sub>3</sub>OH -grown *Ms. mazei* Göl (Figure 1.14) (Abken *et al.* 1998, Beifuss and Tietze 2005). This was the first phenazine compound to be reported from an archaeon, and the first phenazine derivative involved in electron transport of biological systems (Beifus *et al.* 2000, Deppenmeier 2002). Methanophenazine is similar in function to menaquinone, redox active pigments found in

the cell membranes of bacteria, such as *Clostridium thermoaceticum* and *Clostridium thermoautotrophicum*, as well as the sulfate-reducing archaeon *Archaeoglobus fulgidus* (Das et al. 1989, Brüggemann et al. 2000).

Figure 1.14. Structure and reactivity of methanophenazine (Beifuss et al. 2000).

This highly hydrophobic compound is expected to be the link between the  $F_{420}H_2$  dehydrogenase complex and the Hdr, playing an important role in membrane-bound electron transport (Abken *et al.* 1998, Beifuss *et al.* 2000). To date, methanophenzine has not been purified from any other methanoarchaeon outside of the *Methanosarcinaceae* (Beifuss and Tietze 2005). The genes for the synthesis of this cofactor have not been identified.

Because of the extremely hydrophobic character of methanophenazine, and also because of the low yields and time consuming extraction process, the water soluble phenazine derivative 2- hydroxy-phenazine (2-OH-phenazine, -191  $\pm$  8 mV) was chemically synthesized and used in subsequent redox experiments (Beifuss *et al.* 2000, Beifuss and Tietze 2005). 2-OH-phenazine is reduced by the  $F_{420}H_2$  dehydrogenase and by the  $F_{420}$ -nonreducing hydrogenase of *Ms. mazei* Gö1 (Abken *et al.* 1998):

$$F_{420}H_2 + 2\text{-OH-Phen} \rightarrow F_{420} + 2\text{-OH-Phen}H_2$$
  
 $H_2 + 2\text{-OH-Phen} \rightarrow 2\text{-OH-Phen}H_2$ 

It is clear that phenazine compounds can receive and transfer electrons with some of the key enzymes in the methanogenesis pathways in various members of the *Methanosarcinaceae*. Furthermore, the reduced form, dihydro-2-OH-phenazine, acts as a source of electrons for the reductive cleavage of CoM-S-S-CoB (-177  $\pm$  5 mV), via Hdr (Abken et al. 1998, Ide 1999, Murakami et al. 2001):

Reduced phenazine has recently been used as an *in vitro* electron donor for the reduction of CoM-S-S-CoB in *M. blatticola*; phenazine was reduced by membrane-bound hydrogenase activity (Sprenger *et al.* 2005). It is not clear whether methanophenazine is present in this methanoarchaeon.

# 1.5. Elements of electron transport - Redox enzymes

## 1.5.1. Hydrogenases

Hydrogen is a key substrate for methanogenic *Archaea*, providing reducing equivalents for biosynthetic reactions as well as catabolic methane formation, with various hydrogenases being central for this metabolism (Kemner and Zeikus 1994). The existence of this enzyme was discovered by Stephenson and Strickland (1931) who observed the biological activation of H<sub>2</sub> in *E. coli*. Hydrogenases are found in a wide variety of microorganisms and catalyze the simplest chemical reaction in nature:

 $H_2 \leftrightarrow 2H^+ + 2e^-$  (Hedderich 2004). Based on the metal groups found in the various hydrogenases, there are three classes of hydrogenases: [NiFe]-hydrogenases; [Fe]-hydrogenases; and the FeS-free hydrogenases (Vignais *et al.* 2001).

Two roles have been described for hydrogenases in the metabolism of anaerobic bacteria. Fermentative bacteria use hydrogenases to produce H<sub>2</sub> via oxidation of reduced electron carriers during fermentation, allowing the carriers to be recycled for the continuous production of ATP via substrate-level phosphorylation (Adams et al. 1981). Secondly, H<sub>2</sub> is also a source of electrons for the reduction of various terminal electron acceptors, such as sulfate, nitrate, CO<sub>2</sub>, or fumarate, generating ATP via membrane-bound electron transport chains (Adams et al. 1981).

Methanoarchaea possess various hydrogenase activities that function in the reduction of C<sub>1</sub>-moieties to methane, reduction of CoM-S-S-CoB to CoM-SH and CoB-SH, and CO<sub>2</sub> fixation to methanofuran. Most methanogenic hydrogenases are [NiFe] hydrogenases containing the two metals in a bimetallic reaction center involved in the H<sub>2</sub> cleavage reaction, with the exception being the H<sub>2</sub>-dependent methylene-H<sub>4</sub>MPT dehydrogenase (discussed in section 1.3.1.) (Albracht 1994, Sorgenfrei *et al.* 1997, Vignais *et al.* 2001). Hydrogenases that will be discussed in detail include: the coenzyme F<sub>420</sub>-reducing hydrogenase; the coenzyme F<sub>420</sub>-nonreducing (*aka* methyl viologen-reducing) hydrogenase; and the Ech hydrogenase (energy conserving hydrogenase). Each hydrogenase is specialized in its capacity to transfer electrons from H<sub>2</sub> to different electron carriers/acceptors in the methanogenic pathways, with some also being able to transfer from sources other than H<sub>2</sub>.

# 1.5.1.1. $F_{420}$ -reducing hydrogenase

The coenzyme F<sub>420</sub>-reducing hydrogenase has been purified and characterized from a number of methanoarchaea, including Ms. barkeri strains MS and Fusaro, Methanospirillum (Msp.) hungatei GP1, Mtb. thermoautotrophicus, Mtb. marburgensis, Mc. voltae, Mc. jannaschi, Methanococcus (Mc.) vannielii, and Mb. formicicum (Yamazaki 1982, Jacobson et al. 1982, Fauque et al. 1984, Fox et al. 1987, Muth et al. 1987, Sprott et al. 1987, Baron et al. 1989, Fiebig and Friedrich 1989, Shah and Clark 1990, Choquet and Sprott 1991, Michel et al. 1995). F<sub>420</sub>-reducing hydrogenase appears to be universal in all hydrogenotrophic methanoarchaea, with the apparent exceptions of Methanosphaera cuniculi and Methanobacterium strain G2R (McKellar and Sprott 1979, Biavati et al. 1988). This enzyme, which is reversibly inactivated by O2, catalyzes the  $H_2$ -dependent reduction of both coenzyme  $F_{420}$  and synthetic electron acceptors such as methyl viologen. Immunogold labeling studies indicate that this enzyme is located on the inner face of the cytoplasmic membrane (Baron et al. 1989b, Lünsdorf et al. 1991, Braks et al. 1994). While the enzyme has hydrophobic character, the  $F_{420}$ -reducing hydrogenase is not an integral membrane protein, as the sequences encoding for the subunits do not encode the twin arginine signal sequence required for traffic of proteins into the cell membrane (Vignais et al. 2001).

The enzyme is composed of three different subunits,  $\alpha$ ,  $\beta$ , and  $\gamma$ , with the largest subunit ( $\alpha$ ) containing the active [NiFe]-site (Künkel *et al.* 1998, Vaupel and Thauer 1998, Vignais *et al.* 2001). The  $\beta$ -subunit binds FAD, which acts as a 1 e-/2 e- switch during electron transfer from F<sub>420</sub>H<sub>2</sub>, which donates a hydride ion, to the single-electron carrying Fe-S clusters (Thauer *et al.* 1993). The  $\gamma$ -subunit contains FeS clusters which

mediate the transfer of electrons between the  $\alpha$ - and  $\beta$ -subunits (Thauer *et al.* 1993, Vignais *et al.* 2001).

Electrophoresis of the purified  $F_{420}$ -reducing hydrogenase from the respective methanoarchaea on non-denaturing native PAGE gels results in the separation of 2-3 distinct bands, 1-2 large molecular weight bands at 500 kDa -1300 kDa, and a lower molecular weight band of 89 kDa – 115 kDa (Sprott *et al.* 1987, Michel *et al.* 1995). Activity staining using  $F_{420}$  or methyl viologen reveals that both bands possess hydrogenase activity. The smaller band represents the minimal functional subunit (containing  $\alpha$ ,  $\beta$ , and  $\gamma$  subunits) capable of  $F_{420}$ -reduction. The large band is an aggregate containing a number of the functional subunits.

The  $F_{420}$ -reducing hydrogenase of the hydrogenotrophic methanoarchaeon Mtb. thermoautotrophicus, has been well studied (Jacobson et~al. 1982, Jacobson et~al. 1987, Livingston et~al. 1987). This enzyme, encoded by the frhADGB cluster, can be assayed via the reduction of viologen dyes or coenzyme  $F_{420}$  with  $H_2$  (Alex et~al. 1990, Braks et~al. 1994). Activity stained native PAGE gels indicate that there are two distinct protein bands with  $F_{420}$ -reducing activity, consistent with observations of the homologous protein in other methanoarchaea; a high molecular weight band of approximately 800 kDa, and a lower molecular weight band of 115 kDa (Fox et~al.1987). Like most typical  $F_{420}$ -reducing hydrogenases, the enzyme consists of three distinct subunits;  $\alpha$  (FrhA, 47 kDa),  $\beta$  (FrhB, 31 kDa),  $\gamma$  (FrhG, 26 kDa); (Fox et~al.1987). A fourth subunit is also encoded by the frh operon,  $\delta$  (FrhD, 18 kDa); this subunit, found in the homologous gene clusters in other methanoarchaea, does not copurify with the holoenzyme and is expected to encode a protease involved with hydrogenase maturation (Ficke et~al. 2006).

Immunogold labeling studies of a related methanoarchaeon, Mtb. marburgensis, indicated that the majority of the  $F_{420}$ -reducing hydrogenase (90%) was associated with the membrane, bound to the cytoplasmic side with 10-20 kDa linker particles (Braks  $et\ al.$  1994).

The properties of the  $F_{420}$ -reducing hydrogenases are fairly consistent amongst the methanoarchaea, but there are some variations between different genera, and even between strains of the same species.

Mc. voltae possesses two different isoenzymes (Fru and Frc) that differ in that an active-site cysteine is replaced with a selenocysteine (Fru)(Halboth and Klein 1992, Sorgenfrei et al. 1997). The genes encoding the selenium-free hydrogenase (Frc) are expressed in the absence of selenium (Sorgenfrei et al. 1997). Based on information derived from its genome, Mc. jannaschii also possesses two F<sub>420</sub>-reducing isoenzymes (Künkel et al. 1998).

Similarly, the genome of Ms. barkeri Fusaro also encodes two  $F_{420}$ -reducing hydrogenases (frhADGB and freAEGB), although neither isoenzyme contains selenium (Vaupel and Thauer 1998)). Both operons are transcribed during growth on  $H_2$  and  $CO_2$  or  $CH_3OH$ , but not during growth on acetate. While expression of the  $F_{420}$ -reducing hydrogenase is not necessary for  $F_{420}$ -reduction during methylotrophic growth, Ms. barkeri Fusaro produces significant levels of the  $F_{420}$ -reducing hydrogenase, comparable to levels during growth on  $H_2$  and  $CO_2$  (Mukhopadhyay  $et\ al.\ 1993$ ). The physiological significance of the  $F_{420}$ -reducing hydrogenase of Ms. barkeri Fusaro during methylotrophic growth is unclear.

While most of the known F<sub>420</sub>-reducing hydrogenase holoenzymes consist of three separate subunits, the F<sub>420</sub>-reducing hydrogenase from *Ms. barkeri* strain MS consists of one 60 kDa subunit (overall molecular mass 800 kDa): the corresponding enzyme from *Ms. barkeri* strain Fusaro consists of three subunits (kDa) (Fauque *et al.* 1984, Michel *et al.* 1995). Thus, significant variations can exist even within a species.

Immunological studies using antibodies against the purified F<sub>420</sub>-reducing hydrogenase from *Msp. hungatei* GP1 only reacted against the cell extract from *Msp. hungatei* JF1 and *Ms. barkeri* strain MS: no reaction was observed against the extracts of *Methanobacterium* strain G2R, *Mc. voltae*, *Mb. formicicum*, *Ms. mazei* Gö1, two strains each of *Mtb. thermoautotrophicus* and *Mb. bryantii*, *Methanobrevibacter arboriphilus*, or *Methanosaeta concilii* (Sprott *et al.* 1987). These differences may reflect the evolutionary divergence among the methanoarchaea (Sprott *et al.* 1987).

## 1.5.1.2. $F_{420}$ -nonreducing hydrogenase

The F<sub>420</sub>-nonreducing hydrogenase does not interact with coenzyme F<sub>420</sub>, but has been assayed using artificial electron acceptors such as methyl viologen (MV) or benzyl viologen (BV). The F<sub>420</sub>-nonreducing hydrogenase is a [NiFe] hydrogenase that is also reversibly inactivated by O<sub>2</sub> (Thauer *et al.* 1993). This enzyme has been purified from a number of hydrogenotrophic methanoarchaea, including *Mtb. thermoautotrophicus*, *Mb. formicicum*, *Mb. fervidus*, *Mc. jannaschii*, and *Mc. voltae*, as well as from the methylotrophic methanoarchaea, *Ms. barkeri* MS and *Ms. mazei* Gö1 (Jin *et al.* 1983, Adams *et al.* 1986, Shah and Clark 1990, Hedderich *et al.* 1990, Deppenmeier *et al.* 1992, Kemner and Zeikus 1994, Deppenmeier *et al.* 1995). Interestingly, the F<sub>420</sub>-

nonreducing hydrogenase apparently is not found in the hydrogenotrophic methanoarchaeon *Msp. hungatei* GP1 (Sprott *et al.* 1987). As such, the mechanism by which this methanoarchaeon reduces CoM-S-S-CoB is unresolved.

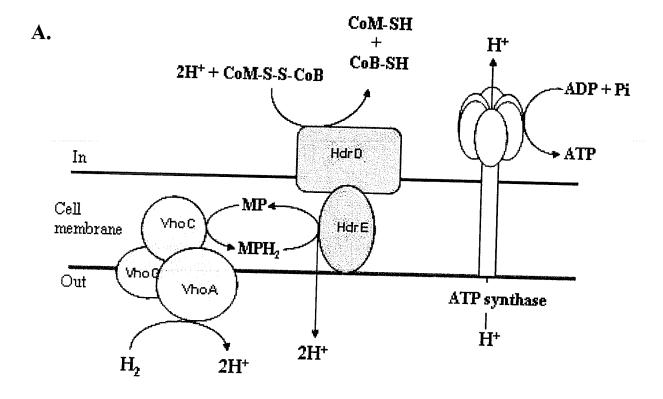
In the hydrogenotrophic methanoarchaea outside of the *Methanosarcinales*, the F<sub>420</sub>-nonreducing hydrogenase has primarily been detected in the cytoplasm after cell breakage, whereas the enzyme isolated from methylotrophic methanoarchaea of the *Methanosarcinales* has been located as an integral membrane protein (Deppenmeier *et al.* 1992, Vignais *et al.* 2001).

The F<sub>420</sub>-nonreducing hydrogenase of Mtb. thermoautotrophicus is encoded by mvhDGAB (Reeve et al. 1989). The MvhA ( $\alpha$ ) (53 kDa) and MvhG ( $\gamma$ ) (33 kDa) subunits form the basic hydrogenase unit typical of all [NiFe] hydrogenases; MvhD ( $\delta$ ) (15.8 kDa) may be the contact site of the hydrogenase for the Hdr (Thauer et al. 1993, Vignais et al. 2001, Stojanowic et al. 2003). MvhB, a 44 kDa subunit, is predicted to be a polyferredoxin (Reeve et al. 1989, Steigerwald et al. 1990). Interestingly, MvhB is not encoded by the mvh operon of Msph. stadtmanae, indicating that the MvhB subunit isn't essential for hydrogenase activity (Fricke et al. 2006).

The  $F_{420}$ -nonreducing hydrogenase has been purified from the cell membranes of  $Ms.\ barkeri$  strain MS and  $Ms.\ mazei$  Gö1 (Deppenmeier  $et\ al.\ 1992$ , Kemner and Zeikus 1994). The enzyme from  $Ms.\ barkeri$  is composed of two subunits, 57 and 35 kDa, with an overall molecular weight estimated as 98 kDa (Kemner and Zeikus 1994). Cytochrome b was reduced in the presence of  $H_2$ , but ferredoxin,  $F_{420}$ , FAD/FMN, and  $NAD(P)^+$  were not. The  $F_{420}$ -nonreducing hydrogenase of  $Ms.\ mazei$  Gö1 has a similar structure, with subunits of 60 and 40 kDa, and a molecular weight between 77 and 79

kDa (Deppenmeier *et al.* 1992). Two separate operons, *vhoGAC* and *vht GAC* encode two nearly identical isoforms of F<sub>420</sub>-nonreducing hydrogenase (92-97% amino acid identity) (Deppenmeier *et al.* 1995). *vhoG/vhtG* and *vhoA/vhtA* encode the 40 and 60 kDa subunits, respectively, while *vhoC/vhtC* encode *b*-type cytochromes; additionally, a twin arginine signal motif (RRxFxK) is encoded at the 5'ends of the *vhoG* and *vhtG* genes from *Ms. mazei* Gö1, which explains the localization of the F<sub>420</sub>-nonreducing hydrogenase to the periplasmic face of the cell membrane (Deppenmeier *et al.* 1996, Vignais *et al.* 2001). F<sub>420</sub>-nonreducing hydrogenase I, encoded by *vhoGAC*, has been purified from cells grown in CH<sub>3</sub>OH, H<sub>2</sub>:CO<sub>2</sub>, or acetate, whereas hydrogenase II, encoded by *vht GAC*, has only been isolated in cells grown in CH<sub>3</sub>OH or H<sub>2</sub>:CO<sub>2</sub> (Deppenmeier *et al.* 1996). As such, it was concluded that hydrogenase I is likely part of the H<sub>2</sub>:heterodisulfide oxidoreductase system as described by Deppenmeier *et al.* (1991), while hydrogenase II is probably involved in electron flow to CO<sub>2</sub> in the course of the formyl-MF dehydrogenase reaction (Schäfer *et al.* 1999).

Abken et al. (1998) have proposed a model in which the Vho hydrogenase extracts electrons from H<sub>2</sub>, which then progress to the Hdr, via methanophenazine; this comprises the proposed H<sub>2</sub>:heterodisulfide oxidoreductase complex in Ms. mazei Göl and for other members of the Methanosarcina spp. (Figure 1.15A) (Ide et al. 1999, Deppenmeier et al. 1999, Deppenmier 2002, 2004). The oxidation of H<sub>2</sub>/reduction of methanophenazine (MP) by Vho, and the oxidation of reduced methanophenazine (MPH<sub>2</sub>)/reduction of CoM-S-S-CoB, generates ΔμH<sup>+</sup> used to synthesize ATP (Deppenmeier et al. 2002, 2004).



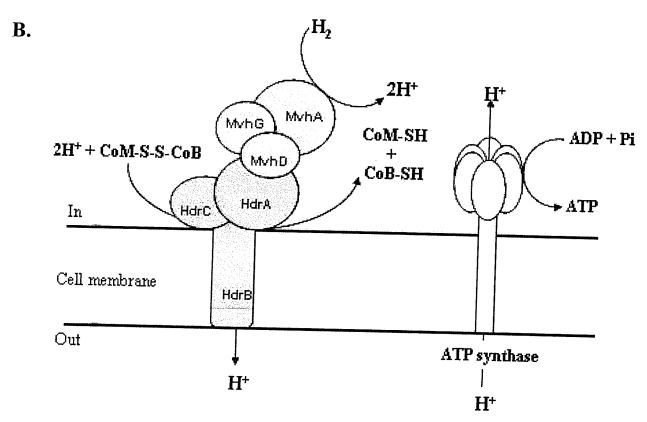


Figure 1.15. Model of electron transport and energy conservation via the H<sub>2</sub>:CoM-S-S-CoB oxidoreductase complex in (A) the *Methanosarcinales*, and (B) the *Methanobacteriales* (Ide et al. 1999, Deppenmeier et al. 2002, 2004, Setzke et al. 1994, Stojanowic et al. 2003, Fricke et al. 2006). MP, methanophenazine. MPH<sub>2</sub>, reduced methanophenazine.

Until recently, the physiological electron acceptor for the F<sub>420</sub>-nonreducing hydrogenase of methanoarchaea outside of the *Methanosarcinales* was not entirely clear. The best studied system amongst the hydrogenotrophic methanoarchaea has been described using *Mtb. thermoautotrophicus* as a model, which contains several proteins that may participate as electron acceptors for the F<sub>420</sub>-nonreducing hydrogenase. These include: polyferredoxin; a flavoprotein (flavoprotein A) that copurifies with the H<sub>2</sub>:heterodisulfide oxidoreductase complex; and possibly rubredoxin (Nölling *et al.* 1995b). Neither polyferredoxin nor flavoprotein A were effectively reduced by the purified hydrogenase (Nölling *et al.* 1995b).

Recent models indicate that there are likely no intermediates between the  $F_{420}$ -nonreducing hydrogenase and Hdr, with the  $F_{420}$ -nonreducing hydrogenase docking directly onto the Hdr. The  $F_{420}$ -nonreducing hydrogenase (Mvh) and Hdr are isolated together as a large, tightly bound complex from the soluble cytoplasmic fraction of *Mtb. thermoautotrophicus* (Figure 1.15B) (Setzke *et al.* 1994, Stojanowic *et al.* 2003). Electron transfer from Mvh to Hdr is likely facilitated at the contact site of the MvhD subunit to HdrA; in the respective genomes of *Methanosarcina* spp. and *Msph. stadtmanae*, an *mvhD* homologue is fused to the 5' end of an *hdrA* homologue, indicating a potential link between the  $F_{420}$ -nonreducing hydrogenase and Hdr without the need for intermediates, although this has not been shown experimentally (Fricke *et al.* 2006).

### 1.5.1.3. Ech hydrogenase

 $H_2$ -evolving respiratory hydrogenases are membrane-bound, multimeric hydrogenases couple the oxidation of a carbonyl group (from formate, acetate, or carbon

monoxide), with the reduction of 2H<sup>+</sup> to produce H<sub>2</sub> (Vignais *et al.* 2001). For example, the *E. coli* hydrogenase 3 is a [NiFe] hydrogenase produced under strictly anaerobic conditions and part of the formate hydrogen lyase (FHL) complex (Künkel *et al.* 1998, Vignais *et al.* 2001). This enzyme, coupled with the formate dehydrogenase, couples the oxidation of formate to CO<sub>2</sub> with the concomitant production of H<sub>2</sub>. The FHL complex is encoded by eight genes in the hyc operon (hycBCDEFG and fdhF), where hycGE encode the hydrogenase portion of the complex: HycE contains the Ni-binding motif found in all [NiFe] hydrogenases (Sauter *et al.* 1992, Hedderich 2004). The remaining hyc-encoded subunits function as membrane anchors and electron carriers between formate dehydrogenase and hydrogenase 3 (Vignais *et al.* 2001). A similar hydrogenase system has been described for *Rhodospirillum rubrum*, where the hydrogenase activity is carbon monoxide-induced: the hydrogenase operates in tandem with the CODH to oxidize CO and to produce H<sub>2</sub> (Fox *et al.* 1996a, b).

Similar hydrogenases have been described from *Ms. barkeri* strain Fusaro, *Mtb. thermoautotrophicus* and *Mtb. marburgensis* (Meuer *et al.* 1999, Tersteegen and Hedderich 1999, Hedderich 2004). The enzyme was purified from the membranes of acetate-grown *Ms. barkeri* and given the designation Ech hydrogenase (energy-conserving hydrogenase) (Künkel *et al.* 1998, Meuer *et al.* 1999). The Ech hydrogenase from *Ms. barkeri* is encoded by a six gene operon, *echABCDEF*. The hydrogenase large subunit, EchE, shares very little sequence similarity with the respective subunits from the F<sub>420</sub>-reducing and F<sub>420</sub>-non reducing hydrogenases, with the exception of the [NiFe]-binding motifs. The hydrogenase small subunit EchC is also poorly conserved, with similarities restricted to an N-terminus cysteine cluster involved in the formation of a

[4Fe-4S] cluster (Künkel et al. 1998, Hedderich 2004). The integral membrane subunits of the Ech hydrogenase have closely related counterparts to the NADH:quinone oxidoreductase (Complex I), and functions as an ion pump with a role in energy-conservation during H<sub>2</sub> evolution, or reverse electron transport during H<sub>2</sub> uptake/reduction of ferredoxin (Fd) (Meuer et al. 2002, Hedderich 2004). Ech provides the cell with reduced Fd that is required as electron donors for the various oxidoreductase activities in the catabolic and anabolic pathways (Meuer et al. 2002).

There have been several hypotheses put forth regarding the physiological importance of Ech in methanoarchaea during growth on the various methanogenic substrates. Anti-Ech serum was used to detect the presence of Ech in the cell extract of *Ms. thermophila*, indicating a potentially essential function of Ech in the acetate metabolism of *Methanosarcina* species (Meuer *et al.* 1999). Cell suspensions of *Ms. barkeri* mutants (Δech) could not grow on acetate; this was attributed to the mutants' inability to catalyze the oxidation of CO to CO<sub>2</sub> and H<sub>2</sub> (Meuer *et al.* 2002). During acetoclastic methanogenesis, it is believed that Ech catalyzes Fd-dependent H<sub>2</sub>-evolution (Meuer *et al.* 2002, Hedderich 2004). The CO-group of acetyl-CoA is oxidized to CO<sub>2</sub>, with the electrons transferred to Fd; Fd-H<sub>2</sub> is then used to provide electrons, in the form of H<sub>2</sub>, for the reduction of CoM-S-S-CoB (Deppenmeier 2004).

During growth on  $H_2$  and  $CO_2$ , it is suggested that Ech couples the consumption of an ion gradient with ferredoxin reduction, thus allowing for a thermodynamically unfavorable reaction to proceed (Meuer *et al.* 1999, 2002, Deppenmeier 2004). Meuer *et al.* (2002) reported that cell suspensions of  $\Delta$ ech *Ms. barkeri* mutants could not grow on  $H_2$  and  $CO_2$ , and that cell extracts from the mutants could not catalyze the reduction of

FD, the electron donor for the reduction of CO<sub>2</sub> to formyl-methanofuran. The genome of *Ms. acetivorans* indicates that this methanoarchaeon does not encode an Ech; this was one explanation as to why this *Ms. acetivorans* could not grow using H<sub>2</sub> and CO<sub>2</sub> (Galagan *et al.* 2002).

Mtb. thermoautotrophicus encodes two Ech-type hydrogenases, Eha and Ehb; based on the abundance of transcripts produced, Ehb is likely involved with catabolic reactions, while Eha is involved with anabolic reactions (Tersteegen and Hedderich 1999). Tersteegen and Hedderich (1999) hypothesized that the function of these hydrogenase complexes may be to act as multisubunit integral membrane proteins that form large electron transfer complexes, driving the reduction of CO<sub>2</sub> to formyl-methanofuran, or aiding in the synthesis of cellular components such as pyruvate, and 2-oxoglutarate.

Analysis of the respective genomes for *Mc. jannaschii*, *M. kandleri*, *Methanococcus maripaludis*, and *Msp. hungatei* JF1 indicates that these methanoarchaea also contain open reading frames that encode for the Ech hydrogenase (Bult *et al.* 1996, Slesarev *et al.* 2002, Porat *et al.* 2006). The genome of *Msph. stadtmanae* also encodes an Ehb hydrogenase, but lacks the gene encoding EhA hydrogenase; Ehb is likely involved with pyruvate synthesis in this methanoarchaeon (Fricke *et al.* 2006).

### 1.5.2. The $F_{420}H_2$ dehydrogenase

The  $F_{420}H_2$  dehydrogenase is part of the methylotrophic methanogenesis pathway (Figure 1.11, section 1.3.2.). The  $F_{420}H_2$  dehydrogenase is an essential component of the membrane-bound  $F_{420}H_2$ :heterodisulfide oxidoreductase system in some members of the *Methanosarcinales*, functioning as the entry point for electrons derived from the

oxidation F<sub>420</sub>H<sub>2</sub>. These electrons are then ferried through a membrane-bound electron transport chain that leads to Hdr and ultimately reduction of CoM-S-S-CoB and generation of an energy conserving, transmembrane proton gradient (Müller *et al.* 1993, Bäumer *et al.* 1998). While the F<sub>420</sub>H<sub>2</sub> dehydrogenase plays an important role in the metabolism of methylotrophic methanoarchaea, its existence and function in non-methylotrophic methanoarchaea has yet to be determined. Electron transport and energy conservation has been well studied in the *Methanosarcinales*, particularly with *Ml. tindarius* and *Ms. mazei* Gö1 (Scheel and Schäfer 1990, Deppenmeier *et al.* 1990a, 1990b, Deppenmeier *et al.* 1996, Deppenmeier 2002, 2004).

Deppenmeier et al. (1990a) analyzed the role of  $F_{420}H_2$  in the reduction of CoM-S-S-CoB by the  $F_{420}H_2$  dehydrogenase, using inverted vesicle preparations of Ms. mazei Gö1 and Ml. tindarius. These vesicles coupled  $F_{420}H_2$  oxidation with the reduction of CoM-S-S-CoB in a concomitant and stoichiometeric reaction (Deppenmeier et al. 1990a):

$$CoM\text{-S-S-CoB} + F_{420}H_2 \rightarrow CoM\text{-SH} + CoB\text{-SH} + F_{420}$$

The  $F_{420}H_2$ -dependent reduction of CoM-S-S-CoB was coupled to the generation of a transmembrane proton gradient  $\Delta\mu H^+$ , which led to the formation of ATP from ADP + Pi  $via~A_1A_0$  ATP synthase: 1 mole of ATP was formed per 2-3 moles of  $F_{420}H_2$  oxidized (Deppenmeier *et al.* 1990b).

Thus far, the  $F_{420}H_2$  dehydrogenase has only been purified from *Ml. tindarius* (Haase *et al.* 1992), *Ms. mazei* Gö1 (Abken and Deppenmeier 1997), and the closely related sulfate-reducing archaeon *A. fulgidus*, which may be a methanoarchaeon that has lost the ability to generate methane (Kunow *et al.* 1994, Bapteste *et al.* 2005).

## 1.5.2.1. Purification of the $F_{420}H_2$ dehydrogenase from Methanolobus tindarius

 $F_{420}H_2$  dehydrogenase was first purified by Haase *et al.* (1992) from *Ml. tindarius*. As described by Haase *et al.* (1992), the oxygen-stable enzyme was extracted from the cell membrane, and contains five subunits with apparent molecular masses of 45, 40, 22, 18, and 17 kDa (total molecular mass: 120 kDa). The enzyme contained  $16 \pm 2$  mol iron and  $16 \pm 3$  mol acid-labile sulfur per mole of enzyme. Enzyme activity was assayed using the methyl viologen/metronidazole system. Methyl viologen is reduced *via* electrons donated from  $F_{420}H_2$  and transferred by  $F_{420}H_2$  dehydrogenase; metronidazole reoxidizes the reduced methyl viologen in order to prevent spectroscopic interference with  $F_{420}H_2$ . Curiously, flavins were not detected, although the electron transfer from the hydride donor  $F_{420}H_2$  to the one-electron acceptor requires a 2-electron/1e-electron switch; as well, the iron-sulfur (FeS) clusters themselves are single electron acceptors, thus necessitating the 2-electron/1-electron switch (Abken and Deppenmeier 1997).

Westenberg et al. (1999), using the N-terminal amino acid sequence of the 40 kDa subunit of the  $F_{420}H_2$  dehydrogenase, discovered the gene ffdB in the Ml.tindarius chromosome. It was apparent that the ffdB gene was part of an operon containing three additional reading frames: ffdA, ffdC, and ffdD. The genes were cloned into E. coli and the protein products were examined. While ffdB encoded the 40 kDa subunit, the products from the other genes were not found to be identical to any of the other known subunits, although it has been surmised that ffdC and ffdD may encode two of the smaller subunits (22, 18, or 17 kDa). However, the purified  $F_{420}H_2$  dehydrogenase from Ml.

tindarius consists of five subunits, indicating that there must be one gene that was not discovered.

1.5.2.2. Purification of the  $F_{420}H_2$  dehydrogenase from Methanosarcina mazei Göl

The discovery of F<sub>420</sub>H<sub>2</sub> dehydrogenase in *Ml. tindarius* was followed by the purification of the homologous enzyme in *Ms. mazei* Gö1 by Abken and Deppenmeier (1997). The oxygen-stable holoenzyme (115 kDa) consisted of five different polypeptide subunits with molecular masses of 40, 37, 22, 20, and 16 kDa (Abken and Deppenmeier 1997). The enzyme from *Ms. mazei* Gö1 contains FeS clusters as well as FAD (0.2 mol/mol enzyme). Abken and Deppenmeier (1997) suggested that the low amounts of FAD may be due to the flavin not being covalently bound and that most may have been lost during the process of purification. It was also suggested that much of flavin from the *Ml. tindarius* F<sub>420</sub>H<sub>2</sub> dehydrogenase may have been lost entirely during purification, since FAD was not detected in the enzyme, with the remaining traces of FAD masked by the buffer components. FAD is essential for the dehydrogenase activity as it acts as a two electron/one electron switch during the transfer of electrons from F<sub>420</sub>H<sub>2</sub> to electron acceptor. The physiological electron carrier for the F<sub>420</sub>H<sub>2</sub> dehydrogenase is believed to be methanophenazine (section 1.4.4.)

1.5.2.3. DPI-mediated inhibition of the phenazine-dependent  $F_{420}H_2$  dehydrogenase

The heterodisulfide reductase (Hdr), the key enzyme for energy conservation, can receive electrons from  $FD_{red}$ ,  $H_2$ , or  $F_{420}H_2$  for the reduction of CoM-S-S-CoB (Deppenmeier *et al.* 1996). In methylotrophic methanoarchaea of the

Methanosarcinacaea, methanophenazine is proposed to be the electronic link to Hdr (Bäumer et al. 1998, Ide et al. 1999, Deppenmeier et al. 2002, Deppenmeier 2004). Brodersen et al. (1999) discovered that the compound diphenyleneiodonium (DPI) was specifically able to inhibit both the  $H_2$ - ( $F_{420}$ -nonreducing hydrogenase) and  $F_{420}H_2$ -dependent ( $F_{420}H_2$  dehydrogenase) heterodisulfide oxidoreductase systems. From this study, it was concluded that DPI competitively inhibits the 2-OH-phenazine-dependent  $F_{420}H_2$  dehydrogenase activity, increasing the Km from 35 to 100  $\mu$ M 2-OH-phenazine (Brodersen et al.1999).

Diphenyleneiodonium Chloride

Phenazine

Figure 1.16. Structures of Diphenyleneiodonium (DPI) chloride and phenazine.

A model was proposed that suggests that the inhibition occurs at the site of 2-OH-phenazine reduction (at the two N atoms). The observed inhibition is not surprising since DPI has a chemical structure that is very similar to phenazine; this provided additional support for the role of methanophenazine as an e- carrier in the oxidation of  $F_{420}H_2$ , and ultimately to the reduction of CoM-S-S-CoB *via* Hdr (Figure 1.16).

## 1.5.2.4. $F_{420}H_2$ dehydrogenase as a redox-driven proton pump

Bäumer *et al.* (2000) reported the discovery of the gene locus in a DNA fragment that contained a cluster of genes encoding the F<sub>420</sub>H<sub>2</sub> dehydrogenase from *Ms. mazei*Gö1; the name given to this locus was *fpo* (<u>F</u><sub>420</sub>H<sub>2</sub>:<u>p</u>henazine <u>o</u>xidoreductase). Analysis of the *fpo* genes and their products revealed similarities between the F<sub>420</sub>H<sub>2</sub> dehydrogenase and the genes (*nuoA-N*) encoding the proton-translocating

NADH:ubiquinone oxidoreductase (NDH-1, complex I) from *E. coli*, and to analogous structures found in the mitochondria of eukaryotic organisms (Yagi *et al.* 1998, Bäumer *et al.* 2000). The purified F<sub>420</sub>H<sub>2</sub> dehydrogenase reported by Abken and Deppenmeier (1997) was actually not the holoenzyme as predicted, but the cytoplasmic portion of a much more sophisticated integral membrane structure.

In accordance with the organization of the gene sequences of the *nuo* operon in *E. coli*, the genes of the *fpo* operon were designated *fpoA*, *B*, *C*, *D*, *H*, *I*, *J*, *K*, *L*, *M*, *N*, and *O*; *fpoF*, which encodes the 37 kDa subunit, is found at a different location in the chromosome away from the *fpo* cluster (Bäumer *et al.* 2000). This gene is believed to encode the electron input device of F<sub>420</sub>H<sub>2</sub> dehydrogenase. Overexpression of *fpoF* from *Ms. mazei* Gö1 in *E. coli* produced a flavoprotein containing non-heme iron and acidlabile sulfur; the FpoF subnit was capable of oxidizing F<sub>420</sub>H<sub>2</sub> in the presence of methyl viologen (Bäumer *et al.* 2000).

A model of electron transport in the membranes of *Ms. mazei* Gö1 was proposed by Bäumer *et al.* (2000) (Figure 1.17). The protein originally isolated by Abken and Deppemeier (1997) from *Ms. mazei* Gö1, consisting of FpoB (22 kDa), C (20 kDa), D (40 kDa), I (16 kDa), and F (37 kDa), are clustered together on the cytoplasmic side of

the cell membrane. FpoF is predicted to be the point of entry for electrons from  $F_{420}H_2$ , which are shuttled through FeS clusters to FpoBCDI. FpoBCDI forms the membrane-associated FeS-containing module that transfers electrons to the membrane-integral FpoAHJKLMN component, and finally to methanophenazine. The role of FpoO is not entirely clear, but does not appear to be essential for activity.

This model is supported by the finding of significant identity and similarity between the primary sequences of 11 proteins encoded by *fpoA* to *fpoN*, excluding *fpoF*, with analogous subunits found in bacterial and mitochondrial NADH dehydrogenase (Bäumer *et al.* 2000). The homologues of complex I, found in the respiratory chains of bacteria, the mitochondria of eukaryotes, and chloroplasts of plants, generate a proton motive force for the synthesis of ATP, and also regenerate NAD<sup>+</sup> by removing excesses of reducing equivalents (Yagi *et al.* 1998, Friedrich 1998, Friedrich and Scheide 2000). The electrons are passed to membrane-bound hydrophobic electron carriers called quinones (Friedrich and Weiss 1997).

The Fpo complex is expected to operate in a similar manner, interacting with  $F_{420}H_2$  instead of NADH, and methanophenazine in place of ubiquinone; ubiquinone is not found in methanoarchaea (Bäumer *et al.* 2000, Deppenmeier 2002). It is speculated that the  $F_{420}H_2$  dehydrogenase and the NADH dehydrogenase both evolved from a common ancestral ion-pumping hydrogenase (Friedrich and Scheide 2000, Hedderich 2004). Indeed, there is sequence similarity of the subunits from the [NiFe] hydrogenases (including the  $F_{420}$ -reducing/nonreducing hydrogenases, and the Ech hydrogenase) with the subunits of the *E. coli* complex I and the Fpo complex (Friedrich and Scheide 2000). FpoF does not show homology to any of the subunits of complex I, but does show

In (cytoplasm)

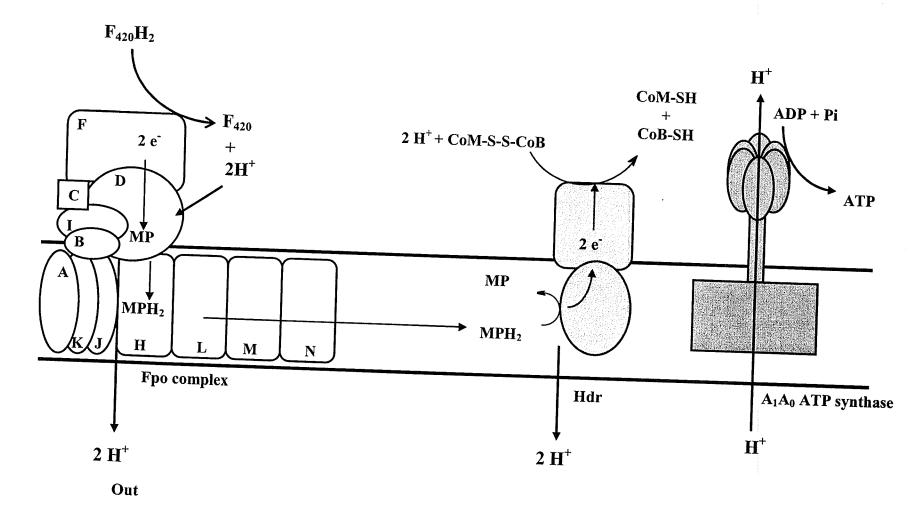


Figure 1.17. Hypothetical model of membrane-bound electron transport and energy conservation in CH<sub>3</sub>OH-grown *Methanosarcina mazei* Gö1 as described in text (Bäumer *et al.* 2000, Deppenmeier *et al.* 2002). Fpo, F<sub>420</sub>H<sub>2</sub> phenazine oxidoreductase. Hdr, heterodisulfide reductase. MP, methanophenazine. MPH<sub>2</sub>, reduced methanophenazine.

homology to the subunits of several methanogenic  $F_{420}$ -dependent enzymes such as the  $F_{420}$ -reducing hydrogenases ( $\beta$ -subunit) and the  $F_{420}$ -dependent formate dehydrogenases, all of which are FAD-containing proteins (Bäumer *et al.* 2000).

Bäumer *et al.* (2000) have also provided experimental evidence that indicates that the activity of the  $F_{420}H_2$  dehydrogenase resembles a proton pump in *Ms. mazei* Gö1. Using inverted membrane vesicles, protons flow into the lumen of the vesicles concurrent with  $F_{420}H_2$ -dependent reduction of 2-OH-phenazine. Two protons are pumped into the lumen during reduction of 2-OH-phenazine *via* Fpo, with another pair of protons translocated during the course of CoM-S-S-CoB reduction. In contrast, 4 protons are pumped out by complex I of *E. coli* (Deppenmeier 2004). A model depicting the activity of the  $F_{420}H_2$  dehydrogenase complex and CoM-S-S-CoB reduction is shown in Figure 1.17.

Several lines of evidence support the homology between the  $F_{420}H_2$  dehydrogenase and the NADH dehydrogenase: both are membrane-bound proteins containing flavins and Fe-S clusters; both  $F_{420}H_2$  (-350 mV) and NADH (-320 mV) are reversible hydride donors with similar midpoint potentials; both proteins use small membrane-bound hydrophobic non-protein electron acceptors (methanophenazine and ubiquinone, respectively); and DPI inhibits  $F_{420}H_2$  dehydrogenase as well as NADH dehydrogenase (Bäumer *et al.* 2000).

### 1.5.2.5. The Fqo complex of Archaeoglobus fulgidus

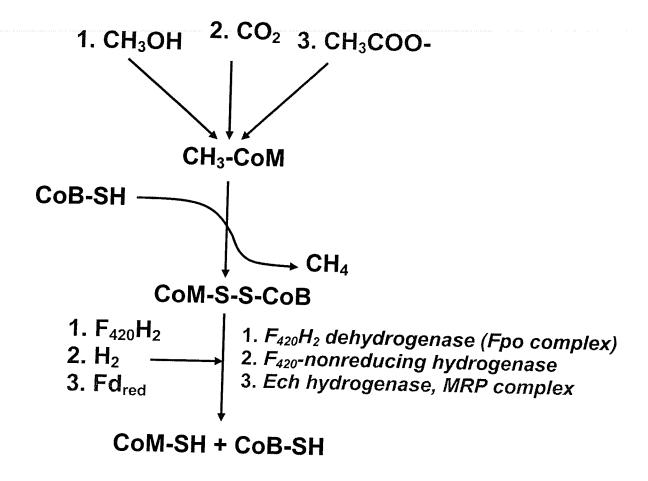
A similar complex was isolated from the membranes of the hyperthermophilic marine archaeon A. fulgidus by Kunow et al. (1994), while Brüggermann et al. (2000)

reported the genes coding for the subunits of the complex. As with the methanogenic homologue, the fqo ( $\underline{F}_{420}H_2$ :quinone oxidoreductase) gene cluster, encompassing 11 genes (A, BC, D, H, I, J, K, L, M, N, and F), shows significant homology to subunits of the prokaryotic and eukaryotic NADH-quinone oxidoreductases (Brüggermann *et al.* 2000). Unlike the fpo operon of Ms. mazei Gö1, the fqo operon contains all of the genes that encode for the Fqo complex, including fqoF. Genes fqoB and fqoC are fused into one gene, whereas the homologous genes in Ms. mazei Gö1 are two separate genes. The structure of the Fqo complex is expected to be similar to the Fpo complex of Ms. mazei Gö1, with FqoBCDIF extruding from the cell membrane into the cytoplasm, and FqoAHJKLMN forming the membrane-integral module (Brüggermann *et al.* 2000). There does not appear to be an analogue for fpoO in the fqo operon.

The Fqo complex oxidizes  $F_{420}H_2$  and likely transfers the electrons to menaquinone, present in the cytoplasmic membrane of A. fulgidus (Brüggermann et al. 2000). Like the analogous component in Ms. mazei Gö1, FqoF is predicted to be the electron input device for the Fqo complex: overexpression of the fqoF gene in E. coli resulted in the production of a 39 kDa protein (containing non-heme iron, acid-labile sulfur, and FAD) that was able to oxidize  $F_{420}$  with the concurrent reduction of methyl viologen (plus metronidazole) (Brüggermann et al. 2000).

## 1.6. $F_{420}H_2$ dehydrogenase activity in other methanoarchaea?

While the methanophenazine-dependent  $F_{420}H_2$  dehydrogenase activity is essential for methylotrophic growth in members of the *Methanosarcinaceae* (Figure 1.18), a similar activity, using methyl viologen as electron acceptor, appears to be



**Figure 1.18**. Methanogenesis, electron transport, and CoM-S-S-CoB reduction in the methanoarchaea. 1. Growth on CH<sub>3</sub>OH. 2. Growth on H<sub>2</sub>:CO<sub>2</sub>. 3. Growth on acetate. (Thauer *et al.* 1993, Shima *et al.* 2002, Deppenmeier 2004, Li *et al.* 2006).

ubiquitous amongst the methanoarchaea, regardless of methanogenic substrate or phylogeny (Wong et al. 1994, Wong PhD thesis 1999). This activity is referred to as F<sub>420</sub>H<sub>2</sub> dehydrogenation, to avoid confusion with the protein purified from the *Ms. mazei* Gö1 and *Ml. tindarius*. The physiological significance of this activity in nonmethylotrophic methanoarchaea is unclear. Based on the studies with *Ml. tindarius* and *Ms. mazei* Gö1, F<sub>420</sub>H<sub>2</sub> dehydrogenase activity does not appear to be essential for methanogenesis on substrates other than CH<sub>3</sub>OH; along with Hdr, the F<sub>420</sub>-nonreducing hydrogenase (H<sub>2</sub>) and the Ech hydrogenase or MRP complex (Fd<sub>red</sub>) should provide the pmf for hydrogenotrophic or acetoclastic methanogenesis, respectively (Figure 1.18). It is currently not known how non-methylotrophic methanoarchaea may couple F<sub>420</sub>H<sub>2</sub> oxidation to CoM-S-S-CoB reduction, or if this activity might be required for some other pathway outside of methanogenesis.

However, the hydrogenotrophic methanoarchaeon *Msp. hungatei* GP1 apparently lacks an F<sub>420</sub>-nonreducing hydrogenase (Sprott *et al.* 1987), which is an important component of the H<sub>2</sub>:heterodisulfide oxidoreductase complex found in hydrogenotrophic methanoarchaea that catalyzes the energy conserving reduction of CoM-S-S-CoB (Thauer *et al.* 1993, Shima *et al.* 2002). In the absence of the F<sub>420</sub>-nonreducing hydrogenase, another mechanism must exist for the reduction of CoM-S-S-CoB. An F<sub>420</sub>H<sub>2</sub> dehydrogenase-like activity has been detected in the cell-free extract of *Msp. hungatei* GP1 (Wong, PhD thesis 1999). Given the close phylogenetic relationship between the *Methanomicrobiales* and *Methanosarcinales* (Bapteste *et al.* 2005), it would not be surprising if an Fpo complex was present in *Msp. hungatei* GP1.

Some methanoarchaea can grow on formate (HCOO<sup>-</sup>) as an energy source (Livingston *et al.* 1987).

$$\text{HCOO}^{-} + \text{F}_{420} \rightarrow \text{CO}_2 + \text{F}_{420}\text{H}_2$$

This reaction is catalyzed by F<sub>420</sub>-dependent formate dehydrogenase, purified from the hydrogenotrophic methanoarchaea *Methanococcus vannielii* (Jones and Stadtman 1980) and *Methanobacterium formicicum* (Schauer and Ferry 1983), and is also found in *Methanococcus voltae* (Brodersen *et al.* 1999b). As F<sub>420</sub>H<sub>2</sub> is the primary carrier of electrons, could it not be used as an alternative to H<sub>2</sub> for CoM-S-S-CoB reduction? Indeed, membrane-bound CoM-S-S-CoB reduction in *Mc. voltae* is coupled to F<sub>420</sub>H<sub>2</sub> oxidation; the source of F<sub>420</sub>H<sub>2</sub> oxidation associated with the F<sub>420</sub>-reducing hydrogenase of this methanoarchaeon (Brodersen *et al.* 1999b).

Ms. barkeri Fusaro is a methylotrophic member of the Methanosarcinaceae, and would be expected to possess a Fpo complex as in Ml. tindarius and Ms. mazei Gö1. Studies by Wong (PhD thesis 1999) indicated that this protein is apparently not produced by Ms. barkeri Fusaro during growth on CH<sub>3</sub>OH, although a similar activity appeared to be associated with an F<sub>420</sub>-reducing hydrogenase activity; in contrast the purified F<sub>420</sub>H<sub>2</sub> dehydrogenases of Ml. tindarius and Ms. mazei Gö1 do not possess F<sub>420</sub>-reducing hydrogenase activity (Haase et al. 1992, Abken and Deppenmeier 1997). Furthermore, significant amounts of F<sub>420</sub>-reducing hydrogenase are produced during growth on CH<sub>3</sub>OH, at levels higher than during growth on H<sub>2</sub>:CO<sub>2</sub> (Mukhopadhyay et al. 1993). High levels of F<sub>420</sub>-reducing hydrogenase would not be necessary during growth on CH<sub>3</sub>OH, as F<sub>420</sub> is produced via CH<sub>3</sub>OH oxidation. Is there a link between the high levels

of  $F_{420}$ -reducing hydrogenase and  $F_{420}H_2$  dehydrogenase activity in this methanoarchaeon?

Msph. stadtmanae, a methanoarchaeon that grows solely on H<sub>2</sub> and CH<sub>3</sub>OH, also possesses an F<sub>420</sub>H<sub>2</sub> dehydrogenation activity, although the source and physiological significance of this activity is unknown (Miller and Wolin 1983, 1985, Wong et al. 1994). F<sub>420</sub>H<sub>2</sub> is not necessary during growth on H<sub>2</sub> and CH<sub>3</sub>OH, but would be required during methanogenesis from H<sub>2</sub> and L-serine, or 2-propanol and CH<sub>3</sub>OH, as well as for anabolic reactions (Wong, Ph.D thesis 1999). An F<sub>420</sub>-reducing hydrogenase was not detected in the cell free extract of Msph. stadtmanae, such that the means for F<sub>420</sub>-reduction in this methanoarchaeon are unclear (Deppenmeier et al. 1989, Wong et al. 1994). It has been suggested that the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of Msph. stadtmanae may be essential for the reduction of F<sub>420</sub>; F<sub>420</sub>-reduction has been observed using cell-free extract of Msph. stadtmanae, phenazine, and H<sub>2</sub>; (Wong PhD thesis 1999). Methanophenazine may be the link between H<sub>2</sub>, the primary source of electrons for Msph. stadtmanae, and the observed F<sub>420</sub>H<sub>2</sub> dehydrogenation activity.

## 2. Research objectives

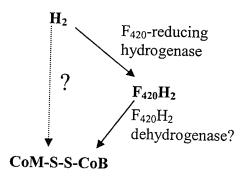
The various electron transport pathways of the methanoarchaea have been well studied, but biochemical and genomic data as presented and discussed in the previous section suggest that there are variations in the electron transport pathways for some methanoarchaea. Thus our laboratory is interested in studying and elucidating these differences and variations in different methanoarchaea, focusing on the role of F<sub>420</sub>H<sub>2</sub> dehydrogenation activity. In some methylotrophic members of the *Methanosarcinaceae*, the F<sub>420</sub>H<sub>2</sub> dehydrogenase plays an important role in electron transport and energy conservation, catalyzing the classic F<sub>420</sub>H<sub>2</sub>-dependent reduction of CoM-S-S-CoB during the final stage of methanogenesis (Deppenmeier 2002, 2004). In *Ms. mazei* Gö1, this activity uses various phenazine-containing compounds as electron carriers/acceptors.

# Objective 1: Determine the extent of phenazine-dependent $F_{420}H_2$ dehydrogenation activity in the methanoarchaea

A similar activity, referred to as  $F_{420}H_2$  dehydrogenation, appears to be ubiquitous in the cell-free extracts of all methanoarchaea. A role for the  $F_{420}H_2$  dehydrogenase activity in non-methylotrophic methanoarchaea containing high levels of  $F_{420}$ -reducing hydrogenase and  $F_{420}$ -nonreducing hydrogenase activities is not readily apparent. However, in the absence of  $F_{420}$ -reducing hydrogenase or  $F_{420}$ -nonreducing hydrogenase, could there be a physiological role for the observed  $F_{420}H_2$  dehydrogenation activity? Could this activity be required for hydrogenotrophic methanoarchaea using electron sources other than  $H_2$ , such as formate or 2-propanol? Could there be an  $F_{420}H_2$ -dependent mechanism for CoM-S-S-CoB reduction? These questions were examined via

study of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activities of *Methanospirillum* hungatei GP1, *Methanosarcina barkeri* Fusaro, and *Methanosphaera stadtmanae*.

Objective 2: Purify and characterize the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of Methanospirillum hungatei GP1

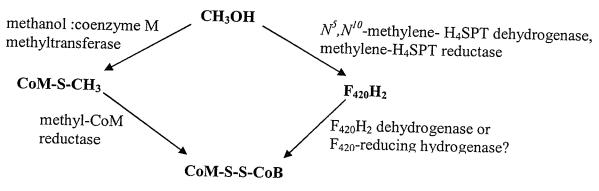


**Figure 2.1**. Electron transport and CoM-S-S-CoB reduction in *Methanospirillum hungatei* GP1.

Msp. hungatei GP1, a hydrogenotrophic methanoarchaeon of the order Methanomicrobiales, possesses an F<sub>420</sub>-reducing hydrogenase, but lacks an F<sub>420</sub>-nonreducing hydrogenase (Figure 2.1, Sprott et al. 1987). In the absence of F<sub>420</sub>-nonreducing hydrogenase, it is unclear how H<sub>2</sub> is coupled to CoM-S-S-CoB reduction in this methanoarchaeon. Methyl viologen-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity had been detected in the membranes, suggesting a possible F<sub>420</sub>H<sub>2</sub>- and methanophenazine-dependent mechanism for CoM-S-S-CoB reduction. Given the close phylogenetic relationship between the orders Methanosarcinales and Methanomicrobiales (Bapteste et al. 2005), the presence of an F<sub>420</sub>H<sub>2</sub> dehydrogenase in Msp. hungatei GP1 would not be surprising. Therefore, one objective of this thesis was to isolate and study the source of

 $F_{420}H_2$  dehydrogenation activity in *Msp. hungatei* GP1, and to determine if this activity could substitute for the Mvh with respect to CoM-S-S-CoB reduction in this methanoarchaeon.

Objective 3: Purify and characterize the F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of methanol-grown *Methanosarcina barkeri* Fusaro

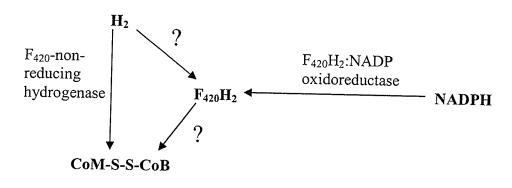


**Figure 2.2**. Electron transport and CoM-S-S-CoB reduction in *Methanosarcina barkeri* Fusaro.

Ms. barkeri Fusaro, grown on methanol as a source of carbon and electrons, would be expected to produce the F<sub>420</sub>H<sub>2</sub> dehydrogenase, but it had previously been reported that benzyl viologen-dependent F<sub>420</sub>H<sub>2</sub>-oxidation was associated with the F<sub>420</sub>-reducing hydrogenase, rather than a distinct F<sub>420</sub>H<sub>2</sub> dehydrogenase (Keltjens and Vogels 1993). Furthermore, a distinct F<sub>420</sub>H<sub>2</sub> dehydrogenase, assayed with methyl viologen, was not detected in this methanoarchaeon by our laboratory (Wong, Ph.D thesis 1999). However, gene sequences coding for a putative F<sub>420</sub>H<sub>2</sub> dehydrogenase were identified in the incomplete genome of Ms. barkeri Fusaro, using sequences coding for subunits of the Fpo complex from Ms. mazei Gö1 (circa. 2003-2004, NCBI website). With the discovery

of methanophenazine as a putative electron acceptor for the  $F_{420}H_2$  dehydrogenase activity of Ms. mazei Gö1 (Abken et al. 1998), it was of great interest to see if a phenazine-dependent  $F_{420}H_2$  dehydrogenase, similar to that found in Ms. mazei Gö1, could be isolated from methanol-grown Ms. barkeri Fusaro.

Objective 4: Purify and characterize the  $F_{420}H_2$  dehydrogenation activity in  $Methanosphaera\ stadtmanae$ 



**Figure 2.3**. Electron transport and CoM-S-S-CoB reduction in *Methanosphaera stadtmanae*.

Various elements of the electron transport pathway have been elucidated in Msph. stadtmanae, including the  $F_{420}H_2$ :NADP oxidoreductase (Fno) and  $F_{420}$ -nonreducing hydrogenase (Mvh) (Figure 2.3, Wong PhD thesis 1999, Elias  $et\ al.\ 2000$ ), but many steps are still unclear, particularly the relevance of a methyl viologen-dependent  $F_{420}H_2$  dehydrogenation activity that was detected in the cell-free extract of  $Msph.\ stadtmanae$  (Wong  $et\ al.\ 1994$ ).  $Msph.\ stadtmanae$ , which grows on  $H_2$  and methanol, possesses an  $F_{420}$ -nonreducing hydrogenase (i.e. Mvh), but an  $F_{420}$ -reducing hydrogenase activity was not detected in this methanoarchaeon (Deppenmeier  $et\ al.\ 1989$ , van de Wijngaard  $et\ al.\ 1989$ 

1991, Wong et al. 1994). F<sub>420</sub> is found at low concentrations in Msph. stadtmanae (0.16 nmol·mg protein<sup>-1</sup>), but F<sub>420</sub>H<sub>2</sub> is required for cellular biosynthesis, and is expected to be necessary during methanogenesis on H<sub>2</sub> and L-serine, or 2-propanol and methanol (Lin and Sparling 1995, Wong et al. 1994). Yet there is no obvious connection between H<sub>2</sub> and F<sub>420</sub>, or F<sub>420</sub>H<sub>2</sub> and Hdr, in the absence of detectable levels of F<sub>420</sub>-reducing hydrogenase (Deppenmeier et al. 1989, Wong et al. 1994). Low levels of H<sub>2</sub>-dependent F<sub>420</sub>-reduction activity were observed in the presence of concentrated cell-free extract, F<sub>420</sub>, and phenazine, suggesting a potential role for the observed F<sub>420</sub>H<sub>2</sub> dehydrogenation activity in methanophenazine-dependent reduction of F<sub>420</sub> in Msph. stadtmanae. The objective was to purify and characterize the activity of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity, and to complete the electron transport pathway for Msph. stadtmanae.

### 3. Materials and Methods

All chemicals used for this thesis were purchased from Sigma Aldrich or Fisher Scientific.

### 3.1. Methanoarchaea used for thesis

The methanoarchaea used for this thesis were obtained from the Deutche Sammlung von Zellkulturen und Mikroooranismen (DSZM), Braunschweig, Germany. The primary methanoarchaea studied during the course of this thesis include *Methanosphaera stadtmanae* (DSZM 3091), *Methanosarcina barkeri* Fusaro (DSZM 804), and *Methanospirillum hungatei* GP1 (DSZM 1101). *Methanococcus thermolithotrophicus* (DSZM 2095) was also grown frequently to provide coenzyme F<sub>420</sub> for various enzymatic assays. *Methanolobus tindarius* (DSZM 2278), *Methanosaeta concilii* GP6 (DSZM 3671), and *Methanothermobacter marburgensis* (formerly *Methanobacterium thermoautotrophicum* Marburg) (DSZM 2133), *Methanobacterium bryantii* (DSZM 863), and *Methanococcus voltae* (DSZM 1533) were also used in this thesis.

#### 3.2. Growth media

Cells were grown in either 1 L Corning bottles, 100 mL serum bottles, or 25 mL Balch tubes; the volume of media depended on the growth substrate used for each methanoarchaeon. Stoppers were wired down to seal the 1 L bottles, while aluminum crimps (Wheaton aluminum seals) were used to seal the Balch tubes and serum bottles.

Media for the respective methanoarchaea were made anaerobic and pressurized to 0.2 atm according to Balch and Wolf (1976), using a water jet vacuum/gas manifold system. Sodium sulfide (Na<sub>2</sub>S·9 H<sub>2</sub>O, 200 mM stock) was added to the anaerobic media (final concentration 2 mM) prior to autoclaving. Additions to the sterile media were made through the rubber stoppers using sterile, anaerobic syringes (treated with anaerobic, sterile 2 mM Na<sub>2</sub>S·9 H<sub>2</sub>O). The volume of inoculum added was 10% of the volume of the medium (i.e. 50 ml of inoculum added to 500 mL of media).

The media formulations of the primary methanoarchaea most frequently studied (Msph. stadtmanae, Ms. barkeri Fusaro, and Msp. hungatei GP1) and Mc. thermolithotrophicus are given here. The media formulations for the other methanoarchaea are given in Appendix A.

Msph. stadtmanae was grown at 37°C according to Miller and Wolin (1985), but in the absence of formate. The media contained (per liter): tryptone (2 g); yeast extract (2 g); K<sub>2</sub>HPO<sub>4</sub> (0.3 g); KH<sub>2</sub>PO<sub>4</sub> (2.8 g); (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (0.3 g); NaCl (0.61 g); MgSO<sub>4</sub>·7H<sub>2</sub>O (0.13 g); CaCl<sub>2</sub>·2 H<sub>2</sub>O (0.08 g); Na acetate (0.5 g); NaSeO<sub>4</sub> (0.04 mg); Na<sub>2</sub>CO<sub>3</sub> (1.7 g); NH<sub>4</sub>Cl (1.0 g); resazurin (0.25 μg); nitrilotriacetic acid (20.2 mg); FeCl<sub>3</sub>·6 H<sub>2</sub>O (2.1 mg); CoCl<sub>2</sub>·6 H<sub>2</sub>O (2.0 mg); MnCl<sub>2</sub>·4 H<sub>2</sub>O (1.0 mg); ZnCl<sub>2</sub> (1.0 mg); NiCl<sub>2</sub>·6 H<sub>2</sub>O (1.0 mg); CuSO<sub>4</sub>·2 H<sub>2</sub>O (0.5 mg); Na<sub>2</sub>MoO<sub>4</sub>·2 H<sub>2</sub>O (0.5 mg); pyridoxine-HCl (0.1 mg); riboflavin (0.5 mg); thiamine (0.05 mg); nicotinic acid (0.05 mg); p-aminobenzoic acid (0.05 mg); lipoic acid (0.05 mg); biotin (0.02 mg); folic acid (0.02 mg); cyanocobalamin (0.01 mg); and Na<sub>2</sub>S·9 H<sub>2</sub>O (0.45 g). The pH of the media prior to gassing/degassing (H<sub>2</sub>:CO<sub>2</sub>) was approximately 7.4 (final pH 7.0). Methanol (CH<sub>3</sub>OH) was added to the media after autoclaving to give a final concentration of 0.4% (100 mM).

Ms. barkeri Fusaro was grown at 37°C according to Kandler and Hippe (1977). The media contained (per liter): yeast extract (2 g); K<sub>2</sub>HPO<sub>4</sub> (0.35 g); KH<sub>2</sub>PO<sub>4</sub> (0.23 g);  $NH_{4}Cl\ (0.5\ g);\ NaCl\ (2.25\ g);\ MgSO_{4}\cdot 7\ H_{2}O\ (0.50\ g);\ CaCl_{2}\cdot 2\ H_{2}O\ (0.25\ g);\ NaSeO_{4}\cdot 10^{-1}$ (0.04 mg); NaHCO<sub>3</sub> (0.85 g); Na<sub>2</sub>CO<sub>3</sub> (1.7 g); resazurin  $(0.25 \text{ }\mu\text{g})$ ; nitrilotriacetic acid  $(20.2 \text{ mg}); FeCl_3 \cdot 6 \text{ H}_2O \text{ } (2.1 \text{ mg}); CoCl_2 \cdot 6 \text{ H}_2O \text{ } (2.0 \text{ mg}); MnCl_2 \cdot 4 \text{ H}_2O \text{ } (1.0 \text{ mg}); ZnCl_2 \cdot 6 \text{ H}_2O \text{ } (2.0 \text{ mg}); MnCl_2 \cdot 4 \text{ H}_2O \text{ } (2.0 \text{ mg}); M$  $(1.0 \text{ mg}); \text{NiCl}_2 \cdot 6 \text{ H}_2\text{O} (1.0 \text{ mg}); \text{CuSO}_4 \cdot 2 \text{ H}_2\text{O} (0.5 \text{ mg}); \text{Na}_2\text{MoO}_4 \cdot 2 \text{ H}_2\text{O} (0.5 \text{ mg});$ pyridoxine-HCl (0.1 mg); riboflavin (0.5 mg); thiamine (0.05 mg); nicotinic acid (0.05 mg); p-aminobenzoic acid (0.05 mg); lipoic acid (0.05 mg); biotin (0.02 mg); folic acid (0.02 mg); cyanocobalamin (0.01 mg); and Na<sub>2</sub>S·9 H<sub>2</sub>O (0.45 g). The pH of the media prior to gassing/degassing was approximately 7.5 (final pH 6.5-6.7). For growth on CH<sub>3</sub>OH, CH<sub>3</sub>OH was added to the media after autoclaving to give a final concentration of 0.4% (100 mM). When acetate was used as methanogenic substrate, NaCH<sub>3</sub>COO was added (prior to gassing/degassing) to a final concentration of 200 mM. For growth on CH<sub>3</sub>OH or acetate the gas phase was N<sub>2</sub>:CO<sub>2</sub>; for growth on CO<sub>2</sub>, the gas phase was  $H_2:CO_2$ .

The hydrogenotrophic methanoarchaeon *Msp. hungatei* GP1, grown on H<sub>2</sub> and CO<sub>2</sub>, was grown at 37°C according to Patel *et al.* (1976). The media contained (per liter); NaCl (0.3 g); MgCl<sub>2</sub>·6 H<sub>2</sub>O (0.064 g); NH<sub>4</sub>Cl (0.4 g); Na acetate (0.42 g); K<sub>2</sub>HPO<sub>4</sub> (0.82 g); CaCl<sub>2</sub>·H<sub>2</sub>O (0.064 g); resazurin (0.25 μg); Na<sub>2</sub>CO<sub>3</sub> (1.0 g); nitrilotriacetic acid (20.2 mg); FeCl<sub>3</sub>·6 H<sub>2</sub>O (2.1 mg); CoCl<sub>2</sub>·6 H<sub>2</sub>O (2.0 mg); MnCl<sub>2</sub>·4 H<sub>2</sub>O (1.0 mg); ZnCl<sub>2</sub> (1.0 mg); NiCl<sub>2</sub>·6 H<sub>2</sub>O (1.0 mg); CuSO<sub>4</sub>·2 H<sub>2</sub>O (0.5 mg); Na<sub>2</sub>MoO<sub>4</sub>·2 H<sub>2</sub>O (0.5 mg); pyridoxine-HCl (0.1 mg); riboflavin (0.05 mg); thiamine (0.05 mg); nicotinic acid (0.05 mg); *p*-aminobenzoic acid (0.05 mg); lipoic acid (0.05 mg); biotin (0.02 mg); folic acid

(0.02 mg); cyanocobalamin (0.01 mg); and  $Na_2S\cdot 9$   $H_2O$  (0.45 g). The pH of the media prior to gassing/degassing ( $H_2:CO_2$ ) was approximately 8.5 (final pH 7.0).

Methanococcus thermolithotrophicus was grown at 63 °C according to R.Sparling (PhD thesis). The media contained (per liter); KCl (0.34 g); NH<sub>4</sub>Cl (1.0); MgCl<sub>2</sub>·6 H<sub>2</sub>O (5.1 g); CaCl<sub>2</sub>·2 H<sub>2</sub>O (0.10 g); K<sub>2</sub>HPO<sub>4</sub> (0.25 g); 2-(N-morpholino)-ethanesulfonic acid (MES, 9.75 g); Piperazine-N,N'-bis-2-ethanesulfonic acid (PIPES, 15.12 g); N-2-Hydroxyethylpiperazine-N'-2-ethanesulfonic acid (HEPES, 11.90 g); resazurin (0.25 μg); NaWO<sub>4</sub>·2 H<sub>2</sub>O (0.07 mg); NaSeO<sub>3</sub> (0.04 mg); NaCl (11.68 g); Na Formate (20.40 g); nitrilotriacetic acid (20.2 mg); FeCl<sub>3</sub>·6 H<sub>2</sub>O (2.1 mg); CoCl<sub>2</sub>·6 H<sub>2</sub>O (2.0 mg); MnCl<sub>2</sub>·4 H<sub>2</sub>O (1.0 mg); ZnCl<sub>2</sub> (1.0 mg); NiCl<sub>2</sub>·6 H<sub>2</sub>O (1.0 mg); CuSO<sub>4</sub>·2 H<sub>2</sub>O (0.5 mg); Na<sub>2</sub>MoO<sub>4</sub>·2 H<sub>2</sub>O (0.5 mg); and Na<sub>2</sub>S·9 H<sub>2</sub>O (0.45 g). The pH of the media prior to gassing/degassing (N<sub>2</sub>:CO<sub>2</sub>) was approximately 6.5 (final pH 5.8-6.2).

Depending on the methanoarchaeon or enzyme being studied, cultures could be harvested either aerobically or anaerobically in the anaerobic chamber (Coy Laboratory Products) containing an atmosphere of 10% H<sub>2</sub>, balance N<sub>2</sub>. In the anaerobic chamber, palladium catalysts were used to remove traces of O<sub>2</sub>. When F<sub>420</sub>H<sub>2</sub> dehydrogenation or F<sub>420</sub>-reducing hydrogenase activities were being studied, cultures were harvested aerobically, with the exception of *Msph. stadtmanae*, which was processed under strictly anaerobic conditions. Harvesting of *Mc. thermolithotrophicus* for F<sub>420</sub> was also conducted under anaerobic conditions. For anaerobic harvesting, all materials were placed in the anaerobic chamber for at least 8 hours prior to usage, usually overnight. Centrifuge bottles capped with rubber o-rings were used to prevent O<sub>2</sub> entry during centrifugation. Cells were centrifuged (15000xg) primarily with a Sorvall RC-5

centrifuge, at 4°C. Cultures were harvested and stored at -60°C in 50 mM HEPES, pH 7.0, containing 15% glycerol,  $\pm$  10 mM dithiothreitol, under an atmosphere of  $N_2$ . Typically, cells were grown, harvested, and frozen until an appropriate amount of cells have accumulated for experimental purposes.

#### 3.3. Cell lysis

With the exception of *Mc. thermolithotrophicus*, all cultures were lysed *via* 3 passages through the French Press (SLM/Aminco, Rochestor) at either 20,000 psi or 40,000 psi (*Msph. stadtmanae*). When anaerobic conditions were required, cold 2 mM Na<sub>2</sub>S was first passed through the press (3 times) to scavenge the O<sub>2</sub> from the pressure cell and from the inlet and outlet tubing. To maintain anaerobic conditions during the lysis process, the inlet and outlet tubing were capped with needles which pierced the rubber stoppers and allowed the extract to pass into the inlet valve and then through the outlet valve into another anaerobic container, minimizing exposure of cell extract to O<sub>2</sub>. *Mc. thermolithotrophicus* did not require passage through the French Press, and could be lysed *via* suspension in low salt buffer.

Post lysis, the extract was centrifuged at low speed (22000xg) in a Sorvall RC-5 centrifuge, at 4°C to remove unlysed cells. This step was followed by high speed centrifugation (120000xg) for 45 minutes at 4°C in L8-80 ultracentrifuge to separate the membranous material from the clarified cell-free extract.

### 3.4. Production of $F_{420}$

 $F_{420}$  was used for a number of assays, including the  $F_{420}H_2$  dehydrogenation activity, F<sub>420</sub>-reducing activity, F<sub>420</sub>H<sub>2</sub> oxidase activity, and for the synthesis of an F<sub>420</sub>affinity column. Mc. thermolithotrophicus was grown in either 1 L bottles (500 mL/bottle) or in an 11 litre fermentor (10 litre of media, New Brunswick Scientific). The cells were grown to late log phase, harvested anaerobically and then lysed in Tris/Cl, pH 8.0 (60% v/v) with CH<sub>3</sub>OH (40% v/v) at 80°C for at least one hour. The lysed cell debris and membranes were removed via centrifugation at 22000xg in a Sorvall RC-5 centrifuge, at 4°C. The supernatant was then loaded onto a column of A-25 DEAE Sephadex and washed with 50 mM Tris/Cl, pH 8.0. The bound  $F_{420}$ /protein was then washed with stepwise increments of ammonium bicarbonate buffer, from 0.3 to 0.5 M; the F<sub>420</sub> was separated from the other bound soluble components found in the supernatant, followed by elution from the column using the 1.0 M buffer. The  $F_{420}$ :ammonium bicarbonate solution was then heated at 80°C, and the ammonium bicarbonate was evaporated using a rotoevaporator. The  $F_{420}$  residue was resuspended in either 50 mM Tris/Cl, pH 8.0, or distilled H<sub>2</sub>O.

# 3.5. Purification of the $F_{420}H_2$ dehydrogenation activity from methanol-grown Methanosarcina barkeri Fusaro and Methanospirillum hungatei GP1

All purification steps were performed aerobically at 4°C; three independent purifications were performed for each methanoarchaeon. Frozen cells were thawed and resuspended in 50 mM HEPES, pH 7.0, containing 5% glycerol, 50  $\mu$ M phenylmethylsulfonyl fluoride (PMSF), and several crystals of RNase and DNase. The cell suspension was lysed *via* three passages through a French Pressure Cell

(SLM/Aminco, Rochestor), at 20000 psi and the resultant cell-free extract was centrifuged at 22000g. The supernatant was then further centrifuged at 180000g for 6 hours, 4°C. The high speed pellet (HSP) formed, which contains both the soluble cytoplasmic and membrane-bound  $F_{420}H_2$  dehydrogenation activities, was homogenized in 10 mM HEPES, pH 7.0, containing 0.5 M NaCl (buffer A).

The homogenate was loaded onto a column of Ni<sup>2+</sup>-charged iminodiacetic acid (45 mL) on Sepharose fast flow resin equilibrated in buffer A. The membranes, containing up to 20% of the overall F<sub>420</sub>H<sub>2</sub> dehydrogenation activity, flowed through the column without binding, while the soluble activity bound to the Ni<sup>2+</sup>-affinity column. The protein-loaded column was washed with 2-3 column volumes of buffer A, followed by washing with 2 column volumes of 10 mM HEPES, 0.5 M NaCl, and 0.05% Triton-X-100 (buffer B). The remaining protein, bound on the column as a dark golden-brown band, was eluted with buffer B supplemented with 25 mM imidazole.

Fractions (3.0 mL) containing  $F_{420}H_2$  dehydrogenation activity were pooled and desalted using an Amicon ultrafiltrator, with an YM30 membrane. The concentrated protein, suspended in 10 mM HEPES, pH 7.0, 0.05% Triton-X-100, and 10% glycerol, was then loaded onto two 3-20% non-denaturing gradient PAGE gels (Hoeffer systems) and separated by electrophoresis overnight (no more than 15 hours) at a constant 150 V. Protein visualized as high molecular weight yellow-brown bands on the unstained PAGE were excised from the gel and cut into 1-2 mm³ pieces, placed into the tubes of a Bio-Rad electro-elutor, and eluted from the gel after 8 hours, with fresh buffer exchanged every 2 hours. After electro-elution, the soluble protein was homogenized in 10 mM HEPES, pH 7.0, 0.05% Triton-X-100, and 10% glycerol, and stored at 4°C under  $N_2$ . Low molecular

weight bands of similar color were processed in the same manner. For the study of proteins that could not be visualized on the native PAGE, molecular markers were placed in the outermost lanes prior to electrophoresis. After electrophoresis, the lanes containing the markers were stained with coomassie blue and then detained until the molecular marker bands could be visualized. The molecular marker gel strips were washed in distilled water and then realigned with the unstained gel. Gel bands at the approximate molecular weight were then excised and processed as described above.

Membrane-bound F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was isolated from the non-binding fractions collected from the Ni <sup>2+</sup>-affinity column. The membranes were sedimented, washed, and solubilized using the detergent CHAPS. The solubilized F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was enriched using either a discontinuous (40-45-50%) sucrose gradient ultracentrifugation step (*Msp. hungatei* GP1) or using a Ni<sup>2+</sup>-affinity column (*Ms. barkeri* Fusaro), followed by desalting/concentration using an Amicon Ultrafiltrator. The concentrated protein was then loaded onto a pair of 3-20% non-denaturing gradient PAGE gels (Hoeffer systems) and electrophoresed overnight (no more than 15 hours) at a constant 150 V.

The protein from the membranes of *Ms. barkeri* Fusaro could be visualized as high molecular weight yellow-brown bands, which were then excised from the gel and cut into 1-2 mm<sup>3</sup> pieces, placed into the tubes of a Bio-Rad electro-elutor, and electrophoresed for 8 hours with fresh buffer exchanged every 2 hours. For the study of proteins that could not be visualized on the native PAGE, molecular markers were placed in the outermost lanes prior to electrophoresis and processed as described above.

The bands of electrophoresed membrane protein from  $Msp.\ hungatei$  GP1 were faint and more difficult to visualize, so the native gel was prepared as described above, using molecular markers in the outer gel lanes. After electro-elution, the protein was homogenized in 10 mM HEPES, pH 7.0, 0.05% Triton-X-100, and 10% glycerol, and stored at 4°C under  $N_2$ .

### 3.6. Enzyme assays

### 3.6.1. $F_{420}H_2$ dehydrogenation assays

Enzyme assays were performed using anaerobic stoppered 12 mm x 75 mm glass cuvettes as described by Daniels and Wessels (1984). The rate of oxidation/reduction of  $F_{420}$  ( $\varepsilon_{420~nm} = 45.5~mM^{-1} \cdot cm^{-1}$ ) or reduction of methyl viologen ( $\varepsilon_{560~nm} = 8~mM^{-1} \cdot cm^{-1}$ ) was followed spectrophotometrically using an LKB Novaspec II spectrophotometer.

The assay buffer for  $F_{420}H_2$  dehydrogenation activity normally consisted of 40 mM potassium phosphate, pH 7.0, 0.5 M sucrose, and 10 mM dithiothreitol (DTT), under an atmosphere of  $N_2$ .  $F_{420}H_2$  dehydrogenation assay tubes for *Msph.stadtmanae* consisted of 200 mM KPO<sub>4</sub> rather than the 40 mM KPO<sub>4</sub> used for other methanoarchaea.

 $F_{420}$  was reduced according to Haase *et al.* (1992). Sodium borohydride was added in small increments to the  $F_{420}$  (flushed with  $N_2$ ) until the solution turned colorless. Concentrated HCl was then added to destroy excess sodium borohydride; HCl was added until the solution just turned cloudy. KOH (10 N) was added dropwise to neutralize excess HCl, until the cloudiness just cleared. Reduced  $F_{420}$  was gassed/degassed under  $N_2$  and then added to the assay mix immediately prior to measurement, followed by addition of purified enzyme or enzyme extract. As a control, a baseline for  $F_{420}H_2$ 

oxidation in the absence of electron acceptors was established, before the reaction was initiated by addition of electron acceptor. No reaction was observed in the absence of  $F_{420}H_2$ .

### 3.6.2. Hydrogenase assays

### 3.6.2.1. $F_{420}$ -reducing hydrogenase activity

The buffer for assay of the  $F_{420}$ -reducing hydrogenase was modified from Choquet and Sprott (1991) and consisted of 100 mM HEPES, pH 7.0, with 10  $\mu$ M  $F_{420}$ , 10  $\mu$ M FAD, 200 mM KCl, 20 mM MgCl<sub>2</sub>·6H<sub>2</sub>O, and 20 mM DTT, under an atmosphere of H<sub>2</sub>. Protein was reductively reactivated in the  $F_{420}$ -reducing hydrogenase assay tube (until the FAD and  $F_{420}$  were reduced), followed by addition of  $F_{420}$  (20-25  $\mu$ M) and measurement of activity. Assays were also performed under an atmosphere of N<sub>2</sub> (control). 50 mM Tris/Cl, pH 8.0, and 200 mM KPO<sub>4</sub>, pH 7.0, were also used to test for optimum  $F_{420}$ -reducing activity of *Msph. stadtmanae*.

## 3.6.2.2. $F_{420}$ -nonreducing hydrogenase activity

Assay tubes to measure  $F_{420}$ -nonreducing hydrogenase activity contained 50 mM Tris/Cl, pH 8.0, 0.05 mM methyl viologen, 20 mM DTT, and 20 mM MgCl<sub>2</sub>·6H<sub>2</sub>O, under an atmosphere of H<sub>2</sub>. Protein was reductively activated in the assay tube (until the tubes turned light blue), followed by addition of methyl viologen (final concentration: 2 mM) and measurement of activity.

### 3.6.2.3. H<sub>2</sub>-dependent phenazine reduction

 $H_2$ -dependent phenazine reduction was observed using  $F_{420}H_2$  dehydrogenation buffer, under an atmosphere of  $H_2$ . Protein was incubated in anaerobic buffer for 30 minutes to ensure that the protein was reductively reduced before phenazine (100  $\mu$ M,  $\epsilon_{365\,\text{nm}}=13\,\text{mM}^{-1}\cdot\text{cm}^{-1}$ ) was added to the reaction mix.

## 3.6.3. $F_{420}H_2$ oxidase activity of Methanosphaera stadtmanae

All manipulations of Msph. stadtmanae cultures and extracts were performed in the anaerobic chamber, with the exception of cell lysis (French press) and centrifugation steps. O<sub>2</sub>-dependent  $F_{420}H_2$  oxidation from the cell-free extract of Msph. stadtmanae was assayed according to Seedorf et al. (2004). The assay buffer consisted of 50 mM Tris/Cl, pH 7.6 dispensed into 12 mm x 75 mm glass cuvettes sealed with rubber stoppers, and made anaerobic by gassing with  $N_2$ .  $F_{420}H_2$  was added to a final concentration of 15  $\mu$ M. Additions of  $O_2$  prior to assay were made using air-saturated water (~250  $\mu$ M  $O_2$ ). Distilled water was bubbled with breathing grade air (Welders Supplies) for at least two hours and then dispensed and sealed into a Balch tube, followed by pressurization with breathing grade air. The reaction was measured following addition of the air-saturated water to assay tubes containing cell-free extract and  $F_{420}H_2$ . A background for  $F_{420}H_2$  oxidation was established prior to addition of the air-saturated water.  $F_{420}H_2$  oxidase assays were performed at ambient room temperature (~25°C).

A problem that arose during the  $F_{420}H_2$  oxidase assays is the presence of  $O_2$  in the tubes prior to addition of  $O_2$ -saturated  $H_2O$ . Unlike the  $F_{420}H_2$  dehydrogenation/ $F_{420}$ -

reducing hydrogenase assays, reducing agents are not added to the  $F_{420}H_2$  oxidase buffer, to prevent the removal of  $O_2$  necessary for assay of the  $F_{420}H_2$  oxidase activity. However, traces of  $O_2$  may be trapped in the assay tubes even after evacuating with  $N_2$ . As a result, the addition of cell-free extract to assay tubes containing  $F_{420}H_2$  leads to significant  $F_{420}H_2$  oxidation in the absence of reducing agent or  $O_2$ -saturated  $O_2$ . To alleviate this problem,  $O_2$ -saturated  $O_2$ 0 was added only after the  $O_2$ 2 present in the tubes was reduced (i.e. a stable baseline for  $O_2$ 2 oxidation was reached in the absence of injected  $O_2$ 2.

### 3.7. Gel electrophoresis

### 3.7.1. SDS-PAGE gel electrophoresis

SDS-PAGE gels were prepared according to Laemmli (1970), with an acrylamide concentration of 12.8%. Protein samples, mixed with SDS loading buffer, were incubated at 100°C for 5 minutes and loaded immediately onto the SDS PAGE gel followed by electrophoresis using a Bio-Rad mini-protean II electrophoresis unit for 1-2 hours at a constant voltage of 150 V. Following electrophoresis, the SDS-PAGE gels were stained with Coomassie Brilliant Blue R, followed by staining with silver nitrate.

### 3.7.1.1. Amino-terminus sequencing

For amino (N-) terminus sequencing of individual subunits, unstained SDS PAGE gels were electroblotted immediately after electrophoresis onto nitrocellulose papers for approximately one hour at 4°C, using a Bio-Rad electro blotter. The paper was then stained using ponceau reagent and the bands were aseptically cut out and stored at -20°C

until they were transferred for sequencing. N-terminal sequencing of the purified subunits was performed by the Center of Proteomics at the University of Victoria.

### 3.7.1.2. Mass Spectroscopy analysis

Mass spectroscopy sequencing of fragments of the purified phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of Msp. hungatei GP1 was performed in the Department of Physics, at the University of Manitoba. A protocol for sample preparation prior to mass spectroscopy analysis is given here. Following electrophoresis the SDS page gel was stained in coomassie blue for 1-2 hours, followed by destaining of the gel to remove background. The stained bands were then cut out of the gel into 1 mm<sup>3</sup> cubes and placed into eppendorf tubes, using a scalpel wiped with 50% CH<sub>3</sub>OH between each sample. Prior to digestion, the gel bands were destained via the following steps (constituting one cycle). The bands were washed with 100 mM ammonium bicarbonate to neutralize the acetic acid remaining after destaining, followed by washing in 40% acetonitrile (in ammonium bicarbonate) to remove the coomassie blue dye, with rotation of the eppendorf tubes for 10 minutes. The tubes were then centrifuged to remove liquid. Acetonitrile was added to the gel pieces, which further removed salt and detergent; the gel pieces were then allowed to sit in the acetonitrile for 5 minutes and then spun to remove the acetonitrile (end of cycle). The cycle was repeated, starting with the addition of 100 mM ammonium bicarbonate, until the dyes on the gel were completely removed.

In-gel digestion of the destained gel pieces proceeded with incubation of the gel in freshly made 10 mM dithiothreiotol (DTT), in 100 mM ammonium bicarbonate (2-3x gel volume), for 45 minutes at 57.5°C. The tubes were then centrifuged, and the volume

of DTT removed was noted. The difference in the volume of DTT added and removed indicated the amount required to swell the gel; this volume is equal to the amount of trypsin that was added for digestion of the protein in the gel. 55 mM iodoacetamide in 100 mM ammonium bicarbonate is added to the eppendorf tubes, followed by incubation in the dark for 30 minutes. Afterwards, the tubes were centrifuged and the liquid discarded. Trypsin (5 ng/ $\mu$ l) in 100 mM ammonium bicarbonate was added to the tubes, and incubated at  $37^{\circ}\text{C}$  overnight.

Following trypsin digestion, the tubes were briefly centrifuged and the liquid (referred to as supernatant) was set aside. 0.01% TFA was added to stop digestion of the protein (5 minute incubation). The liquid was then added to the supernatant previously reserved. The following steps constitute a cycle. Peptide extraction began with the addition of 2-3 volumes of 0.01% TFA in 50% ACN, followed by sonication for 10 minutes in cold water. The tubes were centrifuged, and the liquid was collected and pooled with the supernatant. To complete the cycle, the gel pieces were then incubated for 5 minutes in 40% ACN, followed by centrifugation and removal of the liquid to be added to the supernatant. The cycle was then repeated two more times, beginning with the addition of 0.01% TFA in 50% ACN to the gel pieces in the eppendorf tubes. It is important to note that all of the liquid was reserved and stored with their respective pools.

After the extraction process was completed, the peptide samples were concentrated using a Speed Vac. When necessary, gel pieces remaining in the sample were removed with a ZipTip.

### 3.7.2. Native gradient PAGE gel electrophoresis

Native gradient (3-20% acrylamide) PAGE gels were loaded with 140 µg protein per lane, and electrophoresed for at least 15 hours (no more than 20 hours) at a constant 150 V. The gels were then stained with Coomassie Brilliant Blue R, followed by silver staining according to a protocol ProteoSilver Plus Silver Stain Kit (Prot-Sil2) from Sigma Aldrich.

Activity stains of the native gradient PAGE gels were performed as follows. Following electrophoresis, the native PAGE gels were submerged in anaerobic 50 mM Tris/Cl buffer, pH 8.0, containing 2 mM methyl viologen, 20 mM MgCl<sub>2</sub>, and 5 mM DTT. The gels were then incubated inside an anaerobic chamber under an atmosphere of (approximately) 10% H<sub>2</sub> (balance N<sub>2</sub>) until strong blue bands of reduced methyl viologen could be visualized. Prior to removal of the gels from the anaerobic chamber, 2,3,5-triphenyltetrazolium chloride (1.5 % stock) was added to fix the reduced methyl viologen bands to the gel, producing aerobically-stable red activity-stained bands.

### 3.8. Genome sequence analysis

BLAST (Basic Local Alignment Search Tool) searches of the *Archaea* database were performed at the NCBI website (National Center for Biotechnology Information, <a href="http://www.ncbi.nlm.nih.gov/">http://www.ncbi.nlm.nih.gov/</a>), using the tblastn program (using a protein query to search the translated nucleotide database). Protein queries are as described in the appropriate chapters.

### 3.9. Protein determination

Protein concentration was determined using the Bradford (1976) method, using bovine serum albumin as standard.

## 4. $F_{420}H_2$ dehydrogenation activity in the methanoarchaea

#### 4.1 Introduction

Members of the order *Methanosarcinales* are the most metabolically and physiologically versatile methanoarchaea, containing members capable of metabolizing methylotrophic (methyl-group containing) compounds and/or acetate; some of these methanoarchaea can also grow on H<sub>2</sub> plus CO<sub>2</sub> or CO (Galagan *et al.* 2002, Rother and Metcalf 2004, Maeder *et al.* 2006).

The F<sub>420</sub>H<sub>2</sub> dehydrogenase, also known as the Fpo (E<sub>420</sub>H<sub>2</sub> phenazine oxidoreductase) complex, is a membrane-bound, multi-subunit protein involved with electron transfer reactions and energy conservation in methylotrophic methanoarchaea of the *Methanosarcinaceae* that metabolize methanol or CO (Deppenmeier 2004, J.G. Ferry, personal communication). The complex transfers reducing equivalents from the methanogenic electron carrier coenzyme F<sub>420</sub>H<sub>2</sub> to the electron carrier methanophenazine (Abken *et al.* 1998). Electrons are then transferred from reduced methanophenazine to the heterodisulfide reductase (Hdr), and ultimately leading to reduction of the heterodisulfide complex (CoM-S-S-CoB) to free cofactors CoM-SH and CoB-SH (Figure 17, section 1.5.2.4.) (Deppenmeier 2002, 2004). The two redox reactions are coupled to energy conservation, generating a transmembrane proton gradient used to synthesize ATP *via* A<sub>1</sub>A<sub>0</sub> ATP synthase (Bäumer *et al.* 1998, Brodersen *et al.* 2000).

The F<sub>420</sub>H<sub>2</sub> dehydrogenase of *Methanolobus tindarius* and *Methanosarcina mazei* Gö1 were both initially described as holoenzymes (115-120 kDa) consisting of 5 distinct subunits (FpoBCDFI), containing FeS clusters and FAD as prosthetic groups (Haase *et al.* 1992, Abken and Deppenmeier 1997). Analysis of the genome from *Ms. mazei* Gö1

revealed a much more sophisticated complex consisting of 13 subunits, with gene sequences and operon structure reminiscent of the NADH:ubiquinone oxidoreductase (NDH-1, Complex I) of *E. coli*, and also sharing similarities with the subunits of analogous structures from various eukaryotic organisms (Bäumer *et al.* 2000). It has been suggested that the predicted 400 kDa complex was not isolated during the purification process described by Abken and Deppenmeier (1997) due to the use of the detergent CHAPS to solubilize the protein from the cell membranes, which may have led to the fragmentation of the large protein complex; the use of a milder detergent, octylglycoside, resulted in the partial purification of the 400 kDa Fpo complex (Deppenmeier 2002).

As the amount of methanophenazine isolated from the cell membranes of *Ms. mazei* Gö1 was extremely low, and too hydrophobic to be assayed using standard aqueous buffers, 2-hydroxy (2-OH)-phenazine was synthesized and used to assay the dehydrogenase activity (Beifuss and Tietze 2005). Phenazine could also be used as an electron acceptor to assay the F<sub>420</sub>H<sub>2</sub> dehydrogenase activity, although it was not as effective as 2-OH-phenazine; the K<sub>m</sub> for phenazine was almost 10-fold higher than the K<sub>m</sub> for 2-OH-phenazine. As it is commercially available, phenazine is used in our laboratory to assay the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity. Phenazine is also used to assay membrane-bound hydrogenase activity and CoM-S-S-CoB reduction in the methanoarchaeon *Methanomicrococcus blatticola*, a member of the *Methanosarcinales*; the presence of methanophenazine has not been confirmed in this methanoarchaeon (Sprenger *et al.* 2005). It should be noted that methanophenazine has not been isolated from methanoarchaea other than *Ms. mazei* Gö1; much like cytochromes, this compound

may not exist in methanoarchaea outside of the *Methanosarcinaceae*. The genes coding for methanophenazine biosynthesis proteins have not been identified to date.

The 2-OH phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of *Ms. mazei* Gö1 was found to be competitively inhibited by the presence of diphenyleneiodonium (DPI) chloride, a compound with a planar structure similar to phenazine (Brodersen *et al.* 1999). DPI is also an inhibitor of flavoproteins such as mitochondrial and bacterial NADH:ubiquninone oxidoreductase, and neutrophil NADPH oxidase; DPI inhibits (uncompetitive) these enzymes by binding to the FAD component of the respective enzymes (O'Donnell *et al.* 1993, Majander *et al.* 1994). Thus, inhibition of phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity by DPI could indicate the presence of a protein similar to the F<sub>420</sub>H<sub>2</sub> dehydrogenase of *Ms. mazei* Gö1.

Thus far, the F<sub>420</sub>H<sub>2</sub> dehydrogenase has been isolated from only two methylotrophic methanoarchaea *Ml. tindarius* and *Ms. mazei* Gö1 (Haase *et al.* 1992, Abken and Deppenmeier 1997). The genome of *Methanosarcina* (*Ms.*) *barkeri* Fusaro and *Methanosarcina* (*Ms.*) *acetivorans* C2A, both of which are capable of methylotrophic growth, also contain genes encoding for the subunits of the Fpo complex, (J.G. Ferry, personal communicaton, Maeder *et al.* 2006). A similar complex has also been purified from the sulfate-reducing archaeon *Archaeoglobus fulgidus*, which is phylogenically related to the *Methanosarcinales* (Kunow *et al.* 1994, Brüggermann *et al.* 2000). This complex has not been isolated in methanoarchaea outside of the *Methanosarcinaceae*, and analyses of the published genomes of *Methanocaldococcus jannaschii*, *Methanothermobacter thermoautotrophicus* ΔH and *Methanopyrus kandleri* have not indicated the presence of a distinct Fpo complex, although there are gene sequences

related to complex I that are found in all methanoarchaea, encoding the membrane-bound multi-subunit energy conserving hydrogenase (Ech) (Bult *et al.* 1996, Smith *et al.* 1997, Slesarev *et al.* 2002, Hedderich 2004, Fricke *et al.* 2006).

Research in our laboratory has demonstrated that  $F_{420}H_2$  dehydrogenation activity is not restricted by methanogenic substrate or to the *Methanosarcinaceae*, and was apparently ubiquitous in all of the methanoarchaea studied in our laboratory. Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity has been detected in the cell free extracts from methanoarchaea metabolizing  $H_2$  and  $CO_2$ , acetate,  $H_2$  and methanol. The results of a survey examining the  $F_{420}H_2$  dehydrogenation activity and DPI-inhibition are detailed in this chapter. There are several possible candidates for this activity, including the  $F_{420}$ -reducing hydrogenase, which are discussed in section 4.7.

### 4.2. Materials and methods

For a comprehensive look at the materials and methods used for this chapter, the reader is referred to the following sections in Chapter 3:

- 3.2. Growth media
- 3.3. Cell lysis
- 3.6. Enzyme assays
  - $3.6.1. F_{420}H_2$  dehydrogenation assays

### 4.3. Results and Discussion

The results of a survey examining the phenazine-dependent  $F_{420}H_2$  dehydrogenation activities in cell-free extracts from a number of methanoarchaea across the methanogenic orders are summarized in Table 4.1. This activity is observed in all

methanoarchaea regardless of growth substrate, to varying degrees, but the physiological significance of this activity is unclear. In most instances, phenazine ( $E^{\circ}$ '= -0.265 to -0.165 V) appears to be more effective as an electron acceptor, compared to methyl viologen ( $E^{\circ}$ '= -0.446 V), ranging from a 2- to 10-fold increase in activity. However, methyl viologen and phenazine show little difference in electron accepting capability when looking at the  $F_{420}H_2$  dehydrogenation activities of *Msp. hungatei* GP1 and *Msph. stadtmanae*, which contained low activities.

 $F_{420}H_2$  (E°°= -0.360 V), when mixed with cell-free extract  $\pm$  cell membranes, is oxidized to  $F_{420}$  and  $H_2$  in the absence of added electron acceptor, a function of the 'reverse'  $F_{420}$ -reducing hydrogenase activity. To distinguish between the reverse hydrogenase activity from phenazine- (or another electron acceptor) dependent  $F_{420}H_2$  oxidation, a stable baseline of  $F_{420}H_2$  oxidation in the absence of electron acceptor is established prior to addition of electron acceptor and subtracted from the activity measured upon addition of electron acceptor.

# 4.4. Phenazine-dependent $F_{420}H_2$ dehydrogenation activity in the Methanosarcinales

The highest levels of phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activities were observed with *Ml. tindarius* and CH<sub>3</sub>OH-grown *Ms. barkeri* Fusaro (Table 4.1). By comparison, cultures of *Ms. barkeri* Fusaro grown on H<sub>2</sub>:CO<sub>2</sub> or acetate were approximately 2- and 4-fold lower, respectively. The corresponding activity in *Mst. concilli*, an aceticlastic member of the *Methanosarcinales*, was 5-fold lower. For further study, the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Ml. tindarius* and CH<sub>3</sub>OH-grown *Ms. barkeri* Fusaro were examined.

Table 4.1. Comparison of the  $F_{420}H_2$  dehydrogenation activities found in the cell-free extract of various methanoarchaea. Result are the average of three separate experiments.

Methanoarchaeon	Order	Growth substrate	$F_{420}H_2$ $deH_2ase^{\dagger}$ (phenazine)	$F_{420}H_2$ $deH_2ase^{\dagger}$ $(MV)$	K <sub>m phenazine</sub> ‡	Vmax <sup>‡‡</sup>	Ic <sub>50</sub> (μM DPI)
Ml. tindarius	Methanosarcinales	Methanol	0.379	0.2	$266.7 \pm 44.0$	$0.50 \pm 0.04$	
Ms. barkeri Fusaro	Methanosarcinales	Methanol	0.421	0.048	$95.0 \pm 19.6$	$0.40 \pm 0.03$	3
		H <sub>2</sub> :CO <sub>2</sub>	0.225	0.047	ND	0.40 ± 0.03	>100
		Acetate	0.116	0.019	ND		ND
Mst. concilli	Methanosarcinales	Acetate	0.087			ND	ND
Msp. hungatei GP1	Methanomicrobiales	H <sub>2</sub> :CO <sub>2</sub>		0.042	ND	ND	ND
Mc. voltae	Methanococcales		0.129	0.123	$243.0 \pm 21.1$	$0.39 \pm 0.01$	25
Mtb. marburgensis		H <sub>2</sub> :CO <sub>2</sub>	0.212	0.067	$915.5 \pm 213.4$	$0.60 \pm 0.1$	25
	Methanobacteriales	H <sub>2</sub> :CO <sub>2</sub>	0.270	0.07	$254.2 \pm 60.8$	$0.41 \pm 0.05$	
Mb. bryantii	Methanobacteriales	H <sub>2</sub> :CO <sub>2</sub>	0.296	0.091			30
Msph. stadtmanae	Methanobacteriales	H <sub>2</sub> :methanol			$318.1 \pm 48.5$	$0.92 \pm 0.07$	10
		2modianoj	0.0023	0.005	$138.2 \pm 17.5$	$.003 \pm 0.0002$	2.5

<sup>†</sup> μmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>·mg<sup>-1</sup>
‡ μM phenazine
‡‡ μmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>·mg<sup>-1</sup>
ND, not determined

### 4.4.1. Methanolobus tindarius

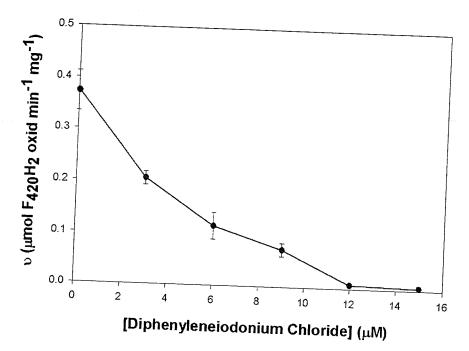
Ml. tindarius is a mesophilic marine, coccoid methanoarchaeon growing solely on methanol and methylamines as carbon and energy source (König and Stetter 1982).

Ml. tindarius is a member of the Methanosarcinaceae family and one of the first methanoarchaea for which the F<sub>420</sub>H<sub>2</sub>-dependent CoM-S-S-CoB reduction was studied, and the first for which the F<sub>420</sub>H<sub>2</sub> dehydrogenase activity was purified (Deppenmeier et al. 1990a, Haase et al. 1992).

 $Ml.\ tindarius$  contains the Fpo complex, but no  $F_{420}$ -reducing hydrogenase, making this methanoarchaeon a suitable model of activity for comparison with the  $F_{420}H_2$  dehydrogenation activity found in other methanoarchaea (Deppnemeier  $et\ al.\ 1989$ , Haase  $et\ al.\ 1992$ ). The  $F_{420}H_2$  dehydrogenase activity was initially assayed using methyl viologen and metronidazole (Haase  $et\ al.\ 1992$ ), but this activity was assayed in our studies using phenazine as an electron acceptor; the phenazine-dependent activity was approximately two-fold higher compared to the activity assayed using the same concentration of methyl viologen (Table 4.1).

The apparent  $K_m$  for phenazine was determined to be 267  $\mu$ M (cell-free extract); the apparent  $K_m$  for phenazine of the purified  $F_{420}H_2$  dehydrogenation activity of  $Ms.\ mazei$  Gö1 is 250  $\mu$ M (Abken  $et\ al.1998$ ). While the presence of methanophenazine has not yet been reported, it is expected that this compound would be found in the membranes of  $Ml.\ tindarius$ . The phenazine-dependent activity was strongly inhibited by the presence of DPI ( $Ic_{50} \sim 3\ \mu$ M DPI) (Figure 4.1A). Addition of 3 and 9  $\mu$ M DPI increased the apparent  $K_m$  for phenazine, respectively, 266  $\mu$ M to 1168 and 2313  $\mu$ M,

A.



B.

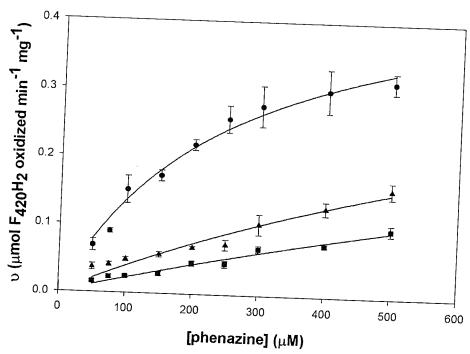


Figure 4.1. (A) Sensitivity of the phenazine-dependent  $F_{420}H_2$  dehydrogenase activity of *Methanolobus tindarius* (in cell-free extract) to diphenyleneiodonium (DPI) chloride and (B) Michaelis Menten kinetics plot of the phenazine-dependent  $F_{420}H_2$  dehydrogenase activity in the absence/presence of DPI (B). •, no DPI added. •, + 3  $\mu$ M DPI. •, + 9  $\mu$ M DPI.

with Vmax =  $0.5 \ \mu mol \ F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> (Figure 4.1B). Similar kinetic parameters were observed were observed when using washed membranes in place of cell-free extract, which contained 15-20% of the overall  $F_{420}H_2$  dehydrogenase activity, indicating competitive inhibition of the  $F_{420}H_2$  dehydrogenase activity by DPI.

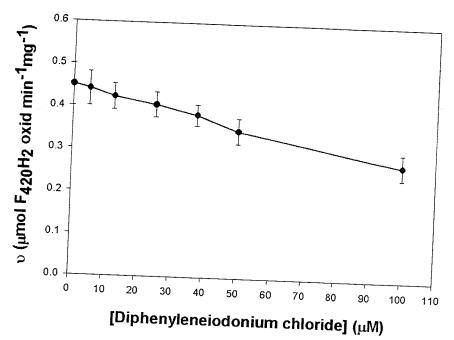
DPI-mediated inhibition of the phenazine-dependent  $F_{420}H_2$  dehydrogenase in Ml. tindarius was similar to observations made using purified  $F_{420}H_2$  dehydrogenase of Ms. mazei Gö1(using 2-OH-phenazine), where the addition of 1  $\mu$ M DPI induced an increase in the apparent  $K_m$  for 2-OH-phenazine from 35 to 100  $\mu$ M (Brodersen *et al.* 1999).

### 4.4.2. Methanosarcina barkeri Fusaro

Ms. barkeri Fusaro is also a member of the Methanosarcinaceae and extremely versatile with respect to its selection of methanogenic substrate; like Ms. mazei Gö1, Ms. barkeri can metabolize methanol  $\pm$  H<sub>2</sub>, H<sub>2</sub>:CO<sub>2</sub>, or acetate for methanogenesis (Maeder et al. 2006). F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was detected in the cell-free extract of Ms. barkeri Fusaro when it is grown on the various substrates, to varying levels (Table 4.1).

Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was measured from the cell-free extract and membranes of methanol-grown Ms. barkeri Fusaro, but the source of this activity was not immediately clear. Work in our laboratory revealed that the  $F_{420}H_2$  dehydrogenation activity of Ms. barkeri Fusaro, grown on methanol, elutes with an  $F_{420}$ -reducing hydrogenase activity (data not shown, Wong, PhD thesis 1999); on the other hand, the Fpo complex does not possess  $F_{420}$ -reducing hydrogenase activity (Abken and Deppenmeier 1997).





B.

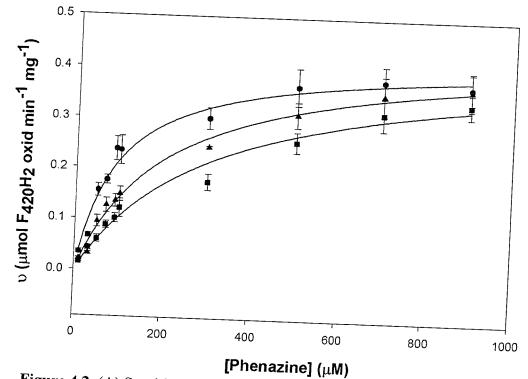


Figure 4.2. (A) Sensitivity of the phenazine-dependent  $F_{420}H_2$  dehydrogenase activity of CH<sub>3</sub>OH-grown *Methanosarcina barkeri* Fusaro (in cell-free extract) to diphenyleneiodonium (DPI) chloride and (B) Michaelis Menten kinetics plot of the phenazine- $F_{420}H_2$  dehydrogenase activity in the absence/presence of DPI (B). •, no DPI added. •, + 50  $\mu$ M DPI. •, + 100  $\mu$ M DPI.

Phenazine was an effective electron acceptor for the  $F_{420}H_2$  dehydrogenation activity, with a 5- to 10-fold increase in  $F_{420}H_2$  dehydrogenation activity relative to methyl viologen (Table 4.1). The apparent  $K_m$  for phenazine of the  $F_{420}H_2$  dehydrogenation activity in cell-free extract from Ms. barkeri Fusaro was 95  $\mu$ M (Vmax = 0.40  $\mu$ mol  $F_{420}H_2$  oxidized min<sup>-1</sup>·mg<sup>-1</sup>), which is approximately 2.5-fold lower than the corresponding  $K_m$  for Ml. tindarius (266  $\mu$ M), indicating a higher affinity for the substrate. Similar to the activity of Ml. tindarius, the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was competitively inhibited by the presence of DPI, though not to the same extent. There was a substantial difference in the degree of inhibition of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity, as can be seen in Figure 4.2A. Addition of 50 and 100  $\mu$ M DPI increases the apparent  $K_m$  value, from 95 to 213 and 317  $\mu$ M, respectively (Figure 4.2B).

It is clear that the while both  $Ml.\ tindarius$  and  $Ms.\ barkeri$  Fusaro possess phenazine-dependent  $F_{420}H_2$  dehydrogenation activities, there are several differences that distinguish the respective activities. Addition of DPI has a more significant effect on the activity of  $Ml.\ tindarius$ , comparable to the inhibition reported for  $Ms.\ mazei$  Gö1 by Brodersen  $et\ al.\ (1999)$ . The inhibition curve for  $Ml.\ tindarius$  ( $Ic_{50}=3\ \mu M\ DPI$ ) shows a much steeper drop in dehydrogenation activity compared to  $Ms.\ barkeri$  Fusaro ( $Ic_{50}>100\ \mu M\ DPI$ ). Low concentrations of DPI (3  $\mu M$ ) were effective in inhibiting the  $F_{420}H_2$  dehydrogenation activity in  $Ml.\ tindarius$ , the same concentration has a negligible effect on the analogous activity in  $Ms.\ barkeri$  Fusaro. The addition of 100  $\mu M$  DPI only decreased the  $F_{420}H_2$  dehydrogenation activity of  $Ms.\ barkeri$  Fusaro by 30% (Figure

4.2A); at that same concentration the corresponding activity of *Ml. tindarius* was already completely inhibited (Figure 4.1A).

The differences between the  $F_{420}H_2$  dehydrogenation activities of the respective methanoarchaea are also emphasized when examining the Michaels Menten kinetic plots in the presence/absence of DPI of the respective methanoarchaeons (Figure 4.1B and 4.2B). The apparent  $K_m$  for phenazine for Ml. tindarius drastically increases from 266  $\mu$ M to 1168  $\mu$ M (3  $\mu$ M DPI) and 2313  $\mu$ M (9  $\mu$ M DPI) (Vmax = 0.5  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>); in contrast, the same concentrations of DPI have a minor effect on the  $F_{420}H_2$  dehydrogenation activity of Ms. barkeri Fusaro, where the  $K_m$  increases from 95  $\mu$ M to 213  $\mu$ M (50  $\mu$ M DPI) and 317  $\mu$ M (100  $\mu$ M DPI), (Vmax = 0.4  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>). The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of Ms. barkeri Fusaro is further examined in Chapter 6.

### 4.5. Phenazine-dependent $F_{420}H_2$ dehydrogenation activity in other methanoarchaea

Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was also detected in methanoarchaea outside of the *Methanosarcinaceae*; this activity is not restricted to the use of methanol as methanogenic substrate since this activity could be detected in the cell-free extract of methanoarchaea growing on  $H_2$  and  $CO_2$  or acetate (Table 4.1). The observed activity was also inhibited by the presence of DPI, to varying degrees.

# 4.5.1. Phenazine-dependent $F_{420}H_2$ dehydrogenation activity in the Methanomicrobiales: Methanospirillum hungatei GP1

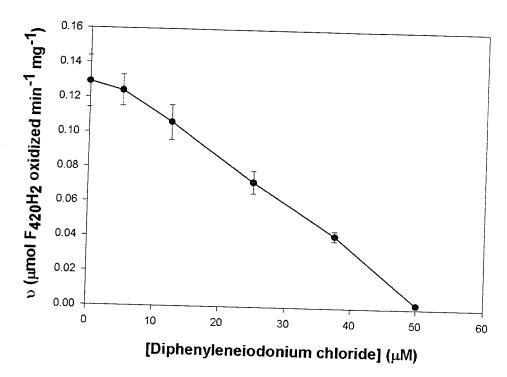
Methanospirillum hungatei GP1 is a hydrogenotrophic methanoarchaeon from the order Methanomicrobiales. An  $F_{420}$ -reducing hydrogenase has been purified from this

methanoarchaeon (Sprott *et al.* 1987, Choquet and Sprott 1991), but unlike other hydrogenotrophic methanoarchaea, an F<sub>420</sub>-nonreducing hydrogenase is apparently not expressed by this methanoarchaeon (Sprott *et al.* 1987). The F<sub>420</sub>-nonreducing hydrogenase is an important component of the H<sub>2</sub>:heterodisulfide oxidoreductase complex which catalyzes the reduction of CoM-S-S-CoB when H<sub>2</sub> and CO<sub>2</sub> are used as methanogenic substrate (Thauer *et al.* 1993, Thauer 1998).

Our studies have detected an  $F_{420}H_2$  dehydrogenation activity in the cell-free extract and membranes of this methanoarchaeon. Phenazine acts as an electron acceptor for the  $F_{420}H_2$  dehydrogenation activity, but its specificity as an electron acceptor was similar to methyl viologen (Table 4.1). In the absence of an  $F_{420}$ -nonreducing hydrogenase, it is of interest that the source of the  $F_{420}H_2$  dehydrogenation activity be investigated. The apparent  $K_m$  for phenazine of the  $F_{420}H_2$  dehydrogenation activity in cell-free extract is 242.6  $\mu$ M (Vmax = 0.39  $\mu$ mol  $F_{420}H_2$  oxidized ·min<sup>-1</sup>·mg<sup>-1</sup>), which is similar to the value obtained for Ms. mazei Gö1 (Abken et al. 1998) and Ml. tindarius (this thesis). The  $F_{420}H_2$  dehydrogenation activity of Msp. hungatei GP1 is inhibited by the addition of DPI ( $Ic_{50} = 25\mu$ M, Figure 4.3A), not to the same extent as Ml. tindarius, but more similar to the inhibition observed with Ms. barkeri Fusaro. Addition of 12.5 and 25  $\mu$ M DPI increased, respectively, the apparent  $K_m$  increased from 242.6 to 308.1 and 403.4  $\mu$ M (Vmax = 0.38 and 0.37 $\mu$ mol  $F_{420}H_2$  oxidized ·min<sup>-1</sup>·mg<sup>-1</sup>, respectively) (Figure 4.3B).

In the absence of the  $F_{420}$ -nonreducing hydrogenase, it is possible that Msp. hungatei uses an  $F_{420}H_2$ -dependent mechanism to reduce CoM-S-S-CoB, in a manner





#### B.

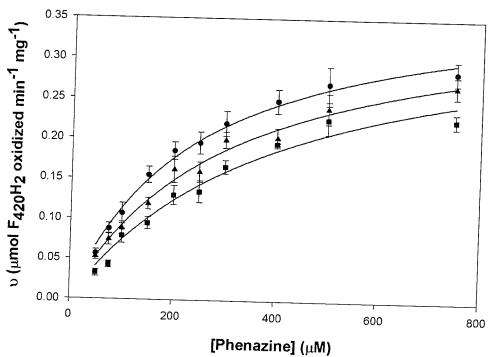


Figure 4.3. (A) Sensitivity of the phenazine-dependent  $F_{420}H_2$  dehydrogenase activity of *Methanospirillum hungatei* GP1 (in cell-free extract) to diphenyleneiodonium (DPI) chloride and (B) Michaelis Menten kinetics plot of the phenazine- $F_{420}H_2$  dehydrogenase activity in the absence/presence of DPI.

•, no DPI added. •, + 12.5  $\mu$ M DPI. •, + 25  $\mu$ M DPI.

similar to that used by *Methanococcus voltae*, as described by Brodersen *et al.* (1999). This notion is further examined in Chapter 5.

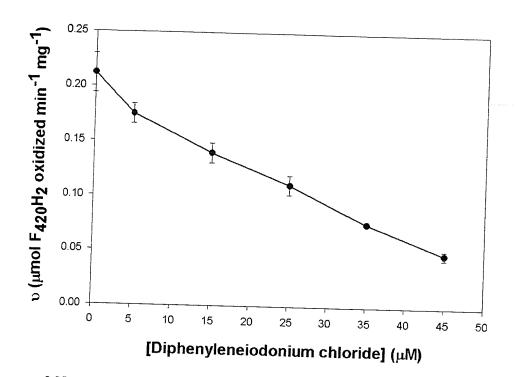
# 4.5.2. Phenazine-dependent $F_{420}H_2$ dehydrogenation activity in the Methanococcales: Methanococcus voltae

The order *Methanococcales* consist of coccoid, marine methanoarchaea that grow using H<sub>2</sub> or formate to reduce CO<sub>2</sub> to CH<sub>4</sub>, including *Methanocaldococcus jannaschi*, *Methanococcus voltae*, and *Methanococcus thermolithotrophicus* (Boone *et al.* 1993). *Mc. voltae*, named for the Italian physicist Alessandro Volta, possesses two forms each of the F<sub>420</sub>-reducing hydrogenase and F<sub>420</sub>-nonreducing hydrogenase (± selenium) (Sorgenfrei *et al.* 1997), but a distinct Fpo complex is not expected to be present in this methanoarchaeon; the genes encoding for the Fpo complex were not found in the genome of *Mc. jannaschi* (Brodersen *et al.* 1999) or *Mc. maripaludis* (NC\_005791).

Brodersen *et al.* (1999b) reported membrane-bound  $F_{420}H_2$ -dependent CoM-S-S-CoB reduction in *Mc. voltae*. The  $F_{420}H_2$  oxidation activity was purified from the membranes of *Mc. voltae*; the source of activity was the  $F_{420}$ -reducing hydrogenase, whose purification had been previously reported by Muth *et al.* (1987). The activity was assayed using methyl viologen and metronidazole by Brodersen *et al.* (1999b), but our findings indicate that phenazine can also used as an electron acceptor, with 3-fold higher activity (Table 4.1).

The apparent  $K_m$  for phenazine for the  $F_{420}H_2$  dehydrogenation activity was found to be 915.4  $\mu$ M phenazine (Vmax = 0.60  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>), which indicates an extremely low affinity for phenazine as an electron acceptor compared to Ms. mazei Gö1 (Abken et al. 1998) or Ml. tindarius and also relative to the  $K_m$  values

A.



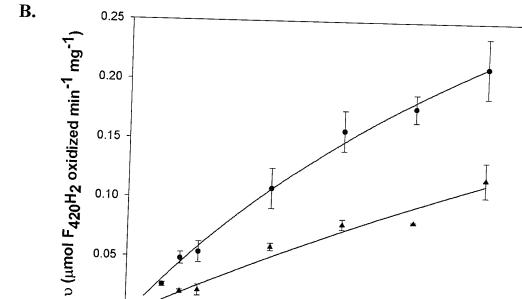


Figure 4.4. (A) Sensitivity of the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of *Methanococcus voltae* (in cell-free extract) to diphenyleneiodonium (DPI) chloride and (B) Michaelis Menten kinetics plot of the phenazine-F<sub>420</sub>H<sub>2</sub> dehydrogenase activity in the absence/presence of DPI.

[Phenazine] (µM)

•, no DPI added.  $\triangle$ , + 15  $\mu$ M DPI.

0.00

derived for the other methanoarchaea. The  $F_{420}H_2$  dehydrogenation activity is also inhibited by the presence of DPI (Ic<sub>50</sub> = 25  $\mu$ M DPI, Figure 4A); addition of 15  $\mu$ M DPI increased the  $K_m$  for phenazine from 915.4 to 2295.3  $\mu$ M phenazine(Vmax = 0.63  $\mu$ mol  $F_{420}H_2$  oxidized ·min<sup>-1</sup>·mg<sup>-1</sup>) (Figure 4.4B).

# 4.5.3. Phenazine-dependent $F_{420}H_2$ dehydrogenation activity in the Methanobacteriales

Members of the order *Methanobacteriales* form a diverse group that typically grow exclusively on H<sub>2</sub> and CO<sub>2</sub> as substrate for methanogenesis, with the exception of *Msph. stadtmanae* and *Msph. cuniculi*, which cannot use H<sub>2</sub> and CO<sub>2</sub>, but reduce methanol to methane only in the presence of H<sub>2</sub> (Miller and Wolin, 1983, 1985, Biavati *et al.* 1988, Boone *et al.* 1993). Several members of this order were tested for phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity, including *Mtb. marburgensis*, *Mb. bryantii*, and *Msph. stadtmanae*.

### 4.5.3.1. Methanothermobacter marburgensis

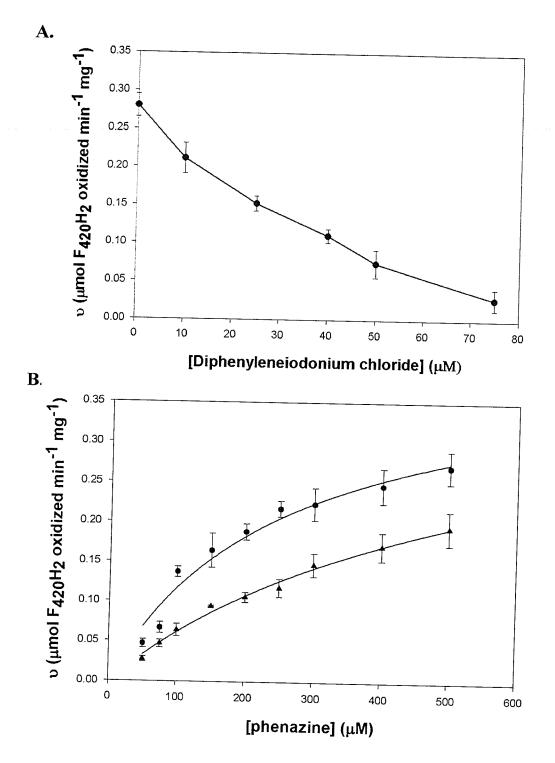
Mtb. marburgensis is a thermophilic methanoarchaeon that uses  $H_2$  and  $CO_2$  as methanogenic substrate. The genome of a related methanoarchaeon, Mtb. thermoautotrophicus  $\Delta H$ , encodes several hydrogenases, including the  $F_{420}$ -reducing hydrogenase (Frh), the  $F_{420}$ -nonreducing hydrogenase (Mvh), and two isoforms of energy-conserving hydrogenase (Ech), EhA and EhB, but does not encode a distinct Fpo complex (Smith et al. 1997, Bäumer et al. 2000).

There is no apparent necessity for the  $F_{420}H_2$  dehydrogenation activity in Mtb. marburgensis, as this methanoarchaeon possesses an  $F_{420}$ -nonreducing hydrogenase in

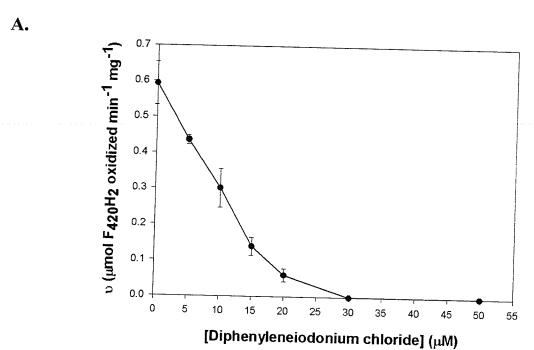
addition to the  $F_{420}$ -reducing hydrogenase (Braks *et al.* 1994). Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity, which is 4-fold higher relative to the methyl viologen-dependent activity, was detected in the cell-free extract, although this activity was approximately 2-fold lower compared to the activity of *Ml. tindarius* (Table 4.1). The apparent  $K_m$  for phenazine of the  $F_{420}H_2$  dehydrogenation activity in cell-free extract was determined to be 254.2  $\mu$ M phenazine (Vmax = 0.41  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>), which is similar to the  $K_m$  value obtained for *Ml. tindarius*. The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was inhibited by the addition of DPI ( $Ic_{50} = 30 \ \mu$ M DPI, Figure 4.5A), although the degree of inhibition observed with *Mtb. marburgensis* was not as acute as that of *Ml. tindarius*. Addition of 30  $\mu$ M DPI increased the apparent  $K_m$  from 254.2 to 578.0  $\mu$ M DPI (Vmax = 0.42  $\mu$ mol  $F_{420}H_2$  oxidized ·min<sup>-1</sup>·mg<sup>-1</sup>) (Figure 4.5B).

#### 4.5.3.2. Methanobacterium bryantii

Like *Mtb. marburgensis*, *Mb. bryantii* is also a hydrogenotrophic methanoarchaeon growing on  $H_2$  and  $CO_2$  that has a phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in cell-free extract. In the absence of inhibitor, the apparent  $K_m$  for phenzine was determined to be 318.1  $\mu$ M (Vmax = 0.92  $\mu$ mol  $F_{420}H_2$  oxidized ·min ·mg ·l·, indicating lower affinity for phenazine compared to *Ml. tindarius* and *Ms. mazei* Gö1, as well as *Ms. barkeri* Fusaro and *Msp. hungatei* GP1. The  $F_{420}H_2$  dehydrogenation activity was more sensitive to DPI than the activity of *Mtb. marburgensis* and *Msp. hungatei* GP1, but less so than *Ml. tindarius* ( $Ic_{50} = 10 \mu$ M DPI, Figure 4.6A). Addition of 7  $\mu$ M DPI increased the apparent  $K_m$  from 318.1 to 415.9  $\mu$ M phenazine ( $Vmax = 0.73 \mu$ mol  $F_{420}H_2$  oxidized ·min ·l·mg ·l·) (Figure 4.6B).



**Figure 4.5**. Sensitivity of the phenazine-dependent  $F_{420}H_2$  dehydrogenase activity of *Methanothermobacter marburgensis* (in cell-free extract) to diphenyleneiodonium (DPI) chloride (A) and Michaelis Menten kinetics plot of the phenazine-dependent  $F_{420}H_2$  dehydrogenase activity in the absence/presence of DPI (B). •, no DPI added. •, + 30  $\mu$ M DPI.



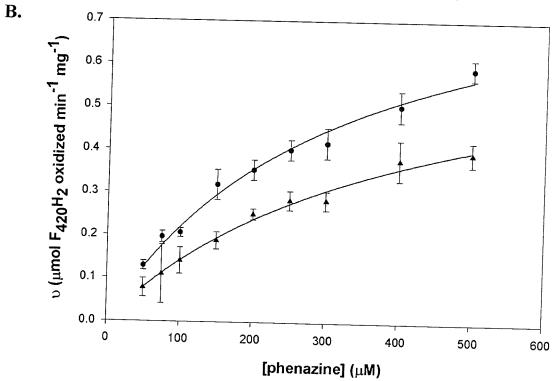


Figure 4.6. (A) Sensitivity of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Methanobacterium bryantii* (in cell-free extract) to diphenyleneiodonium (DPI) chloride and (B) Michaelis Menten kinetics plot of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in the absence/presence of DPI. •, no DPI added. •, + 7  $\mu$ M DPI.

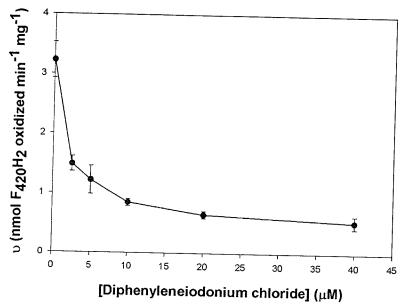
### 4.5.3.3. Methanosphaera stadtmanae

A methyl viologen-dependent  $F_{420}H_2$  dehydrogenation activity is present primarily in the cell-free extract of *Msph. stadtmanae* (Wong *et al.* 1994), although the activity measured is significantly lower than the related activities found in the cell-free extracts of most other methanoarchaea tested for this survey, and is also aerobically unstable.

This activity was reinvestigated using phenazine as electron acceptor; the apparent  $K_m$  for phenazine was quantified as 138.2  $\mu$ M phenazine (Vmax = 0.003  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>), which indicates higher affinity for phenazine relative to  $Ms.\ mazei$  Gö1 (Abken  $et\ al.\ 1998$ ) and  $Ml.\ tindarius$ . The  $F_{420}H_2$  dehydrogenation activity is much lower relative to the other methanoarchaea tested, consistent with the low levels of coenzyme  $F_{420}$  (0.16 nmol mg<sup>-1</sup> protein) found in this methanoarchaeon (Wong  $et\ al.\ 1994$ ). Moreover, this activity was inhibited by DPI (Ic $_{50}$  = 2.5  $\mu$ M DPI Figure 7A). Addition of DPI greater than 10  $\mu$ M did not have a significant effect on the  $F_{420}H_2$  dehydrogenation activity, which remains relatively constant as [DPI] increases up to 40  $\mu$ M (Figure 4.6A).

Addition of 5  $\mu$ M DPI increased the apparent  $K_m$  from 138.2 to 169.9  $\mu$ M phenazine, but the Vmax decreased from 0.003 to 0.001 (as low as 0.0005)  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> (Figure 7B); this was in contrast to the activity of the other methanoarchaea tested where the Vmax remains constant. These results suggest that there may be a different mechanism of inhibition of the  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae* by DPI compared to the inhibition observed in other methanoarchaea, or that there may be multiple phenazine-dependent  $F_{420}H_2$  oxidizing

A.



B. 3.0 υ (nmol F<sub>420</sub>H<sub>2</sub> oxidized min<sup>-1</sup> mg<sup>-1</sup> 2.5 2.0 1.5 1.0 Ī 0.5 0.0 0 50 100 150 200 250 300 350 400 450 500 550 [phenazine] (µM)

Figure 4.7. (A) Sensitivity of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Methanosphaera stadtmanae* (in cell-free extract) to diphenyleneiodonium (DPI) chloride and (B) Michaelis Menten kinetics plot of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in the absence/presence of DPI.  $\bullet$ , no DPI added.  $\blacktriangle$ , + 5  $\mu$ M DPI.

activities in the cell-free extract of Msph. stadtmanae. The  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae is further examined in Chapter 7.

The genome of *Msph. stadtmanae* was recently published, and sequences coding for the subunits of the Fpo complex were not detected in the genome (Fricke *et al.* 2006). However, genes encoding the  $F_{420}$ -reducing hydrogenase and an  $F_{420}H_2$  oxidase are found in the genome. Could one, or both, of these proteins be the source of the low levels of phenazine-dependent  $F_{420}H_2$  dehydrogenation activity?

### 4.6. Phenazine-dependent $F_{420}H_2$ dehydrogenation – a widespread phenomenon?

The presence of a phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity appears to be ubiquitous amongst the methanoarchaea regardless of nutritional requirement or phylogeny. In methylotrophic members of the *Methanosarcinaceae*, the F<sub>420</sub>H<sub>2</sub> dehydrogenase activity is an essential component of the F<sub>420</sub>H<sub>2</sub>:CoM-S-S-CoB oxidoreductase complex (Deppenmeier *et al.* 2002, 2004). The physiological significance of this activity for methanoarchaea outside of the *Methanosarcinaceae* is unclear; it may be an artifact in the same way as the use of methyl viologen as an electron acceptor for assay of the F<sub>420</sub>-reducing hydrogenase (Frh) or F<sub>420</sub>-nonreducing hydrogenase (Mvh, Vho). The genomes of the hydrogenotrophic methanoarchaea *Methanothermobacter thermoautotrophicus* and *Methanocaldococcus jannaschii* indicate that the gene clusters encoding the distinct Fpo complex are not present in these methanoarchaea (Smith *et al.* 1997, Bult *et al.* 1996, Bäumer *et al.* 2000). The methanogenic pathways for these methanoarchaea do not require the presence of this activity, as CoM-S-S-CoB reduction is dependent on the presence of the F<sub>420</sub>-

nonreducing hydrogenase which is found in all of the aforementioned methanoarchaea as part of the H<sub>2</sub>:CoM-S-S-CoB oxidoreductase complex, with the exception of *Msp. hungatei* JF1 which does not possess the F<sub>420</sub>-nonreducing hydrogenase, based on biochemical and genomic analyses (Sprott *et al.* 1987, this thesis).

The significance of phenazine as an electron acceptor is also not clear, as methanophenazine has not been isolated from methanoarchaea outside of the *Methanosarcinaceae* (Beifuss and Tietze 2005). 2-OH-phenazine can be used as electron acceptor for both the  $F_{420}$ -reducing (Frh) and  $F_{420}$ -nonreducing (Mvh) hydrogenases of *Mtb. thermoautotrophicus*  $\Delta H$ , but methanophenazine is not expected to be present in this methanoarchaeon (Meuer *et al.* 1999, Beifuss and Tietze 2005). Although the methanophenazine can be chemically synthesized, the biosynthetic pathway is not known.

# 4.7. Possible sources of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity outside the Methanosarcinales

The purified  $F_{420}H_2$  dehydrogenase of Ms. mazei Gö1 is a hydrophobic membrane-bound flavoprotein capable of accepting and converting hydride ions, via the FAD component, to single electrons. This is essential as the substrates used to assay the  $F_{420}H_2$  dehydrogenase, including methyl viologen, quinones, and various phenazine compounds, are single electron acceptors incapable of reduction via hydride transfer (Abken and Deppenmeier 1997, Abken et al. 1998). The flavoprotein inhibitor DPI, which is structurally similar to phenazine, competitively inhibits the  $F_{420}H_2$  dehydrogenase activity of Ms. mazei Gö1 (Brodersen et al. 1999). A similar inhibition profile was observed with our studies of the phenazine-dependent  $F_{420}H_2$  dehydrogenase

activity of *Ml. tindarius*, a methylotrophic member of the *Methanosarcinaceae* that also contains an Fpo complex (Haase *et al.* 1992).

Methanoarchaea contain several enzymes that can use coenzyme  $F_{420}H_2$  as electron donor, including the  $F_{420}H_2$ :NADP $^+$  oxidoreductase, the methylene- $H_4$ MPT dehydrogenase, the methylene- $H_4$ MPT reductase, the  $F_{420}$ -reducing hydrogenase, and the  $F_{420}H_2$  oxidase.

 $\mathbf{F}_{420}\mathbf{H_2}$ :NADP<sup>+</sup> oxidoreductase can be eliminated as a source of the  $\mathbf{F}_{420}\mathbf{H_2}$  dehydrogenase activity as this enzyme reduces NADP<sup>+</sup> with  $\mathbf{F}_{420}\mathbf{H_2}$  using a ternary, sequential reaction, *via* direct hydride transfer (Warkentin *et al.* 2001), and can not reduce single electron acceptors such as methyl viologen or phenazine.

The cytoplasmic proteins **methylene-H<sub>4</sub>MPT dehydrogenase** and **methylene-H<sub>4</sub>MPT reductase** catalyze, respectively, the reversible reduction of  $N^5$ ,  $N^{10}$ -methenyl-H<sub>4</sub>MPT to  $N^5$ ,  $N^{10}$ -methylene-H<sub>4</sub>MPT and  $N^5$ ,  $N^{10}$ -methylene-H<sub>4</sub>MPT to  $N^5$ -methyl-H<sub>4</sub>MPT (Thauer *et al.* 1993). Both reactions proceed *via* stereospecific hydride transfer, with a ternary complex kinetic mechanism (Ma and Thauer 1990, Vaupel and Thauer 1995). Neither enzyme possesses prosthetic groups such as FAD or FMN, which act as  $2e^{-1}$ 1e switches. As such, these enzymes cannot be the source of the  $F_{420}$ H<sub>2</sub> dehydrogenase activity, since reduction of methyl viologen and phenazine both occur *via* single electron transfers.

One plausible source of  $F_{420}H_2$  dehydrogenation activity may be the  $F_{420}$ reducing hydrogenase. The  $F_{420}$ -reducing hydrogenases are flavoproteins that behave
on hydrophobic column matrices as very hydrophobic proteins, suggesting a functional
association with the cytoplasmic membrane (Sprott *et al.* 1987, Sprott and Beveridge

1993). Immunogold labeling studies of a number of methanoarchaea indicate that the  $F_{420}$ -reducing hydrogenase is located on the inner surface of the cytoplasmic membrane (Baron *et al.* 1989b, Lünsdorf *et al.* 1991, Braks *et al.* 1994). Purified  $F_{420}$ -reducing hydrogenase from *Mtb. thermoautotrophicus* also catalyzes the  $H_2$ -dependent reduction of 2-OH-phenazine (Meuer *et al.* 1999); it would be plausible to envision an  $F_{420}H_2$ -dependent mechanism for 2-OH-phenazine-reduction using this enzyme.

Experiments by Brodersen *et al.* (1999b) demonstrated membrane-bound  $F_{420}H_2$ -dependent reduction of CoM-S-S-CoB in *Mc. voltae*. When the source of the  $F_{420}H_2$  dehydrogenation-like activity was purified from the membranes, the end result was the isolation of the  $F_{420}$ -reducing hydrogenase; this enzyme catalyzed the  $F_{420}H_2$  dependent reduction of methyl viologen. Thus, the  $F_{420}$ -reducing hydrogenase may be a physiologically versatile enzyme, being able to transfer  $e^-$  to  $F_{420}$  from  $H_2$  as a hydrogenase, yet also able to oxidize  $F_{420}H_2$  and transferring the  $e^-$  to a number of electron carriers, such as methyl viologen and phenazine. A similar scenario may be envisioned for *Msp. hungatei* GP1 and *Ms. barkeri* Fusaro; this is examined further in Chapters 5 and 6, respectively.

Recently, an F<sub>420</sub>H<sub>2</sub>-dependent oxidase (FlavoproteinA, FprA) was purified from *Methanobrevibacter arboriphilus*, a mesophilic methanoarchaeon isolated from the hindgut of termites (Seedorf *et al.* 2004). FprA catalyzes the following reaction (Seedorf *et al.* 2004, 2005):

$$2F_{420}H_2 + O_2 \rightarrow 2F_{420} + H_2O$$

The presence of an  $F_{420}H_2$  oxidase enables the strictly anaerobic methanoarchaeon to thrive in microaerophilic environments (Seedorf *et al.* 2004).

The amino acid sequence of FprA from *M. arboriphilus* was found to be similar to the homologous protein from a number of methanoarchaea, including *Mtb*.

marburggensis (71%), *Mc. jannaschii* (69%), and *Ms. acetivorans* (44%) (Seedorf *et al.* 2004). FprA is a cytoplasmic F<sub>420</sub>H<sub>2</sub>-dependent flavoenzyme; it is possible that this protein may be a source of the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity found in the cell-free extract of methanoarchaea outside of the *Methanosarcinaceae*.

Clearly, the observation of phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in cell-free extracts is not an indicator of the importance of this activity in non-methylotrophic methanoarchaea. In order to elucidate a possible physiological role for this activity, the source of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity will have to be purified for further analysis.

# 5. Purification of the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity from *Methanospirillum hungatei* strain GP1

#### 5.1. Introduction

Methanospirillum (Msp.) hungatei is a hydrogenotrophic spiral-shaped methanoarchaeon discovered by P.H. Smith (1966) in a sample of sewer sludge, and characterized by Ferry et al. (1974). Msp. hungatei strain GP1, the strain used in our studies, was described by Patel et al. (1976), who isolated this methanoarchaeon from a pear waste fermenter. Msp. hungatei GP1 is a member of the order Methanomicrobiales, which is closely related to the order Methanosarcinales, based on phylogenetic analyses of ribosomal proteins, methanogenesis enzymes, and methanogenesis coenzyme biosynthesis enzymes (Bapteste et al. 2005). While Msp. hungatei GP1 may share some properties with the non-methylotrophic members of the Methanosarcinales, like other methanoarchaea outside of this order Msp. hungatei GP1 does not possess cytochromes and may share other properties with methanoarchaea from the other methanogenic orders.

In contrast to most other methanoarchaea growing on H<sub>2</sub> and CO<sub>2</sub>, an F<sub>420</sub>nonreducing hydrogenase is apparently not present in *Msp. hungatei* GP1, but an F<sub>420</sub>
H<sub>2</sub>ase has been purified from this methanoarchaeon (Sprott *et al.* 1987, Choquet and
Sprott 1991). The F<sub>420</sub>-nonreducing hydrogenase has been isolated and studied from
hydrogenotrophic methanoarchaea such as *Methanothermobacter* (*Mtb.*)

thermoautotrophicus, *Methanobacterium* (*Mb.*) formicicum, *Methanobacterium* (*Mb.*)

fervidus, *Methanococcus* (*Mc.*) jannaschii, and *Methanococcus* (*Mc.*) voltae, and also
from two methylotrophic methanoarchaea, *Methanosarcina* (*Ms.*) barkeri MS and *Methanosarcina* (*Ms.*) mazei Göl (Jin *et al.* 1983, Adams *et al.* 1986, Shah and Clark

1990, Hedderich *et al.* 1990, Deppenmeier *et al.* 1992, Kemner and Zeikus 1994, Deppenmeier *et al.* 1995). The F<sub>420</sub>-nonreducing hydrogenase is part of the H<sub>2</sub>:heterodisulfide oxidoreductase system, and is essential in most hydrogenotrophic methanoarchaea for the reduction of the heterodisulfide complex (CoM-S-S-CoB), whereas the major physiological role of the F<sub>420</sub>-reducing hydrogenase (F<sub>420</sub> H<sub>2</sub>ase) is to produce F<sub>420</sub>H<sub>2</sub>, which is in turn used to reduce H<sub>4</sub>MPT-bound C<sub>1</sub> intermediates to the level of CH<sub>3</sub>-H<sub>4</sub>MPT (Thauer *et al.* 1993, Shima *et al.* 2002). In the absence of an F<sub>420</sub>-nonreducing H<sub>2</sub>ase, it is not clear how electrons are coupled to the reduction of CoM-S-S-CoB in *Msp. hungatei* GP1.

Work in our laboratory had indicated the possibility of a phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in  $Msp.\ hungatei$  GP1, similar to the respective activities isolated from Methanosarcina (Ms.) mazei Gö1 (Abken  $et\ al.$  1998) and Methanolobus (Ml.) tindarius (Chapter 4). Given the close phylogenetic relationship between the orders Methanosarcinales and the Methanomicrobiales, it was not surprising to find such an activity in  $Msp.\ hungatei$  GP1. Although significant  $F_{420}H_2$  dehydrogenation activities are found in the cytoplasmic fractions ( $\sim$ 80%), the  $F_{420}H_2$  dehydrogenase (aka Fpo ( $F_{420}H_2$  phenazine oxidoreductase complex) of  $Ml.\ tindarius$  and  $Ms.\ mazei$  Gö1 were both purified from the cell membranes of the respective methanoarchaea, the hypothesis being that the  $F_{420}H_2$  dehydrogenase is a major component of the membrane-bound electron transport chain (Deppenmeier  $et\ al.$  1990a, Haase  $et\ al.$ 1992, Abken and Deppenmeier 1997). In these methylotrophic methanoarchaea, methanophenazine is believed to be the physiological electron acceptor for the  $F_{420}H_2$  dehydrogenase activity (Abken  $et\ al.$  1998, Bäumer  $et\ al.$  2000). As a substitute for methanophenazine,

2-hydroxy-phenazine (2-OH-phenazine) was synthesized and used to assay the  $F_{420}H_2$  dehydrogenase activity of  $Ms.\ mazei$  Gö1 in aqueous buffer; this is due to the extremely hydrophobic nature of methanophenazine (Beifuss and Tietze 2005). 2-OH-phenazine is not commercially available and must be synthesized, making phenazine a much more attractive alternative. Our research indicated that phenazine could be used as an electron acceptor when measuring the  $F_{420}H_2$  dehydrogenase activity of  $Ml.\ tindarius$ : the apparent Km for phenazine was determined to be 267  $\mu$ M, which is similar to the value derived for the phenazine-dependent  $F_{420}H_2$  dehydrogenase activity of  $Ms.\ mazei$  Gö1 (Km<sub>phenazine</sub> = 250  $\mu$ M, Abken  $et\ al.\ 1998$ ). Therefore, phenazine was determined to be a suitable replacement for 2-OH-phenazine in our studies.

Purification of the  $F_{420}H_2$  dehydrogenase activity from the cell-free soluble fraction, as well as the membrane fraction, during the course of this thesis did not result in the isolation of a distinct Fpo complex as purified from  $Ms.\ mazei$  Göl. Instead, the purified activity was located on the  $F_{420}$   $H_2$  as of  $Msp.\ hungatei$  GPl.

#### 5.2. Materials and methods

For a comprehensive look at the materials and methods used for this chapter, the reader is referred to the following sections in Chapter 3:

- 3.2. Growth media
- 3.3. Cell lysis
- 3.5. Purification of the  $F_{420}H_2$  dehydrogenation activity from methanol-grown Methanosarcina barkeri Fusaro and Methanospirillum hungatei GP1
- 3.6. Enzyme assays
  - 3.6.1.  $F_{420}H_2$  dehydrogenation assays
  - 3.6.2. Hydrogenase assays

- 3.6.2.1. F<sub>420</sub>-reducing hydrogenase activity
- 3.6.2.2. Methyl viologen reducing hydrogenase activity
- 3.7. Gel electrophoresis
  - 3.7.1. SDS-PAGE gel electrophoresis
    - 3.7.1.2. Mass Spectroscopy analysis
  - 3.7.2. Native gradient PAGE gel electrophoresis
- 3.8. Genome sequence analysis

#### 5.3. Results

# 5.3.1. Preliminary results: Soluble vs membrane-bound $F_{420}H_2$ dehydrogenation activity

Following cell lysis, the vast majority (80-90%) of the  $F_{420}H_2$  dehydrogenation activity was found in the soluble cytoplasmic fraction of  $Msp.\ hungatei$  GP1, with 10-20% of the overall activity located on the washed cell membranes. Various chromatographic techniques were used to resolve the  $F_{420}H_2$  dehydrogenation activity of  $Msp.\ hungatei$  GP1; from the onset, our results indicated that the  $F_{420}H_2$  dehydrogenation activity eluted from the various chromatographic matrices with an  $F_{420}H_2$  as activity. These results were observed on a consistent basis whether the proteins were derived from the soluble cytoplasmic portion, or extracted from solubilized membranes.

DEAE Sephacel (anion exchange chromatography) was a logical first step for the enrichment of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity since this was the initial step used in the purification of the respective F<sub>420</sub>H<sub>2</sub> dehydrogenase activities from *Ml. tindarius* and *Ms. mazei* Gö1 (Haase *et al.* 1992, Abken *et al.* 1997). The F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was initially studied from the cell membranes of *Msp. hungatei* GP1; the activity was extracted from the CHAPS-solubilized cell membranes. The protein extract was

loaded onto the DEAE Sephacel column, and the  $F_{420}H_2$  dehydrogenase activity was eluted using a continuous 0-0.6 M NaCl gradient (Figure 5.1A). Results from previous studies (before this thesis) had demonstrated a small degree of separation of the  $F_{420}H_2$  dehydrogenation and  $F_{420}$   $H_2$  as activities, indicating that the two activities might be two separate entities. The use of DEAE Sephacel during the course of this thesis did not result in any separation of the  $F_{420}H_2$  dehydrogenation and  $F_{420}$   $H_2$  as activities. Manipulation of the NaCl gradient did not yield any further resolution of the activities. The DEAE Sephacel column was further washed with buffer containing 5 M NaCl to remove tightly bound protein;  $F_{420}H_2$  dehydrogenation activity was not detected in the eluate.

The fractions containing  $F_{420}H_2$  dehydrogenation activity were pooled and concentrated for a gel filtration step using G-200 (Figure 5.1B); further resolution of the  $F_{420}H_2$  dehydrogenation and  $F_{420}$   $H_2$  as activities was not obtained using this chromatographic technique. This seemed to indicate that the  $F_{420}H_2$  dehydrogenation activity may be linked to the  $F_{420}$   $H_2$  as of Msp.hungatei GP1.

Choquet and Sprott (1991) reported the enrichment of the  $F_{420}$   $H_2$ ase of Msp. hungatei GP1 using metal chelate affinity chromatography (MCAC), with  $Ni^{2+}$  chelated to the affinity matrix, as the initial step in their purification protocol. This appeared to be a viable step in our protocol to separate the  $F_{420}$   $H_2$ ase from the  $F_{420}H_2$  dehydrogenation activity. However, this step only enriched for the soluble  $F_{420}$   $H_2$ ase, as the cell membranes, containing ~10-20% of the overall activity, did not bind to the column. Thus, the membrane-bound activity passed through the column unimpeded, with no affinity for the  $Ni^{2+}$  (Choquet and Sprott 1991). However, our results demonstrated that

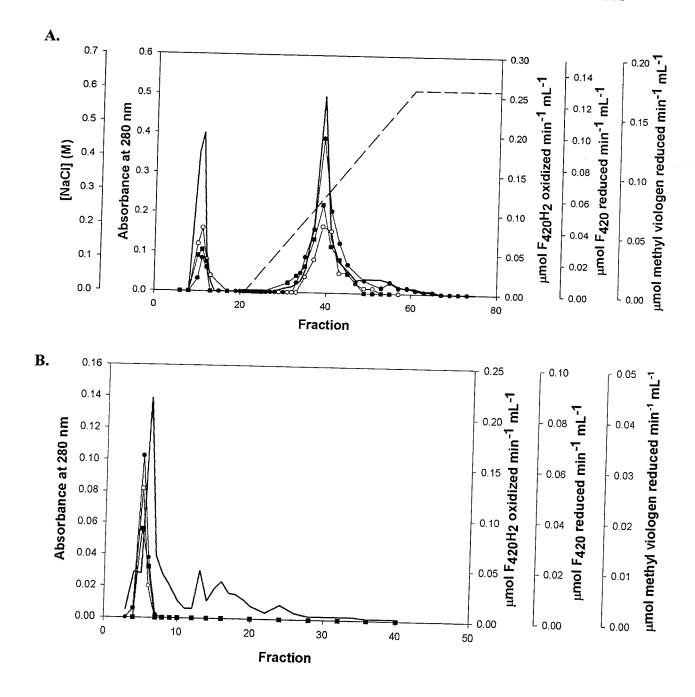


Figure 5.1. (A) Protein and activity profile of fractions collected from a DEAE Sephacel column, eluted using a linear 0-0.6 M NaCl gradient. Protein derived from solubilized cell membranes. Similar profiles observed using lower NaCl gradients (0-0.3 M NaCl). (B) Protein and activity profile of fractions collected from a G-200 column, using pooled and concentrated protein from a DEAE sephacel column. (—),  $A_{280}$ . (----), linear 0-0.6 M NaCl gradient. •, Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. •,  $F_{420}$ -reducing hydrogenase activity. •, Methyl viologen-reducing hydrogenase.

when the membranes are solubilized with detergent, such as CHAPS, the extracted activity does bind to the  $Ni^{2+}$  affinity column.

Since the Fpo complex is expected to be a membrane-bound protein complex, we could not preclude the possibility of an  $F_{420}H_2$  dehydrogenase activity located in the membrane fraction that may not be detected in the soluble fraction of the cell-free extract. Simultaneous experiments were set up to identify the source of  $F_{420}H_2$  dehydrogenation activity in the soluble and membrane portions.

A typical protein and activity profile of fractions collected from a  $Ni^{2+}$  affinity column, using the soluble  $F_{420}H_2$  dehydrogenation activity, is shown in Figure 5.2; similar profiles were observed using CHAPS-solubilized membrane-bound protein. The  $F_{420}H_2$  dehydrogenation,  $F_{420-}$  and methyl viologen-reducing hydrogenase activities all elute in the same peak with no apparent separation. The  $F_{420-}$  reducing and methyl viologen-reducing activities reside on the  $F_{420}$   $H_2$  ase; a distinct  $F_{420-}$  nonreducing  $H_2$  ase was not detected in *Msp. hungatei* GP1 (Sprott *et al.* 1987); this was confirmed by our own analyses.

Further steps were introduced in an attempt to separate and enrich the  $F_{420}H_2$  dehydrogenation activity; the results of subsequent steps using different chromatographic materials are shown in Figures 5.3 and 5.4 (soluble and membrane-bound proteins, respectively); these results also supported the association of the  $F_{420}H_2$  dehydrogenation and  $F_{420}$   $H_2$  as activities on the same protein complex.

Sucrose step gradient profiles are shown in Figures 5.3A and 5.4A, loaded with protein collected and pooled from the Ni<sup>2+</sup> affinity column. A 45-50% sucrose density

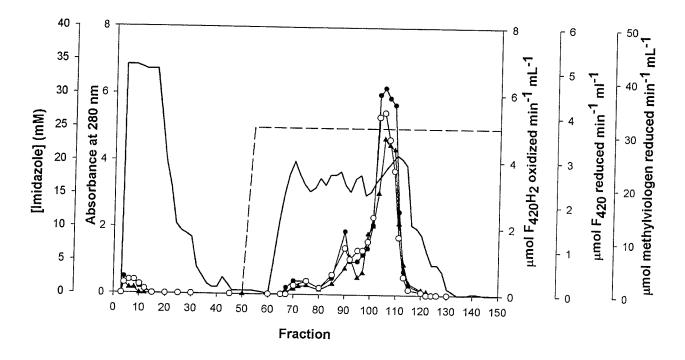
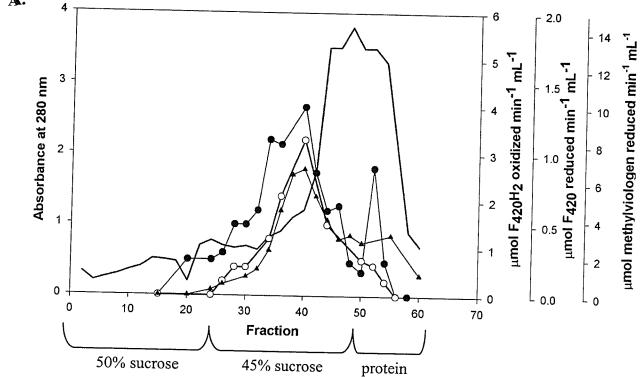


Figure 5.2. Protein and activity profile of fractions collected from a  $Ni^{2+}$  affinity column. Protein derived from the clarified cell-free extract (i.e. soluble fraction). Identical profile observed when using protein extracted from the cell membranes. Elution of protein from the column as described under 'Materials and methods'. (—),  $A_{280}$ . (----), 25 mM imidazole. •, Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. •,  $F_{420}$ -reducing hydrogenase activity. •, Methyl viologen-reducing hydrogenase.

gradient ultracentrifugation step (10 ml per step) was one of the major enrichment steps in the purification of the  $F_{420}$   $H_2$  ase of Msp. hungatei GP1 by Choquet and Sprott (1991). The  $F_{420}H_2$  dehydrogenation activity could not be separated from the  $F_{420}$   $H_2$  ase using the sucrose step gradient; similar results were observed when using a five-step (45-46-47-48-50%) gradient, or from a linear 45-50% sucrose density gradient. Protein pooled and concentrated from the 45-50% sucrose step gradient was then loaded onto a column of Sepharose 6B, an agarose-based gel filtration matrix with broad fractionation range for separating samples with diverse molecular weight  $(10-4x10^3 \text{ kDa})$ , Pharmacia Biotech). The  $F_{420}$   $H_2$  ase is a much larger protein relative to the expected  $F_{420}H_2$  dehydrogenase (7-to 8-fold larger), such that one would expect that the two activities should be well separated using this matrix. Figures 5.3B and 5.4B illustrate that both activities have similar molecular weights as they elute together in the same peaks. Furthermore, there are no apparent differences between the activity profiles of the soluble and membrane associated proteins.

However, the activities were not purified to homogeneity, and separation of the proteins from enriched extracts via electrophoresis indicated the presence of protein bands with molecular weights similar to the  $F_{420}H_2$  dehydrogenase of  $Ms.\ mazei$  Gö1.

Since enrichment of the  $F_{420}H_2$  dehydrogenation activity appears to occur alongside the hydrogenase activity, a pertinent question was raised at this point: was the  $F_{420}H_2$  dehydrogenation activity a distinct entity, or is it located on the  $F_{420}$   $H_2$  as of Msp. hungatei GP1? Based on these results, there was indication that the membrane-bound and soluble phenazine-dependent  $F_{420}H_2$  dehydrogenation activities of Msp. hungatei GP1 were located on the  $F_{420}$   $H_2$  ase. On the basis of the above experiments, a protocol



B.

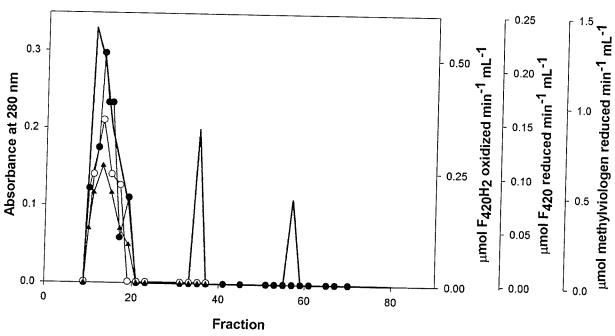


Figure 5.3. Protein and activity profile of fractions collected from: (A) 50-45% sucrose step gradient loaded with protein eluted from a Ni<sup>2+</sup>-affinity column and (B) Sepharose 6B column loaded with protein collected from the 50-45% sucrose step gradient. Protein derived from the clarified cell-free extract (soluble fraction). (—),  $A_{280}$ . •, Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. •,  $F_{420}$ -reducing hydrogenase activity. •, Methyl viologen-reducing hydrogenase activity.

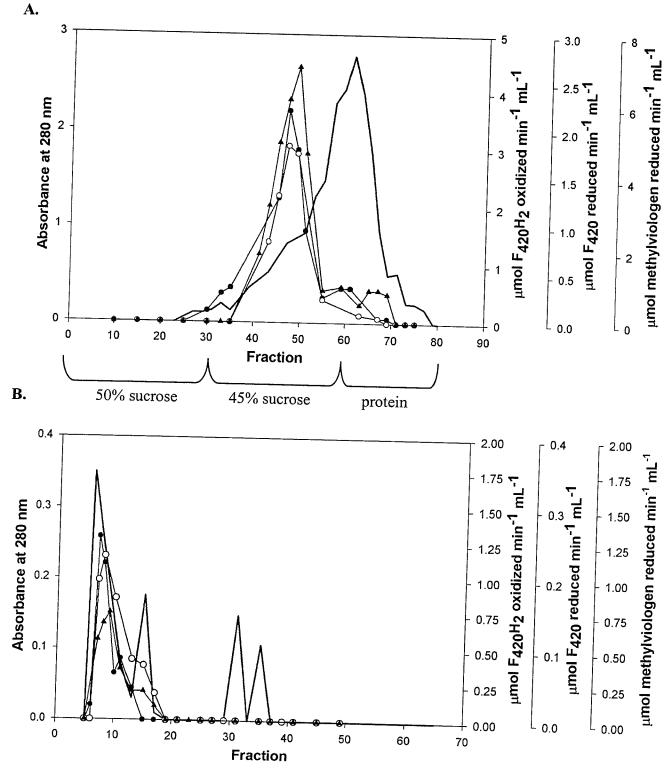


Figure 5.4. Protein and activity profile of fractions collected from: (A) 50-45% sucrose step gradient loaded with protein eluted from a Ni<sup>2+</sup>-affinity column and (B) Sepharose 6B column loaded with protein collected from the 50-45% sucrose step gradient. Protein solubilized and extracted from washed cell membranes. (—),  $A_{280}$ . •, Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. •,  $F_{420}$ -reducing hydrogenase activity. •, Methyl viologen-reducing hydrogenase activity.

was devised for the purification of the  $F_{420}H_2$  dehydrogenation activity to homogeneity, in an attempt to separate both activities. The  $F_{420}$   $H_2$  as also catalyzes  $F_{420}H_2$  oxidation; by selecting for the purification of this protein, we expected to separate this "artifact" activity from any other source of  $F_{420}H_2$  dehydrogenation present in the protein extract.

## 5.3.2. Purification of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity

The purification of the  $F_{420}H_2$  dehydrogenation activity is summarized in Table 5.1; purification of this activity was repeated three times, with representative data presented. Upon cell lysis, and after removal of the unlysed cells and debris, the cell-free extract was further centrifuged at 180000xg for 6 hours. This resulted in the sedimentation of both the soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activities into a tight dark brown pellet, referred to as the high speed pellet (HSP). The HSP was homogenized in 10 mM HEPES, pH 7.0, containing 0.5 M NaCl, and loaded onto the  $Ni^{2+}$  affinity column where the soluble  $F_{420}H_2$  dehydrogenation activity was enriched from the HSP using a single step addition of 25 mM imidazole, along with  $H_2$ -dependent  $F_{420}$ -reducing and methyl viologen-reducing activites (Figure 5.5). The membranes, containing up to 20% of the  $F_{420}H_2$  dehydrogenation activity (and  $F_{420}$   $H_2$  as activity), did not bind to the column and were pooled and stored at 4°C for further analysis.

The pooled fractions containing the soluble  $F_{420}H_2$  dehydrogenation activity were desalted and concentrated using an Amicon ultrafiltrator, and then loaded onto linear 3-20% native gradient PAGE gels (in each available lane); electrophoresis of the gels for

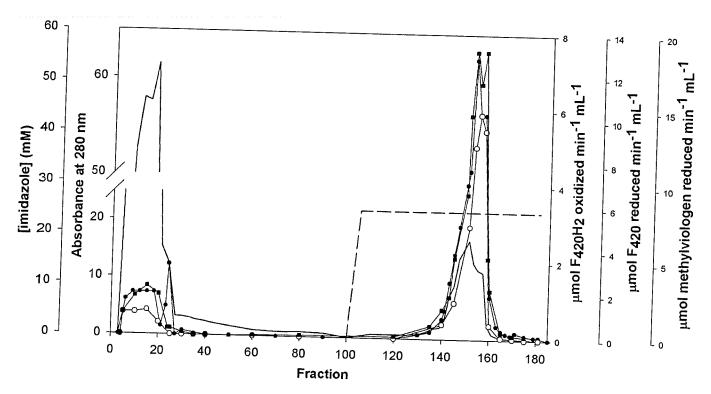
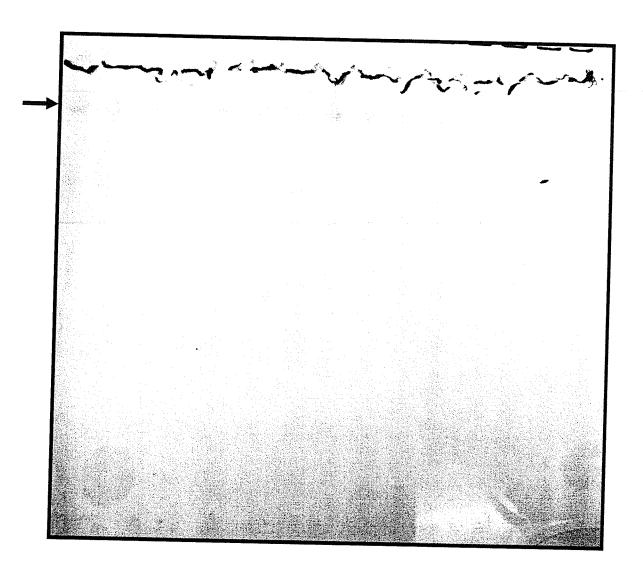


Figure 5.5. Protein and activity profile of fractions collected from a  $Ni^{2+}$  affinity column as described in 'Materials and methods'. Cell-free extract from *Methanospirillum hungatei* GP1 was centrifuged at 50 000 rpm, and the soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activities were separated from the extract into a tight brown pellet (high speed pellet). The pellet was suspended in buffer and applied to the column. Protein binding to the column was eluted from the column by the addition of 25 mM imidazole, while the membrane-bound activity did not have affinity for the  $Ni^{2+}$ . (—),  $A_{280}$ . (----), 25 mM imidazole.  $\bullet$ ,  $F_{420}H_2$  dehydrogenation activity.  $\circ$ ,  $F_{420}$ -reducing activity.  $\blacksquare$ , Methyl viologen-reducing activity.



**Figure 5.6**. Unstained 3-20% native gradient PAGE loaded with protein pooled and concentrated after elution of the soluble F<sub>420</sub>H<sub>2</sub> dehydrogenation activity from Ni<sup>2+</sup> affinity column. Concentrated extract was loaded into each lane, and the protein was electrophoresed overnight; high molecular weight protein band (marked with arrow) were cut from the gel for elution with Bio-Rad electro elutor. Other colored bands were also processed electro-elution.

Table 5.1. Purification of the soluble phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of Methanospirillum hungatei GP1. MCAC, metal chelate chromatography.

Purification stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Sp. activity <sup>‡</sup>	Purification
Cell-free extract	1808.5	547.8	100	0.3	-
High speed pellet	848.5	391.7	81.7	0.46	1.5-fold
MCAC pool	78.8	124.2	25.9	1.58	5.3-fold
Eluted protein <sup>‡‡</sup>	1.7	23.0	4.8	13.5	45.0-fold

approximately 15 hours (i.e. overnight) resulted in gels with a pattern shown in Figure 6. The presence of FAD and iron-sulfur clusters in the protein allows for the  $F_{420}$   $H_2$  as to be visualized without prior staining.

The distinctive, high molecular weight yellow-brown protein bands (marked in Figure 5.6) were excised from the gel and minced into 1-2 mm<sup>3</sup> pieces, followed by elution of the protein from the gel cubes using a Bio-Rad model 422 Electro Eluter. The eluted protein was resuspended in 50 mM HEPES, pH 7.0, with 5% glycerol and 0.05% Triton-X-100, and stored under N<sub>2</sub> at 4°C for further analysis. Other yellow brown bands are also visualized on the native gradient PAGE (Figure 5.6); neither  $F_{420}H_2$ dehydrogenation nor F<sub>420</sub> H<sub>2</sub>ase activities were detected in the protein extracted from the bands.

 $<sup>^{\</sup>dagger}\mu mol\ F_{420}H_2\ oxidized\cdot min^{-1}.$   $^{\ddagger}\mu mol\ F_{420}H_2\ oxidized\cdot min^{-1}\cdot mg^{-1}\ protein$   $^{\ddagger \ddagger}protein\ eluted\ from\ 3-20\%\ native\ gradient\ PAGE$ 

Table 5.2. Enrichment of the soluble F<sub>420</sub> reducing hydrogenase activity during the purification of the phenazine-dependent F420H2 dehydrogenation activity of Methanospirillum hungatei GP1. MCAC, metal chelate affinity chromatography.

Purification stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Sp. activity <sup>‡</sup>	Purification	DeH <sub>2</sub> ase/ H <sub>2</sub> ase
Cell-free extract	1808.5	366.1	100	0.22	-	1.37
High speed pellet	848.5	275.8	75.3	0.33	1.5-fold	1.39
MCAC pool	78.8	179.4	49.0	1.70	7.7-fold	0.93
Eluted protein <sup>‡‡</sup>	1.7	23.0	6.3	13.5	61.4-fold	1.00

The F<sub>420</sub> H<sub>2</sub>ase activity was monitored in tandem with the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity (Table 5.2), confirming the co-migration of the  $F_{420}H_2$ dehydrogenation and F<sub>420</sub> H<sub>2</sub>ase activities. No other source of F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was observed.

The purification of the membrane-bound F<sub>420</sub>H<sub>2</sub> dehydrogenation activity is summarized in Table 5.3. The protocol used for isolation of the membrane-bound  $F_{420}H_2$ dehydrogenation activity was similar to that of the soluble activity, but with the Ni<sup>2+</sup> affinity column step replaced with a three step (40-45-50%) sucrose step gradient. The pooled membranes were solubilized with CHAPS, concentrated with an Amicon ultrafiltrator, and loaded onto the three step sucrose gradient, followed by a 48 hour ultracentrifugation step (23,000 rpm). The phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation

<sup>†</sup>μmol F<sub>420</sub> reduced·min<sup>-1</sup>.

‡μmol F<sub>420</sub> reduced·min<sup>-1</sup>·mg<sup>-1</sup> protein

‡‡protein eluted from 3-20% native gradient PAGE

activity isolated from the solubilized cell membranes eluted from the sucrose step gradient along with methyl viologen- and F<sub>420</sub> H<sub>2</sub>ase activities (Figure 5.7). Similar results are observed when the membrane proteins were separated using various column chromatography techniques with varying chromatographic properties, including DEAE Sephacel (Figure 5.1), Ni<sup>2+</sup> affinity chromatography (Figure 5.2), gel filtration (Figure 5.4), and Phenyl Sepharose Cl-4B (data not shown).

Fractions containing  $F_{420}H_2$  dehydrogenation activity were pooled, diluted of sucrose and concentrated using an Amicon ultrafiltrator, followed by electrophoresis of the concentrated sample on a 3-20% native gradient PAGE. Protein bands could not be directly observed on the unstained gel due to the lower amounts of protein associated with the membrane, but staining of the outermost lanes of the gel revealed the positions of the protein bands in their respective lanes; these bands were cut from the gel, and protein was extracted using the BioRad electro-elutor, followed by homogenization in 50 mM HEPES, pH 7.0, with 5% glycerol and 0.05% Triton-X-100, and storage under  $N_2$  at 4°C. With the exception of the  $F_{420}H_2$  dehydrogenation activity located on the  $F_{420}H_2$  ase, no other source of  $F_{420}H_2$  dehydrogenation activity was detected in the solubilized membrane protein.

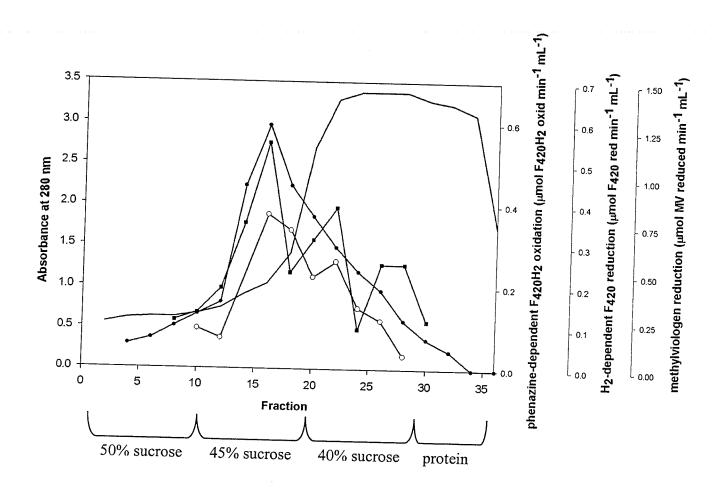


Figure 5.7. Protein and activity profile of fractions collected from a 40-45-50% sucrose gradient ultracentrifugation step. Protein was extracted from CHAPS-solubilized membranes of *Methanospirillum hungatei* GP1. (—),  $A_{280}$ . •, Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. •,  $F_{420}$ -reducing hydrogenase activity. •, Methyl viologen reducing hydrogenase activity.

Table 5.3. Purification of the membrane-bound phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Methanospirillum hungatei* GP1.

Purification stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Sp. activity <sup>‡</sup>	Purification
Washed mem	120.9	27.1	100	0.22	
Solubilized protein	46	9.0	33.2	0.20	0.9-fold
Sucrose pool	9.6	2.6	7.9	0.27	1.2-fold
Eluted protein <sup>‡‡</sup>	0.02	0.2	0.7	12.8	58.3-fold

<sup>†</sup>μmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>.

# 5.3.3. Properties of the $F_{420}H_2$ dehydrogenation activity of Methanospirillum hungatei GP1

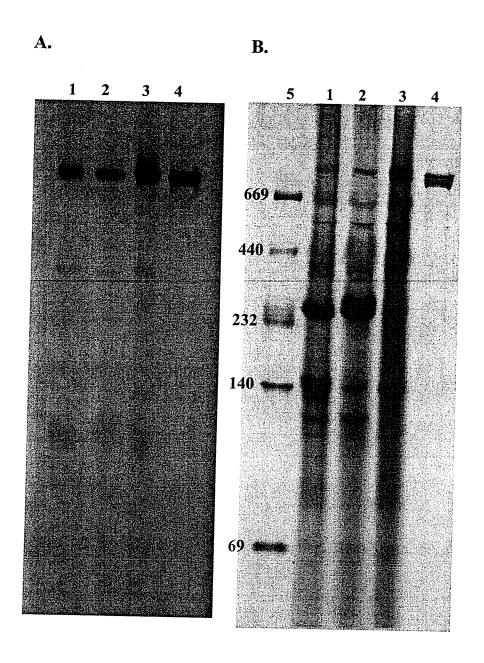
### 5.3.3.1. Molecular properties

Coomassie blue-staining of the purified protein following linear 3-20% native PAGE gel electrophoresis reveal a pair of distinct high molecular weight protein bands (Figure 5.8), both of which exhibit hydrogenase activity when stained with methyl viologen under reducing conditions; this is consistent with the molecular weight of the  $F_{420}$  H<sub>2</sub>ase (720 ± 84 kDa) of *Msp. hungatei* GP1, determined using gel filtration by Sprott *et al.* (1987).

The polypeptide pattern of the subunits on an SDS PAGE appears to be identical to the profile of the purified  $F_{420}$   $H_2$  as described by Choquet and Sprott (1991) (Figure 5.9 and 5.10). The purified protein from Msp. hungatei GP1 consists of 3

<sup>‡</sup> µmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein

<sup>‡‡</sup>protein eluted from 3-20% native gradient PAGE

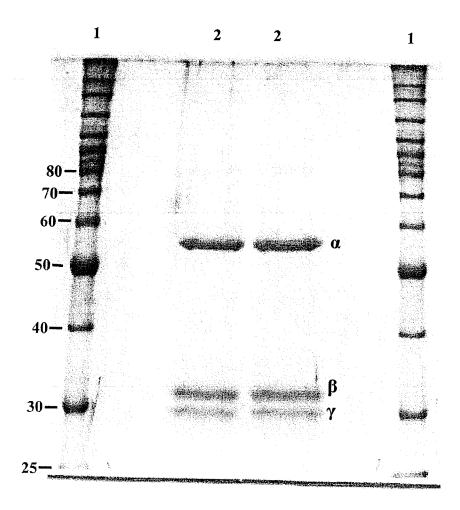


**Figure 5.8**. (A) Hydrogenase activity- and (B) coomassie blue-stained 3-20% native gradient PAGE showing progression of the purification of the soluble phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Methanospirillum hungatei* GP1. 1, Cell-free extract. 2, High speed pellet. 3, Ni<sup>2+</sup> affinity column pool. 4, purified protein. 5, molecular markers (kDa).

subunits of approximately 55 ( $\alpha$ ), 33 ( $\beta$ ), and 30 ( $\gamma$ ) kDa, similar to what was previously reported. While the protein has hydrophobic character, it is not an integral membrane protein but is found attached on the inner face of the cytoplasmic membrane (Sprott et al. 1987, Sprott and Beveridge 1993).

Figure 5.11 shows an activity-stained and double stained (coomassie blue- and silver-stained) 3-20% native gradient PAGE with loaded with the purified soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activities, indicating that the soluble and membrane-bound activities are identical.

From these results, the observed phenazine-dependent  $F_{420}H_2$  dehydrogenation activity from  $Msp.\ hungatei$  GP1 is a property of the  $F_{420}$   $H_2$ ase, and not to a distinct Fpo complex as purified from some methylotrophic members of the Methanosarcinaceae.



**Figure 5.9**. Coomassie blue-stained SDS PAGE (12.8% acrylamide) of the purified soluble phenazine-dependent  $F_{420}H_2$  dehydrogenation activity from *Methanospirillum hungatei* GP1. 1, molecular markers (kDa). 2, purified protein.

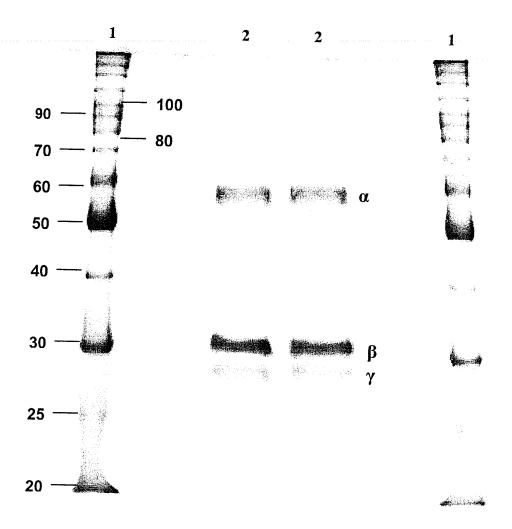


Figure 5.10. Double-stained (Coomassie blue- and silver-stained) SDS PAGE (12.8% acrylamide) of purified soluble phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity from *Methanospirillum hungatei* GP1. 1, molecular markers (kDa). 2, purified protein.

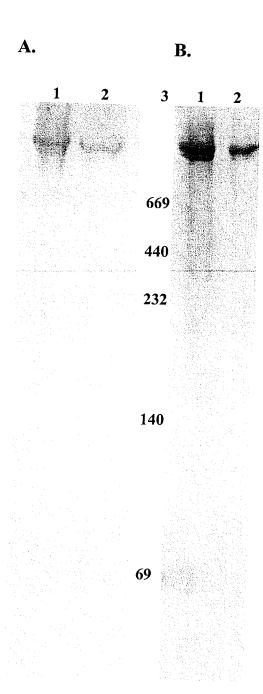


Figure 5.11. (A) Hydrogenase activity-stained and (B) coomassie blue/silver-stained 3-20% native gradient PAGE comparing the purified soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activities. 1, soluble protein. 2, membrane protein. 3, molecular markers (kDa).

## 5.3.3.2. Mass Spectroscopy sequencing of the purified protein

To confirm that the purified  $F_{420}H_2$  dehydrogenation activity was located on the  $F_{420}$   $H_2$ ase, mass spectroscopy (MS) analyses of the purified protein was conducted. Two different methods were used to derive sequences from the purified protein. The first method entailed digestion of the holoenzyme with the enzyme trypsin, followed by separation of the fragments by high performance liquid chromatography (HPLC). When the fragment data was compared against the Sonar MS/MS database, the results indicated high similarities with fragments from  $F_{420}$   $H_2$ ases from several *Methanosarcina* spp. (Table 5.4).

Table 5.4. Mass spectroscopy analysis results for trypsin-digested fragments of the purified soluble phenazine-dependent  $F_{420}H_2$  dehydrogenation activity (holoenzyme) of *Methanospirillum hungatei* GP1. Results compiled using Sonar MS/MS database. Further information regarding the analyses of the holoenzyme is given in Appendix B – Mass Spec data for *Msp. hungatei*.

Expectation	Significant results (Expectation < 0.1)	Methanoarchaeon	Accession number
2.1e-3	F <sub>420</sub> -reducing hydrogenase isoenzyme II, α subunit	Methanosarcina barkeri	CAA74090
2.7e-2	$F_{420}$ -reducing hydrogenase, $\alpha$ subunit	Methanosarcina acetivorans	NP_615927
3.5e-2	$F_{420}$ -reducing hydrogenase isoenzyme I, $\alpha$ subunit	Methanosarcina barkeri	CAA74094
1.8e-1	$F_{420}$ -reducing hydrogenase isoenzyme I, $\beta$ subunit	Methanosarcina barkeri	CAA74097
7.7e-1	F <sub>420</sub> -reducing hydrogenase, B subunit	Methanosarcina mazei Göl	NP_635066

The second technique involved in-gel trypsin digestion for each of the purified subunits, as described in 'Materials and methods'. Of the three subunits, only the  $\alpha$ -subunit provided data of sufficient quality for analysis. Two independent MS analyses were conducted on the  $\alpha$ -subunit. HPLC separation of the digested  $\alpha$ -subunit, followed by MS analysis of the fragments, indicated the presence of a fragment with an amino acid sequence that was identical to a fragment derived from the  $\alpha$ -subunit of the F<sub>420</sub> H<sub>2</sub>ase from *Ms. mazei* Gö1 (Table 5.5).

MALDI TOF (Matrix-Assisted Laser Desorption Ionization Time of Flight) MS analysis of the trypsin-digested subunit also revealed that the predicted sequence of the polypeptide fragment matched with the sequence of the  $\alpha$ -subunit of the F<sub>420</sub> H<sub>2</sub>ase from Ms. mazei Gö1 (Table 5.5). Analysis of the amino acid sequence of the  $\alpha$ -subunit from the F<sub>420</sub> H<sub>2</sub>ase of Msp. hungatei JF1 confirmed the presence of this fragment.

Table 5.5. Mass spectroscopy analysis results for trypsin-digested fragments of the purified soluble phenazine-dependent  $F_{420}H_2$  dehydrogenase activity ( $\alpha$ -subunit) of *Methanospirillum hungatei* GP1. Fragments predicted to be similar to sequences derived from *Methanosarcina mazei* Gö1. Results compiled using Sonar MS/MS database.

Technique	Expectation	Subunit/Sequence	Accession number
HPLC, MS	8.2e-3	α-subunit, <sup>32</sup> GDWLSITPVR <sup>41</sup>	NP_635069
MALDI TOF	3.1e-3	$\alpha$ -subunit, $^{32}$ GDWLSITPVR $^{41}$	NP_635069

BLAST analyses, using amino acid sequences encoding the subunits of the  $F_{420}$   $H_2$  as from various methanoarchaea, confirm that the homologous enzyme from Msp. hungatei is most closely related to the enzymes from the Methanosarcina spp (Appendix C and D).

### 5.3.3.3. Catalytic properties

Although our research focused primarily on the  $F_{420}H_2$  dehydrogenation activity located in the membranes of  $Msp.\ hungatei$  GP1, working with the soluble activity had the benefit of significantly higher yields of purified protein that was indistinguishable from the membrane-bound activity. The soluble phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was purified 45-fold to apparent homogeneity with a yield of 5% and a final specific activity of 13.5  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein (Table 5.1); this value ranges from 10-15  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein.

The  $F_{420}H_2$  dehydrogenation activity does not require reductive reactivation prior to assay, as similar activities were measured under aerobic or anaerobic conditions (data not shown). In contrast, the  $F_{420}$ -reducing activity must be incubated under reducing conditions in the  $F_{420}$   $H_2$ ase reaction buffer before optimal  $F_{420}$ -reducing activity is observed. The reactivation process was modified from Choquet and Sprott (1991), with incubation of the protein in 10  $\mu$ M FAD, 10  $\mu$ M  $F_{420}$  and 20 mM DTT, under an atmosphere of  $H_2$ , prior to assay.

NADPH or NADH could not substitute for  $F_{420}H_2$  as electron donor, and NAD<sup>+</sup>, NADP<sup>+</sup>, and ferredoxin could not substitute for phenazine, while methyl viologen (with metronidazole) and FAD were both suitable electron acceptors for the  $F_{420}H_2$ 

dehydrogenase activity (Table 5.6). The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was inhibited by the presence of diphenyleneiodonium (DPI) chloride. Dimethyl sulfoxide (DMSO), the solvent used to dissolve DPI did not have any effect on the  $F_{420}H_2$  dehydrogenation activity.

DPI is also an inhibitor of flavoproteins, as the purified  $F_{420}$   $H_2$ ase was inhibited by the presence of DPI (at 50  $\mu$ M; the activity was not affected by the addition of 25  $\mu$ M DPI), but the methyl viologen-reducing activity was not inhibited by DPI

Table 5.6. Electron acceptor specificities of the purified  $F_{420}H_2$  dehydrogenase activity of Methanospirillum hungatei GP1.  $[F_{420}H_2] = 10$ -12  $\mu M$ .

E <sup>-</sup> acceptor and/or inhibitor (μ <b>M</b> )	Activity (μmol F <sub>420</sub> H <sub>2</sub> oxidized·min <sup>-1</sup> ·mg <sup>-1</sup> )
-	$0.3 \pm 0.05$
Methylviologen (500) + Metronidazole (1000)	$12.8 \pm 0.7$
Phenazine (500)	$13.5 \pm 1.8$
Phenazine (500) + DMSO (50 μl)	12.4 ± 1.5
Phenazine (500) + DPI (25)	$9.2 \pm 1.2$
Phenazine (500) + DPI (50)	$5.1 \pm 0.4$
FAD (40)	$5.0 \pm 0.4$
NADP <sup>+</sup> (300)	$0.4 \pm 0.05$
NAD <sup>+</sup> (300)	$0.3 \pm 0.05$
Ferredoxin (1 mg/ml)	$0.3 \pm 0.03$

(Table 5.7). Addition of DMSO did not affect either of the F<sub>420</sub>- or methyl viologenreducing  $H_2$  as activities. The presence of phenazine appears to inhibit the  $F_{420}$ reducing activity; this may be due to direct competition for  $H_2(E^{\circ\prime} = -0.420 \text{ V})$  as phenazine ( $E^{\circ\prime} = -0.265$  to -0.165 V) also acts as substrate for the hydrogenase activity.  $F_{420}H_2$  may be oxidized by the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity concurrent with  $F_{420}$ -reduction by the  $H_2$ -dependent  $F_{420}$ - reducing  $H_2$  as activity.

Table 5.7. Effect of DPI and phenazine on the hydrogenase activities of Methanospirillum hungatei GP1. Respective hydrogenase activities were reactivated prior to addition of inhibitor as described under 'Materials and Methods'.  $[F_{420}] = 20-25$  $\mu$ M. [methyl viologen] = 2 mM.

Electron acceptor	Addition (µM)	Activity
$\mathrm{F}_{420}$	_	$14.3 \pm 2.2^{\dagger}$
$\mathbf{F_{420}}$	50 μl DMSO	$14.7 \pm 1.5^{\dagger}$
F <sub>420</sub>	DPI (25)	$14.5\pm1.2^{\dagger}$
$\mathbf{F_{420}}$	DPI (50)	$4.9\pm0.5^{\dagger}$
F <sub>420</sub>	Phenazine (500)	$3.8 \pm 0.5^{\dagger}$
Methyl viologen	-	$50.1 \pm 2.1^{\ddagger}$
Methyl viologen	50 μl DMSO	$57.0 \pm 3.2^{\ddagger}$
Methyl viologen	DPI (50)	$51.0 \pm 2.1^{\ddagger}$
Methyl viologen	phenazine (500)	$54.2 \pm 4.1^{\ddagger,\dagger\dagger}$

<sup>†</sup> μmol F<sub>420</sub> reduced·min<sup>-1</sup>·mg<sup>-1</sup>
‡μmol methyl viologen reduced·min<sup>-1</sup>·mg<sup>-1</sup>
†† full activity after 5-10 minute lag

In the presence of phenazine, methyl viologen (E°' = -0.446 V) reduction lagged for several minutes (~5-10 minutes), before reaching full activity. This may be due to the preference of the hydrogenase activity for phenazine as substrate, or possibly to transfer of electrons from the reduced methyl viologen to phenazine (either chemical or enzymatic). Addition of 100  $\mu$ M phenazine to fully reduced methyl viologen (2 mM) resulted in rapid oxidation of the methyl viologen; the reaction was too quick to quantify.

# 5.3.3.4. Enzyme kinetics and effect of diphenyleneiodonium (DPI) chloride on the $F_{420}H_2$ dehydrogenation activity

The apparent  $K_m$  for  $F_{420}H_2$  was determined to be 6.3  $\mu$ M  $F_{420}H_2$ , with  $V_{max}=31.1~\mu mol~F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> (Lineweaver Burk:  $K_m~F_{420}H_2=6.3~\mu$ M  $F_{420}H_2$ ,  $V_{max}=31.2~\mu mol~F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>) (Figure 5.12). The apparent  $K_m$  for the phenazine-dependent  $F_{420}H_2$  for Ml.~tindarius and Ms.~mazei Gö1 are 5.4 and 7  $\mu$ M  $F_{420}H_2$ , respectively.

The Michaelis Menten and Lineweaver Burk plots used to determine the apparent  $K_m$  for phenazine are shown in Figures 5.13 and 5.14. As determined using the Michaelis Menten kinetics method, the apparent  $K_m$  for phenazine was found to be 288.3  $\mu$ M, while Vmax was 19.5  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>. Similar values were calculated using the Lineweaver Burk method ( $K_m = 243.0 \,\mu$ M phenazine, Vmax = 16.9  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>). The apparent  $K_m$  for phenazine as determined for the dehydrogenation activity of Msp. hungatei GP1 is similar to the values reported for  $F_{420}H_2$  dehydrogenase from Ms. mazei Gö1 and Ml. tindarius (250 and 288.3  $\mu$ M, respectively) (Abken et al. 1998, this thesis). The turnover number for the  $F_{420}H_2$  dehydrogenation activity of Msp. hungatei GP1,  $k_{cat}$ , was determined to be 164.4 s<sup>-1</sup>

 $(k_{cat}/K_m = 5.6 \times 10^5 M^{-1} s^{-1}).$ 

Our studies have also demonstrated inhibition of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of  $Msp.\ hungatei$  GP1 (Figures 5.13 and 5.14). Addition of 25  $\mu$ M DPI increased the  $K_m$  for phenazine from 288.3  $\mu$ M to 370.0  $\mu$ M (Lineweaver Burk:  $K_m$  increased from 243 to 403.5  $\mu$ M phenazine). In contrast, the apparent  $K_m$  increased from 266.7  $\mu$ M to 1168.1  $\mu$ M (3  $\mu$ M DPI) for  $Ml.\ tindarius$  (this thesis). These results indicate that the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of  $Msp.\ hungatei$  GP1 has a similar affinity for phenazine relative to  $Ml.\ tindarius$ , but is much less sensitive to the effects of DPI.

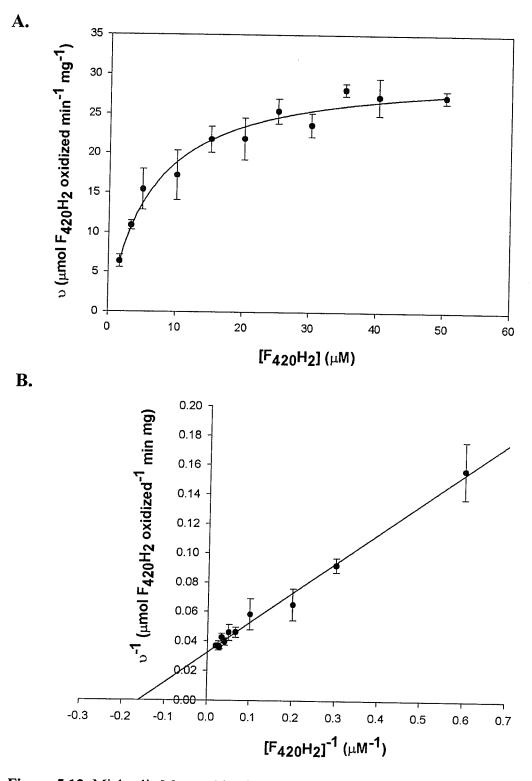


Figure 5.12. Michaelis Menten kinetics (A) and Lineweaver Burk plot for the determination of  $KmF_{420}H_2$  for the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Methanospirillum hungatei* GP1. Each point represents the average of three separate experiments. 500  $\mu$ M phenazine added per tube.

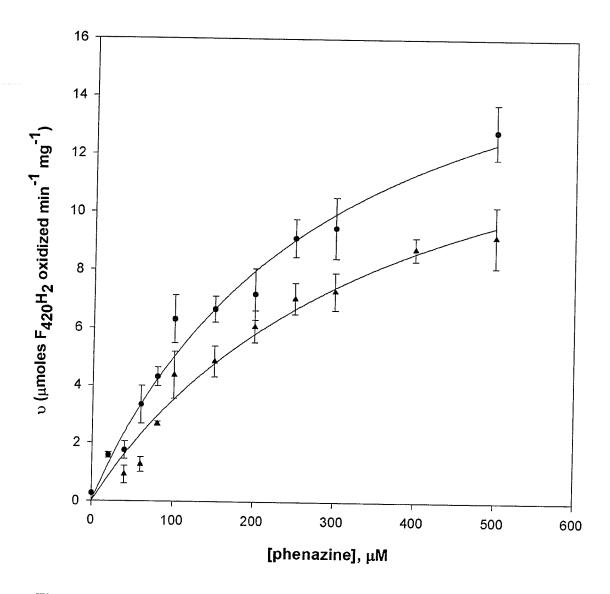


Figure 5.13. Michaelis Menten kinetics plot of the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Methanospirillum hungatei* GP1 in the presence/absence of DPI. All points represent experiments performed in triplicate. Assay conditions as described under 'Materials and Methods'.  $\bullet$ , No DPI.  $\blacktriangle$ , 25  $\mu$ M DPI.

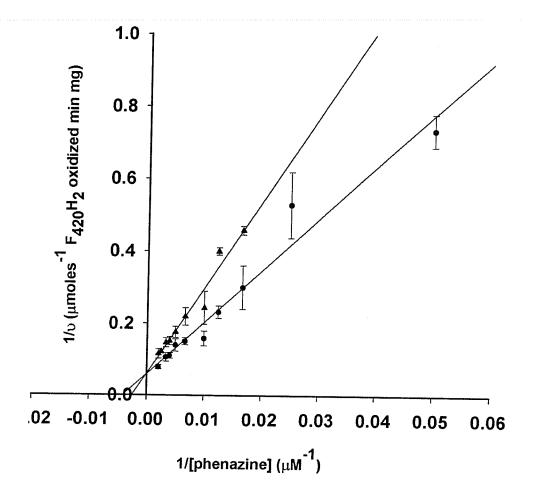


Figure 5.14. Lineweaver Burke kinetics plot of the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Methanospirillum hungatei* GP1 in the presence/absence of DPI. All points represent experiments performed in triplicate. Assay conditions as described under 'Materials and Methods'. •, No DPI.  $\blacktriangle$ , 25  $\mu$ M DPI.

### 5.3.3.5. Analyses of the genome of Methanospirillum hungatei JF1

The annotated genome of *Msp. hungatei* JF1 (NC\_007796) is available for analyses at the National Center for Biotechnology Information (NCBI, <a href="http://www.ncbi.nlm.nih.gov">http://www.ncbi.nlm.nih.gov</a>) and at the Joint Genome Institute (contig 65, JGI, <a href="http://genome.jgi-psf.org">(http://genome.jgi-psf.org</a>)). Examination of the coding sequences located in contig 65 did not reveal the presence of genes coding for the F<sub>420</sub>-nonreducing H<sub>2</sub>ase, while the operon coding for the F<sub>420</sub> H<sub>2</sub>ase (Frh) was present as expected (Table 5.8).

Table 5.8. Frh operon of *Msp. hungatei* JF1. Sequences of subunits from Contig 65 from the Joint Genome Institute Microbial Sequencing Program.

Gene	Sequence location	Subunit
Mhun2329	2584805-2585671	FrhB (β)
Mhun2330	2585674-2586462	FrhG (γ)
Mhun2331	2586503-2587054	FrhD (δ)
Mhun2332	2587068-2588477	FrhA (α)

BLAST analyses of the Archaea database at NCBI, using the amino acid sequence encoding the FpoF subunit (accession number: AAF65743) of the Fpo complex from *Ms. mazei* Gö1, did not reveal the presence of an FpoF homologue in *Msp. hungatei* JF1 (NC\_007796), only sequences coding for the β-subunit (FrhB) of the F<sub>420</sub> H<sub>2</sub>ase were observed (Table 5.9).

Table 5.9. BLAST analyses results using the amino acid sequence coding for the FpoF subunit from Ms. mazei Gö1 against the genome of Msp. hungatei JF1.

Ms. mazei Gö1	Msp. hungatei JF1					
FpoF subunit/ (accession number)	Corresponding subunit/ sequence	Identity	Similarity	E value		
FpoF (AAF65743)	FrhB (β) (899494-900231)	84/248 (33%)	139/248 (56%)	4e-32		
FpoF (AAF65743)	FrhB (β) (2585602-2584817)	95/264 (35%)	147/264 (55%)	9e-41		

BLAST analyses, using the tblastn program, did not indicate the presence of sequences coding for the  $F_{420}$ -nonreducing  $H_2$ ase, consistent with the findings of Sprott *et al.* (1987) and our own biochemical studies. The amino acid sequences encoding the  $F_{420}$ -nonreducing  $H_2$ ase s used as query for the BLAST analyses were derived from the following methanoarchaea: *Mtb. thermoautotrophicus*  $\Delta H$  (Mvh, Table 5.10); *Ms. mazei* Gö1 (Vho, Table 5.11); and *Mc. voltae* (Vhu, Table 5.12). The E values (Expectation) were low, and several of the subunits from the respective  $F_{420}$ -nonreducing  $H_2$ ases corresponded to subunits of the  $F_{420}$   $H_2$ ase of *Msp. hungatei* JF1; this is expected as both enzymes are [NiFe]-hydrogenases (Vignais *et al.* 2001).

Table 5.10. BLAST analyses of the Msp. hungatei JF1 genome using the amino acid sequences coding for the  $F_{420}$ -nonreducing hydrogenase (Mvh) of Mtb. thermoautotrophicus  $\Delta H$ .

Mtb. thermoautotrophicus ΔH	Msp. hungatei JF1				
MvH subunit/ (accession number)	Corresponding subunit/ sequence	Identity	Similarity	E value	
MvhA (Q50783)	FrhA (α) (2588417-2587116)	123/465 (26%)	203/465 (44%)	5e-28	
MvhG (Q50782)	FrhG (γ) (2586462-2585860)	74/248 (29%)	113/248 (45%)	1e-16	
MvhD (Q50781)	MvhD (δ) (2073516-2073926)	75/137 (54%)	97/137 (70%)	8e-39	
MvhD (Q50781)	MvhD (δ) (2075475-2075864)	72/130 (55%)	94/130 (72%)	8e-39	

Table 5.11. BLAST analyses of the  $Msp.\ hungatei$  JF1 genome using the amino acid sequences coding for the  $F_{420}$ -nonreducing hydrogenase (Vho) of  $Ms.\ mazei$  Gö1.

Ms. mazei Gö1	Msp. hungatei JF1				
Vho subunit/ (accession number)	Corresponding subunit/ sequence	Identity	Similarity	E value	
VhoA (AAM32009)	FrhA (α) (2588423-2588052)	41/127 (32%)	65/127 (51%)	6e-12	
VhoA (AAM32009)	FrhA (α) (2587685-2587116)	66/238 (27%)	94/238 (39%)	3e-7	
VhoG (AAM32010)	No match	-	-	-	
VhoC (AAM32008)	ABC transporter (312067-312201)	17/50 (34%)	24/50 (48%)	7.3	

Table 5.12. BLAST analyses of the *Msp. hungatei* JF1 genome using the amino acid sequences coding for the F<sub>420</sub>-nonreducing hydrogenase (Vhu) of *Mc. voltae*.

Mc. voltae	Msp. hungatei JF1					
Vhu subunit/ (accession number)	Corresponding subunit/ sequence	Identity	Similarity	E value		
VhuA	FrhA (α)					
(Q00407)	(2588423-2587155)	115/435 (26%)	194/435 (44%)	6e-27		
VhuG	FrhG (γ)					
(Q00409)	(2586462-2586061)	50/158 (31%)	77/158 (48%)	2e-12		
VhuD	MvhD (δ)					
(Q00408)	(2073531-2073926)	52/132 (39%)	81/132 (61%)	8e-25		
VhuD	MvhD (δ)					
(Q00408)	(2075469-2075804)	52/132 (39%)	81/132 (61%)	6e-25		
VhuU	FrhA (α)					
(Q00410)	(2588417-2587116)	72/130 (55%)	94/130 (72%)	8e-39		

Although the genes encoding the catalytic subunits of the  $F_{420}$ -nonreducing  $H_2$  ase were not isolated, sequences coding for the  $D/\delta$ -subunits from the respective  $F_{420}$ -nonreducing  $H_2$  ases from Mtb. thermoautotrophicus  $\Delta H$  and Mc. voltae are found in the genome of Msp. hungatei JF1. These subunits are distinct from the FrhD subunit, and are not found in the frh operon of Msp. hungatei JF1.

It is interesting to note that on examination of the annotated sequence in Contig 65 (JGI), the genes encoding the D/ $\delta$ -subunits (Mhun 1839 and Mhun1842, contig 65) are identical repeats found immediately adjacent to the genes of the *hdrABC* operon (Mhun1838-Mhun1836), at the 3' end of *hdrA*. These repeats are related to the  $\delta$ -subunit of the F<sub>420</sub>- reducing H<sub>2</sub>ase from *M. kandleri*, but also share approximately 50%

Table 5.13. BLAST search results of the gene encoding the putative  $F_{420}$ -nonreducing hydrogenase  $\delta$ -subunit at the 3' end of hdrA. The gene encoding the hydrogenase  $\delta$ -subunit (Mhun1839, 2073516-2073838) is found at the 3' end of hdrA gene (Mhun1838, 2071501-2073516).

Methanoarchaeon	subunit	hydrogenase	% Identity	% Similarity	E value
M. kandleri	δ	F <sub>420</sub> -reducing	55	70	1e-40
Mc. voltae	VhcD	F <sub>420</sub> - nonreducing	51	72	4e-40
Mc. maripaludis	δ	F <sub>420</sub> - nonreducing	50	70	2e-38
GZFOS26D6 <sup>†</sup>	δ	F <sub>420</sub> -reducing	55	72	9e-38
Mtb. thermoautotrophicus	MvhD	F <sub>420</sub> - nonreducing	54	70	1e-37

<sup>†</sup>uncultured methanogenic archaeon

identity ( $\sim$ 70% similarity) with various  $\delta$ -/D-subunits from the F<sub>420</sub>-nonreducing H<sub>2</sub>ases of various Class I methanoarchaea, including MvhD of *Mtb. thermoautotrophicus*  $\Delta$ H (Table 5.13). MvhD has been suggested to be the electronic link between the F<sub>420</sub>-nonreducing H<sub>2</sub>ase (Mvh) and Hdr in *Mtb. marburgensis* (Figure 1.15B) (Stojanowic *et al.* 2003, Fricke *et al.* 2006). In contrast, FrhD does not copurify with the holoenzyme and is expected to encode a protease involved with maturation (Fricke *et al.* 2006).

#### 5.4. Discussion

## 5.4.1. Purfication of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity in Methanospirillum hungatei GP1

Phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity has been detected in the cell-free extracts of a number of non-methylotrophic methanoarchaea outside of the *Methanosarcinaceae*, including *Msp. hungatei* GP1 (Chapter 4). Given the close phylogenetic relationship between the orders *Methanosarcinales* and *Methanomicrobiales* (Bapteste *et al.* 2005), the observation of a phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activity in *Msp. hungatei* GP1 was not surprising. In the absence of the F<sub>420</sub>-nonreducing H<sub>2</sub>ase, which is typically found in hydrogenotrophic methanoarchaea associated with the H<sub>2</sub>:heterodisulfide oxidoreductase complex, an F<sub>420</sub>H<sub>2</sub>-dependent mechanism for CoM-S-S-CoB reduction can be envisioned in *Msp. hungatei* GP1 (Sprott *et al.* 1987, Thauer *et al.* 1993, this thesis).

Following cell lysis, the vast majority (80-90%) of the  $F_{420}H_2$  dehydrogenation activity is found in the soluble cytoplasmic portion of  $Msp.\ hungatei$  GP1, with 10-20% of the overall activity located on the washed cell membranes. This is consistent with the level of  $F_{420}H_2$  dehydrogenase activity found in the membranes of  $Ml.\ tindarius$  and  $Ms.\ mazei$  Gö1 (Haase  $et\ al.\ 1992$ , Abken and Deppenmeier 1997).

Various chromatographic techniques were used to resolve the  $F_{420}H_2$  dehydrogenation activity of  $Msp.\ hungatei$  GP1; at the onset, our results indicated that the  $F_{420}H_2$  dehydrogenation activity eluted from the various chromatographic matrices with an  $F_{420}$   $H_2$  as activity. These results were observed on a consistent basis whether the proteins were derived from the soluble portion, or extracted from solubilized membranes (Figures 5.1 to 5.4, section 5.3.1).

The profile of the purified  $F_{420}H_2$  dehydrogenation activity of  $Msp.\ hungatei$  GP1, as observed on native gradient PAGE and SDS PAGE gels indicate the isolation of the  $F_{420}$  H<sub>2</sub>ase. The native form of the enzyme has a molecular weight greater than 669 kDa (Figure 5.8), consistent with a molecular weight of  $720 \pm 84$  kDa as reported by Sprott et  $al.\ 1987$ . The protein is made up of three distinct subunits (Figures 5.9 and 5.10) of approximately 55 ( $\alpha$ ), 35 ( $\beta$ ), and 32 ( $\gamma$ ) kDa (Choquet and Sprott 1991). The soluble cytoplasmic- and membrane-bound  $F_{420}H_2$  dehydrogenation activities were both isolated on the  $F_{420}$  H<sub>2</sub>ase (Figure 5.11). No other source of  $F_{420}H_2$  dehydrogenation activity was observed in  $Msp.\ hungatei$  GP1 during our studies.

Mass spectroscopy analyses indicate that fragments of the purified protein share highest similarity with the  $F_{420}$   $H_2$ ase s from the *Methanosarcina* spp. (Tables 5.4 and 5.5). Further analyses of the Archaea database at NCBI indicate that the  $F_{420}$   $H_2$ ase of *Msp. hungatei* JF1 is most closely related to the corresponding enzymes from *Methanosarcina* spp. (Appendix C and D).

In contrast, the  $F_{420}H_2$  dehydrogenase activity purified from Ml. tindarius and Ms. mazei Gö1 each consist of 5 distinct subunits with molecular weights ranging from of 45, 40, 22, 18, and 17 kDa (120 kDa overall, Ml. tindarius) and 40, 37, 22, 20, and 16 kDa (115 KDa overall, Ms. mazei Gö1). The purified activities were not associated with an  $F_{420}$   $H_2$ ase activity (Abken and Deppenmeier 1997).

## 5.4.2. $F_{420}H_2$ dehydrogenase vs. $F_{420}$ -reducing hydrogenase activity

The role of the  $F_{420}$   $H_2$  ase is to transfer electrons from  $H_2$  to  $F_{420}$ :

$$F_{420}+H_2 \rightarrow F_{420}H_2$$

During growth on H<sub>2</sub> and CO<sub>2</sub>, F<sub>420</sub>H<sub>2</sub> is used as a source of electrons for the reduction of C<sub>1</sub>-bound H<sub>4</sub>MPT intermediates to the level of CH<sub>3</sub>-H<sub>4</sub>MPT (Figure 1, Chapter 1) (Thauer *et al.* 1993, Deppenmeier *et al.* 1996, Shima *et al.* 2002). This enzyme is typically isolated from the cytoplasm, although immunogold studies have shown that the enzyme is found attached to the inner face of the cytoplasmic membrane (Baron *et al.* 1989, Lünsdorf *et al.* 1991, Braks *et al.* 1994). A membrane location would be essential if this enzyme is required for energy conservation, as the energy conserving electron transport apparatus is located within the membrane (Schäfer *et al.* 1999).

Similar to the  $F_{420}H_2$  dehydrogenase of Ms. mazei Gö1, the  $F_{420}H_2$  dehydrogenation reaction of Msp. hungatei GP1 could be assayed using phenazine as an electron acceptor:

$$F_{420}H_2$$
 + phenazine  $\rightarrow$   $F_{420}$  + phenazine  $-H_2$ 

Phenazine-dependent  $F_{420}H_2$  dehydrogenation was competitively inhibited by the presence of DPI (25  $\mu$ M). The physiological significance of this activity is unclear, as methanophenazine has not been isolated from methanoarchaea outside of the *Methanosarcinaceae* (Beifuss and Tietze 2005). Also, in contrast to the Fpo complex, the  $F_{420}$   $H_2$ ase does not contain membrane integral components resembling a proton pump, so this protein on its own likely is not a site for energy conservation. The purified  $F_{420}H_2$  dehydrogenase from *Ml. tindarius* and *Ms. mazei* Gö1 were the soluble cytoplasmic portion of a larger multi-subunit protein complex with integral membrane

components (Deppenmeier 2002, 2004). It is possible that the  $F_{420}$   $H_2$  as may dock onto integral membrane components that would not have been purified using our protocol.

#### 5.4.2.1. Differences in requirement for reductive reactivation

The  $F_{420}H_2$  dehydrogenation activity of  $Msp.\ hungatei$  GP1 did not require reactivation prior to assay, even after manipulation under aerobic conditions, when the protein would be in an oxidized form. Addition of protein, whether aerobic or anaerobic, to assay buffer containing  $F_{420}H_2$  and electron acceptor results in immediate enzyme activity. Addition of FAD to column chromatography or storage buffers did not enhance  $F_{420}H_2$  dehydrogenase activity. In contrast, the  $F_{420}$   $H_2$  as activity of  $Msp.\ hungatei$  GP1 requires reactivation of the protein under reducing conditions for optimal activity. These conditions, modified from Choquet and Sprott (1991), require incubation of the protein with reducing agent, FAD, and  $F_{420}$ , under an atmosphere of  $H_2$ . In the absence of reductive reactivation there was a lag in  $F_{420}$ -reducing activity lasting several minutes, and approximately 50% decrease in activity compared to the activated enzyme.

Reductive reactivation is a common procedure for aerobically purified hydrogenase. This entails removal of O<sub>2</sub>, reduction of the active centers of the hydrogenase in the presence of electron donor(s) and traces of active hydrogenase, under an atmosphere of H<sub>2</sub> (Lissolo *et al.* 1984, Teixeira *et al.* 1985). With respect to the F<sub>420</sub> H<sub>2</sub>ase, reduction of the [NiFe] center, FeS clusters, and flavin moiety is essential, particularly after aerobic manipulation (Kojima *et al.* 1983). Active molecules of hydrogenase catalyze the reduction of F<sub>420</sub> to F<sub>420</sub>H<sub>2</sub>, which would then act as electron donors to the oxidized molecules of hydrogenase (Fox *et al.* 1987). Electrons from

 $F_{420}H_2$  enter the protein via the FAD component of the  $\beta$ -subunit, which converts the donated hydride ion into 2 single electrons that are transferred through the FeS clusters in the  $\gamma$ -subunit to the [NiFe] center in the  $\alpha$ -subunit.

While the mechanism for FAD insertion is not known, a possible model could be suggested based on studies with the heterotetrameric NAD<sup>+</sup> reducing hydrogenase of *Ralstonia* (*R*.) *eutropha* (van der Linden *et al*. 2004). This enzyme consists of two modules, an NADH dehydrogenase (HoxF and HoxU) and a hydrogenase module (HoxH and HoxY). Prolonged exposure of the NAD<sup>+</sup>-reducing hydrogenase to NADH leads to a conformational change in the protein, exposing an FMN binding site in HoxY that is normally closed to the environment. NADH is oxidized by HoxF, which contains a fixed molecule of FMN. Electrons are transferred through the FeS clusters from HoxF, through HoxU and HoxY, and finally to HoxH, restoring the protein to a reduced form. At the same time, HoxFU opens up and pivots away from HoxHY, exposing a FMN (bound to HoxY) that slowly dissociates from the protein. This process is reversible, and addition of excess FMN restores the flavin moiety to its site.

Studies with the NAD<sup>+</sup>-reducing hydrogenase of R. eutropha (van der Linden et al. 2004) indicated that FMN could not be restored to the enzyme under oxidized conditions, highlighting the importance of reduced conditions for optimal activation of this protein. Incubation of the oxidized  $F_{420}$  H<sub>2</sub>ase under reducing conditions (H<sub>2</sub> atmosphere, 20 mM DTT, 10  $\mu$ M  $F_{420}$ H<sub>2</sub>) may induce a conformational change to the  $F_{420}$  H<sub>2</sub>ase, perhaps a change in the structure of the  $\beta$ -subunit that exposes the FAD-binding site. The addition of excess FAD may prevent the disassociation of the FAD from its binding site, or more importantly, restore FAD that may have been lost during the course

of purification. This is consistent with the findings of Nelson *et al.* (1984), who observed that FAD dissociated from the  $F_{420}$   $H_2$ ase of *Methanobacterium formicicum* during column chromatography, rendering an inactive enzyme; activity was restored after addition of FAD. Spectrophotometric analysis or FAD quantification could be used to confirm the loss of FAD during purification of the  $F_{420}H_2$  dehydrogenation activity of Msp. hungatei GP1.

## 5.4.3. Mechanism of $F_{420}H_2$ -dependent phenazine reduction via $F_{420}$ -reducing hydrogenase

The  $\alpha$ - and  $\gamma$ - subunits of the methanogenic  $F_{420}$  H<sub>2</sub>ase form a heterodimeric hydrogenase, capable of reducing synthetic electron acceptors such as methyl or benzyl viologen (Thauer *et al.* 1993, Vignais *et al.* 2001). The  $\alpha$ -subunit, which shares homology with the large subunit commonly found in [NiFe] hydrogenases, contains the binding site for the H<sub>2</sub>-activating/splitting bimetallic [NiFe] center (Cammack 1995, Vignais *et al.* 2001).

The  $\gamma$ -subunit contains the FeS clusters that are characteristic of the small hydrogenase subunit, and may also confer electron acceptor specificity to the holoenzyme; the  $\gamma$ -subunit conducts electrons from the H<sub>2</sub>-activating centre of the  $\alpha$ -subunit to the physiological electron acceptor (Vignais *et al.* 2001). Mutation of the FeS clusters from the  $\gamma$ -subunit of the F<sub>420</sub> H<sub>2</sub>ase from *Mc. voltae* results in a 10-fold decrease in F<sub>420</sub>-reduction, but has no effect on methyl viologen-reduction (Bingemann and Klein 2000).

The  $\beta$ -subunit, containing the FAD binding site, confers specificity for  $F_{420}$  to the  $F_{420}$  H<sub>2</sub>ase (Figure 5.15A); absence or denaturation of the  $\beta$ -subunit eliminates the

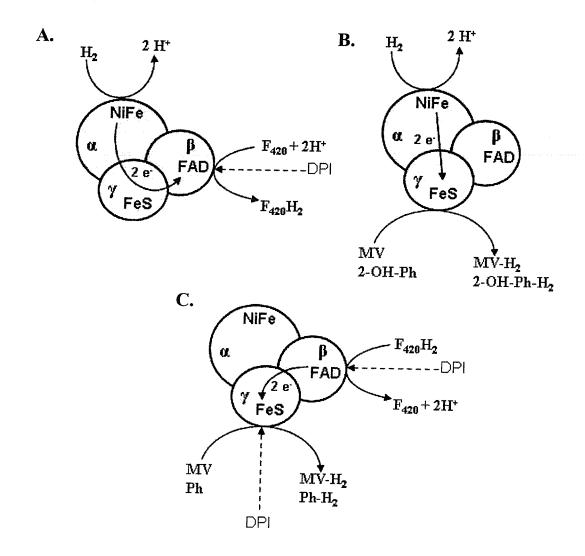


Figure 5.15. Models of electron shuttling with the  $F_{420}$ -reducing hydrogenase, using different substrates. A.  $H_2$ -dependent  $F_{420}$ -reduction. Electrons from the  $H_2$ -splitting [NiFe] center in the α-subunit flow through the FeS clusters in the γ-subunit to the β-subunit. Electrons are converted to a hydride ion via FAD for transfer to  $F_{420}$ . DPI may inhibit  $F_{420}H_2$  reduction by binding to the FAD in the β-subunit. B.  $H_2$ -dependent reduction of methyl viologen and 2-OH-phenazine (in Mtb. thermoautotrophicus  $\Delta H$ , Meuer et al. 1999). Electrons transferred from the  $\alpha$ -subunit through the FeS clusters in the  $\gamma$ -subunit, to the single  $e^-$  acceptors methyl viologen or 2-OH-phenazine. C.  $F_{420}H_2$ -dependent reduction of methyl viologen or phenazine. Hydride ion from  $F_{420}H_2$  converted to single electrons via FAD in  $\beta$ -subunit; electrons shuttled through FeS clusters in  $\gamma$ -subunit to methyl viologen or phenazine. DPI may inhibit phenazine reduction at different sites. MV, methyl viologen. 2-OH-Ph, 2-Oh-phenazine. Ph, phenazine. DPI, Diphenyleneiodonium chloride.

enzymes' ability to react with  $F_{420}$  (Thauer *et al.* 1993). FAD is essential as it allows the hydrogenase to use  $F_{420}$ , which is a hydride donor/acceptor. The $\beta$ -subunits of the  $F_{420}$   $H_2$  as s from various methanoarchaea share homology with the FpoF subunit of the Fpo complex from *Ms. mazei* Gö1 (Bäumer *et al.* 2000). FpoF, which contains FAD and an FeS cluster, is expected to be the point of electron entry from  $F_{420}H_2$  for the  $F_{420}H_2$  dehydrogenases; overexpression of the *fpoF* gene in *E. coli* leads to the production of a protein that oxidizes  $F_{420}H_2$  while reducing methyl viologen (Baümer *et al.* 2000). It is not known if the  $\beta$ -subunit of the  $F_{420}H_2$  ase, which contains FAD but no FeS clusters, can catalyze a similar reaction apart from the holoenzyme.

Based on available knowledge of electron transfer through the  $F_{420}$   $H_2$ ase, a mechanism for  $F_{420}H_2$ -dependent phenazine reduction can be hypothesized. Frh and Mvh of *Mtb. thermoautotrophicus* both catalyze the  $H_2$ -dependent reduction of 2-OH-phenazine in the same manner as methyl viologen; electrons are transferred from the  $\alpha$ -subunit through the FeS-containing  $\gamma$ -subunit to 2-OH-phenazine (Figure 5.15B) (Albracht 1994, Vignais *et al.* 2001, Meuer *et al.* 2002).  $F_{420}H_2$ -dependent reduction would not require the  $\alpha$ -subunit, but would likely require the  $\beta$ - and  $\gamma$ -subunits (Figure 5.15C). Electrons from  $F_{420}H_2$  ( $E^\circ$  = -0.360 V) enter the  $F_{420}$   $F_{420}$  by way of the FAD in the  $\beta$ -subunit, which converts the hydride ion into two single electrons. The electrons are then channeled through the  $\gamma$ -subunit  $\nu ia$  the 3 FeS clusters, shuttling each electron to phenazine ( $E^\circ$  = -0.265 to -0.165 V), which accepts 2 electrons in single transfer steps (Abken *et al.* 1998).

DPI inhibits the  $F_{420}H_2$  dehydrogenase of Ms. mazei Gö1, inhibiting the reduction of 2-OH-phenazine. At low concentrations of DPI (1  $\mu$ M), the  $F_{420}H_2$  dehydrogenase of

Ms. mazei Gö1 was competitively inhibited, increasing the apparent  $K_m$  for 2-OH-phenazine from 35  $\mu$ M to 100  $\mu$ M (Brodersen *et al.* 1999). At 10-fold higher concentrations of DPI,  $F_{420}H_2$ -dependent methyl viologen reduction was inhibited, indicating that there were two sites of DPI inhibition; at higher concentrations it was believed that the FAD component was being inhibited (Brodersen *et al.* 1999).

Addition of 25  $\mu$ M DPI inhibited phenazine-dependent  $F_{420}H_2$  dehydrogenation activity, but not the  $F_{420}$   $H_2$  as activity (Table 5.6 and 5.7); the  $K_m$  for phenazine increased from 288.3  $\mu$ M to 370.0  $\mu$ M,  $V_{max} = 19.5$   $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> (Michaelis Menten kinetics). The inhibition observed in the presence of 25  $\mu$ M DPI appears to be competitive, due to competition with phenazine for the binding site at the  $\gamma$ -subunit. However, addition 50  $\mu$ M DPI inhibited both the phenazine-dependent  $F_{420}H_2$  dehydrogenation and  $F_{420}$ -reducing activities (Table 5.6 and 5.7). This may indicate that the observed inhibition at this concentration of DPI is due to inhibition of the FAD moiety, preventing the interaction of the  $\beta$ -subunit with  $F_{420}/F_{420}H_2$ .

Phenazine was used in our studies, but it is not clear whether methanophenazine is found in  $Msp.\ hungatei$  GP1. It is possible that the  $F_{420}H_2$ -dependent reduction of phenazine may be an artifact of the  $F_{420}$   $H_2$ ase, similar to the  $H_2$ -dependent reduction of methyl viologen by this enzyme. The purified  $F_{420}$   $H_2$ ase of  $Mtb.\ thermoautotrophicus$   $\Delta H$  also catalyzes the reduction of phenazine in place of  $F_{420}$ ; methanophenazine is not found in this methanoarchaeon (Abken  $et\ al.\ 1998$ , Meuer  $et\ al.\ 1999$ , Beifuss and Tietze 2005).

## 5.4.4. Analyses of the genome of Methanospirillum hungatei JF1

The genes encoding for the Fpo complex were not observed in the genome of  $Msp.\ hungatei\ JF1\ (NC_007796)$ . The amino acid sequence of the FpoF subunit (AAF6574) from  $Ms.\ mazei\ G\"o1$  was most closely related to the FrhB ( $\beta$ ) subunit of the F<sub>420</sub> H<sub>2</sub>ase of  $Msp.\ hungatei\ JF1$  (found in the Frh operon), and to a homologue of the  $\beta$ -subunit outside of the operon (Table 5.8). The FpoF and Frh $\beta$  subunits contain the FAD component for the respective enzymes, while FpoF also contains an additional FeS cluster (Bäumer  $et\ al.\ 2000$ ).

An  $F_{420}$ -nonreducing  $H_2$  ase was not detected in Msp. hungatei GP1 by Sprott et al. (1987), nor did we did detect this protein during our own biochemical studies. BLAST analyses of the genome of Msp. hungatei JF1 using the amino acid sequences encoding the subunits of the  $F_{420}$ -nonreducing  $H_2$  ases from various methanoarchaea (Tables 5.10-5.12) did not result in the location of putative genes encoding for an  $F_{420}$ -nonreducing  $H_2$  ase in Msp. hungatei JF1; only the genes sequences of subunits encoding the  $F_{420}$   $H_2$  ase and Ech  $H_2$  ase were found during these searches. However, homologues for the sequences encoding the  $D/\delta$ -subunits from the  $F_{420}$ -nonreducing  $H_2$  ases of Mtb. thermoautotrophicus  $\Delta H$  and Mc. voltae were found in the genome of Msp. hungatei JF1.

An examination of Contig 65 (http://genome.ornl.gov.microbial/mhun/) at JGI indicates the presence of genes coding for the  $\delta$ -subunits of the F<sub>420</sub> H<sub>2</sub>ase (Mhun 1839, 2073516-2073938), at the 3' end of the gene encoding the HdrA subunit (Mhun 1838, 2071501-2073516); this is in addition to the  $\delta$ -subunit encoded within the Frh operon of *Msp. hungatei* JF1. The putative  $\delta$ -subunit at the 3' end of *hdrA* also shares up to 55% identity (72% similarity) with the  $\delta$ -/D-subunit of the F<sub>420</sub>-nonreducing H<sub>2</sub>ase of various

methanoarchaea, including Mc. voltae (E = 4e-40), Mc. maripaludis (E = 2e-38), and Mtb. thermoautotrophicus  $\Delta H$  (E = 1e-37). BLAST analyses at NCBI identify putative subunit as being similar to MvhD and VhuD subunits of the F<sub>420</sub>-nonreducing H<sub>2</sub>ase of Mtb. thermoautotrophicus  $\Delta H$  (Table 5.10) and Mc. voltae (Table 5.12), respectively, indicating a possible electronic link between the F<sub>420</sub> H<sub>2</sub>ase and Hdr of Msp. hungatei GP1.

A homologue of the mvhD gene, which encodes an FeS-containing subunit of the Mvh H<sub>2</sub>ase, is fused to the 3' end of the hdrA genes of Mtb. thermoautotrophicus  $\Delta H$ , Mtb. marburgensis, and Msph. stadtmanae (Stojanowic et al. 2002, Fricke et al. 2006), suggesting a functional link between the Mvh H<sub>2</sub>ase and the Hdr in these methanoarchaea (Figure 1.15B, Chapter 1). It is suggested that electrons flow from the F<sub>420</sub>-nonreducing H<sub>2</sub>ase to the Hdr through contact formed between the MvhD and HdrA subunits of the respective enzymes, this has not been confirmed experimentally (Fricke et al. 2006). In contrast, frhD encodes for a peptidase involved in maturation of the F<sub>420</sub> H<sub>2</sub>ase; FrhD is not involved in electron transport, nor is it a subunit of the holoenzyme.

The  $F_{420}$ -nonreducing  $H_2$  ase plays a key role in the  $H_2$ -dependent reduction of CoM-S-S-CoB (Thauer *et al.* 1993); absence of this protein would require an alternate mechanism for shuttling of electrons to Hdr. A similar arrangement with the  $F_{420}$   $H_2$  ase and the Hdr of Msp. hungatei GP1 may be possible. Braks *et al.* (1994) observed the presence of 10-20 kDa proteins linking the  $F_{420}$   $H_2$  ase of Mtb. marburgensis to the inner face of the cell membrane via microscopic analyses. It is not known if these linkers are the  $D/\delta$ -subunits fused to HdrA, or if these linkers are associated with electron transport

from the  $F_{420}$   $H_2$  ase to proteins in the membrane that may be associated with energy conservation.

#### 5.4.6. Physiological interpretations

It has been proposed that the  $F_{420}$ -nonreducing  $H_2$  as of methanoarchaea outside of the *Methanosarcinaceae* docks directly with Hdr (Figure 1.15B, Chapter 1), with electron transfer directly from the  $\delta$ -/D-subunit of the  $F_{420}$ -nonreducing  $H_2$  as to HdrA (Stojanowic *et al.* 2003, Fricke *et al.* 2006). This is consistent with the findings of Setzke *et al.* (1994), who found that the  $F_{420}$ -nonreducing  $H_2$  as (Mvh) and Hdr of *Mtb.* thermoautotrophicus  $\Delta H$  copurified together as a tightly bound complex.

In the absence of an  $F_{420}$ -nonreducing  $H_2$ ase, a similar role for the  $F_{420}$   $H_2$ ase of  $Msp.\ hungatei$  GP1 may be envisioned, as depicted in Figure 5.16. The  $F_{420}H_2$  dehydrogenation activity, associated with the  $F_{420}$   $H_2$ ase of  $Msp.\ hungatei$  GP1, may be an essential part of an  $F_{420}H_2$ -dependent CoM-S-S-CoB oxidoreductase system, of which the Hdr comprises the second major component of this proposed electron transport system. The  $F_{420}$   $H_2$ ase of  $Msp.\ hungatei$  GP1 has hydrophobic character, with up to 10-20% of the activity associated with the cell membrane (Choquet and Sprott 1991, Sprott and Beveridge 1993, this study). The  $F_{420}$   $H_2$ ases behave on hydrophobic column matrices as very hydrophobic proteins, suggesting a functional association with the cytoplasmic membrane (Sprott and Beveridge 1993). Lipids are also associated with the purified protein, and in fact stimulate the  $F_{420}$   $H_2$ ase activity of  $Msp.\ hungatei$  (Choquet and Sprott1991); it has been suggested that the lipids may be essential for anchoring the  $F_{420}$   $H_2$ ase to the inner face of the cell membrane.

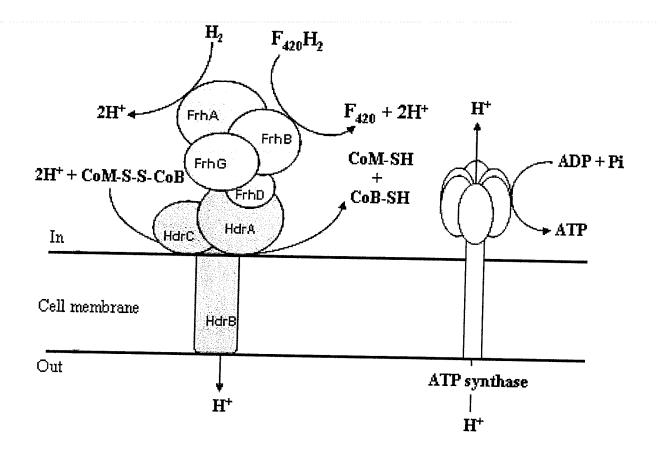


Figure 5.16. Hypothetical model of membrane-bound electron transport in *Methanospirillum hungatei* GP1 during growth on  $H_2$  and  $CO_2$ .  $F_{420}$ -reducing hydrogenase (Frh) is docked onto the heterodisulfide reductase (Hdr). Electrons from  $F_{420}H_2$  transferred from Frh to Hdr via contact between the  $\delta/D$ -subunit of Frh, and HdrA.

Electrons derived from  $F_{420}H_2$  could be shuttled through the  $F_{420}$   $H_2$ ase and transferred to CoM-S-S-CoB via Hdr, to produce the free cofactors CoM-SH and CoB-SH. An energy conserving transmembrane proton gradient would be generated with CoM-S-S-CoB reduction, driving the formation of ATP via ATP synthase.  $H_2$  may also be used as a source of electrons for the reduction of CoM-S-S-CoB (Figure 5.16).

 $H_2$ - and  $F_{420}H_2$ -dependent reduction of CoM-S-S-CoB *via* membrane vesicles of *Ms. mazei* Gö1 leads to the translocation of  $H^+$  across the cytoplasmic membrane and generation of  $\Delta \mu H^+$ , which is used to drive the synthesis of ATP, indicating a requirement for integral membrane components (Peinemann *et al.*1990, Bäumer *et al.* 2000). The Hdr (hdrDE) of methanoarchaea from the *Methanosarcinales* is found integral to the membrane with several transmembrane helices anchoring the protein to the membrane (Deppenmeier *et al.* 1991, Ide *et al.* 1999, Deppenmeier *et al.* 2002). The electron carriers cytochromes and methanophenazine (putative), exclusive to the *Methanosarcinales*, are also found in the cell membranes (Fricke *et al.* 2006).

H<sub>2</sub>-dependent CoM-S-S-CoB reduction also generates ΔμH<sup>+</sup> in non-cytochrome containing methanoarchaea; cell suspensions of *Methanosphaera* (*Msph.*) *stadtmanae* generate ΔμH<sup>+</sup> and ATP sythesis *via* H<sub>2</sub>-dependent reduction of methanol (Sparling *et al.* 1993), while cell suspensions of *Mtb. thermoautotrophicus* couple H<sub>2</sub>-dependent reduction of CoM-S-S-CoB with H<sup>+</sup> translocation (de Poorter *et al.* 2003). However, the Hdr (*hdrABC*) of *Mtb. thermoautotrophicus* has primarily been isolated from the soluble cytoplasmic fraction, although there is evidence that suggests that HdrB may be integral to the membrane (Hedderich and Thauer 1988, Setzke *et al.* 1994, Fricke *et al.* 2006).

While HdrB lacks transmembrane helical segments, Setzke *et al.* (1994) reported that 40% of the H<sub>2</sub>:heterodisulfide oxidoreductase complex was found in the membranes of *Mtb. marburgensis*. Thus there may be alternate means for attachment to the cell membrane. The archaeon *Acidianus ambivalens* possesses a succinate:quinone oxidoreductase (*aka* succinate dehydrogenase), which catalyzes the oxidation of succinate to fumarate, with e<sup>-</sup> transfer to quinones; HdrB of *Msph. stadtmanae* shares sequence similarity with the SdhC subunit (31% identity) (Lemos *et al.* 2001, Fricke *et al.* 2006). This enzyme is an integral membrane protein, but SdhC lacks the usual membrane-anchoring subunits (i.e. transmembrane helical segments). Instead, amphipathic α-helices are present in SdhC that may bind the succinate dehydrogenase to the membranes. A similar phenomenon is observed with mammalian proteins such as the RGS4, a protein that accelerates GTP hydrolysis on α protein (Bernstein *et al.* 2000).

Brodersen *et al.* (1999b) observed that formate-grown Mc. *voltae* cultures possess a membrane-bound  $F_{420}H_2$  dehydrogenation-activity, similar to the  $F_{420}H_2$  dehydrogenase activity isolated from Ml. *tindarius* and Ms. *mazei* Gö1, that was associated with CoM-S-S-CoB reduction. Further examination of the membrane-associated activity lead to the conclusion that the  $F_{420}H_2$  dehydrogenation activity was associated with the membrane-bound  $F_{420}$   $H_2$ ase, identical to the activity previously purified from the soluble cytoplasmic fraction of Mc. *voltae* by Muth *et al.* (1987). While  $F_{420}H_2$  oxidation was measured using the combination of methyl viologen and metronidazole, our results have indicated that phenazine can be used as an electron acceptor for the  $F_{420}H_2$  dehydrogenation activity of Mc. *voltae* (Chapter 4). It remains to be seen if the mechanism of  $F_{420}H_2$ -dependent CoM-S-S-CoB reduction in Mc. *voltae* is as proposed

for *Msp. hungatei* GP1. More research will have to be conducted to determine if the model shown in Figure 5.16 can be confirmed experimentally.

# 6. Purification of the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity from *Methanosarcina barkeri* strain Fusaro

#### 6.1. Introduction

The F<sub>420</sub>H<sub>2</sub> dehydrogenase (also referred to as the Fpo complex (<u>F</u><sub>420</sub>H<sub>2</sub>:<u>p</u>henazine <u>o</u>xidoreductase) has only been isolated from *Methanolobus tindarius* and *Methanosarcina mazei* Gö1, methylotrophic methanoarchaea from the *Methanaosarcinaceae* family, and also from the hyperthermophilic sulfate-reducing archaeon *Archaeoglobus fulgidus* (Haase *et al.* 1992, Abken and Deppenmeier 1997, Brüggermann *et al.* 2000). Although this activity was initially assayed using methyl viologen, the physiological electron acceptor for the methanogenic F<sub>420</sub>H<sub>2</sub> dehydrogenase activity is believed to be methanophenazine, while the acceptor for *A. fulgidus* is a menaquinone (Abken *et al.* 1998, Brüggermann *et al.* 2000).

Most of the present knowledge regarding the methanogenic F<sub>420</sub>H<sub>2</sub> dehydrogenase has been derived from *Ms. mazei* Gö1. Using inverted cell membrane vesicles of *Ms. mazei* Gö1, Deppenmeier *et al.* (1990a, b) observed that F<sub>420</sub>H<sub>2</sub> oxidation and CoM-S-S-CoB reduction was coupled to proton translocation into the lumen of the vesicles. Methanophenazine, which is isolated in small amounts from the cell membranes of *Ms. mazei* Gö1, is extremely hydrophobic, such that 2-OH-phenazine was synthesized as a water soluble alternative (Abken *et at.* 1998, Beifuss and Tietze 2005). Purified F<sub>420</sub>H<sub>2</sub> dehydrogenase from *Ms. mazei* Gö1 catalyzed the reduction of 2-OH-phenazine; the reduced 2-OH-phenazine was then oxidized by heterodisulfide reductase (Hdr) for the reduction of CoM-S-S-CoB (Bäumer *et al.* 1998). Both reactions are coupled to the

generation of a transmembrane proton gradient ( $\Delta \mu H^{+}$ ) and energy conservation (Bäumer et al. 2000, Deppenmeier 2002, 2004).

Comparisons with the gene sequences of the *nuo* operon from *E. coli* suggested that the F<sub>420</sub>H<sub>2</sub> dehydrogenase of *Ms. mazei* Gö1 was an integral membrane-bound multisubunit protein, with a structure similar to NADH:ubiquinone oxidoreductase (NDH-1, complex I) of *E. coli* (Bäumer *et al.* 2000). Analysis of the *fpo* operon revealed that the organization of the gene cluster was ordered as in the *nuo* operon of *E. coli*, with the exception of the subunits required for oxidation of the respective electron carrier (*nuoEFG* for NDH-1 (NADH), *fpoF* for Fpo (F<sub>420</sub>H<sub>2</sub>)). The purified enzymes from the respective methanoarchaea were in fact only a portion of a much larger and more intricate complex, comprising the electron input module for the complex (Deppenmeier *et al.* 1999, Bäumer *et al.* 2000, Deppenmeier 2002).

Like *Ms. mazei* Gö1, *Methanosarcina* (*Ms.*) *barkeri* strain Fusaro is a nutritionally versatile methanoarchaeon from the family *Methanosarcinaceae* that can metabolize H<sub>2</sub>/CO<sub>2</sub>, acetate, and CH<sub>3</sub>OH or methylamines as substrate for methanogenesis. Keltjens and Vogels (1993) noted that F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of *Ms. barkeri* Fusaro appeared to be co-purified along with F<sub>420</sub>-reducing hydrogenase activity (F<sub>420</sub> H<sub>2</sub>ase). The F<sub>420</sub> H<sub>2</sub>ase has been purified (Fiebig and Friedrich 1989, Michel *et al.* 1995); it is a peripherally-bound membrane protein located on the cytoplasmic side of the cell membrane (Lünsdorf *et al.* 1991). Two phylogenetically closely related isoenzymes, Frh and Fre, are encoded in the genome (*frh* and *fre*) of *Ms. barkeri* Fusaro; both isoenzymes are synthesized when *Ms. barkeri* is grown on CH<sub>3</sub>OH or H<sub>2</sub>:CO<sub>2</sub>, but not during growth on acetate (Vaupel and Thauer 1998).

Initial research in our laboratory also detected a MV-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activity in this methanoarchaeon; sucrose density gradient profiles demonstrated distinctly different elution patterns of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activities for *Ml. tindarius* and *Ms. barkeri* Fusaro (Wong, Ph.D thesis 1999). The F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Ms. barkeri* Fusaro appeared to consist of a larger protein mass relative to the respective activity of *Ml. tindarius*, and co-migrated with an F<sub>420</sub> H<sub>2</sub>ase activity. The purified F<sub>420</sub>H<sub>2</sub> dehydrogenase activities from *Ml. tindarius* and *Ms. mazei* Gö1 do not possess F<sub>420</sub> H<sub>2</sub>ase activity (Abken and Deppenmeier 1997). Indeed, *Ml. tindarius* does not possess an F<sub>420</sub> H<sub>2</sub>ase (Deppenmeier *et al.*1989). BLAST analyses of the unpublished genome of *Ms. barkeri* Fusaro (circa. 2003), using the sequences coding for subunits of the Fpo complex from *Ms. mazei* Gö1, indicated the presence of open reading frames (ORFs) coding for a putative Fpo complex. It was not clear whether or not these genes were being expressed.

In this study we report the purification and characterization of a phenazine-dependent  $F_{420}H_2$  dehydrogenation activity from  $Ms.\ barkeri$  Fusaro, grown on  $CH_3OH$ . N-terminus sequencing of the isolated subunits confirmed the identity of the protein as the  $F_{420}$   $H_2$ ase of  $Ms.\ barkeri$  Fusaro. A possible role for the  $F_{420}$   $H_2$ ase during energy conservation is discussed.

#### 6.2. Materials and methods

For a comprehensive look at the materials and methods used for this chapter, the reader is referred to the following sections in Chapter 3:

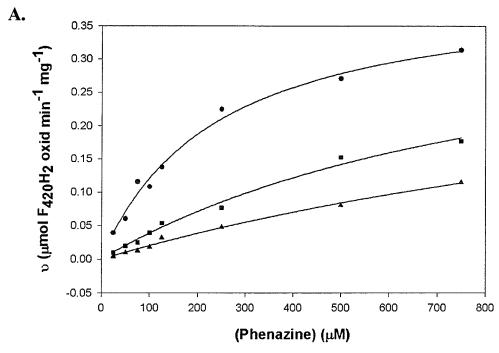
- 3.2. Growth media
- 3.3. Cell lysis

- 3.5. Purification of the  $F_{420}H_2$  dehydrogenation activity from methanol-grown Methanosarcina barkeri Fusaro and Methanospirillum hungatei GP1
- 3.6. Enzyme assays
  - 3.6.1. F<sub>420</sub>H<sub>2</sub> dehydrogenation assays
  - 3.6.2. Hydrogenase assays
    - $3.6.2.1.\ F_{420}$ -reducing-hydrogenase activity
    - 3.6.2.2. Methyl viologen-reducing hydrogenase activity
    - 3.6.2.3. H<sub>2</sub>-dependent phenazine reduction
- 3.7. Gel electrophoresis
  - 3.7.1. SDS-PAGE gel electrophoresis
    - 3.7.1.1. Amino-terminal sequencing
  - 3.7.2. Native gradient PAGE gel electrophoresis

#### 6.3. Results

#### 6.3.1 Preliminary results

The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity assayed using washed membranes of  $CH_3OH$ -grown Ms. barkeri Fusaro was similar to that observed using Ml. tindarius. When assayed in the presence of 3 and 9  $\mu$ M DPI, the  $K_m$  for phenazine for Ml. tindarius increases from 238.9 to 965.6 and 1823.4  $\mu$ M phenazine (Vmax = 0.41  $\mu$ mol  $F_{420}H_2$  oxid·min<sup>-1</sup>·mg<sup>-1</sup>), respectively (Figure 6.1). In contrast, the same concentration of DPI has negligible effect on the  $F_{420}H_2$  dehydrogenation activity of Ms. barkeri Fusaro. Even when 50 and 100  $\mu$ M DPI are added, which is greater than 10-fold the amount used for Ml. tindarius, the  $K_m$  only increases from 106.1 to 204.0 and 342.3  $\mu$ M phenazine (Vmax = 0.32  $\mu$ mol  $F_{420}H_2$  oxid·min<sup>-1</sup>·mg<sup>-1</sup>), respectively.



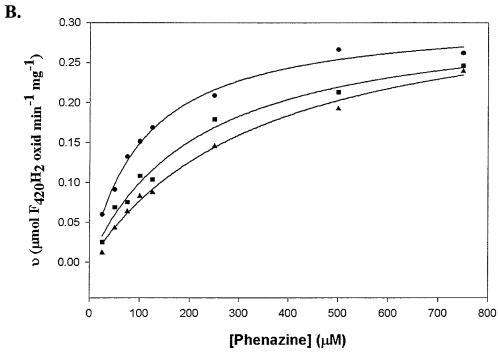


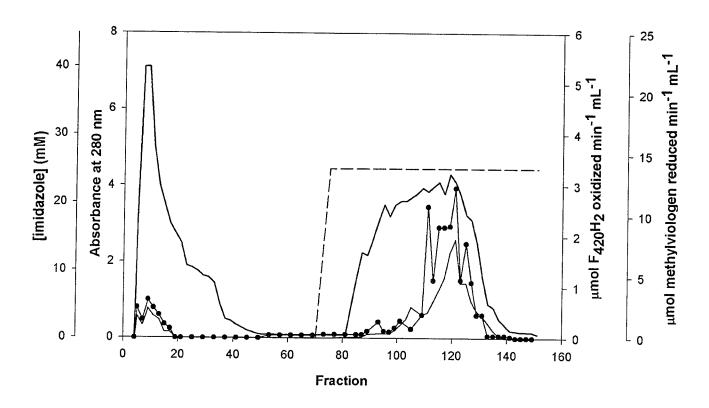
Figure 6.1. Michaelis menten kinetics of the membrane-bound phenazine-dependent  $F_{420}H_2$  dehydrogenase activity of *Ml. tindarius* and *Ms. barkeri* Fusaro. A. *Ml. tindarius*. •, no DPI. •, + 3  $\mu$ M DPI. •, + 9  $\mu$ M DPI. B. *Ms. barkeri* Fusaro, grown on CH<sub>3</sub>OH. •, no DPI. •, + 50  $\mu$ M DPI. •, + 100  $\mu$ M DPI.

The  $F_{420}$   $H_2$  as of *Methanospirillum* (*Msp.*) *hungatei* GPI also catalyzes a reaction similar to the  $F_{420}H_2$  dehydrogenase of *Ms. mazei* Göl (Chapter 5). Thus, we could not discount the possibility that the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of the Fpo complex was being masked by a similar activity catalyzed by the membrane-bound  $F_{420}$   $H_2$  as of *Ms. barkeri* Fusaro that is not found in *Ml. tindarius*.

Studies of the  $F_{420}H_2$  dehydrogenation activity of *Ms. barkeri* Fusaro began with an examination of the membrane-bound proteins. The proteins were extracted from washed membranes (3x) solubilized with CHAPS (1.5 mg CHAPS/mg protein), followed by chromatography of the solubilized protein using various chromatographic matrices; the  $F_{420}H_2$  dehydrogenation activity was assayed using phenazine at each stage of purification.

 $\mathrm{Ni}^{2^+}$  affinity chromatography was used as the initial enrichment step to remove the F<sub>420</sub> H<sub>2</sub>ase, since it is expected to have a high affinity for  $\mathrm{Ni}^{2^+}$  (Choquet and Sprott 1991, this thesis). Unexpectedly, the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity also bound to the column and was eluted along with the F<sub>420</sub> H<sub>2</sub>ase activity using a single step addition of 25 mM imidazole (Figure 6.2). The pooled fractions containing the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity also contained significant hydrogenase activity. Varying the concentration of imidazole or changing from a single step addition of imidazole to a continuous gradient did not resolve the F<sub>420</sub>H<sub>2</sub> dehydrogenation and F<sub>420</sub> H<sub>2</sub>ase activities.

The pooled fractions were desalted, concentrated, and then applied to a 45-50% (10 ml·step<sup>-1</sup>) sucrose density gradient, followed by ultracentrifugation for 48 hours at



**Figure 6.2**. Protein and activity profile of fractions collected from a  $Ni^{2+}$  affinity column, from protein extracted from washed and solubilized cell membranes of *Methanosarcina barkeri* Fusaro. Elution of protein from the column as described under 'Materials and methods'. (—),  $A_{280}$ . (----), imidazole. •, Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. •, Methyl viologen-reducing hydrogenase.

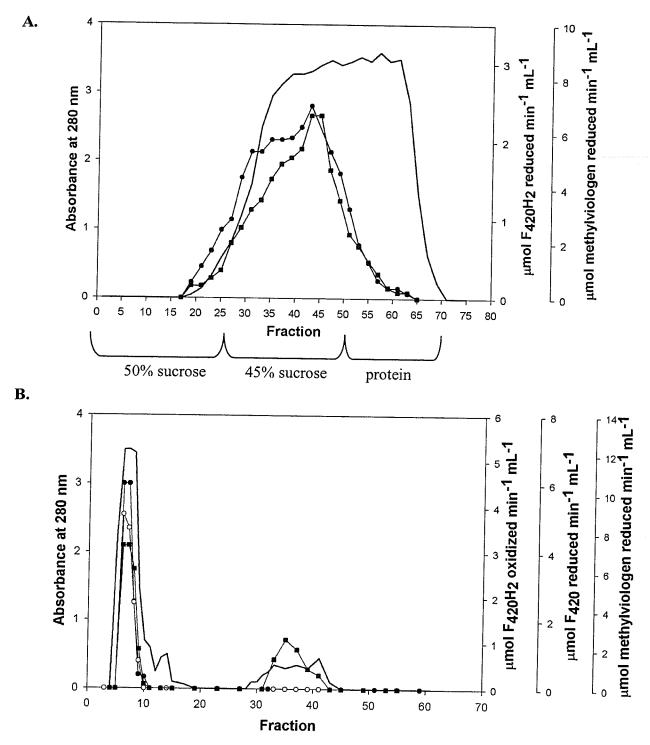


Figure 6.3. Protein and activity profile of fractions collected from: (A) 50-45% sucrose step gradient loaded with protein eluted from a Ni<sup>2+</sup>-affinity column and (B) Sepharose 6B column loaded with protein collected from the 50-45% sucrose step gradient. Protein solubilized and extracted from washed cell membranes. (—), A<sub>280</sub>.

•, Phenazine-dependent  $F_{420}H_2$  dehydrogenation activity.  $\circ$ ,  $F_{420}$ -reducing hydrogenase activity.  $\blacksquare$ , Methyl viologen-reducing hydrogenase activity.

23,000 rpm. The sucrose fractions containing the  $F_{420}H_2$  dehydrogenation activities (Figure 6.3A) were pooled, concentrated, and loaded onto a Sepharose 6B gel filtration column, where the  $F_{420}H_2$  dehydrogenation activity eluted with an  $F_{420}$   $H_2$  as activity (Figure 6.3B). Sepharose 6B resolves proteins with molecular weights ranging from 10 to  $4 \times 10^3$  kDa; the  $F_{420}H_2$  dehydrogenation activity would be expected to have a molecular weight of approximately 115-120 kDa, whereas the  $F_{420}$   $H_2$  as would have a molecular weight of approximately 845 kDa (Fiebig and Friedrich 1989, Abken and Deppenmeier 1997). While the  $F_{420}H_2$  dehydrogenation activity had not been purified to apparent homogeneity, the  $F_{420}H_2$  dehydrogenation and  $F_{420}$   $H_2$  as activities appeared to be closely associated through all the various chromatographic steps. The yield of  $F_{420}H_2$  dehydrogenation activity remaining after the Sepharose 6B gel filtration step was too low to make further analyses feasible.

A possible conclusion was that the steps selected for the purification of the  $F_{420}H_2$  dehydrogenase activity failed to separate it from the  $F_{420}$   $H_2$  as activity. Another possibility is that the observed  $F_{420}H_2$  dehydrogenation activity in Ms. barkeri Fusaro might solely be the property of the  $F_{420}$   $H_2$  ase. Therefore, the same protocol used to purify the  $F_{420}H_2$  dehydrogenation activity from Msp. hungatei GP1 was used to try and separate the respective activity from Ms. barkeri Fusaro. The  $F_{420}$   $H_2$  as also catalyzes the oxidation of  $F_{420}H_2$ ; by selecting for the purification of this protein, we expected to separate the expected Fpo from this "artifact" activity present in the protein extract.

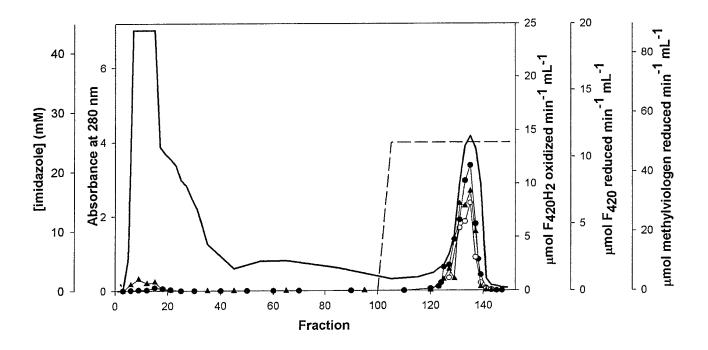
## 6.3.2. Purification of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity

Following cell lysis, and removal of the unlysed cells and debris, the cell-free extract was centrifuged at 180000xg for 6 hours. This resulted in the sedimentation of both the soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activities in a tight dark brown pellet, referred to as the high speed pellet (HSP).

 ${
m Ni}^{2+}$  affinity chromatography was used as the initial step in the enrichment of the  ${
m F}_{420}{
m H}_2$  dehydrogenation activity from the homogenized HSP. The cell membranes, which contain 15-20% of the overall  ${
m F}_{420}{
m H}_2$  dehydrogenation activity, eluted from the column without binding. The soluble  ${
m F}_{420}{
m H}_2$  dehydrogenation activity bound to the  ${
m Ni}^{2+}$ -charged matrix and was eluted using a single step addition of 25 mM imidazole (Figure 6.4). As expected, the  ${
m F}_{420}{
m H}_2$  dehydrogenation and  ${
m F}_{420}{
m H}_2$  as activities elute together from the  ${
m Ni}^{2+}$ -charged column.

Fractions containing the  $F_{420}H_2$  dehydrogenation activity were pooled, desalted and concentrated using an Amicon ultrafiltrator, loaded onto a 3-20% native gradient PAGE, and electrophoresed overnight at 4°C. Intense yellow-brown bands were observed in their respective lanes near the top of the gel; this is due to the presence of FAD and FeS clusters in the protein. Activity staining (as described under 'Materials and methods) indicated that these bands possess hydrogenase activity, suggesting the presence of the  $F_{420}$   $H_2$ ase.

The high molecular weight bands were excised from the gel and cut into 1-2 mm<sup>3</sup> pieces, followed by elution of the protein from the gel pieces using the Bio-Rad Electro-Elutor ( $\sim$ 8 hours). The eluted protein was then dissolved with 10 mM HEPES, pH 7.0, with 0.05% Triton-X-100 and 5% glycerol. Solutions of the purified  $F_{420}H_2$ 



**Figure 6.4**. Protein and activity profile of fractions collected from a  $\mathrm{Ni}^{2+}$  affinity column. Column loaded with homogenized HSP, which contains soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activity. Membrane-bound  $F_{420}H_2$  dehydrogenation activity did not bind to the  $\mathrm{Ni}^{2+}$  column, while the soluble  $F_{420}H_2$  dehydrogenation activity bound and was eluted from the column by the addition of 25 mM imidazole. (—),  $A_{280}$ . (----), imidazole.  $\bullet$ ,  $F_{420}H_2$  dehydrogenation activity.  $\circ$ ,  $F_{420}$ -reducing activity.  $\bullet$ , Methyl viologen reducing activity.

Table 6.1. Purification of the soluble phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of methanol-grown Methanosarcina barkeri Fusaro. MCAC, metal chelate affinity chromatography.

Purification stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Sp. activity <sup>‡</sup>	Purification
Cell-free extract	723.9	466.9	100	0.65	-
High speed pellet	264.3	345.4	74.0	1.31	2.0-fold
MCAC pool	78.8	158.8	34.0	2.02	3.1-fold
Eluted protein <sup>‡‡</sup>	2.9	55.4	11.9	19.10	29.4-fold

μmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>.

dehydrogenation activity were yellowish-brown in color, indicating the presence of flavins and iron-sulfur clusters. The purified enzyme was stored at 4°C under an atmosphere of N<sub>2</sub> for up to 3 months without significant loss of activity.

Excision and electrophoresis of lower molecular weight bands of similar color and/or molecular weight to the F<sub>420</sub>H<sub>2</sub> dehydrogenase of Ml. tindarius and Ms. mazei Gö1 (~120 kDa) from the native PAGE did not reveal protein containing F<sub>420</sub>H<sub>2</sub> dehydrogenation or  $F_{420}$   $H_2$  as activity. The purification of the  $F_{420}H_2$  dehydrogenation activity is summarized in Table 6.1; this process was repeated three times, and representative data are presented here.

Concurrent with F<sub>420</sub>H<sub>2</sub> dehydrogenation purification, the F<sub>420</sub> H<sub>2</sub>ase activity was also tracked and monitored. The ratio of F<sub>420</sub>H<sub>2</sub> dehydrogenation:F<sub>420</sub> hydrogenase activity throughout the enrichment process indicates that the F420 H2ase was enriched at roughly the same rate as the  $F_{420}H_2$  dehydrogenation activity, strengthening the argument

<sup>‡</sup>μmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein ‡‡protein eluted from 3-20% native gradient PAGE

Table 6.2. Enrichment of the soluble  $F_{420}$  reducing hydrogenase activity during the purification of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of methanolgrown *Methanosarcina barkeri* Fusaro. MCAC, metal chelate affinity chromatography.

Purification stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Sp. activity <sup>‡</sup>	Purification	DeH <sub>2</sub> ase/ H <sub>2</sub> ase
Cell-free extract	723.9	661.3	100	0.91	-	0.71
High speed pellet	264.3	547.8	82.8	2.07	2.3-fold	0.63
MCAC pool	78.8	287.1	43.4	3.64	4.0-fold	0.55
Eluted protein <sup>‡‡</sup>	2.9	46.5	7.0	16.03	17.6-fold	1.19

<sup>&</sup>lt;sup>†</sup> μmol F<sub>420</sub> reduced·min<sup>-1</sup>.

<sup>‡</sup>µmol F<sub>420</sub> reduced·min<sup>-1</sup>·mg<sup>-1</sup> protein

for the association of the two enzyme activities.

When the HSP was loaded onto the  $\mathrm{Ni}^{2+}$  affinity column, only the soluble  $\mathrm{F}_{420}\mathrm{H}_2$  dehydrogenation activity binds to the column, while the membrane-bound activity flows through the column unhindered. Analyses of the unpublished and incompletely sequenced genome of Ms. barkeri Fusaro (circa. 2003-2004) indicated the presence of genes encoding for a putative Fpo complex. The electron transport apparatus of the Methanosarcinaceae is located in the cell membranes; to ensure that the activity of the Fpo complex was not being masked by the presence of a similar activity by the  $\mathrm{F}_{420}$   $\mathrm{H}_2$ ase, the phenazine-dependent  $\mathrm{F}_{420}\mathrm{H}_2$  dehydrogenation activity was also purified from membranes collected from the  $\mathrm{Ni}^{2+}$  affinity column.

Protein was extracted from washed and CHAPS-solubilized membranes, and processed in the same manner as the soluble activity. The eluate from the Ni<sup>2+</sup> affinity

<sup>‡‡</sup>protein eluted from 3-20% native gradient PAGE

column containing  $F_{420}H_2$  dehydrogenation activity was desalted, concentrated and electrophoresed on a 3-20% native gradient PAGE. The high molecular weight bands were excised from the gel, minced into 1-2 mm<sup>3</sup> pieces, and the protein extracted from the minced gel pieces. The results are summarized in Tables 6.3 (phenazine-dependent  $F_{420}H_2$  dehydrogenation) and 6.4 ( $F_{420}$   $H_2$ ase). A constant ratio of  $F_{420}H_2$  dehydrogenation:  $F_{420}$   $H_2$ ase activity is maintained throughout the purification process. Electrophoresis of the protein on a native 3-20% gradient PAGE, followed by hydrogenase activity staining, indicates the presence of the  $F_{420}$   $H_2$ ase, and not the presence of the  $F_{420}H_2$  dehydrogenase (Figure 6.6). Protein was also extracted from

Table 6.3. Purification of the membrane-bound phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of methanol-grown *Methanosarcina barkeri* Fusaro. MCAC, metal chelate affinity chromatography.

Purification stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Sp. activity <sup>‡</sup>	Purification
Cell-free extract	979.0	561.9	_	0.57	-
Membranes	385.2	129.2	100	0.34	-
Solubilized protein <sup>††</sup>	42.4	35.7	27.6	0.84	2.5-fold
MCAC pool	18.6	27.8	21.6	1.49	4.4-fold
Eluted protein <sup>‡‡</sup>	0.6	8.0	5.8	12.50	36.8-fold

<sup>†</sup> μmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>.

<sup>&</sup>lt;sup>‡</sup>μmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein

protein extracted from solubilized membranes

ttprotein eluted from 3-20% native gradient PAGE

Table 6.4. Enrichment of the membrane-bound  $F_{420}$  reducing hydrogenase activity during the purification of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of methanolgrown *Methanosarcina barkeri* Fusaro. MCAC, metal chelate affinity chromatography.

Purification stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Sp. activity <sup>‡</sup>	Purification	DeH <sub>2</sub> ase/
Cell-free extract	979.0	899.0	_	0.92	-	0.62
Membranes	385.2	233.7	100	0.61	-	0.56
Solubilized protein <sup>††</sup>	42.4	67.8	27.6	1.60	2.6-fold	0.53
MCAC pool	18.6	54.2	21.6	2.91	4.8-fold	0.51
Eluted protein <sup>‡‡</sup>	0.6	11.6	4.4	19.33	31.7-fold	0.64

<sup>†</sup> μmol F<sub>420</sub> reduced-min<sup>-1</sup>.

<sup>‡</sup>µmol F<sub>420</sub> reduced·min<sup>-1</sup>·mg<sup>-1</sup> protein

‡‡protein eluted from 3-20% native gradient PAGE

gel pieces excised from the native PAGE gel with MW in the range of the  $F_{420}H_2$  dehydrogenase of *Ml. tindarius* and *Ms. mazei* Gö1 (~120 kDA); neither  $F_{420}H_2$  dehydrogenation nor  $F_{420}$ -reducing activity was detected in these fractions.

## 6.3.3. Properties of the $F_{420}H_2$ dehydrogenation activity

### 6.3.3.1. Molecular properties

Electrophoresis of the purified  $F_{420}H_2$  dehydrogenation activity on a nondenaturing 3-20% native PAGE (Figure 6.5) revealed the presence of two prominent bands. One band has a molecular mass greater than 669 kDa, while the second band has a mass of approximately 198 kDa; the profile is identical to the pattern observed during

<sup>††</sup>protein extracted from solubilized membranes

the purification of the F<sub>420</sub> H<sub>2</sub>ase from *Ms. barkeri* Fusaro (Fiebig and Friedrich 1989, Michel *et al.* 1995).

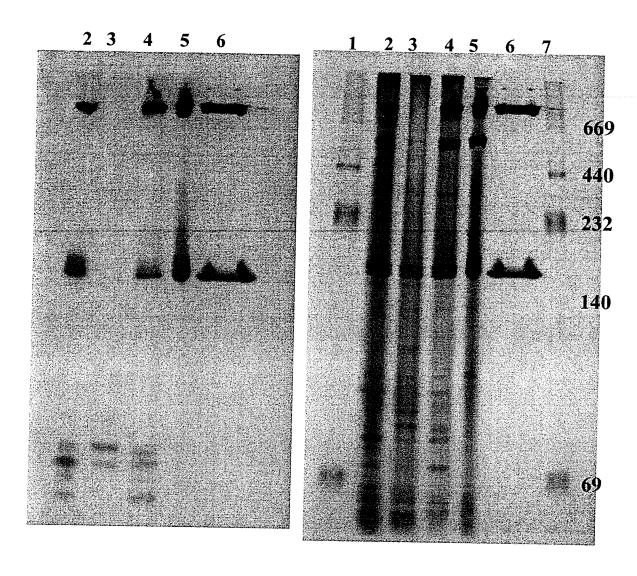
Under an anaerobic atmosphere of 10% H<sub>2</sub> (balance N<sub>2</sub>) and immersion in 50 mM Tris/Cl, pH 8.0, with 20 mM MgCl<sub>2</sub>·6H<sub>2</sub>O, and 20 mM DTT, H<sub>2</sub> uptake-activity by the gel-bound protein was observed, using methyl viologen (2 mM) as electron acceptor (Figure 6.5A). The positions of the activity bands are identical with the bands on the coomassie blue-stained native PAGE gel (Figure 6.5B).

Purification of the membrane-bound  $F_{420}H_2$  dehydrogenation activity alongside the soluble activity reveals that the  $F_{420}$   $H_2$  as catalyzes both activities, and that a distinct Fpo complex is not being detected under the growth conditions. The hydrogenase activity- and coomassie blue-stained native gradient PAGE gels show that the soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activities are identical (Figure 6.6).

A double stained (coomassie blue and silver-stained) SDS PAGE of the purified enzyme (Figure 6.7) indicates the presence of three major polypeptides, corresponding to apparent molecular masses of 48 ( $\alpha$ ), 33 ( $\beta$ ), and 30 ( $\gamma$ ) kDa as reported for the F<sub>420</sub> H<sub>2</sub>ase of *Ms. barkeri* Fusaro by Fiebig and Friedrich (1989).

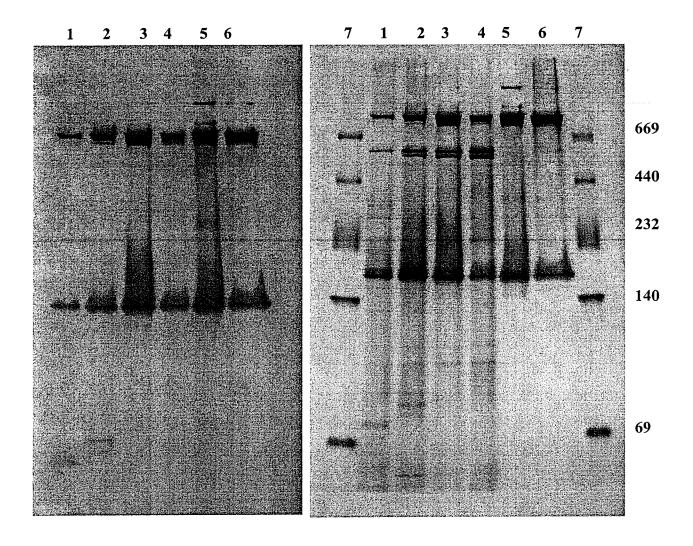
N-terminus sequencing of the  $\alpha$ - and  $\beta$ -subunits confirmed that the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was the  $F_{420}$   $H_2$  as (Table 6.5); sequence information for the  $\gamma$ -subunit was inconclusive. No other source of  $F_{420}H_2$  dehydrogenation activity could be detected over the course of this study, regardless of method of chromatography used to resolve the proteins. The sequences of the  $\alpha$ - and  $\beta$ -subunits, excluding the mismatches, are identical to the sequences reported by Michel *et* 

А. В.

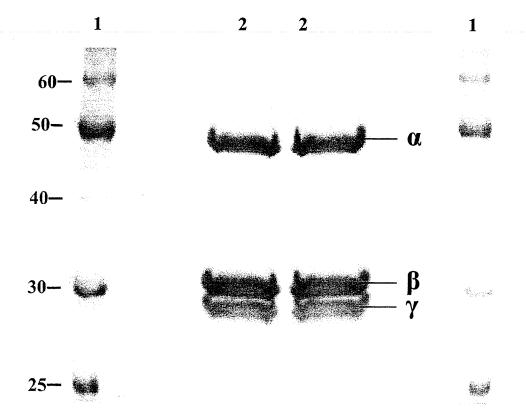


**Figure 6.5**. (A) Activity- and (B) coomassie blue-stained 3-20% native gradient PAGE showing progression of the purification of the soluble phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of  $CH_3OH$ -grown  $Ms.\ barkeri$  Fusaro. Pattern of protein bands observed suggests presence of  $F_{420}$ -reducing hydrogenase. Lanes 1 and 7, ladder, kDa. Lane 2. cell-free extract. Lane 3. clarified cell-free extract. Lane 4. high speed pellet. Lane 5. MCAC pooled protein. Lane 6. purified protein.

A. B.



**Figure 6.6**. Electrophoresis of proteins at various stages of purification of the soluble and membrane-bound phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of CH<sub>3</sub>OH-grown *Ms. barkeri* Fusaro, as visualized on: (A) Activity-stained, and (B) activity- and coomassie blue-stained native gradient 3-20% PAGE gel. Pattern of activity and protein bands observed suggests presence of  $F_{420}$ -reducing hydrogenase. 1. cell-free extract. 2. high speed pellet (HSP). 3. MCAC pooled protein from HSP (soluble fraction). 4. MCAC pooled protein from solubilized membranes. 5. purified soluble  $F_{420}H_2$  dehydrogenation activity. 6. purified (eluted) membrane-bound  $F_{420}H_2$  dehydrogenation activity. 7. molecular markers, kDa.



**Figure 6.7**. SDS PAGE profile of the purified soluble phenazine-dependent  $F_{420}H_2$  dehydrogenation activity from CH<sub>3</sub>OH-grown *Ms. barkeri* Fusaro. Pattern of subunits on coomassie- and silver-stained (12.8% acrylamide) gel suggests presence of  $F_{420}$ -reducing hydrogenase. 1. molecular markers (KDa). 2. purified protein.

al. 1995, and to the corresponding amino acid sequences predicted by the gene sequences of the frh and fre operons, encoding for iso-enzymes of F<sub>420</sub> H<sub>2</sub>ase in Ms. barkeri Fusaro.

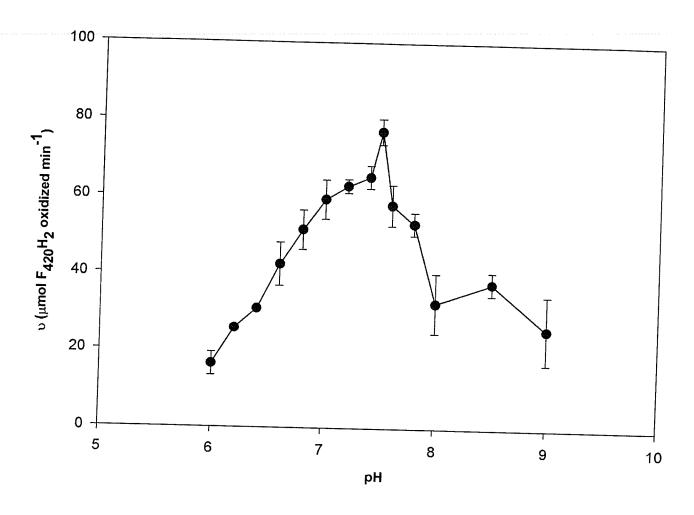
Table 6.5. N-terminus sequences of the  $\alpha$ - (48 kDa ) and  $\beta$ - (33 kDa ) subunits from the purified phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of Methanosarcina barkeri Fusaro (seen in Figure 7.7).

	A	β
This study	Thr-Lys-Val-Val-Glu-Ile-Ser- Pro-Thr-Thr	<b>Gly</b> <sup>†</sup> -Ile-Glu-Asp-Pro-Tyr- Leu <b>-Leu</b> <sup>‡</sup> -Lys- <b>Lys</b> <sup>‡</sup>
F <sub>420</sub> -reducing hydrogenase of <i>Ms. barkeri</i> Fusaro (Michel <i>et al.</i> 1995)	Thr-Lys-Val-Val-Glu-Ile-Ser- Pro-Thr-Thr	Met-Ile-Glu-Asp-Pro-Tyr- Leu-Gly-Lys-Tyr

<sup>‡</sup>ambiguous amino acid

## 6.3.3.2. Catalytic properties

The phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was purified 29.4-fold to apparent homogeneity with a yield of 11.9 % and a final specific activity of 19.1  $\mu$ mol  $F_{420}H_2$  oxidized·min  $^{-1}$  ·mg  $^{-1}$  protein. The  $F_{420}H_2$  dehydrogenation activity (500  $\mu M$ phenazine, room temperature (  $\sim$ 25°C)) was highest at pH 7.5 (Figure 6.8). The pH optimum for the purified F<sub>420</sub>H<sub>2</sub> dehydrogenase activities of Ml. tindarius and Ms. mazei Gö1 are, respectively, 6.8 and 8.5. The pH optimum for the  $F_{420}$   $H_2$  as activity of Ms. barkeri Fusaro was reported as pH 6.5-7.25, measured at 55°C (Michel et al. 1995).



**Figure 6.8**. pH profile of the purified soluble phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of CH<sub>3</sub>OH-grown *Ms. barkeri* Fusaro. Assay conditions as described under 'Materials and Methods'. All points represent experiments performed in triplicate. pH established using 40 mM KPO<sub>4</sub> (pH 6-8) and 50 mM Tris/Cl (pH 8-9), ambient room temperature (~25°C). 500  $\mu$ M phenazine added per tube.

The  $F_{420}H_2$  dehydrogenation activity of Ms. barkeri Fusaro was assayed primarily using phenazine (E°' = -0.265 to -0.165 V), but methyl viologen (E°' = -0.446 V) (with metronidazole) and FAD (E°' = -0.219 V) were also used to assay the  $F_{420}H_2$  dehydrogenation activity (Table 6.6). NADP<sup>+</sup>, NAD<sup>+</sup>, and ferredoxin were not able to substitute for phenazine as electron acceptors. NADH and NADPH could not substitute for  $F_{420}H_2$  as electron donor. The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity

Table 6.6. Electron acceptor specificities of the purified  $F_{420}H_2$  dehydrogenation activity of *Methanosarcina barkeri* Fusaro. 10-15  $\mu$ M  $F_{420}H_2$  added per assay tube.

$E^-$ acceptor / inhibitor ( $\mu$ M)	Activity (µmol F <sub>420</sub> H <sub>2</sub> oxidized·min <sup>-1</sup> ·mg <sup>-1</sup> )		
-	$0.3 \pm 0.05$		
Methylviologen (500) + Metronidazole (1000)	$3.1 \pm 0.4$		
Phenazine (500)	$19.5 \pm 1.2$		
Phenazine (500) + DMSO (100 μL)	$18.0 \pm 1.3$		
DPI (100)	$0.3 \pm 0.05$		
Phenazine (500) + DPI (50)	$15.0 \pm 1.2$		
Phenazine (500) + DPI (100)	$12.4 \pm 1.0$		
FAD (50)	15.2 ± 1.4		
NADP <sup>+</sup> (300)	$0.4 \pm 0.05$		
NAD <sup>+</sup> (300)	$0.3 \pm 0.05$		
Ferredoxin (1 mg/ml)	$0.3 \pm 0.03$		

was inhibited by DPI, dissolved in Dimethyl sulfoxide (DMSO)), an inhibitor of mitochondrial and bacterial electron transport systems that also inhibits the  $F_{420}H_2$  dehydrogenation activity of *Ms. mazei* Göl (Brodersen *et al.* 1999).

DPI was also tested for its inhibitory effect on the  $F_{420}$   $H_2$ ase of Ms. barkeri Fusaro, since DPI is an inhibitor of flavoproteins. The  $F_{420}$   $H_2$ ase was reductively activated prior to assay, incubated under  $H_2$  with 10  $\mu$ M FAD, 10  $\mu$ M  $F_{420}$ , and 20 mM DTT. When using methyl viologen as electron acceptor, the protein was incubated in assay buffer containing 0.05 mM methyl viologen, under  $H_2$ ; the protein was considered reactivated when the buffer turned light blue in color.

Using standard concentrations of  $F_{420}$  (~20  $\mu$ M) or methyl viologen (2 mM),  $H_2$ -dependent  $F_{420}$ - or methyl viologen-reduction were not inhibited in the presence of DPI (Table 6.7). This would indicate that inhibition of the  $F_{420}H_2$  dehydrogenation activity is due to competition with phenazine for the binding site, and not due to direct attack of DPI to the FAD; the concentrations of DPI used inhibited  $F_{420}H_2$  dehydrogenation activity, but not  $F_{420}$   $H_2$  as activity (using equivalent amounts of protein). Reduction of methyl viologen was used as a negative control since the FAD component is not required for electron transfer to this substrate; inhibition of methyl viologen reduction was not observed (Table 6.7).

Hydrogenase activities were also assayed in the presence of phenazine (500  $\mu$ M) (Table 6.7). Phenazine (E°' = -0.265 to -0.165 V) appeared to compete with F<sub>420</sub> for electrons from H<sub>2</sub> (E°' = -0.420 V), since phenazine inhibited F<sub>420</sub>-reduction 10-fold. Another explanation is that as F<sub>420</sub> is reduced by the hydrogenase activity, the F<sub>420</sub>H<sub>2</sub> formed may be used to reduce phenazine by the dehydrogenation activity in a competing

Table 6.7. Effect of DPI on the hydrogenase activities of Methanosarcina barkeri Fusaro.  $[F_{420}] = 20 \mu M$ . [methyl viologen] = 2 mM. Protein reductively reactivated prior to addition of DPI or phenazine.

Electron acceptor	Addition (μM)	Activity	
F <sub>420</sub>	-	$44.0 \pm 3.2^{\dagger}$	
F <sub>420</sub>	100 μl DMSO	$45.0 \pm 2.2^{\dagger}$	
$F_{420}$	DPI (50)	$44.0 \pm 3.2^{\dagger}$	
F <sub>420</sub>	DPI (100)	$39.1 \pm 1.1^{\dagger}$	
$F_{420}$	DPI (150)	$41.0 \pm 4.3^{\dagger}$	
$F_{420}$	Phenazine (500)	$4.4\pm0.4^{\dagger}$	
Methyl viologen	-	$63.0 \pm 6.1^{\ddagger}$	
Methyl viologen	100 μl DMSO	$64.7 \pm 7.4^{\ddagger}$	
Methyl viologen	DPI (100)	$66.5 \pm 4.9^{\ddagger}$	
Methyl viologen	Phenazine (500)	$63.0 \pm 3.1^{\ddagger,\dagger\dagger}$	

reaction. Thus the apparent inhibition of the F<sub>420</sub> H<sub>2</sub>ase activity may be due to competition from the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation reaction.

Phenazine was also a substrate for the hydrogenase activity of the purified protein. Protein was incubated in F<sub>420</sub>H<sub>2</sub> dehydrogenation assay buffer (40 mM KPO<sub>4</sub>, pH 7.0) for 1 hour, under an H<sub>2</sub> atmosphere to insure that the protein was reductively reactivated; phenazine was reduced at a rate of  $14.6 \pm 1.3 \, \mu \text{mol phenazine red} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ .

 $<sup>^{\</sup>dagger}$  µmol F<sub>420</sub> reduced·min<sup>-1</sup>·mg<sup>-1</sup>  $^{\ddagger}$ µmol methyl viologen reduced·min<sup>-1</sup>·mg<sup>-1</sup>  $^{\dagger\dagger}$ full activity after 5-10 minute lag

Phenazine was not reduced under an atmosphere of  $N_2$  or in the absence of protein. The physiological relevance of this activity in Ms. barkeri Fusaro, if any, is unclear. In the presence of phenazine, methyl viologen ( $E^{\circ\prime} = -0.446 \text{ V}$ ) reduction lagged for several minutes ( $\sim 5$ -10 minutes), before reaching full activity. This may be due to the preference of the hydrogenase activity for phenazine as substrate, or possibly to transfer of electrons from the reduced methyl viologen to phenazine (either chemical or enzymatic).

As with *Msp. hungatei* GP1 (Chapter 5), there are differences with respect to the requirement for activation of the  $F_{420}H_2$  dehydrogenation and  $F_{420}$  H<sub>2</sub>ase activities. The  $F_{420}H_2$  dehydrogenation activity did not require a period of reactivation and was assayed immediately, even after aerobic manipulation. In contrast, the  $F_{420}$  H<sub>2</sub>ase activity is reversibly inactivated in the presence of  $O_2$ , and must be reactivated under reducing conditions to prevent a lag in activity (Fiebig and Friedrich 1989). The reactivation process was modified from Choquet and Sprott (1991), with incubation of the protein in  $10 \ \mu M$  FAD,  $10 \ \mu M$  F<sub>420</sub> and  $20 \ mM$  DTT, under an atmosphere of  $H_2$ , prior to assay. This is essential for recovery of the hydrogenase activity, since the protein is purified under aerobic conditions.

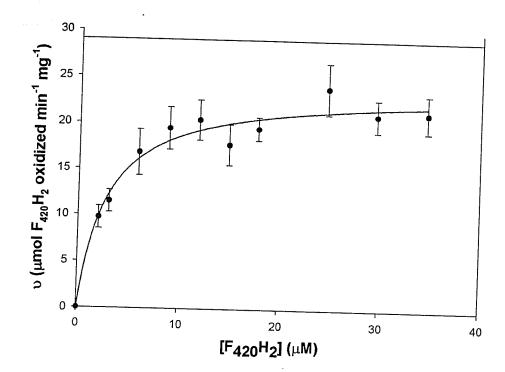
#### 6.3.3.3. Kinetic analyses

The apparent  $K_m$  for  $F_{420}H_2$  was determined to be 2.8  $\mu$ M  $F_{420}H_2$ , with Vmax = 23.7  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein (Michaelis Menten kinetics) (Lineweaver Burk kinetics:  $K_m$   $F_{420}H_2 = 3.0$   $\mu$ mol  $F_{420}H_2$ , Vmax = 23.9  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>) (Figure 6.9). In contrast, the apparent  $K_m$  for  $F_{420}H_2$  for *Ml. tindarius* and *Ms. mazei* Gö1 are 5.4 and 7  $\mu$ M  $F_{420}H_2$ , respectively (Haase *et al.* 1992, Abken and

Deppenmeier 1997). The apparent  $K_m$  for phenazine was determined to be 112.4  $\mu M$  (Figures 6.10 and 6.11; by contrast, the  $K_m$  of the phenazine-dependent activity of the  $F_{420}H_2$  dehydrogenation from Ms. mazei Gö1 (pure enzyme) and Ml. tindarius (membrane-bound activity) was 250 and 238.9  $\mu M$ , respectively (Abken et al. 1998, this thesis). The turnover number, kcat, was determined to be 416.9 s<sup>-1</sup> ( $k_{cat}/K_m = 37.2 \times 10^5 \cdot M^{-1} \cdot s^{-1}$ ).

Inhibition of the purified  $F_{420}H_2$  dehydrogenation activity was similar to that observed with the membrane fraction. Additions of 50 and 100  $\mu$ M DPI increased the  $K_m$  for phenazine from 112.4  $\mu$ M to 238.7 and 299.4  $\mu$ M, respectively, with Vmax = 29.5  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein (Lineweaver Burk kinetics: Additions of 50 and 100  $\mu$ M DPI increased the  $K_m$  for phenazine from 105.0  $\mu$ M to 195.0 and 268.0  $\mu$ M, respectively, with Vmax = 24.6  $\mu$ mol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein) (Figures 6.10 and 6.11).





#### B.

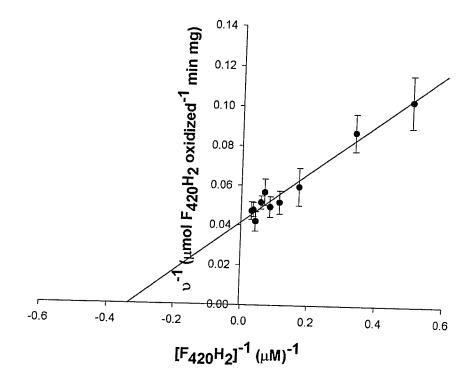


Figure 6.9. Michaelis Menten kinetics (A) and Lineweaver Burk plot (B) for the determination of  $KmF_{420}$  for the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Methanosarcina barkeri* Fusaro. 500  $\mu$ M phenazine added per tube.

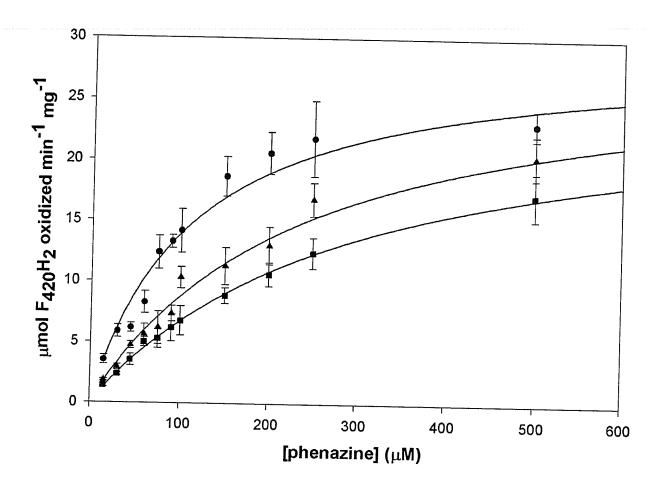


Figure 6.10. Michaelis Menten kinetics plot of the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in the presence/absence of DPI. All points represent experiments performed in triplicate. Assay conditions as described under 'Materials and Methods'. No DPI,  $\bullet$ . 50  $\mu$ M DPI,  $\blacktriangle$ . 100  $\mu$ M DPI,  $\blacksquare$ . 10-15  $\mu$ M  $F_{420}H_2$  added per assay tube.

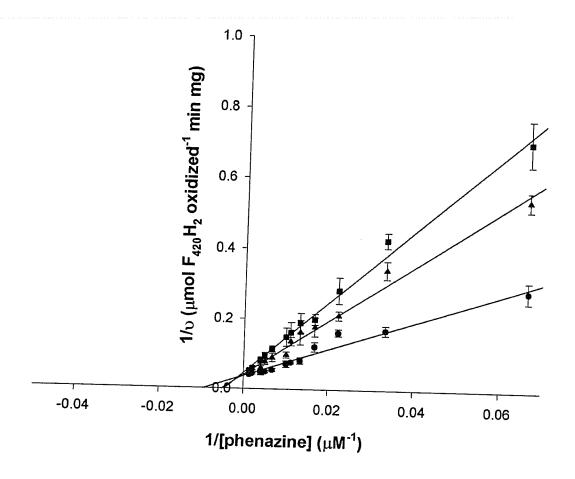


Figure 6.11. Lineweaver Burke kinetics plot of the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in the presence/absence of DPI. All points represent experiments performed in triplicate. Assay conditions as described under 'Materials and Methods'. No DPI, •. 50  $\mu$ M DPI, •. 100  $\mu$ M DPI, •. 10-15  $\mu$ M  $F_{420}H_2$  added per assay tube.

#### 6.4. Discussion

# 6.4.1. Comparing the $F_{420}H_2$ dehydrogenation activity of Methanosarcina barkeri Fusaro with Methanolobus tindarius and Methanosarcina mazei Göl

Comparisons of the genome of *Ms. barkeri* Fusaro with the genomes of *Ms. mazei* Gö1 and *Ms. acetivorans* C2 indicate that this methanoarchaeon contains the full complement of enzymes necessary for the metabolism of the various substrates, with all biosynthetic and catabolic genes conserved, including *fpo* (Deppenmeier *et al.* 2002, Galagan *et al.* 2002, Maeder *et al.* 2006).

Studies were conducted examining the phenazine-dependent  $F_{420}H_2$  dehydrogenase activities in the washed membranes of Ml. tindarius and Ms. barkeri Fusaro (Figure 6.1). The apparent  $K_m$  for phenazine of the  $F_{420}H_2$  dehydrogenation activity of Ml. tindarius was 238.9  $\mu$ M phenazine; the apparent  $K_m$  for phenazine for the homologous activity of Ms. barkeri Fusaro was 106.0  $\mu$ M phenazine, indicating that the activity of Ms. barkeri has a higher affinity for phenazine.

Addition of DPI has a more severe effect on the activity of Ml. tindarius, comparable to the inhibition of the purified  $F_{420}H_2$  dehydrogenase activity of Ms. mazei Gö1. The inhibition curve for Ml. tindarius ( $Ic_{50} = 3 \mu M$  DPI) shows a much steeper drop in dehydrogenation activity compared to Ms. barkeri Fusaro ( $Ic_{50} > 100 \mu M$  DPI) (Chapter 4).

Low concentrations of DPI significantly inhibit the membrane-bound  $F_{420}H_2$  dehydrogenation activity of Ml. tindarius, increasing the  $K_m$  from 238.9 to 965.6 (3  $\mu$ M DPI) and 1823.4  $\mu$ M phenazine (9  $\mu$ M DPI); this is consistent with the observations of DPI-mediated inhibition of the purified  $F_{420}H_2$  dehydrogenase activity of Ms. mazei Gö1 (assayed with 2-OH-phenazine) (Brodersen et al. 1999).

The membrane-bound  $F_{420}H_2$  dehydrogenation activity of Ms. barkeri Fusaro does not respond to DPI as expected under the growth conditions tested; the  $K_m$  increased from 106.0 to 204.0 (50  $\mu$ M DPI) and 342.3  $\mu$ M phenazine (100  $\mu$ M DPI). The total phenazine-dependent  $F_{420}H_2$  dehydrogenation activity measured should be due to the cumulative activity of the Fpo complex and  $F_{420}$   $H_2$ ase. If the Fpo complex were located in the membranes of Ms. barkeri Fusaro, a more prominent effect on the overall  $F_{420}H_2$  dehydrogenation activity would have been observed, as the activity of the Fpo complex should have been completely inhibited at the concentrations of DPI used. The remaining  $F_{420}H_2$  dehydrogenation activity would be due to the  $F_{420}$   $H_2$ ase, and simple Michaelis Menten kinetics would not be observed.

One conclusion that can be drawn from these results is that *Ms. barkeri* Fusaro apparently does not express the genes encoding the Fpo complex under the tested growth conditions; to confirm the apparent lack of expression of the Fpo complex, reverse transcriptase (RT)-PCR could be used to confirm the presence/absence of mRNA encoding for the Fpo complex in *Ms. barkeri* Fusaro under the growth conditions tested.

### 6.4.2. Purification of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity

Various chromatographic techniques were used to enrich for the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity from proteins extracted from solubilized membranes, but in each case the  $F_{420}H_2$  dehydrogenation activity was found to elute with an  $F_{420}$ -reducing activity (Figures 6.2 and 6.3); the purified  $F_{420}H_2$  dehydrogenase activities from *Ml. tindarius* and *Ms. mazei* Göl are devoid of hydrogenase activity

(Abken and Deppenmeier 1997). The membranes of Ml. tindarius do not contain an  $F_{420}$   $H_2$ ase (Deppenmeier et~al. 1989).

Purification of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity from the hydrogenotrophic methanoarchaea  $Msp.\ hungatei$  strain GP1 (Chapter 5) revealed that phenazine-dependent  $F_{420}H_2$  dehydrogenation activity is located on the  $F_{420}$   $H_2$ ase. In light of these findings, we believed that a phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in  $Ms.\ barkeri$  might also reside on its  $F_{420}$   $H_2$ ase.

The use of metal chelate affinity chromatography, sucrose gradient ultracentrifugation, and gel filtration chromatography established that the membrane-associated  $F_{420}H_2$  dehydrogenation activity was eluting with an  $F_{420}$   $H_2$  as activity (Figures 6.2 and 6.3). Other chromatographic techniques yielded similar results. The  $F_{420}$   $H_2$  as also catalyzes the oxidation of  $F_{420}H_2$ ; by selecting for the purification of this protein, we expected to separate this "artifact" activity from any other source of  $F_{420}H_2$  dehydrogenation present in the protein extract.

In an attempt to reduce the amount of time to purify the protein, and to increase the yield, the sucrose gradient ultracentrifugation and gel filtration steps were replaced with electrophoresis of the protein on a native gradient PAGE, followed by electroelution of the bands of interest. Elution of protein from a 3-20% native gradient PAGE confirmed that the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was indeed located on the  $F_{420}$   $H_2$  ase. Other bands excised from the gel, including gel bands with similar MW as the  $F_{420}H_2$  dehydrogenase of *Ml. tindarius* and *Ms. mazei* Gö1, did not contain  $F_{420}H_2$  dehydrogenase or  $F_{420}$   $H_2$  ase activity.

The results were the same whether the activity was isolated from the soluble fraction, or from the membranes of Ms. barkeri Fusaro. No other source of  $F_{420}H_2$  dehydrogenation activity was observed in Ms. barkeri Fusaro. It is not clear whether this activity is an artifact of the  $F_{420}$   $H_2$ ase, or if this activity holds any physiological significance.

### 6.4.3. Molecular properties

As with *Msp. hungatei* GP1 (Chapter 5), the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Ms. barkeri* Fusaro was isolated on the  $F_{420}$  H<sub>2</sub>ase. This was evident after electrophoresis of purified protein on native gradient PAGE (Figures 6.5 and 6.6) and SDS PAGE (Figure 6.7). The protein consists of three distinct subunits with molecular weights of 48, 33, and 30 kDa. The presence of the  $F_{420}$  H<sub>2</sub>ase was confirmed with N-terminal sequencing of the 48 kDa ( $\alpha$ -) and 33 kDa ( $\beta$ -) subunits (Table 6.5).

In contrast, the F<sub>420</sub>H<sub>2</sub> dehydrogenase activity purified from *Ml. tindarius* and *Ms. mazei* Gö1 each consist of 5 distinct subunits with molecular weights ranging from of 45, 40, 22, 18, and 17 kDa (120 kDa overall, *Ml. tindarius*) and 40, 37, 22, 20, and 16 kDa (115 KDa overall, *Ms. mazei* Gö1) (Haase *et al.* 1992, Abken and Deppenmeier 1997).

Ms. barkeri Fusaro possesses two isoenzymes of  $F_{420}$  H<sub>2</sub>ase, Frh and Fre (Vaupel and Thauer 1998), but is not known if which of the isoenzymes was isolated during our studies. The N-terminus sequences derived from the  $\alpha$ - and  $\beta$ -subunits of the purified protein are identical to both isoenzymes. Whether Frh or Fre, or both, are the source of

the observed phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in *Ms. barkeri* Fusaro during growth on  $CH_3OH$  remains unclear.

The only significant difference found between the two isoenzymes are the FrhD  $(\delta-)$  and FreE  $(\epsilon-)$  subunits encoded by the *frhADGB* (F<sub>420</sub> H<sub>2</sub>ase II) and *freAEGB* (F<sub>420</sub> H<sub>2</sub>ase I), respectively. BLAST analyses confirm that FrhD (159 aa) is a hydrogenase maturation factor, while FreE (40 aa) does not show any similarity to any sequences found in the Eukarya, Bacteria, or Archaea databases. Further analyses using the Loctree (Nair and Rost 2005) and Subloc v1.0 (Hua and Sun 2001) software indicate that FreE is an extracellular or periplasmic protein, respectively; FrhD was designated a cytoplasmic protein as expected. Thus, neither the FrhD nor FreE subunits appear to be factors that influence the presence of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity.

### 6.4.4. Catalytic and kinetic analyses

The  $F_{420}H_2$  dehydrogenation activity of Ms. barkeri Fusaro is similar to the activity of the  $F_{420}H_2$  dehydrogenase purified from Ms. mazei Gö1: both activities use  $F_{420}H_2$  as electron donor; both activities can use phenazine as electron acceptor; and both activities are inhibited by the addition of DPI. Methyl viologen and FAD could substitute for phenazine as substrate for the  $F_{420}H_2$  dehydrogenation activity, but  $NAD^+$ ,  $NADP^+$ , and ferredoxin could not replace phenazine for use as electron acceptors.

The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of  $Ms.\ barkeri$  Fusaro was competitively inhibited by DPI, but the degree of inhibition was not as great as observed in  $Ms.\ mazei$  Gö1 (Brodersen  $et\ al.\ 1999$ ) or  $Ml.\ tindarius$ . DPI inhibition of the  $F_{420}H_2$  dehydrogenation activity of  $Ms.\ barkeri$  Fusaro follows simple Michaelis-

Menten enzyme kinetics. Additions of 50 and 100  $\mu$ M DPI increased the  $K_m$  for phenazine from 112.4  $\mu$ M to 241.3 and 301.9  $\mu$ M, respectively, and  $V_{max} = 29.5 \ \mu mol$   $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> protein.

The  $F_{420}$  H<sub>2</sub>ase activity of Ms. barkeri Fusaro was not inhibited by the addition of DPI, indicating that DPI was not interacting directly with the FAD component. Phenazine could substitute for  $F_{420}$  as electron acceptor for the  $F_{420}$  H<sub>2</sub>ase (Table 6.7). The physiological significance of this activity, if any, is unclear. Meuer  $et\ al.\ 1999$  reported that the  $F_{420}$  H<sub>2</sub>ase from Mtb.  $thermoautotrophicus\ \Delta H$  also catalyzes the H<sub>2</sub>-dependent reduction of phenazine, yet methanophenazine is not found in this methanoarchaeon. Phenazine may be functioning solely as an artificial electron acceptor, like methyl or benzyl viologen.

# 6.4.5. Location of the $F_{420}H_2$ dehydrogenation activity on the $F_{420}$ -reducing hydrogenase – physiological ramifications

During growth on  $H_2$  and  $CO_2$ , the  $F_{420}$   $H_2$  ase of Ms. barkeri Fusaro catalyzes the reversible reduction of  $F_{420}$  with  $H_2$  (Thauer et al. 1993, Shima et al. 2002, Deppenmeier 2004). Electrons from  $F_{420}H_2$  are transferred to  $H_4MPT$ -bound  $C_1$  intermediates in order to reduce  $CO_2$  to  $CH_3$ - $H_4MPT$  (Thauer et al. 1993):

(1) 
$$F_{420}H_2 + N^5$$
,  $N^{10}$ -methenyl- $H_4MPT \rightarrow N^5$ ,  $N^{10}$ -methylene- $H_4MPT + F_{420}$ 

(2) 
$$F_{420}H_2 + N^5$$
,  $N^{10}$ -methylene- $H_4MPT \rightarrow N^5$ -methyl- $H_4MPT + F_{420}$ 

However, when grown on  $CH_3OH$  there is no apparent need for  $H_2$ -dependent reduction of  $F_{420}$ , as  $F_{420}H_2$  is synthesized through the course of  $CH_3OH$  disproportionation to  $CH_4$ , using the same reactions as in  $CO_2$  reduction, but in the reverse direction. 1 mol of  $CH_3OH$  is oxidized to  $CO_2$  to produce  $F_{420}H_2$  for the reduction of

CoM-S-S-CoB generated by the remaining 3 mol of CH<sub>3</sub>OH (Deppenmeier *et al.* 1996, Bäumer *et al.* 1998).

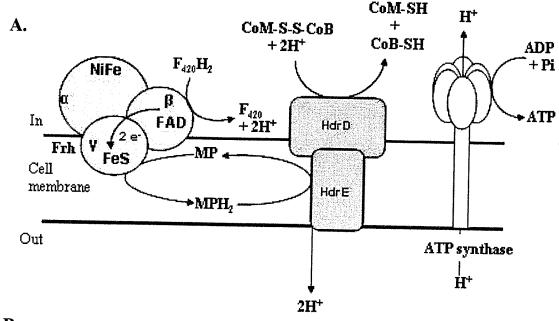
 $F_{420}H_2$ -dependent CoM-S-S-CoB reduction is catalyzed by the Fpo complex and Hdr in the membranes of Ms. mazei Gö1; methanophenazine is the purported electronic link between Fpo and Hdr (Deppenmeier 2002, 2004). We did not detect the presence of a similar complex in Ms. barkeri Fusaro, despite the presence of genes encoding for an Fpo complex in the genome of this methanoarchaeon. The membrane-bound  $F_{420}H_2$  dehydrogenation activity was found solely on the  $F_{420}$   $H_2$  ase, and no other source of  $F_{420}H_2$  dehydrogenation activity was detected. Furthermore, the  $F_{420}H_2$  dehydrogenation activity was phenazine-dependent; the cytoplasmic portion of the Fpo complex of Ms. mazei Gö1 was also assayed using phenazine (Abken et al. 1998).

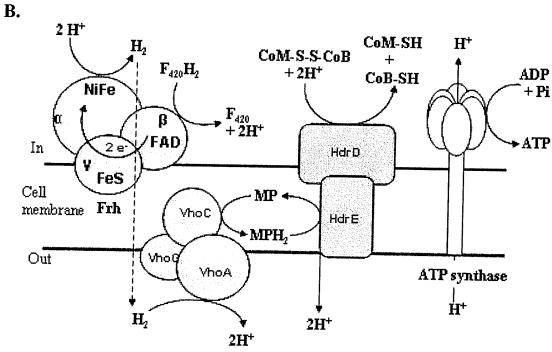
Our studies indicate that  $F_{420}H_2$  dehydrogenation activity is highest when  $Ms.\ barkeri$  Fusaro is grown with CH<sub>3</sub>OH (Table 4.1, Chapter 4); this is consistent with the findings of Mukhopadhyay  $et\ al.$  (1993), who reported differential expression of the  $F_{420}$  H<sub>2</sub>ase of  $Ms.\ barkeri$  when grown using different methanogenic substrates. Production of the  $F_{420}$  H<sub>2</sub>ase was highest when grown on CH<sub>3</sub>OH (0.437  $\pm$  0.0507  $\mu$ mol  $F_{420}$ -reduced·min<sup>-1</sup>·mg<sup>-1</sup> protein), compared to growth on H<sub>2</sub>/CO<sub>2</sub> (0.340  $\pm$  0.0386  $\mu$ mol  $F_{420}$ -reduced·min<sup>-1</sup>·mg<sup>-1</sup> protein) and acetate (0.080  $\pm$  0.0131  $\mu$ mol  $F_{420}$ -reduced·min<sup>-1</sup>·mg<sup>-1</sup> protein) (cultures grown to mid-log phase, Mukhopadhyay  $et\ al.$  (1993)).

Brodersen *et al.* (1999b) discovered that the membrane-bound  $F_{420}$   $H_2$  as of the hydrogenotrophic methanoarchaeon *Mc. voltae* was the source of an  $F_{420}H_2$  dehydrogenation activity coupled to membrane-bound CoM-S-S-CoB reduction.

 $Mc.\ voltae$  can grow either on  $H_2/CO_2$  or on formate as substrate for methanogenesis. During growth on  $H_2/CO_2$   $F_{420}H_2$  is not required as an electron donor for CoM-S-S-CoB reduction, since  $H_2$  can be used directly by the  $F_{420}$ -nonreducing  $H_2$  ase: however, when growing on formate,  $F_{420}H_2$ , generated from formate dehydrogenation, is the source of reductant (Brodersen  $et\ al.\ 1999b$ ). The identity of the intermediate electron carrier linking the passage of electrons from  $F_{420}H_2$ -oxidizing activity (of the  $F_{420}\ H_2$  ase) and the Hdr in  $Mc.\ voltae$  is currently not known.

It has been suggested that the  $F_{420}$   $H_2$ ase or  $F_{420}$ -nonreducing  $H_2$ ase of Ms. barkeri Fusaro could be involved in energy conservation (Lünsdorf et al. 1991). From the results of our studies, it is plausible that electrons may be transferred from  $F_{420}H_2$  to an electron acceptor such as methanophenazine via the  $F_{420}H_2$  dehydrogenation activity associated with the  $F_{420}$   $H_2$ ase (Figure 6.12A). Methanophenazine, which would act as the electronic link between the  $F_{420}$   $H_2$ ase and  $F_{420}$   $H_2$  would then transfer the electrons from reduced methanophenazine to CoM-S-S-COB. In contrast to the Fpo complex, the  $F_{420}$   $H_2$ ase does not act as a proton pump, as it lacks a transmembrane proton channel (Lünsdorf et al. 1991). It is not known whether the  $F_{420}$   $H_2$ ase may be attached to another integral membrane protein that may function as a proton pump. Protons would be pumped out of the cell during the reduction of CoM-S-S-COB, with electrons derived from reduced methanophenazine; the  $\Delta \tilde{u}H^+$  generated would drive the synthesis of ATP from ADP and  $F_{11}$   $F_{12}$   $F_{13}$   $F_{14}$   $F_{15}$   $F_{15}$  F





**Figure 6.12**. Possible models of  $F_{420}H_2$ -dependent CoM-S-S-CoB reduction in  $CH_3OH$ -grown *Methanosarcina barkeri* Fusaro. A. Model 1:  $F_{420}$ -reducing hydrogenase (Frh) catalyzes the  $F_{420}H_2$ -dependent reduction of methanophenazine (MP). MPH<sub>2</sub> used to reduce CoM-S-S-CoB *via* Hdr.  $\Delta \tilde{u}H^+$  generated used to synthesize ATP. B. Model 2:  $F_{420}H_2$  oxidized to  $H_2$  *via* Frh, which diffuses through cell membrane.  $H_2$  is then oxidized *via* the  $F_{420}$ -nonreducing hydrogenase (Vho), and electrons are transferred to MP. MPH<sub>2</sub> used to reduce CoM-S-S-CoB *via* Hdr.

It has been suggested that the  $F_{420}$  H<sub>2</sub>ase, during growth on CH<sub>3</sub>OH, may be a redox valve to regulate the levels of  $F_{420}$ H<sub>2</sub> and H<sub>2</sub> in cultures of *Ms. barkeri* Fusaro (Keltjens and Vogels 1993, de Poorter *et al.* 2005). With respect to energy conservation, the  $F_{420}$  H<sub>2</sub>ase may oxidize  $F_{420}$ H<sub>2</sub> to  $F_{420}$  and H<sub>2</sub>, producing a source of electrons used by the  $F_{420}$ -nonreducing H<sub>2</sub>ase (Vho) to reduce methanophenazine (Figure 6.12B). The  $F_{420}$ -nonreducing H<sub>2</sub>ase is produced when *Ms. barkeri* Fusaro is grown on CH<sub>3</sub>OH (Figure 6.5); the homologous membrane-bound extra-cytoplasmic enzyme in *Ms. mazei* Gö1 is also expressed during growth on CH<sub>3</sub>OH (Deppenmeier 1995, Deppenmeier *et al.* 1996). Vho catalyzes the reduction of 2-OH-phenazine, which generates an energy conserving  $\Delta\mu$ H<sup>+</sup>; the reduced methanophenazine is then used for the reduction of CoM-S-S-CoB *via* Hdr, also generating an energy conserving  $\Delta\mu$ H<sup>+</sup> (Ide *et al.* 1999, Deppenmeier 2002, 2004).

A model was suggested for *Msp. hungatei* GP1 (Chapter 5), where electrons are shuttled from the F<sub>420</sub> H<sub>2</sub>ase directly to the Hdr, and then to CoM-S-S-CoB (Figure 5.16, Chapter 5); this is likely not the case for *Ms. barkeri* Fusaro. Electron transport in the *Methanosarcinaceae* is dependent on cytochromes (VhoC and HdrE) and methanophenazine, electron carriers found within the membranes; these electron carriers are not found in methanoarchaea outside of the *Methanosarcinaceae* (Ide *et al.* 1999, Deppenmeier 2002, 2004).

It should be noted that in our studies, using a protocol derived from Abken et al. (1998), we have not detected the presence of methanophenazine from lyophilized membranes of Ms. barkeri Fusaro or Ml. tindarius; it is possible that the technique used to isolate the compound from the membranes of Ms. barkeri Fusaro or Ml. tindarius was

not sensitive enough detect to the low amounts of methanophenazine present; Beifuss and Tietze (2005) reported that there were very low amounts of methanophenazine isolated from the membranes of *Ms. mazei* Gö1.

Further studies are required to conclusively resolve the mechanism of  $F_{420}H_2$ -dependent CoM-S-S-CoB reduction in *Ms. barkeri* Fusaro, as well as the implications with respect to bioenergetics in this methanoarchaeon.

# 7. Phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity in *Methanosphaera stadtmanae*

#### 7.1. Introduction

Methanosphaera stadtmanae is a coccoid methanoarchaeon that resides in the human large intestine (Miller and Wolin 1983, 1985). Along with Methanosphaera cuniculi, a closely related methanoarchaeon isolated from the intestinal tract of rabbits (Biavati et al. 1988), Msph. stadtmanae belongs to the order Methanobacteriales (Miller and Wolin 1985). Msph. stadtmanae has strict growth requirements for acetate, CO<sub>2</sub>, isoleucine, and thiamin, and can only produce methane (CH<sub>4</sub>) in the presence of H<sub>2</sub> and methanol (CH<sub>3</sub>OH); this is in contrast to other members of the Methanobacteriales, which metabolize H<sub>2</sub> and CO<sub>2</sub> as methanogenic substrate (van de Wijngaard et al. 1991). Msph. stadtmanae can also produce CH<sub>4</sub> using H<sub>2</sub> and serine as methanogenic precursors, although the amount of CH<sub>4</sub> produced is approximately 50-fold lower compared to cells grown on H<sub>2</sub> and CH<sub>3</sub>OH (Lin and Sparling 1995, 1998).

H<sub>2</sub> and CH<sub>3</sub>OH are both metabolic waste products produced by other microorganisms indigenous to the large intestine: H<sub>2</sub> is a common byproduct of fermentation by the intestinal microbial community, and CH<sub>3</sub>OH is obtained in the gut via degradation of pectin by *Bacteroides thetaiotaomicron* (Miller and Wolin 1983, Dongowski *et al.* 2000). Reduction of CH<sub>3</sub>OH to CH<sub>4</sub> is coupled to energy conservation and ATP synthesis (Sparling *et al.* 1993).

## 7.1.1. Msph. stadtmanae and the Methanosarcinales: key differences

Msph. stadtmanae produces CH<sub>4</sub> when growing on CH<sub>3</sub>OH and H<sub>2</sub> in a manner similar to Methanosarcina barkeri, a member of the order Methanosarcinales, according to the following equation (van de Wijngaard et al. 1991):

$$H_2 + CH_3OH \rightarrow CH_4 + H_2O$$
,  $\Delta G^{\circ}$ '=-113 kJ/mol

Unlike the CH<sub>3</sub>OH-metabolizing methanoarchaea from the order *Methanosarcinales*, *Msph. stadtmanae* cannot disproportionate CH<sub>3</sub>OH or metabolise methylamines (Miller and Wolin 1983, 85). *Msph. stadtmanae* also does not possess cytochromes, which are characteristic of most CH<sub>3</sub>OH-metabolizing methanoarchaea (Miller and Wolin 1985).

In some members of the *Methanosarcinales* capable of CH<sub>3</sub>OH disproportionation, the  $F_{420}H_2$  dehydrogenase (Fpo complex,  $F_{420}H_2$  phenazine oxidoreductase) is expected to be an essential component of the  $F_{420}H_2$ :heterodisulfide oxidoreductase system (Deppenmeier *et al.* 1999, Deppenmeier 2002). Electrons from  $F_{420}H_2$  are transferred to the putative electron carrier methanophenazine *via* the  $F_{420}H_2$  dehydrogenase; the electrons are then transferred from reduced methanophenazine to the heterodisulfide reductase (Hdr) and ultimately to the heterodisulfide complex (CoM-S-S-CoB), generating an energy conserving transmembrane proton gradient ( $\Delta \mu H^+$ ) (Deppenmeier 2002, 2004).

Msph. stadtmanae possesses an  $F_{420}$ -nonreducing hydrogenase (MV  $H_2$ ase, MvhADG), which is found in most hydrogenotrophic methanoarchaea and is an essential component of the  $H_2$ :heterodisulfide oxidoreductase complex, transferring electrons from  $H_2$  to Hdr for the reduction of CoM-S-S-CoB and generation of  $\Delta\mu H^+$  (Thauer et al.

1993, Wong Ph.D thesis 1999, Shima *et al.* 2002). As such, neither F<sub>420</sub>H<sub>2</sub> nor F<sub>420</sub>H<sub>2</sub> dehydrogenase would be necessary during growth on H<sub>2</sub> and CH<sub>3</sub>OH. However, low levels of F<sub>420</sub>H<sub>2</sub> dehydrogenation activity have been detected in the cell-free extract of *Msph. stadtmanae* (Wong *et al.* 1994). The F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae* was studied primarily using the synthetic electron acceptor methyl viologen (with metronidazole) (Wong, Ph.D thesis 1999). Unlike the F<sub>420</sub>H<sub>2</sub> dehydrogenase purified from *Ml. tindarius* and *Ms. mazei* Gö1, the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity in *Msph. stadtmanae* is aerobically unstable (Wong *et al.* 1994).

#### 7.1.2. $F_{420}H_2$ Dehydrogenation activity of Methanosphaera stadtmanae

It has been proposed that the role of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity in *Msph. stadtmanae* is actually to couple the transfer of electrons from H<sub>2</sub> to the reduction of F<sub>420</sub>, as the presence of a H<sub>2</sub>-dependent F<sub>420</sub>-reducing activity had not been detected in cell-free extract (Deppenmeier *et al.* 1989, van de Wijngaard *et al.* 1991, Wong *et al.* 1994). F<sub>420</sub> has been detected in very low amounts in *Msph. stadtmanae* (0.16 nmol F<sub>420</sub>·mg<sup>-1</sup> protein); while F<sub>420</sub>H<sub>2</sub> is not required for methanogenesis from H<sub>2</sub> and CH<sub>3</sub>OH, F<sub>420</sub>H<sub>2</sub> is required for methanogenesis from H<sub>2</sub> and L-serine, or 2-propanol and CH<sub>3</sub>OH, and for cellular biosynthesis (Wong *et al.* 1994, Lin and Sparling 1995). One of the objectives of this thesis was to purify this activity, and to determine if this activity is could be linked to F<sub>420</sub>-reduction in *Msph. stadtmanae* as proposed. Studies of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae* focused on the use of phenazine as the electron acceptor, since methanophenazine is the putative physiological electron acceptor for the F<sub>420</sub>H<sub>2</sub> dehydrogenase activity in the *Methanosarcinaceae* (Abken *et al.* 1998).

#### 7.2. Materials and methods

For a comprehensive look at the materials and methods used for this chapter, the reader is referred to the following sections in Chapter 3:

- 3.2. Growth media
- 3.3. Cell lysis
- 3.6. Enzyme assays
  - 3.6.1. F<sub>420</sub>H<sub>2</sub> dehydrogenation assays
  - 3.6.2. Hydrogenase assays
    - 3.6.2.1. F<sub>420</sub>-reducing hydrogenase activity
    - 3.6.2.2. Methyl viologen reducing hydrogenase activity
    - 3.6.2.3. H<sub>2</sub>-dependent phenazine reduction
  - 3.6.3. F<sub>420</sub>H<sub>2</sub> oxidase activity of Methanosphaera stadtmanae
- 3.7. Gel electrophoresis
  - 3.7.2. Native gradient PAGE gel electrophoresis

In addition, several column chromatography techniques used to study the  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae* are described in Appendix E.

#### 7.3. Results

### 7.3.1. Catalytic properties of the $F_{420}H_2$ dehydrogenation activity in cell-free extract

The vast majority of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was found in the soluble cytoplasmic fraction (clarified cell-free extract, membranes removed *via* ultracentrifugation) of *Msph. stadtmanae*, with approximately 10-15% of the overall activity located in the membrane (Table 7.1). Therefore, the experimental work

in this chapter was performed using either clarified cell-free extract or cell-free extract (where noted), under anaerobic conditions.

Table 7.1. Location of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Methanosphaera stadtmanae*. Activity measured using 500  $\mu$ M phenazine, 10  $\mu$ M  $F_{420}H_2$ .

Cell fraction	Total activity (nmol F <sub>420</sub> H <sub>2</sub> oxid·min <sup>-1</sup> )	% activity	
Cell-free extract	4164.4 (3.2) <sup>†</sup>	100	
Clarified cell-free extract	3446.2 (3.5) <sup>†</sup>	82.7	
Membranes	518.2 (1.6) <sup>†</sup>	12.4	

<sup>\*</sup> specific activity: nmol F<sub>420</sub>H<sub>2</sub> oxid·min<sup>-1</sup>·mg<sup>-1</sup>

Several different substrates were tested in comparison to phenazine as electron acceptors for the  $F_{420}H_2$  dehydrogenation activity in cell-free extract. The results are summarized in Table 7.2. Phenazine is not the optimal electron acceptor in the reaction buffer used, as greater  $F_{420}H_2$  dehydrogenation activity is observed when using methyl viologen or FAD. The rate of  $F_{420}H_2$  oxidation using methyl viologen in place of phenazine is up to 3-fold higher compared to the activity when the same concentration of phenazine is used. FAD can also substitute for phenazine as electron acceptor; FAD catalyzes the chemical oxidation of  $F_{420}H_2$ , this amount is subtracted from the activity observed in the presence of clarified cell-free extract.

Table 7.2. Electron acceptor specificity of the  $F_{420}H_2$  dehydrogenation activity of *Msph.stadtmanae* in clarified cell-free extract. All values represent the average of three separate assays, each performed in triplicate. 10-12  $\mu$ M  $F_{420}H_2$  added per assay tube.

Electron acceptor (μM)	Activity (nmol F <sub>420</sub> H <sub>2</sub> oxidized·min <sup>-1</sup> ·mg <sup>-1</sup> )		
Phenazine (25)	$1.22 \pm 0.5$		
Phenazine (500)	$3.82 \pm 0.2$		
Phenazine (500) <sup>†</sup>	$1.24 \pm 0.2$		
Methyl viologen (500)	$7.3 \pm 0.4$		
Ferredoxin (1.0 mg/ml)	No activity detected		
FAD (40)	$5.69 \pm 0.6$		
NAD <sup>+</sup> (300)	No activity detected		
$O_2 (25)^{\ddagger}$	$3.5 \pm 0.3$		

<sup>&</sup>lt;sup>†</sup> 40 mM KPO<sub>4</sub> dehydrogenation assay buffer

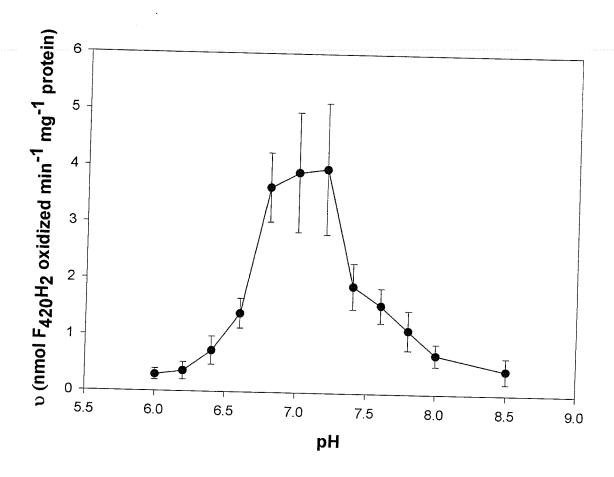
NADH could not substitute for  $F_{420}H_2$  as electron donor; similarly,  $NAD^+$  could not substitute for phenazine as an electron acceptor.  $NADP^+$  was not tested, but it is expected that  $F_{420}H_2$  oxidation would occur, as *Msph. stadtmanae* possesses an  $F_{420}H_2$ : $NADP^+$  oxidoreductase (Elias *et al.* 2000). Ferredoxin (from *Clostridium pasteuranium*) was also tested for its effectiveness as an electron acceptor, but  $F_{420}H_2$  dehydrogenation activity was not detected with this substrate.  $O_2$  (breathing grade air dissolved in Milli-Q  $H_2O$ , ~250  $\mu$ M) was also tested as an electron acceptor and was found to be an effective electron acceptor; the Km for  $O_2$  was approximately 10-fold

<sup>&</sup>lt;sup>‡</sup> breathing grade air-saturated MilliQ  $H_2O$  ([O<sub>2</sub>] = ~250  $\mu$ M), 200  $\mu$ L added per tube

lower than that of phenazine, indicating a higher affinity for  $O_2$  as an electron acceptor for the  $F_{420}H_2$  dehydrogenation activity (See section 7.3.10). In the absence of protein,  $F_{420}H_2$  oxidation was not observed upon addition of  $O_2$ .  $O_2$  and the possible physiological function of the  $F_{420}H_2$  dehydrogenation activity will be discussed at the conclusion of this chapter.

### 7.3.2. pH Profile of the $F_{420}H_2$ Dehydrogenation of Msph.stadtmanae

A pH profile had been compiled for the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity in cell-free extract, using methyl viologen as electron acceptor; the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was greatest at pH 7.0 (Denny Wong, PhD thesis, 1999). A similar pH profile was observed when using phenazine as electron acceptor, with the optimum activity found between pH 6.8-7.3 (Figure 7.1). This is similar to the pH optimum of the purified F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of *Ml. tindarius*, whose activity was highest at pH 6.8 (Haase *et al.* 1992), and of the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Ms. barkeri* Fusaro (pH 7.5, this thesis). In contrast, the pH optimum for the purified F<sub>420</sub>H<sub>2</sub> dehydrogenase activities from *Ms. mazei* Gö1 and sulfate-reducing archaeon *Archaeoglobus fulgidus* were pH 8.5 and pH 8.0, respectively (Kunow *et al.* 1994, Abken and Deppenmeier 1997). In the case of *Ml. tindarius*, *Ms. mazei* Gö1, and *A. fulgidus*, the pH profiles of the respective F<sub>420</sub>H<sub>2</sub> dehydrogenation activities were determined using methyl viologen as electron acceptor.



**Figure 7.1**. pH profile of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in clarified cell-free extract. pH of assay buffer established by varying concentration of  $K_2HPO_4$  and  $KH_2PO_4$ . Final  $[KPO_4] = 200$  mM.  $[F_{420}] = 10$ -15  $\mu$ M

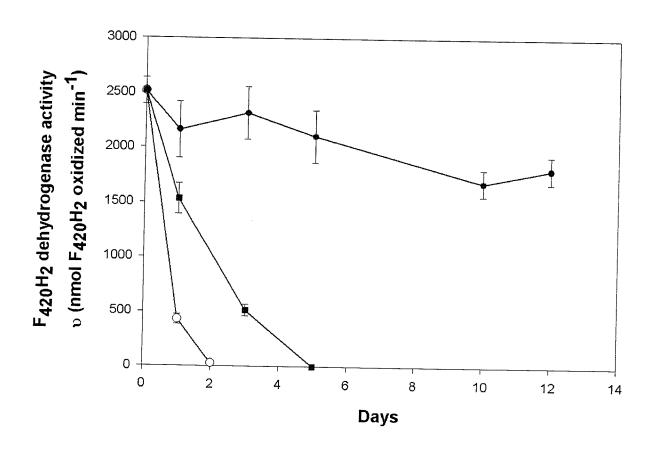


Figure 7.2. Stability of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in clarified cell-free extract (CCFE). ( $\circ$ ), aerobic, 4 $^{\circ}$ C. ( $\bullet$ ), anaerobic, 4 $^{\circ}$ C. ( $\bullet$ ), anaerobic, ambient room temperature.  $O_2$  in aerobic sample purged with  $N_2$  prior to assay.

### 7.3.3. Stability of the $F_{420}H_2$ Dehydrogenation activity

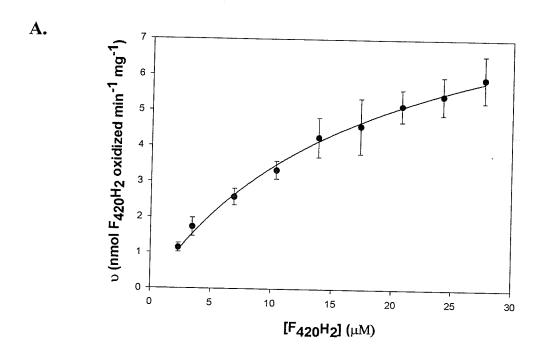
The  $F_{420}H_2$  dehydrogenation activity was stable under refrigerated, anaerobic conditions, in unprocessed clarified cell-free extract (Figure 7.2). The activity was stable for up to two weeks, with or without the presence of  $(NH_4)_2SO_4$ . This is in contrast to the stability of the activity under aerobic conditions, where the enzyme activity was greatly diminished after one day; assay of the  $F_{420}H_2$  dehydrogenation activity after incubation for 24 hour under aerobic conditions (4°C) resulted in approximately 80 to 85% loss of activity, and inactivity after 48 hours. Storage of the CCFE at ambient room temperature  $(25^{\circ}C)$  results in rapid loss of activity, with the protein rendered inactive after 4-5 days.

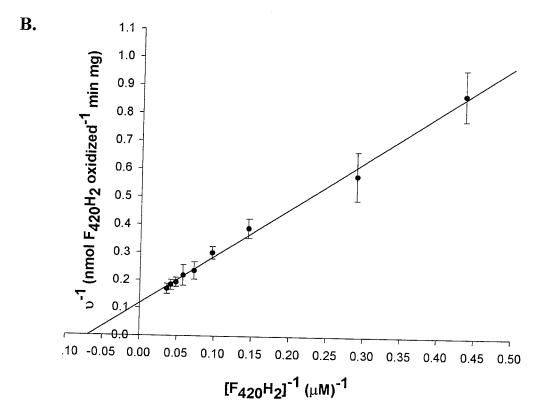
By comparison, the phenazine-dependent  $F_{420}H_2$  dehydrogenation activities of *Ms. barkeri* Fusaro and *Msp. hungatei* GP1, associated with the respective  $F_{420}$   $H_2$ ase, were stable under aerobic conditions, stored at 4°C (chapters 5 and 6).

## 7.3.4. Kinetic analyses of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity

The  $F_{420}H_2$  dehydrogenation activity as catalyzed by clarified cell-free extract follows simple Michaelis-Menten kinetics. The apparent  $K_m$  for  $F_{420}H_2$  was determined to be 18.4  $\mu$ M, with  $V_{max} = 9.7$  nmol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> (Lineweaver Burk:  $K_m$   $F_{420}H_2 = 14.5$   $\mu$ M,  $V_{max} = 8.5$  nmol  $F_{420}H_2$  oxid·min<sup>-1</sup>·mg<sup>-1</sup>) (Figure 7.3).

The apparent  $K_m$  for phenazine was calculated to be 138  $\mu$ M, with  $V_{max} = 3.0$  nmol  $F_{420}H_2$  oxidized· min<sup>-1</sup>·mg<sup>-1</sup> (Lineweaver Burk:  $K_{m \; phenazine} = 156 \; \mu$ M,  $V_{max} = 3.2$  nmol  $F_{420}H_2$  oxid·min<sup>-1</sup>·mg<sup>-1</sup>) (Figures 7.4 and 7.5). This value is lower than the  $K_m$  determined for the  $F_{420}H_2$  dehydrogenase of Ml. tindarius (cell-free extract) (266  $\mu$ M phenazine, Chapter 4).





**Figure 7.3**. Determination of  $Km_{F_{420H_2}}$  for the phenazine-dependent  $F_{420H_2}$  dehydrogenation activity in clarified cell-free extract of *Methanosphaera stadtmanae*. (A) Michaelis Menten kinetics plot. (B) Lineweaver Burk kinetics plot. Activity measured using 500 $\mu$ M phenazine.

Diphenyleneiodonium (DPI) chloride (1  $\mu$ M) was found to be a competitive inhibitor of the purified  $F_{420}H_2$  dehydrogenase activity of Ms. mazei Gö1(Brodersen et al. 1999). Our studies also show that the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of Ml. tindarius (in cell-free extract) is also inhibited in the presence of DPI, as are the respective activities of Ms. barkeri Fusaro and Msp. hungatei GP1 (Chapters 5 and 6).

The  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae is also inhibited by DPI, but the nature of inhibition is unclear. Addition of 5  $\mu$ M resulted in an increase in the apparent  $K_m$  from 138.0 to 170.0  $\mu$ M, but the apparent Vmax decreases from 3.0 to 1.2 nmol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> (Lineweaver Burk kinetics: Addition of 5  $\mu$ M resulted in an increase in the apparent  $K_m$  from 156.0 to 166.5  $\mu$ M phenazine, apparent Vmax decreases from 3.2 to 1.1 nmol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>) (Figures 7.4 and 7.5). This would appear to indicate non-competitive inhibition of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae.

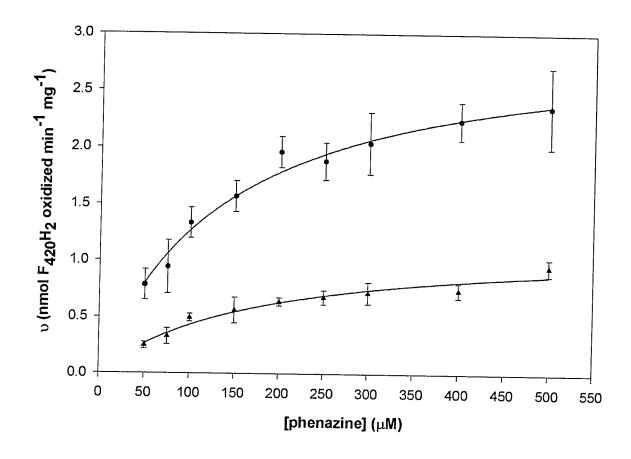


Figure 7.4. Michaelis Menten kinetics plot of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in clarified cell-free extract of *Methanosphaera stadtmanae*, in the presence/absence of DPI. All points represent experiments performed in triplicate. Assay conditions as described under 'Materials and Methods'. •, No DPI. •, 5  $\mu$ M DPI.

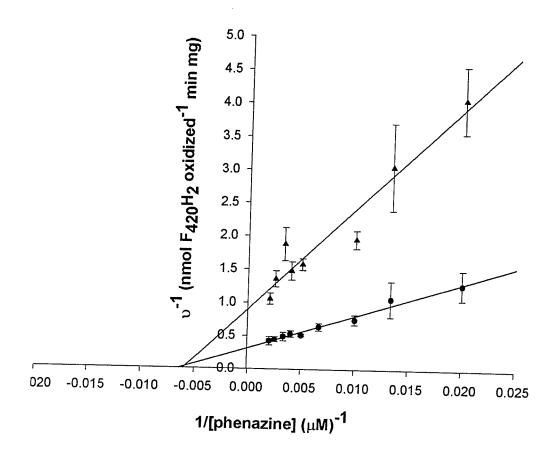


Figure 7.5. Lineweaver Burke kinetics plot of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in clarified cell-free extract of *Methanosphaera stadtmanae*, in the presence/absence of diphenyleneiodonium (DPI)chloride. All points represent experiments performed in triplicate. Assay conditions as described under 'Materials and Methods'.  $\bullet$ , No DPI.  $\blacktriangle$ , 5  $\mu$ M DPI.

## 7.3.5. Comparison of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity with other methanoarchaea

The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae*, in cell-free extract, is significantly lower when compared to the activities from a number of methanoarchaea, ranging from 30- to 140-fold difference in activity (Table 7.3). This is not surprising, since the amount of  $F_{420}$  found in *Msph. stadtmanae* is also low (0.16 nmol  $F_{420}$ ·mg<sup>-1</sup> protein, (Wong *et al.* 1994)), consistent with a role in biosynthetic pathways or  $O_2$ -detoxification.

The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae* is over 100-fold lower than the activity of *Ml. tindarius*, a methanoarchaeon containing the Fpo complex (Haase *et al.* 1992), and *Ms. barkeri* Fusaro (grown on  $CH_3OH$ ).

The activity from *Msph. stadtmanae* is also 65- to 100-fold lower than the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activities measured from other hydrogenotrophic methanoarchaea from the order *Methanobacteriales*, and from the orders *Methanococcales* and *Methanomicrobiales*.

Table 7.3. Comparison of the  $F_{420}H_2$  dehydrogenation ( $F_{420}H_2$  de $H_2$ ase) activity in cellfree extract of various methanoarchaea. [phenazine] = 500  $\mu$ M. [methyl viologen (mv)] = 500  $\mu$ M (note – 1 mM metronidazole added to assay tubes using methyl viologen as electron acceptor). [ $F_{420}$ ] = 10-15  $\mu$ M. Values are the average of triplicate assays. (Reprinted from Chapter 4).

Methanoarchaeon	Order	Growth substrate	$F_{420}H_2 \\ deH_2 ase^{\dagger} \\ (phenazine)$	$\begin{array}{c c} F_{420}H_2 \\ deH_2 ase^{\dagger} \\ (mv) \end{array}$
Ml. tindarius	Methanosarcinales	СН₃ОН	0.379	0.2
<i>Ms. barkeri</i> Fusaro	Methanosarcinales	СН₃ОН	0.421	0.048
		H <sub>2</sub> :CO <sub>2</sub>	0.225	0.019
		Acetate	0.116	0.047
Mst. concilli	Methanosarcinales	Acetate	0.087	0.042
Msp. hungatei GP1	Methanomicrobiales	H <sub>2</sub> :CO <sub>2</sub>	0.129	0.123
Mc. voltae	Methanococcales	H <sub>2</sub> :CO <sub>2</sub>	0.212	0.067
Mtb. marburgensis	Methanobacteriales	H <sub>2</sub> :CO <sub>2</sub>	0.270	0.070
Mb. bryantii	Methanobacteriales	H <sub>2</sub> :CO <sub>2</sub>	0.296	0.091
Msph. stadtmanae	Methanobacteriales	H <sub>2</sub> : CH <sub>3</sub> OH	0.002	0.005

 $<sup>^{\</sup>dagger}$  µmol  $F_{420}H_2$  oxidized·min $^{-1}$ ·mg $^{-1}$ 

# 7.3.6. Development of a protocol for the enrichment of the $F_{420}H_2$ dehydrogenation activity of Methanosphaera stadtmanae

Various column chromatography techniques were tested in order to enrich the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae*, some with very mitigated success. All steps were performed under anaerobic conditions, at ambient room temperature. The results of some of these steps are summarized in Appendix E.

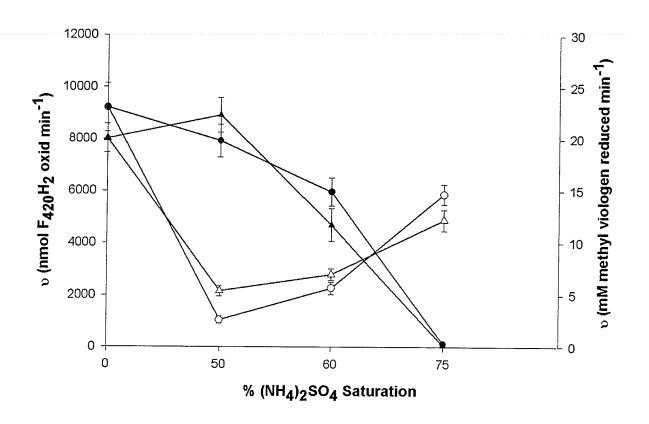
Since the activity is best preserved under high salt conditions after (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> precipitation, a strategy had to be developed that could be used under high salt conditions, at least during the initial enrichment step. A protocol utilizing several column chromatography techniques in tandem was devised to purify the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity. All steps were performed under anaerobic conditions at ambient room temperature. Ice gel packs were used to keep the protein cool when possible. Purification of the activity was not successful, but the end results appeared to be promising, suggesting that the potential for purification of this activity exists. The results are summarized in Table 7.4.

#### 7.3.6.1. Ammonium sulfate $((NH_4)_2SO_4)$ precipitation

 $F_{420}H_2$  dehydrogenation activity was enriched *via* 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> saturation of the clarified cell-free extract (Figure 7.6). A methyl viologen reducing hydrogenase activity was also enriched along with the  $F_{420}H_2$  dehydrogenation activity.

#### 7.3.6.2. Phenyl Sepharose CL-4B

Crystals of anaerobic (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> were added to the resuspended 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet to insure a final saturation of 60%; the concentration of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> remaining in



**Figure 7.6**. Ammonium sulfate precipitation profile of the  $F_{420}H_2$  dehydrogenation activity from clarified cell-free extract.

- $\bullet$ ,  $F_{420}H_2$  dehydrogenation activity of  $(NH_4)_2SO_4$  supernatant.
- o, F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet.
- $\triangle$ , methyl viologen-reducing hydrogenase activity of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> supernatant.  $\triangle$ , methyl viologen reducing hydrogenase activity of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet.

the protein pellet would be approximately 0.01 M. The protein solution was then diluted ½ with 60% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> saturated 50 mM HEPES, pH 7.0, containing 5% glycerol and 10 mM DTT (buffer A) and applied to a 45 mL Phenyl Sepharose CL-4B column equilibrated with buffer A; the protein and activity profile is shown in Figure 7.7A.

3 mL aliquots were collected in 4 inch glass tubes, using a Bio-Rad fraction collector. Filled tubes were removed from the fraction collector, sealed with rubber stoppers, and cooled on ice prior to removal from the anaerobic chamber for activity assay. Once the protein was fully loaded, the column was washed with two bed volumes of buffer A + 0.05% Triton-X-100, followed by a continuous 60%-0% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> gradient (3 bed volumes). At the end of the gradient, the column was washed with 2 bed volumes of 50 mM HEPES, pH 7.0, containing 5% glycerol, 0.05% Triton-X-100, and 10 mM DTT.

The fractions containing the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity were pooled and tested for activity. Two peaks of activity were observed; only the activity eluted in the absence of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (fractions 95-110) were processed further. When dialysis or gel filtration was employed to desalt the (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>—containing fractions the activity was lost upon desalting. Therefore, only the salt free fractions could be saved for further processing.

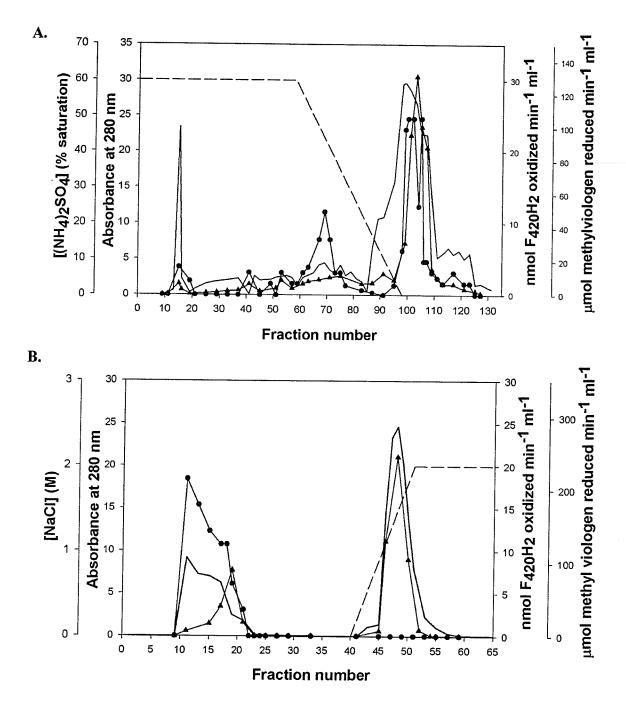


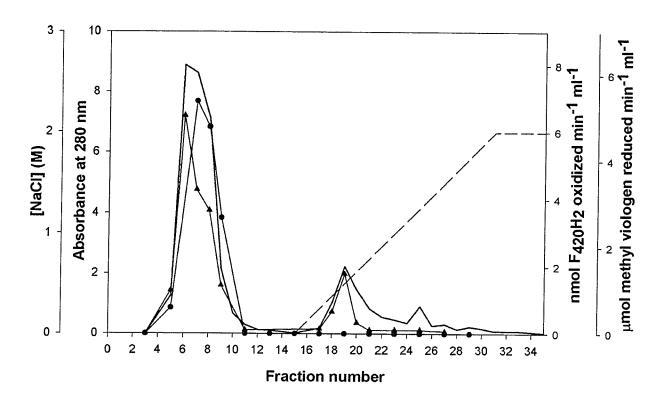
Figure 7.7. A. Protein and activity profile of fractions eluted from a Phenyl Sepharose CL 4B chromatography column. (—),  $A_{280}$ . (---) 60-0% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>. •,  $F_{420}H_2$  dehydrogenation activity. •, Methyl viologen reducing hydrogenase activity. B. Protein and activity profile of fractions eluted from a DEAE Sephacel chromatography column. (—),  $A_{280}$ . (---) 0-2 M NaCl. •,  $F_{420}H_2$  dehydrogenation activity. •, Methyl viologen reducing hydrogenase activity. Conditions for elution as described in text.  $F_{420}H_2$  dehydrogenation activity assayed using 500  $\mu$ M phenazine.

#### 7.3.6.3. DEAE Sephacel

The salt-free eluted protein collected from the Phenyl Sepharose CL-4B column was then applied to DEAE Sephacel column equilibrated with 50 mM HEPES, pH 7.0, containing 5% glycerol, 0.05% Triton-X-100, and 10 mM DTT (buffer B). The column was then washed with 2 bed volumes of buffer B before a gradient of 0-2 M NaCl was applied (3 bed volumes). The elution and activity profile of the DEAE Sephacel column is shown in Figure 7.7B. Only one peak of F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was observed; the F<sub>420</sub>H<sub>2</sub> dehydrogenation did not have an affinity for the DEAE Sephacel and passed through the column without binding to the matrix (fractions 15-25).

#### 7.3.6.4. $F_{420}$ -Affinity Column

The  $F_{420}$ -affinity chromatography steps were completed in diffuse light (i.e. ambient light only) to prevent the bound  $F_{420}$  from photolysing. The protein was loaded onto  $F_{420}$ -bound EAH sepharose (8 mL) equilibrated with buffer B (50 mM HEPES, pH 7.0, with 5% glycerol, 0.05% Triton-X-100, and 10 mM DTT). After the protein was loaded onto the column, the column was washed with 2 bed volumes of buffer B, followed by a continuous 0-2 M NaCl gradient (made using buffer B). The protein and activity profile is shown in Figure 7.8. The  $F_{420}H_2$  dehydrogenation activity did not appear to have any affinity for the bound  $F_{420}$  and eluted from the column with the buffer front. The fractions containing  $F_{420}H_2$  dehydrogenation activity (fractions 5-11) were collected, pooled and tested for activity. Surprisingly, a significant  $H_2$ -dependent methyl viologen reducing activity devoid of  $F_{420}H_2$  dehydrogenation activity was eluted from the



**Figure 7.8**. Protein and activity profile of fractions collected from an  $F_{420}$ -affinity column. (—),  $A_{280}$ . (---), linear 0-2 M NaCl.. •,  $F_{420}H_2$  dehydrogenation activity. •, Methyl viologen reducing hydrogenase activity.  $F_{420}H_2$  dehydrogenation activity assayed with 500  $\mu$ M phenazine.

column during the 0-2 M NaCl gradient – no further analyses were performed on these fractions.

A summary of the attempted purification of the  $F_{420}H_2$  dehydrogenation is shown in Table 7.4. The highest degree of purification occurred at the Phenyl-Sepharose CL-4B stage. Further purification attempts beyond this stage did not yield satisfactory results, as the stability of the  $F_{420}H_2$  dehydrogenation activity decreased quickly as the purification process progressed. This is in contrast to the cell-free extract (stored at 4°C), which was stable throughout the length of the enrichment procedure. As the  $F_{420}H_2$  dehydrogenation activity is  $O_2$  labile, the chromatography steps were performed in the anaerobic chamber,

Table 7.4. Summary of the enrichment process for the  $F_{420}H_2$  dehydrogenation activity of *Methanosphaera stadtmanae*. Process was repeated three times, with representative data presented here.  $[F_{420}H_2] = 10$ -15  $\mu$ M. [phenazine] = 500  $\mu$ M.

Stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Specific activity <sup>‡</sup>	Purification (-fold)
CCFE	929.3	3235.2	100	3.5	-
75% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> pellet	583.2	4125.6	127.5	7.1	2.0
Phenyl-Sepharose CL-4B	20.4	456.7	14.1	22.1	6.3
DEAE-Sephacel <sup>††</sup>	6.1	86.5	2.7	14.3	4.1
$F_{420}$ -affinity $^{\dagger\dagger}$	2.1	14.3	0.4	6.8	1.9

 $<sup>^{\</sup>dagger}$ nmol  $F_{420}H_2$  oxidized·min $^{-1}$ 

<sup>†</sup> nmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>·mg<sup>-1</sup>

<sup>††</sup> activity measured from pooled non-binding fractions

at room temperature, but the protein extracts were kept cool using cold gel packs, and fractions were kept on ice prior to removal from the anaerobic chamber; pooled extracts were refrigerated between chromatographic steps, such that temperature should not have been an issue.

A coomassie blue- and silver-stained linear 0-20% native PAGE gel is shown in Figure 9.9. As the protein passes through the various column chromatography steps it is evident that certain protein bands are being enriched, while others are lost, indicating that the  $F_{420}H_2$  dehydrogenation activity is being enriched.

## 7.3.7. Reactivation of the $F_{420}H_2$ Dehydrogenation activity of Methanosphaera stadtmanae

As the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae goes through the various chromatographic steps, the activity yield at each stage drops, ultimately to a point where further chromatographic steps/analyses were not feasible. Furthermore, the  $F_{420}H_2$  dehydrogenation activity was inactive after 3-4 days storage at 4°C, under reduced, anaerobic conditions. Several assays were performed to determine if the  $F_{420}H_2$  dehydrogenation activity could be restored after apparent loss of this activity. The protein was mixed with one of the following:  $(NH_4)_2SO_4$  (60% saturation); 200 mM KPO<sub>4</sub>  $F_{420}H_2$  dehydrogenation assay buffer; 20  $\mu$ M FAD.

The results are summarized in Table 7.5. The addition of solid  $(NH_4)_2SO_4$  crystals to protein extract to yield 60% saturation did not restore  $F_{420}H_2$  dehydrogenation activity. While  $(NH_4)_2SO_4$  is apparently able to maintain/enhance the  $F_{420}H_2$  dehydrogenation activity (in cell-free extract), it cannot reactivate the activity after it is lost. The addition of KPO<sub>4</sub> buffer (200 mM KPO<sub>4</sub>, with 0.5 M sucrose, 10 mM DTT, pH

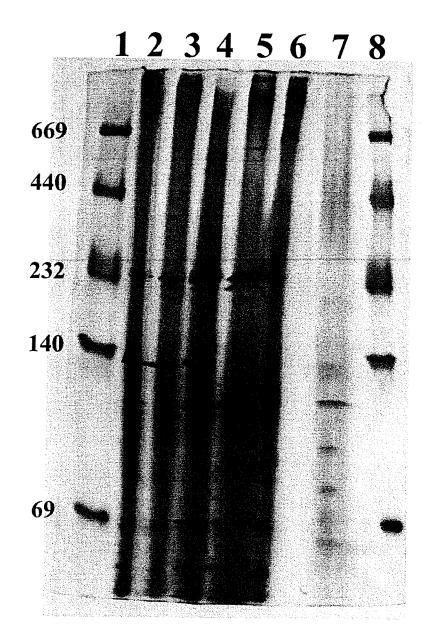


Figure 7.9. Double-stained 3-20% native gradient PAGE detailing the enrichment of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity from the clarified cell-free extract of *Methanosphaera stadtmanae.* 1, 8. Molecular markers, kDa. 2. Clarified cell-free extract. 3. re-suspended 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet. 4. Phenyl Sepharose CL-4B pool. 5. DEAE Sephacel pool 1. 6. DEAE Sephacel pool 2. 7. F<sub>420</sub>-affinity column pool.

7.0, final concentration), the standard buffer system used to assay the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae*, did not have any noticeable effect on the inactive  $F_{420}H_2$  dehydrogenation activity.

Since FAD appears to be essential for the reactivation of the  $F_{420}$  H<sub>2</sub>ase activity of Msph. stadtmanae, the use of this compound is a logical step for reactivation of the  $F_{420}$ H<sub>2</sub> dehydrogenation activity, which would also require FAD. Incubation of the protein with 20  $\mu$ M FAD for 3-4 hours at 4°C lead to reactivation of the phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activity. Moreover, in the presence of 20  $\mu$ M FAD, the phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activity increased almost 3-fold (2.6x) higher as compared to the initial activity of the non reactivated protein.

Table 7.5. Reactivation of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity under different conditions. Protein reactivated approximately 3 days after initial measurement after collection from  $F_{420}$ -affinity column.

Incubation condition (all at 4°C)	Activity (nmol F <sub>420</sub> H <sub>2</sub> oxidized min <sup>-1</sup> mg <sup>-1</sup> )
Initial activity (no addition)	$6.8 \pm 0.8$
No addition (3 days later)	No activity detected
+ 20 μM FAD	$19.0 \pm 0.1$
+ 200 mM KPO <sub>4</sub> , 0.5 mM sucrose, 10 mM DTT	No activity detected
+ 60% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	No activity detected

#### 7.3.8. Enrichment of an $H_2$ -dependent $F_{420}$ -reducing activity

Early attempts to detect H<sub>2</sub>-dependent F<sub>420</sub>-reduction in cell-free extracts of *Msph. stadtmanae* indicated that this activity appeared to be absent (< 0.01 μmol F<sub>420</sub> reduced·min<sup>-1</sup>·mg<sup>-1</sup>), although a methyl viologen-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was present in this methanoarchaeon (Wong *et al.* 1994, Wong, Ph.D thesis 1999). With the discovery of methanophenazine as an intermediate in F<sub>420</sub> recycling in *Ms. mazei* Gö1 (Abken *et al.* 1998), it was believed that a similar compound might be present in *Msph. stadtmanae*. Indeed, F<sub>420</sub>-reduction with H<sub>2</sub> was observed when phenazine was included in the reaction buffer: a 2 step mechanism for F<sub>420</sub> reduction was suggested, involving (1) H<sub>2</sub>-dependent reduction of an unknown intermediate, followed by (2) F<sub>420</sub>-dependent oxidation of this intermediate (Wong, Ph.D thesis 1999). As such, it had been proposed (Wong, Ph.D thesis 1999) that the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity observed in the cell-free extract of *Msph. stadtmanae* may be operating in the reverse direction *in vivo*, with F<sub>420</sub> being reduced using electrons

Table 7.6.  $H_2$ -dependent  $F_{420}$ -reducing activity using cell-free extract of Msph. stadtmanae. Activity assayed using 200 mM KPO<sub>4</sub>, 0.5 M sucrose, 10 mM DTT, under  $H_2$  (except where noted).  $[F_{420}] = 20 \, \mu M$ . Note – results from one assay, in triplicate.

Mediator	μmol F <sub>420</sub> reduced·min <sup>-1</sup> ·mg <sup>-1</sup>
none	$0.003 \pm 0.0001$
none (under N <sub>2</sub> )	0.0004
Phenazine (10 μM)	$0.003 \pm 0.0001$
Ferredoxin (25 μg/ml)	$0.002 \pm 0.001$

extracted from an alternate electron donor, potentially reduced-methanophenazine, via the  $F_{420}H_2$  dehydrogenation activity (Wong, Ph.D thesis 1999).

 $H_2$ -dependent  $F_{420}$ -reduction in cell-free extract, using redox mediators such as phenazine and ferredoxin, was also studied simultaneously with the ongoing enrichment experiments of the  $F_{420}H_2$  dehydrogenation activity. These assays indicated that  $H_2$ -dependent  $F_{420}$ -reducing activity was not dependent on the presence of phenazine or ferredoxin (Table 7.6). Subsequent assays demonstrated a requirement for incubation under reducing conditions, with the inclusion of FAD, for optimal activity (Chapter 8).

The  $F_{420}$  H<sub>2</sub>ases from  $Msp.\ hungatei$  GP1 (Chapter 5) and  $Ms.\ barkeri$  Fusaro (Chapter 6) were the sole source of a phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activity in the respective methanoarchaeon. In order to elucidate whether the  $F_{420}$ H<sub>2</sub> dehydrogenation activity of  $Msph.\ stadtmanae$  might also be associated with its putative  $F_{420}$  H<sub>2</sub>ase,  $F_{420}$ -reducing activity of was also monitored after each chromatographic step. The  $F_{420}$ -reducing activity (Table 7.7) follows a similar pattern of purification comparable to the  $F_{420}$ H<sub>2</sub> dehydrogenation (Table 7.4), indicating possible association of the phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activity with the  $F_{420}$  H<sub>2</sub>ase activity.

Table 7.7. Summary of the enrichment process for the  $F_{420}$ -reducing hydrogenase activity of Methanosphaera stadtmanae.  $F_{420}$   $H_2$ ase activated with 10  $\mu M$  FAD and 10  $\mu$ M F<sub>420</sub> prior to assay. [F<sub>420</sub>] = 25-30  $\mu$ M.

Stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Specific <sup>‡</sup> activity	Purification (-fold)
CCFE	929.2	193.9	100	0.21 <sup>‡‡</sup>	-
75% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> pellet	583.2	235.2	121.3	0.40	1.9
Phenyl-Sepharose	20.4	41.4	21.4	2.03	9.7
<b>DEAE</b> Sephacel <sup>††</sup>	6.1	13.5	7.0	2.21	10.6
F <sub>420</sub> -affinity <sup>††</sup>	2.1	2.1	1.1	1.00	4.8

 $F_{420}H_2$  as activity of CCFE = 0.005  $\mu$ mol  $F_{420}$  reduced·min<sup>-1</sup>·mg<sup>-1</sup>

#### 7.3.9. $F_{420}H_2$ oxidase – source of a phenazine-dependent $F_{420}H_2$ dehydrogenation activity?

The recently published genome of Msph. stadtmanae indicates the presence of a putative F<sub>420</sub>H<sub>2</sub> oxidase (FprA), which catalyzes the F<sub>420</sub>H<sub>2</sub>-dependent reduction of O<sub>2</sub> (Seedorf et al. 2004, Fricke et al. 2006). As seen in Table 7.3 (section 7.3.1.), O2 is able to substitute for phenazine as substrate for the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity. This activity is primarily located in the soluble cytoplasmic fraction (i.e. clarified cell-free extract, CCFE) (>95% of the activity), with only 2-5% of the activity measured using

 $<sup>^{\</sup>dagger}\mu mol~F_{420}~reduced \cdot min^{-1}$   $^{\ddagger}\mu mol~F_{420}~reduced \cdot min^{-1} \cdot mg^{-1}$   $^{\dagger\dagger}$  activity measured from pooled non-binding fractions

<sup>&</sup>lt;sup>‡‡</sup>in the absence of FAD,

washed membranes (Table 7.8). This is similar to the  $F_{420}H_2$  oxidase isolated from *Methanobrevibacter arboriphilus*, which is a cytoplasmic protein (Seedorf *et al.* 2004). Similarly, 85-90% of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity is detected in the CCFE (10-15% from the washed membranes); this is comparable to the level of  $F_{420}$   $H_2$  as activity detected in *Msph. stadtmanae* (Table 7.8).

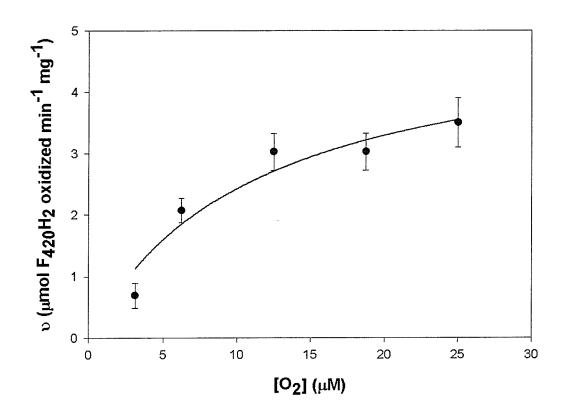
Table 7.8. Comparison of the phenazine- and  $O_2$ -dependent  $F_{420}H_2$  dehydrogenation, and  $H_2$ -dependent  $F_{420}$ -reducing hydrogenase activities of *Methanosphaera stadtmanae*. Activities measured from cell-free extract, clarified cell-free extract, and washed membranes.  $F_{420}$   $H_2$ ase activated with 10  $\mu$ M FAD and 10  $\mu$ M  $F_{420}$  prior to assay. 18-20  $\mu$ M  $F_{420}$  added to  $F_{420}H_2$  de $H_2$ ase or  $F_{420}H_2$  oxidase assay tubes.

Fraction	$F_{420}H_2 deH_2 ase^{\dagger}$ [phenazine] = 500 $\mu M^{\dagger}$	$F_{420}H_2$ oxidase $[O_2] = 25 \mu M^{\dagger}$	$F_{420} H_2 ase^{\ddagger}$ $[F_{420}] = 20-25 \mu M$
Cell-free extract	$1.4 \pm 0.2$	$1.1 \pm 0.2$	$93.6 \pm 9.5$
Clarified cell-free extract	$1.2 \pm 0.1$	$1.1 \pm 0.1$	$81.5 \pm 3.3$
Washed membranes	$0.2 \pm 0.02$	$0.02\pm0.005$	$9.2 \pm 2.2$

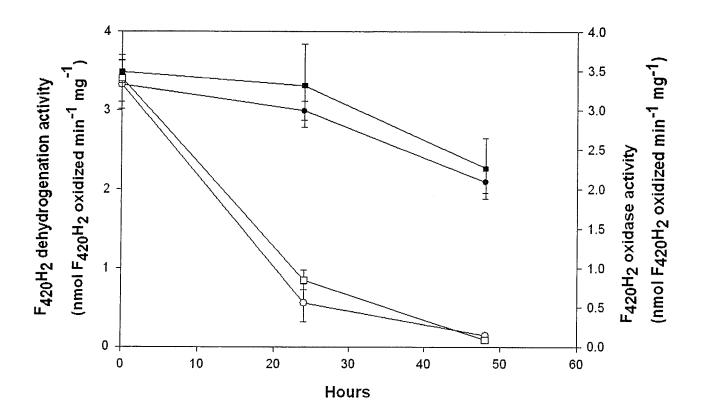
<sup>†</sup> nmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>

The apparent  $K_m$  for  $O_2$  was determined to be 11  $\mu$ M  $O_2$ , with  $V_{max} = 5.1$  nmol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup> (Figure 7.10).  $O_2$ -dependent  $F_{420}H_2$  oxidation was observed with the addition of clarified cell-free extract; this activity was not observed in the absence of added protein, indicating that the activity is enzymatic in nature.  $O_2$ -dependent  $F_{420}H_2$  oxidation activity is aerobically unstable, with only 23 % activity remaining after 24 hours storage under aerobic conditions at 4°C, and 2.3% activity

 $<sup>^{\</sup>ddagger}$  µmol F<sub>420</sub> reduced·min<sup>-1</sup>; O<sub>2</sub> added as aliquot from air-saturated H<sub>2</sub>0, ~250 µM



**Figure 7.10**. Michaelis Menten kinetics plot for the  $F_{420}H_2$  oxidase activity in clarified cell-free extract. O<sub>2</sub> added to anaerobic assay tubes as air-saturated MilliQ H<sub>2</sub>O.  $[F_{420}H_2] = 18-20 \ \mu\text{M}$ .



**Figure 7.11.** Comparison of the stability of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity and the  $F_{420}H_2$  oxidase activity in the cell-free extract of *Methanosphaera stadtmanae*. Activities were assayed over a 48 hour period. Aerobically stored extract purged with  $N_2$  prior to assay.  $[F_{420}H_2] = 15 \ \mu\text{M}$ . [phenazine] = 500  $\mu\text{M}$ .  $[O_2] = 18.75 \ \mu\text{M}$ •,  $F_{420}H_2$  dehydrogenation activity (anaerobic).  $\bigcirc$ ,  $F_{420}H_2$  dehydrogenation activity (aerobic).  $\square$ ,  $F_{420}H_2$  oxidase activity (aerobic).

detected after 48 hours. The decrease in  $F_{420}H_2$  oxidase activity is comparable to the loss of activity observed with the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity when stored under aerobic conditions; only 17% of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was remaining after 24 hours, and 4% of the activity after 48 hours (Figure 7.11).

#### 7.4. Discussion

#### 7.4.1. $F_{420}H_2$ dehydrogenation activity in Methanosphaera stadtmanae

In methylotrophic members of the *Methanosarcinales*, such as *Ms. mazei* Gö1, the  $F_{420}H_2$  dehydrogenase (Fpo complex,  $F_{420}H_2$  phenazine oxidoreductase), in tandem with the heterodisulfide reductase (Hdr) is an essential component of the  $F_{420}H_2$ : heterodisulfide oxidoreductase complex, catalyzing the reduction of CoM-S-S-CoB and generating an energy conserving transmembrane proton gradient (Deppenmeier 2002, 2004). Methanophenazine is believed to be the physiological electron acceptor for the  $F_{420}H_2$  dehydrogenation activity; this compound has not been isolated in methanoarchaea.

A similar activity is observed in the cell free extract of *Msph. stadtmanae*, a member of the *Methanobacteriales* that grows solely on H<sub>2</sub> and CH<sub>3</sub>OH (Miller and Wolin 1983, 1985). However, the Fpo complex is not encoded in the recently published genome of *Msph. stadtmanae*, nor does it appear to be essential for methanogenesis on H<sub>2</sub> and CH<sub>3</sub>OH as *Msph. stadtmanae* possesses an MV H<sub>2</sub>ase (Mvh) (Fricke *et al.* 2006). For most hydrogenotrophic methanoarchaea, the MV H<sub>2</sub>ase is part of the H<sub>2</sub>:CoM-S-S-CoB oxidoreductase complex, catalyzing the H<sub>2</sub>-dependent reduction of CoM-S-S-CoB

(Thauer et al. 1993, Shima et al. 2002). In the absence of the Fpo complex, what is the source of the  $F_{420}H_2$  dehydrogenation activity in Msph. stadtmanae?

The recently published genome of Msph. stadtmanae indicates the presence of several different putative genes encoding for  $F_{420}H_2$ -dependent enzymes. These include methylene- $H_4MPT$  dehydrogenase and methylene- $H_4MPT$  reductase, the  $F_{420}H_2$ :NADP oxidoreductase (Fno), and a putative  $F_{420}H_2$  oxidase (FprA) (Fricke et~al.~2006). The genome also indicates the presence of genes encoding for a putative  $F_{420}$   $H_2$ ase; our studies have demonstrated that the purified  $F_{420}$   $H_2$ ase from various methanoarchaea possess phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. A natural question to ask would be whether the observed phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in Msph.~stadtmanae is associated with any of the above mentioned proteins as potential candidates for this activity, or if it is a distinct protein on its own?

Methylene- $H_4$ MPT dehydrogenase catalyzes the reversible reduction of  $N^5$ , $N^{I0}$ -methylene- $H_4$ MPT, while methylene- $H_4$ MPT reductase catalyzes the reduction of  $N^5$ , $N^{I0}$ -methylene- $H_4$ MPT to  $N^5$ -methyl- $H_4$ MPT (Shima *et al.* 2002). Both reactions proceed *via* stereospecific hydride transfer, with a ternary complex kinetic mechanism (Ma and Thauer 1990, Thauer *et al.* 1993). As such, these enzymes cannot be the source of the  $F_{420}H_2$  dehydrogenation activity, since reduction of methyl viologen and phenazine both occur *via* single electron transfers.

The  $F_{420}H_2$ :NADP oxidoreductase (Fno) was purified and characterized by Elias et al. (2000); this enzyme catalyzes the reversible  $F_{420}H_2$ -dependent reduction of NADP to NADPH. Several properties appear to rule out this protein as a source of the  $F_{420}H_2$  dehydrogenation activity. The Fno is aerobically stable, whereas the  $F_{420}H_2$ 

dehydrogenation activity rapidly diminishes under aerobic conditions. The bulk of the Fno activity precipitates at 95% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> saturation, while the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity precipitates at 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (Wong *et al.* 1994, this thesis). Fno reduces NADP<sup>+</sup> with F<sub>420</sub>H<sub>2</sub> using a ternary sequential reaction, *via* direct hydride transfer (Warkentin *et al.* 2001); the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity catalyzes the reduction of substrates such as methyl viologen or phenazine *via* single e<sup>-</sup> transfers (Wong *et al.* 1994, this thesis).

The putative proteins  $F_{420}H_2$  oxidase and  $F_{420}$   $H_2$  as are both potential candidates for sources of phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. Both are flavoproteins:  $F_{420}H_2$  oxidase contains FMN, while the  $F_{420}$   $H_2$  as contains FAD.  $F_{420}H_2$  is a hydride donor, such that 2 electrons are transferred to the protein in a single step (Jacobson and Walsh 1984). Reduction of phenazine occurs in two single electron transfer steps (Abken *et al.* 1998), such that a 2 e<sup>-</sup>/1 e<sup>-</sup> switch is necessary for the transit of electrons from  $F_{420}H_2$  through the protein, and then to phenazine. Finally, both enzymes can use the low potential electron donor coenzyme  $F_{420}H_2$  to reduce higher potential substrates such as  $O_2$  or phenazine.

### 7.4.2. Electron acceptor specificity and kinetics studies of the $F_{420}H_2$ dehydrogenation activity

NADH (E°'= -0.320 V) could not substitute for  $F_{420}H_2$  (E°'= -0.360 V) as electron donor for the  $F_{420}H_2$  dehydrogenation activity, although both are hydride donors with similar midpoint potentials. Several substrates were able to substitute for phenazine (E°'= -0.265 to -0.165 V) as electron acceptor for the  $F_{420}H_2$  dehydrogenation activity, including FAD (E°' = -0.219 V), methyl viologen (E°'= -0.446 V), and  $O_2$  (added as

breathing grade air-saturated  $H_2O$ ,  $E^{\circ \prime}=+1.229~V$ ) (Table 7.2). The use of  $O_2$  as electron acceptor is particularly interesting as a putative  $F_{420}H_2$  oxidase is encoded by the genome of Msph. stadtmanae (Fricke et~al. 2006). The presence of  $F_{420}H_2$ -dependent  $O_2$ -reduction is not necessarily indicative of the presence of an  $F_{420}H_2$  oxidase; the  $F_{420}$   $H_2$  ase of Mtb.  $thermoautotrophicus~\Delta H$  also catalyzes the  $F_{420}H_2$ -dependent reduction of  $O_2$  to  $H_2O_2$  (Jacobson et~al. 1982). Neither NAD<sup>+</sup> nor ferredoxin could replace phenazine as electron acceptor.

The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity, in cell-free extract, of *Msph. stadtmanae* appears to follow simple Michaels Menten kinetics, similar to what has been reported for *Ms. mazei* Gö1 (Abken *et al.* 1998), and for other methanoarchaea possessing this activity (Chapter 4). The  $K_{mphenazine}$  for the  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae* was 138  $\mu$ M, indicating a higher affinity for phenazine relative to the *Ml. tindarius* (266  $\mu$ M, cell-free extract) and *Ms. mazei* Gö1 (250  $\mu$ M, purified enzyme, Brodersen *et al.* 1999)  $F_{420}H_2$  dehydrogenase activities. By comparison, the  $K_{mphenazine}$  values determined for the purified phenazine-dependent  $F_{420}H_2$  dehydrogenation activity (associated with the  $F_{420}$   $H_2$ ase) in *Msp. hungatei* GP1 and *Ms. barkeri* Fusaro were determined to be approximately 290  $\mu$ M and 95  $\mu$ M, respectively (Chapters 5 and 6).

DPI has been shown to be a competitive inhibitor for the purified  $F_{420}H_2$  dehydrogenase of Ms. mazei Gö1, with low concentrations of DPI (1  $\mu$ M) increasing the apparent  $K_m$  from 35 to 100  $\mu$ M 2-OH-phenazine (Brodersen et al. 1999). In contrast, DPI does not appear to act as a competitive inhibitor of the  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae. While there is a small increase in the apparent  $K_m$  for

phenazine upon addition of 5  $\mu$ M DPI (138  $\mu$ M  $\rightarrow$  170  $\mu$ M), the apparent Vmax decreases from 3.0 to 1.2 nmol  $F_{420}H_2$  oxidized·min<sup>-1</sup>·mg<sup>-1</sup>, indicating noncompetitive inhibition. This is likely due to DPI binding to the FAD component of the protein bearing  $F_{420}H_2$  dehydrogenation activity; the  $F_{420}$   $H_2$  as activity is also inhibited when 5  $\mu$ M DPI is included in the reaction mixture, which would suggest noncompetitive inhibition (see Chapter 8).

Another explanation for the observed non-competitive inhibition may have to do with the presence of two sources of phenazine-dependent  $F_{420}H_2$  dehydrogenation activities in the cell-free extract: the  $F_{420}$   $H_2$  ase and  $F_{420}H_2$  oxidase. One of the activities may be inhibited more strongly by the addition of DPI, likely the  $F_{420}$   $H_2$  ase, leaving the  $F_{420}H_2$  oxidase as the major source of  $F_{420}H_2$  dehydrogenation activity remaining. Initial studies indicated that addition of DPI greater than  $10~\mu M$  did not have further effect on the  $F_{420}H_2$  dehydrogenation activity, which remained constant ( $\sim 0.6~\text{nmol}\ F_{420}H_2$  oxidized min<sup>-1</sup>·mg<sup>-1</sup>) without a significant decrease in activity as the [DPI] increases up to  $40~\mu M$  (Figure 6A, Chapter 4). This would indicate that one of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activities in the cell-free extract is less susceptible to inhibition by DPI.

# 7.4.3. Attempts at purification of the $F_{420}H_2$ dehydrogenation activity of Methanosphaera stadtmanae

The strategies used to enrich for the  $F_{420}H_2$  dehydrogenation activity from Msp. hungatei GP1 and Ms. barkeri Fusaro could not be applied here, for several reasons. The  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae is aerobically unstable, in contrast to the respective activity of Msp. hungatei GP1 and Ms. barkeri Fusaro, and must be handled under anaerobic conditions at all times. The protocols used in Msp. hungatei GP1 and Ms. barkeri Fusaro either did not enrich for the dehydrogenation activity of Msph. stadtmanae, or destroyed it. This resulted in a "trial and error" approach to the purification, using chromatographic matrices with different binding properties.

The majority of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae* is located in the soluble cytoplasmic fraction (85-90%). To get an initial enrichment of the activity, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, to 75% saturation, was used to precipitate the activity. Removal of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> from the resuspended protein pellet, *via* gel filtration or dialysis, resulted in loss of activity up to 60-75% (or greater). However, in the absence of a desalting step, the resuspended protein pellet (~0.01 M (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>) did not have affinity for matrices requiring low salt concentrations for binding. At this point it was clearly beneficial to find a chromatographic technique that could take advantage of the high salt content.

Enrichment of the  $F_{420}H_2$  dehydrogenation activity was observed with the hydrophobic matrix Phenyl Sepharose CL-4B. The use of Phenyl Sepharose CL-4B makes sense as it makes use of the high salt content of the homogenized 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet, which contains the bulk of the  $F_{420}H_2$  dehydrogenation activity. Interestingly, there were two separate peaks of  $F_{420}H_2$  dehydrogenation activity, with one peak eluting within the linear 60-0% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> gradient, and the other displaced when the column was washed with (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>-free buffer, which desalted the protein and allowed for the use of chromatographic matrices that require low salt conditions. Once it became available, the genome of *Msph. stadtmanae* indicated that there were potentially two proteins that could be the source of the observed phenazine-dependent  $F_{420}H_2$ 

dehydrogenation activity: the  $F_{420}$   $H_2$  as and the  $F_{420}H_2$  oxidase (Fricke *et al.* 2006). Thus, the observations of two peaks of dehydrogenation activity are consistent with the findings in the genome of *Msph. stadtmanae*.

Initial chromatography results using DEAE Sephacel loaded with cell-free extract of *Msph. stadtmanae* also indicated the presence of 2 distinct F<sub>420</sub>H<sub>2</sub> dehydrogenation activity peaks, one with no apparent affinity for the DEAE Sephacel matrix, and another that elutes within the 0-2 M NaCl gradient (Figure 3A, Appendix E). When the pooled fractions collected from the Phenyl Sepharose CL-4B column are loaded onto the DEAE Sephacel column (Figure 7.7B), only one activity peak is observed, the non-binding activity. This particular F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae* did not have an affinity for the DEAE matrix material, but other extraneous proteins did bind, thereby reducing the protein content of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity eluate.

The  $F_{420}H_2$  dehydrogenation activity is enriched after elution from the Phenyl Sepharose column, but the specific activity decreases after passing through the DEAE Sephacel column. An  $F_{420}$  H<sub>2</sub>ase activity is also enriched in the pooled fractions collected from the Phenyl Sepharose CL-4B and DEAE Sephacel chromatography steps (Table 7.7). Taken together, the results indicate an enrichment of the  $F_{420}$  H<sub>2</sub>ase, accounting for one of the observed phenazine-dependent  $F_{420}H_2$  dehydrogenation activities. Further discussion of the  $F_{420}$  H<sub>2</sub>ase continues in Chapter 8.

The pooled fractions collected from the DEAE Sephacel column were then loaded onto an  $F_{420}$ -affinity column (Figure 8), the results of which are rather puzzling. The inability of the  $F_{420}H_2$  dehydrogenation activity to bind to the coenzyme  $F_{420}$ -affinity column came as an unexpected surprise, since this column was expected to be one of the

more potent steps in the purification; this step had been used in the purification of the  $F_{420}H_2$  dehydrogenation activity of Ml. tindarius by Haase et al. (1992). One possible explanation for the apparent non-affinity of the  $F_{420}H_2$  dehydrogenation activity for  $F_{420}$  may be that the protein does not have an affinity for oxidized  $F_{420}$ , since the primary substrate would be the reduced form,  $F_{420}H_2$ . Since the oxidized form would in effect be a product of the reaction, the protein would not be expected to have a high affinity for  $F_{420}$  relative to  $F_{420}H_2$ . On the other hand, if the  $F_{420}$   $H_2$  ase is the source of the  $F_{420}H_2$  dehydrogenation activity, it would be expected to bind to the  $F_{420}$ -affinity column, yet the  $F_{420}$   $H_2$  as activity is also collected in the non-binding fractions.

While the  $F_{420}H_2$  dehydrogenation activity did not bind to the  $F_{420}$ -bound matrix, a portion of the MV  $H_2$ ase activity did have apparent affinity for the matrix. Interestingly, passage of the protein through the column led to the reduction of the  $F_{420}$  bound on the matrix; the MV  $H_2$ ase does not use  $F_{420}$  as substrate. The eluted protein did not possess  $F_{420}H_2$  dehydrogenation or  $F_{420}$   $H_2$ ase activity. It is possible that by the time the protein was collected from the column it emerged in a destabilized form that could not be reactivated in the presence of FAD or  $F_{420}$ . During the enrichment process, extended exposure of the protein extract to the reducing conditions in the anaerobic chamber (filled with 10%  $H_2$ , balance  $N_2$ ) may have lead to destabilization of the protein. In contrast, the unprocessed cell-free extract still retains greater than 70% activity after 48 hours (Figures 7.2 and 7.11). It has been suggested that storage of the  $F_{420}$   $H_2$ ase under reduced, anaerobic conditions may lead to the reduction of FAD to FADH2, which may be less tightly bound and lost from the holoprotein (Fox *et al.* 1987). Sprott *et al.* (1987) found that the purified  $F_{420}$   $H_2$ ase activity of *Msp. hungatei* GP1 decreased sharply

after 2 hours incubation under reducing conditions. Fiebig and Friedrich (1989) reported that when the F<sub>420</sub> H<sub>2</sub>ase of *Ms. barkeri* Fusaro was maintained under reducing conditions, the activity was unstable with 15% loss of activity after 50 minutes, and could not be stablilized.

The results are also similar to the findings of Schink and Probst (1980), who reported that storage of solubilized or purified hydrogenase from *Alcaligenes eutrophus* under  $O_2$ -free  $H_2$  or  $N_2$ , or prolonged exposure to reducing agents, such as  $\beta$ -mercaptoethanol or DTT, leads to rapid and complete loss of activity. The  $F_{420}$ -dependent formate dehydrogenase of *Mb. formicicum* is a FAD-dependent enzyme which is susceptible to loss of the FAD component when stored with reducing agents such as dithionite; incubation of the protein with excess FAD restores activity (Schauer and Ferry 1985).

The purification process was not successful, but the native gradient PAGE (Figure 7.11) indicates that there is potential for Enrichment of this activity. Purification of the  $F_{420}H_2$  dehydrogenation activity was greatly hampered by the instability of the protein(s) through the various stages.

#### 7.4.4. Reactivation of the $F_{420}H_2$ dehydrogenation activity

A significant problem with respect to the enrichment of the  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae was the instability of the activity during the various column chromatography steps. While the unprocessed cell-free extract was stable for at least two weeks, or in  $(NH_4)_2SO_4$ , the activity was rapidly lost during the process of enrichment through the various column chromatography steps. Several additives were tested to try and stabilize/restore the  $F_{420}H_2$  dehydrogenation activity.

Reactivation of the  $F_{420}H_2$  dehydrogenation activity *via* the addition of crystals of  $(NH_4)_2SO_4$  to 60% saturation appeared to be plausible, since the activity of the  $(NH_4)_2SO_4$  precipitated (75% saturation)  $F_{420}H_2$  dehydrogenation activity appeared to be enhanced on a consistent basis (from 130-150%) compared to the activity of the cell-free extract from which it was precipitated (Table 7.4); the activity of the resuspended protein pellet was also stable for 2 weeks or more. This is comparable to observations of the activity of the Fno, which was enhanced and stable after  $(NH_4)_2SO_4$  precipitation (Wong *et al.* 1994). Addition of  $(NH_4)_2SO_4$  to 60 or 75% saturation did not restore activity to the inactive  $F_{420}H_2$  dehydrogenation activity to any degree.

High concentrations of KPO<sub>4</sub> (200 mM) have been shown to increase the stability and activity of the methyl viologen-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph*. *stadtmanae* (Wong, Ph.D thesis 1999). However, incubation of the inactive F<sub>420</sub>H<sub>2</sub> dehydrogenation activity in 200 mM KPO<sub>4</sub> (final concentration) also did not restore activity.

The inactive  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae was reactivated when the protein was incubated with 20  $\mu$ M FAD. The protein might also have been destabilized over time, with loss of FAD, resulting in the loss of  $F_{420}H_2$  dehydrogenation activity. Indeed, part of the reactivation process for the  $F_{420}$   $H_2$  as activity of Msph. stadtmanae requires the addition of FAD, under reducing conditions, prior to assay; in the absence of added FAD, the observed  $F_{420}$   $H_2$  as activity is 40- to 50-fold lower, compared to protein reactivated with FAD (Chapter 8).

In contrast, addition of FAD did not lead to an increase in the specific activity of the purified F<sub>420</sub>H<sub>2</sub> dehydrogenase of *Ms. mazei* Gö1 (Abken and Deppenmeier 1997).

The purified  $F_{420}H_2$  dehydrogenation activities (located on the  $F_{420}$   $H_2$ ase) of Msp. hungatei GP1 and Ms. barkeri Fusaro did not require addition of FAD; incubation of the  $F_{420}H_2$  dehydrogenation activity of Msp. hungatei GP1 with FAD did not increase the yield or specific activity of the  $F_{420}H_2$  dehydrogenation activity.

However, there are several examples of proteins requiring reactivation with FAD.

Nelson et al. (1984) reported that the loss of activity of the F<sub>420</sub> H<sub>2</sub>ase of

Methanobacterium formicicum during its purification via anaerobic hydrophobic column chromatography was due to loss of FAD; addition of FAD restored the F<sub>420</sub> H<sub>2</sub>ase activity. Addition of FAD is not typically used in the reactivation of the F<sub>420</sub> H<sub>2</sub>ase activity of Mc. voltae, but FAD is essential for reactivation after passage of the F<sub>420</sub> H<sub>2</sub>ase through hydrophobic chromatography matrices, and also after inactivation via heat treatment (Muth et al. 1987). The F<sub>420</sub> H<sub>2</sub>ase of Msp. hungatei GP1 was routinely reactivated in the presence of FAD by Choquet and Sprott (1991).

Incubation of protein extract of Msph. stadtmanae with FAD, under reducing conditions, may allow for reintegration of FAD into the protein and activation of the  $F_{420}$   $H_2$  as activity. A similar step may be necessary to fully activate the  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae, particularly as the activity progresses through various chromatography steps.

## 7.4.5. Resolution of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity in Methanosphaera stadtmanae

Negligible  $F_{420}$   $H_2$  as activity had been detected in cell-free extract (< 0.01  $\mu$ mol  $F_{420}$  reduced·min<sup>-1</sup>·mg<sup>-1</sup>), such that the means for  $F_{420}$ -reduction in Msph. stadtmanae

were unclear (van de Wijngaard et al. 1991, Wong et al. 1994, Wong, Ph.D thesis 1999). It had been suggested that the role of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity in *Msph. stadtmanae* may be associated with methanophenazine-dependent F<sub>420</sub> reduction, since F<sub>420</sub>-reduction was observed in the presence of phenazine, F<sub>420</sub>, and cell-free extract, under an atmosphere of H<sub>2</sub> (Wong, Ph.D thesis 1999). A mechanism for F<sub>420</sub>-reduction was proposed for *Msph. stadtmanae*.

In the initial step, MV H<sub>2</sub>ase (Mvh), which is typically associated with CoM-S-S-CoB reduction, would catalyze the reduction of methanophenazine (MP) to MPH<sub>2</sub> (Wong, Ph.D thesis 1999). The F<sub>420</sub>-nonreducing hydrogenase (Vho) of *Ms. mazei* Gö1, which is part of the H<sub>2</sub>:CoM-S-S-CoB oxidoreductase complex, catalyzes the reduction of 2-OH-phenazine (Abken *et al.* 1998, Ide *et al.* 1999). The purified F<sub>420</sub> H<sub>2</sub>ase and MV H<sub>2</sub>ase of *Mtb. thermoautotrophicus* ΔH are both able to catalyze the H<sub>2</sub>-dependent reduction of 2-OH-phenazine (Meuer *et al.* 1999). Finally, MP-H<sub>2</sub> would become the source of electrons for the reduction of F<sub>420</sub> *via* the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activity detected in the cell-free extract of *Msph. stadtmanae* (Wong PhD thesis 1999).

Neither the  $F_{420}H_2$  dehydrogenase (Fpo complex) or methanophenazine are found in *Msph. stadtmanae* (Fricke *et al.* 2006). The question remains, what is the source of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity in *Msph. stadtmanae*? The results from our studies indicate the presence of two possible sources of phenazine-dependent  $F_{420}H_2$  dehydrogenation activities.

One source of  $F_{420}H_2$  dehydrogenation activity appears to be consistent with the presence of an  $F_{420}$   $H_2$  ase: enzymatic analyses indicate the presence of this activity in

Msph. stadtmanae. The recently published genome also indicates the presence of genes (frhADGB) encoding for a putative F<sub>420</sub> H<sub>2</sub>ase in Msph. stadtmanae (Fricke et al. 2006). This activity has been detected in the cell-free extract and co-elutes with an F<sub>420</sub>H<sub>2</sub> dehydrogenation activity throughout the chromatographic steps (Table 7.6). The F<sub>420</sub> H<sub>2</sub>ase of Mc. voltae also possesses an F<sub>420</sub>H<sub>2</sub> dehydrogenation activity linked to CoM-S-S-CoB reduction (Brodersen et al. 1999). Our studies have shown that the respective F<sub>420</sub> H<sub>2</sub>ases from Msp. hungatei GP1 (Chapter 5) and Ms. barkeri Fusaro (Chapter 6) both possess phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity. However, analyses of the F<sub>420</sub> H<sub>2</sub>ase and F<sub>420</sub>H<sub>2</sub> dehydrogenation activities of Msph. stadtmanae show some striking differences.

When incubated under anaerobic and optimal conditions, the  $F_{420}$  H<sub>2</sub>ase activity of Msph. stadtmanae (in cell-free extract) is approximately 60-fold greater than the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity (Chapter 8); with respect to Msp. hungatei GP1 and Ms. barkeri Fusaro, under identical conditions there is less than a four-fold difference in the anaerobic  $F_{420}$  H<sub>2</sub>ase: $F_{420}H_2$  dehydrogenation activity ratios ( $F_{420}$  H<sub>2</sub>ase/ $F_{420}H_2$  DeH<sub>2</sub>ase = 3.5 for Msp. hungatei GP1, Ms. barkeri Fusaro). The  $F_{420}$  H<sub>2</sub>ase: $F_{420}H_2$  dehydrogenation activity ratio for Methanobacterium bryantii, a member of the Methanobacteriales, is 3.93 (See Chapter 4 and 8).

In contrast to the homologous activities in Msp. hungatei GP1 and Ms. barkeri Fusaro, the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae is aerobically unstable. However, although the  $F_{420}$   $H_2$  as activity was enriched under anaerobic conditions, the activity was stable when stored under aerobic conditions (Chapter 8); on this basis, it would seem that the phenazine-dependent  $F_{420}H_2$ 

dehydrogenation activity is not associated with the  $F_{420}$   $H_2$  as activity in Msph. stadtmanae, although the activities appeared to co-migrate. Whether this is an anomaly specific for the  $F_{420}$   $H_2$  as of Msph. stadtmanae has yet to be resolved. Further study of the  $F_{420}$   $H_2$  as activity of Msph. stadtmanae is required to elucidate the nature of  $F_{420}H_2$  dehydrogenation activity associated with this putative protein.

Genomic data of Msph. stadtmanae indicates the presence of a gene encoding a putative  $F_{420}H_2$  oxidase (Fricke et~al. 2006). An  $F_{420}H_2$  oxidase (FprA, 45 kDa), which contains FMN, was purified from the cell-free extract of Methanobrevibacter arboriphilus, a methano-archaeon found in the intestinal tract of termites (Seedorf et~al. 2004).  $F_{420}H_2$  oxidase catalyzes the  $F_{420}H_2$ -dependent reduction of  $O_2$  to form 2  $H_2O$  and 2  $F_{420}$  (Seedorf et~al. 2004). The presence of an  $F_{420}H_2$  oxidase in Msph. stadtmanae would not be surprising, as the large intestine contains enough oxygen ( $PO_2 = \sim 15$  mm Hg) to inhibit the growth of strictly anaerobic microorganisms, and this methanoarchaeon does not possess a superoxide dismutase or a catalase (Levitt 1970, Fricke et~al. 2006).

The  $F_{420}H_2$  oxidase is a possible source of the observed phenazine-dependent  $F_{420}H_2$  dehydrogenation activity. Like the  $F_{420}H_2$  dehydrogenation activity, the  $F_{420}H_2$  oxidase activity is also aerobically unstable, with activity declining at a similar rate as the  $F_{420}H_2$  dehydrogenation activity when stored under aerobic conditions (Figure 7.11). The phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was restored/enhanced when the protein was incubated with FAD, under reducing conditions. As the  $F_{420}H_2$  oxidase contains FMN instead of FAD, one way to determine whether the  $F_{420}H_2$  oxidase is a source of the observed phenazine-dependent  $F_{420}H_2$  dehydrogenation activity would be to see if the activity is enhanced when the protein is incubated with FMN.

It should be noted that the  $F_{420}$   $H_2$  as of Mtb. thermoautotrophicus  $\Delta H$  is also capable of catalyzing the  $F_{420}H_2$  dependent reduction of  $O_2$  (Jacobson et al. 1982, Livingston et al. 1987), so we cannot conclude that the  $F_{420}H_2$  oxidase is being expressed without further enrichment of this activity in Msph. stadtmanae.

Reverse transcription polymerase chain reaction (RT PCR) may be one method to elucidate whether the  $F_{420}$   $H_2$  ase or  $F_{420}H_2$  oxidase are being expressed. RT PCR can also be used to quantify transcription and determine the levels at which frh and fprA are expressed. Once it has become clear that the genes encoding for the respective proteins are being expressed, strategies can be developed for purification of the specific protein.

# 8. $F_{420}$ -reducing hydrogenase activity in *Methanosphaera stadtmanae* 8.1. Introduction

According to recently published genomic information, Methanosphaera (Msph.) stadtmanae possesses the genes encoding several distinct putative hydrogenases; an  $F_{420}$ -nonreducing hydrogenase (MV  $H_2$ ase, mvhADG); an Ech-type hydrogenase (ehb); and an  $F_{420}$ -reducing hydrogenase ( $F_{420}$   $H_2$ ase, frhADGB) (Fricke et al. 2006).

The MV H<sub>2</sub>ase from *Msph. stadtmanae* has been characterized (Wong, PhD thesis 1999); as with most hydrogenotrophic methanoarchaea, the expected function of the MV H<sub>2</sub>ase, in conjuction with Hdr, is to catalyze the H<sub>2</sub>-dependent reduction of CoM-S-S-CoB to the free cofactors CoM-SH and CoB-SH (Fricke *et al.* 2006). The *mvh* operon of *Mtb. thermoautotrophicus* also encodes the electron carrier polyferredoxin (*mvhB*) (Reeve *et al.* 1989); this is apparently not the case for *Msph. stadtmanae*, where *mvhB* is not found in the genome (Fricke *et al.* 2006). The significance of this gene is unclear, as polyferredoxin is not a substrate for the Mvh hydrogenase of *Mtb. thermoautotrophicus*, nor does it seem to be essential for its activity.

The genome of *Msph. stadtmanae* reveals the presence of a single Ech hydrogenase (Ehb), which is expected to catalyze the reduction of Fd for the reductive carboxylation of acetyl-CoA to pyruvate (Hedderich 2004, Fricke *et al.* 2006):

$$H_2 + Fd \rightarrow Fd-H_2$$

$$Acetyl\text{-}CoA + CO_2 + Fd\text{-}H_2 \rightarrow pyruvate + CoA + Fd$$

The appearance of an  $F_{420}$   $H_2$  as has not been detected in *Msph. stadtmanae*, and the hypothesis was that this activity did not exist in this methanoarchaeon (Deppenmeier *et al.* 1989, van de Wijngaard *et al.* 1991, Wong *et al.* 1994). Recent observations

indicate the presence of an  $F_{420}$ -reducing  $H_2$  as activity in the cell-free extract of Msph. stadtmanae; this activity was enriched during studies of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity (Chapter 7). The recently sequenced genome of Msph. stadtmanae also indicates the presence of genes encoding sequences consistent with the  $F_{420}$   $H_2$  as (frhADBG); however it was unclear to Fricke et al. (2006) whether the genes were silent, or if the activity itself was just very low, since this activity was not reported from 3 different laboratories (Deppenmeier et al. 1989, van de Wijngaard et al. 1991, Wong et al. 1994).

While there is no apparent need for this activity during methanogenesis when grown on  $H_2$  and methanol (CH<sub>3</sub>OH), *Msph. stadtmanae* can produce limited quantities of CH<sub>4</sub> in the presence of  $H_2$  and L-serine (Lin and Sparling 1995, 1998), and 2-propanol and CH<sub>3</sub>OH (Wong *et al.* 1994). During methanogenesis from  $H_2$  and L-serine, the hydroxymethyl group from serine is transferred to  $H_4$ MPT, catalyzed by  $H_4$ MPT-dependent serine hydroxymethyl transferase.  $N^5N^{10}$ -methylene- $H_4$ MPT is then reduced to CH<sub>3</sub>- $H_4$ MPT, a step requiring  $F_{420}H_2$ . 2-propanol can be used as the source of electrons for methanogenesis with methanol. 2-propanol is oxidized to acetone *via* 2-propanol ADH, with the electrons transferred to NADP<sup>+</sup>. Electrons from NADPH are then transferred to  $F_{420}$  *via*  $F_{420}H_2$ :NADP<sup>+</sup> oxidoreductase (Fno);  $F_{420}H_2$  is then used to reduce CoM-S-S-CoB (Wong *et al.* 1994).

 $F_{420}H_2$  is also required for cellular biosynthesis, specifically the reduction of NADP<sup>+</sup> to NADPH (Wong *et al.* 1994). Fno catalyzes the  $F_{420}H_2$ -dependent reduction of NADP<sup>+</sup> (Elias *et al.* 2000):

$$F_{420}H_2 + NADP^+ \longrightarrow F_{420} + NADPH + H^+$$

The low cellular concentration of  $F_{420}$  (0.16 nmol  $F_{420}$ ·mg<sup>-1</sup> protein) supports the biosynthetic role of  $F_{420}H_2$  in *Msph. stadtmanae* (Wong *et al.* 1994).

Previous studies of the  $F_{420}$ -reducing activity of Msph. stadtmanae indicated extremely low  $F_{420}$ -reducing activity (<0.01 µmol  $F_{420}$  reduced·min<sup>-1</sup>·mg<sup>-1</sup> protein) in this methanoarchaeon (Wong et~al. 1994). Slow  $H_2$ -dependent reduction of  $F_{420}$  using cell-free extract had been observed, but appeared to be dependent on the presence of phenazine in a high phosphate (200 mM KPO<sub>4</sub>, pH 7.0) buffer. It was suggested that phenazine may act as an intermediate between  $H_2$  and  $F_{420}$ ; the observed  $F_{420}$ -reducing activity under these conditions was reportedly very low (Wong, PhD thesis 1999).

Subsequent studies indicated that the  $F_{420}$   $H_2$  as activity was not dependent on the presence of mediators such as phenazine or Fd (Table 7.5, Chapter 7). In this chapter we present an initial description and characterization of an  $F_{420}$   $H_2$  as activity in the cell-free extract of Msph. stadtmanae.

#### 8.2. Materials and methods

For a comprehensive look at the materials and methods used for this chapter, the reader is referred to the following sections in Chapter 3:

- 3.2. Growth media
- 3.3. Cell lysis
- 3.6. Enzyme assays
  - 3.6.1. F<sub>420</sub>H<sub>2</sub> dehydrogenation assays
  - 3.6.2. Hydrogenase assays
    - 3.6.2.1. F<sub>420</sub>-reducing hydrogenase activity
    - 3.6.2.2. Methyl viologen reducing hydrogenase activity

Several buffer systems were tested to determine optimal conditions for assay of the F<sub>420</sub>-reducing activity: 50 mM Tris/Cl, pH 8.0, containing 20 mM MgCl<sub>2</sub>•6H<sub>2</sub>0 and 20 mM DTT; 200 mM potassium phosphate, pH 7.0, with 0.5 M sucrose and 20 mM DTT; and 100 mM HEPES, pH 7.0, with 200 mM KCl, 20 mM MgCl<sub>2</sub> 6H<sub>2</sub>0 and 20 mM DTT. All manipulations of the cell-free extract were conducted under anaerobic conditions, with the extract being stored at 4°C under an atmosphere of N<sub>2</sub>.

#### 3.7. Gel electrophoresis

3.7.2. Native gradient PAGE gel electrophoresis

#### 3.8. Genome sequence analysis

Clustal W software (Chenna et al. 2003) was used to generate a phylogenetic tree of the amino acid sequences coding for the  $\alpha$ -subunit (FrhA) from various methanoarchaea.

#### 8.3. Results

#### 8.3.1. Reactivation of the $F_{420}$ -reducing hydrogenase activity

Preliminary analyses of the  $F_{420}$   $H_2$  as activity were initiated with an examination of several buffers to determine optimal conditions for assay of the  $F_{420}$ -reducing activity; the results are shown in Table 8.1. The highest levels of  $F_{420}$ -reducing activity were observed when using 100 mM HEPES or 200 mM KPO<sub>4</sub> buffer; the former was used by Choquet and Sprott (1991) to assay the  $F_{420}$   $H_2$  as activity of Msp. hungatei GP1, and the latter is typically used to assay the  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae, with  $N_2$  in place of  $H_2$  (Chapter 7). When measurement of the  $F_{420}$   $H_2$  as activity was preceded by incubation of the protein extract under reducing conditions (in either 100 mM HEPES or 200 mM KPO<sub>4</sub> assay buffer), with 10  $\mu$ M  $F_{420}$  and 10  $\mu$ M FAD at room

Table 8.1. Assay of the  $F_{420}$ -reducing hydrogenase activity of *Msph. stadtmanae* in different buffers, with or without incubation with 10  $\mu$ M FAD and 10  $\mu$ M  $F_{420}$ .

Assay buffer	Activity (μmol F <sub>420</sub> reduced min <sup>-1</sup> mg <sup>-1</sup> )	
100 mM HEPES, pH 7.0	$0.26 \pm 0.02$	
100 mM HEPES, pH 7.0 <sup>†</sup>	$0.005 \pm 0.0005$	
200 mM KPO <sub>4</sub> , pH 7.0	$0.21 \pm 0.01$	
200 mM KPO <sub>4</sub> , pH 7.0 <sup>†</sup>	$0.005 \pm 0.001$	
50 mM Tris/Cl, pH 8.0 <sup>†</sup>	< 0.0004	

 $<sup>^{\</sup>dagger}$  no reactivation with FAD or F<sub>420</sub>

temperature, a 40- to 50-fold increase in activity was observed relative to those tubes without prior reactivation with FAD.

50 mM Tris/Cl buffer, pH 8.0, had initially been used to survey the membrane-bound F<sub>420</sub> H<sub>2</sub>ase activity of a variety of methanoarchaea, including *Msph. stadtmanae*; F<sub>420</sub>-reduction in cell-free extract of all tested methanoarchaea was observed, with the exceptions of *Methanolobus tindarius* and *Msph. stadtmanae* using this buffer (Deppenmeier *et al.* 1989); our results using 50 mM Tris/Cl, pH 8.0 are consistent with previous results.. It should be noted, however, that the optimal pH for measurement of F<sub>420</sub>-reduction is typically lower, from pH 7.0 – 7.5 (Fox *et al.* 1987, Muth *et al.* 1987, Fiebig and Friedrich 1989, Choquet and Sprott 1991). In contrast, significant levels of the F<sub>420</sub>- reducing H<sub>2</sub>ase activities of *Msp. hungatei* GP1 and *Ms. barkeri* Fusaro (in cell-free extract or purified enzyme, non-reactivated) could be observed using 50 mM Tris/Cl, pH 8.0, buffer with little loss of activity.

From this point forward, the  $F_{420}$  H<sub>2</sub>ase activity of *Msph. stadtmanae* was assayed in 100 mM HEPES, pH 7.0, with 200 mM KCl, 20 mM MgCl<sub>2</sub>·6H<sub>2</sub>0 and 20 mM DTT, with protein preincubated in the reaction buffer-filled assay tubes with 10  $\mu$ M FAD and 10  $\mu$ M F<sub>420</sub> at room temperature prior to assay; measurement of activity began with the addition of F<sub>420</sub> (final concentration 25-30  $\mu$ M).

## 8.3.2. Detection of the $F_{420}$ -reducing hydrogenase activity via native PAGE electrophoresis

Studies of the  $F_{420}$   $H_2$  as activity of *Msph. stadtmanae* were also conducted *via* native gel electrophoresis simultaneously with the enzyme activity assays in the different buffers. Activity staining was performed using methyl viologen, a synthetic electron acceptor also used as substrate by  $F_{420}$   $H_2$  asses.

On initial analysis, the results following electrophoresis of cell-free extract on activity-stained native PAGE gels appeared to be consistent with previous reports of low or no expression of the  $F_{420}$  H<sub>2</sub>ase in *Msph. stadtmanae*. High molecular weight (MW) bands (600–1300 kDa) possessing hydrogenase activity, attributed to the presence of the  $F_{420}$  H<sub>2</sub>ase, are not found in the expected positions on activity-stained 3- 20% native gradient PAGE gels (Figure 8.1). However, a single band with an approximate molecular weight of 232 kDa can be observed (Figure 1); this band has a higher apparent MW than the MV H<sub>2</sub>ase (> 100 kDa), yet a lower MW than the high MW  $F_{420}$  H<sub>2</sub>ase band (< 600 kDa). For comparison, the MW of the MV H<sub>2</sub>ase described by Wong (Ph.D thesis, 1999) was 49 kDa.

The  $F_{420}$   $H_2$  as was not expected to comprise a major portion of the overall protein in *Msph. stadtmanae*, since this activity was likely associated with anabolic

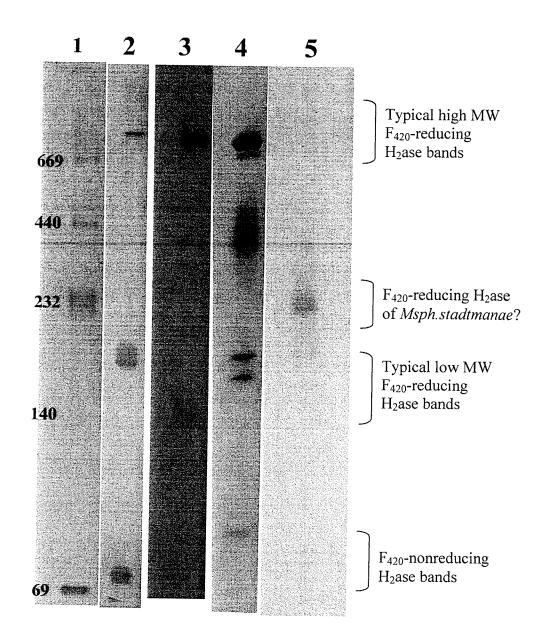
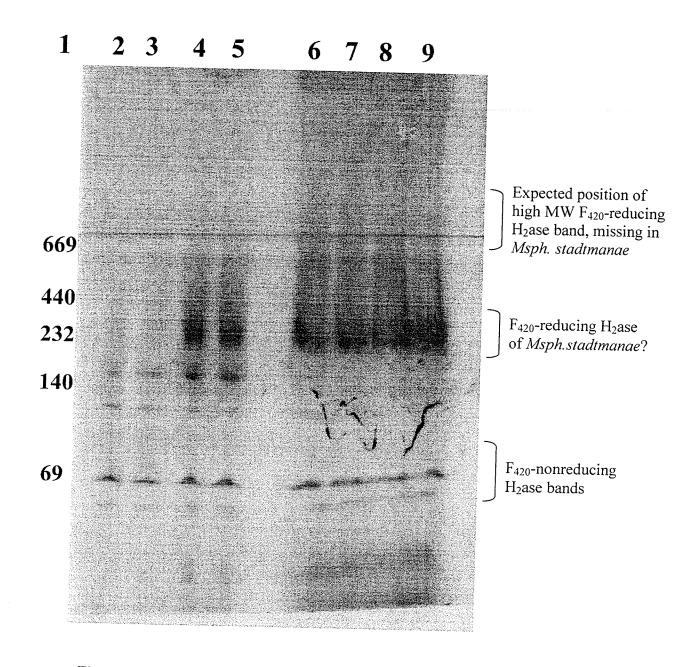


Figure 8.1. A comparison of the hydrogenase activities from the cell-free extract (CFE) of several methanoarchaea on activity-stained 3-20% native gradient PAGE gels. 140 μg of CFE protein added per lane. Gel incubated in 50 mM Tris/Cl, pH 8.0, with 20 mM MgCl<sub>2</sub>·6H<sub>2</sub>0, 2 mM methyl viologen. 1. Molecular markers, kDa. 2. *Methanosarcina barkeri* Fusaro (grown on CH<sub>3</sub>OH). 3. *Methanospirillum hungatei* GP1. 4. *Methanococcus voltae*. 5. *Methanosphaera stadtmanae*.

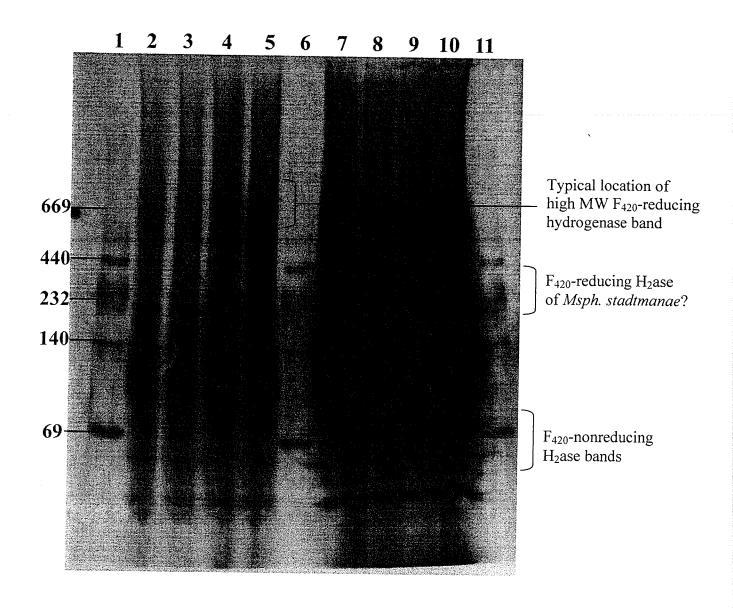
functions rather than methanogenesis. As such, the protein loaded ( $\sim$ 140  $\mu$ g) may not have contained detectable amounts of the  $F_{420}$   $H_2$ ase. To determine the minimal amount of protein necessary to produce detectable hydrogenase activity in the gel, a series of dilutions were made of the clarified cell-free extract (membranes removed) (final protein content loaded: 1100, 552, 276, and 138  $\mu$ g) and loaded onto a 3-20% native gradient PAGE (Figures 8.2 and 8.3).

Based on the banding patterns, two observations were made of these gels. First, the typical high molecular weight F<sub>420</sub> H<sub>2</sub>ase activity bands were not visualized on the activity-stained gel, regardless of the amount of protein loaded onto the gel (Figure 8.2). The coomassie blue-stained gel indicates the presence of a high molecular weight protein band that typically displays hydrogenase activity (from methanoarchaea that possess the F<sub>420</sub> H<sub>2</sub>ase) (Figure 8.3). One can speculate that this band may represent the F<sub>420</sub> H<sub>2</sub>ase of *Msph. stadtmanae*, as this band lies at a position typically occupied by the F<sub>420</sub> H<sub>2</sub>ase of methanoarchaea such as *Mtb. thermoautotrophicus*, *Ms. barkeri* Fusaro, *Msp. hungatei*, *Mc. voltae*, etc, when electrophoresed on native PAGE gels. However, there is no evidence to suggest that this is the F<sub>420</sub> H<sub>2</sub>ase, since this band does not stain for hydrogenase activity. Potential loss of FAD is not expected to be a factor here, as methyl viologen reduction is not dependent on the presence of FAD (Muth *et al.* 1987, Fox *et al.* 1987, Thauer *et al.* 1993)

The second observation was the appearance of an activity stained band with a molecular weight of approximately 232 kDa (Figure 8.2). This band was faint when the standard amount of protein (140  $\mu$ g) was electrophoresed. The molecular weight of the band would seem to disqualify it as the MV H<sub>2</sub>ase (Mvh), which has a molecular weight



**Figure 8.2**. A comparison of the hydrogenase activities from the clarified cell-free extract of *Methanosphaera stadtmanae*, with varying amounts of protein, as observed on activity-stained 3-20% native gradient PAGE gels. 1. Molecular markers, kDa. 2,3. 138 μg. 4,5, 276 μg. 6,7. 552 μg. 8,9. 1100 μg.



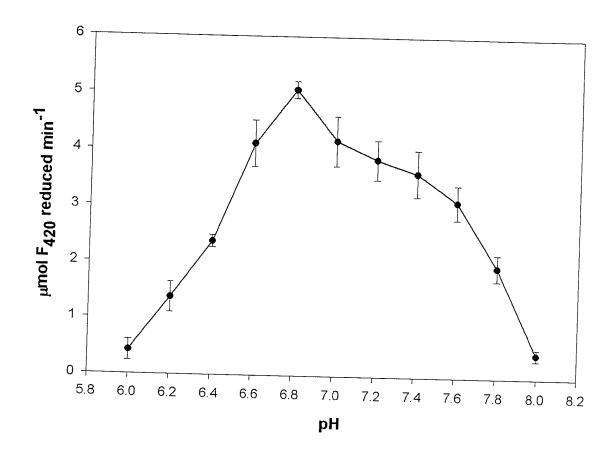
**Figure 8.3**. Activity (red)- and coomassie blue-stained 3-20% native gradient PAGE comparison of the hydrogenase activities from the clarified cell-free extract of *Methanosphaera stadtmanae*, with varying amounts of protein. 1,6,11. Molecular markers, kDa. 2,3. 138  $\mu$ g. 4,5. 276  $\mu$ g. 7,8. 552  $\mu$ g. 9,10. 1100  $\mu$ g.

of approximately 49 kDa when electrophoresed on a native gradient PAGE (Wong, PhD thesis 1999). However, the apparent molecular weight of the band is also much lower than the typical high molecular weight band typically observed for the  $F_{420}$   $H_2$ ase (600–1300 kDa) (Michel *et al.* 1995).

## 8.3.3. Catalytic and kinetic properties of the $F_{420}$ -reducing hydrogenase activity in the cell-free extract of Methanosphaera stadtmanae

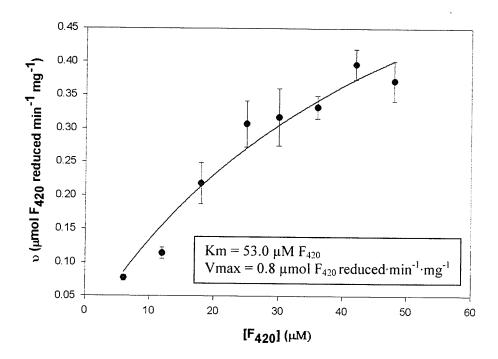
The  $F_{420}$  H<sub>2</sub>ase activity of Msph. stadtmanae, in clarified cell-free extract, appears to be highest at pH 6.8 (Figure 8.4). NAD<sup>+</sup> and NADP<sup>+</sup>, both hydride donors/acceptors, were also tested for use as electron acceptors but could not substitute for coenzyme  $F_{420}$ . H<sub>2</sub>-dependent phenazine reduction was observed using cell-free extract of Msph. stadtmanae; phenazine is reduced at a rate of 0.16  $\mu$ mol phenazine·min<sup>-1</sup>·mg<sup>-1</sup>. Phenazine is also reduced by cell-free extract under an atmosphere of  $N_2$ , presumably using reducing equivalents remaining in the cell-free extract; this value was taken into account when calculating the H<sub>2</sub>-dependent activity. The significance of this activity is unclear, as methanophenazine is apparently not present in Msph. stadtmanae.

A kinetic profile of the  $F_{420}$ -reducing activity in cell-free extract is shown in Figure 8.5. The activity follows simple Michaelis Menten kinetics, with an apparent Km for  $F_{420} = 53 \mu M$ , and  $V_{max} = 0.8 \mu mol F_{420} reduced min^{-1} mg^{-1}$  (Lineweaver Burk: apparent Km= 89  $\mu$ M  $F_{420}H_2$ ,  $V_{max} = 1.2 \mu mol F_{420} reduced min^{-1} mg^{-1}$ ).

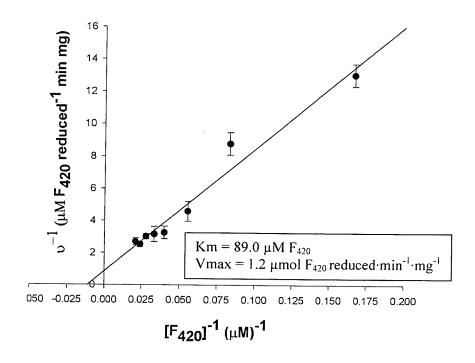


**Figure 8.4.** pH profile of the  $F_{420}$ -reducing hydrogenase activity in clarified cell-free extract. CCFE reactivated with FAD and  $F_{420}$  prior to assay. pH of assay buffer established by varying concentration of  $K_2HPO_4$  and  $KH_2PO_4$ . Final  $[KPO_4] = 200$  mM.  $[F_{420}] = 20-25 \ \mu M$ .

A.



B.



**Figure 8.5**. Kinetics plots of the  $F_{420}$ -reducing hydrogenase activity in the cell-free extract of *Methanosphaera stadtmanae*. (A) Michaelis Menten kinetics plot (B) Lineweaver Burk kinetics plot.  $[F_{420}] = 20\text{-}25~\mu\text{M}$ 

### 8.3.4. Inhibitor analyses of the $F_{420}$ -reducing hydrogenase activity

Prior to publication of the genome of Msph. stadtmanae, it was not clear whether the  $F_{420}$ -reducing activity in the cell-free extract was due to the presence of a distinct  $F_{420}$   $H_2$ ase, or if the observed activity was the coupled reaction of two or more proteins, as suggested by Wong 1999 (PhD thesis). A study had been conducted examining the effect of various compounds on the MV  $H_2$ ase activity by Denny Wong (PhD thesis, 1999). Various metals ( $Ni^{2+}$ ,  $Hg^{2+}$ , etc), toxic anions ( $CN^-$ ,  $N_3^-$ ), and chemicals such as the metal chelator EDTA, were tested for their ability to inhibit the MV  $H_2$ ase activity.

To help determine whether the  $F_{420}$ -reduction was indeed the activity of a distinct  $F_{420}$   $H_2$ ase, a similar study was performed to determine the extent, if any, of inhibition by these chemicals on the  $F_{420}$   $H_2$ ase activity in cell-free extract. The potentially inhibitory substances were added to the assay tubes after reactivation of the  $F_{420}$   $H_2$ ase activity.

The results are shown in Table 8.2. Heavy metal ions inhibit the  $F_{420}$   $H_2$  as activity, the most inhibitory being  $Cu^{2+}$  (17.9% activity remaining).  $CN^-$  and  $N_3^-$  stimulated the  $F_{420}$   $H_2$  as activity, as did EDTA and the methanogenic inhibitor 2-bromoethanesulfonic acid (2-BES), a structural analog of CoM-SH.

Diphenyleneiodonium (DPI) chloride also inhibits  $F_{420}$  reduction; in contrast, the methylviologen-reducing hydrogenase activity of the cell-free extract was not inhibited in the presence of DPI.

Table 8.2. Inhibition of the  $F_{420}$ -reducing hydrogenase activity of the clarified cell-free extract of *Methanosphaera stadtmanae*. Hydrogenase activity was reactivated with 10  $\mu$ M FAD and 10  $\mu$ M  $F_{420}$  prior to addition of inhibitor.  $[F_{420}] = 25$ -30  $\mu$ M.

Inhibitor (mM)	Activity (μmol F <sub>420</sub> reduced min <sup>-1</sup> mg <sup>-1</sup> )	% activity		
-	$\boldsymbol{0.28 \pm 0.03}$	100.0		
CN <sup>-</sup> (1)	$0.31 \pm 0.04$	109.4		
N <sub>3</sub> -(3)	$0.35 \pm 0.09$	124.5		
2-BES (4)	$0.31 \pm 0.01$	111.3		
EDTA (10)	$0.36 \pm 0.01$	128.3		
Cd <sup>2+</sup> (1)	$0.07 \pm 0.02$	24.5		
Cu <sup>2+</sup> (1)	$0.05 \pm 0.01$	17.9		
Zn <sup>2+</sup> (1)	$0.07 \pm 0.01$	26.4		
DPI (0.005)	$0.18 \pm 0.04$	66.0		
DPI (0.010)	$0.06 \pm 0.02$	22.6		

# 8.3.5. Comparison of the $F_{420}$ -reducing hydrogenase activity of Methanosphaera stadtmanae in cell-free extract with the homologous activity from various methanoarchaea.

For comparison purposes, the  $F_{420}$  H<sub>2</sub>ase activities (in cell-free extract) from several methanoarchaea were measured.  $F_{420}$  H<sub>2</sub>ase activity from each methanoarchaeon was assayed under conditions modified from Choquet and Sprott (1991): protein extract was reactivated and assayed in 'standard buffer' of 100 mM HEPES, pH 7.0, with 10  $\mu$ M FAD, 10  $\mu$ M F<sub>420</sub>, 20 mM MgCl<sub>2</sub>·6H<sub>2</sub>O and 200 mM KCl, at room temperature (~25°C). The results of this brief survey are shown in Table 8.3. *Methanolobus tindarius* was also

included as a negative control, as this methanoarchaeon apparently does not possess an  $F_{420}$   $H_2$ ase (Deppenmeier *et al.* 1989).

Table 8.3. Comparison of the  $F_{420}$ -reducing hydrogenase activities in the clarified cell-free extract of various methanoarchaea. All assays performed at room temperature as described in text except where noted, following activation as described, with modification, by Choquet and Sprott (1991). Values determined using Michaelis Menten kinetics.

Methanoarchaeon	Order	Growth substrate	Km (μΜ F <sub>420</sub> )	Vmax <sup>†</sup>	
Ml. tindarius	Methanosarcinales	СН₃ОН	-	-	
<i>Ms. barkeri</i> Fusaro	Methanosarcinales	СН₃ОН	$26.3 \pm 3.0$	$2.1 \pm 0.1$	
Msp. hungatei GP1	Methanomicrobiales	H <sub>2</sub> :CO <sub>2</sub>	$73.5 \pm 18.5$	$1.8 \pm 0.3$	
Mc. voltae	Methanococcales	H <sub>2</sub> :CO <sub>2</sub>	$50.0 \pm 9.3$	$1.3 \pm 0.1$	
Mtb. marburgensis <sup>‡</sup>	Methanobacteriales	H <sub>2</sub> :CO <sub>2</sub>	2.4x10 <sup>6</sup>	2.5x10 <sup>4</sup>	
Mtb. marburgensis**	Methanobacteriales	H <sub>2</sub> :CO <sub>2</sub>	$36.5 \pm 9.1$	$1.6 \pm 0.1$	
Mb. bryantii	Methanobacteriales	H <sub>2</sub> :CO <sub>2</sub>	59.4 ± 23.5	5.6 ± 1.4	
Msph. stadtmanae	Methanobacteriales	H <sub>2</sub> :CH <sub>3</sub> OH	53.2 ± 22.6	$0.8 \pm 0.2$	

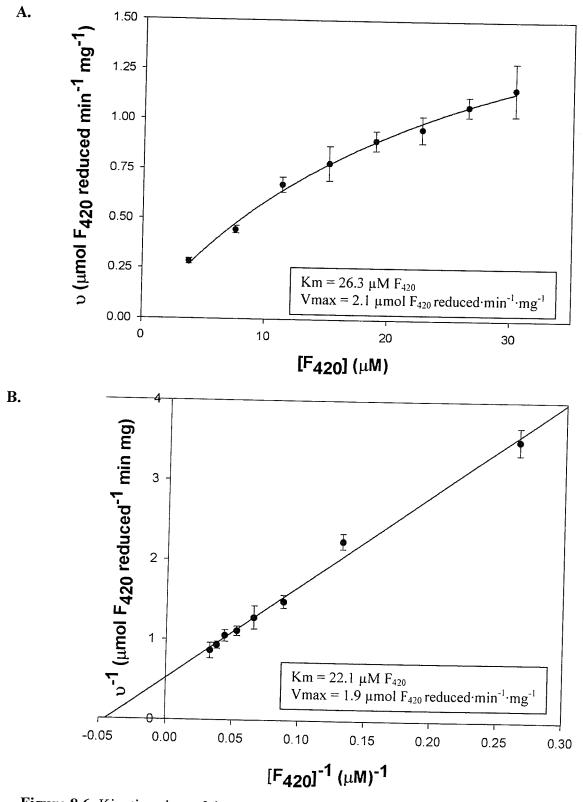
 $<sup>^{\</sup>dagger}$  µmol  $F_{420}$  reduced min  $^{-1}$  mg protein  $^{-1}$ 

<sup>\*</sup>activity of *Mtb. marburgensis* could not be fully activated using the modified method of Choquet and Sprott when incubated at room temperature

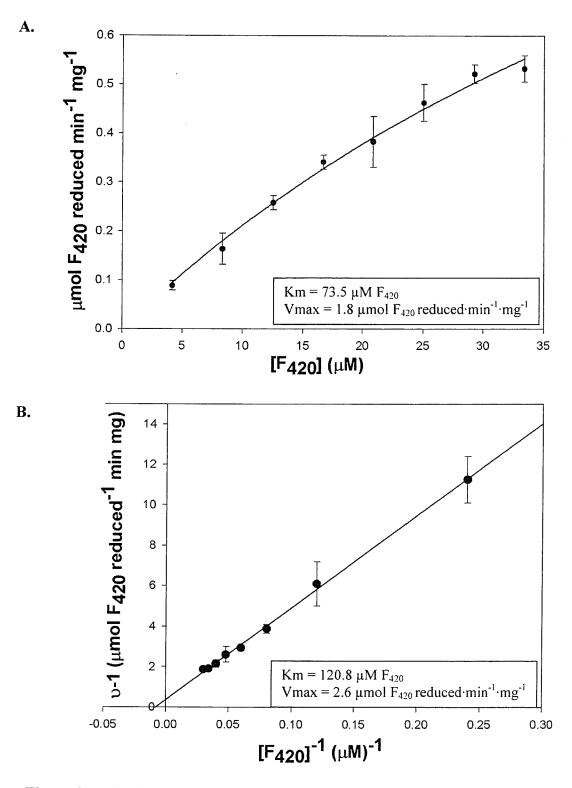
activity of *Mtb. marburgensis* reactivated using the modified method of Choquet and Sprott at 45°C for 30 minutes prior to assay

The apparent Km for  $F_{420}$  of the  $F_{420}$  H<sub>2</sub>ase activity of *Msph. stadtmanae* is similar to the values (with the exception of *Mtb. marburgensis*) obtained for the homologous activity from the other methanoarchaea in the survey (at room temperature). Excluding *Mtb. marburgensis*, the maximum  $F_{420}$  H<sub>2</sub>ase activity in the cell-free extract of *Msph. stadtmanae* is less than 2- to 7-fold lower than the corresponding activity from the cell-free extract of a number of methanoarchaea (Figures 8.6-8.10). Comparisons of  $F_{420}$  H<sub>2</sub>ase activity could not be accurately made with the  $F_{420}$  H<sub>2</sub>ase activity of *Mtb. marburgensis* when using the reactivation procedure of Choquet and Sprott (1991); this is likely due to the different requirements for reactivation of the  $F_{420}$  H<sub>2</sub>ase activity of *Mtb. marburgensis*.

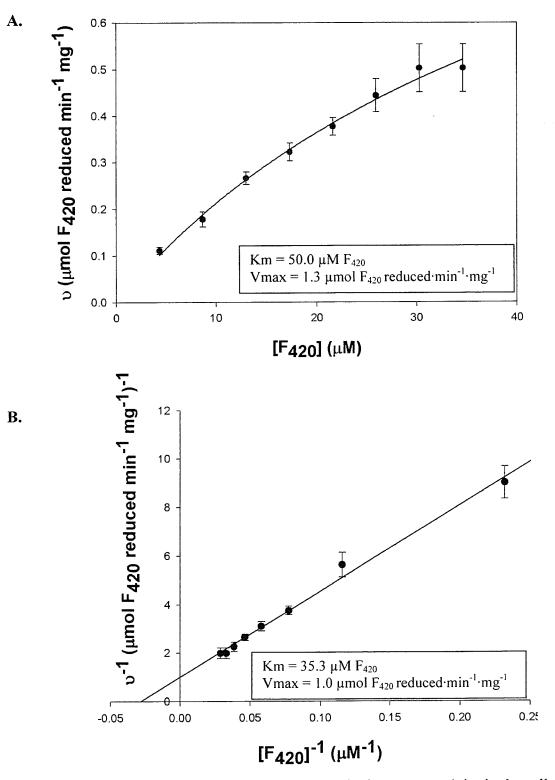
According to Jacobson *et al.* (1981), the  $F_{420}$  H<sub>2</sub>ase activity of *Mtb*. *thermoautotrophicus*  $\Delta H$  is reactivated *via* incubation of protein extract or purified enzyme at 45°C for 1 hour, in 1M salt (K<sup>+</sup> or Na<sup>+</sup>); under these conditions, the apparent Km was reported as 19  $\mu$ M F<sub>420</sub>. Using our standard reaction buffer, at room temperature, the apparent Km was determined to be  $2.4 \times 10^6$   $\mu$ M; when the cell-free extract was incubated in standard reaction buffer at 45°C for ½ hour, the apparent Km<sub>F420</sub> was 36.5  $\mu$ M, considerably closer to the Km value reported by Jacobson *et al.* (1981).



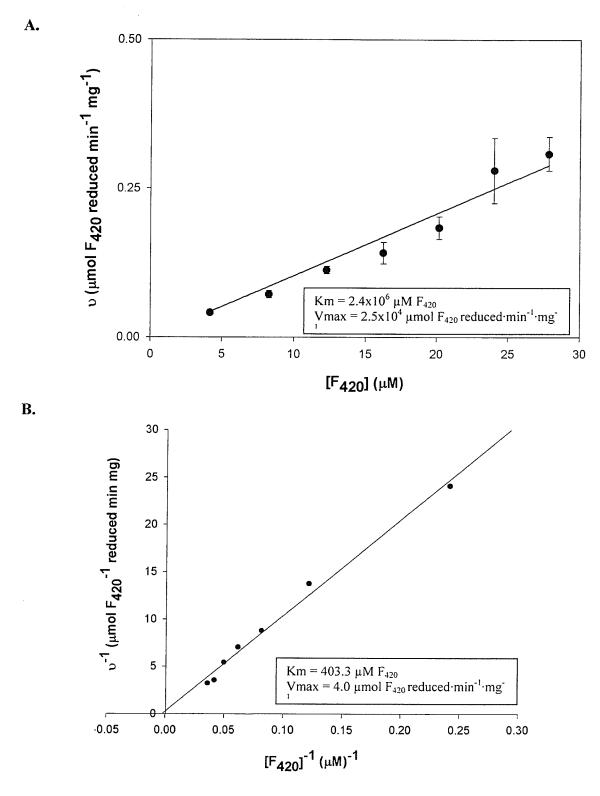
**Figure 8.6**. Kinetics plots of the  $F_{420}$ -reducing hydrogenase activity in the cell-free extract of *Methanosarcina barkeri* Fusaro. (A) Michaelis Menten kinetics plot (B) Lineweaver Burk kinetics plot.  $[F_{420}] = 20$ -25  $\mu$ M



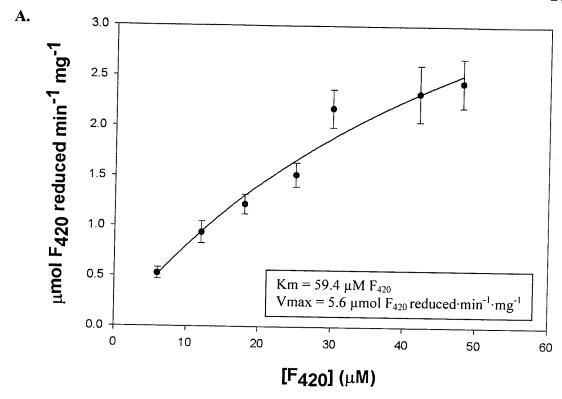
**Figure 8.7**. Kinetics plots of the  $F_{420}$ -reducing hydrogenase activity in the cell-free extract of *Methanospirillum hungatei GP1*. (A) Michaelis Menten kinetics plot (B) Lineweaver Burk kinetics plot.  $[F_{420}] = 20-25 \mu M$ 



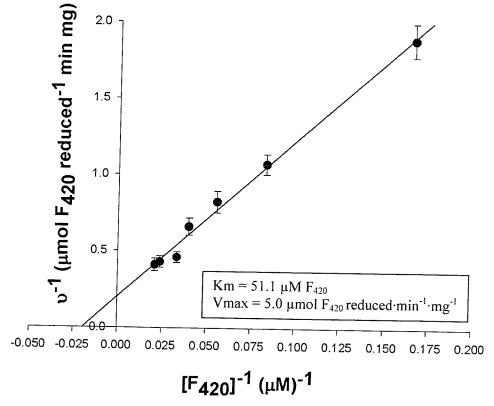
**Figure 8.8.** Kinetics plots of the  $F_{420}$ -reducing hydrogenase activity in the cell-free extract of *Methanococcus voltae*. (A) Michaelis Menten kinetics plot (B) Lineweaver Burk kinetics plot.  $[F_{420}] = 20-25 \mu M$ 



**Figure 8.9**. Kinetics plots of the  $F_{420}$ -reducing hydrogenase activity in the cell-free extract of *Methanothermobacter marburgensis*. (A) Michaelis Menten kinetics plot (B) Lineweaver Burk kinetics plot.  $[F_{420}] = 20$ -25  $\mu$ M



B.

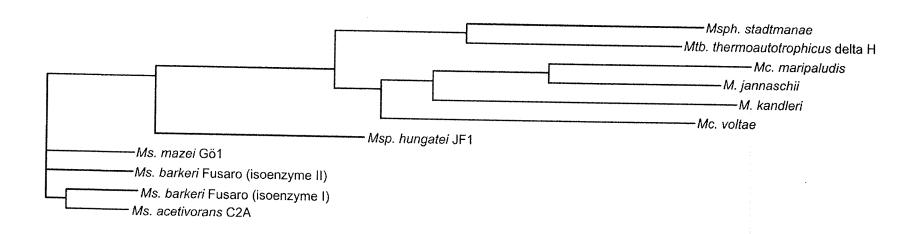


**Figure 8.10**. Kinetics plots of the  $F_{420}$ -reducing hydrogenase activity in the cell-free extract of *Methanobacterium bryantii*. (A) Michaelis Menten kinetics plot (B) Lineweaver Burk kinetics plot.  $[F_{420}] = 20-25 \mu M$ 

### 8.3.6. Genetic relatedness to known methanogenic $F_{420}$ -reducing hydrogenases

BLAST analyses, using the tblastn program, of existing methanogenic genome sequences in the Archaea database, using the amino acid sequences coding for the  $\alpha$ -,  $\beta$ -,  $\gamma$ -, and  $\delta$ -subunits of the putative  $F_{420}$   $H_2$ ase of Msph. stadtmanae (NC\_007681), were performed through the National Center for Biotechnology Information website (NCBI, <a href="http://www.ncbi.nlm.nih.gov/">http://www.ncbi.nlm.nih.gov/</a>). The results indicate that the  $F_{420}$   $H_2$ ase of Msph. stadtmanae is most closely related to the homologous enzyme of Mtb. thermoautotrophicus  $\Delta H$ , with identity ranging from 36-60%, depending on the subunit. As with Mtb. thermoautotrophicus  $\Delta H$ , a thermophilic member of the Methanobacteriales, the putative holoenzyme of  $F_{420}$   $H_2$ ase from Msph. stadtmanae is also composed of three distinct subunits, FrhAGB; FrhD is a peptidase involved in hydrogenase maturation that does not purify with the  $F_{420}$   $H_2$ ase (Fricke et al. 2006).

The [NiFe]-containing subunit, FrhA ( $\alpha$ ), is 60% identical (240/406) and 78% similar (317/406) with the homologous subunit from *Mtb. thermoautotrophicus*  $\Delta$ H (E = 5e-153). FrhG ( $\gamma$ ), the subunit containing the e<sup>-</sup> transporting iron-sulfur (FeS) clusters, is 53% identical (123/229) (78% (164/229) similar) to the corresponding subunit of an F<sub>420</sub> H<sub>2</sub>ase from an uncultured methanoarcheon (E = 2e-69). The FAD-containing subunit FrhB ( $\beta$ ) is 55% identical (154/280) (72% (202/280) similar) with the subunit from *Mtb. thermoautotrophicus*  $\Delta$ H (E = 6e-86). FrhD ( $\delta$ ), the hydrogenase maturation peptidase protein, is only 36% identical (55/152) (67% (102/152) similar) to the homologous subunit in *Mtb. thermoautotrophicus*  $\Delta$ H (E = 3e-28).



**Figure 8.11**. Phylogenetic tree showing the evolutionary relationships of the  $F_{420}$ -reducing hydrogenase from various methanoarchaea, including the putative  $F_{420}$  H<sub>2</sub>ase of *Msph. stadtmanae*. ClustalW software (Chenna *et al.* 2003) was used to produce a multiple sequence alignment from the amino acid sequences encoding the α-subunit from the  $F_{420}$  H<sub>2</sub>ase of various methanoarchaea. The sequence alignment was then used to produce the phylogenetic tree with ClustalW, in the Phylip output format.

Multiple sequence alignments of the amino acid sequences coding for the  $\alpha$ -subunit (FrhA) from various methanoarchaea were also used to generate a phylogenetic tree that indicates the close evolutionary relationship between the  $F_{420}$   $H_2$ ase of *Mtb. thermoautotrophicus*  $\Delta H$  (Figure 8.11).

### 8.3.7. Enrichment of the $F_{420}$ -reducing hydrogenase

 $F_{420}$ -reducing activity was monitored during the enrichment of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae using various column chromatography steps (Chapter 7); the results are summarized in Table 8.4. There are some signs indicating the enrichment of the  $F_{420}$ -reducing activity. The specific activity of the  $F_{420}$   $H_2$ ase activity increased, as did the degree of purification, during progression from the homogenized 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet through the Phenyl Sepharose CL-4B and DEAE Sephacel column chromatography steps.

Hydrogenase and F<sub>420</sub>H<sub>2</sub> dehydrogenation activities co-eluted from the Phenyl Sepharose CL-4B column when washed with (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>-free buffer (Figure 7.7A); at this point, the F<sub>420</sub> H<sub>2</sub>ase activity was enriched 5-fold compared to the 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> saturated protein pellet activity. An examination of lane 5 of the activity- stained native PAGE (Figure 8.12B) indicates the presence of a band containing hydrogenase activity with a molecular weight of approximately 232 kDa; this band displays greater activity staining intensity than the clarified cell-free extract (lane 3) or the re-suspended 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet (lane 4); the corresponding lane of the coomassie blue-stained native PAGE also indicates the presence of a protein band that similarly

Table 8.4. Progression of the  $F_{420}$ -reducing hydrogenase activity from Methanosphaera stadtmanae through various column chromatography steps (reprinted from Chapter 7). Protein reactivated with 10  $\mu$ M FAD and 10  $\mu$ M F<sub>420</sub> prior to assay.  $[F_{420}] = 25-30 \mu M.$ 

Purification stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Specific <sup>‡</sup> activity	Purification (-fold)	
Clarified cell- free extract	979 7		100	0.21	-	
75% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> pellet	583.2	235.2	121.3	0.40	1.9	
Phenyl-Sepharose CL-4B	20.4	41.4	21.4	2.03	9.7	
DEAE Sephacel <sup>‡‡</sup>	6.1	13.5	7.0	2.21	10.6	
F <sub>420</sub> -affinity <sup>‡‡</sup>	$\mathbf{F}_{420}$ -affinity <sup>‡‡</sup> 2.1		1.1	1.00	4.8	

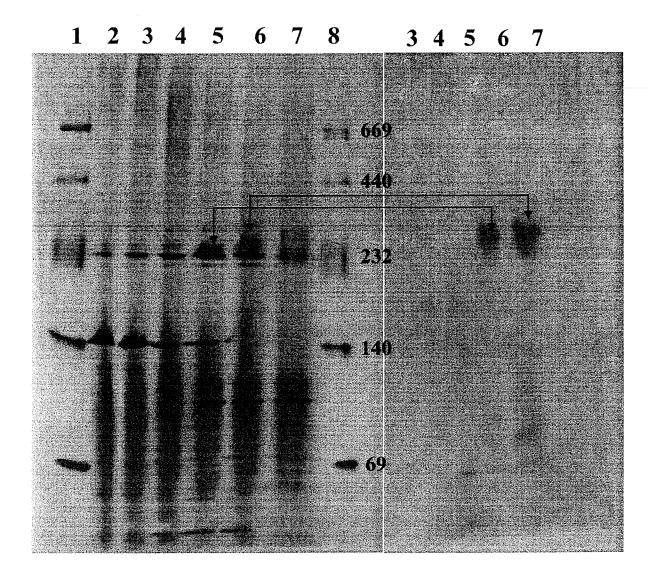
<sup>†</sup>μmol F<sub>420</sub> reduced·min<sup>-1</sup>

shows a greater stained intensity relative to the previous purification stage (Figure 8.12A).

The pooled fractions collected from the Phenyl Sepharose CL-4B column were then loaded onto a DEAE Sephacel column; the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity did not have any apparent affinity for the matrix and flowed through the column uninhibited (Figure 7.7B, Chapter 7). A significant portion of the hydrogenase activity remained bound to the DEAE Sephacel column, while a fraction (~30%) of the hydrogenase activity passed through the column without binding to the column. The protein content of the non-binding hydrogenase activity decreased approximately 3.5-fold from the

<sup>‡</sup>µmol F<sub>420</sub> reduced·min<sup>-1</sup>·mg<sup>-1</sup> ‡‡ protein pooled from non-binding fractions

A. B.



**Figure 8.12**. (A.) Coomassie blue- and (B.) activity stained (using methyl viologen) 3-20% native gradient PAGE showing enrichment of hydrogenase activity at ~232 kDa at various purification stages of the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of *Methanosphaera stadtmanae*. 1, 8. Molecular markers, kDa. 2. Cell-free extract (CFE). 3. Clarified CFE . 4. Re-suspended (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet. 5. Phenyl Sepharose CL-4B pool. 6. DEAE Sephacel pool. 7. F<sub>420</sub>-affinity column pool. Approximately 90 μg of protein loaded per lane.

amount loaded onto the column, with a 10% increase in the specific activity of the  $F_{420}$   $H_2$ ase activity (Table 8.4). The 232 kDa band (lane 6, Figure 8.12A) displays greater coomassie blue-staining intensity compared to the Phenyl Sepharose CL-4B band; the corresponding hydrogenase activity-stained native PAGE also shows an increase in staining intensity at ~232 kDa (Figure 8.12B).

Finally, the pooled non-binding protein collected from the DEAE Sephacel column was loaded onto an  $F_{420}$ -affinity column, where neither the  $F_{420}H_2$  dehydrogenation activity nor the hydrogenase activity had any significant affinity for the matrix-bound  $F_{420}$ , and co-elute from the column in the non-binding fractions (Figure 7.8, Chapter 7);  $F_{420}$   $H_2$  as activity was also detected in the pooled fractions. Approximately 10-20% of the hydrogenase activity remained bound to the  $F_{420}$ -affinity column and was eluted using the 0-2M NaCl gradient, although no  $F_{420}$ -reducing activity was detected in the eluate. The amount of protein isolated at this step, (measured from the pooled, non-binding fractions) decreases 3-fold, but the specific activity of the  $F_{420}$   $H_2$  as activity decreases by more than 50%. Furthermore, activity- and coomassie blue-staining of the 232 kDa bands is greatly reduced, with a significant loss in staining intensity (Figure 8.12, lane 7).

8.3.7.1.  $F_{420}$ -reducing hydrogenase activity vs  $F_{420}$ -nonreducing hydrogenase activity Table 8.5 summarizes the enrichment of the MV H<sub>2</sub>ase activity of Msph. stadtmanae. The  $F_{420}$  H<sub>2</sub>ase can reduce  $F_{420}$  as well as artificial electron acceptors such as methyl or benzyl viologen; the MV H<sub>2</sub>ase can not reduce  $F_{420}$ , but is assayed using methyl or benzyl viologen. Based on genomic evidence, Msph. stadtmanae is expected

Table 8.5. Progression of the F<sub>420</sub>-nonreducing hydrogenase activity from Methanosphaera stadtmanae through various column chromatography steps.  $[methyl\ viologen] = 2\ mM$ 

Stage	Protein (mg)	Activity <sup>†</sup>	Yield (%)	Sp. activity <sup>‡</sup>	Purification
Clarified cell-free extract	929.2	4181.4	100	4.5	-
75% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> pellet	583.2	2755.2	65.9	4.7	1.0
PhenylSepharose CL-4B	20.4	663.8	15.8	32.5	7.2
DEAE Sephacel <sup>‡‡</sup>	6.1	52.2	1.2	8.6	1.9
F <sub>420</sub> -affinity <sup>‡‡</sup>	2.1	2.4	0.06	1.1	0.3

to possess the  $F_{420}$   $H_2$ ase and MV  $H_2$ ase as separate and distinct enzyme activities (Fricke et al. 2006).

Enrichment of the MV  $H_2$  as activity did not progress at the same rate as the  $F_{420}$ H<sub>2</sub>ase activity. Figure 7.7B (Chapter 7) shows there are two peaks of methyl viologenreducing activity that elute from the DEAE Sephacel column; the first peak, collected with the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity, elutes in the void volume, before commencement of the linear NaCl gradient; the pooled fractions contain  $F_{420}$   $H_2$ ase activity. The second peak of methyl viologen-reducing activity elutes during the application of the linear 0-2M NaCl gradient, without apparent F<sub>420</sub>H<sub>2</sub> dehydrogenation or  $F_{420}$   $H_2$ ase activity. The yield of the MV  $H_2$ ase hydrogenase activity (MV  $H_2$ ase +  $F_{420}$ 

<sup>†</sup> μmol methyl viologen reduced·min<sup>-1</sup> ‡μmol methyl viologen reduced·min<sup>-1</sup>·mg<sup>-1</sup> protein pooled from non-binding fractions

 $H_2$ ase) measured from the pooled non-binding fractions dropped nearly 12-fold; by comparison, the  $F_{420}$   $H_2$ ase activity decreased  $\sim$ 3-fold relative to the activity loaded onto the column.

The MV  $H_2$ ase activity of the pooled non-binding fractions collected from the  $F_{420}$ -affinity column decreased 22-fold, whereas the  $F_{420}$ -reducing activity decreased 6-fold. The protein and activity profile of fractions collected from the  $F_{420}$ -affinity column (Figure 7.8, Chapter 7) shows that there are two separate peaks of methyl viologen-reducing hydrogenase activity. The major hydrogenase activity peak (80-90%) elutes in the void volume (i.e. non-binding fractions) along with the  $F_{420}$ -affinity column activity, while the other MV  $F_{20}$ -activity peak elutes from the  $F_{420}$ -affinity column within the linear NaCl gradient. The non-binding protein fractions contained  $F_{420}$   $F_{20}$ -activity along with  $F_{420}$ -activity dehydrogenation activity, while the NaCl-eluted fractions from the  $F_{420}$ -affinity column contained MV  $F_{20}$ -activity only. Table 8.6 highlights the change in ratio of the methyl viologen-reducing hydrogenase activity to  $F_{420}$ -ase activity as the protein extract progresses through the column chromatography steps described above, indicating the enrichment of the  $F_{420}$ -ase activity relative to the methyl viologen-reducing activity.

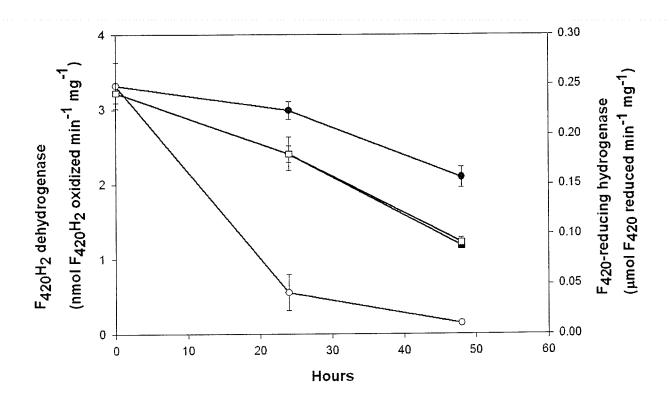
The results indicate a consistently observed trend, a steady increase in the  $F_{420}$   $H_2$ ase activity relative to the MV  $H_2$ ase activity, demonstrating an enrichment of the  $F_{420}$   $H_2$ ase, and separation of the MV  $H_2$ ase throughout the various chromatographic stages.

Table 8.6. Ratio of the  $F_{420}$   $H_2$  as activity to the MV  $H_2$  as activity and  $F_{420}H_2$  dehydrogenation activity of *Methanosphaera stadtmanae* after various column chromatographic stages.  $F_{420}$ - $H_2$  as activated with 10  $\mu$ M FAD and 10  $\mu$ M  $F_{420}$  prior to assay. MV  $H_2$  as activated with 0.05 mM MV prior to assay. All assays performed at room temperature.

Purification stage	F <sub>420</sub> H <sub>2</sub> ase/MV H <sub>2</sub> ase	$F_{420}H_2$ ase/ $F_{420}H_2$ De $H_2$ ase		
Cell-free extract	0.046	58.8		
75% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> pellet	0.085	55.6		
Phenyl Sepharose CL-4B	0.062	90.9		
DEAE Sephacel	0.259	166.7		
F <sub>420</sub> -affinity	0.875	142.9		

#### 8.3.7.2. $F_{420}$ -reducing hydrogenase vs $F_{420}H_2$ dehydrogenation activity

In cell-free extract, the  $F_{420}$  H<sub>2</sub>ase activity is ~60-fold greater than the phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activity, and the ratio of  $F_{420}$  H<sub>2</sub>ase activity to  $F_{420}$ H<sub>2</sub> dehydrogenation activity increases as the protein progresses through the column chromatography steps, although not at the same pace as the  $F_{420}$  H<sub>2</sub>ase:MV H<sub>2</sub>ase ratio (Table 8.6). This may suggest partial separation of the  $F_{420}$  H<sub>2</sub>ase activity from the  $F_{420}$ H<sub>2</sub> dehydrogenation activity, which was not observed during the purification of the  $F_{420}$ H<sub>2</sub> dehydrogenation activity from *Msp. hungatei* GP1 and *Ms. barkeri* Fusaro, where a constant  $F_{420}$  H<sub>2</sub>ase: $F_{420}$ H<sub>2</sub> dehydrogenation ratio was maintained throughout the process.



**Figure 8.13**. Comparison of the stability of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity and the  $F_{420}$ -reducing hydrogenase activity in the cell-free extract of *Methanosphaera stadtmanae*, under aerobic and anaerobic conditions. Activities were assayed over a 48 hour period. Aerobic extract was made anaerobic with  $N_2$  prior to assay.  $F_{420}$   $H_2$  ase activity reactivated with  $10~\mu M$  FAD and  $10~\mu M$   $F_{420}$  in reaction buffer prior to assay.  $\bullet$ ,  $F_{420}H_2$  dehydrogenation activity (anaerobic).  $\circ$ ,  $F_{420}H_2$  dehydrogenation activity (aerobic).  $\circ$ ,  $F_{420}$ -reducing hydrogenase activity (aerobic).

As with other methanogenic  $F_{420}$   $H_2$ ases, the  $F_{420}$   $H_2$ ase activity of Msph. stadtmanae is stable under aerobic or anaerobic conditions (Figure 8.13); aerobic samples are flushed with  $N_2$ , and then incubated in the reaction buffer, under reducing conditions with FAD, prior to assay. In contrast, and unexpectedly, the  $F_{420}H_2$  dehydrogenation activity is  $O_2$  labile, with approximately 75-95% lost after 24 hours (Figure 8.13); the enriched  $F_{420}$   $H_2$ ase is more air stable, losing only 50% of its activity after 48 hours, which is also observed for the anaerobically stored activity.

These observations differ from the observations with  $Msp.\ hungatei$  GP1 and  $Ms.\ barkeri$  Fusaro, where both the  $F_{420}$  H<sub>2</sub>ase and phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activities are aerobically stable.

#### 8.4. Discussion

## 8.4.1. Discovery of an $F_{420}$ -reducing hydrogenase activity in Methanosphaera stadtmanae

As a methanoarchaeon growing solely on H<sub>2</sub> and CH<sub>3</sub>OH, *Msph. stadtmanae* does not metabolize CH<sub>3</sub>OH in the same manner as the methylotrophic members of the *Methanosarcinales* are able to do, such that F<sub>420</sub>H<sub>2</sub> cannot be produced using the methylotrophic pathway. There is no apparent need for F<sub>420</sub>H<sub>2</sub> during methanogenesis on CH<sub>3</sub>OH and H<sub>2</sub>, and unlike most members of the *Methanobacteriales*, *Msph. stadtmanae* does not metabolize CO<sub>2</sub> as substrate for methanogenesis: metabolism of CO<sub>2</sub> requires F<sub>420</sub>H<sub>2</sub> for the reduction of C<sub>1</sub>-bound H<sub>4</sub>MPT intermediates to the level of CH<sub>3</sub>-H<sub>4</sub>MPT (Miller and Wolin 1985, Thauer *et al.* 1993, Shima *et al.* 2002).

Msph. stadtmanae is also capable of producing CH<sub>4</sub> from H<sub>2</sub> and L-serine, although attempts to grow Msph. stadtmanae on H<sub>2</sub> and serine were unsuccessful (Lin

and Sparling 1995, 1998). The hydroxymethyl group of L-serine forms  $N^5, N^{10}$ methylene-H<sub>4</sub>MPT; this would imply that F<sub>420</sub>H<sub>2</sub> would be required for the reduction of  $N^5, N^{10}$ -methylene-H<sub>4</sub>MPT to methyl-H<sub>4</sub>MPT as in other hydrogenotrophic methanoarchaea growing on H<sub>2</sub> and CO<sub>2</sub> (Thauer *et al.* 1993, Lin and Sparling 1995). In the absence of H<sub>2</sub>, 2-propanol can be used as a source of electrons for the reduction of CH<sub>3</sub>OH; 2-propanol is oxidized to provide electrons for the reduction of NADP<sup>+</sup> to NADPH. Fno then catalyzes the NADPH-dependent reduction of F<sub>420</sub>, with the F<sub>420</sub>H<sub>2</sub> formed expected to be used for the reduction of CoM-S-S-CoB (Wong *et al.* 1994).

The low levels of  $F_{420}H_2$  found in *Msph. stadtmanae* (0.16 nmol  $F_{420}$ ·mg<sup>-1</sup> protein) would indicate that the cofactor is likely used for biosynthetic purposes, in conjunction with the Fno, and possibly for the detoxification of  $O_2$  (Wong *et al.* 1994, Elias *et al.* 2000, Fricke *et al.* 2006).

Until recently, the means of F<sub>420</sub>-reduction in *Msph. stadtmanae* were largely unknown, since the F<sub>420</sub> H<sub>2</sub>ase had not been detected in this methanoarchaeon (Deppenmeier *et al.* 1989, van de Wijngaard *et al.* 1991, Wong *et al.* 1994). *Msph. stadtmanae* possesses formate dehydrogenase (Lin and Sparling 1995), but this activity is not F<sub>420</sub>-dependent as it is in some methanoarchaea such as *Methanobacterium* (*Mb.*) *formicicum* (Schauer and Ferry 1985). Interest in the process of H<sub>2</sub>-dependent F<sub>420</sub>-reduction in *Msph. stadtmanae* was reinvigorated by the observation of a phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity in the cell-free extract of *Msph. stadtmanae*, albeit at low levels of activity (Wong, Ph.D thesis 1999).

As part of the enrichment protocol for the  $F_{420}H_2$  dehydrogenation activity (Chapter 7), passage of cell-free extract through an  $F_{420}$ -affinity column (in an anaerobic

chamber containing  $10\%~H_2$ ,  $90\%~N_2$ ) resulted in reduction of the  $F_{420}$  bound to the matrix material. In the absence of protein, the matrix-bound  $F_{420}$  is not reduced in the anaerobic chamber. It was not clear if  $F_{420}$  was being reduced directly from  $H_2$ , or if there was another electron carrier acting as an intermediate between  $H_2$  and  $F_{420}$ . Follow up studies demonstrated that  $F_{420}$ -reduction occurred at the same rate in the presence/absence of mediators such as phenazine or ferredoxin (Table 7.5, Chapter 7).

# 8.4.2. Reactivation of the $F_{420}$ -reducing hydrogenase activity of Methanosphaera stadtmanae

The  $F_{420}$  H<sub>2</sub>ase activity of *Msph. stadtmanae*, located primarily in the clarified cell-free extract (85-90%), required reactivation under reducing conditions, with the addition of FAD and  $F_{420}$ , to achieve optimal activity. Addition of FAD appears to be particularly important in the activation of the  $F_{420}$  H<sub>2</sub>ase (and  $F_{420}$ H<sub>2</sub> dehydrogenation) activity of *Msph. stadtmanae*; in the absence of incubation with FAD and  $F_{420}$ , the  $F_{420}$  H<sub>2</sub>ase activity was approximately 40- to 50-fold lower compared to the reactivated activity (Table 8.1). The low  $F_{420}$ -reducing activity observed in the absence of reactivation is consistent with the observations made by Wong *et al.* (1994).

The need for exogenous FAD for stability is not unique to the putative  $F_{420}$  H<sub>2</sub>ase of *Msph. stadtmanae*, but appears to be required at a level that is distinctive for this methanoarchaeon relative to  $F_{420}$  H<sub>2</sub>ase from other methanoarchaea. FAD is not typically used in the reductive reactivation protocols for the  $F_{420}$  H<sub>2</sub>ase activities of *Mtb*. *thermoautotrophicus*, *Mc. voltae*, or *Ms. barkeri* Fusaro (Fox *et al.* 1987, Muth *et al.* 1987, Fiebig and Friedrich 1989, Michel *et al.* 1995), but is included for the  $F_{420}$  H<sub>2</sub>ase activity of *Msp. hungatei* GP1 (Choquet and Sprott 1991).

FAD is included under some conditions. Nelson *et al.* (1984) reported that the FAD moiety dissociated from the  $F_{420}$  H<sub>2</sub>ase of *Mb. formicicum* during anaerobic hydrophobic column chromatography, resulting in loss of  $F_{420}$ -reducing activity that was restored when the protein was incubated with FAD. Addition of FAD also prevents loss of activity from heat treated  $F_{420}$  H<sub>2</sub>ase of *Mc. voltae*; in the absence of added FAD, only the methyl viologen-reducing activity remained (Muth *et al.* 1987).

The pH of the assay medium also appears to be important, as  $F_{420}$  H<sub>2</sub>ase activity is not detected using 50 mM Tris/Cl buffer, pH 8.0; while this pH is often used to assay methyl viologen-dependent  $F_{420}$  H<sub>2</sub>ase and MV H<sub>2</sub>ase activity (Fiebig and Friedrich 1989, Wong PhD thesis 1999), the pH for measurement of  $F_{420}$ -reduction is typically lower, from pH 7.0–7.5 (Fox *et al.* 1987, Muth *et al.* 1987, Fiebig and Friedrich 1989, Choquet and Sprott 1991). Our results indicate a pH optimum of approximately 6.8, with low activity at pH 8.0 (Figure 4). Ultimately, the buffer modified from Choquet and Sprott (1991), consisting of 100 mM HEPES, pH 7.0, with 200 mM KCl, 20 mM MgCl<sub>2</sub>·6H<sub>2</sub>O,  $10 \mu$ M FAD,  $10 \mu$ M F<sub>420</sub>, and  $20 \mu$ M DTT, was used to assay the F<sub>420</sub> H<sub>2</sub>ase activity of *Msph. stadtmanae*.

## 8.4.3. Electrophoretic properties of the $F_{420}$ -reducing hydrogenase

The F<sub>420</sub> H<sub>2</sub>ases typically exhibit at least two distinct F<sub>420</sub>-reducing forms on activity-stained native PAGE gels (small form: 89–115 kDa, large form(s): 600-1300 kDa); the small forms aggregate to form the larger macromolecular complexes (Michel *et al.* 1995). The F<sub>420</sub> H<sub>2</sub>ase of *Msp. hungatei* GP1 has a MW of 720 kDa (Figure 5.8, Sprott *et al.* 1987), while *Ms. barkeri* Fusaro exhibits a large heteromultimeric form at

845 kDa, and a smaller band at 198 kDa (Figure 6.5, Fiebig and Friedrich 1989). Electrophoresis of clarified cell free extract of *Msph. stadtmanae* on native PAGE gels did not yield the expected activity patterns as observed with other methanoarchaea (Figure 1). It seems unusual that the pattern formed by the F<sub>420</sub> H<sub>2</sub>ase activity from *Msph. stadtmanae* would deviate so drastically from the pattern observed from other methanoarchaea.

Lower MW bands containing hydrogenase activity are routinely seen at approximately 69 kDa; this is consistent with the presence of the MV H<sub>2</sub>ase described by Denny Wong (PhD thesis 1999). However, a distinct activity-stained band, with a molecular weight of approximately 232 kDa, is also observed on the same gels.

The aerobically purified  $F_{420}$   $H_2$  ase from Mtb. thermoautotrophicus  $\Delta H$  is heterotrimeric flavoprotein with an overall MW of 800 kDa, consisting of three subunits;  $\alpha$  (47 kDa),  $\beta$  (31 kDa), and  $\gamma$  (26 kDa) (Fox et al. 1987, Braks et al. 1994). In addition to the 800 kDa band, which contains the majority of the  $F_{420}$ -reducing activity, activity-staining also reveals the presence of a minor band at 115 kDa which also contained  $F_{420}$ -reducing activity; a similar profile was observed when the activity was purified under anaerobic conditions (Fox et al. 1987). Based on the presence of the 115 kDa form, it was suggested that the subunit stoichiometry ( $\alpha$ : $\beta$ : $\gamma$ ) was a 1:1:1 ratio, with an overall MW of 105 kDa (Fox et al. 1987). A fourth subunit, FrhD ( $\delta$ -, 17 kDa), is encoded by the genome, but is not found with the purified holoenzyme after separation of the subunits on SDS PAGE (Fox et al. 1987). Proteins (10-20 kDa) that link the  $F_{420}$   $H_2$  ase to the cell membrane have been observed in a related methanoarchaeon, Mtb. marburgensis, but have not been isolated (Braks et al. 1994).

BLAST analyses of the Archaea database, using the sequences encoding the putative  $F_{420}$  H<sub>2</sub>ase of *Msph. stadtmanae*, and multiple sequence alignments of the sequences encoding the  $\alpha$ -subunit of the  $F_{420}$  H<sub>2</sub>ase from various methanoarchaea, indicate that the putative  $F_{420}$  H<sub>2</sub>ase of *Msph. stadtmanae* is most closely related to the homologous enzyme from *Mtb. thermoautotrophicus*  $\Delta H$  (see section 8.3.5). The MW of the  $\alpha$ -,  $\beta$ -, and  $\gamma$ -subunits of the putative  $F_{420}$  H<sub>2</sub>ase from *Msph. stadtmanae*, estimated using the Compute pI/MW software (Bjellqvist *et al.* 1993, 1994), are, respectively, 45, 33, and 28 kDa (overall MW: 106 kDa). As the  $F_{420}$  H<sub>2</sub>ase of *Msph. stadtmanae* is most closely related to the enzyme from *Mtb. thermoautotrophicus*  $\Delta H$  (section 8.3.5.), the observed 232 kDa band may be representative of a dimer of two ~115 kDa  $F_{420}$  H<sub>2</sub>ases in *Msph. stadtmanae*.

It is interesting to note that the yellow-brown colored bands typically associated with the  $F_{420}$   $H_2$ ase cannot be observed on the unstained gel. One explanation is that the protein may be more susceptible to loss of FAD relative to other  $F_{420}$   $H_2$ ases from other methanoarchaea. Indeed, our findings reveal a requirement for FAD during reactivation of the  $F_{420}$   $H_2$ ase activity of Msph. stadtmanae (Table 8.1).

## 8.4.4. Inhibitor analysis of the $F_{420}$ -reducing hydrogenase activity

Several known inhibitors of hydrogenase activity were used to help elucidate whether the observed  $F_{420}$ -reduction in cell-free extract was associated with a distinct hydrogenase (Table 8.2).

Divalent metal ions are strong inhibitors of hydrogenase activity in various microorganisms (Woo *et al.* 1993).  $F_{420}$ -reducing activity in cell-free extract of *Msph*.

stadtmanae was inhibited by the addition of various metal ions, most significantly by  $Cu^{2+}$ .  $Cu^{2+}$ , when used as a potential ligand for MCAC, rendered the  $F_{420}$   $H_2$ ase of Msp. hungatei GP1 inactive (Choquet and Sprott 1991); inhibition was also observed after the  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae was passed through a  $Cu^{2+}$ -charged column (Appendix E).

F<sub>420</sub>-reduction was stimulated in the presence of the anionic metabolic inhibitors N<sub>3</sub><sup>-</sup> and CN<sup>-</sup>, as well as in the presence of the metal chelating agent EDTA. Azide (N<sub>3</sub><sup>-</sup>) and cyanide (CN<sup>-</sup>) are both electron transport inhibitors that also inhibit hydrogenases (Seefeldt and Arp 1989), while EDTA chelates divalent cations, including Ni<sup>2+</sup> which is an essential component of [NiFe] hydrogenases (Friedrich *et al.* 1981).

2-Bromoethanesulfonate (2-BES), a methanogenic inhibitor with a similar structure to CoM-SH, was used as a negative control, and does not have any significant impact on the F<sub>420</sub>-reducing activity. 2-BES inhibits the methylreductase, which catalyzes the reductive demethylation of CH<sub>3</sub>-S-CoM with CoB-SH, forming CoM-S-S-CoB and CH<sub>4</sub> (DiMarco *et al.* 1990). These results are consistent with the inhibition/ stimulation observed when measuring methyl viologen-reducing activity in the cell-free extract of *Msph. stadtmanae* (Wong, Ph.D thesis 1999).

Addition of DPI also inhibits  $F_{420}$ -reduction from the cell-free extract of Msph. stadtmanae. DPI likely binds to the FAD component of the  $\beta$ -subunit, preventing access of  $F_{420}$  to the active site. Inhibition of  $F_{420}$  H<sub>2</sub>ase activity is consistent with the observed inhibition of the phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activity of Msph. stadtmanae, indicating that attack of the FAD component by DPI is the likely basis for

inhibition of both activities in *Msph. stadtmanae*, suggesting non-competitive inhibition. DPI is an inhibitor of flavoproteins such as mitochondrial and bacterial NADH: ubiquinone oxidoreductase and NADH oxidase (O'Donnell *et al.* 1993, Majander *et al.* 1994), as well as the 2-OH-phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of *Ms. mazei* Gö1 (Brodersen *et al.* 1999). DPI acts as a competitive inhibitor to the 2-OH-phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of *Ms. mazei* Gö1, due to its planar structure which is similar to phenazine (Figure 1.15) (Brodersen *et al.* 1999). Methyl viologen-reduction in *Msph. stadtmanae* is not affected by the presence of DPI, as FAD is not necessary for methyl viologen-reduction by the F<sub>420</sub> H<sub>2</sub>ase (Thauer *et al.* 1993). This is consistent with our observations with the corresponding activity of *Msp. hungatei* GP1 (Table 5.7, Chapter 5) and *Ms. barkeri* Fusaro (Table 6.7, Chapter 6)

For comparison, the purified  $F_{420}H_2$  dehydrogenation activity (associated with the  $F_{420}$   $H_2$ ase) of Msp. hungatei GP1 was inhibited by DPI (25 and 50  $\mu$ M), while the  $F_{420}$   $H_2$ ase was inhibited by 50  $\mu$ M DPI (Chapter 5), indicating competitive inhibition at lower [DPI] (25  $\mu$ M), but non-competitive at higher [DPI] (50  $\mu$ M). The purified  $F_{420}H_2$  dehydrogenation activity (associated with the  $F_{420}$   $H_2$ ase) of Ms. barkeri Fusaro was competitively inhibited by additions of 50 and 100  $\mu$ M DPI, while the purified  $F_{420}$   $H_2$ ase activity of Ms. barkeri Fusaro was not inhibited (Chapter 6).

# 8.4.5. Comparison of the $F_{420}$ -reducing hydrogenase activity of Methanosphaera stadtmanae with homologous activities in other methanoarchaea

The catalytic/kinetic properties of the  $F_{420}$   $H_2$ ase activity were examined from clarified cell-free extract of *Msph. stadtmanae*, and compared with the  $F_{420}$   $H_2$ ase activity from other methanoarchaea (Table 8.3). Using an  $F_{420}$   $H_2$ ase assay protocol modified

from Choquet and Sprott (1991) (described in section 8.3.4), the apparent  $K_m$  for  $F_{420}$  of the  $F_{420}$  H<sub>2</sub>ase activity of *Msph. stadtmanae* was found to be similar to the activities observed in other methanogenic archaea, with the exception of *Mtb. marburgensis* (discussed further below). Once reactivated in the presence of FAD, the level of activity was similar to the  $F_{420}$  H<sub>2</sub>ase activities in the clarified cell-free extracts of the methanoarchaea tested. It should be noted that assay conditions for the respective  $F_{420}$  H<sub>2</sub>ase activities, particularly with respect to reactivation, vary from one methanoarchaeon to the next, such that the conditions may not have been optimal for each activity. Standardized conditions (i.e. the activation/assay conditions used for *Msph. stadtmanae*) were used to provide a means for comparison between the  $F_{420}$ -reducing activities of *Msph. stadtmanae* and other methanoarchaea.

While the putative  $F_{420}$   $H_2$  ase of Msph. stadtmanae is most closely related to the enzyme of Mtb. thermoautotrophicus  $\Delta H$ , the observed activities of a closely related methanoarchaeon, Mtb. marburgensis, and Msph. stadtmanae were quite distinct. This is due to the conditions under which the activities were assayed. As described by Jacobson  $et\ al.\ (1981)$ , the  $F_{420}$   $H_2$  ase activity of Mtb.  $thermoautotrophicus\ \Delta H$ , at various stages of purification, is optimized via incubation at 45°C for 1 hour, in buffer containing 1M salt  $(K^+$  or  $Na^+$ ). The apparent  $K_m$  for  $F_{420}$  of the  $F_{420}$   $H_2$  ase of Mtb.  $thermoautotrophicus\ \Delta H$  was determined to be 19  $\mu$ M  $F_{420}$  by Jacobson  $et\ al.\ (1981)$ .

Under our standardized conditions (incubation of protein under  $H_2$ , in 100 mM HEPES, pH 7.0, with 200 mM KCl, 20 mM MgCl<sub>2</sub>·6H<sub>2</sub>O, 10  $\mu$ M FAD, 10  $\mu$ M F<sub>420</sub>, and 20 mM DTT, room temperature), the apparent  $K_m$  for F<sub>420</sub> for *Mtb. marburgensis* was determined to be 2.4x10<sup>6</sup>  $\mu$ M (Michaelis Menten kinetics); however, when the clarified

cell-free extract was incubated at 45°C for ½ hour, the apparent  $K_m$  was determined to be 36.5  $\mu$ M  $F_{420}$ , which is considerably closer to the value reported for Mtb. thermoautotrophicus  $\Delta H$  by Jacobson et al. (1981). Thus, despite the phylogenetic similarities between the  $F_{420}$   $H_2$ ase of Msph. stadtmanae and Mtb. thermoautotrophicus  $\Delta H$ , there are clearly differences in requirements for activation of the respective  $F_{420}$   $H_2$ ase activities.

## 8.4.6. Partial enrichment of the $F_{420}$ -reducing hydrogenase activity

The F<sub>420</sub> H<sub>2</sub>ase activity of *Msph. stadtmanae* was only recently observed in our laboratory. It is apparent that the protocol used to isolate the F<sub>420</sub> H<sub>2</sub>ase from *Ms. barkeri* Fusaro, *Msp. hungatei* GP1, and *Mc. voltae* cannot be directly applied without some modification. The protocol (described in greater detail in Chapters 5 and 6) involved: enrichment of the F<sub>420</sub>H<sub>2</sub> dehydrogenation/ F<sub>420</sub> H<sub>2</sub>ase activity from a Ni<sup>2+</sup>-affinity column; electrophoresis of the pooled desalted/concentrated protein on a 3-20% native gradient PAGE; and electro-elution of the protein from gel bands excised from the native PAGE.

There is no significant affinity for the  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae* to the Ni<sup>2+</sup>-charged sepharose fast-flow resin (i.e. metal chelate affinity chromatography), such that the  $F_{420}$ -reducing activity cannot be enriched using this step. A distinct, high MW band consistent with the presence of  $F_{420}$   $H_2$ ase is not visualized on the native gradient PAGE gel after electrophoresis of protein (over 1 mg of protein loaded) from *Msph. stadtmanae*, in the absence or presence of activity stain. This was an important step in the isolation of the  $F_{420}$   $H_2$ ases of methanoarchaea studied in this

thesis, since the  $F_{420}$   $H_2$ ase can be visualized (without staining) as a distinct, high MW, yellow-brown protein band separated from most other proteins. A protein band of similar high MW can only be visualized after coomassie blue-staining of the native PAGE, this band does not stain for hydrogenase activity (Figures 8.2 and 8.3).

Phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity is associated with the F<sub>420</sub> H<sub>2</sub>ase of *Msp. hungatei* GP1 and *Ms. barkeri* Fusaro; it is possible that a similar association could be observed in *Msph. stadtmanae*. F<sub>420</sub> H<sub>2</sub>ase activity was monitored during the enrichment of the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae*. The F<sub>420</sub> H<sub>2</sub>ase activity was enriched after several chromatographic steps (Table 8.4); this is corroborated by the enrichment of hydrogenase activity- and coomassie blue-stained bands of approximately 232 kDa on a 3-20% native gradient PAGE (Figure 8.12). The F<sub>420</sub> H<sub>2</sub>ase activity appeared to be distinct from the MV H<sub>2</sub>ase activity, since the ratio of MV H<sub>2</sub>ase to F<sub>420</sub> H<sub>2</sub>ase activity increased throughout the chromatographic progression (Table 8.6). However, the F<sub>420</sub> H<sub>2</sub>ase also appeared to be distinct from the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity, since the ratio of F<sub>420</sub> H<sub>2</sub>ase to F<sub>420</sub>H<sub>2</sub> dehydrogenation activity, since the ratio of F<sub>420</sub> H<sub>2</sub>ase to F<sub>420</sub>H<sub>2</sub> dehydrogenation activity increased throughout the chromatographic progression (Table 8.6).

Present research indicates that there is an  $F_{420}$  H<sub>2</sub>ase activity in the cell-free extract of *Msph. stadtmanae*; activity at similar levels as other methanoarchaea was observed when the cell-free extract was incubated, under reducing conditions, with 10  $\mu$ M FAD and 10  $\mu$ M F<sub>420</sub>, at neutral pH (pH 7.0) and ambient room temperature (~25°C) (Table 8.4). Table 8.7 highlights some of the similarities and differences of the putative

Table 8.7. Comparison of properties of the putative  $F_{420}$ -reducing hydrogenase of Methanosphaera stadtmanae with the  $F_{420}$ reducing hydrogenase of various methanoarchaea.

Methanoarchaeon	Order	FAD requirement	pH optimum	Km (μΜ F <sub>420</sub> )	Molecular mass (kDa)	Subunit composition (kDa)	Reference
Ms. barkeri Fusaro	Methanosarcinales	No	6.5-7.25	25	845, 198	48, 33, 30	Fiebig and Friedrich 1989, Michel et al. 1995
Msp. hungatei GP1	Methanomicrobiales	No <sup>a</sup>	7.0	N.D.	720	50.7, 32.9, 30.7	Sprott <i>et al.</i> 1987, Choquet and Sprott 1991
Mc. voltae	Methanococcales	No <sup>b</sup>	7.0	16	1300, 745, 105	55 <sup>c</sup> , 45, 37, 27	Muth <i>et al</i> . 1987
Mtb. thermoautotrophicus ΔΗ	Methanobacteriales	No	6.5-7.5	26	800, 115	47, 31, 26	Fox <i>et al</i> . 1987, Livingston <i>et al</i> . 1987
Msph. stadtmanae <sup>d</sup>	Methanobacteriales	Yes <sup>e</sup>	6.8 <sup>f</sup>	53	232 <sup>d</sup>	(45,33,28) <sup>g</sup>	This thesis

<sup>&</sup>lt;sup>a</sup> FAD used but not needed, according to Sprott et al. 1987, Choquet and Sprott 1991

<sup>&</sup>lt;sup>b</sup> FAD added after hydrophobic chromatography

<sup>&</sup>lt;sup>c</sup> dimer of 27 kDa subunits

d putative F<sub>420</sub> H<sub>2</sub>ase

e 50- to 60-fold lower activity in the absence of FAD

f no activity detected in Tris/Cl pH 8.0, absence of FAD

g estimated via analysis of frhAGB using Compute pI/MW software

N.D., not determined in original references

F<sub>420</sub> H<sub>2</sub>ase of *Msph. stadtmanae* with the corresponding enzyme from various methanoarchaea.

In summary, the  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae was investigated as a possible source of phenazine-dependent  $F_{420}$ -reducing activity, in the apparent absence of an  $F_{420}$   $H_2$ ase. The relationship between the  $F_{420}H_2$  dehydrogenation activity and the  $F_{420}$   $H_2$ ase activity of Msph. stadtmanae has yet to be resolved. While more work is required to isolate and to characterize this putative enzyme, the observation of an  $F_{420}$   $H_2$ ase activity in Msph. stadtmanae is significant.

### 8.4.7. Resolution of the $F_{420}H_2$ dehydrogenation, $F_{420}$ -reducing hydrogenase, and $F_{420}$ -nonreducing hydrogenase activities

While the  $F_{420}$  H<sub>2</sub>ase was not initially believed to be present in *Msph. stadtmanae*, our enrichment of this activity from cell-free extract led to some interesting observations. Table 8.6 indicates that as the cell-free extract of *Msph. stadtmanae* progresses through the various column chromatography stages, the ratio of  $F_{420}$  H<sub>2</sub>ase activity to MV H<sub>2</sub>ase activity increases, indicating a decrease in methyl viologen-reducing activity with a concurrent enrichment of  $F_{420}$ -reducing activity. This is consistent with the presence of genes encoding for the both hydrogenases in the genome of *Msph. stadtmanae* (Fricke *et al.* 2006).

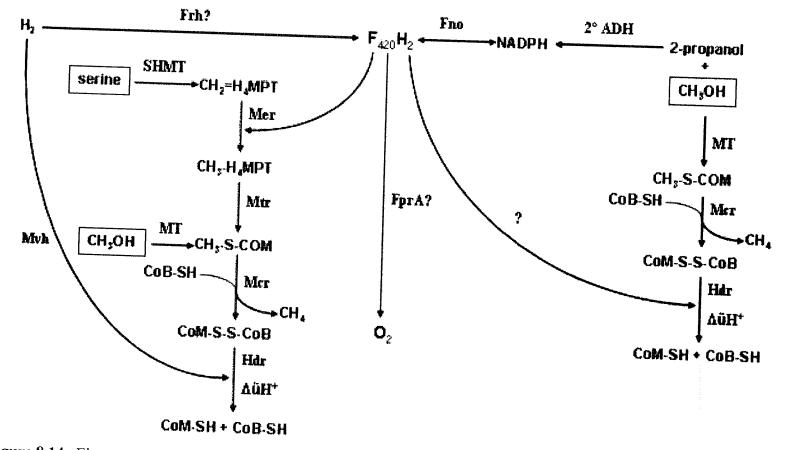
Is the putative  $F_{420}$   $H_2$  ase of Msph. stadtmanae a source of the  $F_{420}H_2$  dehydrogenation activity? Resolution of the  $F_{420}$   $H_2$  ase and  $F_{420}H_2$  dehydrogenation activities is more difficult. As the cell-free extract passes through various column chromatography steps, the  $F_{420}$   $H_2$  ase activity is enriched, while the  $F_{420}H_2$  dehydrogenation activity decreases (Table 8.6). The observed  $F_{420}H_2$  dehydrogenation

activity of Msph. stadtmanae (in cell-free extract) is O2 labile, and activity declines rapidly; the  $F_{420}$   $H_2$  as activity is itself aerobically stable (Figure 8.13). Based on these results, it appears that the F<sub>420</sub> H<sub>2</sub>ase of Msph. stadtmanae is not the sole source of the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity, if at all. In contrast, the phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity purified from Msp. hungatei GP1 and  $\it Ms.\ barkeri\ Fusaro$  is directly associated with the respective  $F_{420}$ -reducing hydrogenases, and is also aerobically stable (Chapter 5 and 6, respectively). However, we also must consider that the differences in the  $F_{420}$   $H_2$  ase:  $F_{420}H_2$  dehydrogenation activity ratios could also be due to loss of the FAD component during the column chromatography process, which was not taken into account during assays of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity. Incubation of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity with FAD was not a typical step during our studies, as the addition of FAD did not increase the specific activity of the F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of Ml. tindarius or Ms. mazei Gö1 (Haase et al. 1992, Abken and Deppenmeier 1997). Our results indicate that incubation of protein of Msph. stadtmanae with FAD can stimulate the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity after apparent loss of activity (Table 7.6). In the absence of further analysis, one cannot discount the possibility that the aerobic instability of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity may be an anomaly of the  $F_{420}$   $H_2$  as of *Msph. stadtmanae*.

A putative  $F_{420}H_2$  oxidase (FprA) is encoded in the genome of *Msph. stadtmanae* (Fricke *et al.* 2006). Similar to the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity, the  $F_{420}H_2$  oxidase activity is also  $O_2$  labile, and  $F_{420}H_2$ -dependent  $O_2$ -reduction declines at a similar pace as the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity when stored under aerobic conditions (Figure 7.11). These results should be interpreted with

caution, as the  $F_{420}$   $H_2$ ase of *Mtb. thermoautotrophicus*  $\Delta H$  also catalyzes the  $F_{420}H_2$ -dependent reduction of  $O_2$  (Jacobson *et al.* 1982).

Recent biochemical and genomic studies in our laboratory have lead to new insights into the electron transport pathway of Msph. stadtmanae, in particular to reactions involving the reduction of  $F_{420}H_2$ . A model highlighting the findings of our recent studies on electron transport in Msph. stadtmanae are shown in Figure 8.14.



**Figure 8.14**. Elements of electron transport in *Methanosphaera stadtmanae*. CH<sub>3</sub>OH, methanol; H<sub>4</sub>MPT, tetrahydromethanopterin; CoM-SH, coenzyme M; CoB-SH, coenzyme B; CoM-S-S-CoB, heterodisulfide complex; Mer, methylene-H<sub>4</sub>MPT reductase; Mtr, methyl-H<sub>4</sub>MPT:coenzyme M methyltransferase; Mt, methyltransferase; Mcr, methyl-coenzyme M reductase; Hdr, heterodisulfide reductase; Mvh, methyl viologen-reducing hydrogenase; Frh, F<sub>420</sub>-reducing hydrogenase (putative); Fno, F<sub>420</sub>H<sub>2</sub>:NADP<sup>+</sup> oxididoreductase; 2° ADH, secondary alcohol dehydrogenase; SHMT, serine hydroxymethyl transferase; FprA, flavoprotein A (F<sub>420</sub>H<sub>2</sub> oxidase, putative).

#### 9. Conclusions and Future Considerations

#### 9.1. Introduction

The F<sub>420</sub>H<sub>2</sub> dehydrogenase (Fpo (F<sub>420</sub>H<sub>2</sub>:phenazine oxidoreductase complex) has been purified from the methylotrophic methanoarchaea *Methanolobus tindarius* and *Methanosarcina mazei* Gö1; this enzyme is expected to be an essential component of the F<sub>420</sub>H<sub>2</sub>:heterodisulfide oxidoreductase complex (Haase *et al.* 1992, Abken and Deppenmeier 1997). Studies with *Ms. mazei* Gö1 indicate that methanophenazine is the purported electron carrier for this activity, but this activity is can also be assayed using various phenazine-containing compounds (Abken *et al.* 1998). While the Fpo complex has only been isolated from methylotrophic members of the *Methanosarcinaceae*, phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity appeared to be ubiquitous amongst the methanoarchaea, independent of growth substrate or phylogeny. This activity was further examined in *Methanospirillum hungatei* GP1, CH<sub>3</sub>OH-grown *Methanosarcina barkeri* Fusaro, and *Methanosphaera stadtmanae*.

Some of the differences and similarities of the methanogenic  $F_{420}H_2$  dehydrogenation activity studied during the course of this thesis are summarized in Table 9.1 (from Chapters 4, 5, 6, and 7). Unlike the activity purified from *Ml. tindarius* (Haase *et al.* 1992), the  $F_{420}H_2$  dehydrogenation activities of *Msp. hungatei* GP1 and *Ms. barkeri Fusaro* are both located on the respective  $F_{420}$   $H_2$ ase; these activities are further distinguished from the activity of Fpo complex of *Ms. mazei* Gö1 (Abken *et al.* 1998) and *Ml. tindarius* (this thesis) by the low sensitivity to DPI. The  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae* is among the lowest of all the activities observed. Unlike

the activity measured from other methanoarchaea, this activity was anaerobically unstable. The source of this activity is not immediately clear.

Table 9.1. Properties of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of methanoarchaea studied in this thesis.

Methano- archaeon	Substrate	Source of activity	Km <sub>phenazine</sub> ‡	Vmax <sup>‡‡</sup>	Ιc <sub>50</sub> (μΜ DPI) <sup>†</sup>	Reference
Ml. tindarius	СН₃ОН	Fpo complex	266	0.5 <sup>†</sup>	3	Haase <i>et al</i> . 1992, This thesis
Msp. hungatei GP1	H <sub>2</sub> /CO <sub>2</sub>	F <sub>420</sub> H <sub>2</sub> ase	288.3	19.5 <sup>††</sup>	25	This thesis
<i>Ms. barkeri</i> Fusaro	СН₃ОН	F <sub>420</sub> -H <sub>2</sub> ase	112.4	29.5 <sup>††</sup>	>100	This thesis
Msph. stadtmanae	H₂/ CH₃OH	$F_{420}$ - $H_2$ ase <sup>†</sup> or $F_{420}H_2$ oxidase <sup>†</sup>	138.2	0.003 <sup>†</sup>	2.5	Fricke <i>et al.</i> 2007, This thesis

measured from cell-free or clarified cell-free extract

## 9.2. Phenazine-dependent $F_{420}H_2$ dehydrogenation activity in Methanospirillum hungatei GP1

An F<sub>420</sub>-nonreducing hydrogenase (MV H<sub>2</sub>ase) was not detected in *Msp. hungatei* GP1 (Sprott *et al.* 1987); this enzyme is an essential component of H<sub>2</sub>:CoM-S-S-CoB oxidoreductase complex in hydrogenotrophic methanoarchaea, catalyzing the H<sub>2</sub>-dependent reduction of CoM-S-S-CoB (Figure 1.16) (Thauer *et al.* 1993, Shima *et al.* 2002). In the absence of MV H<sub>2</sub>ase, alternate means for CoM-S-S-CoB reduction were investigated. Phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity detected in the

<sup>††</sup>purified activity

 $<sup>^{\</sup>ddagger}\mu M$  phenazine

<sup>&</sup>lt;sup>‡‡</sup>µmol F<sub>420</sub>H<sub>2</sub> oxidized·min<sup>-1</sup>·mg<sup>-1</sup>

soluble and membrane fractions was isolated and found associated with the  $F_{420}$ -reducing hydrogenase ( $F_{420}$  H<sub>2</sub>ase), and not with an  $F_{420}$ H<sub>2</sub> dehydrogenase. Biochemical and genomic analyses during the course of this thesis confirmed the absence of an  $F_{420}$ H<sub>2</sub> dehydrogenase and MV H<sub>2</sub>ase in *Msp. hungatei* JF1. The  $F_{420}$  H<sub>2</sub>ase may provide an alternate means for CoM-S-S-CoB reduction in *Msp. hungatei* GP1, using  $F_{420}$ H<sub>2</sub>; a similar mechanism has been suggested for *Methanococcus voltae*, grown on formate (Brodersen *et al.* 1999). Genomic analyses of *Msp. hungatei* JF1 indicate the presence of noncatalytic MV H<sub>2</sub>ase subunits (MvhD) that may allow for direct transfer of electrons from H<sub>2</sub> or  $F_{420}$ H<sub>2</sub> to the heterodisulfide reductase (Hdr) (Figure 5.16).

More studies must be conducted to determine if the F<sub>420</sub> H<sub>2</sub>ase is associated with CoM-S-S-CoB reduction. This could be accomplished using membrane vesicle preparations of *Msp. hungatei* GP1, which would contain both the F<sub>420</sub> H<sub>2</sub>ase and Hdr. Spheroplasts of *Msp. hungatei* GP1 can be formed according to Sprott *et al.* (1987); gentle lysing of these cells would allow for study of the F<sub>420</sub> H<sub>2</sub>ase and Hdr in the cell membrane vesicles. Peinemann *et al.* (1990) were able to demonstrate ATP synthesis using H<sub>2</sub>-dependent reduction of CoM-S-S-CoB using membrane vesicles of *Ms. mazei* Gö1. H<sub>2</sub> or F<sub>420</sub>H<sub>2</sub>-dependent CoM-S-S-CoB reduction, using vesicle preparations of *Msp. hungatei* GP1, could be monitored, as well as H<sup>+</sup> translocation and ATP synthesis.

# 9.3 Phenazine-dependent $F_{420}H_2$ dehydrogenation activity in methanol-grown Methanosarcina barkeri Fusaro

BLAST analyses of the unpublished genome of *Ms. barkeri* Fusaro indicated the presence of genes coding for a putative Fpo complex. However, kinetic profiles of the membrane-bound phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activities of CH<sub>3</sub>OH-

grown  $Ms.\ barkeri$  Fusaro and  $Ml.\ tindarius$  are significantly different, indicating that the respective enzymes are likely dissimilar. Purification of the soluble and membrane-bound  $F_{420}H_2$  dehydrogenation activity of  $Ms.\ barkeri$  Fusaro did not result in the isolation of the  $F_{420}H_2$  dehydrogenase, but instead produced the  $F_{420}$   $H_2$ ase. Our results indicate that the  $F_{420}H_2$  dehydrogenation activity is highest when  $Ms.\ barkeri$  Fusaro is grown with  $CH_3OH$  as methanogenic substrate (Table 4.1), consistent with the findings of Mukhopadhyay  $et\ al.\ (1993)$ , who reported that  $F_{420}\ H_2$ ase activity was highest in  $CH_3OH$ -grown cultures. The  $F_{420}H_2$  dehydrogenation activity of  $Ms.\ barkeri$  Fusaro, associated with the  $F_{420}\ H_2$ ase, shares some similarities with the respective  $F_{420}H_2$  dehydrogenase from  $Ml.\ tindarius$  and  $Ms.\ mazei\ Gö1$ .

While F<sub>420</sub> H<sub>2</sub>ase activity is comparable in *Ms. barkeri* Fusaro (0.034 μmol F<sub>420</sub> reduced·min<sup>-1</sup>) and *Ms. mazei* Gö1 (0.027 μmol F<sub>420</sub> reduced·min<sup>-1</sup>), neither *Ml. tindarius* nor *Methanosarcina acetivorans* have detectable F<sub>420</sub> H<sub>2</sub>ase activity (Deppenmeier *et al.* 1989, Galagan *et al.* 2002). It is not clear why significant levels of F<sub>420</sub> H<sub>2</sub>ase activity would be required in *Ms. barkeri* Fusaro during growth on CH<sub>3</sub>OH, as F<sub>420</sub>H<sub>2</sub> is produced during CH<sub>3</sub>OH disproportionation (Deppenmeier *et al.* 1996). H<sub>2</sub> is produced during methanogenesis of CH<sub>3</sub>OH-grown *Ms. barkeri* Fusaro (Ranalli *et al.* 1986), and it has been suggested that the function of F<sub>420</sub> H<sub>2</sub>ase in *Ms. barkeri* Fusaro may be to regulate levels of F<sub>420</sub>H<sub>2</sub> and H<sub>2</sub> in CH<sub>3</sub>OH-grown cultures, by acting as a redox valve (de Poorter *et al.* 2005). As one of the possible means for CoM-S-S-CoB-reduction during growth of *Ms. barkeri* Fusaro on CH<sub>3</sub>OH, the H<sub>2</sub> produced *via* the F<sub>420</sub> H<sub>2</sub>ase could be obtained by the F<sub>420</sub>-nonreducing hydrogenase (Vho) for CoM-S-S-CoB reduction (Figure 6.12B, Chapter 6), such that an F<sub>420</sub>H<sub>2</sub> dehydrogenase would not be required.

It is not clear why a distinct F<sub>420</sub>H<sub>2</sub> dehydrogenase is not produced under the growth conditions tested. It's possible that the H<sub>2</sub> produced during growth of CH<sub>3</sub>OH-grown *Ms. barkeri* Fusaro may repress the expression of the *fpo* operon; since the cultures are grown in sealed bottles, H<sub>2</sub> would accumulate in the bottles and may be used as an electron source. Reverse transcriptase (RT) PCR could be used to confirm the absence of the Fpo complex in *Ms. barkeri* Fusaro. It is also possible that Fpo is being produced at levels to low for detection by our methods; RT PCR could be used to quantify the level of transcripts produced from the *fpo* operon.

### 9.4. The phenazine-dependent $F_{420}H_2$ dehydrogenation activity and electron transport in Methanosphaera stadtmanae

Our studies with Msph. stadtmanae focused on a phenazine-dependent  $F_{420}H_2$  dehydrogenation activity, in the apparent absence of an  $F_{420}$   $H_2$ ase (Deppenmeier et~al. 1989, Wong et~al. 1994, this thesis). This  $F_{420}H_2$  dehydrogenation activity was distinct from similar activities in other methanoarchaea, in that the  $F_{420}H_2$  dehydrogenation activity of Msph. stadtmanae is aerobically unstable, and much lower compared to other methanoarchaea (Table 4.1).

Due to its instability, the proteins involved in F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae* could not be fully purified. In cell-free extract, the activity is stable for up to two weeks under anaerobic conditions, but rapidly declines during enrichment *via* column chromatography. The phenazine-dependent F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was partially characterized during the course of study, and sources for this activity in cell-free extract may have been revealed.

An  $F_{420}$  H<sub>2</sub>ase activity was observed using the cell-free extract of *Msph. stadtmanae*, independent of the requirement for phenazine as coupling agent, as previously suggested (Table 7.6) (Wong, PhD thesis 1999). This activity required incubation under reducing conditions in the presence of FAD, at neutral pH. the putative  $F_{420}$  H<sub>2</sub>ase could be a source of phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activity; the corresponding enzyme in *Msp. hungatei* GP1 and *Ms. barkeri* Fusaro contain phenazine-dependent  $F_{420}$ H<sub>2</sub> dehydrogenation activity. However, in contrast to similar activities in other methanoarchaea, the  $F_{420}$ H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae* is  $O_2$  labile; this brings into question whether the observed  $F_{420}$ H<sub>2</sub> dehydrogenation activity is truly associated with the putative  $F_{420}$  H<sub>2</sub>ase of *Msph. stadtmanae*.

Another source of phenazine-dependent  $F_{420}H_2$  dehydrogenation activity may be the putative  $F_{420}H_2$  oxidase. The genome of *Msph. stadtmanae* contains a gene encoding for  $F_{420}H_2$  oxidase, an FMN-containing protein that detoxifies  $O_2$  using  $F_{420}H_2$  (Seedorf *et al.* 2004, Fricke *et al.* 2006); our studies have demonstrated  $O_2$ -dependent  $F_{420}H_2$  oxidation using cell-free extract of *Msph. stadtmanae*. The proteins catalyzing  $F_{420}H_2$ -dependent phenazine-reduction or  $O_2$ -reduction, in cell-free extract, are aerobically unstable, and the respective activities decline at similar rates. While the activity is consistent with the presence of an  $F_{420}H_2$  oxidase, these results must be interpreted with caution; the  $F_{420}H_2$  ase (from *Mtb. thermoautotrophicus*  $\Delta H$ ) also catalyzes the reduction of  $O_2$  with  $F_{420}H_2$  (Jacobson *et al.* 1982). Without further analysis, we cannot conclude that the observed  $F_{420}H_2$ -dependent  $O_2$ -reduction is the property of an  $F_{420}H_2$  oxidase; RT

PCR could be a means to confirm and to quantify the presence of an  $F_{420}H_2$  oxidase in Msph. stadtmanae.

The  $F_{420}$   $H_2$  as activity of Msph. stadtmanae was further examined. The properties of the  $F_{420}$   $H_2$  as activity, in cell-free extract, are similar to the corresponding activities of other methanoarchaea (Table 8.2); genomic analyses indicates that the genes encoding the putative  $F_{420}$   $H_2$  as are most closely related to the frhAGDB operon of Mtb. thermoautotrophicus  $\Delta H$ . There are several features of the  $F_{420}$ -reducing activity that distinguish it from the activities of other methanoarchaea. Electrophoresis of protein extracts of Msph. stadtmanae on native PAGE gels do not indicate the presence of bands typical of an  $F_{420}$   $H_2$  as observed with other methano-archaea (Figure 8.1, Chaper 8).

The recently published genome of *Msph. stadtmanae* contains the gene sequences consistent with the presence of a putative F<sub>420</sub> H<sub>2</sub>ase (Fricke *et al.* (2006). It is not immediately clear if the observed F<sub>420</sub>-reduction is due to the presence of the putative F<sub>420</sub> H<sub>2</sub>ase; RT PCR could be used to identify and to quantify the presence of transcripts coding for the F<sub>420</sub> H<sub>2</sub>ase in *Msph. stadtmanae*. Given the unexpected appearance of the F<sub>420</sub> H<sub>2</sub>ase activity in *Msph. stadtmanae*, it may also be worthwhile to use RT PCR to confirm the absence of F<sub>420</sub> H<sub>2</sub>ase in the *Methanosphaera cuniculi*, a closely related methanoarchaeon that also lacks F<sub>420</sub> H<sub>2</sub>ase activity (Biavati *et al.* 1988). There are other methanoarchaea that are not expected to possess an F<sub>420</sub> H<sub>2</sub>ase, including *Methanobacterium* strain G2R (McKellar and Sprott 1987), and *Ml. tindarius* (Deppenmeier *et al.* 1989); RT PCR could potentially be used to screen for the presence of transcripts encoding for a putative F<sub>420</sub> H<sub>2</sub>ase.

The results of this thesis show that while much of the methanogenic and electron transport pathways of the methanoarchaea have been elucidated, there may still be variations that arise in these pathways. After all, the five methanogenic orders contain widely diverse taxa with vast differences in morphology, physiological characteristics, and adaptations to different ecological niches (Bapteste *et al.* 2005). It should come as no surprise that a group as diverse as the methanoarchaea would have differences that may be unseen when information is derived from limited numbers of model organisms. The advent of increasingly sophisticated biochemical and genomic techniques through the years has aided in the study of these microorganisms and their unique physiologies, and will very likely lead to more interesting discoveries in the years to come.

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## Appendix A - Other methanogenic media

Methanobacterium bryantii M.o.H. (DSM 863) (Jarrell et al. 1982)	
KH <sub>2</sub> PO <sub>4</sub>	0.42 g/L
K <sub>2</sub> HPO <sub>4</sub>	0.23
MgCl <sub>2</sub> ·6H <sub>2</sub> O	0.064
CaCl <sub>2</sub> ·2H <sub>2</sub> O	0.03
NaCl	0.30
NH <sub>4</sub> Cl	0.70
Na <sub>2</sub> CO <sub>3</sub>	0.16
Resazurin	1.0 mL from 0.025% stock solution
Mineral elixir	10 mL/L (or 1 mL/L from 10x stock)
Na <sub>2</sub> S·9H <sub>2</sub> O	1% (v/v) from 200 mM stock after gassing
Growth at pH 7.0 under H <sub>2</sub> :CO <sub>2</sub> (pH 8.5 before gassing), growth temperature: 37°C	

Methanothermobacter marburgensis (DSM 2133) (Fuchs et al. 1978)	
KH <sub>2</sub> PO <sub>4</sub>	0.42 g/L
K <sub>2</sub> HPO <sub>4</sub>	0.23
MgCl <sub>2</sub> ·6H <sub>2</sub> O	0.04
CaCl <sub>2</sub> ·2H <sub>2</sub> O	0.03
NaCl	0.595
NH <sub>4</sub> Cl	0.70
Na <sub>2</sub> CO <sub>3</sub>	0.16
Resazurin	1.0 mL from 0.025% stock solution
Mineral elixir	10 mL/L (or 1 mL/L from 10x stock)
Na <sub>2</sub> S·9H <sub>2</sub> O	1% (v/v) from 200 mM stock after gassing
Growth at pH 7.0 under H <sub>2</sub> :CO <sub>2</sub> (pH 8.5 before gassing), growth temperature: 63°C	

Methanosaeta concilii (DSM 3671) (Patel 1984)	
KH <sub>2</sub> PO <sub>4</sub>	0.30 g/L
NaCl	0.60
MgCl <sub>2</sub> ·6H <sub>2</sub> O	0.10
CaCl <sub>2</sub> ·2H <sub>2</sub> O	0.08
NH <sub>4</sub> Cl	1.00
KCO <sub>3</sub>	4.00
Na acetate	6.80
Resazurin	1.0 mL from 0.025% stock solution
Mineral elixir	10 mL/L (or 1 mL/L from 10x stock)
Vitamin elixir	10 mL/L (or 1 mL/L from 10x stock)
Na <sub>2</sub> S·9H <sub>2</sub> O	1% (v/v) from 200 mM stock after gassing
Growth at pH 7.0 under N <sub>2</sub> :CO <sub>2</sub> (pH 7.7 before gassing), growth temperature: 37°C	

Methanolobus tind	Varius (DSM 2278) (König and Stetter 1982)
KCl	0.335 g/L
K <sub>2</sub> HPO <sub>4</sub>	0.14
MgCl <sub>2</sub> ·6H <sub>2</sub> O	4.00
MgSO <sub>4</sub> ·7H <sub>2</sub> O	3.45
CaCl <sub>2</sub> ·2H <sub>2</sub> O	0.14
NaCl	18.00
NH <sub>4</sub> Cl	0.25
NaHCO <sub>3</sub>	1.00
Fe(NH <sub>4</sub> ) <sub>2</sub> (SO <sub>4</sub> ) <sub>2</sub> ·7H <sub>2</sub> O	0.002
Resazurin	1.0 mL from 0.025% stock solution
Mineral elixir	10 mL/L (or 1 mL/L from 10x stock)
Vitamin supplement	10 mL/L (or 1 mL/L from 10x stock)
Na <sub>2</sub> S·9H <sub>2</sub> O	1% (v/v) from 200 mM stock after gassing
Methanol	1% (v/v) from 40% stock after autoclaving
Growth at pH 6.5 under N <sub>2</sub> :CO <sub>2</sub> (p	pH 8.3 before gassing), growth temperature: 28°C

Methanococcus voltae (1	DSM 1537) (DSZM catalogue of strains, 1993)
KCl	0.335 g/L
MgCl <sub>2</sub> ·6 H <sub>2</sub> O	4.00
MgSO <sub>4</sub> ·7 H <sub>2</sub> O	3.45
NH <sub>4</sub> Cl	0.25
CaCl <sub>2</sub> ·2 H <sub>2</sub> O	0.14
K <sub>2</sub> HPO <sub>4</sub>	0.14
Yeast extract	2.00
Trypticase/Tryptone	2.00
NaCl	18.00
NaHCO <sub>3</sub>	5.00
Fe(NH <sub>4</sub> ) <sub>2</sub> (SO <sub>4</sub> ) <sub>2</sub> ·7H <sub>2</sub> O	0.002
Resazurin	1.0 mL from 0.025% stock solution
Na <sub>2</sub> S·9 H <sub>2</sub> O	1% (v/v) from 200 mM stock after gassing
Na Acetate	1.00
Mineral elixir	10 mL/L
Vitamin supplement	10 mL/L
Growth at pH 7.0 under H <sub>2</sub> :CO <sub>2</sub>	(pH 7.5 before gassing), growth temperature: 37°C

Mineral elixir stock solution		
Trisodium nitrilacetate	2.02 g/L	
FeCl <sub>3</sub> ·6H <sub>2</sub> O	0.21	
CoCl <sub>2</sub> ·6H <sub>2</sub> O	0.20	
MnCl <sub>2</sub> ·4H <sub>2</sub> O	0.10	
$ZnCl_2$	0.10	
NiCl <sub>2</sub> ·6H <sub>2</sub> O	0.10	
CaCl <sub>2</sub> ·2H <sub>2</sub> O	0.05	-
CuSO <sub>4</sub> ·2H <sub>2</sub> O	0.05	
Na <sub>2</sub> MoO <sub>4</sub> ·2H <sub>2</sub> O	0.05	

Vitamin supplement stock solution	
Pyridoxine-HCl	10 mg/L
Riboflavin	5
Thiamine	5
Nicotinic acid	5
p-Aminobenzoic acid	5
Lipoic acid (aka Thioctic acid)	5
Biotin	2
Folic acid	2
Cyanocobalamin	1

Appendix B - Mass Spectroscopy data for *Methanospirillum hungatei* GPI. Confirmation of the isolation of the  $F_{420}$ -reducing hydrogenase during the purification of the phenazine-dependent  $F_{420}H_2$  dehydrogenase activity.

1.  $F_{420}$ -reducing hydrogenase isoenzyme II alpha subunit, *Methanosarcina barkeri* (Expect =  $2.1 \times 10^{-3}$ )

Expectation	Sequence
3.6 x 10 <sup>-3</sup>	<sup>327</sup> GTFAQHIAR <sup>335</sup>
2.1 x 10 <sup>-2</sup>	<sup>381</sup> GSNIHLAR <sup>388</sup>
0.76	<sup>32</sup> GDWLSITPVR <sup>41</sup>
1.5	<sup>291</sup> ANPQMEACTGVPTYDGQPVEVGPR <sup>314</sup>
125.4	<sup>4</sup> VVEISPTTR <sup>12</sup>

**2.**  $F_{420}$ -reducing hydrogenase, *Methanosarcina acetovorans* C2A (Expect =  $2.7 \times 10^{-2}$ )

Expectation	Sequence
3.6 x 10 <sup>-3</sup>	<sup>327</sup> GTFAQHIAR <sup>335</sup>
0.27	<sup>381</sup> GTDVHIAR <sup>388</sup>
0.76	<sup>32</sup> GDWLSITPVR <sup>41</sup>

#### <sup>4</sup>VVEISPTTR<sup>12</sup>

3.  $F_{420}$ -reducing hydrogenase isoenzyme I, *Methanosarcina barkeri* (Expect =  $3.5 \times 10^{-2}$ )

Expectation	Sequence
3.6 x 10 <sup>-3</sup>	<sup>327</sup> GTFAQHIAR <sup>335</sup>
0.27	<sup>381</sup> GTDVHIAR <sup>388</sup>
125.4	<sup>4</sup> VVEISPTTR <sup>12</sup>

**4.**  $F_{420}$ -reducing hydrogenase isoenzyme I beta subunit, *Methanosarcina barkeri* (Expect = 0.18)

Expectation	Sequence
1.5	<sup>230</sup> VGNEIWSK <sup>237</sup>
0.77	<sup>1</sup> MIEDPYLGK <sup>9</sup>

5.  $F_{420}$ -reducing hydrogenase beta subunit, *Methanosarcina mazei* Gö1 (Expect = 0.77)

Expectation	Sequence
2.1 x 10 <sup>-2</sup>	<sup>195</sup> YEQPGCHVCLDYVSSLA- DISTGSVGSPDGWSTY <sup>227</sup>

Appendix C - Similarity of the sequences encoding for the subunits of the  $F_{420}$ reducing hydrogenase from  $Msp.\ hungatei\ JF1$  with corresponding sequences from  $Ms.\ mazei\ G\"o1,\ Mtb.\ thermoautotrophicus\ \Delta H,$  and  $Mc.\ voltae.$ 

Ms. mazei Gö1 (accession number)	Msp. hungatei JF1		
	Identity	Positives	E value
FrhB (β) (NP_635066)	190/292 (65%)	238/292 (81%)	4e-113
FrhG (γ) (NP_635067)	183/267 (68%)	214/267 (80%)	1e-105
FrhD (δ) (NP_635068)	90/156 (57%)	120/156 (76%)	2e-49
FrhA (α) (NP_635069)	310/455 (68%)	369/455 (81%)	0.0

$Mtb.$ $thermoautotrophicus$ $\Delta H$ (accession number)	Msp. hungatei JF1		
	Identity	Positives	E value
FrhB (β) (P19499)	147/276 (53%)	201/276 (72%)	5e-81
FrhG (γ) (P19498)	122/235 (51%)	153/235 (65%)	2e-64
FrhD (δ) (P19497)	64/155 (41%)	103/155 (66%)	1e-35
FrhA (α) (P19496)	172/456 (37%)	266/456 (58%)	1e-86

Mc. voltae (accession number)	Msp. hungatei JF1		
	Identity	Positives	E value
FrcB (β) (Q00391)	149/275 (54%)	194/275 (70%)	2e-84
FreG (γ) (Q00393)	121/250 (48%)	150/250 (60%)	2e-56
FrcD (δ) (CAA43497)	57/158 (36%)	94/158 (59%)	2e-20
FrcA (α) (Q00390)	180/439 (41%)	249/439 (56%)	5e-81

## Appendix D - Similarity of the $F_{420}$ -reducing hydrogenase of $Msp.\ hungatei\ JF1$ with $F_{420}$ -reducing hydrogenases from $Methanosarcina\ spp.$

Blast searches performed at the National Centre for Biotechnology Information (NCBI, http://www.ncbi.nlm.nih.gov/) using sequences encoding the subunits of the  $F_{420}$ -reducing hydrogenase of Ms.mazei Gö1.

Sequences producing significant alignments ( $\alpha$ -subunit):	(Bits)	Value
gi 20986624 gb AE008384.1 Methanosarcina mazei strain Goel, com gi 19918815 gb AE010299.1 Methanosarcina acetivorans str. C2A, gi 72394721 gb CP000099.1 Methanosarcina barkeri str. fusaro gi 2463278 emb Y13763.1 MSFHMIII Methanosarcina barkeri fhm a, d gi 2463273 emb Y13764.1 MSFHMII Methanosarcina barkeri fha a, e, gi 88186784 gb CP000254.1 Methanospirillum hungatei JF-1, compl gi 6626257 gb AE000666.1 Methanothermobacter thermautotrophi	934 842 824 816 670 330	0.0 0.0 0.0 0.0 0.0 0.0 5e-90
Sequences producing significant alignments ( $\gamma$ -subunit):	(Bits)	Value
gi 20986624 gb AE008384.1 Methanosarcina mazei strain Goel, com gi 72394721 gb CP000099.1 Methanosarcina barkeri str. fusaro gi 2463278 emb Y13763.1 MSFHMIII Methanosarcina barkeri fhm a, d gi 2463273 emb Y13764.1 MSFHMII Methanosarcina barkeri fha a, e, gi 19918815 gb AE010299.1 Methanosarcina acetivorans str. C2A, gi 88186784 gb CP000254.1 Methanospirillum hungatei JF-1, compl gi 6626257 gb AE000666.1 Methanothermobacter thermautotrophi	589 509 509 474 473 381	2e-168 3e-144 3e-144 7e-134 3e-133 1e-105

# Appendix D - Similarity of the $F_{420}$ -reducing hydrogenase of Msp. hungatei JF1 with $F_{420}$ -reducing hydrogenases from Methanosarcina spp. (cont'd)

Sequences producing significant alignments ( $\delta$ -subunit):	(Bits)	Value
gi   20986624   gb   AE008384.1   Methanosarcina mazei strain Goel, com gi   19918815   gb   AE010299.1   Methanosarcina acetivorans str. C2A, gi   72394721   gb   CP000099.1   Methanosarcina barkeri str. fusaro gi   2463278   emb   Y13763.1   MSFHMIII   Methanosarcina barkeri fhm a, d gi   88186784   gb   CP000254.1   Methanospirillum hungatei JF-1, compl gi   116077928   emb   AM114193.2   Uncultured methanogenic archaeon RC	328 286 281 281 192 159	2e-90 2e-77 3e-76 3e-76 2e-49 2e-39
Sequences producing significant alignments ( $\beta$ -subunit):	(Bits)	Value
gi   20986624   gb   AE008384.1   Methanosarcina mazei strain Goel, com gi   72394721   gb   CP000099.1   Methanosarcina barkeri str. fusaro gi   2463278   emb   Y13763.1   MSFHMII   Methanosarcina barkeri fhm a, d gi   2463273   emb   Y13764.1   MSFHMII   Methanosarcina barkeri fha a, e, gi   19918815   gb   AE010299.1   Methanosarcina acetivorans str. C2A, gi   88186784   gb   CP000254.1   Methanospirillum hungatei JF-1, compl gi   45047480   emb   BX957222.1   Methanococcus maripaludis S2 complet	593 541 541 520 499 405 315	2e-169 6e-154 6e-154 1e-147 3e-141 4e-113 6e-86

## Appendix E - Development of a protocol for the purification of the phenazine-dependent $F_{420}H_2$ dehydrogenation activity of Methanosphaera stadtmanae

Various column chromatography techniques were tested in order to enrich the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity of *Msph. stadtmanae*. The results of some of the steps are summarized here. All steps were performed under anaerobic conditions, at ambient room temperature.

## 1. Ammonium sulfate ((NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>) precipitation

 $(NH_4)_2SO_4$  was added to the clarified cell-free extract (membranes removed) in successive increments to yield  $(NH_4)_2SO_4$  saturation of 50, 60, and 75%. 65-75% of the phenazine-dependent  $F_{420}H_2$  dehydrogenation activity was recovered in the 75%  $(NH_4)_2SO_4$  pellet (Figure 6). The pellet was initially re-suspended in 200 mM KPO<sub>4</sub>, pH 7.0, with 10 mM DTT and 5% glycerol, but in later experiments was switched to 100 mM HEPES, pH 7.0, with 5% glycerol to accommodate salt sensitive chromatographic techniques. Methyl viologen (shown here) and  $F_{420}$  H<sub>2</sub>ase activities also precipitated out in a similar manner to the  $F_{420}H_2$  dehydrogenation activity, maximally at 75%  $(NH_4)_2SO_4$  saturation.

A quick word on desalting of the  $F_{420}H_2$  dehydrogenation activity is discussed here. The bulk of the  $F_{420}H_2$  dehydrogenation activity from *Msph. stadtmanae* preciptitates at 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> saturation (~2.8 M): to make use of various column chromatography steps, such as affinity or ion-exchange chromatography, the protein must be desalted prior to application of the extract onto the respective column matrices. The desired desalting method would be to use ultrafiltration, but this technique would have to

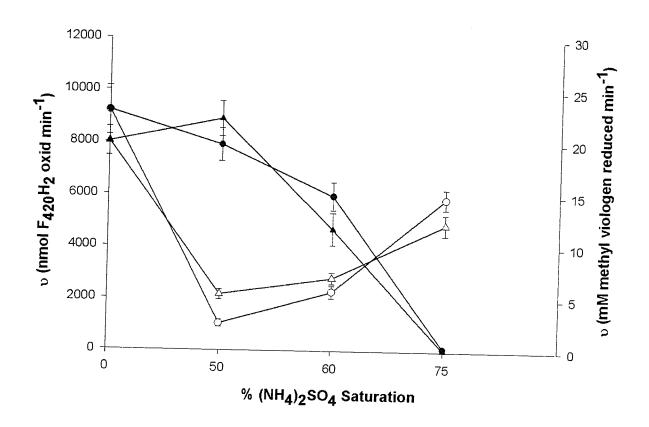


Figure 1. Ammonium sulfate ((NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>) precipitation profile of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity from clarified cell-free extract. •, F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> supernatant.

 $\circ$ ,  $F_{420}H_2$  dehydrogenation activity of  $(NH_4)_2SO_4$  pellet.

▲, methyl viologen-reducing hydrogenase activity of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> supernatant. Δ, methyl viologen reducing hydrogenase activity of  $(NH_4)_2SO_4$  pellet.

be conducted in the anaerobic chamber at room temperature, possibly leading to denaturation of the protein. Instead of ultra-filtration, two different desalting methods were tested, gel filtation and dialysis.

During gel filtration, the salt enters the pores of the gel beads and moves slowly through the column, while the proteins are excluded from the pores and move freely around the beads (Preneta 1995). The homogenized 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet (~12 mL) was loaded onto a Sephadex G-25 column (40 ml), in the anaerobic chamber at ambient room temperature; 50 mM HEPES, pH 7.0, containing 5% glycerol, 0.05% Triton-X-100, and 10 mM DTT, was used as the exchange buffer. The pooled protein was tested for activity and only 15-25% of the loaded F<sub>420</sub>H<sub>2</sub> dehydrogenation activity was retained. Of note, several colored fractions separated from the main protein band during passage through the gel filtration column. These fractions were added back to the desalted protein mix prior to assay; F<sub>420</sub>H<sub>2</sub> dehydrogenation activity could not be reactivated even after addition of the separated fractions.

Dialysis of the 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet was attempted, but this method was not anymore useful than the gel filtration step. The protein sample was loaded into semipermeable membrane tubing and placed into the desired salt-free buffer; small molecules pass through the membrane while large molecules are retained (Harris 1995). The resuspended 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> pellet was placed into pretreated Fisherbrand cellulose dialysis tubing (6000-8000 Da cut-off) and dialyzed in 50 mM HEPES, pH 7.0, containing 5% glycerol, 0.05% Triton-X-100, and 10 mM DTT. Dialysis was performed either in the anaerobic chamber (with frequent exchanges of fresh, cold buffer), or in an anaerobic 1 L Corning bottle at 4°C. The samples were dialyzed for 8 hours: for dialysis

in the anaerobic chamber the buffer was exchanged every hour; the buffer was changed after one hour, and then every two hours thereafter when dialyzed at 4°C in an anaerobic corning bottle. In either case, only 25-30% (or less) of the activity was retained after completion of dialysis. As such, dialysis was deemed to be an unsatisfactory step.

### 2. Metal chelate affinity chromatography

The technique of immobilized metal affinity chromatography was used in an attempt to isolate the  $F_{420}H_2$  dehydrogenation activity from *Msph. stadtmanae*. This process, using Ni<sup>2+</sup>-bound iminodiacetic acid, was used successfully in our studies to enrich for the phenazine-dependent  $F_{420}H_2$  dehydrogenation activities from *Ms. barkeri* Fusaro and *Msp. hungatei* GP, though ultimately the observed  $F_{420}H_2$  dehydrogenation activity was associated with the  $F_{420}$   $H_2$  ase isolated from the respective methanoarchaeon.

Two separate metals were used in our studies, Ni<sup>2+</sup> and Cu<sup>2+</sup>. All buffers used in these processes are the same as those previously described for the purification of the dehydrogenation activities for *Msp. hungatei* GP1 and *Ms. barkeri* Fusaro, except for the addition of 10 mM DTT. All buffers were made anaerobic *via* the gassing manifold and refrigerated until use. The protein, during the loading, washing, and elution steps, was kept as cool as possible using ice gel packs wrapped around the column to prevent denaturation, and all fractions were stoppered and kept cool with ice packs prior to removal from the anaerobic chamber.

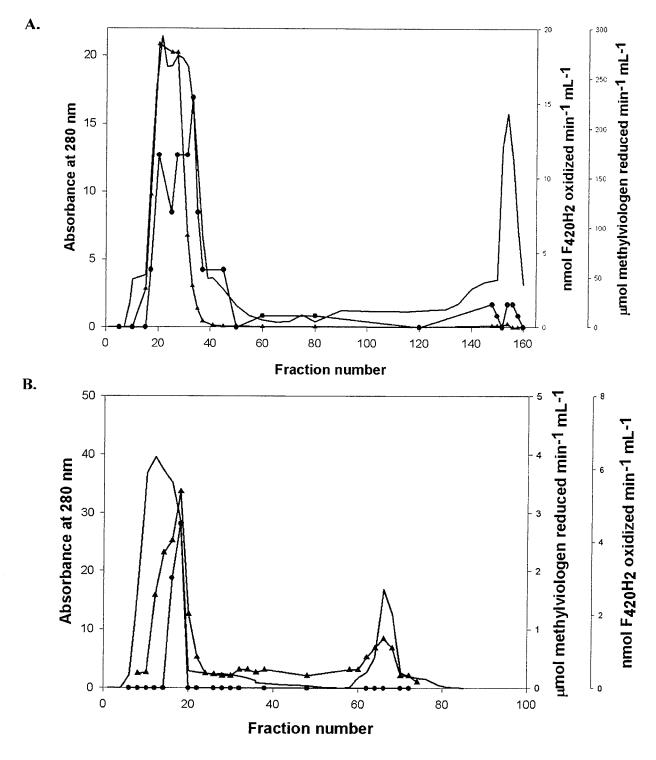
 $\mathrm{Ni}^{2+}$  was the first metal to be tested. The results are shown in Figure 2A. The  $\mathrm{F}_{420}\mathrm{H}_2$  dehydrogenation activity did not have any significant affinity for  $\mathrm{Ni}^{2+}$ , with only 14% of the loaded activity binding to the column matrix. To ensure that the lack of

binding by the protein was not due to over-saturation of the column, the unbound fraction was reloaded onto a freshly recharged Ni<sup>2+</sup> column; the subsequently reloaded protein did not have any apparent affinity for the Ni<sup>2+</sup>-bound matrix.

 ${
m Cu}^{2+}$  was also tested for its ability to bind the *Msph. stadtmanae*  ${
m F}_{420}{
m H}_2$  dehydrogenation activity (Figure 2B).  ${
m Cu}^{2+}$  was tested by Denny Wong as a possible purification step to isolate the methyl viologen reducing hydrogenase from *Msph. stadtmanae*, which ultimately did not have an affinity for this particular metal ion (Wong, PhD thesis 1998).  ${
m Cu}^{2+}$  was also used by Kemner and Zeikus (1994b) to separate the  ${
m F}_{420}$   ${
m H}_2$ ase from the  ${
m F}_{420}$ -nonreducing hydrogenase activities of *Ms. barkeri* MS; in this study, the  ${
m F}_{420}$ -nonreducing hydrogenase bound to the  ${
m Cu}^{2+}$ , whereas the  ${
m F}_{420}$   ${
m H}_2$ ase did not bind to the column. An experiment by Choquet and Sprott (1991) found that the  ${
m F}_{420}$   ${
m H}_2$ ase had affinity for  ${
m Cu}^{2+}$ , but the enzyme was found to be inactive upon elution from the immobilized  ${
m Cu}^{2+}$  column.

The results of our studies are shown in Figure 2B. As expected the methyl viologen reducing activity did not have any significant affinity for the immobilized  $Cu^{2+}$ , nor did the  $F_{420}H_2$  dehydrogenation activity. Passage through the  $Cu^{2+}$  column resulted in over 4-fold loss in the pooled activity.

It remains to be seen whether or not other metal ions, such as  $Zn^{2+}$  or  $Fe^{2+}$ , may be useful in the enrichment of the  $F_{420}H_2$  dehydrogenation activity from *Msph. stadtmanae*.



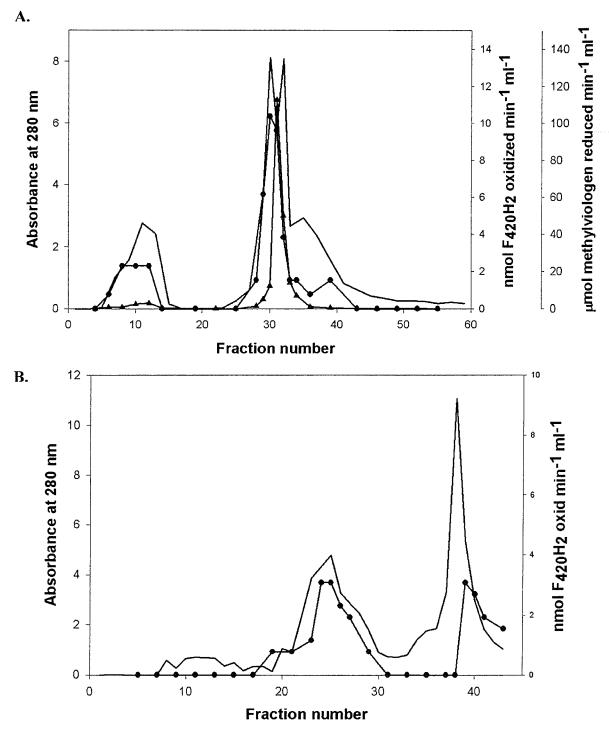
**Figure 2**. (A) Protein and activity profile of fractions eluted from a Ni<sup>+</sup> affinity chromatography column. (B) Protein and activity profile of fractions eluted from a Cu<sup>2+</sup> affinity column.  $A_{280}$ , no symbol.  $\bullet$ ,  $F_{420}H_2$  dehydrogenation activity.  $\blacktriangle$ , Methyl viologen reducing hydrogenase activity.  $F_{420}H_2$  dehydrogenation activity assayed using 500  $\mu$ M phenazine.

#### 3. DEAE Sephacel (Anion exhange chromatography)

Ion-exchange chromatography is the separation of proteins on the basis of their charge (Roe 1995). Proteins carry both positive and negative charges on their surface, due in large part to the side chains of acidic and basic amino acids (Roe 1995). In low-ionic strength buffer, positively or negatively charged protein binds to ion-exchange matrices; the bound protein is eluted by increasing the ionic strength of the buffer.

DEAE Sephacel, an anion exchanging matrix, has been used in the initial enrichment process of the F<sub>420</sub>H<sub>2</sub> dehydrogenase activity of *Ms. mazei* Gö1 and *Ml. tindarius*, so it was applied as step in the enrichment of the F<sub>420</sub>H<sub>2</sub> dehydrogenation activity of *Msph. stadtmanae* (Figure 3A). Following (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> precipitation, the protein is desalted *via* gel filtration and then applied to the DEAE Sephacel column. Two separate peaks of activity are observed. One activity elutes without binding to the column; this activity does not appear to elute with significant hydrogenase activity. The second peak elutes within the linear 0-2 M NaCl gradient, along with a methyl-viologen reducing hydrogenase activity.

This is not an ideal method for the initial chromatography step as the protein must first be desalted; the resuspended 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> protein pellet does not bind to the DEAE Sephacel matrix. The desalted protein does bind to the matrix, but the activity prior to application to the column is greatly decreased. Clarified cell-free extract (protein suspended in 200 mM KPO<sub>4</sub>) in the absence of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> precipitation did not have any affinity for the DEAE Sephacel.



**Figure 3**. A. Protein and activity profile of fractions eluted from a DEAE Sephacel chromatography column.  $A_{280}$ , no symbol.  $F_{420}H_2$  dehydrogenation activity,  $\bullet$ . Methyl viologen reduction,  $\blacktriangle$ . B. Protein and activity profile of fractions eluted from a Phenyl sepharose CL 4B column.  $A_{280}$ , no symbol.  $F_{420}H_2$  dehydrogenation activity,  $\bullet$ . Conditions for elution as described in text.  $F_{420}H_2$  dehydrogenation activity assayed using 500  $\mu$ M phenazine.

#### 4. Phenyl Sepharose CL-4B (Hydrophobic interaction chromatography)

Hydrophobic interaction chromatography involves the adsorption of protein to non-ionic groups, such as octyl or phenylalanine functionalities bound to inert matrices (Roe 1995). This requires the presence of hydrophobic patches on the protein, including amino acids such as alanine, trytophan, or phenylalanine. The addition of salts would remove the water molecules that form an ordered shell around the hydrophobic patches. The hydrophobic patches are exposed and may interact with non-ionic groups bound to the inert matrix. The protein is eluted from the matrix by decreasing the salt content (or other water miscible solvents).

Clarified cell-free extract (protein suspended in 200 mM KPO<sub>4</sub>) was applied to Phenyl Sepharose and to Octyl Sepharose columns, with no apparent affinity. Later experiments made use of the presence of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, at sub-precipitation saturation. The protein and activity profile of a Phenyl Sepharose column loaded with a 75% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> saturated pellet, re-suspended in 60% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> saturated buffer, is shown in Figure 3B. This has proven to be a convenient starting step, as it makes use of the high salt content of the protein sample. Elution of the protein also leads to desalting of the protein, enabling the use of other techniques after elution (i.e. ion exchange, affinity chromatography). The bound protein was eluted using a continuous 60-0% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> gradient, in 50 mM HEPES, pH 7.0, with 0.05% Triton-X-100, 5% glycerol, and 10 mM DTT (Buffer A), followed by a single washing step of Buffer A to remove tightly bound protein. Two peaks of F<sub>420</sub>H<sub>2</sub> dehydrogenation activity were observed during elution. One peak eluted off the column along during the continuous 60-0% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> gradient, while the second peak eluted when the column was washed with (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> –free buffer.