Mineralogy, Lithostratigraphy and Geochemistry of North Ingebright Lake, Saskatchewan, Canada

By

Yuqiang Shang

A Thesis Submitted to the Faculty of Graduate Studies in Partial Fulfillment of the Requirements for the Degree of

Doctor of Philosophy

Department of Geological Sciences University of Manitoba Winnipeg, Manitoba

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A Thesis/Practicum submitted to the Faculty of Graduate Studies of The University

of Manitoba in partial fulfillment of the requirements of the degree

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ABSTRACT

North Ingebright Lake, containing Canada's thickest known Holocene terrestrial salt sequence, is a small (1.25 km²), hypersaline playa basin located in the Great Sand Hills area of southwestern Saskatchewan. Because of its remarkably thick lacustrine sedimentary fill, the basin offers considerable potential for providing a high resolution record of past environmental changes in a region that has few other sources of Holocene paleoenvironmental information. Until recently, efforts to acquire cores of this thick sequence of well-indurated but very soluble salt have been unsuccessful using conventional coring techniques. However, exceptional sediment recovery was achieved using a compressed-air diamond drilling technique that permitted collection of continuous, large-diameter, undisturbed and chemically unaltered cores.

The 10,000 year long lacustrine sequence recovered from the basin consists of well-indurated salt, with only minor dispersed mud and organic debris. Indeed, the section is remarkable in its lack of obvious bedding, colour variation or other visible sedimentary structures. The mineral suite of these Holocene salts consists mainly of hydrated Na, Mg, Ca, and Mg+Na sulfates, carbonates, and chlorides. Minor amounts of nitrate evaporitic minerals also occur. This long uninterrupted sequence of soluble salts implies that the lake was characterized by high salinity throughout its Holocene history. However, variations in the mineralogy of the salts suggest that the basin experienced significant changes in brine composition. These evaporites, which allow a detailed reconstruction of specific ion activities and ratios in the brine through time, provide our evidence of the complexity of brine evolution in a closed-basin lacustrine environment in western Canada.

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Chapter 1 Introduction

1.1 Overview

Limnology is a well established science, but paleolimnology is relatively new. Paleolimnology is generally considered to be a multidisciplinary science dealing with the reconstruction of past lacustrine environments through the interpretation of fossil records and the study of sediments. Most of the older (pre-1970's) published work addresses the biological aspects of paleolimnology, whereas comparatively few papers attempt to integrate studies of the lacustrine deposits from which various parameters have been extracted.

Saline and hypersaline lakes have been largely ignored by paleolimnologists because they are generally deficient in paleobiological parameters such as ostracodes, chrysophytes, diatoms, chironomids, and plant macrofossils which are conventionally employed to reconstruct past environments. However, non-biological sediment parameters, such as endogenic mineral composition, geochemistry, and isotopes provide important information about paleolimnological, paleochemical, and paleohydrological changes (Last, 1990; 1999; Sullivan and Charles, 1994; Fernandez, 1994; Zuniga et al., 1991). The Canadian Prairies contain thousands of saline and hypersaline lakes and playas. The modern biological and hydrochemical characteristics of many of these lakes have been well studied (Hammer, 1978a; 1978b; Hammer and Haynes, 1978) but, until recently, little work has been done in the sedimentary realm of these lakes (Last and Schweyen, 1983).

1.2 General Characteristics of Saline Lakes

Salt lakes are usually located in arid to semi-arid regions of the world. The volume of the world's saline lake water (about 100,000 km³) is nearly equal to the volume of its fresh water (125,000 km³) (Hammer, 1986). Saline water bodies range from a few m² to thousands of km² in area. For example, both the Caspian Sea (436,400 km²) and the Aral Sea (62,00 km²) in eastern Europe/Asia are large salt lakes.

Saline lakes have a wide range of salinities, from 3 to over 400 g/l. Williams (1964) defined saline waters as those which have a salinity greater than 3 g/l, whereas water with a salinity less than 3 g/l is considered to be fresh. This definition has been well accepted as an arbitrary division point. Figure 1.1 shows classifications of saline lakes by several different workers (SIL, 1959; Hammer, 1978; Hammer et al., 1988; Williams, 1964; 1991; Last, 1992; 1993). The salinity of hyposaline lakes ranges from 3 to 20 g/l; mesosaline lakes range from 20 to 50 g/l; and hypersaline lakes have salinities above 50 g/l.

Saline waters display a remarkable diversity in their chemical composition: some are carbonate-rich, many are chloride-rich, or sulphate-rich. Regardless of whether calcium, magnesium, or potassium are the dominant cations in the water, sodium is usually the most abundant cation. Anion constituents are much more diverse (Hardie et al., 1978; Eugster, 1970; 1982; Eugster et al., 1975; 1978). Solutes in saline lake brines may be derived from the soils and bedrock of the watershed, through precipitation and fallout of salts in the atmosphere, or from wind-blown material. Normally, bedrock lithology is a key factor which controls the

	hypersaline >50		mesosaline		hyposaline				freshwater		Last (1992; 1993)
	hypersaline >50	>50 mesohaline			salt water				freshwater		Williams (1964; 1991)
	hypersaline >50		mesohaline			hyposaline			subsallne	freshwater	Hammer (1978)
	hvnersaline	>40			mesohaline			euhaline		freshwater	Venice System (SIL 1959)
Salinity (g/l)	Ş	2	40		20 -	0	S.	ي ا		;	

Figure 1.1 Classification of Saline Lakes

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final composition of brines (Hardie et al., 1978).

1.3 Conditions of Lake Formation in the Canadian Prairies

Saline lakes can form wherever evaporation exceeds precipitation and the lake basin is hydrologically closed (Hammer, 1986; Mungoma, 1990; Williams, 1991; Seaman et al., 1991). In Canada, after the last episode of glaciation and deglaciation, millions of closed-basins were left in the Prairie region. These lakes show a tremendous diversity in size, basin morphology, hydrochemistry, and sedimentary characteristics (Last, 1988; 1994; 1999). Most of these lake basins are the result of either depositional or erosional processes associated with the glaciation and deglaciation of the region. The most common ways in which such lake basins originated include: (a) ice erosion of bedrock; (b) drainage barriers composed of moraine or outwash dams; (c) depressions formed in glacial deposits; and (d) ice dams (Smith and Ashley, 1985).

As the Lauretide ice sheet began to retreat from the Canadian Prairies, extensive proglacial lakes developed at or near the front of the ice. Proglacial lakes ranged in size from a few km² to hundreds of thousands of km². Most of these were formed when ice blocked preglacial valleys (drainage routes). For example, North Ingebright Lake is part of the drift-filled channel of a preglacial valley (Cole, 1926; Rueffel, 1968). Other lakes developed when the weight of the ice depressed the earth's crust sufficiently so that closure between the solid rock margin and ice margin formed. Most proglacial lakes have vanished either because the ice has retreated and the ice barrier has disappeared, or because the land has rebounded

to its preglacial position. During deglaciation, meltwater from the retreating glacier carved numerous ice-marginal channels and spillways. Although now abandoned or buried under more recent sediment, these depressions are often sites of saline lakes. For example, Ceylon Lake in southern Saskatchewan is presently a shallow, hypersaline playa. The basin originated as a glacial meltwater spillway (Last, 1989; Cole, 1926).

As the margin of the Laurentide Ice Sheet retreated to the northeast during deglaciation, altering drainage patterns, and changes in the extent of these proglacial lakes produced major changes in the Prairies. During its retreat, the Laurentide Ice Sheet blocked the preglacial drainage that flowed to Hudson Bay. The older lakes in Alberta tended to be long and narrow, while those in Saskatchewan and Manitoba were far more extensive because the relief of the Prairies is generally greater in the west than in the east. In Alberta there were lakes centred around the major valleys: these included Lake Edmonton, Lake Red Deer, Lake Bycroft, and Lake Peace. In general, the Saskatchewan proglacial lakes were larger and fewer than the southern Alberta lakes. These include Lake Swift Current, Lake Saskatoon, and Lake Rosetown which all drained either southeast or east. In Manitoba, Lake Agassiz was formed by the damming of glacial melt waters between the ice margin of the continental ice sheet and the Manitoba Escarpment (Meneley et al., 1957; Christiansen, 1967; 1979; Bird, 1972). Lake Agassiz initially drained southwards and eastwards and finally drained to Hudson Bay as the ice barrier to the north disappeared about 7700 B.P.(Andrews, 1987; Klassen, 1972; 1989; Teller, 1985; 1987; 1990; Teller and Thorleifson, 1983).

Although most of the proglacial lakes related to ice dams have disappeared, widely-distributed glacio-lacustrine deposits were left in the Canadian Prairies (Teller, 1987; Stalker, 1976). Sometimes beaches also were left. In the deeper parts of the lake, laminated clay and silts were deposited and formed extremely level clay plains, such as Lake Regina.

In shallow, nearshore lake waters, sands were deposited as deltas or fans; the large sand underflow fan (delta) of the Assiniboine River along the western shores of former glacial Lake Agassiz is a large-scale example of this. The coarsegrained Little Manitou fan, which was deposited in the western shores of glacial Last Mountain Lake during an outburst event near Watrous, Saskatchewan is an example a small fan (Kehew and Teller, 1993). During the last deglaciation, Lake Saskatchewan, Lake Elstow, and Last Mountain Lake were impounded in lowlands between ice margin to the north and upland tracts including the Hawarden Hills and the Allen Hills. These proglacial lakes often released outburst floods as ice dams failed in the edges of the upland areas. The drainage bursts deeply eroded the spillway channels and commonly emptied the lakes. During an outburst flood event from Lake Elstow, Watrous spillway was incised more than 30 m into the drift. The coarse sediment eroded at this time was deposited as the Little Manitou fan at the mouth of the spillway in the Last Mountain Lake basin (Kehew and Teller, 1993).

1.4 Economic Significance of Salt Lakes

In continental environments, sediments of closed-basin lakes often contain

the best records of past hydrochemical, hydrological, and climatic variations. Closed-basin lakes potentially yield the most sensitive and explicit record of changes in past environmental conditions and thus are valuable indicators of past climates. In addition, the sediments themselves are an important industrial mineral resource. For example, sodium sulphate is the second largest non-metallic mineral resource in western Canada (Last and Slezak, 1986; Douglas, 1970). Therefore, a knowledge of sedimentary processes and the mineralogy of these lacustrine sodium sulphate deposits has science and economic importance.

Commercial extraction of salts has occurred from numerous saline wetlands and playas in the Northern Great Plains region for over 80 years (Tomkins, 1953, 1954; Last and Slezak, 1987). One of North America's largest sodium sulfate mines and processing plants is located a few kilometres to the south of Ingebright Lake. Salts have also been extracted from nearby Chain, Snakehole, Corral, Boot, Verlo, and Vincent Lakes, all within 30 km of the North Ingebright basin. Although North Ingebright Lake has, to date, not been mined, the lake is presently under mineral lease. Cole (1926) calculated a total anhydrous Na₂SO₄ reserve of about 3 million tonnes.

Early geological work by Cole (1926) on the salt lakes of the Canadian Prairies was mainly based on their potential as a reserve for sodium and magnesium sulphates. Subsequently the biological and hydrochemical characteristics of many of these lakes were documented by the works of Rawson and Moore (1944), Tomkins (1953; 1954), and Hammer (1978). By comparison, relatively little work has been done on the sedimentary realm of the salt lakes. Up

to the early 1990s, the stratigraphic records of only eighteen of the basins had been examined (Last, 1992). Extensive research on the stratigraphic records of lakes in this region was part of a large multidisciplinary Palliser Triangle project by the Geological Survey of Canada (Last, 1996; Last et al., 1997; Lemmen, 1996; Vance et al., 1992, 1993, 1995; Vance; Last, 1994; Last, 1999; and Shang and Last, 1999).

1.5 Previous Work on North Ingebright Lake

North Ingebright Lake, also known locally as Ingebright North or Ingebright No.2, was named after the Norwegian immigrant Ingebright Selseth who homesteaded in 1912 near the lake (Rueffel, 1968). This sodium sulphate deposit was first reported in the Annual Report of the Topographical Survey, 1913-1914 (Tomkins, 1954; Rueffel, 1968); however, the deposit has not been exploited/mined to date (Last and Slezak, 1986).

Cole (1926) undertook an extensive investigation on the chemical composition of the brine and sediment of North Ingebright Lake. His study provides valuable scientific data for our understanding of chemical sedimentation history in the lake. In addition, North Ingebright Lake was also discussed by Tomkins (1954) and Rueffel (1968).

The glacial history, surficial geology, and stratigraphy of the North Ingebright Lake region have been discussed by a number of authors. Comprehensive reports and maps of the surficial geology of the Prelate area (72K), including North Ingebright Lake were done by David (1964). Christiansen (1968, 1979) described

the Quaternary history and stratigraphy of southern Saskatchewan including the Great Sand Hills area. Klassen (1991, 1994) discussed, in detail, the deglaciation history and proglacial lake development of southwestern Saskatchewan. Epp and Townley-Smith (1980) addressed modern morphology and terrestrial ecology of the Great Sand Hills area, including North Ingebright Lake area.

1.6 Objective of Research

The Ingebright Lake chain in southwestern Saskatchewan, comprising North Ingebright Lake and Ingebright Lake, contains the thickest known sequence of Holocene salts in Canada and provides a relatively complete record of Holocene environmental change. A detailed examination of the stratigraphic record of North Ingebright Lake may yield important information that will enhance our understanding of the paleoclimate and paleohydrology of the northern Great Plains.

In order to conduct this Ph.D research within the monetary and temporal constraints imposed by available funding, this project focuses on selected aspects of the North Ingebright Lake record, namely the lithostratigraphy, sedimentology, mineralogy, and geochemistry of the sediments. The general objective of this research project is to contribute the understanding of the geolimnology and history of North Ingebright Lake. The specific objectives of this research project are to: (1) acquire cores and samples from North Ingebright Lake; (2) describe and document the characteristics of the Holocene sediments in the lake basin; (3) identify the major spatial and temporal variations within this sediment record; and (4) relate these changes to hydrological fluctuations, climatic variation as well as other

changes.

1.7 Acknowledgements

I would like to thank my supervisor, Dr. W. M. Last, for his guidance, support, encouragement and patience courteously extended throughout this thesis research. I am also grateful to Dr. J. T. Teller for his many helpful insights into the study of Quaternary geology, as well as his help with specific details and problems. Sincere thanks are due to Drs. B. J. Hann and R. E. Vance for their critical reading and assistance in preparing this dissertation.

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Chapter 2 Regional Setting

2.1 Physiography

The northern Great Plains of western Canada is bounded by the Precambrian Shield on the north and east, and by the Rocky Mountain foothills and Cordillera on the west. This is a vast region of flat to gently rolling terrain interrupted by occasional steeply sided river valleys and punctuated by scattered groups of low hills or uplands (DMR, 1969; Richards and Fung, 1969 Prest, 1970; 1984; Bird, 1972). The major geomorphological features are the result of differential fluvial erosion of the weak bedrock and of Pleistocene glaciation. Prior to glaciation, the Prairies had undergone a prolonged interval of erosion. This produced broad northeast-trending valleys separated by low uplands. Stream deposits in the valleys were derived from the Cordillera or reworked from older Tertiary and Cretaceous deposits (DMR, 1969; Richards and Fung, 1969; Bird, 1972).

The gradual up slope east to west trend exhibited by the Canadian Prairies is interrupted by a number of sharp rises or steps. In general, these northwestward trending rises are most prominent in the south and become lower towards the north. The Manitoba Escarpment separates the Manitoba Plain on the east from the Saskatchewan Plain. The Missouri Coteau rises about 100 m above the Saskatchewan Plain on the east to the Alberta Plain on the west (Prest, 1984).

Glaciation is responsible for most superimposed features on the broader

bedrock physiography. When ice advanced, the northeast drainage was dammed so that lakes developed in the valleys and depressions, and water was diverted to the south. When ice retreated, ice marginal lakes developed and steep-walled valleys were cut because meltwater flowed from one lake basin to the next or the flow was channelled southward along the ice margin. Repeated glacial and nonglacial intervals produced a complex of valley fills of different ages (Fenton, 1984; Fulton et al, 1984; Teller, 1985; 1987; Gravenor et al., 1959). Normally these glacial valleys trend east or southeast, in contrast with the large nonglacial valleys that typically trend northeastward.

2.2 Summary of Late Deglaciation History of the Canadian Prairies

The Laurentide Ice Sheet advanced to its last maximum position across South Dakota and Iowa about 14 ka ago. After this maximum advance, the ice began to retreat northward, interrupted by both stoppages and minor readvances of major ice Iobes between 14 and 8 ka (Klassen, 1979; Teller, 1985; 1987; Teller and Thorleifson, 1983; Teller and Fenton, 1980). In the Canadian Prairies, a series of proglacial lakes formed mainly by ice dams. Widely distributed glaciolacustrine deposits were left on the Prairies (Wolfe and Teller, 1993). Although Late Wisconsinan deposits make up much of the surface of the Canadian Prairies and outcrops are abundant, both the location and timing of the Late Wisconsinan deglaciation limits remain controversial. In the past decade a number of regional scale histories of the deglaciation in the Prairies have been reported (Christiansen, 1979; Clayton and Moran, 1982; Klassen, 1989; Teller and Fenton, 1980). The differences among these reports are significant.

There are two general approaches to chronology and the use of radiocarbon dates among these reports. One approach advocates using all of the available stratigraphically significant dates. For example, nine ice-marginal positions have been recognized across the Prairies based on all available dates (Christiansen, 1979). The other approach suggests using only dates from wood (Teller and Fenton, 1980a; Teller et al, 1980b; Clayton and Moran, 1982). Teller and Fenton (1980) stated that dates on very fine organic matter are unreliable in areas where pre-Quaternary organic matter could contaminate the sediments. Therefore, dates based on finely-disseminated organic matter may be too old. They identified five lithostratigraphic units of Late Wisconsinan till and established the history of Late Wisconsinan glaciation and deglaciation in southwestern Manitoba.

The retreat of Late Wisconsinan ice from the Prairies appears to have proceeded by the melting of stagnant ice along broad marginal belts. Christiansen (1979) proposed an average of 60 m/a for the early stages and 275 m/a for final stages of retreat; Klassen (1983) derived retreat rates of 250 m/a and 300 m/a in northern Manitoba from varve counts and radiocarbon dates, respectively. Figure 2.1 shows the change in area of the ice sheet, area of ice-marginal lakes in southern Saskatchewan through late glacial time between 14 and 8 ka. Kehew and Teller (1994) show the history of proglacial lakes in southern Saskatchewan as related to the ice margin during deglaciation.

In southwestern Saskatchewan, by 11-10 ka, virtually all of the area was ice free. Residuals of drift-mantled stagnant ice may have remained within parts of the



Figure 2.1 Sequence of proglacial lakes in Saskatchewan and adjacent areas (from Teller, 1987, after Quigley, 1980). The phase 2 glacial boundary represents the late Wisconsin position about 13 ka. By about 11 ka the ice had retreated to the phase 8 position hummocky moraine belts in the area.

2.3 Study Area

2.3.1 Location

North Ingebright Lake occupies a narrow riverine basin located within the Great Sand Hills region of southwestern Saskatchewan, about 100 km west of Swift Current (lat 50°24'N, long 110°22' W) (Figure 2.2). The closed, riverine-shaped basin of North Ingebright Lake is about 3 km long and relatively narrow (<500 m), with a surface area (at normal high water level) of 1.25 km² (Table 2.1).

The basin probably originated about 11,000 years ago as a glacial meltwater spillway that was superposed on a buried, preglacial valley (David, 1964; Christiansen, 1979). In the 1960's 65 test holes were drilled in the Ingebright Lake area in an attempt to locate an adequate water supply for the sodium sulfate plant operation (Rueffel, 1968). The results of this comprehensive drilling program revealed that the lowest-lying areas in which the two Ingebright Lakes and a number of potholes are located is a "V"-shaped valley, incised into the bedrock, approximately 60 to 90 m deep and 240 m wide at the top. This valley may have originated as a preglacial valley or by the melting of ice during the last glacial retreat. Besides Ingebright Lakes, many other salt lakes in south Saskatchewan are also located in the preglacial valleys (Christiansen, 1967). Basal radiocarbon dates of nonglacial sediments in southwestern Saskatchewan show an average of 10,000 B.P. (Table 2.2). Therefore, it is likely that sedimentation in North Ingebright Lake



Figure 2.2 Map showing the sediment thickness of North Ingebright Lake Inset shows the location of North Ingebright Lake in the northern Great Plains of western Canada

Maximum length	3 km					
Maximum width	0.5 km					
Area	1.25 km²					
Maximum thickness of salt sediments	>23 m					
Mean daily temperature	-15°C (January)					
	20°C (July)					
Mean annual temperature	3.3°C					
Extreme temperatures	-46°C to 45°C					
Average annual precipitation	<30 cm					
Average annual evaporation	>125 cm					

Table 2.1Selected Morphometrical, Climatic, and Sediment Parameters of
North Ingebright Lake

9,120 ± 250 B.P.	Harris Lake (Sauchyn, 1990; Sauchyn & Sauchyn 1991)
9,240 ± 260 B.P.	Harris Lake (Last & Sauchyn, 1993)
9,500 B.P. 10,000 B.P.	Horseman Site (Klassen, 1994)
12,600 B.P.	Ham Site (Vreeken, 1994)
10,250 B.P.	Metiskow Lake (Wallick, 1981)
11,460 B.P.	Val Marie Site (Christiansen & Sauer, 1988)
9,000 B.P. 9,989 ±70B.P.	Clearwater Lake (Vance, 1994; Last et al., 1997)

Table 2.2 Basal Radiocarbon Dates of Nonglacial Sediments inSouthwestern Saskatchewan and Adjacent Area

was initiated at about 10 ka and possibly contains a complete record of Holocene environment change.

The drainage basin of the lake is characterized by hummock to gently rolling topography. The landforms surrounding North Ingebright Lake consists of sand flats, stabilized dunes, and active dunes (Figure 2.3). Most of the sand flats and dunes have developed no drainage system, and precipitation usually percolates through the sand directly into the water table instead of accumulating on the surface (Epp and Townley-Smith, 1980). Therefore, surface runoff is weak in the North Ingebright Lake area, and sedimentation is dominated by endogenic processes.

2.3.2 Climate

The region has a cold continental climate. In winter, in response to the equatorward displacement of the Equatorial Trough and the zonal westerlies, a cold, dry Arctic air mass dominates the area. Penetration of Pacific air into the continental interior is blocked by this cold, stable air. The influence of moist tropical air is also reduced. The lowest temperature during January can reach -46°C (Connor, 1920; Currie, 1953). In summer, as the Equatorial Trough and westerlies move northwards, the influence of the Arctic air mass is limited. Both Pacific and maritime tropical air masses penetrate further into the continental interior. The climate is typically dry and windy, characterized by great temperature variation, both daily and seasonally (Currie, 1953). The average annual temperature for the Great Sand Hills region for the period 1941 to 1970 was 3.3° C (CNC/IHD, 1978), and mean annual precipitation for this same period was less than 300



Figure 2.3 Surface deposits of the Great Sand Hills region (From Epp and Townley-Smith, 1980)

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	Jan.	Feb.	Mar.	Apr.	Мау	June	July	Aug.	Sep.	Oct.	Nov.	Dec,	Annual
Mean daily temperature (Cº)	-14	-11	-6	3.7	10.3	15	19	18	12	6.1	-4	-10	3.3
Mean rainfall (mm)	1	0.5	0.8	12.7	32.8	78	52	46	33	10	2.3	0	268.9
Mean snowfall (cm)	22	17.3	18	13.5	3.3	0.3	0	0	2.3	10	17	20	123.8
Mean total precipitation(mm)	22	17.3	18	26.2	36.1	78	52	46	35	21	19	19	299

 Table 2.3 Temperature and Precipitation Data for the North Ingebright Lake Region (1941-70)

Data from *Environment Canada* (1980)

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mm (Table 2.3). The town of Fox Valley, 20 km north of the basin, receives approximately 300 mm of precipitation annually, and has a mean daily temperature of -14°C in January and +21°C in July. Precipitation occurs as sporadic rain during the summer and as snow in the winter. Most of the precipitation occurs between May and September at the time of the most intensive evaporation. More than 125 centimetres of water evaporate annually from open bodies of water (CNC/IHD, 1978). Average annual moisture deficits are in the range of 20 to 30 centimetres as evaporation greatly exceedsprecipitation (Fremlin, 1974; Hammer, 1978).

2.3.3 Regional Geology

Southwestern Saskatchewan is underlain by nearly horizontal Phanerozoic sedimentary rocks. The Paleozoic section mainly consists of carbonate-evaporite cycles. The overlying Mesozoic and Cenozoic bedrock is dominantly a sand-shale sequence. The bedrock of the region is mantled by unconsolidated Quaternary sediment over 300 m thick in places. These deposits consist of till, fluvial sands and gravels, and lacustrine silts and clay (DMR, 1969; Richards and Fung, 1969).

Bedrock outcrops in the Great Sand Hills area are mainly concentrated on the north side along the South Saskatchewan Valley. Bedrock topography is reflected by the present day topography. The bedrock highs, which could have obstructed glacier flow and caused the development of hummocky moraines and ice-thrust ridges, have the highest elevation today. Preglacial valleys were usually filled with thick drift, with a series of sinkholes comparable in size to that of the valley. Most of these sinkholes are filled with

(from David, 1964)						
Time	Formations					
Pleistocene or Late Pliocene	Saskatchewan gravels					
Miocene-Oligocene	Cypress Hills					
Eocene	Swift Current Creek					
Paleocene	Ravenscrag					
Late Cretaceous	Eastend Bearpaw Belly River Lea Park					

Table 2.4 Bedrock Stratigraphy of Great Sand Hills Area

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shallow lakes, marking the former positions of preglacial valleys. For example, Bigstick Lake, South Ingebright Lake, North Ingebright Lake, and the sags around Johnsborough outline the buried Johnsborough valley.

The major bedrock formations are listed in Table 2.4. The Lea Park Formation consists of dark grey to black, non-calcareous shale, sandy shale, and fine-grained sand. The Belly River beds lie between the underlying Lea Park and overlying Bearpaw shales. They are comprised of fine-grained sandstone, shale, and sandy shale and uncemented sands. Along the preglacial valley, the Belly River beds directly underlie the drift.

The Bearpaw Formation consists mainly of dark grey marine shales, with layers of fine-grained sands and shaly sands. It overlies the Belly River beds and underlies the drift in most of the area. Eastend and Ravenscrag Formations appear to overlie the Bearpaw Formation and underlie the Swift Current Creek and Cypress Hills Formations. They consist chiefly of yellowish green fine-grained sand, coarse-grained silts, and grey shales (David, 1964).

Swift Current Creek and Cypress Hills Formations consist mainly of light coloured fine-grained sandstones and conglomeratic sandstones. These sandstones contain large amounts of quartzite pebbles and cobbles (David, 1964).

The Saskatchewan gravels are late Pliocene to early Pleistocene in age and are clearly preglacial. They consist of pebbles that are hard, well-rounded, and similar in appearance to the Cypress Hills Formation, from which they were derived (David, 1964).

The best preserved and most complete Pleistocene stratigraphic sections are in the buried preglacial valleys, such as Johnsborough Valleys in the Ingebright Lake area. Fine-grained preglacial sand and silt were commonly found at the base of valley fill. These sediments may have been deposited when the first ice in the area blocked the northward drainage and a preglacial lake formed in the valley (David, 1964).

2.3.4 Hydrogeology and Hydrologic Setting

North Ingebright Lake is one of a series of distinct, separate, riverine salt playas occupying a 50 km long, north-south oriented meltwater channel. Closure of the North Ingebright basin is small and a water level of more than approximately 1.5 m above today's surface will spill over to Ingebright Lake less than a kilometre to the south. The W-E spill point for the entire Ingebright-North Ingebright chain of basins is approximately 20 m above the present-day surface of the valley.

David (1964), Ruffel (1968), and Rutherford (1967, 1970) provide important regional and local groundwater and hydrochemical data of lakes in the region. Typical of most of the shallow, salt-dominated playas in western Canada, the hydrologic budget of North Ingebright Lake is dominated by groundwater inflow and evaporation. There is no perennial streamflow into this topographically closed basin and evaporation exceeds mean annual precipitation by a factor of 3 (CNC/IHD, 1978). Groundwater enters the lake via both diffuse inflow and through several discrete springs. The lake exhibits typical playa characteristics, filling with water during the spring and early summer and usually drying completely by late summer or fall. Maximum water depths are generally less than 50 cm.

David (1964) investigated the shallow groundwater and deep groundwater chemistry in North Ingebright Lake region; his data are summarised in Figures 2.4, 2.5 and 2.6. Cole (1926) analysed the chemical compositions of lake brines and spring water surrounding North Ingebright Lake basin (Figures 2.4, 2.5, and 2.6). The shallow



Figure 2.4 Mean water composition (milli-equivalent %) of North Ingebright Lake (■), inflowing springs(●), deep groundwater (♥) and shallow groundwater (♦)



Figure 2.5 Trilinear diagrams of ionic composition (milli-equivalent %) of shallow groundwater in North Ingebright Lake region



Figure 2.6 Trilinear diagrams of ionic composition (millii-equivalent %) of deep groundwater in North Ingebright Lake region

groundwater in North Ingebright Lake region has relatively low salinity, ~1.7 gl⁻¹ total dissolved solids. These waters vary in composition; dominant anions tend to be SO_4^{2-} and CO_3^{2-} +HCO₃⁻ and the dominant cations, Na⁺ and Mg²⁺ (Figure 2.5). These shallow groundwaters are slightly alkaline, with a pH of 7.8. Deep groundwaters contain 2.2 gl⁻¹ total dissolved solids with a pH of 7.5. The dominant anions are SO_4^{2-} and the dominant cation is Na⁺, sometimes with appreciable Ca²⁺ and Mg²⁺ (Figure 2.6).

North Ingebright Lake water is usually hypersaline, with salinities normally in the range of 100 to 300 g l⁻¹ total dissolved solids (TDS). Although strong seasonal variation occurs in brine chemistry, the water is alkaline (pH > 8), and strongly dominated by Na⁺, SO_4^{-2-} , and Cl⁻ and nearly completely depleted of Ca²⁺ and HCO₃⁻.

2.4 Modern Sedimentology of North Ingebright Lake

North Ingebright Lake is a salt-dominated playa. Except for a narrow, near shore fringe of organic-rich, fine grained and poorly sorted clastics, the modern sediments are almost entirely very soluble evaporitic salts (Figure 2.7). These endogenic precipitates of the modern salt pan sedimentary facies are comprised mainly of hydrated sodium and magnesium sulfates, including mirabilite (Na₂SO₄ 10H₂O), bloedite [Na₂Mg(SO₄)₂ 4H₂O], epsomite (MgSO₄ 7H₂O), and leonhardtite (MgSO₄ 4H₂O). Gypsum (CaSO₄ 2H₂O), halite (NaCl), despujolsite [Ca₃Mn(SO₄)₂ (OH)₆ 3H₂O], and protodolomite [disordered, nonstoichiometric MgCa(CO₃)₂] are also commonly present in small to trace amounts in the upper few centimeters. The inorganic fraction of the detrital sediments of the mudflat, sand flat and surrounding colluvium facies are composed of clay minerals (mostly illite and



Figure 2.7 Map showing the generalized spatial distribution of modern sedimentary facies in North Ingebright Lake

expandable lattice clays), feldspars, carbonate minerals (mainly dolomite), quartz, and ferromagnesian minerals.

Modern sedimentation in North Ingebright Lake is controlled to a major degree by the seasonal (or periodic) sequence of flooding, evaporative concentration, and desiccation of the playa surface. The process sedimentology of the basin, like many other saline playas in the region, is dominated by: (a) formation of salt crusts, efflorescent crusts, spring deposits, and intrasedimentary salts, (b) subaqueous cumulate and bottom salt precipitation, (c) physical reworking and redistribution of clastic salt and accretionary salt grains, (d) formation of salt cements, (e) irregular dissolution of surface salt crusts and formation of solution pits on the playa surface and within the near-surface sediment, (f) formation of karst chimneys, and (g) mud diapirism and subsequent reworking of finegrained siliciclastic material. These processes are discussed in detail in Last (1984, 1989, 1993).

Chapter 3 Methodology

3.1 Core Acquisition

3.1.1 Drilling Site Selection

Coring locations were determined by the following factors: expected sediment thickness, site distribution, and the need to avoid salt karst. Two drilling sites were initially chosen on the basis of a preliminary isopach map of the total salt thickness in North Ingebright Lake constructed by Cole (1926). In order to maximize the possibility of obtaining a continuous record, both sites are located on the thickest part of lake sediments where it was anticipated that the most complete and continuous stratigraphic record would likely exist (Figure 3.1).

3.1.2 Drilling Techniques

Several different drilling and coring methods were available for obtaining core samples. The decision of which to use was dependent on a number of criteria. The most important of these were: (a) the types of sediment to be recovered; (b) the types of data required; and (c) the depth of interval required for data collection. The major techniques commonly used in core acquisition for paleolimnological research are listed in Table 3.1.

Hand-operated Livingstone coring and hand augering are usually used to retrieve cores of relatively short length (4 m or less). Bombardier-mounted and power driven Shelby tube coring is designed to retrieve cores in long sequences; however, both are limited to coring relatively soft, unconsolidated muds (Last,



Figure 3.1 Map showing the sediment thickness distribution and the location of coring sites in North Ingebright Lake. North is toward the top of the map. Inset shows the location of the lake within the northern Great Plains of western Canada.

Table 3.1 Core Acquisition Techniques

1.	Hand	Operation	Technique
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Livingstone Coring Hand Auger

- 2. Shelby Tube Coring
- 3. Rotasonic Drilling
- 4. Vibra-Coring Technique
- 5. Diamond Drilling Technique

Diamond Drilling Water Cooled Diamond Drilling Air Cooled 1980). Because the sediments of North Ingebright Lake consist of very well consolidated salt, hand-operated Livingstone coring and Shelby tube coring techniques would not be able to penetrate the consolidated sediments.

Rotasonic drilling is a widely-used overburden drilling method for stratigraphic investigations and mineral exploration. With this drilling technique, a tungsten carbide-tipped core barrel string is drilled into the sediment, and then a tungsten carbide-tipped casing string is drilled outside of the core barrel string (Thorleifson and Kristjansson, 1993). The drill utilizes a combination of downhole pressure, rotary motion and ultrasonic vibration to penetrate both unconsolidated sediment and rock. However, rig time and tool rental costs make rotasonic drilling an expensive operation (Thorleifson and Kristjansson, 1993).

Vibra-coring technique, as the name implies, involves the use of intensive vibration to penetrate the sediment. This system consists of a powered concrete vibrator, drill pipe and liner or thin-walled aluminum irrigation tubing (Lanesky et al., 1979). The vibrator produces a low amplitude (0.1-1.0 mm) standing wave in the core tube. This standing wave fluidizes and displaces sediment adjacent to the core tube and allows it to pass through the sediment without resistance. Vibra-coring can only be used for acquisition of continuous cores up to 13 m in length in unconsolidated sediments (Lanesky et al., 1979).

After an extensive investigation of various drilling and coring techniques employed over the past ten years (Lanesky et al., 1979; Last, 1980; Thorleifson and Kristjansson, 1993), the wireline diamond drilling technique was chosen on the basis of equipment capability and cost.

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3.1.3 Diamond Drilling

The drilling assembly for conventional coring consists of two parts: the core head or bit and the core barrel. Unlike drill bits, core bits do not cut a complete cylinder of formation. Instead, they are designed to cut an annular ring of formation, leaving a solid cylinder of uncut formation to pass through the centre of the bit into the core barrel.

Conventional diamond drilling uses a water-based circulation fluid to carry the cuttings away from the bit and to cool it. This water-based fluid would have dissolved the salt, resulting in poor core recovery. Therefore, I used a drilling system that circulates air, rather than water, to carry the cuttings and cool the bit. This method uses standard diamond drilling drill rods and wire-line core barrel. The bit has large surface set diamonds, modified water ways, and cuts a slightly larger borehole diameter than standard bits. These modifications of the bit are made to allow the use of air, rather than water, as the circulation medium. Because of the high cost and comparatively infrequent use of diamond core bits, a drilling operation is rarely equipped with a supply in all types and sizes. Commonly the core size is 63.5 mm.

A core barrel has a reaming shell following the bit near the bottom of the core barrel, and a stabilizer at the top of the core barrel (Figure 3.2). The purpose of these is to contact the borehole walls, and provide stability to the core barrel while it is rotated. The drilling apparatus uses a sequence of casing bits and core barrels in one or more 5 ft.(1.5 m) lengths. The drill utilizes varying intensities of rotation and downhole pressure to obtain a continuous core of material ranging from soft

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Figure 3.2 Schematic diagram of a diamond drill with air circulation system.

mud to hard salts. Any changes are made slowly and carefully. Sudden changes in drilling parameters (rotary speed and pump pressure) may damage the core bit or cause the core to break. The top unconsolidated sediment was cored without pumping air, but hard salt required the core to be recovered and drilling to be resumed with air pumped down the rods and core barrels to cool the bit. In tough, compact sediments, resistance against both the outer and inner walls of the core barrel dictated that the bit must occasionally be raised so that air reaches the bottom of the hole and "lubricated" drilling. Coring was completed in 5 ft.(1.5 m) increments as the casing and rods were added. Before the core was recovered, the casing was advanced to, or nearly to, the depth of the bit. The rods and then the core barrels were recovered and the core was extruded from each successive 5 ft.(1.5 m) core barrel into 5 ft.(1.5 m) long core boxes. The core barrels were raised and the core was extruded in most cases at 5 ft. intervals. These abnormally short drilling runs were dictated by the resistance exerted by the highly compact salt against the inner wall of the core barrel. The recovered core was usually found to be shorter than drilled depth due to missing or dissolving. Depths were assigned by interpolating between the top and bottom of each run.

3.1.4 Coring

In February, 1994, two boreholes were drilled at North Ingebright Lake. The sediment cores were retrieved using Mobile S-61 Drill, mounted on a Nodwell (all terrain) carrier (Figure 3.3). A core, approximately 23 m in length, was retrieved from the first borehole, and one 9 m in length was taken from the second borehole. Core



Figure 3.3 Compressed-air diamond drill at North Ingebright Læke (February 17, 1994)

recovery averaged about 85%. This wireline diamond coring work was done by Paddock Drilling Ltd of Brandon.

Immediately upon retrieval of the core from the lake, the cores were stored in the sample box (5 feet long) and labelled with respect to site number and depth below the surface (sediment/water interface). Sealed cores were transported to the University of Manitoba for physical, chemical, and mineralogical analyses. Prior to detailed subsampling, the cores were placed in cold storage in the Department of Geological Sciences, University of Manitoba.

3.2 Laboratory Analyses

The cores were photographed, described, and subsampled in the Sedimentology lab. Figure 3.4 shows a generalized procedure of sediment description, subsampling, and lab analysis. A 2 cm subsampling interval was used for the two cores in this study. About one-half of the core was retained to provide raw, unprocessed sediment for ¹⁴C dating and additional analyses. This section of the core is also to be used for future paleolimnological research such as fluid inclusion research or biological research. No doubt the sediment record contains a wide variety of additional climatic information of which we are not yet aware, or which cannot be interpreted at present.

The other half of the core was subsampled for physical, chemical, and mineralogical analyses. Total of 1338 subsamples were prepared. These subsamples were analysed for bulk mineralogy, detailed carbonate mineralogy, detailed evaporite mineralogy, organic matter content, and total carbonate content.



Figure 3.4 Generalized procedure of sediment description, subsampling lab analysis, and data treatment.

microstructure and morphology.

The number and types of sediment parameters selected for investigation were mainly based on: (a) potential importance of decoding and characterizing the paleohydrological fluctuation and paleoclimatic change; (b) availability of equipment and technical expertise within the departmental laboratories; and (c) budgetary restraints. Table 3.2 summarizes major sediment parameters which were analysed in this research.

3.2.1 Organic Matter Content

Organic matter content was determined by measuring weight loss upon heating, known as weight loss-on-ignition (LOI). Air dried sediment subsamples were heated for one hour at 400°C-500°C. Organic matter content is calculated using the following equation: [(weight of sample before heating)-(weight of sample after heating to 500°C)/(weight of sample before heating)] x 100 (Last, 1980).

3.2.2 Total Carbonate Content

After measurement of organic matter content, the subsample was heated further to 1000°C for another hour. The total amount of carbonate minerals in the sediments was measured by the weight loss-on-combustion (LOC) of a portion of the oven-dried sediment: [(weight of sample LOI)-(weight of sample after heating to 1000°C)/(weight of dry sample before heating)] x 100. The weight loss is expressed as a percentage of the oven-dry-weight (Last, 1980).

Thermal Analyses	Mineralogy	SEM	Geochemistry	lsotope	Statistics
Organic matter	Allogenic mineralogy	Microstructure	Major ion ratio	C ¹⁴	Factor analysis
content	Carbonate mineralogy			dating	Correlative analysis
Total carbonate content	Evaporite mineralogy	Morphology	Numerical simulation		

Table 3.2 Summary of Evaluated Sediment Parameters

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Selected samples were examined using scanning electron microscopy to determine

3.2.3 Chronology

The chronostratigraphy of the recovered sequence of salts from North Ingebright Lake is only poorly constrained owing to the paucity of datable material. Large quantities of core were processed and examined in an unsuccessful attempt to find datable, well-preserved upland and shoreline macrofossils. The absence of macroscopic organic remains in the core necessitated the use of finely disseminated organic matter from two samples in order to establish a preliminary chronostratigraphy for the recovered sedimentary sequence. There are 3¹⁴C dates from the lake sequence. A ¹⁴C age of 5544 ±66 years was determined on disseminated organics from the base of the Muddy Mg-Calcite--Na-Sulfate Unit at 3.2-3.4 m depth and a sample of disseminated organics from the base of the Na-Sulfate Unit at 22.1-22.9 m depth yielded a ¹⁴C age of 10,250 ±150 years. In addition to the two radiocarbon dates from this core, a sample of endogenic carbonate material from the Ca-Sulfate--Na-Sulfate Unit at the base of a 9 m long core in the southern portion of the basin yielded a ¹⁴C date of 8240 ±120 years. Further temporal control is provided by the presence of a borate mineral zone (inderborite) at 5.6 m depth which may be an indication of a diagenetically-altered volcanic ash interval corresponding to the 6800 year B.P. Mazama eruption.

The organic matter samples from the cores were prepared for C¹⁴ dating by separating the finely disseminated organic material and clay-size material from the soluble evaporite minerals. The procedure is as follows: first, the sediment sample

is heated at 32°C until completely dehydrated and remove liquid brine. Distilled water is added and the sediment is allowed to sock for one hour. The water is then removed and the sediment sample dehydrated again. This procedure is repeated until all the soluble salt is removed.

3.2.4 Mineralogy

The mineralogy of the sediment of North Ingebright Lake was quantitatively measured by X-ray diffractometry (Philips PW1710). An unoriented sample was prepared by packing the finely powdered subsample onto a glass slide. Traces were run by using nickel-filtered copper radiation generated at 40 kilovolts and 40 milliampere, 1° beam slit and a 0.006 inch detector slit. Each slide was irradiated in one direction from 3° two theta to 65° two theta.

A diffracted beam is produced when the Bragg law or the Laue equations are satisfied. The resulting diffraction pattern of a crystal, comprising both the positions and intensities of the diffraction effects, is a fundamental physical property of substances. Each substance in the mixture produces its pattern independently of the others. The diffraction pattern indicates the state of chemical combination of the elements in the sample. It serves not only for the speedy identification but also for the complete elucidation of mineral structure (Last, 1980; Klug and Alexander, 1974; Rosyse et al., 1971; Schultz, 1964).

After the X-ray diffraction pattern was obtained, the three strongest refection peaks were converted from 20 angles to d-spacing values. A search to match the d-spacing values of the unknown pattern to those in given index book (PDF) is undertaken. The proper mineral identification can be made on the basis of the best match.

The abundance of minerals is determined from the strength of the strongest X-ray diffraction peak for each mineral on the X-ray trace of the unoriented powder. Mineral determination is aided by the use of an automated peak-match computer program-µPDSM (micro Powder Diffraction Search/Match). This computer program can carry out the entire searching-and-matching process using the reference data collected by the JCPDS-International Centre for Diffraction Data. It contains a complete package of functions designed to assist researchers in the analysis of diffraction patterns. The input information consists of the d (d-spacing) and I (intensities) values for the lines in the pattern and elemental information about the sample. The output of program lists the experimental diffraction data juxtaposed with 10-20 of the most intense lines for each standard matched (uPDSM User's Manual, 1992; Chen, 1977).

3.2.5 Geochemical Simulation

In the past decades, the thermodynamic modelling of equilibrium-reaction has experienced two generations of development. Early computer programs of chemical models include WATEQF (Truesdell and Jones, 1974; Plummer et al., 1976), WATEQ2 (Ball et al., 1979), and SOLMNEQ (Kharaka and Barnes, 1973). These models use extended Debye - Huckel activity coefficients and can match experimental data up to ionic strengths of about 10; however, for highly concentrated solutions, these extended Debye-Huckel models are in poor agreement with experimental data (Harvie and Weare, 1980; Weare, 1987).

Recently, Harvie et al. (1980; 1984) developed a new interaction model based on Pitzer's phenomenological equations which can be used to describe complex aqueous systems to high concentration and for a range of temperatures up to 250°C (Weare, 1987). This model places a larger emphasis on solubility data than other models. It is intended for the prediction of mineral solubility in natural water systems. The agreement of model phase diagram with measurements is within experimental error. With extensive validation with laboratory data, the model can be applied with confidence to field settings.

At present, there are two computer programs available for this model: ACTIVITY and PHRQPITZ. ACTIVITY was written by J. S. Hanor, Louisiana State University; PHRQPITZ was developed by L.N. Plummer and others (1988). They can be run so that a solution can be kept in equilibrium with a designated set of minerals. The model monitors how much of each mineral precipitates or dissolve. Different options, such as solute input and leakage ratio, can be chosen to fit the specific study. When the regional hydrological system surrounding a lake is well known, ACTIVITY/PHRQPITZ can generate various mineral sequences with different thickness by changing the ratio of the different source inputs. By comparing the results of the chemical model with that in the field sequence, a more lealistic scenario of the hydrological history of lake basin can be developed. In this study, the computer code PHRQPITZ was used to establish the sequence of evaporite mineral precipitation and hydrological history of North Ingebright Lake.

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3.2.6 Statistical Methods

Various statistical analyses were used to evaluate the interrelationships among the sediment parameters of North Ingebright Lake in this study. First, the linear correlation coefficient matrix (Table F-1) was calculated from a data set consisting of twenty-one mineralogical and geochemical variables from the cores of North Ingebright Lake. Second, the same data set underwent cluster analysis to test the homogeneity of the data population in each lithostratigraphic unit. Using the results of this analysis, a factor analysis of seventeen mineralogical variables was carried out. The results are discussed in Appendix F. All these multivariate statistical analyses of the sediment data set were undertaken using the various subroutines of SAS package (University of Manitoba Unix system).

Chapter 4 Lithostratigraphy of North Ingebright Lake

4.1 General Description

Playa lakes in southwestern Saskatchewan show a tremendous diversity in size, basin morphology, hydrochemistry, and sedimentary characteristics. No two playa basins have an identical stratigraphy (Last and Slezak, 1987; Last, 1983; Egan, 1984). Despite this broad variation, some generalized stratigraphic associations still can be made due to their approximately synchronous glaciation-related origins. The stratigraphy of a typical Saskatchewan hypersaline playa lake exhibits a three-layer structure: black mud at the bottom, indurated crystalline salts in the middle, and muddy salt at the top (Last and Slezak, 1987; Egan, 1984). The stratigraphic sequence retrieved from North Ingebright Lake also shows a similar three-layer structure. One core, the 23 m core, is mainly composed of indurated sulfates in the bottom and organic-rich muddy salts in the top. This core does not extend down to the bottom black mud association; however, the second core (9 m) constitutes a complete three-layer stratigraphic structure. The black mud unit directly underlies the salt unit and consists of highly reducing, fine-grained sand, silt and clay.

In general, the two sediment cores recovered from North Ingebright Lake are composed of structureless, colourless, and well indurated salt, with small amounts of mud and organic debris (Figure 4.1). The section is remarkable in its lack of obvious bedding, colour variation or other visible sedimentary features (Figure 4.1,



Figure 4.1 Structureless, colourless, and indurated salt (north core: 20.5-20.66 m) in North Ingebright Lake

Appendix E). The degree of compaction and crystal size generally increase downward. The upper part of the core is dominated by friable and fine crystalline salts, whereas at depth the sediment is very well consolidated and coarsely crystalline salts.

Based on the sediment texture, bulk mineralogy, detailed carbonate and evaporite mineralogy, and organic matter content of the cores, eight lithostratigraphic units were recognized and named on the basis of their dominant composition. Usually, the contacts between units are not visually distinctive and are gradational. Unit boundaries were defined on the basis of variation in the mineralogy/mineralogical composition. Table 4.1 and Figures 4.2-4.11 summarize the variation of major and minor mineralogical composition in eight salt-dominated stratigraphic units. When examined and analysed in detail, the salt sequence (units 1-8) can be further subdivided into 31 individual zones based on the composition of the later stage evaporate minerals (Table 4.2, Appendix C). Each zone is distinguished by a specific suite of chemical precipitates that indicate significant differences in the chemical nature of the lake at the time of their deposition.

In the following sections, characteristics of different lithostratigraphic units are summarized from two cores. However, determination and description of each mineral zone in the salt sequence is largely based on the 23 m long north core because of its continuity and central location in the lake basin. Table 4.1. Summary of average characteristics of mineral composition in North Ingebright Lake Cores (includes north and south cores; nc: not calculated; nd: not detected)

				Strat	igraphic (Jnit			
	-	2	3	4	5	9	7	8	Average
% Total Clay Minerals	40.9	0.3	6.7	12.2	6.2	0.6	6.9	3.4	5.2
% Quartz	25.2	<0.1	0.6	1.2	1.5	1.2	3.6	9.9	2.6
%Total Feldspar Minerals	14.6	0.2	1.8	0.5	11.8	3.4	1.1	2.7	3.1
% Potassic Feldspars	1.6	<0.1	0.1	0.1	0.4	<0.1	0.5	1.8	0.4
%Plagioclase	13	0.2	1.7	0.5	11.4	3.4	0.6	0.9	2.7
% Total Calcite	pu	<0.1	0.1	0.1	0.1	<0.1	0.7	0.3	0.2
% Low-Mg Calcite	pu	<0.1	<0.1	<0.1	0.1	<0.1	0.1	<0.1	<0.1
% Magnesian Calcite	pu	<0.1	0.07	0.1	0.1	<0.1	0.7	0.3	0.2
mol% MgCO ₃ in Mg-calcite	nc	nc	8.7	6.3	9.4	12.2	13.4	11.4	10.4
% Total Dolomite	4.5	0.1	1.6	2.0	1.3	0.4	3.4	11.9	3.1
% Stoichiometric Dolomite	4.1	0.1	0.3	0.6	0.6	0.2	1.8	2.0	0.8
% Protodolomite	0.4	0.1	1.2	1,4	1.7	0.3	1.6	9.9	2.3
mol% CaCO ₃ in Protodol.	nc	nc	62.6	63.8	62.5	60.3	54.3	60.1	60.9
% Magnesite	pu	0.1	1.0	1.0	0.4	<0.1	0.1	pu	0.4
% Aragonite	pu	<0.1	0.3	0.2	0.1	<0.1	0.1	<0.1	0.1
%Ca-Sulfates	3.2	1.3	3.5	17.2	5.6	0.6	6.8	1.8	5.1

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% Na-Sulfates	12.8	98.9	83.9	65.5	72.1	93.7	76.3	65.8	79.5
% Mg-Sulfates	pu	0.2	0.4	0.1	<0.1	<0.1	0.3	4.2	0.7
% Chlorides	ри	<0.1	0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1
% Pyrite	pu	pu	<0.1	ри	pu	ри	р	nd	<0.1
% Allogenic Fraction	83.2	0.53	9.4	14.5	20.1	5.4	13.3	18.2	116
% Endogenic + Authigenic	16.8	99.5	90.6	85.5	79.9	94.6	85.8	81.8	88.2
% Total Organic Matter	22.7	0.34	1.4	2.5	1.9	0.6	2.5	5.3	2.1

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	MINERAL ZONE		SALT MINERAL ASSEMBLAGES			
UNII			Major	Ancillary	Minor & Trace	
8 Dolomitic Na-	3	0.1-0.4	Bloedite Protodolomite Thenardite	Gypsum Mg-calcite	Ankerite Benstonite Leightonite Minrecordite	
Sulfate	2	0.4-0.8	Protodolomite Thenardite	Gypsum Mg-calcite	Bichofite	
	1	0.8-1.2	Thenardite	Bloedite Mg-calcite Protodolomite	Hanksite Niter	
	3	1.2-1.6	ProtodolomiteThenardite	Gypsum Mg-calcite	Arcanite Artinite Hanksite	
7 Muddy Mg-calcite	2	1.6-2.4	Gypsum Thenardite	Aragonite Bloedite Gypsum Magnesite Protodolomite	Arcanite Hanksite Mercallite Rapidcreekite	
Na-Sulfate	1	2.4-3.4	Gypsum Thenardite	Mg-calcite Protodolomite	Benstonite Epsomite Görgeyite Hanksite Mercallite Polyhalite	
6	3	3.4-3.9	Thenardite	Aragonite Gypsum Protodolomite	Hanksite Tychile	
Na-Sultate	2	3.9-4.1	Thenardite	Gypsum Protodolomite	Niter	
	1	4.1-4.5		Protodolomite	Hanksite Niter	
	5	4.5-4.9	Thenardite	Mg-calcite Magnesite Protodolomite	Hanksite	
5	4	4.9-5.6	Gypsum Thenardite	Aragonite Mg-calcite Protodolomite	Arcanite Hanksite Inderborite	
Muddy Na-Sulfate	3	5.6-6.2	Thenardite	Gypsum Magnesite Protodolomite	Arcanite Gaylussite Hanksite Sepiolite Wattevilleite	
	2	6.2-7.2	Gypsum Thenardite	Aragonite Protodolomite	Arcanite Hanksite Polyhalite Sepiolite	
			Gypsum Protodolomite	Aragonite Magnesite Mg-	Burkeite Epsomite Hanksite	
	1	7.2-8.2	Thenardite	calcite	Mercallite Görgeyite Sepiolite	

Table 4.2 Stratigraphic units and mineral zones in North Ingebright Lake cores.

.

	5	8,2-9,4	Gypsum Thenardite	Aragonite Gypsum Magnesite Protodolomite	Carnallite Hanksite Polyhalite
	4	9.4-10.2	Gypsum Magnesite Thenardite	Aragonite Halite Mg-calcite Protodolomite	Hanksite Mercallite Polyhalite Görgeyite Rapidcreekite Sepiolite
4 Ca-Sulfate Na-Sulfate	3	10.2-10.5	Gypsum Thenardite	Aragonite Magnesite Protodolomite	Hanksite Mercallite Polyhalite
	2	10.5-10.9	Gypsum Thenardite	Aragonite Magnesite Protodolomite	Hanksite Polyhalite
	1	10.9-11.7	Gypsum Thenardite	Aragonite Halite Magnesite Mg-calcite Protodolomite	Hanksite Rapidcreeksite Sepiolite
	7	11,7-13,4	Gypsum Magnesite Thenardite	Aragonite Protodolomite	Hanksite
3	6	13.4-14.1	Gypsum Magnesite Thenardite	Aragonite Bloedite Halite Mg- calcite Protodolomite	Arcanite Eugsterite Hanksite Rapidcreekite
Mg-Carbonate Na-Sulfate	5	14.1-15.1	Gypsum Protodolomite Thenardite	Aragonite Magnesite	Hanksite Görgeyite
	4	15.1-15.8	Gypsum Thenardite	Aragonite Magnesite Protodolomite	Gaylussite
	3	15,8-16,4	Magnesite Thenardite	Aragonite Protodolomite	Bradleyite Hanksite Pyrite
	2	16,4-17,4	Gypsum Magnesite Protodolomite Thenardite	Mg-calcite	Hanksite Rapidcreekite
	1	17.4-17.6	Thenardite		Görgeyite
	5	17.6-19.5	Thenardite/gypsum	Magnesite Protodolomite	Hanksite
	4	19.5-20.6	Thenardite	Halite	Hanksite Nitratine Nitrobarite
2 Na-Sulfate	3	20.6-21.1	Thenardite		Nitratine Protodolomite

	2	21.1-22.2	Thenardite	Gypsum	
	1	22.2-23.0	Thenardite		
1 Basal Black Mud		8.96-9.16 m (south core)	Gypsum	Thenardite	

4.2 General Stratigraphic Variation of Mineralogical Constituents

The mineralogy of detrital (allogenic) components in sequence shows a varying trend; it is high in Units 1, 5 and 8, low in Units 2 and 6 (Figures 4.2 and 4.3). The content of quartz and K-feldspar in the sediment core displays an upward increasing trend (Figures 4.4 and 4.5). The percentage of plagioclase and total clay minerals do not show any systematic change with depth. Overall, the content of plagioclase and total clay minerals is high in Units 1, 3, 5 and 8, low in Units 2 and 6 (Figures 4.4 and 4.5).

The endogenic+authigenic fraction, crystallized in-situ in the basin, dominates the sedimentary sequence (average: 88%; Figures 4.2 and 4.3; Table 4.1), and consists of a complex assemblage of mainly hydrated Na, Mg, Na+Mg, and Ca sulfates, with smaller amounts of Mg, Ca+Mg, Na, and Ca carbonates and chlorides (Figures 4.6, 4.10 and 4.11). The concentrations of endogenic+authigenic components display systematic changes with depth; they are relatively low in Units 1, 5, and 8, high in Units 2 and 6 (Figures 4.2 and 4.3). As can be seen in Figures 4.6 and 4.7, the amount of thenardite is high in Units 2 and 6, low in Units 1, 5, and 8, and average in Units 3, 5 and 7 (Figures 4.6 and 4.7). The amount of gypsum fluctuates significantly but there is no obvious trend with depth (Figure 4.6). On average, it is low in Units 2 and 8 and high in Units 1 and 4. Anhydrite occurs only in the upper 3.4 m of sediment and at 8.75 m (Figure 4.6) and its concentration increases upward.



Figure 4.2 Stratigraphic variation in proportion of allogenic material versus endogenic plus authigenic components in the north core of North Ingebright lake



Figure 4.3 Stratigraphic variation in proportion of allogenic material versus endogenic plus authigenic components in the south core of North Ingebright lake core
Bloedite is common in sediment above 3.4 m and below 8.2 m in the north core, but is only sporadically present below that and shows no consistent trends with depth (Figures 4.6 and 4.7).

The carbonate mineral suite of North Ingebright Lake consists of protodolomite, magnesite, dolomite, high-Mg calcite, aragonite, and low-Mg calcite. The total amount of carbonates increases slightly upward and is well correlated with the organic matter content (Figures 4.8 and 4.9). Most of the carbonate minerals show no stratigraphic variation trend. Mg-carbonates are most abundant in Units 3, 4, and 5, whereas the carbonates in the upper meter of section are dominated by disordered, nonstoichiometric dolomite (Figures 4.10 and 4.11). The trace amounts of low-Mg calcite and aragonite are only present in Units 3, 4, 5, and 6 (Figures 4.10 and 4.11). The percentages of high-Mg calcite and magnesite show little systematic fluctuation with depth. Magnesite occurs only in the sediment of Units 1, 3, 4, and 5, whereas high-Mg calcite is present in each of eight units and increases slightly upward (Figures 4.10 and 4.11). The contents of both dolomite and protodolomite increase upward in the cores and are significantly correlated with quartz and K-feldspar (Figures 4.4, 4.5, 4.10, and 4.11).

4.3 Lithostratigraphy

4.3.1 Unit 1 Basal Black Mud (south core: 8.96-9.16 m)

Unit 1 was penetrated only in the 9 m core. Approximately 20 cm of black mud of Unit 1 was retrieved and consists mainly of black, highly reducing, finegrained sand, silt and clay. It is easily recognized by a high ratio of detrital to



Figure 4.4 Stratigraphic variation in percentages of quartz, K-feldspar, plagioclase, and clay minerals in the north core of North Ingebright Lake







in the north core of North Ingebright Lake







Figure 4.8 Variation of organic matter and carbonate content with depth in the north core of North Ingebright Lake



Figure 4.9 Stratigraphic variation in percentages of organic matter and carbonate content in the south core of North Ingebright Lake







endogenic components and the high content of organic matter (up to 25%) relative to the overlying sediments. Minerals in the unit are dominantly clays (40.9%), quartz (25.2%), and feldspar (27.6%), but also present are gypsum, thenardite/mirabilite, quartz, feldspar, and dolomite (Table 4.1).

4.3.2 Unit 2 Na-Sulfate Salt (north core: 17.60-23.00 m; south core: 7.00-8.96 m)

Unit 2 is characterized by a simple mineral composition. It consists almost entirely (98.9%) of Na sulfate, with 1.3% Ca sulfate, 0.2% Mg sulfate, and <0.3% of all other components (Table 4.1). The major minerals are thenardite/mirabilite and gypsum with minor amounts of halite, protodolomite, and dolomite (Figures 4.2-4.11). The endogenic/allogenic ratio of Unit 2 is the highest one among the eight stratigraphic units (Figure 4.2 and 4.3). Relative to the overlying units of North Ingebright Lake evaporites, this Na-sulfate salt unit is much thicker (1.9-5.4 m). The Na-sulfate salt unit is divided into five stratigraphic zones whose boundaries separate deposits characterized by specific assemblages of evaporite minerals, as can be seen in Table 4.2).

Zone 1 Mirabilite zone (22.20-23.00 m)

Zone 1 consists mostly of pure thenardite/mirabilite. Traces of nitratine and hanksite occur in a few samples. The preponderance of thenardite/mirabilite over other saline minerals and lack of variation distinguish this zone from zones above it.

Zone 2 Gypsum zone (21.10-22.20 m)

Gypsum appears in the lower part of zone 2, and this characteristic distinguishes zone 2 from zone 1. Locally there are trace quantities of nitratine and hanksite.

Zone 3 Mirabilite zone (21.00-21.10 m)

Salts in zone 3 are dominated by thenardite/mirabilite and contain traces of protodolomite and nitratine.

Zone 4 Halite zone (19.50-21.00 m)

In zone 4, organic matter content is relatively low and carbonate minerals are absent. Most samples contain minor amounts of halite, and this distinguishes the deposits from those of zone 3. Traces of hanksite, nitratine and nitrobarite also occur in several samples.

Zone 5 Magnesite zone (17.60-19.50 m)

Zone 5 consists chiefly of thenardite/mirabilite and gypsum. Protodolomite, magnesite, and fairchildite are commonly found. Trace amounts of hanksite are also detected in several samples.

4.3.3 Unit 3 Mg-Carbonate--Na-Sulfate Salt (north core: 11.70-17.60 m; south core: 5.00-7.00 m)

Unit 3 is dominantly Na sulfate (83.9%) and clay (12.2%); there is <5% other silicate minerals and 3.5% Ca sulfate (Table 4.1). It is recognized by a relatively high content of magnesite, gypsum, bloedite, and feldspar (Figures 4.4-4.11). The

endogenic/allogenic ratio (Figures 4.2 and 4.3) is relatively low, and the organic matter content is considerably higher than that of Unit 2 (Figures 4.8 and 4.9). The total carbonate content is also higher than that of the underlying sediments. Carbonates are dominated by magnesite and protodolomite, with small amounts of high-Mg calcite and dolomite (Figures 4.10 and 4.11). Trace amounts of starkeyite, eugsterite, hanksite, and niter occur in few samples (Appendix C). Small quantities of quartz and clay minerals also appear in the mud.

Unit 3 in the north core is divided into seven mineral zones on the basis of detailed evaporite mineralogy. The most significant changes in mineral composition of Unit 3 are as follows: the lower four zones consist mainly of thenardite/mirabilite; zones 5, 6, and 7 consist of thenardite/mirabilite and gypsum with some detrital materials.

Zone 1 Mirabilite zone (17.40-17.60 m)

Zone 1 is almost entirely composed of mirabilite/thenardite with traces of görgeyite. Fine grained detrital materials commonly appear built form less than 10%.

Zone 2 Starkeyite-gypsum zone (16.40-17.40 m)

The main saline minerals in zone 2 are thenardite/mirabilite, gypsum, protodolomite, aragonite, and magnesite. Gypsum and magnesite are concentrated in the middle part, whereas thenardite/mirabilite, pro-todolomite, and aragonite appear in the lower and upper parts. Calcite and deolomite are found in several samples. Trace amounts of starkeyite, nitratine, hanksite,

and rapidcreekite are also determined in few samples.

Zone 3 Magnesite zone (15.80-16.40 m)

Gypsum is absent in zone 3, and this feature distinguishes it from both zone 2 and zone 4. Thenardite/mirabilite and magnesite are predominant minerals; the average percentage of protodolomite is much higher than that of the underlying zones. Small amounts of aragonite occur in a number of samples. Trace minerals include nitratine, hanksite, and bradleyite.

Zone 4 Gaylussite zone (15.10-15.80 m)

Zone 4 is mainly composed of thenardite/mirabilite, gypsum, protodolomite, aragonite, and magnesite. Mud impurities are common in this zone. Trace amounts of gaylussite and fairchildite are found respectively at 15.5 m, 15.1 m, and 15.2 m depths.

Zone 5 Starkeyite zone (14.10-15.10 m)

Zone 5 is similar to zone 2 in terms of mineral composition. Like zone 2, it consists largely of thenardite/mirabilite, gypsum, protodolomite, aragonite, and magnesite, with traces of hanksite, görgeyite, starkeyite, fairchildite, and nitratine.

Zone 6 Bloedite-eugsterite (13.40-14.10 m)

Zone 6 is characterized by relatively high organic matter, carbonates, and bloedite contents. Sulfate minerals consist mainly of thenardite/mirabilite, gypsum, and bloedite; carbonate minerals include high Mg calcite, aragonite, protodolomite, and magnesite. Magnesite reaches its highest concentration in this zone; much of the magnesite is concentrated near the base of the zone. Trace amounts of halite, eugsterite, görgeyite, hanksite, rapidcreekite, arcanite, inderborite, nitrobarite, and phosphate minerals are also detected.

Zone 7 Gypsum zone (11.70-13.40 cm)

Like zone 6, zone 7 consists chiefly of thenardite, gypsum, protodolomite and magnesite, with small amounts of aragonite and dolomite. Bloedite is absent. Trace minerals include hanksite and nitratine.

4.3.4 Unit 4 Ca-Sulfate--Na-Sulfate Salt (north core: 8.20-11.70 m; south core: 3.90-5.00 m)

Unit 4 is characterized by high gypsum (3.%) and total clay mineral (12.2%), relatively low thenardite/mirabilite (65.5%) and plagioclase contents (0.5%), and high magnesite content (Figures 4.2-4.11 and Table 4.1). Trace amounts of hanksite, polyhalite, gaylussite, carnallite, and burkeite are found mixed with the other saline minerals. Unit 4 also displays relatively high organic matter and total carbonate contents (Figures 4.8 and 4.9). Generally, organic matter content, quartz, and total clay minerals increase upward, whereas gypsum and thenardite/mirabilite decrease upward. Unit 4 has been subdivided into five zones on the basis of compositional changes in the saline mineral assemblage.

Zone 1 Halite zone (10.90-11.70 m)

Thenardite/mirabilite and gypsum are the major mineral components of zone 1, together forming more than 85% of the total mineral composition. Small amounts of quartz, clay minerals, high-Mg calcite, aragonite, protodolomite, and magnesite occur. Sepiolite, fairchildite, halite, hanksite, rapidcreekite, nitratine, and inderborite constitute the trace mineral assemblage.

Zone 2 Polyhalite zone (10.50-10.90 m)

Similar to zone 1, zone 2 is composed mainly of Na and Ca sulfate salts (mirabilite/thenardite, gypsum) with lesser amounts of Ca, Mg and K carbonates (aragonite, protodolomite, dolomite, magnesite, and fairchildite). A wide variety of other soluble minerals have been identified in trace amounts including polyhalite, starkeyite, and hanksite.

Zone 3 Gaylussite zone (10.20-10.50 m)

Zone 3 is distinguished from the underlying zones by the presence of trace amounts of gaylussite at the base. This zone is composed of thenardite/mirabilite, clay minerals, gypsum, and protodolomite. Minor minerals include aragonite, dolomite, and magnesite. Mercallite, polyhalite, hanksite, and phosphate minerals constitute rare mineral components.

Zone 4 Polyhalite zone (9.40-10.20 m)

Zone 4 is characterized by the common occurrences of trace polyhalite, hanksite, and halite. Quantities of quartz, protodolomite, and thenardite/mirabilite are relatively low, but gypsum and magnesite are very high. Other minor minerals include high Mg calcite, aragonite, dolomite, sepiolite, mercallite, görgeyite, and rapidereekite.

Zone 5 Carnallite zone (8.20-9.40 m)

Main minerals include carnallite, burkeite, and bloedite. Thenardite/mirabilite, gypsum, and clay minerals constitute major minerals. Minor minerals include aragonite, protodolomite, dolomite, magnesite, and anhydrite. Hanksite, polyhalite, inderborite, arcanite, and phosphate minerals are present in trace amounts. Organic matter and detrital components are higher than any of the other zones within this unit.

4.3.5 Unit 5 Muddy Na-Sulfate Salt (north core: 4.50-8.20 m; south core: 2.50-3.90 m)

The sediment of Unit 5 is dominated by mirabilite/thenardite and plagioclase, with higher protodolomite and dolomite contents relative to the underlying unit (Figures 4.6-4.11). The organic matter content of Unit 5 is also relatively high (Figures 4.8 and 4.9). Muddy material, approximately 10%, is scattered among the structureless and colourless salts. Five stratigraphic zones are identified on the basis of the appearance of specific late-stage evaporite minerals.

Zone 1 Epsomite zone (7.20-8.20 m)

Zone 1 is characterized by the presence of epsomite, starkeyite, and burkeite. Thenardite/mirabilite, gypsum, protodolomite, and clay minerals constitute major minerals. Minor minerals include high Mg calcite, aragonite, dolomite, and magnesite. Sepiolite, görgeyite, mercallite, hanksite, and phosphate minerals are present in trace amounts.

Zone 2 Polyhalite zone (6.20-7.20 m)

Zone 2 is distinguished from zone 1 by the occurrence of polyhalite and hanksite. This zone is mainly composed of thenardite/mirabilite and gypsum, with minor amounts of aragonite, protodolomite, and dolomite. Trace minerals include sepiolite, arcanite, and phosphate minerals.

Zone 3 Gaylussite zone (5.60-6.20 m)

Zone 3 is identified by the appearance of gaylussite. Thenardite/mirabilite and clay minerals are major components. Minor minerals include protodolomite, dolomite, magnesite, and gypsum. Hanksite, wattevilleite, arcanite, and sepiolite are present in trace amounts.

Zone 4 Inderborite zone (4.90-5.60 m)

Zone 4 is characterized by the occurrence of inderborite. Major minerals include thenardite/mirabilite, gypsum, plagioclase, and clay minerals. High Mg calcite, aragonite, protodolomite, and dolomite constitute minor minerals. Trace minerals include arcanite and hanksite.

Zone 5 Mirabilite zone (4.50-4.90 m)

Zone 5 is distinguished from zone 4 by the presence of görgeyite and hanksite. This zone consists mainly of thenardite/mirabilite, with minor amounts of high Mg calcite, protodolomite, dolomite, and magnesite.

4.3.6 Unit 6 Na-Sulfate Salt (north core: 3.40-4.50 m; south core: 1.20-2.50 m)

Similar to Unit 2, Unit 6 is composed almost entirely of thenardite/mirabilite

(93%), with lesser amounts of plagioclase (3.4%). The sediment of Unit 6 is also recognized by its relatively low protodolomite and gypsum contents, the absence of high-Mg calcite and magnesite, and a low organic matter content (Figures 4.2-4.11). The endogenic/allogenic ratio is 17 (Figures 4.2 and 4.3). Three sedimentation zones have been delimited on the basis of the trace mineral assemblage.

Zone 1 Kainite zone (4.10-4.50 m)

Zone 1 is mainly composed of thenardite/mirabilite, plagioclase and other minerals, including protodolomite and dolomite. Trace minerals consist of kainite, hanksite, and niter.

Zone 2 Niter zone (3.90-4.10 m)

Zone 2 is distinguished from zone 1 by the common occurrence of trace niter. Major mineral is thenardite/mirabilite, with minor amounts of gypsum and protodolomite.

Zone 3 Tychite zone (3.40-3.90 m)

This zone is characterized by thenardite/mirabilite with disseminated minerals of protodolomite, aragonite and gypsum. Tychite and hanksite are present in trace amounts.

4.3.7 Unit 7 Muddy Mg-calcite-Na-Sulfate Salt (north core: 1.20-3.40 m; south core: 0.50-1.20 m)

Unit 7 is black, relatively poorly consolidated, and muddy salts. Non-evaporite material content is approximately 30-40%. Sediment of Unit 7 is characterized by

higher protodolomite, high-Mg calcite, and gypsum contents (Figures 4.5, 4.6, 4.10, and 4.11). The organic matter and carbonate contents are also high compared to the underlying stratigraphic units (Figures 4.8 and 4.9). Three stratigraphic zones are recognized within Unit 7.

Zone 1 Polyhalite zone (2.40-3.40 m)

Zone 1 consists chiefly of thenardite, gypsum, protodolomite, and clay minerals, with minor amounts of dolomite and high Mg calcite. Traces of fairchildite, benstonite, kutnohorite, epsomite, mercallite, görgeyite, polyhalite, and hanksite also occur. The organic matter and carbonate contents are relatively high.

Zone 2 Anhydrite zone (1.60-2.40 m)

Zone 2 consists thenardite/mirabilite, gypsum, protodolomite, with small amounts of aragonite, magnesite, anhydrite, detrital, and organic matter, and traces of niter, mercallite, arcanite, leightonite, rapidereekite, and hanksite. The amount of the dominant mineral, thenardite/mirabilite, is low relative to the underlying and overlying zones. Bloedite occurs at 1.7 and 1.8 m depths.

Zone 3 Gypsum zone (1.20-1.60 m)

Thenardite and protodolomite are major components. High-Mg calcite, dolomite, and gypsum constitute minor components. Trace minerals include artinite, hanksite, leightonite, and arcanite.

4.3.8 Unit 8 Dolomitic Na-Sulfate Salt (0.12-1.20 m)

This is the uppermost unit in the sequence, and is black, nonbedded, and loosely consolidated muddy salt. The mineral composition of Unit 8 is distinctively different from the other units in this sequence. The top 25 cm of Unit 8 consists mainly of non-evaporite materials. Mud and organic matter not only fill the spaces between crystalline salt grains but sediment and organic debris inclusions are also common within individual crystals.

Unit 8 is characterized by several general trends. Quartz, feldspar, clay minerals, organic matter, protodolomite, and bloedite contents all increase markedly upward. Conversely, thenardite and gypsum contents decrease upward. The endogenic/allogenic ratio is relatively low (4.5)(Figures 4.2 and 4.3). The organic matter and carbonate contents are much higher than that of underlying units (Figures 4.8 and 4.9). On the basis of mineralogical composition, three stratigraphic zones are recognized in Unit 8.

Zone 1 Thenardite zone (0.80-1.20 m)

Zone 1 is distinguished by high percentages of thenardite/mirabilite. Extremely high bloedite content occurs at 1.1 m depth. High-Mg calcite, protodolomite, and detrital components are generally low, but variable.

Zone 2 Bichofite zone (0.40-0.80 m)

Zone 2 is identified by the appearance of trace amounts of bichofite. The main minerals are thenardite and protodolomite, with minor amounts of high Mg calcite and anhydrite.

Zone 3 Anhydrite zone (0.10-0.40 m)

Zone 3 is distinguished from the underlying zones by its wide variety of mineral constituents: major components include thenardite/mirabilite, bloedite, and protodolomite; minor components consist of high-Mg calcite, dolomite, gypsum, and anhydrite; and trace components are niter, starkeyite, minrecordite, benstonite, and leightonite. The thenardite/mirabilite reaches its lowest in the entire section at zone 3. Conversely, organic matter, quartz, and carbonate contents reach their highest percentage.

4.4 Stratigraphic Correlation Within North Ingebright Lake Basin

As is well known, evaporites form in a continuum of depositional environments, ranging from nonmarine to marine. Three main environmental groups of evaporite precipitation are identified (Shreiber et al., 1976; Kendall, 1984; Warren, 1986; Hsü et al., 1969; Hsü, 1972): continental, coastal sabkha, and subaqueous. However, very few examples of ancient continental facies evaporites have been identified in the geological record (Hardie et al., 1978, Rosen, 1991). Most ancient evaporites have been interpreted as marine in origin (Kinsman, 1969). The establishment of a facies model for many bittern salt deposits in continental environments is extremely difficult because of the lack of original fabrics and structures. No single facies model can be applied to diversified evaporites. Accordingly, stratigraphic correlations of evaporite sequences are very difficult to determine. Stratigraphic correlations within evaporite sequences, and between evaporites and other facies are affected by the lack of biostratigraphic control, rapidity of evaporite deposition, complexity of stratigraphic relationships, and removal of soluble evaporites by groundwaters and surface solution (Kendall, 1988; Last, 1990; Handford, 1982). The complex products resulting from the effect of these factors make stratigraphic correlations of an evaporite sequences one of most challenging and controversial subjects in sedimentology.

The difficulty is best illustrated by the attempt to make stratigraphic correlations in the evaporite sequences of Bristol Dry Lake Basin (Rosen, 1991). Bristol Dry Lake is a large playa (155 km²) in the Mojave Desert, USA, filled with over 500 m of sediment. Two continuous cores over 400 m in length were retrieved from lake basin less than 3 km apart. Unfortunately, no beds in the salt sequences of the two cores can be correlated. Correlation with other short cores taken virtually in the same site is also nearly impossible.

Two cores were retrieved from the North Ingebright Lake Basin. The sediments consist of well-indurated salt, with only minor mud and organic debris. The general lack of organic matter, obvious bedding, colour variation or other visible sedimentary structures in the sediments indicates that oxygenated, well-mixed, and shallow water conditions dominated in lake basin. Further, the poorly developed marginal facies and steeply-walled morphology of the North Ingebright Lake Basin suggest that the vertical sequences in the evaporites most likely reflect basin-wide changes in evaporite depositional environments, rather than lateral facies

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relationships. Therefore, stratigraphic correlations can be established by comparing variation in mineralogical composition reflected in the cores. Figure 4.12 shows the stratigraphic correlation of two cores based on the mineral composition. The good correlation of the salt beds between the two cores suggests that the sequence found in the two cores depicts a series of changes that probably affected the entire basin simultaneously.



Figure 4.12 Subsurface Section Showing Relationship between Stratigraphic Units of North Ingebright Lake

Chapter 5 Mineralogy of North Ingebright Lake Core

5.1 General Description

The mineral suite of the sediment retrieved from the basin of North Ingebright Lake consists mainly of thenardite/mirabilite, gypsum, carbonates, clay minerals, feldspar minerals, and quartz (Tables 4.1 and 4.2). Thenardite and gypsum are two dominant minerals in most samples throughout the entire stratigraphic section. Carbonate minerals present in the sediment core include protodolomite, dolomite, magnesite, aragonite, high-Mg calcite, and calcite. The feldspars comprise an average of 3% of the mineral suite, primarily plagioclase, with minor amounts of potassic feldspar. Small quantities of halite and pyrite are present in many samples. Anhydrite and bloedite were also identified in a small number of samples. Most of these samples are from the upper part of the sediment core. Trace quantities of nitrate, borate, carbonate-sulfate, carbonate-sulfate-chloride minerals, Na+Ca (or K+Ca, Ca+Fe, Ca+Mn) carbonates, and Na+Mg (or Mg, K, Fe, Na+Ca, K+Ca+Mg) sulfates occur locally (Table 5.1, Appendix C).

5.2 Origin of Mineralogical Constituents

The mineral components of the North Ingebright Lake sediment cores can be divided into three genetic types: allogenic, endogenic, and authigenic. Minerals derived from outside the lake are referred to as allogenic (Jones and Bowser, 1978). Allogenic components in the North Ingebright lake sediments consist of clay minerals, quartz, feldspar and some of the carbonate minerals. These minerals are derived from the surrounding watershed sediments by means of fluvial, sheetwash, and aeolian transport and closely reflect the composition of the surrounding glacial materials (David, 1964). The proportions of all of these detrital minerals are significantly correlated (linear correlation coefficients, γ , are more than 0.5).

The endogenic fraction is that portion of the sediment that originated from within the water column of the lake, whereas authigenic components are derived from syndepositional reactions of early formed minerals with the evolving brine or pore water (Jones and Bowser, 1978; Last, 1993, 1997). The paragenesis of the endogenic and authigenic mineral suite in North Ingebright Lake is exceedingly complex. In a groundwater-dominated, hypersaline salt pan/playa environment such as is present in the modern basin, the distinction between endogenic and authigenic mineral components becomes obscure (see also discussions in Smoot and Lowenstein, 1991; and Sonnenfeld, 1984; Taibot and Kelts, 1986). Salts that may have formed as true endogenic precipitates are very likely dissolved and reprecipitated numerous times at the sediment-water interface as the composition of the overlying brine changes during the year, or at the surface of the playa upon subaerial exposure and desiccation (Last, 1997). Likewise, both fresh and saline pore waters within the upper few centimetres of the playa surface can readily modify the precipitated salts. Although seasonal and longer-term mineralogic changes and accompanying diagenesis of the non-detrital suite in North Ingebright sediment is a complex process, observations from numerous other saline playas in western Canada and other areas of North America (e.g., Last, 1989; Smith and Friedman,

1986) indicate that the evaporite mineral suite is largely stabilized within a few tens of centimetres below the playa surface.

The carbonate mineral suite of North Ingebright Lake consists mainly of magnesite, protodolomite, dolomite, high-Mg calcite, aragonite, and low-Mg calcite. Non-stoichiometric dolomite (protodolomite), high-Mg calcite, aragonite, and magnesite are absent from the bedrock, till, and surface soils of Western Canada (Last, 1982). For this reason, it is reasonable to conclude that these metastable minerals in the sediment of North Ingebright Lake were derived from inorganic precipitation within the water column in response to supersaturated conditions or by diagenetic reactions. In this study, stoichiometric dolomite and low-Mg calcite are considered detrital in origin, being derived from erosion of the surrounding glacial deposits or bedrock. Most of the carbonate minerals and the soluble late stage evaporitic minerals in the North Ingebright Lake sediment are considered as either endogenic or early authigenic in origin.

5.3 Significance of Carbonate Mineralogy in North Ingebright Lake

Origins of the carbonates in lacustrine sediments can be detrital, endogenic, and/or authigenic. Carbonate minerals can be produced by inorganic precipitation, by biologically-induced inorganic precipitation, and/or by purely organic precipitation (Last, 1982; Last et al., 1994; Last and Deckker, 1992). For hypersaline lakes, carbonate minerals are dominantly derived by inorganic precipitation because these waters are generally deficient in microorganisms. With evaporative concentration and calcite precipitation, the Mg content of the lake water and the Mg/Ca ratio will increase. Müller et al. (1972) have outlined the relationship between the Mg/Ca ratio in waters and associated carbonate formation. With an increase in the Mg/Ca ratio, calcite, low-Mg calcite, high-Mg calcite, aragonite and dolomite may precipitate in turn (Müller, 1970). Thus, it is commonplace to use the stratigraphic variation of these species in a lacustrine basin to deduce past ionic ratios and the salinity of the lake water (Last, 1982; Last and Schweyen, 1985).

The presence of endogenic carbonates throughout the entire sediment core of North Ingebright Lake indicates the past presence of a pH of at least 7.0. There are several high peaks of high-Mg calcite, protodolomite, and magnesite contents in the sediment core, which imply that North Ingebright Lake experienced periodic or extended episodes where the Mg/Ca ratio was high. During the deposition of the north core at 8.2-17.6 m depths, the Mg/Ca ratio was around 100 because carbonate minerals in these sections of the core mainly consist of magnesite. During the deposition of the north core at 0.1-3.4 m depths, the Mg/Ca ratio was also very high (>10) as indicated by the presence of high-Mg calcite and protodolomite. Generally, the Mg/Ca ratio displays a slightly systematic upward increase, as shown by the variation of MgCO₃ in protodolomite and high-Mg calcite with depth (Figure 5.1). Therefore, the water chemistry experienced an overall gradual shift from a sodium dominated brine to one of mixed sodium and magnesium.

White/dark laminae may form in perennial closed-basin lakes. The white laminae are usually calcareous sediment (calcite or aragonite), whereas the darker laminae are organic-rich with admixtures of clay, quartz and calcite (Eugster and

MgCO3 CONTENT



Figure 5.1 Variation of MgCO₃ content in high Mg-calcite (A) and protodolomite (B) with depth at North Ingebright Lake

Kelts, 1983; Kelts and Hsü, 1978; Lambert and Hsü, 1979). In ephemeral North Ingebright Lake basin, however, seasonal lowering of the lake waters might lead to subaerial exposure, mudcracking and intrasedimentary growth of saline minerals. These processes can completely destroy laminations (Eugster and Kelts, 1983). Therefore, the subaerial exposure, mudcracking and saline minerals intrasedimentary growth probably best explains the lack of carbonate laminae in North Ingebright Lake, which are commonly found in other closed-basin lakes in southern Saskatchewan (e.g. Ceylon Lake, Little Manitou Lake)(Last, 1990; Sack and Last, 1994). Overall, the carbonate content in the sediments of North Ingebright Lake is very low (3%). The periodic appearance of the small percentage of aragonite, protodolomite, and magnesite probably resulted from brief changes in the chemical composition of lake waters caused by an influx of freshwater.

5.4 Soluble Evaporitic Mineralogy in North Ingebright Lake

Over 40 species of endogenic+authigenic minerals have been identified in North Ingebright Lake deposits (Table 5.1). Most of the late stage evaporitic minerals in North Ingebright Lake were formed by syndepositional reaction of early formed minerals with the evolving brine. These reaction minerals should be regarded as a proper part of a primary depositional sequence (Braitsch, 1971; Dean, 1978). However, exact origins (primary or syndepositional) of these minerals are not essential for the deduction of hydrological and chemical conditions. The important thing is their existence because each mineral represents particular hydrological-chemical conditions (Braitsch, 1971; Bradley et al., 1969; Dean, 1978;

Mineral Name	Composition	Occurrence ¹	
Carbonate Minerals			
Aragonite	CaCO ₃	common	
Artinite	Mg ₂ CO ₃ (OH) ₂ 3H ₂ O	very rare	
Ankerite	Ca(Fe,Mg)(CO ₃) ₂	rare	
Benstonite	Ca ₇ Ba ₆ (CO ₃) ₁₃	very rare	
Calcite	CaCO ₃	rare	
Fairchildite	K ₂ Ca(CO ₃) ₂	very rare	
Gaylussite	Na ₂ Ca(CO ₃) ₂ 5H ₂ O	very rare	
Kutnohorite	Ca(Mn,Mg)(CO ₃) ₂	very rare	
Magnesite ²	MgCO ₃	common	
Magnesian Calcite	(Mg _x Ca _{1-x})CO ₃	common	
Minrecordite	CaZn(CO ₃) ₂	very rare	
Protodolomite	CaMg(CO ₃) ₂	very common	
Zemkorite	Na ₂ Ca(CO ₃) ₂	very rare	

Table 5.1. Endogenic and Authigenic Mineralology of North IngebrightLake Sediments Identified by XRD

Sulfate Minerals

Arcanite	$(K, NH_4)_2 SO_4$	rare
Bloedite	$Na_2Mg(SO_4)_2 4H_2O$	very common
Epsomite	MgSO₄ 7H₂O	common
Eugsterite	$Na_4Ca(SO_4)_3 2H_2O$	very rare
Görgeyite	K₂Ca₅(SO₄) ₆ 4H₂O	very rare
Gypsum	CaSO₄ 2H₂O	very common
Krausite	$Fe_2(SO_4)_3 2H_2O$	very rare
Leightonite	K₂Ca₂Cu(SO₄)₄2H₂O	very rare

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Mercallite	KHSO₄	very rare
Mirabilite ³	Na₂SO₄ 10H₂O	very common(?)
Polyhalite	K₂Ca₂Mg(SO₄)₄2H₂O	rare
Starkeyite	MgSO₄ 4H₂O	very rare
Thenardite	Na₂SO₄	very common
Wattevilleite	$Na_2Ca(SO_4)_2 4H_2O$	very rare

.

Carbonate-Sulfate, Carbonate-Sulfate-Chloride, and Carbonate-Phosphate

Minerals		
Bradleyite	Na₃Mg(PO₄)(CO₃)	very rare
Burkeite	2(Na₂SO₄)Na₂CO₃	very rare
Hanksite	KNa ₂₂ (CO ₃) ₂ (SO ₄) ₉ Cl	common
Kainite	KMg(SO₄)Cl3H₂O	rare
Rapidcreekite	Ca ₂ (CO ₃)SO ₄ 4H ₂ O	very rare
Tychite	Na₀Mg₂SO₄(CO₃)₀	very rare

Chloride Minerals

Bischofite	MgCl ₂ 6H ₂ O	very rare
Carnallite	KMgCl ₃ 6H ₂ O	very rare
Halite	NaCl	rare
Korshunovskite	Mg ₂ Cl(OH) ₃ 4H ₂ O	very rare

Nitrate and Borate Minerals

Inderborite	CaMgB ₆ O ₁₁ H ₂ O	rare
Niter	KNO ₃	very rare
Nitratine	NaNO ₃	very rare

	Other	
Pyrite	FeS₂	very rare
Sepiolite	Mg₄Si ₆ O₁₅(OH)₂6H₂O	very rare

¹Very common: mineral occurs in greater than 75% of the samples analyzed; common: mineral occurs in 25-75% of the samples analyzed; rare: mineral occurs in 10-25% of the samples analyzed; very rare: mineral occurs in less than 10% of the samples analyzed. These qualifications apply only to occurrence within the core and are not meant to imply any information about the relative abundance of the particular mineral. For example, some of the phases that are listed as having rare or very rare occurrence can, nonetheless, be relatively abundant in a particular sample, and minerals that may be listed as common occurrence can have very low abundances.

²Includes hydromagnesite and psuedohydromagnesite.

³Because of its extreme instability and very rapid dehydration, mirabilite was rarely identified by X-ray diffraction.

Hosler, 1979; Kendall, 1988; Last, 1990). In the following section, the formative physical-chemical conditions of evaporitic minerals in Table 5.1 are noted.

Arcanite ((K,NH₄)₂SO₄)

Arcanite is a rare salt mineral. It was first found in the Santa Ana mine, Orange County, California (Palache et al., 1951; Frondel, 1950). It occurs as thin, pseudo-hexagonal tablets. Origin and paragenesis of the arcanite are unclear. However, the arcanite crystals can be artificially made in lab and its crystal morphology and cell parameters have be well studied by McGinnety (1972).

Bloedite $(Na_2Mg(SO_4)_24H_2O)$

Bloedite is commonly found together with gypsum, thenardite, and magnesite in North Ingebright Lake sediments. As discussed by Madsen (1966), dehydration of bloedite only occurs at above 120°C. It is a stable mineral (its metastable phase is konyaite (Na₂Mg(SO₄)₂H₂O)) under room temperature and primarily precipitated from the concentrated brine containing Na₂SO₄ and MgSO₄ in a molar ratio of 1:1 (Van Doesburg, 1982).

Epsomite (MgSO₄7H₂O)/hexahydrite (MgSO₄6H₂O)/kieserite (MgSO₄H₂O)

Kieserite is the most common Mg sulfate in evaporites, while both epsomite and hexahydrite are rarely found in salt deposits (Holser, 1979; Braitsch, 1971). Usually, kieserite, epsomite and hexahydrite coprecipitate from Na-Mg-Cl-SO₄ type brine with carnallite and polyhalite (Eugster et al., 1980).

Eugsterite (Na₄Ca(SO₄)₃2H₂O)

Eugsterite was first reported to occur in Kenya and Turkey and named after Hans P. Eugster (Vergouwen, 1981). The mineral is commonly found in the salt efflorescence in most playa lakes, where it originates from Na-SO₄-Cl type waters. Experimental results indicate that eugsterite can form together with thenardite and gypsum in considerable amounts in solutions with a molar ratio Na/Ca more than 4 (Vergouwen, 1981).

Görgeyite $(K_2Ca_5(SO_4)_64H_2O)$ and starkeyite $(MgSO_44H_2O)$

Görgeyite and starkeyite are rare secondary minerals. Görgeyite could formed by the reaction of brines with previously deposited gypsum: $[CaSO_42H_2O + K^+ \sim K_2Ca_5(SO_4)_64H_2O]$. Starkeyite might be a result of dehydration of epsomite (MgSO_47H_2O) under the extremely concentrated conditions (Eugster and Smith, 1965).

$Gypsum(CaSO_42H_2O)$ and $anhydrite(CaSO_4)$

Anhydrite can be produced by either direct precipitation or by gypsum dehydration. Gypsum is, however, reported to exist down to a depth of 1200 m in other area (Sonnenfeld, 1984), which indicates that neither the geothermal gradient nor the overburden pressure is the primary agent of anhydritization in the North Ingebright Lake basin. Anhydrite precipitation and gypsum-dehydration in modern lake environments are primarily controlled by the brine concentration. Contact between the previously precipitated gypsum and brines is necessary to alter gypsum to anhydrite (Braitsch, 1971). Primary anhydrite precipitation usually occurs when the brine becomes sufficiently concentrated and the water activity has been greatly reduced (0.2-0.7)(Sonnenfeld, 1984). Neutron diffraction studies (Atoji, 1958) have shown that the water molecules around gypsum are so weakly bound that the hygroscopic bitterns can readily dehydrate gypsum. Gypsum dehydration usually starts when a brine is concentrated beyond the stage of the saturation with respect to sodium chloride (Eugster et al., 1980).

Krausite ($Fe_2(SO_4)_32H_2O$)

Krausite was found in borate deposits in San Bernardino County, California (Foshag, 1931). It occurs in the alunite as large crystals or in the nests and irregular cavities in the fine clay-like alunite and coquimbite. Krausite is commonly associated with jarosite, fibroferrite, gypsum and anhydrite (Foshag, 1931). The mineral probably originated from the oxidation of pyrite (Foshag, 1931).

Leightonite $(K_2Ca_2Cu(SO_4)_42H_2O)$

Leightonite was first identified in the Chuquicamata mine in Atacama province, Chile (Palache, 1938; Van Loan, 1962; Cook, 1978). It is named in honor of Dr. Tomas Leighton, University of Santiago, Chile (Palache, 1938). Leightonite is commonly associated with gypsum, bloedite, atacamite and natrochalcite (Van Loan, 1962).
Mercallite (KHSO₄)

Mercallite is named after Giuseppe Mercalli (1850-1914), a former director of Vesuvius Observatory (Palache, 1951). It was identified in a saline efforescence from Vesuvius as minute orthorhombic crystals (Embrey and Fuller, 1979). It can primarily precipitate from the concentrated solutions (Palache et al., 1951).

Mirabilite($Na_2SO_410H_2O$) and then ard ite (Na_2SO_4)

Mirabilite and thenardite constitute one mineral pair which depends on the brine composition, temperature, and activity of water. At one atmosphere, in the presence of a pure sodium sulfate solution, the transition temperature of mirabilite to thenardite is 32° C (Cole, 1926)(Figure 5.2); mirabilite can crystallize at temperatures near 0°C from brines having total salinities between 30 and 70 ppt (Smith, 1979). However, the presence of NaCl and Na₂CO₃ dissolved in the brine can lower the mirabilite/thenardite transition temperature. For example, the transition temperature decreases from 3.2° C for a pure sodium-sulfate solution to 17.9°C for a solution saturated with respect to both sodium sulfate and sodium chloride (Figure 5.3). In brines containing 16.3% Na₂CO₃, the transition occurs at about 27° C (Eugster and Smith, 1965). In a simple ternary system of calcium sulfate, sodium sulfate, and water, mirabilite and thenardite usually are precipitated when CaSO₄ concentration reaches 22% and 34% at 25° C (Hill and Wills, 1938). The pronounced effect of common-ion compounds in solution on the transition



Figure 5.2 Solubility of sodium sulfate in water (from Cole, 1926)



Figure 5.3 Diagram showing the effect of dissolved NaCl on the mirabilite-thenardite transition temperature (from Eugster and Smith, 1965)

temperature of mirabilite and thenardite shows that the phase of sodium sulfate precipitated from brines is strongly controlled not only by the temperature and water activity, but also by the chemical composition of brines.

Polyhalite (Ca₂K₂Mg(SO₄)₄2H₂O)

Origin of polyhalite can be both secondary and primary. In lacustrine environments, most of the polyhalite is secondarily formed by the conversion of gypsum or anhydrite, such as [CaSO₄2H₂O + K⁺+ Mg²⁺ \Rightarrow Ca₂K₂Mg(SO₄)₄2H₂O + 2Ca²⁺](Hite, 1983; Sonnenfeld, 1984). Polyhalite may also be formed at the expense of glauberite by reaction: [Na₂Ca(SO₄)₂H₂O + CaSO₄ + 2K⁺+ Mg²⁺ + SO₄ + 2H₂O \Rightarrow Ca₂K₂Mg(SO₄)₄2H₂O + 2Na⁺]. The occurrence of polyhalite implies that brines are relatively enriched in K and Mg and new brine-solid phase equilibrium is reached by the back-reaction with previously deposited sulfates.

Wattevilleite ($Na_2Ca(SO_4)_2 4H_2O$)

Wattevilleite was found in pyritic lignite in the Einigkeit mine, near Bischofsheim, Germany (Palache et al., 1951; Roberts et al., 1990). However, origin and paragenetic data of this mineral are not available in the published literature.

Artinite $(Mg_2CO_3(OH)_23H_2O)$

Artinite was found in Val Malenco, Italy and in San Benito County, California (Cisneros et al., 1977). It occurs as crusts of acicular crystals, cross-fiber veinlets, and as spherical aggregates of radiating fibers coating fracture surfaces in serpentinized volcanic rocks. The mineral is commonly associated with hydromagnesite, magnesite, calcite, pyroaurite and dypingite (Cisneros et al., 1977).

Ankerite (Ca(Fe,Mg)(CO₃)₂)

Ankerite (also called ferroan dolomite) is one member of $CaCO_3$ -MgCO_3-FeCO_3 mineral system (Goldsmith et al., 1962). The effect of the substitution of the magnesium ions by the larger ferrous iron ion in the ankerite is an increase in both the *a* and *c* dimensions of the cell parameters (Howie et al., 1958). This mineral commonly occurs in sedimentary rocks and ore deposits, such as at the Antwerp iron mine, New York and at Oldham, Lancashire, England (Roberts et al., 1990; Broadhurst et al., 1958). Ankerite is most likely resulted from the alteration of previously precipitated calcite or dolomite (Broadhurst et al., 1958).

Benstonite $(Ca_7Ba_6(CO_3)_{13})$

Benstonite was found at the large barite deposit in Hot Spring County, Kansas (Lippmann, 1962). The beds of deposit lie between the Devonian-Mississippian novaculite and the Pennsylvanian shale. The barite rock is finegrained with varying amounts of interspersed shaly materials. Benstonite occurs as irregular white veins in the barite rock and is commonly associated with calcite and barite (Lippmann, 1962). It is likely that the barium of benstonite came from the barite, whereas calcium was introduced by percolating water. Benstonite is primarily precipitated from the concentrated solution as a carbonate mineral (Lippmann, 1962).

Fairchildite $(K_2Ca(CO_3)_2)$

Fairchildite is named in honor of John G. Fairchild, chemist of Geological Survey, United States Department of the Interior (Milton and Axelrod, 1947). It has been found in wood ash of partly burned forest trees in the western United States (Milton and Axelrod, 1947). Fairchildite has also been identified in ash from the biomass combustion (Navrotsky et al., 1997). Experiental works indicate that fairchildite can primarily precipitate from the concentrated K⁺-Ca⁺²-CO₃⁻² solutions or form by the reaction of previously precipitated gaylussite with the K⁺-CO₃⁻² aqueous solutions (Milton and Axelrod, 1947).

Gaylussite (Na₂Ca(CO₃)₂5H₂O)

In the lacustrine environment, gaylussite is usually result of back-reaction of evaporating alkaline brine with early-formed calcite (or aragonite), controlled by the magnitudes of the activity of water (Eugster and Smith, 1965): [CaCO₃ + 2Na⁺+ $CO_3^{2^-}$ + 5H₂O \Rightarrow Na₂Ca(CO₃)₂5H₂O]. The presence of gaylussite rather than pirssonite (Na₂Ca(CO₃)₂3H₂O) indicates a relatively low temperature and high ^aH₂O conditions.

Kutnohorite (Ca(Mn,Mg)(CO₃)₂)

Differential thermal and X-ray analysis indicate that kutnohorite is a maganese member of the dolomite group (Frondel et al., 1954). Kutnohorite occurs

as anhedral masses with curved cleavage surfaces up to three centimeters in Franklin and Sterling Hill, New Jersy. It is associated with calcite and rhodochrosite (Frondel et al., 1954). Recently, kutnohorite was commonly identified in modern lake deposits (Radovanovic, 2000; Radovanovic and Last, 2000; and Stevens et al., 2000). In the Big Watab Lake, MN, USA, the kutnohorite occurs as a true authigenic mineral associated with calcite in a reducing environemnt (Stevens et al., 2000). In the West Basin Lake, Victoria, Australia, the kutnohorite occurs as cement in beach rocks and lithified microbialites (Radovanovic and Last, 2000).

Minrecordite $(CaZn(CO_3)_2)$

Minrecordite is the Ca and Zn member of the dolomite series, found in the dioptase deposit of Tsumeb, Namibia (Garavelli et al., 1982). It normally occurs as two distinct variants: nearly pure $CaZn(CO_3)_2$ with only minor substitution of Mg, Fe, and Mn for Zn and a magnesian variety. However, these two minrecordite variants differ only slightly in chemical composition, unit-cell parameters and physical properties (Garavelli et al., 1982). The paragenetic mineral assemblage includes calcite, zincian dolomite, dolomite, magnesian minrecordite and minrecordite.

Zemkorite $(Na_2Ca(CO_3)_2)$

Zemkorite was first identified in cores, Yakutiya, Russia (Yegorov et al., 1990). It is a secondary mineral, resulted from the alteration of alkaline volcanic rock by highly concentrated brine solutions. Associated minerals include shortite and

halite (Yegorov et al., 1990).

Bradleyite (Na₃MgPO₄CO₃)

Bradleyite, sodium phosphate-magnesium carbonæte, was first found in the oil shale of Green River formation, Sweetwater County, Wyoming (Fahey and Tunell, 1941). It is commonly associated with trona, shortite, pirssonite, gaylussite, bromlite, and northupite. It was determined that bradleyite in the oil shale of Green River formation is a secondary mineral, originated from the alteration of sodium-calcium carbonate mineral, shortite [Na₂Ca₂(CO₃)₃] (Fahey and Tunell, 1941).

Burkeite (2(Na₂SO₄)Na₂CO₃)

Burkeite is a common mineral in lacustrine sediments; it is formed by the reaction of previously precipitated thenardite with CO_3^{-2} becaring brine: [2NaHCO_3 + 2Na₂SO₄ = 2(Na₂SO₄)(Na₂CO₃) + H₂O + CO₂]. Usually the occurrence of burkeite implies existence of the following conditions: solid carbonattes are present in excess over solid sulfates (Eugster and Smith, 1965).

Hanksite (9Na₂SO₄2Na₂CO₃KCI)

Hanksite is a secondary mineral formed by the reaction of previously precipitated thenardite, halite, or burkeite with K or CO₃ bearing brine: [12NaHCO₃ + K₃Na(SO₄)₂ + 3NaCl + 25Na₂SO₄ \Rightarrow 3(2Na₂CO₃ 9Na₂SO₄ KCl) + 6H₂O + 6CO₂] (Smith, 1979). The occurrence of hanksite usually means that brine composition is characterized by high pH (Smith, 1979). The ratios of Na_2CO_3 to Na_2SO_4 that allow hanksite to form, when the solution is saturated in halite and contains the correct percentage of K, are 0.3-4 (Smith, 1979).

Kainite (KClMgSO₄3H₂O)

Kainite is formed only if the ratio between magnesium sulfate and potassium chloride in the brine exceeds 0.1 by the following reaction: $[MgSO_4 7H_2O + K^+ + CI^- \\ \Leftrightarrow KCIMgSO_4 3H_2O + H_2O]$. At a ratio between 0.1-0.3, kainite coprecipitates with carnallite; at a ratio < 0.1, kainite occurs mixed with epsomite (Sonnenfeld, 1984). The Harvie-Weare model (1982) predicts kainite saturation at water activity of 0.42-0.62. Therefore, the presence of kainite in sediments usually shows the existence of the arid condition during deposition of sediments.

Rapidcreekite (Ca₂SO₄CO₃4H₂O)

Rapidcreekite was first identified by Roberts et al. (1986) from a sequence of Lower Cretaceous (Albian) ironstones and shales in the Rapid Creek area, northern Yukon Territory. It is originated from recent surface weathering and occurs as a secondary phase on dilated joint-surfaces and bedding planes in a blocky quartz-rich bed in sideritic iron-formation. The mineral is commonly associated with aragonite, gypsum, hexahydrite, epsomite and diadochite (Roberts et al., 1986).

Tychite $(Na_6Mg_2SO_4(CO_3)_6)$

Tychite is a secondary mineral. It precipitates at the expense of thenardite,

nahcolite and dolomite by the following reaction: $[Na_2SO_4 + 4NaHCO3 + 2CaMg(CO_3)2 \approx 2CaCO_3 + Na_6Mg_2SO_4(CO_3)_6 + 2H_2O + 2CO_3]$. Occurence of tychite usually implies the existence of the following conditions: solid carbonates are present in excess over solid sulfates and consistently contact with brines (Eugster and Smith, 1965).

Bischofite (MgCl₂6H₂O)

Bischofite is a late stage evaporite mineral. It can primarily precipitate from a concentrating brine after the potassium supply has been exhausted. Bischofite can also form by diagenetic alteration of previously precipitated carnallite minerals (Sonnefeld, 1984).

Carnallite (KMgCl₃ 6H₂O)

Carnallite occurs mainly as a component of sedimentary saline deposits, often associated with kieserite, halite, sylvite, and polyhalite. Carnallite can precipitate primarily from the concentrated brine when the the solution reach the the point of carnallite saturation. It can also form by secondary alternation after kainite as a result of the following reaction (Braitsch, 1971; Sonnefeld, 1984): [KMg(SO₄)Cl3H₂O + 4MgCl₂ + 32H₂O \Rightarrow 4MgSO₄H₂O + 4 KMgCl₃ 6H₂O + 15H₂O].

Korshunovskite (Mg₂Cl(OH)₃4H₂O)

Korshunovskite was first identified from the Korshunov iron-ore deposit, Irkutsk region, Russia (Dunn and Fleischer, 1983). Paragenetic minerals include magnesite, dolomite and antigorite. However, the origin data of Korshunovskite is incomplete at present.

Inderborite (CaMgB₆O₁₁H₂O)

Borate minerals are usually present in potassium-magnesium salt sequences. This suggests that borate minerals precipitate only during the late evaporation stage. For example, in Germany borate minerals only appear in the carnallite or polyhalite-kieserite region of the evaporite sequence (Sonnefeld, 1984). In most cases, borate deposits are also closely associated with volcanic rocks. The thick Miocene borate deposits at Kramer, California are underlain by the Saddleback Basalt and overlain by beds of volcanic ash. Borate minerals have been shown to have originated from diagenetically altered volcanic rocks (Barnard and Kistler, 1966). In Mississippian evaporites of southern New Brunswick, Canada, a number of borate minerals are found in sylvite, halite, anhydrite and red mudstone. The addition of boron from volcanic exhalation is postulated to account for these borate mineral deposits (Rcuslston and Waugh, 1981). Therefore, the presence of borate minerals can be explained by the occurrence of syndepositional volcanic activity, which is a source of boron for the evaporite basin.

Nitrate (KNO₃) and nitratine (NaNO₃)

Niter and nitratine are commonly found in some lacustrine deposits. The largest known deposits of lacustrine nitrates are located in the deserts of Atacama and Tarapaca, in the northern part of Chile. Niter and nitratine are also reported to occur in large quantities in Searles Lake, California (Smith, 1979). The nitrates are usually associated with anhydrite, gypsum, thenardite, mirabilite, bloedite, and epsomite. The nitrates possibly are precipitated from the concentrated brine or formed by the oxidation of organic matter (Clarke, 1924).

Chapter 6 Geochemistry of Sedimentation

6.1 Introduction

Evaporite formation and related physical-chemical conditions in the sedimentary realm are subjects of long-standing interest and study over the last two centuries (Eugster, 1971; Multhauf, 1978; Warren, 1989; Rosen, 1991). Numerous chemical models have been established to predict the sequence of salts precipitated during evaporative concentration of a brine. In general, two approaches have been used to determine the stable mineral-solution assemblages for the various component systems in evaporating brines: (i) direct evaporation experiments, and (ii) thermodynamic and solubility calculations. Since the early experimental work of Van't Hoff (Eugster, 1971) and his successors, a large number of solubility measurements of salts in different aqueous systems has accumulated (see Assarsson, 1950a; 1950b; 1950c; Braitsch, 1971; Eugster, 1971; Stewart, 1963). These solubility data have been used to predict successfully the sequence of major minerals precipitated from evaporating brines by means of concentrationtemperature diagrams and phase (or Jänecke) diagrams. Clearly, this graphical approach is best applied to simple binary and common-ion aqueous systems and cannot readily manage the more complex ionic systems that commonly occur in natural brines (Hardie, 1984). To solve this problem, several generations of computer models based on equilibrium thermodynamics have emerged over the past two decades, such as WATEQF (Truesdell and Jones, 1974; Plummer et al.,

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1976), WATEQ2 (Ball, Jenne, and Nordstrom, 1979), SOLMNEQ (Kharaka and Barnes, 1973), and ACTIVITY and PHRQPITZ (Plummer et al., 1988). Although many of these models are limited to dilute aqueous solutions with ionic strengths less than 0.7, some are valid to high ionic strengths (Harvie et al., 1984; Weare, 1987).

These experimental and theoretical geochemical models provide powerful tools that can assist in deciphering past chemical and hydrologic conditions and depositional processes during evaporite formation in a sedimentary basin. In this chapter, these mineral solubility models are used as the theoretical basis to interpret the observed mineral assemblages and chemical compositions of brines in North Ingebright Lake, Saskatchewan, and the hydrologic conditions under which these evaporites were deposited. Although commonly applied to brine reconstruction in marine and lacustrine settings elsewhere, this is the first such effort to apply these models to lake deposits of the Great Plains of North America.

6.2 Basis of Chemical Composition Reconstruction

Two approaches have been used to determine the chemical sedimentation history of North Ingebright Lake: mass balance calculations and physical-chemistry reasoning. For the bulk chemical composition determination of lake water, a basic tenet is that in a hydrologically closed system undergoing evaporative concentration, a complete assemblage of nondetrital minerals must have the same bulk ionic composition as the original water (Hardie, 1984). As discussed elsewhere (Nesbitt, 1974; Wasson et al., 1984; Torgersen et al., 1986; Teller and Last, 1990; Spencer et al., 1985a; 1985b; Last, 1994; 1999), there are many assumptions and limitations that must be accepted in the application of this straightforward principle to paleolimnology.

For North Ingebright Lake, no perennial streams enter or exit the lake, and the catchment area is very small, which limits clastic input. Sediments are dominantly endogenic/authigenic rather than detrital, and chemical sedimentation is the main process. High density of subsampling (2 cm interval) in the North Ingebright Lake cores provides a good basis for the proper identification of the composition of the parent brine. For each sample, the complete assemblage of saline minerals (qualitative) and the abundance of the major saline minerals (semiquantitative) have been determined. From simple mass balance considerations, the major ion ratios of the parent brine for North Ingebright Lake evaporite sequence can be related to the mineral contents as follows (in molar proportion) (Hardie, 1984):

Ca/Na \approx (protodolomite+gypsum+anhydrite+calcite+aragonite)/(thenardite+2bloedite+halite) Mg/Na \approx (magnesite+protodolomite+bloedite)/(thenardite+2bloedite+halite) SO₄/Na \approx (gypsum+anhydrite+bloedite+thenardite)/(thenardite+2bloedite+halite) CO₃/Na \approx (aragonite+magnesite+2protodolomite+calcite+Mg calcite)/(thenardite+2bloedite+halite) CO₃/SO₄ \approx (aragonite+magnesite+2protodolomite+calcite+Mg calcite)/(gypsum+anhydrite+bloedite+thenardite)

Table 6.1 and Figure 6.1 show changes in the major dissolved components in the precipitating brine in North Ingebright Lake through the Holocene. Further refinement of the brine composition is accomplished by the occurrence of minor and trace minerals. Over 40 species of primary and authigenic minerals have been

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	Stratigraphic Unit							
	2 Na-Sulfate	3 Mg-Carbonate Na-Sulfate	4 Ca-Sulfate Na-Sulfate	5 Muddy Na- Sulfate	6 Na- Sulfate	7 Muddy Mg-calcite- Na-Sulfate	8 Dolomitic Na- Sulfate	Average
Ca/Na	0.1	0.1	0.3	0.1	0.1	0.1	0.2	0.2
Mg/Na	<0.1	0.1	0.1	<0.1	<0.1	<0.1	0.2	0.1
SO₄/Na	1	1.1	1.3	1.1	1	1.1	0.9	1.1
CO₃/Na	0.1	0.1	0.1	<0.1	<0.1	0.1	0.6	0.2
CO₃/SO₄	<0.1	0.1	0.1	<0.1	<0.1	0.1	0,5	0.1

Table 6.1 Variations of average ion ratios in different stratigraphic units of north core at North Ingebright Lake

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identified in North Ingebright Lake deposits (Table 5.1; Appendix C). Many of these minerals may provide important information on the hydrochemical and climatic conditions at the time of precipitation (Hosler, 1979; Last, 1990). The physical-chemical conditions for the formation of trace minerals, such as gaylussite, epsomite, eugsterite, polyhalite, bischofite, carnallite, and burkeite, follow Bradley and Eugster (1969); Braitsch (1971); Eugster and Smith (1965); Eugster et al. (1984); Hardie and Eugster (1970); and Smith (1979). However, modeling data on some trace minerals detected in North Ingebright Lake, such as fairchildite, starkeyite, and rapidcreekite are less complete or unavailable. By combining results from bulk composition determination and physical-chemistry reasoning, the generalized evolutionary paths of brine types at North Ingebright Lake are reconstructed.

6.3 Geochemical History of North Ingebright Lake

The mineral assemblage of North Ingebright lake salts can be represented by phase relationships in the reciprocal quinary system Na-Mg-Ca-Cl-SO₄-H₂O and its subsystems (4-, 3-, or 2-component systems) (Figure 6.2). In that system, at 25°C, gypsum, anhydrite, and mirabilite occupy the bulk of the prism, with thin slices for thenardite, bloedite, and halite (Harvie and Weare, 1980; Harvie et al., 1980; Eugster et al., 1980). The general composition change and evaporative stage of the brine during deposition of the salts are deduced with the aid of these phase diagrams. However, this reconstruction only represents an extremely simplified version of the chemical events that occurred in the lake. In certain cases theoretical



Figure 6.2 The reciprocal quinary system Na-Mg-Ca-Cl-SO,-H₂O and its subsystems at 25°C. Modified from Harvie and Weare (1980), Harvie, Eugster, and Weare (1982), and Eugster, Harvie and Weare (1980) For abbreviations see Table 6.2.

A	anhydrite	CaSO₄
Ар	aphthitalite	NaK ₃ (SO ₄) ₂
Ant	antarcticite	CaCl ₂ 6H ₂ O
Bi	bischofite	MgCl ₂ 6H ₂ O
BI	bloedite	Na ₂ Mg(SO ₄) ₂ 4H ₂ O
Car	carnallite	KMgCl₃6H₂O
Ep	epsomite	MgSO₄7H₂O
G	gypsum .	CaSO₄2H₂O
GI	glauberite	Na₂Ca(SO₄)₂
н	halite	NaCl
Hx	hexahydrite	MgSO₄6H₂O
Ka	kainite	KMgClSO₄3H₂O
Ki	kieserite	MgSO₄H₂O
Le	leonite	K ₂ Mg(SO ₄) ₂ 4H ₂ O
Μ	mirabilite	Na₂SO₄10H₂O
Pic	picromerite	K₂Mg(SO₄)₂6H₂O
Po	polyhalite	K ₂ MgCa ₂ (SO ₄) ₂ 2H ₂ O
Syl	sylvite	KCI
Syn	syngenite	K₂Ca(SO₄)₂6H₂O
Тс	tachyhydrite	Mg ₂ CaCl ₆ 12H ₂ O
Th	thenardite	Na ₂ SO ₄

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Table 6.2 Mineral abbreviations

equilibrium models cannot predict the observed mineral sequence so more complicated geological processes have to be proposed to explain the mineral formations.

Unit 1 was only penetrated in the south core. It consists mainly of gypsum, fine-grained sand, silt, clay and organic materials. Unlike Units 2-8, Unit 1 was dominated by physical sedimentation process. Therefore, only chemical sedimentation processes of Units 2-8 are discussed in the following section.

6.3.1 Unit 2 Na-Sulfate Salt

The saline minerals in Unit 2 are mostly mirabilite or thenardite with minor amounts of gypsum, bloedite, and halite. This simple mineral assemblage indicates that the saline lake waters were chemically simple and dominated by Na and SO₄ (SO₄/Na molar ratio = 1) (Table 6.1, Figure 6.1). The small percentage of Ca, Mg, and HCO₃+CO₃ ions may indicate brief changes in the chemical compositions of the lake waters perhaps caused by episodes of freshwater.

The evaporites in Unit 2 are composed chiefly of minerals that represent phases in the quaternary system $CaSO_4$ - Na_2SO_4 - $NaCl-H_2O$. The stability fields of the solid phase assemblages in terms of the anhydrous composition of the coexisting solutions at 25°C and 1 atm total pressure are shown in Figure 6.3 (Hardie, 1968). The data in Unit 2 show that of the related minerals in Figure 6.3, only mirabilite-gypsum, mirabilite-thenardite, and thenardite-halite coexist in appreciable quantities; glauberite and anhydrite are not present.

Zones 1 and 3 are similar in that they consist solely of monomineralic beds



Figure 6.3 Schematic phase relations in CaSO₄-Na₂SO₄-NaCl-H₂O system of Unit 2. Modified from Hardie (1968). For abbreviations see Table 6.2.

of mirabilite, while zones 2 and 5 are also similar because both of them contain gypsum (see Section 4.3.2). Halite is only found in zone 4, which may indicate that the brines of zone 4 were desiccated more extensively than those that produced zones 1, 2, 3, and 5. In the $CaSO_4$ - Na_2SO_4 - $NaCl-H_2O$ system, the mirabilite and thenardite stability fields are small and close to the $NaCl-Na_2SO_4$ boundary (Figure 6.3). This implies that the brines of zones 1 and 3 were very close to the Na_2SO_4 corner at the time crystallization began. The crystallization of mirabilite forced the composition of the remaining brine toward the $CaSO_4$ -NaCl edge of the phase diagram. However, it never reached the boundary of the mirabilite and gypsum adjoining field. Zones 2 and 5 both contain mirabilite and gypsum, which indicates that the initial brines were represented by points in the mirabilite or gypsum field but close to the phase boundary.

Zone 4 is composed mainly of mirabilite and thenardite with traces of halite. This mineral composition shows that the brines that produced salts in zone 4 experienced a short-term shift toward the NaCl corner along the Na_2SO_4 -NaCl edge of the diagram shown in Figure 6.3.

It is evident, therefore, that the brines that produced salts in zones 1 and 3 had an initial composition of Na-SO₄, the brines that desiccated to produce salts in zone 4 were dominated by Na-SO₄-Cl, and the initial composition of zones 2 and 5 brines were Na-Ca-SO₄. Thus, the whole salt sequence of Unit 2 is obviously not a result of one evaporating event. However, the mineral assemblage points toward three evaporating cycles from zone 1 to zone 5: Na-SO₄-Na-Ca-SO₄ (zone 1-zone 2), Na-SO₄-Cl (zone 3), and Na-SO₄-Na-Ca-SO₄ (zone 4-zone 5).

The existence of nitrates in zone 4 (also little in 1, 2, and 3) provides further insight into the depositional environments. Niter and nitratine are commonly believed to be either (i) precipitated from a highly concentrated brine or (ii) formed by the oxidation of organic matter. The trace amounts of nitrates of zone 4 were most likely produced by the oxidation of organic matter, which is supported by a general low organic matter content (average: 0.4%) in Unit 2 (Figures 4.8 and 4.9). Therefore, the presence of nitrates might indicate that oxygenated shallow water conditions existed during the deposition of Unit 2.

6.3.2 Unit 3 Mg-Carbonate--Na-Sulfate Salt

The evaporite minerals in the sediments of Unit 3 consist of high-Mg calcite, aragonite, protodolomite, magnesite, gypsum, mirabilite, thenardite, eugsterite, bloedite, starkeyite, and halite. The number of components involved in the mineral assemblages of most zones in Unit 3 is large and makes complete graphic representation of the phase relations impossible. However, the post-carbonate mineral assemblage can be represented by phases in the quinary system: Na-Mg-Ca-Cl-SO₄-H₂O. A complete analysis of the phases in this six-component system is rather complex. According to Gibbs phase rule, a maximum of four solids and one liquid phase can coexist at an arbitrary pressure (P) and temperature (T) (Harvie et al., 1982). Based on the semi-empirical equations for the excess solution free energy, Harvie and Weare (1980) and Harvie et al. (1982) determined the invariant points and associated solution composition in this quinary component system and subsystems at 25°C. The stability fields of the solid phase assemblages have been

plotted in a trigonal prism diagram representing the reciprocal system. Ten single solid phase fields have been delimited. Gypsum dominates the phase volume; mirabilite, thenardite, bloedite, and epsomite occupy very small percentage of the phase volume. Halite, anhydrite, hexahydrite, and kieserite occur in smaller volume near the base (Figure 6.2).

The evolution of brine composition during deposition of salts in Unit 3 can be qualitatively described through the phase relations in the Na₂SO₄-CaSO₄-MgSO₄ system (Figure 6.4). The salts in zones 1 and 3 are mostly pure mirabilite and the mud content is relatively high (Figures 4.4-4.6). Therefore, the initial composition must have been close to the Na₂SO₄ corner in Figure 6.4. These zones were probably the result of winter cooling of the lake during periods when its concentration reached 30-70 ppt. In solutions dominated by sodium-sulfate, mirabilite solubility is very sensitive to temperature, and when cooled, mirabilite could be "frozen out" exclusive of the other minerals. During the following warmer seasons, freshwater influxes carry a new group of dissolved solutes (Mg, Ca, and HCO_3+CO_3) and suspended materials to the lake. The deposition of carbonates and clastic mud cover the floor of the lake and protect the mirabilite salts from being dissolved by warmer, summer season waters (Last, 1989).

Zones 2, 5, and 7 consist mainly of mirabilite and gypsum. The brines that produced these salts must have shifted in composition toward the boundary of the mirabilite and gypsum fields, therefore, the initial brines were represented by points in either the mirabilite or gypsum field but close to the phase boundary as noted in Figure 6.4. Zones 2 and 5 contain trace amounts of görgeyite $[K_2Ca_5(SO_4)_64H_2O]$



Figure 6.4 Phase relations in NaSO₄-CaSO₄-MgSO₄ system, projected from the H_2O apex (Harvie et al., 1982). For abbreviations see Table 6.2.

and starkeyite (MgSO₄4H₂O), which are rare secondary minerals. The presence of starkeyite and görgeyite indicates the lake might have been completely dry for a short period of time. Although as discussed by Harvie et al. (1982), stable phase diagrams cannot be readily applied to the minerals that result from complete desiccation of the solution phase to track changing compositions, the existence of starkeyite and görgeyite does show that the brines during deposition of zones 2 and 5 contain K⁺ and Mg⁺².

Small amounts of gaylussite $[Na_2Ca(CO_3)_25H_2O]$ have been detected in zone 4 of Unit 3. The presence of gaylussite, rather than pirssonite, indicates relative low temperature and high ${}^{a}H_2O$ conditions (Eugster and Smith, 1965). The occurrence of sodium-carbonate minerals also implies that the brine composition was a high pH, alkaline brine, poor in Ca and Mg (i.e., Na-K-CO₃-SO₄-Cl) during the deposition of zone 4.

The initial composition of brines during deposition of zone 6 is much more complicated than that of zones 1, 2, 3, 4, 5, and 7, as shown by its saline assemblage of mirabilite, gypsum, bloedite, eugsterite, arcanite, and halite. The appearance of eugsterite and bloedite provides significant information for understanding the chemical composition of brines that produced the salts of zone 6. Eugsterite [Na₄Ca(SO₄)₃2H₂O] is commonly found in salt efflorescence in playa lakes, where it originates from Na-SO₄-Cl type waters with Na/Ca molar ratios more than 4 (Vergouwen, 1981). Bloedite [Na₂Mg(SO₄)₂4H₂O] is primarily precipitated from concentrated brines containing Na, Mg and SO₄ in a molar ratio of 1:1 (Van Doesburg, 1982). These show that saline lake waters were dominated by Na, Ca,

Mg, and SO₄, with trace K and Cl indicated by the existence of arcanite and halite.

6.3.3 Unit 4 Ca-Sulfate--Na-Sulfate Salt

Unit 4 is largely composed of Ca-Mg carbonate and Na-Ca sulfate minerals, which indicate that the starting compositions of the solution that produced these salts in North Ingebright Lake consisted mainly of Na, Ca, Mg, SO₄, and CO₃ However, a wide variety of minor mineral assemblages in the different zones show that the lake waters experienced diverse variations in the chemical composition, but no general pattern is discernable. These minor minerals include: halite, hanksite, polyhalite, burkeite, carnallite, and gaylussite; most of these are formed by the syndepositional reaction of early formed minerals with the evolving brine. These reaction minerals may be regarded as part of a primary depositional sequence.

Polyhalite $[Ca_2K_2Mg(SO_4)_4H_2O]$ occurs in all salt zones of Unit 4, except zone 1, but mainly as a trace mineral. The occurrence of polyhalite implies that brines were relatively enriched in K and Mg and the new brine-solid phase equilibrium was reached by the back-reaction with previously deposited sulfates. Another trace mineral commonly found in Unit 4 is hanksite (9Na₂SO₄2Na₂CO₃KCI). The occurrence of hanksite indicates that brine composition was characterized by a high pH (alkaline) brine, which is also evidenced by the presence of fairchildite in zones 1 and 2, gaylusite in zone 3, and burkeite in zone 5. The ratios of Na₂CO₃ to Na₂SO₄ that allow hanksite to form are 0.3-4, when the solution is saturated with halite and contains the proper percentage of K (Smith, 1979). Minor amounts of anhydrite occur in zone 5, indicating that ^aH₂O has been decreased to below 0.7

(Braitsch, 1971).

Physical-chemical conditions deduced from the above mineral composition suggest that the minerals making up Unit 4 precipitated over an extreme range of brine concentrations from nearly fresh water to concetrated brines. Each zone approximately represents one continuous evaporating event. In general, brine compositions were dominated by Na-Ca-Mg-CO₃-SO₄ in early stages, and shifted to Na-K-SO₄-Cl in late stages.

6.3.4 Unit 5 Muddy Na-Sulfate Salt

Compared to previous stages, the saline components in Unit 5 suggest that unusual lake conditions existed at this time. The lake was relatively stable and contained more suspended sediment than previous lakes. The high organic matter content and dark colouration of Unit 5 indicate that deposition occurred subaqueously under reducing conditions. However, the lack of lamination suggests that lake water body was not stratified.

Unit 5 is composed mainly of mirabilite/thenardite and gypsum, with lesser amounts of Ca-Mg carbonates. The post-carbonate composition of the brines that formed the salts can be represented by points around the boundaries between the mirabilite and gypsum phase fields and the thenardite and mirabilite phase fields in the CaSO₄-Na₂SO₄-NaCl-H₂O system (shaded area in Figure 6.5). Although the lake was consistently occupied by brines saturated in respect with gypsum and mirabilite/thenardite at the time of zones 1-5 deposition, there were probably several short episodes of nearly complete desiccation.



Figure 6.5 Schematic phase relations in CaSO₄-Na₂SO₄-NaCl-H₂O system. Modified from Hardie (1968). For abbreviations see Table 6.2.

conditions. In addition, polyhalite is found in zone 2. As discussed in Chapter 5, polyhalite likely resulted from the back-reaction of evaporating K-Mg-SO₄ brine with early-formed gypsum, anhydrite, or glauberite. It is evident, therefore, that the brine composition was dominated by Na, Ca, Mg, and SO₄ (Appendix D) and climate was dry during the deposition of zones 1 and 2.

In contrast with zones 1 and 2, salts in zones 3, 4, and 5 were deposited under relatively dilute waters and cool conditions, as indicated by the presence of gaylussite in zone 3, inderborite in zone 4, and pure mirabilite in zone 5. Gaylusite usually forms under a relative low temperature and high ${}^{a}H_{2}O$ environments by back-reaction of evaporating alkaline brine with early-formed carbonates (Eugster and Smith, 1965). Inderborite (CaMgB₆O₁₁H₂O) crystallization is also favored by cool temperatures (Smith, 1979), while mirabilite can precipitate from pure sodiumsulfate solution at low temperatures . It is clear that relatively wet and cool conditions are well documented in zones 3, 4, and 5.

6.3.5 Unit 6 Na-Sulfate Salt

Changes in bulk brine composition during the crystallization of zones 1, 2, and 3 are rather gradual. The starting composition of the brine that produced the salts in zone 1 must have been close to the Na_2SO_4 corner along the boundary between NaCl and Na_2SO_4 of Figure 6.6, which is shown by its nearly pure mirabilite component. The saline minerals in zones 2 and 3 are mostly mirabilite and gypsum, implying that the composition of the brine responsible for zones 2 and 3 gradually shifted from the Na_2SO_4 corner to the boundary between the gypsum and mirabilite

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Figure 6.6 Schematic phase relations in CaSO₄-Na₂SO₄-NaCl-H₂O system. Modified from Hardie (1968). For abbreviations see Table 6.2

Trace amounts of epsomite (MgSO₄7H₂O) were detected in zone 1 as an admixture to mirabilite and gypsum, suggesting the existence of the concentrated phase fields. In addition, trace amounts of kainite (4KCl4MgSO₄11H₂O) are in zone 1. As discussed by Sonnenfeld (1984), kainite is formed only if the ratio between magnesium sulfate and potassium chloride in the brine exceeds 1:9; therefore, the brine of zone 1 also contained a small amount of K and Cl ions. In comparison with other units, Ca, Mg, and CO₃ ions are generally low; Na ion is mostly high. However, variations in the Ca, Mg, and CO₃ profiles shown in Figure 6.1 and Appendix D are rather erratic and do not show any overall trends with depth.

The salts in Unit 6 were likely deposited under oxygenated and shallow water conditions. This resulted in a general low content of organic matter and absence of sulfide minerals, which are commonly found in other units (see Appendix C). Niter (KNO_3) is detected in Unit 6; it is most likely a product of organic matter decomposition under this oxygenated shallow water condition.

6.3.6 Unit 7 Muddy Mg-calcite--Na-Sulfate Salt

The evaporite minerals making up this unit consist primarily of mirabilite, thenardite, and gypsum. Carbonate, polyhalite, epsomite, bloedite, and anhydrite occur in lesser amounts. Similar to Unit 6, the composition of the brines that formed the salts of Unit 7 are dominated by Na, Ca, and SO₄ ions, with small amounts of Mg and HCO_3+CO_3 ions. Figure 6.1 and Appendix D show that Na and SO₄ concentrations remain constantly high from zones 1 to 3. Ca, Mg, and CO_3 concentrations vary broadly. In general, Ca and CO_3 decrease and Mg increases

upward.

Although the depth of the brine is difficult to assess, the lake was likely quite deep during deposition of salts in Unit 7. The salt was deposited subaqueously under reducing conditions, which is indicated by its relatively high organic matter content and black coloration.

6.3.7 Unit 8 Dolomitic Na-Sulfate Salt

The saline components in Unit 8 consist mainly of high-Mg calcite, protodolomite, mirabilite, thenardite, bloedite, and gypsum. The most abundant mineral assemblage is mirabilite, thenardite, and bloedite, which suggests that the initial composition during the crystallization of Unit 8 can be represented by points in NaCl-Na₂SO₄-bl triangle region (Figure 6.7, shaded area). In general, as shown in Figure 6.1, Mg, Ca, SO₄, and CO₃ ion concentrations increase upward. Magnesium calcite with 4-9 mol% MgCO₃ is dominant in zone 1, whereas zones 2 and 3 contain high Mg calcite with 11-15 mol% MgCO₃ as a major phase. Concurrently with this change, the molar Mg/Na ratio in the waters increases from 0.01 at zone 1 to 1% at zone 3.

Most evaporite beds in Unit 8 show only the earlier and very late parts of the expected succession with some repetition of sequence, such as protodolomitebloedite, mirabilite-starkeyite, and thenardite-bichofite assemblages. Many transitional mineral phases in these sequences are not present. One possible reason is that dramatic changes of climate caused the rapid drying of the lake. If the brines were concentrated in a short period, there may simply not have been enough



Figure 6.7 Phase relations in the reciprocal system Na-Mg-SO₄-Cl (after Harvie, Eugster, and Weare, 1982). For abbreviations see Table 6.2.

time for equilibrium precipitation (Smoot and Lowenstein, 1991).

6.4 Summary of Chemical Sedimentation

Understanding the intricate processes controlling relationships between hydrological condition and lake chemical sedimentation is a difficult task. In general, variations in mineralogical composition in sediments are often viewed as indices of hydrological fluctuation (see overviews in Vance, 1997; Xia et al., 1997; Last, 1999; Van Stempvoort at al., 1993). Because there is no permanent inflow, North Ingebright Lake was fed by groundwater, direct precipitation, and diffuse overland runoff throughout its history. These input sources contributed at differing proportions over time, and these variations are recorded by the mineral assemblage. Among these different sources, groundwater plays a key role in maintaining the existence of North Ingebright Lake. Analyses of the lake waters and sediment mineralogy suggest that North Ingebright Lake may be fed by two distinct groundwater sources. The first is a deep groundwater aquifer with a very high content of SO₄ and Na; the second is a shallow groundwater aquifer which is less evolved, less saline, and has relatively lower content of SO₄ and Na. The shallow groundwater has very high Mg, Ca, and CO₃ contents, resulting from contact with Mg-rich bedrock (till) and early precipitation of calcite cements in soils.

The initial sediments of Unit 1 (Basal Black Mud) were deposited as the lake developed in the remnant of preglacial valley. This stage of North Ingebright Lake is represented by clastic deposits under freshwater lacustrine conditions. After this early clastic sedimentation in Unit 1, the basin evolved under an endoreic regime,
where seasonal recharge and the recycling of previously-formed soluble minerals led to the deposition of thick salt sequence.

During the deposition of Units 2 and 6, the lake level dropped dramatically and is interpreted to have been fed by deep groundwater. Lake water was saturated with Na and Na+Mg sulfates. This represents the most concentrated period of lake water history.

During the deposition of Units 3, 5, and 7, lake-levels were relatively high. The spring and deep groundwater inputs were negligible during this period. The increase in surface runoff and shallow groundwater inputs resulted in dilution of SO_4 and Na, while Ca, Mg, and CO_3 increased slightly and remained nearly constant due to carbonate precipitation. Protodolomite and magnesite are the two main carbonate minerals found in Units 3, 5, and 7, but magnesite is the more abundant of the two minerals. These proportions suggest that the lake in which the Units 3, 5, and 7 were deposited was characterized by a high Mg/Ca ratio. The presence of trace gaylussite in these three units suggests a positive moisture condition. In addition, the coexistence of gaylussite, aragonite, and magnesite with soluble evaporitic minerals in Units 3, 5, and 7 means there were rapid fluctuations in lake levels.

During the deposition of Units 4 and 8, the lake is interpreted to be fed by both shallow and deep groundwater sources. The continued high input of SO₄ and Na from the deep groundwater resulted in a stable high concentration of SO₄ and Na in the lake. The concentration of Ca and Mg in solution also increased as a result of the relatively high input from surface inflow. The uppermost section of Unit 8 is a mottled coloration mud layer. These mottled mud materials are inferred to

represent times when the lake was dry and the sediments were exposed to oxidation by the atmosphere. However, mottled mud layers are rarely found in Unit 4. This indirectly indicates that the lake never experienced long-term desiccation in the central area during much of the time Unit 4 was deposited, rather a fluctuating lake that desiccated only for relatively brief periods is indicated by the very soluble evaporate minerals in Unit 4.

In its entire history, North Ingebright Lake appears to have been saturated with Na-SO₄. It was diluted during occasional wet periods and concentrated during dry periods. It is evident that the thick accumulations of evaporites in North Ingebright Lake did not result from one evaporative event but are the result of continuous input and evaporation.

Chapter 7 North Ingebright Lake: A Dynamic Closed-Basin Lake System

7.1 Introduction

Salt minerals in North Ingebright Lake basin result from the complex interaction of water, solute, and energy fluxes. The number of factors that control the precipitation of salt minerals is large but mainly include: temperature, relative humidity, salinity, water activity, bedrock lithology, hydrological balance, and energy budget. The relationships among these factors are rather complex and are not readily resolvable with presently available data. In this chapter, the relationships of several major factors that control the deposition of North Ingebright Lake sequence are discussed.

7.2 Thermodynamic Activity of H₂O and Salinity

The thermodynamic activity of water (${}^{a}H_{2}O$) changes insignificantly in most dilute solutions. It can be regarded as a constant with a value of one in many near surface geochemical processes. However, in concentrated brines, the effects of changing the activity of water on evaporation rate and reaction coefficients must be considered (Helgeson et al., 1970). The thermodynamic activity of water in a simple electrolyte solution at a given temperature and pressure can be represented by

$$\ln^{a}H_{2}O = -0.018\Theta_{\pm e}m_{e}\Psi_{e}$$

where m_e is the total molality of the solute, $\vartheta_{t,e}$ refers to the stoichiometric number

of moles of ions in one mole of the eth electrolyte, and Ψ_{e} is the molal osmotic coefficient of the electrolyte, which can be determined from an extended Debye-Hückel power function of the ionic strength (I) of the solution (Helgeson et al., 1970). When the total molality of the solute, m_e, is zero (pure water) the factivity of water equals 1.

The addition of ions (such as sodium and sulphate) two water has a substantial electrostrictive effect on water structure, and they introduce disorder into the normal tetrahedral coordination of water molecules. Because of the electronic polarity of water molecules, some water molecules become structurally bound in hydration sheaths around ions and the thermodynamic activity of the water molecules will be significantly reduced (Dingman, 1994; Kinsman, 1976; Harbeck, 1955). Salts commonly found in natural waters, such as sodium =sulfate, sodium carbonate, and sodium chloride, have a similar effect on the water activity in varying concentrations, as shown in Figure 7.1 (Harbeck, 1955; Eugster arad Jones, 1979; Langbein, 1961). The dissolved solids in North Ingebright Lake largely consist of sodium sulfate; thus, the curve in Figure 7.1 best represents the relationship between salinity and water activity in North Ingebright Lake brine at high concentration. Overall, the water activity of brine decreases about 1 percent for a change in salinity of 10 ppt.

7.3 Water Activity, Relative Humidity, and Mineral Facies

As discussed by Kinsman (1976), the water activity in a solution can also be represented by the saturated vapour pressure of the solution. Essentially, the





saturated vapour pressure of the solution is another way to reflect the same property of the solution: thermodynamic activity of water. It is generally held that the amount of evaporation is directly proportional to the vapour pressure difference between the saturated vapour pressure of the solution and the vapour pressure of the air. Dissolved solids affect evaporation by reducing the saturated vapour pressure of solution/thermodynamic activity of water (Harbeck, 1955; Kinsman, 1976).

Relative humidity is the ratio of the actual water vapour pressure of the air to the saturation or equilibrium water vapour pressure of the air. The ratio is normally expressed as a percentage (Dingman, 1994). As is well known, the moisture content of the atmosphere (better known as relative humidity) exerts a significant influence on weather. Relative humidity determines cloud formation and precipitation, but also, through changes of vapour pressure, is intimately involved in the precipitation of salt minerals in a lake (Dingman, 1994; Kinsman, 1976). It is the vapour pressure that determines the movement of moisture across the interface between the lake water surface and air above it. If the saturated water vapour pressure (thermodynamic activity of water) of the lake water is greater than the vapour pressure of the atmosphere, then there will be a net transfer of water molecules from lake water to the atmosphere across interface. The evaporation rate from the lake surface depends upon the vapour pressure difference between the evaporating surface and the air into which the moisture is moving. As a result of the transfer, the atmospheric water vapour pressure will increase until equilibrium is achieved. Thus, the most important prerequisite for net evaporation across the

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water-air interface is that the atmospheric relative humidity must be less than the saturated water vapour pressures exerted by the lake water. In other words, in order to maintain continuous evaporation from saline water, saturation vapour pressure (thermodynamic activity of water) of the solution must be close to vapour pressure of the air. At each establishment of the equilibrium between the saturated vapour pressure (thermodynamic activity of water) of the solution and vapour pressure of the air, particular evaporitic minerals may precipitate from the solution. Therefore, water activities deduced from evaporitic minerals are approximately equal to the atmospheric vapour pressure of the air/humidity.

In the past, the occurrence of evaporite sequences has simply been explained as resulting from evaporation exceeding precipitation and a closed basin. The constraining control of atmospheric relative humidity has been largely ignored. For the formation of late stage evaporation facies minerals, very low mean relative humidities in a locality are required. In general, the average relative humidity is high both in the equatorial region and in the polar regions. It is lowest in the subtropical regions since moisture moves outward in both directions from these zones by trade winds and westerlies (Lydolph, 1985). Mean annual relative humidity in the equatorial zone, as well as parts of higher latitudes, mainly over the oceans, is higher than 86%. The interiors of middle-latitude continents, particularly in North America, are occupied by air with low relative humidity values (below 60-50%) (Lydolph, 1985). Therefore, late stage evaporitic minerals can be precipitated only in evaporite basins in areas of high continentality.

Although the relationships among relative humidity, salinity, evaporation rate,

temperature, and other physical processes are extremely complex, and it is impossible to define a simple relationship between any two variables, a general relationship between the average relative humidity of a location and the ultimate evaporite mineral suite should exist. Figure 7.2 shows a generalized sequence of mineral precipitation with the decrease of relative humidity. The relative humidity values are mainly compiled from Kinsman (1976), Eugster et al. (1980), and Harvie et al. (1982).

7.4 Hydrological Balance and Mineral Precipitation

7.4.1 Hydrological balance

The interaction between precipitation, evaporation, groundwater and surface water in a lake determines not only its hydrological balance but also affects its hydrochemistry, which in turn affects the mineral precipitation. Relatively small changes in the balance between groundwater or surface water and precipitation inputs may induce shifts in the mineral sequences. The hydrogeology in North Ingebright Lake region is controlled by a three-layer system: glacial till overlies fluvial sand, which in turn rests on the erosional surface of Cretaceous bedrock (David, 1964). A buried preglacial valley along the course of the present-day potholes was identified by David (1964) and Ruffel (1968). The buried valley was infilled with alluvial sands and silts containing lenses of gravel. The groundwater contribution to the water balance of North Ingebright Lake basin arises from this unique local geological situation that allows the vertical movement of groundwater to the surface under an upward head gradient (David, 1964; Ruffel, 1968).

Despite the importance of groundwater inputs to playa lakes, very few studies



Relative Humidity

have been undertaken in the Canadian Prairies (Price, 1993; van der Kamp and Maathuis, 1991; Woo, 1992; Birks and Remenda, 1999). In most cases, groundwater inflow has been omitted because of the difficulty of its measurements, or calculated as a residual term. In this study, the groundwater contribution is considered to be an important component of the water balance and the hydrogeological mechanisms operating within North Ingebright Lake.

Playa lakes in hummocky moraine are either expressions of groundwater discharge from miniature flow systems or infiltration ponds from which groundwater is being replenished. Meyboom (1966) presented the annual fluctuation patterns of groundwater flows underneath a typical playa lake in hummocky moraine of southcentral Saskatchewan. One annual sequence of groundwater flow conditions approximately consists of three flow patterns. From January to April, groundwater flow in the playa lake area is downward and the lake basin acts as a major recharge area for deeper aquifers. As soon as the melt water accumulated in the lake basin starts to infiltrate, a complete flow pattern of lateral and vertical dissipation forms around the playa basin. During the summer and fall a reversal of groundwater flow occurs. This condition of inverted water flow is characterized by radial flow toward the lake basin and groundwater discharges to the lake basin. This annual rechargedischarge cycle actually also operates on the long term scale in playa lakes. During the dry period, the lake basin is maintained by groundwater discharge. The lack of lunette dunes around the playa basin further indicates that the lake basin has never experienced long-term desiccation. During the wet period, lake water recharges to the groundwater aquifers.

The thermodynamic properties of the ionic components of the groundwaters and brine have been calculated using a computer program PHRQPITZ (Plummer et al, 1988). The ionic strengths, activity coefficients of major ions and water, and saturation indices with respect to major salt minerals are listed in Table 7.1. The results reveal that shallow groundwater, deep groundwater, and spring waters are all strongly supersaturated with respect to calcite, dolomite, and aragonite. Magnesite is at or near saturation in all water sources, suggesting precipitation of magnesite is thermodynamically favoured in the lake basin. Gypsum, amhydrite, mirabilite and halite are undersaturated everywhere, and therefore should not be precipitating at present conditions.

7.4.2 Numerical simulation-mineral precipitation

Although the general water activity and evaporite mineral facies relationship shown in Figure 7.2 are well established by experiments and theoretical calculations (see Assarsson, 1950a; 1950b; 1950c; Braitsch, 1971; Eugster, 1971; Stewart, 1963; Harvie et al., 1984; Weare, 1987), it is still necessary to examine the actual order of mineral precipitation in an individual lake basin before evaporite mineral facies can be used as indicators of relative humidity. As discussed previous**I**y, North Ingebright Lake is maintained by either shallow groundwater influx, deep groundwater discharge or mixtures of both. Significant mineral composition wariation in the evaporite mineral suite is a result of complex alternating inflow processes of shallow groundwater influx and deep groundwater discharge. In this research, the computer program, PHRQPITZ has been used to simulate evaporation pæths

			·		
	Deep groundwater	Shallow groundwater	Spring water	Lake water	
Ca (mol)	4.69e-03	4.37e-03	3.39e-03	n.d.	
Mg (mol)	2.98e-03	5.38e-03	9.42e-03	1.35e-01	
Na (mol)	1.71e-02	7.36e-03	6.01e-02	5.07e-01	
Cl (mol)	1.84e-03	1.71e-03	8.62e-03	1.59e-01	
CO ₃ +HCO ₃ (mol)	4.70e-03	4.91e-03	1.81 e -02	1.79 e -02	
SO₄(mol)	1.23e-02	8.98e-03	5.52e-03	3.00e-01	
pH	7.5	7.8	7.4	7.9	
⁴H₂O	0.9993	0.9995	0.9983	0.9853	
I (Ionic strength)	0.0515	0.0443	0.0793	1.2117	

Table 7.1 Chemical parameters of deep groundwater, shallow groundwater,spring water, and lake water from North Ingebright lake (from Cole;1926 and David, 1968)

Saturation Index: log (IAP/KT)

Calcite	0.2	0.6	0.5	n.c.
Aragonite	0.0	0.4	0.3	n.c.
Dolomite	0.4	1.4	1.6	n.c.
Magnesite	-0.6	0.0	0.3	1.3
Gaylussite	-7.8	-7.9	-6.0	n.c.
Pirssonite	-17.2	-8.0	-6.2	n.c.
Natron	-8.3	-8.6	-6.8	n.c.
Gypsum	-0.5	-0.6	-1.1	n.c.
Anhydrite	-0.8	-0.9	-1.3	n.c.
Mirablite	-4.3	-5.2	-3.7	-1.0
Bloedite	-8.9	-9.6	-8.3	-3.7
Epsomite	-3.3	-3.1	-3.2	-1.4
Hexanhydrite	-3.5	-3.4	-3.5	-1.7

Kieserite	-5.1	-5.0	-5.1	-3.3
Halite	-6.2	-6.6	-5.0	-3.1
Bischofite	-13.1	-12.8	-11.3	-8.3
Glauberite	-5.9	-6.8	-5.8	n.c.
Burkeite	-20.4	-22.5	-17.6	-10.2

n.d. - not detected; n.c - not calculated.

(Plummer et al., 1988). PHRQPITZ uses thermodynamic equilibrium and mass balance principles and contains the Pitzer virial-coefficient approach for calculating the thermodynamics of minerals in solutions with high ionic strength developed by Harvie et al. (1980 and 1984).

Throughout most of its history, the basin is assumed to have been hydrologically closed and evolved with a constant volume of water. Hence, the evaporated volume of water was replaced by shallow groundwater or deep groundwater, and complete mixing was assumed. Using PHRQPITZ to simulate this closed system is accomplished by a two-step procedure. In the first step, shallow and deep groundwaters are speciated in order to calculate ionic strengths, thermodynamic activity, and saturation indices with respect to various minerals. Program options within PHRQPITZ were used to mix shallow groundwater and deep groundwater at ratios of 1:9, 3:7, 5:5, 7:3, and 9:1. Results of the speciation calculation for these mixing solution are presented in Table 7.2. They constitute a spectrum of inflow waters to North Ingebright Lake basin in different stages of the basin's history.

In the second step, evaporation to complete dryness of each of inflow waters in Table 7.2 is simulated using the computer code PHRQPITZ. Mineral precipitation during evaporation is governed by the thermodynamic equilibrium between the minerals and the brine at a constant temperature of 15°C. Due to the complex relationships between multiple solute components and mineral precipitation, numerical methods must be used to solve the mass balance problems. Thus, the simulation process is undertaken in a series of evaporation steps. At each step, the

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	Deep	Shallow	Mixing solut	ions 1 and 2 in di	fferent proportion	ns (ratio=solution	1/solution 2)
	groundwater (solution 2)	groundwater (solution 1)	9:1	7:3	5:5	3:7	1:9
Ca (mol)	4.69e-03	4.37e-03	4.66e-03	4.59e-03	4.53e-03	4.47e-03	4.40e-03
Mg (mol)	2.98e-03	5.38e-03	3.22e-03	3.70e-03	4.18e-03	4.66e-03	5.14e-03
Na (mol)	1.71e-02	7.36e-03	1.61e-02	1.42e-02	1.22e-02	1.03e-02	8.33e-03
Cl (mol)	1.84e-03	1.71e-03	1.82e-03	1.80e-03	1.77e-03	1,75e-03	1.72e-03
CO3 +HCO3(mol)	4.70e-03	4.91e-03	4.72e-03	4.76e-03	4.28e-03	4.85e-03	4.89e-03
SO₄(mol)	1.23e-02	8.98e-03	1.19e-02	1.13e-02	1,06e-02	9.96e-03	9,30e-03
рН	7.5000	7.8000	7.4885	7.4667	7.4462	7,4269	7.4087
"H ₂ O	0.9993	0.9995	0.9994	0,9994	0.9994	0.9995	0.9995
I (Ionic strength)	0.0515	0.0443	0.0508	0.0493	0.0479	0.0464	0.0450
Saturation Inde.	x: log (IAP/KT)						
Calcite	0.2	0.6	0.2	0.2	0.2	0.2	0,1
Aragonite	0.0	0.4	0.0	0.0	0.0	0.0	-0.1
Dolomite	0.4	1.4	0.4	0.4	0.4	0.5	0.5

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 Table 7.2 Chemistry composition of various mixing solutions

Magnesite	-0.6	0.0	-0.6	-0.6	-0.5	-0,5	-0.5
Gaylussite	-7.8	-7.9	-7.9	-8.0	-8.2	-8.4	-8.6
Pirssonite	-17.2	-8.0	-8.1	-8.2	-8.4	-8.6	-8.8
Natron	-8.3	-8.6	-8.3	-8.5	-8.6	-8.8	-9.0
Gypsum	-0.5	-0.6	-0,5	-0.5	-0.6	-0.6	-0.6
Anhydrite	-0.8	-0.9	-0,8	-0.8	-0.8	-0.8	-0.9
Mirablite	-4.3	-5,2	-4.4	-4.5	-4.7	-4,9	-5.1
Bloedite	-8.9	-9.6	-9,0	-9.1	-9,2	-9.3	-9.5
Epsomite	-3.3	-3,1	-3.2	-3.2	-3.1	-3.1	-3.1
Hexanhydrite	-3.5	-3,4	-3.5	-3,4	-3,4	-3.4	-3.4
Kieserite	-5.1	-5.0	-5.1	-5.1	-5.0	-5.0	-5.0
Halite	-6.2	-6.6	-6.3	-6.3	-6,4	-6.5	-6,6
Bischofite	-13.1	-12.8	-13.0	-13.0	-12.9	-12.9	-12.8
Glauberite	-5.9	-6.8	-5,9	-6.1	-6,3	-6.5	-6.7
Burkeite	-20.4	-22.5	-20.5	-20,9	-21.4	-21,9	-22.5

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evaporation of a certain volume of water is achieved by simply adding the solute mass that is representative of solutes in the input water to the evaporating brine rather than by the removal of water from the solution (Ayora et al., 1994; Wood and Sanford, 1990; Sanford and Wood, 1991). The inflow water may be either shallow groundwater, deep groundwater or mixtures of both. During each series of evaporation steps, the number of basin volumes evaporated at that step depends on the concentration of the starting solution and the stage of evaporation. For relatively dilute shallow groundwater, a time-step size of 1-50 evaporated basin volumes was chosen. A time-step size of 0.1-20 evaporated basin volumes was used for the deep groundwater evaporation simulation because of its relatively high concentration. The result indicates that the choice of time-step size in this study is adequate for the evaporation simulation to illustrate the effects of hydrological balance on the sequence of mineral precipitation.

Output from PHRQPITZ is comprehensive. Following a printout of the input data set, the starting solution is speciated for each evaporation step. The ionic strengths, activity coefficients of major ions and water, and the saturation state of minerals are also included. By visually inspecting the saturation indices of the minerals at the different evaporation steps, the point in time when any new mineral would begin to precipitate can be identified. Correspondingly, the sequence of mineral precipitation for each inflow waters in Table 7.2 can be established.

The sequence of mineral precipitation, with respect to different inflow waters, are listed as a function of number of evaporated lake volume in Tables 7.3a-g. For most of the inflow waters, the temporal succession of minerals remains basically the

Table	7.3a	Minera	l facies	s as a f	unctio	n of eva at 1	porateo	i lake v	olumes	s for s	hallov	v grou	ndwater		
Mineral Facies	Cal	Doł	Mag	G	A	Ga	Pi	Ga	М	Bl	H	Ер	Hx	Ki	Bi
Evaporated Lake Volume	0	0	l	3	6	30	35	125	350	550	780		880		
Water Activity	0.9995	0.9995	0,9991	0.9983	0,9971	0,9884	0.9866	0,9528	0,8289	0,702	0,574		0.5291		
Ionic Strength	0.0443	0.0443	0,0866	0.17	0.2955	1.327	1.5466	5.5854	15.2153	22.73	30,96		35.5979		

Table 7.3b Mineral facies as a function of evaporated lake volumes for mixing solution of shallow and deep groundwaters (Ratio = 9:1) at 15°C

Mineral Facies	Cal	Dol	Mag	G	А	Ga	Pi	Gl	М	Bl	Н	Ер	Hx	Ki	Bi
Evaporated Lake Volume	0	0	I	3	5	30	35	115	350	530	750		880		
Water Activity	0,9995	0,9995	0.999	0.9982	0.9974	0.9879	0.986	0.9552	0,8244	0,709	0.587		0.5293		
Ionic Strength	0.045	0.045	0.0882	0.1733	0.2587	1.3522	1.5754	5.2117	15.4145	22,36	30,51		35,4011		

				g	roundw	vaters (Ratio =	7:3) at	15°C					•	
Mineral Facies	Cal	Dol	Mag	G	Α	Ga	Pi	Gl	Μ	Bl	Н	Ер	Нх	Ki	Bi
Evaporated Lake Volume	0	0	1	3	5	25	30	100	340	510	700	880	880	>880	
Water Activity	0.9995	0,9995	0.999	0.9981	0.9972	0.9889	0.9869	0.9585	0,8223	0.712	0.605	0.5303	0.5303	>0.5303	
Ionic Strength	0.0464	0.0464	0,0913	0,18	0.2689	1.1736	1,4033	4.6734	15.4407	22,33	29,79	36.9592	36,9592	>36,959	

Table 7.3c Mineral facies as a function of evaporated lake volumes for mixing solution of shallow and deep groundwaters (Ratio = 7:3) at 15°C

Table 7.3d Mineral facies as a function of evaporated lake volumes for mixing solution of shallow and deep groundwaters (Ratio = 5:5) at 15°C

Mineral Facies	Cal	Dol	Mag	G	A	Ga	Pi	Gl	Μ	Bl	Н	Ep	Нх	Ki	Bi
Evaporated Lake Volume	0	0	1	3	5	25	30	90	320	490	670	830	830	>830	
Water Activity	0.9994	0.9994	0.9989	0.9979	0.997	0,9881	0,986	0.9603	0.8275	0.715	0.613	0.3848	0.3848	>0.3848	
Ionic Strength	0.0479	0.0479	0.0945	0.1867	0.2791	1.2165	1.4538	4.3377	15.023	22,26	29.72	27.3787	27.3787	>27.378	

·····	groundwaters (Ratio = 3:7) at 15°C														
Mineral Facies	Cal	Dol	Mag	G	Α	Ga	Pi	Gl	Μ	Bl	Н	Ер	Нх	Ki	Bi
Evaporated Lake Volume	0	0	7	2	4			70	170	175	770	175	175	175	175
Water Activity	0,9994	0.9994	0,9956	0.9982	0.9972			0.9639	0.9032	0.901	0.411	0,9016	0.9016	0,9016	0,9016
Ionic Strength	0.0493	0.0493	0.4113	0.1528	0.2562			3.6687	6.0787	4,690	24.95	4.6902	4.6902	4.6902	4.6902

Table 7.3e Mineral facies as a function of evaporated lake volumes for mixing solution of shallow and deep groundwaters (Ratio = 3:7) at 15°C

Table 7.3f	Mineral facies as a function of evaporat	ed lake volumes	for mixing	solution of s	hallow and dee	эþ
	groundwaters	(Ratio = 1:9) at $^{\prime}$	15°C			-

Mineral Facies	Cal	Dol	Mag	G	A	Ga	Pi	Gl	М	Bl	Н	Ер	Нх	Ki	Bi
Evaporated Lake Volume	0	0	1	2	4	12	13	75	180	450	590	720	. 720	>720	
Water Activity	0,9994	0.9994	0.9988	0,9982	0.9971	0.989	0,9866	0.9629	0.9055	0,722	0,637	0,4442	0.4442	>0.4442	
Ionic Strength	0,0508	0.0508	0,1008	0.1503	0.2494	1.0491	1.3011	3.8426	9,1458	22.00	28.39	24,9171	24.9171	>24.917	

Table 7.3	g min	erai ta	icies a	s a tur	iction o	of evapo	brated	lake vo	lumes 1	for dee	ep gro	undwa	ater at 1	5°C	
Mineral Facies	Cal	Dol	Mag	G	А	Ga	Pi	Gl	М	Bl	Н	Ер	Hx	Ki	Bi
Evaporated Lake Volume	0	0	0.2	2	4	25	30	80	180	450	600	>730	>730	>730	
Water Activity	0,9993	0,9993	0.9992	0.9982	0.9970	0.9862	0.9837	0.9619	0,9015	0.711	0.614	>0.43	>0.4341	>0.4341	
Ionic Strength	0.0515	0.0515	0.0618	0.1527	0.2536	1.3219	1,5782	3.5587	9,1451	23,00	27,36	>24.1	>24.159	>24.159	

Note: Cal-calcite, Dol-dolomite, Mag-magnesite, G-gypsum, A-anhydrite, Ga-gaylussite, Pi-pirssonite, Gl-glauberite, M-mirabilite, Bl-bloedite, H-halite, Ep-epsomite, Hx-hexahydrite, Ki-kieserite, Bi-bischofite

same. This result confirms the validity of using evaporite mineral facies and relative humidity relationship, shown in Figure 7.2, for determining the relative humidity from North Ingebright Lake evaporites.

7.5 Hydrological Conditions of North Ingebright Lake

The sediment cores recovered from the basin of North Ingebright Lake indicate that the basin has experienced water level and water chemistry fluctuations, as indicated by variations in the mineral composition and organic matter content of sediment record. Overall, the entire sediment sequence was deposited in shallow water and hypersaline conditions.

There is no evidence that North Ingebright Lake was ever a deep-water lake. Shallow water and oxygenated bottom conditions were present throughout much of the Holocene, as indicated by general low organic content and absence of bedding (Teller and Last, 1981; 1990). Common occurrences of halite and K-Mg sulfates in the evaporite deposits of North Ingebright Lake imply that the salinity of lake water has frequently reached a value of approximately 400 g/l, as compared to present day values between 100 to 300 g/l.

The general lack of bedding in the sedimentary fill could have been caused by many factors. The bedding was probably disrupted and destroyed by later groundwater discharges (creating porous surfaces) or by the growth and dissolution of ephemeral evaporite crystals and crusts (Kendall, 1984). It also may have been destroyed by wind generated waves which resuspend and physically mix the offshore bottom sediments (Last and Sauchyn, 1993). In addition, the absence of bedding suggests that this lake's water column has remained mainly nonstratified through much of its history.

The lake sediment is almost completely dominated by colourless salt. The lack of original fabric and textures in the salt make it very difficulty to interpret the depth of the brine in this hypersaline lake because both deep and shallow water salt deposition has been found in modern Saskatchewan lakes (Last, 1984). However, the final purpose of lake level determination is to reconstruct climatic conditions, so the evaporite mineral assemblage is used to establish mean surface humidity condition rather than lake level.

Figure 7.3 depicts the pattern of changes in relative regional humidity reconstructed for the past 10,000 years. During the episodes of 8700-10,350 yr B.P., 5500-6500 yr B.P., and 1000-2500 yr B.P., the mean relative humidity of the North Ingebright Lake basin was considerably lower than it is today. In contrast, comparatively humid conditions (but not wet enough to cause a basic change in the hydrologic setting of the playa) are suggested by higher relative humidities during the periods of 6500-7200 yr B.P. and 5500-3000 yr B.P. The 8000 to 7000 yr B.P. period was characterized by considerable variability with rapid, short-term fluctuations from high to low relative humidity conditions.

7.6 North Ingebright Lake Relative to Other Holocene Lacustrine Records

Although the purpose of this project was not to reconstruct the regional climatic history, nonetheless, it is tempting to compare the changes in relative humidity interpreted for North Ingebright Lake with paleolimnological fluctuations at



Figure 7.3 Variation in mean atmospheric relative humidity at North Ingebright Lake as estimated by the evaporitic mineral suite. Numbers on the right refer to the lithostratigraphic unit . Temporal scale is in ¹⁴ C years B.P.

other sites in the region having long Holocene records (Tables 7.4a and 7.4b).

The 7500 year long sediment record from Chappice Lake, Alberta, a hypersaline playa located about 80 km west of North Ingebright, is one of the longest and best documented lacustrine paleohydrological records in the entire northern Great Plains region (Vance et al., 1992, 1993, 1995). Even though the temporal aspect of the North Ingebright record is much less tightly constrained than that of Chappice Lake, broad similarities do exist. For example, both basins show an early Holocene episode (prior to 6000 yr BP in Chappice and 7000-8500 yr BP in North Ingebright) of significant fluctuation in humidity and water level/salinity conditions. Similar early Holocene periods of rapidly changing conditions, from high to low water levels and from hypersaline to fresh water, are noted in the stratigraphic records from Ceylon Lake, a salt playa located on the Missouri Coteau about 400 km southeast of Ingebright (Last, 1990) and from Clearwater Lake, a perennial freshwater lake 100 km to the northeast (Last et al., 1997). Similarly, a mid-Holocene interval of low water, saline to hypersaline conditions correlative to the very low mean atmospheric relative humidity at 5000-6000 yr in North Ingebright is also documented at Chappice and Ceylon, as well as several other lacustrine sites in the Prairie region (Vance et al., 1995).

The 10,000 year long, uninterprupted sediment sequence from Oro Lake in southern Saskatchewan, a perennial saline lake located about 160 km southeast of North Ingebright, also contains the well-preserved paleolimnological fluctuation records (Last and Vance, 2000). Early Holocene transitions from fresh to saline and/or hydrologically open to closed, evaporitic conditions have been well

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¹⁴ C YR BP	North Ingebright Mean Relative Humidity	Chappice Water Level & Salinity	Oro Water Level & Salinity	Clearwater Water Level & Salinity	Ceylon Water Level & Salinity	
Present		FLUCTUATION FROM LOW			LOW & HYPERSALINE	
	HIGH	SALINE	DEEP & SALINE TO - HYPERSALINE		HIGH & FRESH WATER	
1000		LOW WATER		-	STABLE CONDITIONS	
2000	LOW			DEEP & FRESH WATER STABLE CONDITIONS	LOW WATER & HYPERSALINE	
		DEEP & RELATIVELY FRESH WATER	STABLE CONDITIONS		HIGH & FRESH WATER	
3000	HIGH		DEEP & SALINE		STABLE CONDITIONS LOW WATER & HYPERSALINE	
4000			•			
5000		LOW WATER &			VERY LOW TO DRY & HYPERSALINE	
3000	·			an a		
6000	VERY LOW		FRESH WATER			
	HIGH	REPEATED FLUCTUATION FROM DRY				
7000		TO DEEP & FRESH TO HYPERSALINE				
	FROM LOW TO HIGH		-	DEEP & FRESH WATER	FLUCTUATION FROM	
8000	·	Base of Record	STABLE CONDITIONS		SHALLOW TO DEEP & FRESH TO HYPERSALINE	
9000	HIGH		HYPERSALINE	DEEP TO SHALLOW (DRY AT 8200 BP) & SALINE TO HYPERSALINE		
10000	VERY LOW TO LOW		DEEP & FRESH WATER	SHALLOW & HYPERSALINE	DEEP & FRESH WATER	
11000	HIGH			SHALLOW & FRESH WATER		

Table 7.4a Comparison of mean relative humidity at North Ingebright Lake to interpreted water levels and salinities at Chappice Lake, Alberta, and Oro lake, Clearwater and Ceylon Lakes, Saskatchewan

¹⁴ C YR BP	North ingebright Mean Relative Humidity	Pickerel Water Level & Salinity	Medicine Water Level & Salinity	Williams Water Level & Salinity		
Present						
	HIGH	REPEATED FLUCTUATION FROM				
1000		LOW TO DEEP & FRESH TO SALINE	LOW TO DEEP SALINE TO HYPERSALINE			
	LOW			SALINE TO HYPERSALINE		
2000		· _				
3000			REPEATED FLUCTUATION			
	HIGH	LOW WATER & SALINE	FROM LOW TO DEEP & FRESH TO SALINE	LOW WATER &		
4000						
5000			LOW WATER SALINE TO HYPERSALINE			
	VERY LOW	-		REPEATED FLUCTUATION		
6000		LOW & SALINE TO HYPERSALINE		SALINE TO HYPERSALINE		
	HIGH	-				
7000	REPEATED FLUCTUATION		STABLE DEEP WATER MEROMITIC & SALINE TO			
0000	FROM LOW TO HIGH		HYPERSALINE			
8000				HTPERSALINE		
9000	HIGH					
10000	VERY LOW TO LOW	DEEP & FRESHWATER	STABLE CONDITIONS	STABLE CONDITIONS DEEP & FRESHWATER		
11000	HIGH					

 Table 7.4b Comparison of mean relative humidity at North Ingebright Lake to interpreted water levels and salinities at Pickerel Lake, Medicine Lake and Williams Lake in United States

documented at Oro Lake. Sediments (prior to 9300 yr BP) in Oro Lake were deposited in a relatively deep, freshwater lake. However, this early Holocene freshwater lake changed rapidlly to a saline lake with very high Mg/Ca ratio by about 9300 yr BP. During the two thousand year period between 9300 and 7400 yr BP, deep and saline conditions dominated the lake, with occasional excursions to hypersaline conditions. Similarly, low humid conditions are also recorded in North Ingebright Lake during 10350-8700 yr BP.

The mid-Holocene (4000-7400 yr BP) aragonite-laminated mud deposition in the Oro Lake basin reflects a positive hydrological budget. This mid-Holocene freshening of the brines in Oro Lake corresponds well with the occurrence of relatively humid conditions (6500-7200 yr BP) at North Ingebright lake.

In addition, the transitions from relatively deep and freshwater to shallow and higher salinity conditions in lakes of the northern Great Plains during the early and mid-Holocene period have also been documented at several lakes in United States. For example, dramatic increases in values of both δ^{18} O and δ^{13} C of carbonate at Pickerel Lake, South Dakota between 6000 and 10,000 yr BP indicate the shifts from hydrologically open conditions to more or less hydrological closed conditions (Dean and Schwalb, 2000). In Coldwater Lake, North Dakota, values of δ^{18} O and Sr/Ca ratios in ostracode shells in sediments suggest a transition from relatively deep and freshwater to shallow and higher salinity conditions between 6500 and 10800 yr BP (Xia et al., 1997). Shallow and higher salinity conditions during the early to middle Holocene have also been noted in Medicine Lake, South Dakota (Valero-Garces and Kelts, 1995) and Williams Lake in west-central Minnesota

(Schwalb et al., 1995).

In general, relatively humid conditions were dominant at North Ingebright Lake basin during the late Holocene period. This lengthy period of stable and relatively moist conditions between about 5000 and 2000 years ago is also recorded in numerous other basins in the northern Great Plains (see overviews in Vance, 1997; Xia et al., 1997; Lemmen, 1996; Lemmen and Vance, 1999).

Chapter 8 Summary

North Ingebright Lake originated in a remnant of a preglacial valley at about 11 ka. The continued collapse of buried ice blocks along the preglacial valley enhanced the elongate shape of the basin, and a series of smaller bodies of water coalesced to form a relatively large, freshwater lake. This stage of North Ingebright Lake is represented by clastic deposits; sediments of fine-grained sand, silt, and clay deposited under freshwater lacustrine conditions. At the close of the deposition of the Basal Black Mud Unit, climatic changes in the basin area resulted in a decrease in the size of North Ingebright Lake and a corresponding increase in chemical deposition. Playa conditions dominated the rest of its history.

Generally, the mineral suite of the sediment retrieved from the basin of North Ingebright Lake consists mainly of thenardite/mirabilite (80%), gypsum (5%), carbonates (4%), clay minerals (5%), and feldspar (3%). The most distinctive feature in North Ingebright Lake sediment is that large amounts of thenardite/mirabilite are present in most subsamples. The content of thenardite/mirabilite in samples ranges from 6 to 100%. Other very soluble evaporitic minerals, niter, polyhalite, epsomite, bloedite, anhydrite, tychite, and halite were identified in a small number of samples. Trace quantities of gaylussite, fairchildite, zemborite, and hanksite occur in some samples. The carbonate minerals present in the sediment core include protodolomite, dolomite, aragonite, high-Mg calcite, and magnesite. Minor quantities of calcite were identified in a small

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number of samples. Clastic silt and clay, and organic materials are always present but subordinate. The feldspars comprise an average of 3% of the mineral suite. Most of this amount is plagioclase (2.7%), with minor amounts of potassic feldspar (0.4%).

The mineral assemblage of North Ingebright Lake salts can be represented by phase relationships in the reciprocal quinary system Na-Mg-Ca-Cl-SO₄-H₂O and its subsystems - quaternary systems: three reciprocal systems (Na-Mg-Cl-SO₄, Mg-Ca-Cl-SO₄, and Na-Ca-Cl-SO₄) and two tetrahedra systems (Na-Ca-Mg-Cl and Na-Ca-Mg-SO₄). The bulk chemical composition calculation and physical-chemistry reasoning indicate that, through its history, North Ingebright Lake appears to have been dominated by Na-SO₄. During wet periods the brine was diluted and dominated by Na-Ca-Mg-SO₄; during dry periods it concentrated and shifted to Na-SO₄-Cl or Na-K-SO₄ type brines.

Overall, the entire sediment sequence was deposited in shallow water and saline to hypersaline conditions. There is no evidence that the lake basin ever contained a deep water lake. Shallow water and oxygenated bottom conditions were present throughout much of the Holocene as indicated by general low organic content, lack of colour, and absence of bedding. Common occurrences of halite and K-Mg sulfates in the evaporite deposits of North Ingebright Lake imply that the salinity of lake water has frequently approached 400 g/l. However, the lack of pedogenic horizons, erosional contacts, and coarse grained indicates that the lake has never undergone complete drying and desiccation.

The evaporite minerals were used as indicators of relative humidity

throughout the stratigraphic column. The wide range of relative humidity (34-98%) deduced from the saline mineral assemblage indicates that dramatic changes of climatic and hydrological conditions occurred during the deposition of North Ingebright Lake evaporates. During the episodes of 8700-10,350 yr B.P., 5500-6500 yr B.P., and 1000-2500 yr B.P., the North Ingebright Lake basin experienced climatic conditions in which the mean relative humidity was considerably lower than that of today. In contrast, comparatively humid conditions (but not wet enough to cause a basic change in the hydrologic setting of the playa) are suggested by higher relative humidities during the periods of 6500-7200 yr B.P. and 5500-3000 yr B.P. Finally, the millennium between 8000 and 7000 yr B.P. was characterized by considerable variability with rapid, short-term fluctuations from high to low relative humidity conditions.

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Appendix A Organic Matter and Carbonate Content Data

.

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
. ,					
12	12.8	3.9	108	0.4	0.5
14	11.7	4.2	110	0.6	0.0
16	8.5	7.6	112	0.6	0.1
18	8.8	9.2	114	0.6	0.2
20	12.6	5.6	116	0.6	1.5
22	14.5	4.5	118	0.1	0.6
24	9.2	8.4	120	0.1	0.7
26	13.2	4.6	122	0.4	1.0
28	13.8	6.8	124	0.4	1.0
30	12.5	7.3	126	0.2	0.5
32	14.8	3.2	128	0.1	0.6
34	12.3	5.6	130	0.1	0.7
36	13.9	3.4	132	2.3	1.5
38	12.6	3.9	134	0.9	0.7
40	10.9	6.5	136	1.0	1.0
42	13.2	2.3	138	0.5	0.9
44	12.2	3.1	140	0.8	1.9
46	8.7	3.2	142	0.9	1.6
48	11.1	1.4	144	0.1	0.6
50	4.3	5.1	146	3.6	5.6
52	4.4	I.4	148	4.8	6.1
54	3.2	2.0	150	4.6	2.0
56	4.1	0.8	152	5.9	2.1
58	1.8	1.3	154	0.7	0.3
60	2.7	0.5	156	0.5	0.8
62	1.7	1.5	158	0.5	0.5
64	2.4	0.9	160	0.5	0.6
66	1.7	1.7	162	0.6	0.4
68	1.6	2.8	164	0.6	0.4
70	2.1	0.3	166	1.0	0.6
72	2.6	0.4	168	0.3	0.1
74	1.9	0.9	170	1.0	0.0
76	1.8	2.3	172	0.8	0.4
78	1.7	0.5	174	0.2	0.1
80	1.9	7.2	176	0.3	0.0
82	1.4	0.4	178	1.2	0.5
84	1.8	0.6	180	1.5	0.2
86	1.2	0.6	182	1.1	0.3
88	2.3	1.2	184	0.8	0.0
90	2.6	0.6	186	1.5	0.5
92	1.2	0.1	188	1.4	0.7
94	1.2	0.4	190	0.8	0.2
96	1.5	0.4	192	3.3	0.2
98	0.5	1.0	194	1.9	0.0
100	1.7	0.4	196	2.3	0.6
106	1.0	0.6	198	2.6	0.0
200	2.5	0.2	292	2.4	0.7
202	2.7	0.0	294	2.0	12
204	2.6	0.3	296	2.5	0.2
206	2.1	1.0	298	1.9	0.5
208	2.7	0.6	300	3.1	14

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
210	2.7	0.7	302	3.6	0.8
212	2.9	0.2	304	2.4	1.1
214	3.2	0.8	306	2.0	1.0
216	2.5	0.3	308	2.0	0.1
218	3.7	0.6	310	1.6	1.1
220	2.9	0.8	312	1.2	1.8
222	3.9	0.5	314	1.7	0.3
224	4.7	0.9	316	1.6	0.4
226	3.4	0.7	318	0.9	1.6
228	4.3	0.7	320	0.9	1.0
230	3.5	0.9	322	1.2	0.6
232	3.6	0.5	324	1.4	0.7
234	4.5	0.8	326	1.0	0.7
236	5.5	0.0	328	10	0.8
238	4.5	0.4	330	1.0	1.0
240	5.0	0.0	332	13	0.5
242	3.8	0.9	334	13	0.3
244	3.9	12	336	0.7	0.5
246	4.5	0.9	338	1.6	0.4
248	51	10	340	21	1.0
250	49	0.5	347	0.8	0.5
252	41	1.8	344	0.0	0.5
254	55	13	346	1.0	03
256	45	0.7	348	0.5	0.3
258	37	2.0	350	29	0.5
260	37	0.8	357	0.7	0.5
267	51	1.4	354	1.0	10.0
264	4.5	2.5	356	0.0	0.0
266	4.8	0.0	358	0.9	0.4
268	4.5	27	360	1.1	0.4
270	4.1	67	367	1.0	0.0
270	3.8	1.6	364	1.4	0.9
274	3.9	1.0	366	1.7	0.0
276	41	1.4	368	1.2	5.4
278	3.4	0.9	370	1.5	0.5
280	5. 4 A A	0.9	370	0.9	3.4
200	3.7	0.9	372	0.8	0.1
202	3.7	1.0	374	0.8	3.9
204	3.2	1.5	370	0.0	4.2
200	5.0 7 /	1.0	81 C	1.3	0.3
200	2.4	0.6	00C 202	0.0	0.0
384	0.8	0.0	362	0.9	0.4
386	0.8	50	470	4.1 2 2	U.I
382	12	0.0	470 500	2.2	10.4
300	1.0	0.0	200	1.9	0.2
300	0.0	0.1	502	1.8	1.3
394 204	U_4 0.6	0.3	504	2.4	6.6
374 206	0.0	0.4	506	2.3	0.4
206	0.5	0.9	508	1.5	0.1
378	0.2	0.8	510	3.0	17.1
400	0.4	0.3	512	1.9	0.5
402	U.4	U.I	514	4.0	2.0

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
、 /					
404	0.1	0.2	516	5.5	0.2
406	0.0	0.2	518	5.3	2.3
408	0.1	0.6	520	5.0	0.9
410	0.1	0.1	522	4.0	0.3
412	0.7	0.1	524	19.6	8.8
414	0.1	0.5	526	1.0	0.3
416	0.1	0.4	528	1.2	8.6
418	0.4	0.1	530	1.5	1.2
420	1.0	0.1	532	1.2	1.0
422	2.0	4.2	534	1.0	0.8
424	0.1	0.5	536	2.4	0.7
426	0.4	0.2	538	0.8	0.0
428	0.4	0.1	540	1.0	0.2
450	0.4	0.2	542	1.4	29.4
452	0.4	2.6	544	2.0	0.1
454	0.7	0.5	546	0.8	0.0
456	0.7	0.1	548	0.2	0.7
458	0.5	0.2	550	2.4	0.5
460	0.9	0.7	552	1.0	0.2
462	0.4	0.5	554	1.7	1.8
464	0.5	0.3	556	1.0	0.1
466	0.7	18.3	558	3.4	0.5
468	0.5	0.3	560	1.5	0.7
470	0.5	0.1	562	2.4	0.3
472	0.7	0.4	564	2.6	3.6
474	0.7	0.2	566	1.6	0.2
476	2.7	1.4	568	2.4	0.9
478	3.0	0.3	570	1.5	4.9
480	1.5	0.2	572	2.0	0.1
482	1.7	0.5	574	3.2	1.0
484	2.7	0.8	576	1.1	0.4
486	1.7	0.2	578	0.8	0.2
488	2.1	0.1	580	0.9	1.6
490	1.4	0.3	582	1.5	0.8
492	2.0	0.0	584	1.7	0.6
494	2.6	0.2	586	0.6	0.1
588	0.5	0.6	684	2.2	0.8
590	0.9	0.2	686	1.7	0.0
592	1.0	0.5	688	1.4	0.3
594	1.4	0.0	690	0.8	0.4
596	1.4	74.2	692	0.8	1.3
598	2.3	2.8	694	1.1	0.0
600	4.9	1.5	696	0.5	1.1
602	2.7	5.4	698	1.1	0.9
604	3.4	0.7	700	1.3	11.8
606	4.5	0.6	702	1.6	0.5
608	2.9	1.1	704	2.7	1.4
612	2.9	0.6	706	0.6	0.4
616	0.7	0.9	710	0.5	0.5
618	0.5	0.0	712	1.3	1.0
620	1.0	0.7	714	0.5	0.5

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%).	Content(%)
677	1 3	0.0	716	1 0	1 3
624	1.3	0.0	710	1.8	1.5
024	1.2	3.7	718	1.2	1.0
620	1.0	1.0	720	1.0	0.3
670	2.0	12.3	722	2.0	0.2
672	3.0	1.0	724	2.0	0.0
634	2.2	0.9	720	1.0	0.4
676	4.3	0.9	728	1.5	1.0
679	4.5	0.8	730	2.0	0.7
640	1.0	0.2	734	5.0	0.9
642	4.7	3.7 0.6	736	1.2	0.4
644	2.4	0.0	730	1.0	2.5
646	2.0	0.0	738	5.0	2.5
640	4.0	0.0	740	1.0	1.9
650	4.2	0.0	742	1.5	1.0
657	20	0.0	744	1.9	1_7
654	3.0	0.8	740	2.1	1.1
656	2.5	1.4	740	0.0	0.0
650	1.2	0.1	750	2.2	1.0
660	1.2	0.4	752	2.0	19.5
000	1.0	0.7	754	3.0	0.4
002	1.9	2.2	756	1.7	0.1
004	1.7	0.4	758	1.0	0.0
000	1.6	0.3	760	1.0	1.2
008	1.2	1.8	762	1.4	0.6
670	1.4	0.0	764	1.4	2.0
6/2	2.0	0.0	792	0.9	0.0
0/4	2.2	0.0	794	1.6	27.8
676	1.2	0.0	/96	1.9	1.0
6/8	1.8	0.1	/98	1.5	0.4
680	1.6	0.0	800	2.0	0.0
082	1.3	0.5	802	1.0	1.0
804	0.9	0.6	900	2.2	3.1
806	1.7	0.6	902	3.6	2.0
808	4.0	1.9	904	2.6	0.0
810	16.7	3.1	906	3.0	0.2
812	3.7	0.0	908	2.9	0.2
814	2.1	0.5	910	3.0	0.5
816	2.3	1.9	912	4.1	0.2
818	3.0	0.4	914	5.8	1.4
820	2.8	0.9	916	7.1	1.9
822	5.0	0.7	918	5.8	1.6
824	2.8	6.4	920	1.9	0.6
826	3.2	0.0	922	2.1	1.5
828	6.4	0.0	924	4.4	1.2
830	4.1	0.7	926	3.0	3.1
832	3.1	1.0	928	3.1	0.6
834	4.2	0.5	930	3.0	0.1
836	3.8	0.6	932	3.6	0.2
838	5.8	1.1	934	3.1	1.8
840	3.3	0.4	936	3.0	0.6
842	4.7	0.2	938	2.6	0.6

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
					0.6
844	0.3	1.3	940	2.7	0.6
840	4.1	0.7	942	3.1	1.5
848	4.1	1.0	944	2.1	0.7
850	3.9	0.7	946	3.0	0.8
852	3.5	59.2	948	3.1	0.7
820	4.3	0.0	950	2.0	0.7
860	4.3	0.9	952	4.1	1.3
862	4.0	0.6	954	29.1	5.3
864	4.0	.0.7	956	1.6	0.6
866	4.1	0.3	958	1.8	0.5
868	2.2	14.2	960	1.4	0.8
870	1.5	2.6	968	1.6	0.7
872	4.1	0.7	970	1.6	0.9
874	2.6	1.6	972	2.3	0.9
876	4.4	1.4	974	1.5	0.7
878	1.9	0.0	978	2.1	0.6
880	1.5	3.7	980	1.8	0.9
882	3.4	0.4	982	1.7	0.4
884	3.2	1.3	984	2.0	0.7
886	3.1	0.8	986	3.4	0.8
888	2.7	1.0	988	1.2	0.7
890	3.0	0.9	990	1.7	0.4
892	2.8	0.6	992	1.2	0.4
894	2.6	0.4	994	3.2	0.9
896	2.0	0.2	996	4.2	0.9
898	1.9	1.6	998	4.7	1.0
1,000	3.4	0.9	1,096	1.6	0.0
1,002	1.3	0.8	1,098	1.0	0.8
1,004	1.0	0.5	1,100	1.2	0.0
1,006	1.3	0.0	1,102	0.6	0.0
1,008	2.0	0.0	1,104	1.1	0.2
1,010	2.3	0.5	1,106	1.5	0.0
1,012	1.2	0.2	1,108	1.1	0.1
1,014	2.3	0.9	1,110	0.8	0.0
1,016	3.2	0.9	1,112	0.7	0.0
1,018	3.2	0.8	1,114	0.9	0.1
1,020	1.6	0.8	1,116	0.9	0.1
1,022	1.3	1.7	1,118	0.9	0.0
1,024	2.5	0.8	1,120	1.1	0.0
1,026	2.1	0.7	1,122	0.3	0.1
1,028	1.5	0.2	1,124	0.9	0.3
1,030	3.1	1.3	1,126	1.3	0.3
1,032	1.6	. 0.6	1,128	0.8	8.4
1,034	1.3	0.9	1,130	2.1	0.1
1,038	2.5	1.5	1,132	I.4	0.0
1,040	0.9	0.4	1,134	1.6	0.0
1,042	1.5	0.8	1,136	1.2	0.0
1,044	1.4	5.7	1,138	1.1	0.0
1,046	0.6	0.9	1,140	0.8	0.5
1,048	1.1	0.4	1,142	1.3	0.0
1.050	1.4	1.1	1,144	2.5	0.2

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
1,052	1.5	0.6	1 148	10	0.0
1.054	24	0.0	1,140	13	0.0
1,054	1.4	0.0	1 157	12	0.0
1.058	1.4	0.0	1,154	2.0	0.2
1.060	16	0.0	1,156	2.0	0.2
1.062	1.0	0.9	1,150	19	0.3
1,064	1.5	0.2	1,150	19	0.8
1,066	1.5	0.4	1,162	21	0.0
1.068	1.4	0.4	1,162	31	0.4
1.070	2.6	0.2	1,165	25	0.4
1,070	3.2	0.2	1,168	2.5	0.0
1.074	3.2	0.4	1,108	2.4	0.0
1,074	2.2	0.7	1,170	5.0	0.5
1.078	2.2	0.7	1,174	20	0.0
1,078	2.0	0.4	1,174	2.9	0.5
1,080	2.5	0.7	1,170	0.9	0.4
1,082	1.9	7.1	1,170	0.9	0.5
1 088	1.4	0.1	1,100	5.0	0.5
1,000	1.4	0.0	1,102	2.0	0.4
1,090	1.7	0.9	1,104	2.0	0.8
1,092	1.0	0.5	1,100	1.2	0.1
1,094	0.7	0.0	1,100	0.8	0.0
1,190	0.7	0.0	1,364	3.2	0.8
1,192	0.0	0.2	1,380	3.5	0.0
1,194	0.7	0.0	1,388	4.4	0.9
1 200	0.5	0.0	1,390	5.0	0.8
1,200	0.0	0.5	1,392	5.4	0.2
1,202	0.4	0.2	1,394	4.4	0.0
1,204	0.6	0.2	1,390	1.7	0.0
1,200	5.0	0.0	1,398	1.2	1.0
1,200	5.2	2.0	1,400	1.5	0.0
1,210	0.0	0.0	1,402	1.5	0.0
1,212	0.0	0.0	1,404	J.I 4 0	0.0
1,214	0.0	0.0	1,400	4.8	0.0
1,210	0.0	0.0	1,408	1.1	0.0
1,210	0.0	0.0	1,410	1.9	0.0
1,220	0.2	0.0	1,412	2.5	0.0
1 224	0.0	0.0	1,414	1.0	0.0
1,326	0.0	0.0	1,410	0.8	0.0
1,320	0.4	0.6	1,410	0.2	0.3
1,320	0.4	0.3	1,420	0.1	0.1
1 2 2 2	0.4	0.5	1,422	0.0	0.3
1 334	0.1	0.0	1,424	0.0	0,2
1,334	1.0	0.0	1,420	0.0	0.0
1 3 3 5	1.4	0.0	1,428	0.0	0.0
1 340	1.1	U.0 8 7	1,450	0.7	0.2
1,340	د ۲۰۲	0.1	1,452	0.0	2.4
1 344	1.5	0.5	1,434	0.0	U.I
1 344	<u>ئ</u> ين ۲ ۲	0.9	1,430	U.0	0.0
1,340	2.0	C.1	1,428	1.5	0.2
1350	1.0	0.0	1,940	0.8	0.1
1,000	J.0	1)	1.442	U.ð	01

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
. ,					
1 3 5 3	4.0	2.1			0.0
1,352	4.9	2.1	1,444	0.9	0.0
1,354	4.3	1.9	1,440	1.3	0.1
1 260	3.0	0.7	1,448	1.0	0.4
1,258	3.4 2.4	1.9	1,450	2.3	0.0
1,200	3.4	0.8	1,452	2.3	1.1
1,302	4.0	1.4	1,454	1.2	0.2
1,304	3.7	1.9	1,456	1.2	1.5
1,300	2.0	1.1	1,458	0.6	0.3
1,368	2.2	0.7	1,460	0.9	0.1
1,370	3.8	0.0	1,462	1.1	0.3
1,2/2	3.8	2.0	1,464	1.4	0.2
1,374	3.3	0.5	1,466	0.4	12.2
1,376	4.4	1.0	1,468	2.5	0.6
1,378	3.2	3.4	1,470	0.6	0.3
1,380	3.4	0.6	1,472	0.7	0.3
1,382	3.7	4.1	1,474	0.5	0.5
1,476	0.6	0.0	1,568	1.8	0.0
1,478	0.5	11.1	1,570	0.9	2.5
1,480	0.8	0.0	1,572	1.7	0.3
1,482	0.0	0.0	1,574	0.6	0.2
1,484	1.1	0.1	1,576	0.6	0.1
1,486	0.6	0.2	1,578	0.8	0.2
1,488	0.8	0.1	1,580	0.1	0.0
1,490	0.6	0.1	1,582	0.0	0.0
1,492	2.4	0.2	1,584	0.0	0.0
1,494	0.6	0.1	1,586	1.9	0.3
1,496	0.8	0.1	1,588	4.9	0.3
1,498	0.7	0.0	1,590	0.7	5.5
1,500	1.3	0.2	1,592	1.4	0.3
1,502	0.5	0.8	1,594	1.3	0.0
1,504	1.1	0.1	1,596	2.1	0.0
1,506	7.5	0.0	1,598	1.0	0.0
1,508	0.8	0.0	1,600	0.8	0.4
1,510	0.7	0.1	1,602	0.6	0.0
1,512	1.9	0.4	1,604	0.9	0.3
1,514	0.9	0.2	1,606	1.1	0.0
1,516	0.8	0.2	1,604	0.9	0.3
1,518	0.9	0.2	1,608	1.2	0.0
1,520	1.1	0.1	1,610	0.7	0.0
1,522	0.7	0.4	1,612	0.7	0.0
1,524	1.0	0.0	1,614	0.9	0.2
1,526	1.2	0.1	1,616	0.7	0.0
1,528	2.2	0.4	1,618	5.1	0.0
1,530	0.5	0.0	1,620	1.1	0.0
1,532	0.7	0.1	1,622	1.1	0.7
1,534	1.7	0.2	1,624	1.7	0.0
1,536	0.4	0.3	1,626	0.8	0.2
1,538	0.9	0.3	1,628	0.5	0.2
1,540	0.7	0.0	1,630	0.0	0.0
1,542	0.0	0.0	1,632	0.2	0.0
1,544	0.0	11.7	1,634	0.5	0.0

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

.

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
1 546	0.4				
1,540	0.4	0.1	1,030	0.4	0.0
1,548	1.1	0.0	1,638	0.0	0.0
1,550	0.3	0.0	1,640	0.4	0.3
1,552	0.4	0.1	1,042	0.0	0.0
1,554	1.7	0.0	1,044	11.0	0.0
1,550	2.1	0.0	1,040	1.2	0.6
1,330	1.1	0.1	1,048	0.7	0.4
1,500	0.7	0.1	1,050	0.3	0.0
1,502	1.5	0.2	1,652	0.1	19.9
1,504	0.8	0.1	1,654	0.0	0.5
1,300	1.1	0.1	1,000	0.0	0.0
1,038	0.7	0.0	1,798	0.6	0.0
1,000	1.4	0.0	1,800	0.7	0.3
1,002	2.7	0.4	1,802	0.2	0.9
1,004	1.5	13.5	1,804	0.2	0.2
1,000	1.0	1.1	1,800	0.4	0.2
1,000	0.7	2.5	1,808	0.2	0.4
1,070	0.2	0.0	1,810	0.2	0.6
1,072	0.8	0.0	1,812	1.1	0.0
1,074	1.0	0.0	1,814	0.4	0.0
1,070	0.9	0.5	1,810	0.2	0.7
1,070	0.4	1.4	1,010	0.2	0.2
1,000	0.4	0.1	1,820	0.0	0.0
1,730	1.9	0.3	1,022	0.0	0.0
1 734	0.9	0.2	1,024	0.0	0.2
1,734	18	0.0	1,020	0.0	0.0
1,739	7.0	0.0	1,828	0.0	0.0
1,730	2.0	1.8	1,030	0.1	0.0
1,740	0.8	1.0	1,054	0.0	0.0
1,742	0.7	0.2	1,034	0.0	0.4
1,744	0.9	0.0	1,050	0.7	0.0
1,740	1.0	0.4	1,912	0.7	0.0
1,750	0.5	0.0	1,914	0.4	0.0
1,750	10	0.4	1,910	0.1	0.2
1,752	0.8	0.3	1,918	0.5	0.0
1,756	0.0	0.7	1,920	0.1	0.1
1,758	0.0	0.0	1,922	0.0	0.8
1,750	0.0	0.0	1,924	0.0	1.0
1,700	0.3	0.0	1,920	0.9	13.3
1,702	0.2	0.0	1,920	0.0	0.0
1 766	0.0	0.4	1,930	0.2	0.0
1 768	01	2.0	[934	0.2	0.0
1 770	0.0	12	1 036	72	0.0
1 772	0.2	0.0	1 038	0.6	0.4
1 774	0.2	0.0	1,990	0.0	0.0 5 A
1 776	0.0	0.3	1,340	0.0	J.4 0.0
1,778	0.0	0.3	1,742	0.2	0.0
1 780	0.3	0.3	1,244	0.4	0.2
.,,00	0.0	0.0	1,740	0.0	0.1
1 782	0.6	0.7	I OA V	0.0	<u> </u>

Organic Matter and Carbonate Content of North Core at North	(ngenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
1 786	10	0.3	1.052	0.0	0.1
1,788	0.4	0.3	1,932	0.0	0.1
1,700	0.4	0.3	1,954	0.0	0.4
1 792	0.0	2.0	1 958	0.0	0.0
1,792	0.0	2.0	1,558	0.0	0.0
1,796	1.0	0.7	1,500	0.5	0.0
1,750	0.3	0.7	2,502	0.2	0.0
1,966	1.4	0.2	2,058	0.0	0.0
1.968	0.9	1.2	2,000	0.0	0.0
1 970	0.5	1.5	2,002	0.0	0.5
1 972	0.7	0.0	2,004	0.0	0.0
1 974	0.5	0.2	2,000	0.0	0.0
1 076	0.5	0.0	2,008	0.0	0.0
1.078	0.5	0.0	2,070	0.0	0.0
1 090	0.0	0.5	2,072	0.0	0.2
1 092	0.4	0.2	2,074	0.0	0.7
1,902	0.7	0.0	2,070	0.0	0.1
1,904	0.0	0.0	2,076	1.4	3.2
1,900	0.5	0.4	2,080	0.0	0.2
1,900	0.0	0.0	2,002	0.0	0.4
1,000	0.4	0.4	2,004	0.0	2.2
1,992	0.0	0.8	2,080	0.0	0.1
1,994	0.7	0.0	2,000	0.0	0.5
1,008	0.7	0.0	2,090	0.0	0.5
7,000	0.0	0.9	2,092	1.1	0.0
2,000	0.3	0.2	2,094	0.0	0.0
2,002	0.0	0.3	2,090	0.1	0.0
2,004	0.0	0.3	2,098	0.0	0.1
2,000	0.0	0.4	2,100	0.0	0.0
2,000	0.0	0.3	2,102	9.2	0.0
2,010	0.1	0.7	2,104	0.0	0.0
2,012	0.1	0.2	2,100	0.0	0.0
2,014	0.4	0.2	2,108	2.8	23.4
2,010	0.3	0.3	2.110	0.1	0.2
2,010	0.2	0.5	2,112	0.5	0.0
2,020	0.0	0.6	2,114	0.5	0.0
2,022	0.0	0.0	2,110	0.1	0.1
2,024	0.1	0.8	2,118	0.1	0.0
2,020	0.0	0.5	2,120	0.2	1.4
2,028	0.0	0.5	2,122	0.2	0.0
2,030	0.1	0.0	2,124	0.0	0.5
2,032	0.4	0.7	2,120	0.0	0.0
2,034	0.1	0.4	2,128	0.2	0.0
2,030	0.1	0.2	2,120	0.5	0.8
2,030	0.1	0.0	2,132	0.1	0.0
2,040	0.0	0.0	2,134	0.2	0.1
2,042	0.1	0.0	2,130	0.1	0.0
2,044 7 049	0.0	0.0	2,138	0.2	0.0
2,040	0.0	0.0	2,140	0.2	0.0
2,030	0.0	0.0	2,142	0.1	0.0
2,032	0.0	0.0	2,144	0.2	0.4
2,054	0.0	0.0	2,146	0.0	20.6

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
- (cm)	Matter(%)	Content(%)	(cm)	Matter(%)	Content(%)
(0)		Content(70)	(cm)		country (70)
2,056	0.0	0.0	2,148	0.2	0.5
2,150	0.2	0.0	2,242	0.2	0.0
2,152	0.2	0.4	2,244	0.1	8.6
2,154	0.2	0.2	2,246	0.0	0.0
2,156	0.0	0.7	2,248	0.1	0.0
2,158	0.2	0.2	2,250	0.1	0.3
2,160	1.1	0.0	2,252	0.0	0.0
2,162	0.2	1.1	2,254	0.5	0.0
2,164	0.9	0.4	2,256	0.0	0.0
2,166	0.4	0.4	2,258	0.1	0.0
2,168	0.7	0.0	2,260	0.4	0.0
2,170	0.2	0.3	2,262	0.3	0.0
2,172	0.5	0.3	2,264	0.5	0.0
2,174	0.4	0.9	2,266	0.3	0.6
2,176	0.3	0.1	2,268	0.1	0.4
2,178	0.3	0.0	2,270	1.1	0.1
2,180	0.2	0.5	2,272	0.3	0.0
2,182	0.4	0.9	2,274	0.3	0.2
2,184	0.6	0.0	2,276	0.6	0.0
2,186	0.6	0.5	2,278	2.7	0.4
2,188	0.4	0.2	2,280	1.2	0.0
2,190	1.4	0.1	2,282	0.2	0.0
2,192	0.3	0.1	2,284	0.3	0.3
2,194	0.5	0.1	2,286	0.2	0.1
2,196	0.0	0.0	2,288	0.6	0.0
2,198	0.1	0.4	2,290	0.0	0.2
2,200	0.0	0.0	2,292	0.2	0.0
2,202	0.0	0.0	2,294	0.0	0.2
2,204	0.0	0.0			
2,206	0.0	0.0			
2,208	0.0	0.2			
2,210	0.0	0.0			
2,212	0.0	0.0			
2,214	0.0	0.0			
2,216	0.0	0.0			
2,218	0.2	0.0			
2,220	0.0	0.0			
2,222	0.0	0.0			
2,224	1.1	0.1			
2,226	0.6	0.2			
2,228	0.0	0.3			
2,230	0.0	0.0			
2,232	0.0	0.0			
2,234	0.0	0.0			
2,236	0.1	0.0			
2,238	0.2	0.2			
2,240	0.0	0.1			

Organic Matter and Carbonate Content of North Core at North Ingenbright Lake

Depth (cm)	Organic Matter(%)	Carbonate Content(%)	Depth (cm)	Organic Matter(%)	Carbonate
			(em)		· · · · · · · · · · · · · · · · · · ·
16	2.9	11.8	108	0.3	0.5
18	7.6	9.4	110	0.0	0.4
20	11.3	5.5	112	0.3	0.4
22	4.8	14.0	114	0.1	0.5
24	7.6	6.2	116	0.1	0.3
26	8.7	5.3	118	0.4	0.3
28	11.7	3.7	120	0.3	0.7
30	14.6	7.6	122	0.4	2.1
32	12.3	5.1	124	0.4	0.2
34	11.5	6.1	126	0.7	0.2
36	3.1	12.6	128	0.5	0.5
38	11.6	8.2	130	0.4	0.3
40	12.3	7.0	132	0.3	0.3
42	7.4	7.9	134	0.4	0.1
44	10.5	5.4	136	0.0	0.2
46	8.5	5.3	138	0.3	1.0
48	11.3	3.5	140	0.3	0.6
50	0.5	1.2	142	0.2	0.8
52	1.7	1.0	144	0.7	0.8
54	0.3	5.1	146	0.7	0.2
56	I.I	0.1	148	0.2	0.6
58	0.5	0.5	150	0.3	0.1
60	0.1	2.3	152	0.0	0.2
62	0.0	0.0	154	0.2	0.2
64	0.0	0.0	156	0.0	0.8
66	0.0	0.0	158	0.4	0.4
68	- 0.1	0.0	160	0.0	0.0
70	0.1	0.5	162	0.0	0.0
72	0.0	0.3	164	0.0	0.0
74	0.0	0.3	166	0.0	0.0
76	0.0	0.2	168	0.1	0.1
78	0.0	0.0	170	0.0	0.0
80	0.0	0.1	172	0.0	0.0
82	3.6	8.2	174	0.4	0.0
84	0.0	0.0	176	0.0	0.0
86	0.3	0.0	178	0.1	0.3
88	0.0	0.2	180	0.0	0.0
90	0.0	0.3	182	0.0	0.2
92	0.2	0.4	184	0.8	0.8
94	0.2	0.3	186	0.0	0.0
96	0.1	0.0	188	0.2	0.9
98	0.2	8.1	190	0.0	0.0
100	0.4	0.0	192	0.0	0.0
102	0.2	0.7	194	0.3	0.0
104	0.4	0.0	196	0.2	0.0
106	1.0	0.0	198	0.0	1.0
200	0.3	0.0	292	2.9	0.9
202	0.2	0.1	294	3.2	1.6
204	0.5	0.9	296	2.5	0.9
206	0.7	0.6	298	2.0	0.0
208	0.0	3.0	300	1.7	1.5

Organic Matter and Carbonate	Content of South Core	at North Ingenbright Lake

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Depth (cm)	Organic Matter(%)	Carbonate Content(%)	Depth (cm)	Organic Matter(%)	Carbonate Content(%)
210	0.6	0.0	307	19	0.4
212	14	0.0	304	1.5	1.6
214	16	0.0	306	3.0	07
214	0.9	57	308	2 3	13
210	0.9	<u> </u>	310	0.8	1.5
210	0.0	0.0	310	0.8	0.6
220	10	0.1	314	1.0	0.5
274	1.0	0.0	316	0.7	0.5
224	0.6	0.1	319	14	1.0
220	1.2	0.2	320	1.4	1.0
220	0.7	0.5	320	0.4	21
230	13	0.5	324	0.4	0.0
734	1.9	0.0	324	0.8	0.3
234	1.0	0.0	320	2.0	19
230	0.5	0.0	328	1.9	2.0
230	0.2	0.2	330	0.9	0.7
240	0.7	0.0	332	1.2	0.7
242	1.1	0.0	334	1.1	0.8
244	0.9	0.0	330	0.0	1.2
240	1.4	0.0	338	1.2	0.7
248	1.0	10.8	340	0.5	0.7
250	1.8	0.0	348	1.0	0.8
252	1.3	0.0	350	0.0	0.0
254	1.2	1.2	352	0.1	0.8
256	2.8	0.0	354	1.4	0.4
258	0.9	0.0	320	0.0	0.6
260	0.0	1.2	358	0.2	0.8
262	1.4	2.4	360	2.0	0.6
264	0.0	1.5	362	0.9	0.4
266	0.1	1.5	364	1.1	0.7
208	0.5	1.0	300	1.4	0.4
270	0.9	2.0	368	1.8	0.4
272	1.3	0.8	370	2.3	0.1
274	1.4	2.7	372	0.8	0.3
276	1.5	0.6	374	0.7	0.5
278	2.0	1.1 1.7	376	1.0	0.2
280	2.0	1.0	378	1.7	0.1
282	3.2	1.7	380	1.4	0.8
284	2.9	0.5	382	1.5	0.0
286	2.0	1.1 1.5	384	2.3	0.0
288	1.9	1.5	386	1.1	0.6
290	2.3	2.1	388	1.4	0.4
390	2.0	1.1	566	3.7	2.3
392	2.3	3.3	568	2.7	1.6
394	3.2	1.3	570	3.8	0.0
396	3.4	<i>د</i> .ا	572	3.0	3.4
398	1.7	0.6	574	3.2	3.8
400	2.7	0.0	576	2.9	0.3
402	3.2	0.6	578	3.5	2.6
404	3.3	0.4	580	3.6	1.4
406	1.7	0.8	582	3.5	0.3
408	4.1	1.9	584	4.0	0.1

(Organic Matter a	nd Carbonate	Content of S	South Core	e at North	Ingenbright Lake
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Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	. Content(%)
410	2.2	1.4	586	41	0.0
412	2.6	0.2	588	3.9	0.5
414	1.7	1.0	590	3.3	0.5
416	0.7	1.0	592	0.3	2.5
500	2.5	0.9	594	21	1.0
502	1.0	1.4	596	1.7	2.6
504	2.7	0.9	598	24	2.4
506	3.3	01	600	3.4	17
508	3.4	01	602	3.8	0.8
510	36	14	604	4.0	0.8
512	10.8	5.4	606	37	5.0
514	68	0.0	608	3.0	1.1
516	8.6	11	610	29	0.8
518	73	1.1	612	13	1.4
520	5.2	0.7	614	33	1.4
522	10.7	0.0	616	29	5.4
524	40	0.0	618	0.9	0.8
526	63	13	620	2.6	4.0
528	51	1.1	620	2.0	7.0
530	23	7.9	674	1.9	2.5
537	49	0.4	626	5.2	J.J 4 1
534	4.9	0.4	678	J.2 2 <i>1</i>	4.1
536	53	0.7	630	2.4	
538	50	0.8	630	2.0	13
540	66	0.9 T A	634	2.0	1.5
542	4.4	0.4	676	2.0	1.9
546	35	0.4	630	1.2	1.7
548	62	4.1	640	1.2	1.8
550	62	0.7	647	1.5	1.5
550	4.8	0.2	644	1.2	1.0
554	43	1.5	646	0.0	0.7
556	50	0.0	648	0.5	0.2
558	5.0	2.7	040	1.0	0.4
560	20	5.7	650	5.7	1.4
567	2.5	1.0	654	0.8	0.7
564	2.5	1.9	656	0.8	0.0
658	0.3	0.2	020	1.0	1.0
660	0.5	0.2	00 4 996	23.5	1.0
667	0.0 7 A	0.4	000	20.0	3.0
664	1.8	0.5	800	10.1	4.5
666	1.0	0.2	890	17.0	2-1
668	3.5	0.4	804	10.2	4.0
670	2.0	1.1	074 204	19.2	3.7
670	51	0.0	070 202	44-1 73 0	1.3
674	2.1	0.9	070	25.0	1.5
676	2.7	0.9 A t	300	20.0	1.0
679	5.8	17	910	20.1	5.1
610 620	3.0	0.4			
682	3.0	47			
602	3.7	4.1 0.7			
686	27	1.7			
000	I	4 - 4			

Organic Matter and Carbonate Content of South Core at North Ingenbright Lake

Depth	Organic	Carbonate	Depth	Organic	Carbonate
(cm)	Matter(%)	Content(%)	(cm)	Matter(%)	. Content(%)
824	2.2	0.5		<u> </u>	
826	0.9	1.0			
828	0.4	2.1			
830	1.6	1.2			
832	0.8	1.4			
834	0.1	4.6			
836	1.8	1.0			
838	1.8	1.3			
840	2.3	1.4			
842	0.8	5.7			
844	3.3	4.0			
846	6.1	1.5			
848	1.2	1.2			
850	1.9	2.0			
852	1.3	0.9			
854	5.3	1.2			
856	1.3	2.2			
858	1.1	1.3			
860	1.1	0.9			
862	0.6	1.0			
864	1.6	2.7			
866	3.2	1.7			
868	5.2	2.2			
870	17.8	4.8			
872	18.5	4.1			
874	19.2	2.6			
876	16.7	3.7			
878	16.2	2.2			
880	17.8	2.8			
882	14.5	4.0			

Organic Matter and Carbonate Content of South Core at North Ingenbright Lake

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Appendix B Bulk Mineralogical Data from XRD

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Depth	Quartz	Х	Plag.	Total	Calcite	Hieh Me	Aragonite	Proto	Dolomite
(cm)		Feldspar)	Clay Miner.		Calcite	0	Dolomite	
12	23.5	10.5	5.3	0.0	0.0	0.0	0.0	40.0	00
14	73.3	16.0	0.0	0.0	0.0	0.0	0.0	10.7	00
16	30.2	0.0	6.6	0.0	0.0	4.0	0.0	0.0	39.0
18	10.1	4.3	0'0	0.0	0.0	0.0	0.0	16.0	0.0
20	12.3	0.0	0.0	53.0	0.0	1.6	0.0	25.4	0.0
22	27.4	0:0	4.8	6.6	0.0	0.0	0.0	33.1	0.0
24	26.6	0.01	4.5	0.0	0.0	0.0	0.0	42.1	0.0
26	9.3	3.3	2.1	27.4	0.0	1.0	0.0	0.0	14.8
28	11.9	3.7	2.0	26.5	0.0	0.0	0.0	17.1	00
90	22.4	9.9	4.4	0.0	0.0	0.0	0.0	51.1	0.0
32	28.5	0.0	4.5	0.0	0.0	0.0	0.0	0.0	41.8
34	25.4	12.6	0.0	0.0	0.0	8.3	0.0	37.5	0.0
36	18.6	5,9	0.0	39.3	0.0	0.0	0.0	18.5	0.0
38	26.6	6.9	0.0	0.0	0.0	0.0	0.0	31.9	0.0
40	14.9	4.5	3.1	0'0	0.0	0.0	0.0	23.3	0.0
42	5.1	2.1	0.9	14.1	0.0	0.0	0.0	6.6	0.0
44	24.1	0,0	4.1	0,0	0.0	0.0	0.0	39.5	0.0
46	19.0	0.0	0.0	0.0	0.0	1.3	0.0	14.6	0.0
48	21.8	0'0	5.0	0.0	0.0	0.0	0.0	30.2	0.0
20	6.9	0,0	0.0	0.0	0'0	0.0	0.0	8.5	0.0
52	4.7	0.0	0.0	17.7	0.0	0.0	0.0	9.8	0.0
54	3.7	0.0	0.0	0.0	0.0	0.0	0.0	3,2	0.0
20 20	4.7	0.0	0.0	0.0	0.0	0.0	0.0	3.9	0.0
28	3.2	0.0	0.0	0.0	0.0	0.0	0.0	4.9	0.0
09	5.7	0.0	0.0	0.0	0.0	0.0	0.0	4.7	0.0
62	3.9	0.0	0.0	0.0	0.0	0.0	0.0	5.6	0.0
ξ (3.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
3 (4.9	0.0	0.0	0.0	0.0	0.0	0.0	5.0	0.0
89 8	3.9	0.0	0.0	0.0	0.0	0.0	0.0	5.6	0.0
2 2	4.6	0.0	0.0	0.0	0.0	0.0	0.0	3.7	0.0
21	2.6	0.0	0.0	0.0	0'0	0.0	0.0	3.4	0.0
4	L.C.	0.0	0.0	0.0	0.0	0'0	0.0	3.4	0.0
16	3.0	0.0	0.0	0.0	0.0	0'0	0.0	4.0	0.0
78	2.8	0.0	0.0	0.0	0.0	0.0	0.0	3,5	0.0
08	4.2	0.0	0.0	0.0	0.0	0.9	0.0	3.6	0.0
82	1.4	0'0	0.0	0.0	0.0	0.0	0.0	3.4	0.0
84	3.9	0.0	0.0	0.0	0.0	0.0	0.0	3.2	0.0
80	1.9	0.0	0.0	0.0	0.0	0.0	0.0	2.5	0.0
88	5.7	0.0	1.5	0.0	0.0	0.0	0.0	0.0	0.0

Mineralogical Composition of North Core at North Ingebright Lake

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Denth	Onart <i>s</i>	K	Dlag	Tatal	0.1-14				
(cm)		Feldspar		Clay Miner.	Calche	rugn mg Calcite	Aragonite	Proto Dolomite	Dolomite
96	2.9	0.0	00	0 U	00	00			
92	1.2	00		0.0	0.0	0.0	0.0	2.1	3.6
94	2.1	0.0	0.0	0.0		0.0	0.0	0.0	1.9
96	5.5	00	00	0.0	0.0	0.0	0.0	0,0	3.6
98	2.2	0.0	0.0	0.0		0.0	0.0	3.2	0.0
100	2.1	0.0		0.0	0.0	0.U	0.0	2.7	0.0
901	4.5	0.0	0.0	0.0	0.0	0.0	0.0	2.5	0.0
108	80	0.0	91	0.0	0.0	0.U 0	0'0	0.0	2.9
011	0 C		0.1	0.0	0.0	0.0	0.0	0.0	0.0
	9.7 2.0	0.0	1.9	0.0	0.0	0.0	0.0	0.0	0.0
711	C.U	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
411	c.v	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.4
011	4.1	0.0	0.0	0.0	0.0	0'0	0.0	1.9	0.0
811	c.0	0.0	0.0	0.0	0.0	0,0	0'0	0.0	0.0
120	0.6	0.0	0.0	0.0	0.0	0.0	0,0	0.0	0.0
122	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
124	0.6	0.0	0.11	0.0	0.0	0.0	0.0	0.0	88.4
126	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
128	2.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
130	0.9	0.0	0.0	0.0	0.0	0.0	0.0		0.0
132	1.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
134	2.2	0.0	0.0	0.0	0.0	0.0	0'0	2.4	0.0
136	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
138	1.3	0.0	0.0	0.0	0.0	0.0	0.0	2.7	0.0
140	1.6	0.0	0.0	0.0	0.0	0.0	0,0	3.6	0.0
142	2.1	0.0	0.0	0.0	0.0	0.0	0.0	3.3	0.0
144	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
146	12.3	3.4	3.4	0'0	0.0	0.0	0.0	8.6	0.0
148	4.5	4,1	2.0	0.0	0.0	0.0	0.0	8.9	0.0
150	7.3	0.0	2.3	0.0	0.0	0.0	0.0	12.5	0.0
152	5.9	0.0	0.0	19.2	0.0	1.3	0.0	8.8	0.0
154	1.1	0.0	0.0	0.0	0.0	0.0	0.0	2.8	0.0
156	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
158	0.9	0,0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
091	1.4	3.0	0,0	0.0	0.0	0.0	0.0	0.0	0.0
162	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
40 1	4.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
166	1.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
168	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1 /0	1.9	ij	0.0	0.0	0.0	0.0	0.0	0.9	0.0

Denth	Ouartz	X	Plag	Total	Calaita	Utet Me			
(cm)		Feldspar	0	Clay Miner.		Calcite	Alagomic	Dolomite	Dolomite
172	1.9	0.0	0.0	0.0	0.0	00	00	~~~	
174	0.0	0.0	0.0	0.0	0.0	00	0.0	0,0	0.0
176	0.5	0.0	0.0	0.0	0.0	1.2	00	0.0	0.0
178	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
180	1.7	0.0	0.0	0.0	0.0	0.0	0.0	 - -	0.0
182	2.7	0.0	1.3	0.0	0.0	0.0	0.0	<u>:</u> :	4.1 0 0
184	1.0	0.0	0.0	0.0	0.0	0.0	0.0	7 O	0.0
188	1.7	0.0	0.0	0.0	0.0	0.0	0.0	2.1	0.0
190	1.7	0.0	0.0	0.0	0.0	0.0	0.0	.00	0.0
192	3.1	0.0	2.3	24.8	0.0	0.0	0.0	61	0,0
194	1.6	0:0	0.0	0.0	0.0	0.0	0.0	5	7 0
196	3.5	2.7	L .1	0.0	0.0	0.0	0.0	0.0	0.0
198	3.0	2.8	0.0	0'0	0.0	0.0	0.0	2.5	0.0
200	2.4	0.0	0.0	18.8	0.0	0.0	0.0	0.0	0.0
202	2.8	2.7	0.0	0.0	0.0	0.0	0.0	2.0	00
204	2.8	2.7	0.0	0.0	0.0	0.0	0.0	2.0	00
206	2.9	0.0	0.0	16.8	0.0	0.0	0.0	1.5	1.7
208	5.1	0.0	1.7	23.5	0.0	0.0	0.0	2.8	00
210	3.8	0.0	0.0	0.0	0.0	2.3	0.0	0.0	8.1
212	2.7	0.0	0.0	0.0	0.0	0.0	0.0	2.0	00
214	2.9	0.0	0.0	24.9	0.0	0.0	0.0	2.0	00
216	5.1	0.0	0.0	0.0	0.0	0.0	0.0	2.1	00
218	3.4	0.0	0.0	29.9	0.0	0.0	0.0	2.2	00
220	2.8	0.0	0.0	0.0	0.0	0.0	0.0	61	8.0
222	3.7	0.0	1.3	21.4	0.0	0'0	0.0	1.8	2.5
224	5.6	0.0	4.5	21.8	0.0	0.0	0.0	2.6	2.6
226	6.5	0.0	0.0	0.0	0.0	0.0	0.0	3.7	0.0
877	4.9	3.3	1.6	0.0	0.0	0.0	0.0	2.2	4.3
250	4.5 0	0.0	0.0	0.0	0.0	0.0	8.0	0.6	0.0
727	2.2	0.0	1.1	18.0	0.0	0.0	0.0	1.7	1.8
234	4,0	0.0	0.0	19.9	0.0	0.0	0.0	2.8	0.0
230	4.5	0.0	0.0	23.2	0.0	0.0	0.0	0.0	2.5
2.58	5,1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.1
240	J.4	2.1	0.0	14.0	0.0	0.0	0.0	2.1	2.2
242	0.6	0.0	1.6	0.0	2.0	1.5	0.0	2.1	3.8
244	4.3	0.0	0.0	0.0	0.0	1.6	0.0	0.0	2.6
240	0.7	1.6	0.0	19.2	0.0	0.0	0.0	0.0	2.1
248	4.7	0.0	0.0	24,8	0.0	2.3	0.0	2.8	2.6
007	4.0	0.0	1.0	0.0	0.0	1.9	0.0	2.2	2.5

Depth (cm)	Quartz	K Feldspar	Plag.	Total Clay Miner.	Calcite	High Mg Calcite	Aragonite	Proto Dolomite	Dolomite
252	4.6	0.0	10	0.0	0.0	19	0.0	22	25
252	63	0.0	1.0	30.8	0.0	1.8	0.0	17	2.9
256	37	0.0	0.0	16.7	0.0	0.8	0.0	28	4.8
250	J.1 A 5	0.0	2.0	20.6	0.0	13	0.0	2.0	7,0 7 A
250	4.5	0.0	2,0	20.0	0.0	1.5	0.0	2,3	2,4
200	4.3	0.0	0.0	24.8	0.0	1.3	0.0	2.7	2.2
202	4.3	2.3	0.0	32.3	0,0	1,1	0.0	2.2	2,2
204	0.1	0.0	0.0	28.1	0.0	1.1	0.0	3.3	3,2
266	3.9	0.0	0.0	23.4	0.0	0.0	0.0	2.2	1.9
268	5.5	3.5	0.0	24.7	0.0	0.0	0.0	1.9	2.6
270	6.2	2.6	1.1	31.7	0,0	0.0	0.0	2.7	2,5
272	3.9	2.5	0,0	18.0	0.0	1.5	0.0	2,2	2.9
274	7.3	0.0	1.5	0.0	0.0	2.2	0.0	3.0	3.0
276	4.1	2.9	9,9	27.0	0.0	0.0	0,0	2.1	2.6
278	10.2	0.0	0.0	24.1	0.0	1.7	0.0	0.0	2.7
280	4.4	0.0	1.2	3.2	0.0	2.0	0.0	2.3	2.6
282	6.4	0.0	1.6	30.6	0.0	2.4	0.0	2.8	2.9
284	2.6	1.9	0.9	18.5	0.0	1.1	0.0	1.4	3.0
286	5.4	0.0	0.0	0.0	0.0	2.2	0.0	0.0	0.0
288	69.7	0.0	0.0	0.0	0.0	30.3	0.0	0.0	0.0
290	2.3	0.0	1.1	0.0	0.0	0.0	0.0	0.0	1.8
292	1.8	0.0	0.0	0.0	0.0	1.5	0.0	1.7	1.7
294	2.0	0.0	1.6	0.0	0.0	0.0	0.0	0.0	2,9
296	2.1	0.0	0.0	0.0	0.0	2.0	0.0	6.3	0,0
298	1.6	0.0	• 0.0	0.0	0.0	1.2	0.0	0,0	0.0
300	2.2	0.0	0,0	25.6	0.0	1.5	0.0	1,6	2.9
302	3.1	0.0	0.0	0.0	0.0	1.9	0.0	1.7	0,0
304	2,3	0.0	0.0	0.0	0.0	0.0	0.0	1.6	0.0
306	1.4	0.0	0.0	0.0	0.0	0.9	0.0	0.0	0.0
308	1.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
310	1.3	0.0	0.0	22.0	0.0	1.0	0.0	0.0	0.0
312	0.9	0.0	0.0	0.0	0.0	0.0	0.0	2.0	.0.0
314	1.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
316	0.7	0.0	0.0	9.6	0.0	0.0	0.0	0.0	0.0
318	1.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	13
320	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
220	1.9	0.0	0.0	10.0	0.0	0.0	0.0	1 4	1 1
346	1.0	0.0 A 6	0.0	12.2	0.0	0.0	0,0	0.0	1.1
324	1,9	4.U A A	0.0	0.0	0.0	0.0	0.0	0.0	0.0
320	0.0	0.0	4.1	0.0	0.0	0,0	0.0	0.0	0.0
528	υ,δ	0.0	0.0	0.0	0.0	0,0	0,0	1.7	0.0

Mineralogical Composition of North Core at North Ingebright Lake

Depth	Quartz	X	Plag.	Total	Calcite	High Mg	Aragonite	Proto	Dolomite
(cm)		Feldspar		Clay Miner.		Calcite		Dolomite	
330	1.1	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
332	0.7	0.0	0.0	0.0	0.0	0.0	0.0	1.3	0.0
334	0.5	0.0	0.0	0'0	0.0	0.0	0.0	0.8	0,0
336	1.0	0.0	0.0	0.0	0.0	0.0	0.0	1.2	0.0
338	0.9	0.0	0.0	0.0	0.0	0.0	0.0	1.1	0.0
340	1.S	0.0	0.0	0.0	0.0	0.0	0.0	1.8	0.0
342	0.9	0.0	0.0	14.9	0.0	0.0	0.0	0.0	1,4
344	0.6	0.0	11.2	0.0	0.0	0.0	0.0	0.0	0.0
346	0.8	0.0	0.0	0.0	0.0	0.0	0.0	1.2	0.0
348	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0.0
350	0.6	0.0	1.4	0.0	0.0	0.0	0.0	0.0	0.0
352	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
354	1.4	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
356	1.0	0.0	0.0	0.0	0.0	0.0	0.0	1.9	0.0
358	1.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
360	2.3	0.0	15.2	0.0	0.0	0.0	0.0	0.0	1,8
362	1.4	0.0	0.0	0.0	0.0	0.0	0'0	1.8	0.0
364	11.3	0.0	1.8	0.0	0.0	0.0	0.0	1.0	0.0
368	1.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.1
370	0.8	0.0	0.0	0.0	0.0	0.0	0.0	1.5	0.0
372	1.0	0.0	0.0	0.0	0.0	0.0	0'1	1.3	0.0
374	0.6	0.0	0.0	19,8	0.0	0.0	0.0	0.0	0.0
376	1.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
378	4.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.3
380	0.8	0.0	9.4	0.0	0.0	0'0	0.0	1.2	0.0
382	1.2	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
384	1.4	0.0	0.0	0.0	0.0	0.0	0.0	1.6	0.0
386	0.9	0'0	0.0	0.0	0.0	0.0	0'0	0'0	0.0
388	0,6	0.0	0.0	0:0	0'0	0.0	0.0	0.0	0.0
390	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
392	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
394	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
396	1.0	0.0	10.8	0.0	0.0	0.0	0.0	0.0	0.0
398	0.6	0.0	9.0	0.0	0.0	0.0	0.0	0.0	0.0
400	1.2	0.0	8.2	0.0	0.0	0.0	0.0	0.0	0.0
402	1.5	0.0	0.0	0.0	0.0	0.0	0.0	1.4	0.0
404	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
406	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
408	0.7	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0

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Ingebright Lake
Core at North
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cal Composit
Mineralogi

Depth	Ouartz	Х	Plac.	Total	Calaita	Uich Me	A		
(cm)	,	Feldenar	-	Clay Minor	CAILIE	Star ingirt	Alagonic	r7010	Dolomite
		wiens.		Ciay Miller.		Calcite		Dolomite	
410	0.6	0.0	0.0	0.0	0.0	0.0	00	00	¢
412	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0.0
414	0'0	0.0	15.2	0.0	0.0	00	0.0	0.0	0.0
416	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
418	1.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
420	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
122	0.4	0.0	1.2	0.0	0.0	0.0	00	0.0	0.0
124	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
426	0.4	0.0	0.0	0.0	0.0	00			0.0
428	5.2	0.0	10.4	0.0	0.0	0.0	0.0	0.0	0.0
430	0.6	0.0	8.0	0.0	0.0	0.0		0.0	0.0
432	0.6	0.0	1.11	0.0	0.0	0.0	0.0	0.0	0.0
434	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
436	0.7	0.0	9.9	0.0	0.0	0.0	0.0	00	0.0
438	0.3	0.0	6.8	0.0	0.0	0.0	0.0	0.0	0.0
440	0.6	0.0	9.5	0.0	0.0	0.0	0.0	0.0	0.0
442	2.4	0.0	15.1	0.0	0.0	00	0.0	0.0	0.1
444	0.5	0.0	12.0	0.0	0.0	0.0	0.0	0'0	0.0
446	0.4	0.0	9.6	0.0	0.0	00	0.0	0.0	0.0
448	1.9	0.0	9.3	0.0	0.0	0.0	0.0	0.0	0.0
450	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
452	0.5	0.0	10.5	0.0	0.0	0.0	0.0	0.0	0.0
454	0.3	0.0	9.4	0.0	0.0	0.0	0.0	0.0	0.0
456	0.6	0.0	14.0	0.0	0.0	0.0	0.0	0.0	0.0
458	0.5	0.0	8.6	0.0	0.0	0.0	0.0	0.0	0.0
460	2.3	0.0	1:1	0.0	0.0	0.0	0.0	2.2	23.0
462	2.5	0.0	0'0	5.0	0.0	1.7	0.0	0.0	0.0
404	3.6	0.0	0.0	0.0	0.0	0.0	0.0	1.3	0.0
466	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	8.5
468	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
470	C.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0 .
472	1.9	2.4	0.0	0.0	0.0	0.0	0.0	6'1	1.9
4/4	9.1	0.0	0.0	20.9	0.0	0.0	0.0	1.6	61
476	1.2	0.0	3.8	13.3	0.0	0.0	0.0	0.1	2.2
4/8	5.5	0.0	1.0	18.8	0.0	0.0	0.0	1.9	0.0
480		0.0	1.1	17.4	0.0	0'0	0.0	0.0	1.4
482	9.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
484	0.2	0.0	0.0	0.0	0.0	0.0	0,0	2.9	0,0
480	0.1	0.0	0.0	0.0	0.0	0.0	0.0	2.6	0.0

Denth	Onorta	4	2						
	Zuar 12	4	riag.	Total	Calcite	High Mg	Aragonite	Proto	Dalamite
(cm)		Feldspar		Clay Miner.		Calcite	D	Dolomite	
488	2.2	0.0	0.0	0.0	00	00	~~~		• • • • • • • • • • • • • • • • • • • •
490	2,4	2.7	0.0	0.0	0.0		0.0	1.5	0.0
492	1.7	0.0	0.0	0.0	0.0		0.0	0.0	0.0
494	2.0	0.0	0.0	0.0	0.0		0.0	0.0	0.0
496	1.4	0.0	0.0	0.0	0.0		0'0	4.1	1.4
498	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
500	2.2	0.0	1.9	0.0	0.0	0.0	0.0	0,0	0.0
502	1.5	0.0	00		0.0	1.1	0.0	0.0	0.0
504	1.4	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
506	4.1	0.0	0.0 1 C	0.0	0.0	0.0	0.0	0.0	0.0
508	2.3	0.0	0.0	0.0	0.0	0.0	0.0	2.1	0.0
510	1.0	00	0.0	0.0	0.0	0.0	0.0	1.6	0.0
512	61	0.0	0.0	0.0	0.0	0.0	0'0	2.3	0.0
614	01	0.0	0.0	9.1c	0.0	0.0	0.0	2.5	2.6
215	0,0	0.0	0.0	0.0	0.0	0.0	0.0	3.9	00
815	<u>, , , , , , , , , , , , , , , , , , , </u>	0.0	0.0	31.9	0.0	0.0	0.0	2.5	2.6
0005	2.1 2.5	0.0	2.4	34.0	0.0	0.0	0.0	2.4	0.0
522	n e F -	0.0	0.0	22.6	0.0	0.0	0.0	2.6	00
776		0.0	0.0	0.0	0.0	0.0	0.0	2.5	0.0
805		0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
527		C.1	0.0	0.0	0.0	0.0	0.0	0.0	00
234	2.C	0.0	0.0 0.0	0.0	0.0	0.0	0.0	0.0	0.0
985	2.1 7	C.2	0.0	0.0	0.0	0.0	0.0	0.0	2.4
825		0'1	0.0	26.0	0.0	0.0	0.0	2.3	0.0
000		C.0	0.0	0.0	0.0	0.0	1.4	0.0	0.0
242		0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
544		0.0	0.0	0.0	0.0	0.0	0.0	2.6	0.0
546	0.0	0.0	0.0	C.42	0.0	0.0	1.3	2.1	0.0
548	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
550	1.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
552		0.0	0.0	C.12	0.0	0.0	2.1	2.4	0.0
554	01	0.0	0.0	0.0	0.0	0.0	0.0	1.8	. 2.2
556	1.7	0.0	0.0	8.U2	0.0	0.0	0.0	0.0	0.0
558	2.9	0.0		د./د مور	0.0	0.0	0.0	2.6	0.0
560	1.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
564	0.0	0.0	0.0 7	0.0	0.0	0.0	0.0	4.4	0.0
566	80	0.0	+.12	22.5	0.0	0.0	0.0	1.7	0.0
568	0.0	0.0	0.10	14.0	0.0	0.0	0.0	0.8	0.0
570	1.0	0.0	0.72	0.0	0.0	0.0	0.0	1.4	0.0
	!	2.2	מימר	14./	0.0	0.0	0.0	0.0	0.0

Depth	Quartz	×	Plag.	Total	Calcite	Hieh Me	Aragonite	Proto	Dalamite
(cm)		Feldspar)	Clay Miner.		Calcite	D	Dolomite	
572	0.6	0.0	30.0	14.9	0.0	0.0	0.0	1.5	0.0
574	1.0	0'0	27.5	1.91	0.0	0.0	0.0	8.1	0.0
576	0.8	0.0	35.7	0.0	0.0	0.0	0.0	0.0	0.0
578	0.6	0.0	33.9	12.3	0.0	0.0	0.0	0.8	0.0
580	1.2	1.7	32.0	0.0	0.0	0.0	0.0	0.1	0.0
582	0.6	0.0	36.8	0.0	0.0	0,0	0.0	1.2	0.0
584	1.7	0.0	35.9	0.0	0.0	0.0	0.0	1.5	0.0
586	0.6	0.0	45.3	0.0	0.0	0.0	0.0	0.0	0.0
588	0.5	0.0	46.4	0.0	0.0	0.0	0.0	1.2	0.0
590	0.9	0.0	42.4	0.0	0.0	0.0	0.0	0.0	0.0
592	0.6	0.0	30.2	0.0	0.0	0.0	0.0	0.0	0'0
594	0.6	37.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0
596	2.6	0.0	40.1	0.0	0.0	0.0	0.0	0.0	0.0
598	2.4	2.6	17.2	27.6	0.0	0.0	0.0	3.1	2.1
600	1.0	0.0	22.1	0.0	0.0	0.0	0.0	2.2	0.0
602		2.1	20.7	24.7	0.0	0.0	0.0	2.0	1.8
604	1.5	0.0	15.4	41.2	0.0	0.0	0.0	3.0	2.2
606	1.3	2.1	33.8	0.0	0.0	0.0	0.0	1.2	1.4
608	1.1	2.0	29.0	0.0	0.0	0.0	0.0	2.1	1.7
616	0.8	0.0	34.8	0.0	0.0	0.0	0.0	0,0	0'0
618	0.9	1.6	36.1	0.0	0.0	0.0	0.0	0.0	0.0
620	0.7	0.0	33.2	0.0	0.0	0.0	0.0	0.9	0.0
622	1.0	0.0	35.9	0.0	0.0	0.0	0.0	0.0	0.0
624	0.5	0.0	28.5	14.0	0,0	0.0	0.0	1.1	0.0
626	1.3	0.0	28.8	16,9	0.0	0.0	0.0	0.0	0.0
628	1.0	0.0	26.2	0.0	0.0	0.0	0.0	2.1	7.0
630	1.1	0.0	30.9	0.0	0.0	0.0	0.0	2.1	2.2
634	2.6	0.0	30.5	0.0	0.0	0.0	0.0	4.2	0.0
636	0.5	0.0	84.8	0.0	0.0	0.0	0.0	0,6	0.0
638	2.6	0.0	34.2	0.0	0.0	0.0	0.0	1.0	0'0
640	0.7	0.0	40.1	0.0	0.0	0.0	0.0	0.0	· 1.5
642	0.8	0.0	28.6	18.0	0.0	0.0	0.0	1.3	1.5
644	0.9	0.0	42.2	0.0	0.0	0.0	0.0	0.0	0.0
646	1.4	0'0	28.8	0.0	0.0	0.0	0.0	1.8	0.0
648	1.2	0.0	37.7	0.0	0.0	0:0	0.0	1.6	0.0
650	0.4	0,0	41.4	0.0	0.0	0.0	0.0	0.0	0.0
652	0.9	0,0	25.0	15,2	0.0	0.0	0.0	1.5	1.0
654	1.3	0,0	34.3	0.0	0.0	0.0	0.0	1.2	0.0
656	0.5	0.0	37.0	0.0	0.0	0.0	0.0	0.0	0.0

Dolomite	Proto	Stagonite	SM AgiH	Calcite	letoT	Plag.	Ж	Quartz	Depth
	Dolomite		etiole. Calcite	••••••	Clay Miner.	• * * * • • • • • • • • • • • •	Feldspar		(ເພວ)
0.0	11	0.0	0.0	0'0	0.0	0.55	0.0	8.0	859
0.0	1.2	0'0	0.0	0'0	0.0	0'0	0.0	5.2	099
6'0	1.1	0.0	0.0	9'8	0.0	24.2	0.0	0.1	799
0.0	0.0	0.0	0.0	0.0	0.0	\$'95	0.0	9.0	6 99
0,0	0.0	0,0	0.0	0.0	0.0	5.45	0,0	2'I	000
0,0	0'0	0.0	0.0	0.0	0.0	4.66	0.0	8.0	800
0.0	Z'1	0.0	0.0	0'0	0.0	0.00	0.0	0.0	0/0
0.0	6'0	0.0	0.0	0,0	0.0	8.82	0.0	/.0	7/0
0,0	0.0	6 1	0.0	0.0	011	6,14 C 7C	0.0	50	9L9 #/0
0.0	17	0.0	0.0	0.0	611	0 CL 7'07	0.0	C.U	8L9 0/0
11	<i>L</i> 1	0.0	0,0	0.0	0.0	0.20	0.0	0.1	089
0.0	C 1 /*1	0.0	0.0	0.0	0.0	L I 0'0	0.0	8 U 0'7	689
0.0	2'1	0.0	0.0	0.0	0.0	2 L /1	0.0	0'A	789
0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0	57	989
0.0	0.0	00	0.0	0.0	0.0	0.0	0.0	ኒ ሮቴ	889
0.0	0.0	0.0	0,0	0.0	0.0	0.0	0.0	 	069
0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	٤ £ 	609 060
0.0	0'0 7 I	0.0	0.0	0.0	00	0.0	0.0	6 l ere	709 760
0.0	0.0	0.0	0.0	01	0.0	0.0	0.0	U I 7'1	909 560
0.0	2 I 2 I	0.0	0.0	0.0	0.0	0.0	0.0	εi 01	869
0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.8	002
0.0	0.001	0.0	0.0	0.0	0.0	0.0	0.0	0.0	202
0.0	8.1	0.0	0.0	0.0	54.1	0.0	0.0	4.1	407
0.0	0.0	0.0	0.0	0.0	0.0	þ.l	0.0	0'1	902
0.0	٤,1	2.1	0.0	0.0	0.0	0'0	0'0	6.0	80 <i>L</i>
0.0	0.0	0.0	0.0	0.0	0.0	0.0	0,0	<i>L</i> .0	01 <i>L</i>
0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	8,0	417
0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	£.1	91 <i>L</i>
0.0	£.1	0.0	0.0	0.0	0.0	0.0	0.0	9.1	812
0.0 ·	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1'1	07 <i>L</i>
<i>L</i> *1	0.0	0,0	0,0	0.0	0.0	0.0	0.0	9.1	72 <i>1</i>
0.0	9.1	0.0	0.0	0.0	1,22,1	0.0	0.0	6'1	42T
8.4	0.0	0,0	0.0	0.0	0.0	0.0	0.0	6'1	87 <i>L</i>
0.0	1,2	0.0	0.0	0.0	9.61	0.0	0.0	9.1	0£L
0,0	2.2	0,0	0'0	0.0	52'0	0.0	0.0	6.1	25 <i>L</i>
0.0	1 ,1	0,0	0.0	0.0	4,52	0.0	0.0	0.1	457
L'T	0.0	0.0	0,0	0.0	0'0	0.0	0.0	1'Z	9EL

0'0

0.0

č.£

0.0

8£L

0'0

9.2

0.0

0.0

Depth	Quartz	Х	Plag.	Total	Calcite	Hieh Me	Argonite	Proto	Dolomite
(cm)		Feldspar	•	Clay Miner.		Calcite	9	Dolomite	
740	2.2	0.0	0.0	0.0	0.0	0.0	0.0	υu	00
746	2.6	0.0	0.0	20.3	0.0	0.0	4	0,0	00
748	1.0	0.0	0'0	0.0	0.0	0.0	0.0		0.0 1
750	3.0	0.0	0.0	24.9	0.0	0.0	0.0	1.6	4 O O
752	4.5	0.0	0.0	16.3	0.0	1.0	0.0	51	0.0
754	2.3	0.0	0.0	21.5	0.0	0.0	0.0	- C	<u>.</u> .
756	2.6	0.0	0.0	0.0	0.0	0.0	0.0	2.7	0.0
758	1.3	0.0	0.0	19.0	0.0	0.0	0.0	1.6	0.0
760	1.5	0.0	0.0	0.0	0.0	0.0	00	01	
762	1.2	0.0	0.0	15.2	0.0	0.0	1.1	<u>}</u>	0.0
764	4.3	0.0	0.0	0.0	0.0	0.0	12	0.0	0.0
792	1.0	0.0	0.0	0.0	0.0	0.0	0.0	8	0.0
794	0.9	0.0	0.0	20.6	0.0	0.0	0.0	0.0	0.0
796	0.7	0.0	0.0	15.3	0.0	0.0	0.0	1.9	0.0
798	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
800	0.8	0.0	0.0	18.1	0.0	0.0	1.2	8.1	8
802	0.6	0.0	0.0	0.0	0.0	0.0	0.0	14	00
804	0.6	0.0	0.0	0.0	0.0	0,0	0.0	0.0	5.1
806	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
808	2.3	0.0	0.0	21.9	0.0	0.0	1.6	1.9	91
810	1,4	0.0	0.0	0.0	0.0	0.0	0.0	2.0	0.0
812	1.8	0.0	0.0	25.2	0.0	0.0	0.0	2.0	0.0
814	0.7	0.0	0.0	18.0	0.0	0.0	0.0	0.0	1.2
816	2.7	0.0	0.0	17.3	0.0	0.0	1.5	0.0	1.4
818	1.9	0.0	0.0	0.0	0.0	0.0	0.0	2,3	0.0
820	1.1	0.0	0.0	22.8	0.0	0.0	0.0	1.6	0.0
822	2.5	0.0	0'0	0.0	0.0	0.0	0.0	2.8	0.0
824	1.8	0.0	0.0	17.4	0.0	0.0	1.7	1.6	0.0
826	2.0	0.0	0.0	23.5	0.0	0.0	0.0	1.6	2.2
828	2.1	0.0	0.0	27.3	0.0	0.0	0.0	2.3	0.0
830	3.7	0.0	0.0	24.6	0.0	0.0	0.0	2.6	2.8
832	2.3	0.0	0.0	0.0	0.0	0.0	2.0	2.9	2.2
834	2.7		0.0	24,4	0.0	0.0	2.0	1.9	0.0
836	1.9	0.0	0.0	28.9	0'0	0.0	0.0	3.0	0.0
838	2.4	0.0	0.0	30.9	0.0	0.0	2.6	3,4	2.6
840	1:1	0.0	0:0	21.0	0.0	0.0	0.0	2.3	0.0
842	2.5	0.0	0.0	25.8	0.0	0.0	0.0	2.2	0.0
844	2.0	0.0	0.0	25.2	0.0	0.0	0.0	2.6	2.3
840	3.4	0.0	0.0	24.9	0.0	0.0	0.0	1.8	1.9

Danth	0.04	2					,		
nudaon	Quartz	4	Plag.	Total	Calcite	High Mg	Aragonite	Proto	Dolomite
(cm)		Feldspar		Clay Miner.		Calcite	0	Dolomite	
848	4.4	0.0	0.0	29.6	00	00	00		
850	3.6	0.0	1.1	23.0	00	0.0	0.0	7.7	3.3
852	3.9	0.0	4.0	00	0.0		0.0	8.1	1.8
856	2.8	0.0	0.0	0.0	0.0	0.0	0.0	2.7	0.0
860	1.4	0.0	0.0	10.5	0.0	0.0	0.0	2.5	0.0
862	2.1	0.0	0.0	0.01	0.0	0.0	0.0	1.8	1.8
864	2.9	0.0		8.12	0.0	0.0	0.0	1.5	2.1
866	20	0'0	4 C	18.1	0.0	0.0	0.0	1.8	1.6
878	0,0	0.0	0.0	60.4	0.0	0.0	0.0	0.8	0.0
000		0.0	0.0	29.6	0.0	0.0	0.0	1.7	61
010	0.0	0.0	0.0	14.6	0.0	0.0	0.0	1.3	
7/0	· · ·	0.0	0.0	0.0	0.0	0.0	0.0	2.1	. c
4/0 876		0.0	0.0	14.6	0.0	0.0	0.0	1.2	00
0/0	4.7 2 I	4.7	0.0	22.9	0.0	0.0	0.0	6.1	2.4
0/0	<u>.</u> -	0.0	0.0	25.0	0.0	0.0	0.0	1.7	
000		0.0	0.0	0.0	0.0	0.0	0.0	10	0.0
799	0.9	0.0	0.0	17.0	0.0	0.0	0.0	0.0	0.0 1
884 222	6.1	0.0	0.0	0.0	0.0	0.0	00		
880	1.7	0.0	0.0	22.3	0.0	0.0	0.0	1.5	0.0 - c
888	7.1	0.0	0.0	21.3	0.0	0.0	0.0	0	1.2
0.68	č.0	0.0	0.0	0.0	0.0	0.0	0.0	0.7	0.0
892	0.1	0.0	0.0	16.6	0.0	0.0	00	6 1	
894	1.1	0.0	0'0	20.6	0.0	0.0	00	7'T	n'n -
896	1.1	0.0	0.0	0.0	0.0	00	0.0		
868	1.0	0.0	0.0	12.7	0.0	0.0	00	0.0	0.0
900	0.8	0.0	0.0	20.7	0.0	00		0.1	<u>.</u>
902	2.0	0.0	0.0	21.2	0.0	0.0	0.0	נ. 	8.1
904	1.1	0.0	0.8	17.6	0.0	0.0	0.0	0.1	ני <u>-</u>
906	1.5	0.0	3.4	0.0	0.0	0.0	0.0		7.1
908	1.2	0.0	0.0	0.0	0.0	0.0		<u>.</u>	<u>.</u>
016	1.2	0.0	0.0	14.3	0.0	00	0,0	<u>.</u>	0.0
912	2.2	0.0	0.0	19.3	00	0.0	0.0	4	0.0
914	4.3	0.0	0.0	40.2	00	0.0	n o	4.1	0.0
916	3.5	0.0	1.2	33.8	0.0	0.0		7.1 1.6	1.9
918	4.0	0.0	0.0	38.2	0.0	0.0	0.0	2.4	2.1
920	0.5	0.0	0.0	20.6	00	00	0.0	4.2	2.7
924	0.6	0.0	0.0	24.0	0.0	0.0	1.1	0.0	0.0
926	0.9	0.0	0.0	00	0.0	0.0	0.0	1.8	0.0
928	2.2	0.0	00	2.0	0.0	0.0	2.1	2.4	0.0
930	0.6	0.0	0.0	0.01	0.0	0.0	0.0	1.5	0.0
			0.0	19.6	0.0	0.0	0.0	6.1	0'0

Depth (cm)	Quartz	K Feldspar	Plag.	Total Clay Miner.	Calcite	High Mg Calcite	Aragonite	Proto Dolomite	Dolomite
012	0.6	0.0	00	21.6	0.0	0.0	0.0	18	00
934	0.7	0.0	0.0	20.6	0.0	0.0	0.0	1.8	0.0
936	1.1	0.0	0.0	22.4	0.0	0.0	0.0	1.9	0.0
938	0.5	0.0	0.0	17.4	0.0	0.0	0.0	19	0.0
940	0.8	0.0	0.0	18.0	0.0	0.0	0.0	1.6	0.0
942	1.7	0.0	0.0	21.6	0.0	0.0	0.0	2.8	0.0
944	0.7	0.0	0.0	0.0	0.0	0.0	0.0	2.4	0.0
946	0.6	0.0	0.0	21.0	0.0	0.0	0.0	1.7	0.0
948	0.9	0.0	0.0	0.0	0.0	0.0	0.0	3.6	0.0
950	2.0	0.0	0.0	0.0	0.0	0.0	1.6	0.0	0.0
952	0.9	0.0	0.0	22.3	0.0	0.0	0.0	2.0	0.0
954	1.3	0.0	0.0	22.2	0.0	0.0	0.0	1.6	0.0
956	0,8	0.0	0.0	20,5	0.0	0.0	0,0	1.3	0.0
958	0.7	0.0	0.0	25.7	0.0	0.0	0.0	1.4	0.0
960	1.3	0.0	2.1	0.0	0.0	0.0	0.0	1.8	0.0
962	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
964	0.3	0.0	0.0	51.4	0.0	0,0	0.0	0.0	0.3
972	0.8	0,0	0.0	16.1	0.0	0.0	0.0	1.4	0.0
974	1.0	0.0	0.0	11.8	0.0	0.0	0,0	0.0	0.0
976	1.7	0.0	0.0	20.0	0.0	0.0	0.0	1.6	0.0
978	1.0	0.0	0.0	21.2	0.0	0.0	1.1	1.7	0.0
980	0.7	0.0	0.0	0.0	0.0	0.0	0.0	1.1	1.4
982	0.7	0.0	0,0	16.8	0.0	0.0	0.0	1.5	0.0
984	1.0	0.0	0.0	22.3	0.0	0.0	0.0	0.0	0.0
986	1.2	0.0	0.0	20.2	0.0	0.0	0.0	1.6	1.9
988	0.7	0.0	0.0	0.0	0.0	0.0	0.0	1.8	0.0
990	0.6	0.0	0.0	15.3	0.0	0.0	0.0	2.9	0.0
992	0.4	0.0	0.0	14.0	0.0	0.0	0.0	1.5	0.0
994	1.0	0.0	0.0	19.8	0.0	1.0	0.0	2.0	1.7
996	0.5	0.0	0.0	20.1	0.0	0.0	0.0	1.7	1.5
998	0.8	0.0	0.0	16.1	0.0	0.0	0.0	1.4	2.0
1,000	1.1	0.0	0.0	30.3	0.0	0.0	0,0	3.5	3.3
1,002	0.6	0.0	0.0	23.3	0.0	0.0	0.0	1.5	0.0
1,004	0.6	0.0	0.0	20.1	0.0	0.0	0.0	1.8	0.0
1,006	1.9	0.0	0.0	0.0	0.0	2.5	0.0	0.0	1.8
1,008	0.6	0.0	0.0	20.5	0.0	0.0	0,0	0.0	0.0
1,010	3.7	0.0	0.0	0.0	0.0	0.0	0.0	2.5	0.0
1,012	1.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,014	1.8	0.0	0.0	22.5	0.0	0.0	0.0	2.2	0.0

Mineralogical Composition of North Core at North Ingebright Lake

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Depth	Quartz	×	Plag.	Total	Calcite	High Mg	Aragonite	Proto	Dolomite
(cm)		Feldspar		Clay Miner.		Calcite)	Dolomite	
1,016	4.5	0.0	0.0	29.0	0,0	0.0	0.0	6.1	0.0
1,018	1.7	0.0	0.0	25.8	0.0	0.0	0.0	2.2	0.0
1,020	0.8	0.0	0.0	21.6	0.0	0.0	0.0	1.5	E.1
1,022	1.0	0.0	0.0	17.9	0.0	0.0	0.0	1.7	1.1
1,024	1.3	0.0	0.0	20.3	0.0	0.0	0.0	1.5	1,4
1,026	0.7	0.0	0.0	22.6	0.0	0.0	0.0	1,6	0.0
1,028	0.5	0.0	0.0	1.9.1	0.0	0.0	1.3	1.4	0.0
1,030	0.5	0'0	0.0	22.2	0.0	0.0	1.5	1.8	1.2
1,032	0.5	0.0	0.0	7.4	0.0	0.0	0.0	0.0	0.5
1,034	0.9	6.5	0.0	17.7	0.0	0.0	0.0	1.4	0.0
1,036	0.4	0.0	0.0	0.0	0.0	0.0	0.0	1.3	0.0
1,038	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,040	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,042	1.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0,0
1,044	2.7	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
1,046	0.4	0.0	0.0	15.7	0.0	0.0	0.0	0'0	0'0
1,048	0.7	0'0	0'0	17.7	0.0	0'0	2.0	2.0	1.3
1,050	0.5	0.0	0.0	20,4	0.0	0.0	0.0	1.7	6.1
1,052	2.4	1.7	0'0	0.0	0.0	0.0	0.0	0.0	0.0
1,054	0.5	0.0	0.0	17.9	0.0	0.0	0.0	1.6	0.0
1,056	0.5	0.0	0.0	16.9	0.0	0.0	0.0	2.2	0.0
1,058	0.9	0.0	0'0	0'0	0.0	0.0	0.0	1.6	1.7
1,060	2.5	0.0	0.0	0.0	0.0	0.0	0.0	1.3	0.0
1,062	2.3	0.0	0.0	0.0	0.0	0'0	1.7	1.8	2.3
1,064	2.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,066	0,4	0.0	0.0	0.0	0.0	0.0	0.0	0.5	0.4
1,068	1.1	0.0	0'0	18.5	0.0	0.0	1.8	2.2	0.0
1,070	2.9	0.0	0.0	3.9	0.0	0.0	6.1	2.0	1.1
1,072	0.6	0.0	0.0	0.0	0.0	0.0	0.0	2.4	0.0
1,074	0.4	0.0	0.0	20.2	0.0	0.0	1.7	1.8	0.0
1,076	0.4	0.0	0.0	15.5	0.0	0.0	1.2	1.3	0.0
1,078	0.6	0.0	0.0	0.0	0.0	0.0	1.2	1.5	0.0
1,080	0.9	0.0	0.0	0.0	0.0	0:0	0.0	2.2	0.0
1,082	0.4	0.0	0.0	12.8	0.0	0.0	0.0	0.0	0.0
1,084	0.1	0.0	0.0	0.0	0.0	0'0	1.8	2.6	0.0
1,086	1.2	0.0	0.0	25.8	0.0	0'0	0.0	1.9	1,6
1,088	0.9	0.0	0.0	14,9	0.0	0.0	0.0	1.4	1.4
060'1	1.3	0:0	0.0	0.0	0.0	0.0	0.0	2.7	1.6
1,092	0.5	0.0	0'0	0.0	0.0	0.0	0.0	0.0	2.6

Depth	Quartz	K	Plag.	Total	Calcite	High Mg	Araponite	Proto	Dolomite
(cm)		Feldspar		Clay Miner.		Calcite		Dolomite	
1,094	2.5	0.0	0.0	0.0	0.0	0.0	0.0	63	UU
1,096	0.0	0.0	0.0	0.0	0.0	0.0	1.4	00	0.0
1,098	0.0	0.0	1.4	0.0	0.0	0.0	0.0	0.0	0.0
1,100	0.2	0.0	0.0	0.0	0.0	0.9	0.0	0.0	00
1,102	0.0	0.0	0.0	0.0	0.0	0.0	0.0	8.0	00
1,104	0.0	0.0	0.0	0.0	0.0	0.0	00	0.0	
1,106	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,108	0.3	0.0	0.0	0.0	0.0	0.0	0.0	0.4	01
1,110	0.0	0.0	9.8	0.0	0.0	0.0	0.0	0.0	
1,112	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.5	0.0
1,114	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00	0.0
1,116	1.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,118	0.4	0.0	0.0	0.0	0.0	0.0	0.0	1.4	1.7
1,120	0.9	0.0	0.0	0.0	0.0	0.0	0.0	1.3	00
1,122	0.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,124	0.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,126	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
1,128	0.0	0.0	10.7	0.0	0.0	0.0	0.0	0.0	
1,130	0.5	0.0	0.0	17,0	0'0	0.0	0.0	0.0	1.6
1,132	0.0	0.0	9.7	0.0	0.0	0.0	0.0	0.0	0.0
1,134	0.1	0.0	2.7	0'0	0.0	0.0	0.0	0.0	0.0
1,130	0.2	0.0	8.0	0.0	0'0	0.0	0.0	0.0	0.0
1,138	0.5	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0
1,140	0.0	0.0	7.9	0.0	0.0	0.0	0.0	0.0	0.0
1,142	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0,0	1.1
1,144	0.7	0.0	0.0	0.0	0.0	0.0	0.0	2.5	0.0
1,148	0.4	0.0	0.0	0.0	0.0	0.0	0.9	0.0	0.0
00111	0.0	0.0	12.9	0.0	0.0	0.0	0.0	0.0	0.0
761,1	5.U 0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
+011	0.0	0.0	0.0	0.0	0.0	0.0	1.0	0.9	0.0
00111	0.4 0.2	0.0	0.0	0'0	0.0	0.0	0.0	1.7	0.0
901'1	0.J	0.0	0.0	0'0	0.0	0.0	0.0	0.0	1.3
1,160	0.8	0.0	0.0	14.1	0.0	0.0	0.0	1.1	1,4
1,102	0.9	0.0	0.0	0.0	0.0	0.0	0.0	1,4	0,0
1,164	0.4	0.0	0.0	15.6	0.0	0.0	1.6	1.7	0.0
1,100	0.0	0.0	0.0	15.0	0.0	1.3	0.0	1.6	0.0
1,108	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.5	1.6
0/1'1	0.7	0.0	0.0	0.0	0.0	0.0	0.0	1.2	1.3
1,172	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0

Depth (cm)	Quartz	K Feldspar	Plag.	Total Clay Miner.	Calcite	High Mg Calcite	Aragonite	Proto Dolomite	Dolomite
1.174	1.6	0.0	0.0	24.0	0.0	0.0	0.0	1.9	0.0
1.176	0.4	0.0	5.3	0.0	0.0	0.0	0.0	1.5	0.0
1,178	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,180	0.4	0.0	0.0	0.0	0.0	0.0	0.0	1.6	1.8
1,182	0.0	0.0	0.0	0,0	0.0	0.0	1,3	0.0	0.0
1,184	0.0	0.0	0.0	0.0	0.0	0.0	2,3	2.4	0.0
1,186	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.5	0.0
1,188	0.0	0,0	13.6	0,0	0.0	0,0	0.0	0.0	0.0
1,190	0.0	0.0	0.5	0.0	0.0	0.0	0.0	0.0	0,0
1,192	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,194	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.1
1,198	0.0	0.0	11.5	0.0	0.0	0.0	0.0	0,0	0.0
1,200	0.0	0.0	7.5	0.0	0.0	0.0	0.0	0.0	0.0
1,202	0.3	0,0	0.0	0.0	0.0	0.0	0.0	1.2	0.0
1,204	0.0	0,0	1.8	0.0	0.0	0.0	0.0	0.0	0.0
1,206	0.0	0.0	19.7	0.0	0.0	0.0	0.0	0.0	0.0
1,208	0.0	0.0	12.1	0.0	0.0	0.0	0.0	0.0	0.0
1,210	0.0	0.0	10.9	0.0	0.0	0.0	0.0	0.0	0.0
1,212	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0	1.2
1,214	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0	0.0
1,216	0.0	0.0	10,3	0.0	0.0	0.0	0.0	0.0	0.0
1,218	0.0	0.0	10.8	0.0	0.0	0.0	0.0	0.0	0.0
1,220	0.0	0.0	9.3	0.0	0.0	0.0	0.0	0.0	0.0
1,222	0.0	0.0	14,4	0.0	0.0	0.0	0.0	0.0	0.0
1,224	0,0	0.0	12.1	0.0	0.0	0.0	0.0	0.0	0.0
1,326	0.6	0.0	0.0	21.9	0.0	0.0	0.0	1.9	2.0
1,328	0.0	0.0	12.7	0.0	0.0	0.0	0.0	4.1	0.0
1,330	0.3	0.0	11.8	0.0	0.0	0.0	0.0	1.2	0.0
1,332	0.0	0.0	9.4	0.0	0.0	0.0	0.0	1.1	0.0
1,334	0.4	0.0	12.6	0.0	0.0	0.0	0.0	0.0	0.0
1,336	1.2	0.0	10.5	0.0	0.0	0.0	0.0	0.0	- 1.9
1,338	0.7	0.0	37.6	0.0	0.0	0.0	0.0	0.0	1.0
1,340	2.9	0.0	0.0	0.0	0.0	1.4	0.0	1.6	0.5
1,342	3.4	0.0	11.8	0.0	0.0	0.0	0.0	1.7	2.1
1,344	2.3	0.0	2.3	0.0	0.0	0.0	0.0	0.0	2.5
1,346	3.5	0.0	0,0	0.0	0.0	1.4	0.0	0.0	2.5
1,348	0.9	0.0	0.0	0.0	0.0	0.0	1.7	0.0	0,0
1,350	4.6	0.0	0.0	0.0	0.0	0.0	0.0	8,3	0.0
1,352	6.8	0.0	0.0	0.0	0.0	0.0	0.0	14.0	0,0

Mineralogical Composition of North Core at North Ingebright Lake

Depth	Quartz	×	Plag.	Total	Calcite	Hieh Me	Aragonite	Proto	Dolomite
(cm)		Feldspar	•	Clay Miner.		Calcite	0	Dolomite	
1,354	3.3	0.0	0.0	0.0	0.0	0:0	0.0	4.6	0.0
1,356	3.0	0.0	0.0	0.0	0.0	3.0	0.0	0.0	0.0
1,358	1.8	1.9	0.0	13.2	0'0	0.7	0.0	6'1	2.4
1,360	2.4	2.2	0.0	0.0	0.0	0.0	0'0	2.0	0.0
1,362	3.5	2.5	0.0	0.0	0'0	0.0	0.0	2.8	0.0
1,364	3.5	2.8	0.0	0.0	1.1	0.0	0.0	0.0	2.2
1,366	2.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	6.8
1,368	2.3	0.0	0.0	0'0	0.0	0.0	0.0	1.9	2.4
1,370	3.1	0.0	0.0	17.2	0.0	0.0	0.0	0.0	0'0
1,372	4.3	3,1	0.0	25.3	0.0	0.0	0.0	2.1	2.6
1,374	3.4	0.0	0.0	0.0	0.0	0.0	0.0	2.1	0.0
1,376	3.0	2.0	0,9	32.3	0.0	0.0	0.0	2.3	0.0
1,380	1.9	0.0	0.0	27.5	0.0	0.0	0.0	2.4	2.0
1,382	3.0	0.0	0.0	27.3	0.0	1.5	2.0	1.2	2.0
1,384	2.2	1.7	0.0	30.0	0.0	0.0	2.1	2.5	3.0
1,386	3.7	0.0	0.0	0.0	0.0	0.0	3.8	4.3	0.0
1,388	3.4	2.8	0.0	38.1	0.0	0.0	0.0	3.0	2.0
1,390	2.4	1.7	0.0	24.9	0.0	0.0	0.0	2.1	0'0
1,392	6.1	0.0	0.0	0.0	0.0	0.0	0.0	7.0	0.0
1,394	1.6	0.0	0.0	36.9	0.0	1.2	0.0	4.2	2.7
1,396	0.0	0.0	0.0	22.3	0.0	0,0	0.0	1.4	0.0
1,398	0.3	0.0	0.0	18.8	0.0	0.0	0.0	1.5	0.0
1,400	0.7	0.0	1.4	17.7	0.0	0.0	0.0	2.1	0.0
1,402	0.5	0'0	0.0	1.9.1	0.0	0.0	0.0	2.0	0.0
1,404	0.8	0.0	0.0	26.5	0.0	0.0	0.0	2.1	1.4
1,406	1.3	0.0	0.0	29.8	0.0	0.0	0.0	2.9	2.3
1,408	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,410	0.1	0.0	0.0	3.1	0.0	0.0	0.0	0.4	0.2
1,412	0.7	0.0	0.0	0.0	0.0	0.0	0.0	3.7	2.3
1,414	0.7	0.0	0.0	0.0	0.0	0.0	1.5	0.0	0.0
1,416	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	.0.0
1,418	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,420	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,422	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,424	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0	0.0
1,426	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.5	0.0
1,428	0.0	0.0	0.0	0.0	0.0	0.0	0.0	3.5	0.0
1,430	0.0	0.0	0.0	0.0	0.0	0.0	0.0	3.8	0.0
1,432	0.0	0.0	0.0	0.0	0.0	0.0	0.0	15	0.0

Depth	Ouartz	×	Plad	Total	Calaita	Utch Mc			
(cm)	,	Feldspar	9	Clay Miner.	Calute	Calcite	Aragomte	Proto Dolomite	Dolomite
1,434	0.0	0.0	0.0	0.0	0.0	0.0	00	1.0	
1,436	0.0	0.0	0.0	0.0	0.0	0.0	2.3	2.5	0.0
1,438	0.0	0.0	0.0	18.5	0.0	0.0	2.2	2.6	0.0
1,440	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,442	0.0	0.0	0.0	14.3	0.0	0.0	1.6	1.7	0.0
1,444	0.0	0.0	0.0	27.6	0.0	0.0	0.0	2.6	0.0
1,446	0.0	0.0	0'0	24.1	0.0	0.8	0.0	5.0	0.0
1,448	0.0	0.0	0.0	19.9	0.0	0.0	1.2	C 1	0.0
1,450	0.0	0.0	0.0	29.4	0.0	0'0	0.0	3.3	0.0
1,452	0.0	0.0	0.0	20.1	0.0	0.0	0.0	2.0	00
1,454	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.9	0.0
1,456	0.4	0.0	0.0	19,9	0.0	0.0	0.0	6.1	0.0
1,458	0.0	0.0	0.0	13.2	0.0	0.0	0.0	1.1	0.0
1,460	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.2	0.0
1,462	0.8	0.0	0.0	14.3	0.0	0.0	1.2	1.5	1.7
1,464	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,466	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,468	1.6	0.0	0.0	26.9	0.0	0.0	0.0	1.7	2.1
1,470	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,472	0.4	0.0	0.0	20.4	0.0	0.0	0.0	0.0	0.0
1,474	0.5	0.0	0,0	0.0	0.0	0.0	0.0	0.0	0.0
1,476	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
1,478	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,480	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,482	0.3	0.0	0.0	18.9	0.0	0.0	0.0	2.4	0.0
1,484	0.0	0.0	0.0	17.3	0.0	0.0	0.0	1.7	0.0
1,480	0.0	0.0	0.0	0.0	0.0	0.0	2.2	1.9	0.0
1,488	0.5	0.0	0'0	0.0	0.0	0.0	2.0	1.6	0.0
1,490	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,492	0.0	0.0	0.0	0.0	0.0	0,9	1.8	1.4	0.0
1,494	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,496	0.3	0.0	0.0	0.0	0.0	0.0	0'0	1.5	0.0
1,498	0.4	0.0	0.0	22.7	0.0	0.0	0.0	1.9	0.0
1,500	0.7	0.0	0.0	0.0	0.0	0.0	0.0	5.2	0.0
+0c'1	0.8	0.0	0.0	0.0	0.0	1.4	0.0	2.6	0.0
00c'1	0.4	0.0	0.0	0.0	0.0	0.0	0.0	2.5	0.0
80C,1	0'0	0.0	0.0	0'0	0.0	0'0	0.0	1.4	0.0
016,1	0'0	0.0	0.0	0.0	0.0	0.0	0.0	1.2	0.0
71C'1	0.4	0.0	0.0	23.0	0.0	0.0	2.0	2.2	0.0

Danth		:	1						
(cm)	Xual LL	Feldspar	riag.	Total Clav Miner.	Calcite	High Mg Caloite	Aragonite	Proto	Dolomite
						Calcile		Dolomite	
1,514	0.4	0.0	0.0	23.0	0'0	0.0	00	13	00
1,516	0.3	0.0	0.0	0.0	0.0	0.0	2.5 C	<u>;</u> -	0.0
1,518	0.0	0.0	0.0	0.0	0.0	0.0	7 T	1.7	0.0
1,520	1.0	0.0	0.0	0.0	0.0	0.0		1 C	0'0
1,522	0.5	0.0	0.0	0.0	0.0	00			0.0
1,524	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,526	0.2	0.0	0.0	8.9	00	0.0	0.0	0.0	0.0
1,528	0.9	0.0	0.0	26.7	0.0	0.0	0.0	0.0	0.0
1,530	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.1	0.0
1,532	0.3	0.0	0.0	921	0.0	0.0	0.0	0.0	0.0
1,534	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,536	0.3	0.0	00	12.2		0.0	0.0	2.9	0.0
1,538	0.6	0.0	00	0.0	0,0	0.0	0.0	0.1	0.0
1.540	0.0	00	0.0	00	0.0	0.0	5.1	0.0	0.0
1.542	0.0	0.0		0.0	0.0	0.0	1.2	0.0	0.0
544	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.7	0.0
1.546	0.0	0.0		0.0	0.0	0.0	0.0	0.0	0.0
1.548	0.0	0.0	0.0	18.2	0.0	0.0	0'0	0.0	0.0
035 1	100	0.0	0.0	0.0	0.0	0.0	1.4	1.7	0.0
543 [0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.5	0.0
7001	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0.0	0,0
400 ⁴ 1	1 .0	0.0	0.0	25.6	0.0	0,0	2.0	2.4	0.0
0001	0.1	0.0	0.0	22.5	0.0	0.0	2.5	2.4	0.0
0001	C.U	0.0	0.0	0.0	0.0	0.0	1,6	2.3	0.0
000,1	0.8	0.0	0.0	0.0	0.0	0'0	0.0	0.0	0.0
70041	c.0 c.0	0.0	0.0	0.0	0.0	0.0	1.5	1.7	0.0
+0c*1	0.0	0.0	0.0	17.5	0.0	0'0	1.5	1.2	0.0
000-1	r 7	0.0	0.0	20.4	0,0	0'0	0.0	1.7	1.1
0251	7.0	0.0	0.0	8.02	0.0	0.0	2.0	2.0	0.0
0/01	0.0	0.0	0.0	20.9	0.0	0.0	0.0	2.1	0.0
7/01	C.D	0.0	0.0	27.6	0.0	0'0	0.0	2.5	0.0
+/C'I	c.0	0'0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
0/01	0.0	0'0	0.0	16.6	0.0	0.0	0.0	1.7	0.0
8/C'I	0.0	0.0	0.0	23.3	0.0	0.0	1.9	2.1	0.0
1,500	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
700'1	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0	0.0
080,1	0.7	0.0	0.0	20.4	0.0	0.0	0.0	2.0	0.0
880,1	0.4	0.0	0.0	38.8	0.0	0.0	0.0	4.5	2.2
060,1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.6	0.0
740'1	0.0	0.0	0.0	0.0	0.0	0.0	2.4	2.5	0.0

Denth	Ollartz	2		1					
(und)	Xual Lo		riag.	lotal	Calcite	High Mg	Aragonite	Proto	Dolomite
(cm)		Feldspar		Clay Miner.		Calcite		Dolomite	
1,594	0.0	0.0	0.0	25.4	00	00	~~~~		
1,596	0.0	0.0	0.0	28.4	0.0	0.0	0.0 C	2.0	0.0
1,598	0.0	0.0	0.0	0.0	0.0	0.0	1.2	2.3	0.0
1,600	0.0	0.0	0.0	00	0.0	0.0	0.0	1'0	0.0
1,602	0.0	0.0	00	0.0	0.0	0.0	0.0	0.0	0.0
1,604	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,606	0.0	0.0	0.0	0.0	0.0	0,0	0.0	1.1	0.0
1,608	0.0	00	0.0	0.0	0.0	0.0	1.0	0.0	0.0
1,610	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1.612	0.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1.614	200		0.0	0.0	0.0	0.0	0.0	0.0	0.0
1 616	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1 618	0.0	0.0	12.0	0.0	0.0	0.0	0.0	1.4	0.0
1 620		0.0	0.0	14.0	0.0	0.0	0.0	1.5	0.0
1 627	0.0	0.0	0.0	13.4	0.0	0.0	0.1	1.5	0.0
770'1	0.0 C	0.0	0.0	16,9	0.0	0.0	1.1	1.6	0.0
2091	2.0	0.0	0.0	0.0	0.0	0.0	0,0	2.2	0.0
070'1	c.0	0.0	0.0	18.5	0.0	0.0	0.0	2.0	00
0701	0.0	0.0	0.0	19.4	0.0	0.0	2.2	2.1	0.0
0001	0.0	0.0	0.0	0.0	0.0	0.0	1.2	5 1	00
1,032	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2	0.0
4cn'i	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
0001	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
0001	0.0	0.0	12.5	0.0	0.0	0.0	0.0	0.0	0.0
1,040	c.u	0.0	0.0	0.0	0.0	0.0	1.3	0.0	0.0
1 644	0.0	0.0	13.9	0.0	0.0	0.0	0.0	0.0	0.0
440'1	1.0	0.0	11.7	0.0	0.0	0.0	0.0	. I	0.0
1 648	+ O	0.0	0.11	0.0	0.0	0.0	0.0	1.3	2.2
1.650	1.0	1 , C	9.9	0.0	0.0	0.0	0.0	0.9	0.0
2591	0.0	0.0	14.7	0.0	0.0	0.0	0.0	0.0	0.0
7591	0.0	0.0	8.21	0.0	0.0	0.0	0.0	0.0	0.0
259 I	0.0	0.0	12,8	0.0	0.0	0.0	0.0	0.0	0.0
059 1	0.0	0.0	9.4	0.0	0.0	0.0	0.0	0.0	0.0
0001	4.0	0.0	0.0	16.2	0.0	0.0	1.3	0.0	00
000'1	c.0	0.0	0.0	0.0	0.0	0.0	2.3	2.2	0.0
7001		0.0	0.0	20.8	0.0	0.0	0.0	2.1	51
1 666	+	0.0	0.0	27.1	0.0	0.0	0.0	2.1	0.0
0001		0.0	0.0	22.8	0.0	0.0	1.8	1.4	
0001	C. 0	0.0	0.0	18.8	0.0	0.0	0.0	1.5	0.0
2224	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0

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_	Quartz	K	Plag.	Total	Calcite	High Mg	Aragonite	Proto	Dolomite
		Feldspar		Clay Miner.		Calcite		Dolomite	
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
	0.3	0.0	0.0	0.0	0.0	1.3	0.0	0.0	0.0
	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.5	0.0	0.0	0.0	0.0	0.0	0,0	1.5	1.5
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
	1.1	0:0	0.0	16.2	0'0	0.0	0.0	1.8	0.0
	1.5	0:0	0.0	13.8	0.0	0.0	1.5	1.6	0.0
	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	1.2	0.0	0.0	0.0	0.0	0.0	1.7	1.9	0.0
	0.7	0.0	0.0	18.6	0.0	0.0	0.0	1.3	1.2
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0,0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0,0
	0.4	0.0	0.0	0.0	0.0	0.7	0.0	2.2	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
	0.4	0.0	0.0	23.7	0.0	0.0	0.0	2.1	1.9
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	11,4	0.0	0.0	1.1	0'0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
	0.0	0.0	0.0	18.5	0.0	0.0	0.0	1.4	1.2
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.3
	0.0	0'0	0.0	0:0	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.3	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
	0.9	0.0	0.0	0.0	0.0	0.0	0'0	1.9	0.0
	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0

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Depth	Quartz	Х	Plag.	Total	Calcite	High Mg	Aragonite	Proto	Dolomite
(cm)		Feldspar		Clay Miner.		Calcite)	Dolomite	
1,800	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	3.0
1,802	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,804	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
1,806	0.0	0.0	2.2	0.0	0.0	0.0	0.0	0.0	0.0
1,808	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,810	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,812	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,814	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,816	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,818	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,820	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,822	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0,0
1,824	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
1,826	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,828	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,830	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,832	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,834	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,836	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,912	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,914	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,916	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,918	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,920	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0.0
1,922	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,924	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,926	0.0	0.0	0.0	10.9	0.0	0.0	0.0	1.8	0.0
1,928	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,930	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,932	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0.0
1,934	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,936	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,938	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,940	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,942	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,944	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,946	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,948	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,950	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0

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Mineralogical Composition of North Core at North Ingebright Lake

Depth	Quartz	×	Plag.	Total	Calcite	Hiah Ma	Arganita	Drata	Polonite.
(cm)		Feldspar		Clay Miner.		Calcite		Dolomite	
1,952	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00	VV
1,954	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.5	0.0
1,956	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,958	0.0	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0
1,960	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
1,962	0.0	0.0	0'0	0.0	0.0	0.0	00	0.0	
1,964	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00	0.0
1,966	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
1,968	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00	0.0
1,970	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,974	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
1,976	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,978	0.0	0.0	0'0	0.0	0.0	0.0	0.0	00	0.0
1,980	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,982	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00	0.0
1,984	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,986	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00	0.0
1,988	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
066'1	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
1,992	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,994	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
1,996	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,000	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,00,2	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
2,004	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,000	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
2,005	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
21012	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
71017	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
21014	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,010	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
21018	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
7,020	0.0	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0
770'7	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
470'Z	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
2,026	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,028	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0,0	0.0
050,2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
750,2	0'N	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0

Depth	Ouartz	X	Plao.	Total	Calcite	Hiah Ma	Arganite	Proto	Dalamita
(u)		Feldspar	0	Clay Miner.		Calcite		Dolomite	
2,034	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,036	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,038	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,040	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,042	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
2,044	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,046	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,048	0.0	0.0	40.5	0.0	0.0	0.0	0.0	0.0	0.0
2,052	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
2,054	0.0	0:0	0'0	0.0	0.0	0.0	0.0	0'0	0.0
2,056	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
2,058	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,060	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,062	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0
2,064	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,066	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.9	0.0
2,068	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,070	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,072	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,074	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.2	0'0
2,076	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
2,078	0.0	0.0	1.6	0.0	0.0	0.0	1.3	0'0	0.0
2,080	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,082	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,084	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.7
2,086	0.0	0.0	0.0	0.0	0,0	0'0	0.0	0.0	0.0
2,088	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,090	0,0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,092	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0'0	0.0
2,094	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
2,096	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,098	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0,0
2,100	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,102	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,104	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.0	0.0
2,108	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,110	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
2,112	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,114	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0

Mineralogical Composition of North Core at North Ingebright Lake

Depth	Quartz	×	Plag.	Total	Calcite	Hieh Me	Argonite	Proto	Delamita
(cm)		Feldspar)	Clay Miner.		Calcite	9	Dolomite	
2,116	0.0	0.0	0.0	0.0	0.0	00	VV	VV	00
2,118	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,120	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,122	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,124	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,126	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,128	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,130	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,132	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,134	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,136	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,138	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
2,140	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,142	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,144	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
2,146	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
2,148	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
2,150	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
2,152	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,154	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,150	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
2,158	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
2,160	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,162	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,164	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,100	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,168	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
0/1/7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
7/1/7	0.0	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0
4/1/7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
0/1/7	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0.0	. 0.0
2,1,5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,180	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
7,182	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
491.2	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
2,180	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
001.2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
061,2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
76147	0.0	0.0	0.0	0.0	0,0	0.0	0.0	0.0	0.0

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Depth	Quartz	X	Plag.	Total	Calcite	High Mg	Aragonite	Proto	Dolomite
(cm)		Feldspar		Clay Miner.		Calcite)	Dolomite	
2,194	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,196	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,198	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,200	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,202	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,204	0.0	0.0	0.0	0.0	0.0	0,0	0.0	0.0	0.0
2,206	0.0	0.0	0.0	0.0	0.0	0,0	0.0	0.0	0.0
2,208	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,210	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,212	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,214	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,216	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
2,218	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0
2,220	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,222	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0,0
2,224	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,226	0.0	0,0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,228	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,230	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
2,232	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,234	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,236	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,238	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,240	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,242	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,244	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,246	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,250	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,252	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,254	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,256	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	. 0.0
2,258	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
2,260	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,262	0,0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,264	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,266	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,268	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,270	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
2,272	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0'0	0.0

-									
(cm)	Quartz	K Feldspar	Plag.	Total Clay Miner.	Calcite	High Mg Calcite	Aragonite	Proto Dolomite	Dolomite
2,274	0.0	0.0	0.0	0.0	0.0	0.0	00	00	VV
2,276	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
2,280	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
2,282	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
2,284	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
2,286	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00	00
2,288	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
2,290	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,292	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
2,294	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
2,296	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0

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Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
(em)		•• · · ·					ME CALCIC	
12	0,0	0.0	0.0	20.6	0.0	0.0		5.6
14	0.0	0.0	0.0	0.0	0.0	0.0		71.8
16	0.0	0.0	0.0	20.3	0.0	0.0	15.9	
18	0.0	0.0	6.4	6.5	56.8	0.0		52.8
20	0.0	0.0	0.0	7.6	0.0	0.0	8.4	56.6
22	0.0	0.0	17.4	10,6	0.0	0.0		52.3
24	0.0	0.0	0.0	16.9	0.0	0,0		52.5
26	0.0	0.0	4.3	5.6	32.2	0.0	18.9	
28	0.0	0.0	0.0	4.9	34,0	0.0		54.6
30	0.0	0.0	0.0	12.2	0.0	0.0		51.9
32	0.0	11.9	0.0	13.4	0.0	0.0		
34	0.0	16,1	0.0	0.0	0.0	0.0	11.6	56.5
36	0.0	0.0	9,2	8.5	0.0	0.0		56.3
38	0.0	0.0	0.0	34.6	0.0	0,0		55,7
40	0.0	0.0	0.0	14.5	39.7	0.0		52.7
42	0.0	20.2	0.0	6.7	44.3	0.0		52.5
44	0.0	0.0	0.0	32.2	0.0	0.0		49.5
46	0.0	0.0	0.0	59.1	0.0	0.0	7.6	53,2
48	0.0	0.0	0,0	42.9	0.0	0.0		52.4
50	0.0	0.0	0.0	84.7	0.0	0.0		53.8
52	0.0	0.0	0.0	67.8	0.0	0.0		5,3
54	0.0	0.0	0.0	93.1	0.0	0.0		67.4
56	0.0	0.0	0,0	91.4	0.0	0.0		63.0
58	0.0	0.0	0.0	91.9	0.0	0.0		55,3
60	0.0	0.0	0.0	89.6	0.0	0.0		52,4
62	0.0	0.0	0.0	90.5	0.0	0,0		52.6
64	0.0	0.0	9.0	87.9	0.0	0.0		
66	0.0	0.0	0.0	90.1	0.0	0.0		54.6
68	0.0	0.0	0.0	90.4	0.0	0.0		53.7
70	0.0	0.0	0.0	91.7	0.0	0.0		51.2
72	0.0	0.0	0.0	94.0	0.0	0.0		54.6
74	0.0	0.0	0.0	91.2	0,0	0.0		55,9
76	0.0	0.0	0.0	93.0	0.0	0.0		54.6
78	0.0	0.0	0.0	93.7	0.0	0,0		5.8
80	0.0	0.0	0.0	91.2	0.0	0.0	18.5	53.3
82	0.0	0.0	0.0	95.3	0.0	0.0		56.6
84	0.0	0.0	0.0	92.9	0.0	0.0		51.4
86	0.0	0.0	0.0	95.6	0.0	0.0		51,4
88	0.0	0.0	0.0	92.8	0.0	0.0		

Mineralogical Composition of North Core at North Ingebright Lake

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(cm) 90 92 93 92 94 92 92 93 92 93 93 93 93 93 93 93 93 93 93 93 93 93	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	91.3 96.8 94.4		ment të dhamanit i e na 🕳	Mg Calcite	Duckadalandar
90 90 94 92 94 96 96 96 98 96 98 98 99 98 90 90 91 98 98 98 99 98 90 90 91 90 91 90 91 90 91 90 91 90 91 90 92 90 93 90 90 90 91 90 120 90 121 90 122 90 123 90 134 90 155 90 156 90 157 90 158 90 150 90 150 90	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	91.3 96.8 94.4				F'rotogotomic
92 94 95 96 96 96 96 96 96 90 111 112 112 112 112 112 112 112 112 11	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	96.8 94.4	0.0	0.0		613
 94 96 98 98 98 98 99 99 99 90 	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0 0.0 0 0.0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	94,4	0.0	0.0		1.10
96 98 98 100 106 108 108 111 112 1112 1120 1120 1120 112	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0 0.0 0 0 0 0 0		0.0	0.0		
 98 106 106 108 108 108 109 109 114 118 111 <	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	0.0 0.0 0.0 0.0 0 0 0 0 0 0 0 0 0 0 0 0	51.3	0.0	0.0		0 Y
100 0.0 106 0.0 108 0.0 118 0.0 119 0.0 111 0.0 111 0.0 111 0.0 111 0.0 111 0.0 111 0.0 111 0.0 112 0.0 120 0.0 121 0.0 132 0.0 133 0.0 134 0.0 135 0.0 136 0.0 157 0.0 158 0.0 150 0.0 151 0.0 152 0.0 153 0.0 154 0.0 155 0.0 156 0.0 157 0.0 158 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0.0 0.0 0.0 0.0 0.0 0 0 0 0 0 0 0 0 0 0	95.1	0.0	0.0		0'r
106 0.0 108 0.0 110 0.0 111 0.0 112 0.0 113 0.0 114 0.0 118 0.0 118 0.0 118 0.0 120 0.0 131 0.0 132 0.0 133 0.0 134 0.0 135 0.0 136 0.0 137 0.0 138 0.0 139 0.0 136 0.0 157 0.0 158 0.0 150 0.0 151 0.0 152 0.0 153 0.0 154 0.0 150 0.0 151 0.0 152 0.0 153 0.0 154 0.0 155 0.0 150 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0.0 0.0 0.0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	95.3	0.0	00).cr
108 0.0 110 0.0 111 0.0 112 0.0 113 0.0 114 0.0 126 0.0 131 0.0 132 0.0 133 0.0 134 0.0 135 0.0 136 0.0 137 0.0 138 0.0 139 0.0 136 0.0 137 0.0 138 0.0 139 0.0 138 0.0 138 0.0 150 0.0 151 0.0 152 0.0 153 0.0 154 0.0 155 0.0 150 0.0 151 0.0 152 0.0 153 0.0 154 0.0 155 0.0 150 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0 0 0 0 0 0	0.0 0.0 0.0 0.0 0.0 0 0 0 0	92.6	0.0	00		0'IC
110 0.0 112 0.0 114 0.0 115 0.0 116 0.0 117 0.0 118 0.0 118 0.0 120 0.0 121 0.0 122 0.0 123 0.0 134 0.0 135 0.0 136 0.0 137 0.0 138 0.0 139 0.0 136 0.0 137 0.0 138 0.0 139 0.0 150 0.0 151 0.0 152 0.0 153 0.0 154 0.0 155 0.0 150 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0 0 0 0 0 0 0 0	0.0 0.0 0.0 0.0 0 0 0 0	97.6	0.0	000		
112 0.0 114 0.0 116 0.0 118 0.0 118 0.0 120 0.0 121 0.0 122 0.0 123 0.0 124 0.0 135 0.0 136 0.0 137 0.0 138 0.0 139 0.0 136 0.0 137 0.0 138 0.0 139 0.0 138 0.0 139 0.0 150 0.0 151 0.0 152 0.0 153 0.0 154 0.0 156 0.0 157 0.0	0.0 0.0 0.0 0.0 0.0	0.0 0.0 0.0 0.0	95.3	0.0	0.0		
114 0.0 116 0.0 120 0.0 121 0.0 122 0.0 123 0.0 124 0.0 137 0.0 138 0.0 139 0.0 131 0.0 132 0.0 133 0.0 134 0.0 135 0.0 140 0.0 151 0.0 152 0.0 153 0.0 154 0.0 155 0.0 156 0.0 157 0.0	0.0 0.0 0.0 0.0	0.0 0.0 0.0	99.5	0.0	0.0		
116 0.0 120 0.0 122 0.0 123 0.0 124 0.0 128 0.0 134 0.0 138 0.0 138 0.0 138 0.0 138 0.0 138 0.0 138 0.0 138 0.0 140 0.0 151 0.0 152 0.0 153 0.0 154 0.0 156 0.0 157 0.0 158 0.0 150 0.0	0.0 0.0 0.0	0.0 0.0	7.67	18.4	0.0		
118 0.0 120 0.0 122 0.0 124 0.0 128 0.0 129 0.0 129 0.0 130 0.0 131 0.0 132 0.0 133 0.0 134 0.0 135 0.0 146 0.0 147 0.0 156 0.0 157 0.0 158 0.0 158 0.0 158 0.0 158 0.0 158 0.0	0.0	0.0	96.7	0.0	00		10.5
120 122 122 124 126 128 128 139 130 130 132 132 130 132 132 132 132 132 144 142 133 144 142 142 136 144 142 136 144 142 150 156 156 156 156 156 156 156 156 156 156	0.0	0.0	99.5	0.0	00		C. 7.4
122 0.0 124 0.0 126 0.0 128 0.0 133 0.0 134 0.0 136 0.0 144 0.0 146 0.0 148 0.0 148 0.0 148 0.0 156 0.0 156 0.0 156 0.0 156 0.0	00		99.4	0.0	0.0		
124 0.0 126 0.0 128 0.0 130 0.0 132 0.0 134 0.0 136 0.0 144 0.0 144 0.0 148 0.0 148 0.0 148 0.0 158 0.0 158 0.0 158 0.0 158 0.0	2.2	0.0	100.0	0.0	0.0		
126 0.0 128 0.0 130 0.0 134 0.0 135 0.0 136 0.0 144 0.0 148 0.0 148 0.0 148 0.0 150 0.0 156 0.0 156 0.0 156 0.0 156 0.0	0.0	0.0	0.0	0.0	0.0		
128 0.0 130 0.0 132 0.0 134 0.0 136 0.0 142 0.0 148 0.0 148 0.0 148 0.0 150 0.0 156 0.0 156 0.0 158 0.0 158 0.0	0.0	0.0	1,99	0.0	0.0		
130 0.0 134 0.0 135 0.0 136 0.0 137 0.0 138 0.0 149 0.0 146 0.0 147 0.0 150 0.0 151 148 152 0.0 153 0.0 154 0.0 155 0.0 156 0.0 157 0.0 158 0.0 150 0.0	0.0	0.0	7.79	0.0	0.0		
132 0.0 134 0.0 136 0.0 137 0.0 138 0.0 149 0.0 140 0.0 141 0.0 142 0.0 148 0.0 150 0.0 152 0.0 153 0.0 154 0.0 156 0.0 157 0.0 158 0.0 159 0.0	0.0	0.0	97.8	0,0	0.0		5 5
134 0.0 136 0.0 138 0.0 140 0.0 142 0.0 146 0.0 150 0.0 156 0.0 158 0.0 158 0.0 158 0.0 158 0.0	11.4	0.0	87.4	0.0	0.0		2
136 0.0 138 0.0 140 0.0 142 0.0 146 0.0 150 0.0 156 0.0 158 0.0 158 0.0 158 0.0 158 0.0	0.0	0.0	95.5	0.0	0.0		515
138 0.0 140 0.0 142 0.0 146 0.0 150 0.0 158 0.0 158 0.0 158 0.0 158 0.0 158 0.0	10.7	0,0	88.7	0.0	0.0		2
140 0.0 142 0.0 146 0.0 148 0.0 150 0.0 152 0.0 154 0.0 158 0.0 158 0.0 158 0.0	0.0	0.0	96.0	0.0	0.0		57.0
142 0.0 144 0.0 146 0.0 148 0.0 150 0.0 151 0.0 152 0.0 153 0.0 154 0.0 152 0.0 153 0.0 154 0.0 156 0.0 158 0.0 158 0.0 150 0.0	0.0	0.0	94.8	0.0	0.0		517
144 0.0 146 0.0 148 0.0 150 0.0 152 0.0 154 0.0 158 0.0 158 0.0	0.0	0.0	94.5	0.0	0.0		55.0
146 0.0 148 0.0 150 0.0 154 0.0 156 0.0 158 0.0 158 0.0	0.0	0.0	100.0	0.0	0.0		
148 0.0 150 0.0 152 0.0 154 0.0 156 0.0 158 0.0 160 0.0	0.0	0.0	72.3	0.0	0.0		52.0
150 0.0 152 0.0 154 0.0 156 0.0 158 0.0 160 0.0	0.0	0.0	80.5	0.0	0.0		5. A
152 0.0 154 0.0 156 0.0 158 0.0 160 0.0	0.0	0.0	78.0	0.0	0.0		F:20
154 0.0 156 0.0 158 0.0 160 0.0	0.0	0.0	64.8	0.0	0.0	17.3	1.20
156 0.0 158 0.0 160 0.0	0.0	0.0	96.1	0.0	00		0 U V
158 0.0 160 0.0	0.0	0.0	5.00	0.0	0.0		+,6+
160 0.0	0.0	0.0	1.66	0.0	00		
	0.0	0.0	95.6	0.0	0.0		
162 0.0	0.0	0.0	99.4	00			
164 0.0	0.0	0.0	98.6	0.0	0.0		
166 0.0	0.0	6.3	92.1	0.0	0.0		
168 0.0	0.0	0.0	100.0	0.0	0.0		
170 0.0	0.0	0.0	82.0	13.7	0.0		50.8

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Mineralogical Composition of North Core at North Ingebright Lake

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172 176 176 178 188 188 192 196 196 196 196 200 200 200	0.0				Dioeulie	LIAIILC	Mg in Hign Mg Calcite	Protodolomit
174 176 176 188 188 192 196 196 196 202 202 202 206		0.0	0.0	98.1	0.0	0.0		
176 178 188 188 190 196 196 198 200 202 202 202	0.0	0.0	0.0	100.0	0.0	0.0		
178 180 182 184 192 196 196 198 202 202 202 202	0.0	0.0	0.0	98.2	. 0.0	0.0	18.8	
180 182 184 192 196 196 198 202 202 202 204	0.0	6.4	0.0	0.16	0.0	0'0		69.2
182 184 198 198 198 198 202 202 202 202 206	0.0	0.0	0'0	94.7	0.0	0.0		52.9
184 190 196 196 198 202 202 202 206	0.0	0.0	0.0	77.5	17.2	0.0		61.9
188 190 194 196 200 202 202 204	0.0	7.4	0.0	91.6	0.0	0.0		
190 192 194 196 202 202 204 208	0.0	7.1	0.0	89,0	0.0	0.0		5.8
192 194 196 200 204 204 206	0.0	19.8	0.0	78.5	0.0	0.0		
194 196 200 204 204 206	0.0	6.3	0.0	59.4	0.0	0.0		59.9
196 198 200 204 204 206	0.0	0.0	0.0	96.9	0.0	0.0		64.6
198 200 202 204 206	0.0	10.7	0.0	81.8	0'0	0.0		
200 202 204 206	0.0	11.3	0.0	80.4	0.0	0.0		56.5
202 204 206	0.0	10.5	0.0	65,3	0.0	0.0		
204 206 208	0.0	11.4	0.0	81.1	0.0	0'0		61.7
206 208	0'0	11.4	0.0	81.1	0.0	0.0		61.7
208	0.0	17.8	0.0	59.4	0.0	0.0		62.2
500	0.0	0.11	0.0	54.8	0.0	0.0		49.6
210	0.0	15,1	0.0	77.1	0.0	0.0	15.6	
212	0.0	26.1	0.0	69.2	0.0	0.0		61.8
214	0.0	12.8	0.0	57.5	0.0	0.0		6.8
216	0.0	8.6	0.0	84.2	0.0	0.0		52.4
218	0.0	7.6	0.0	56.9	0.0	0.0		62.7
220	0.0	13.2	0.0	79.4	0.0	0.0		6.6
222	2.9	1.5.1	0.0	51.4	0.0	0.0		63.2
224	2.6	12.2	0.0	48.1	0.0	0.0		62.9
226	0.0	0,0	0.0	89.7	0.0	0.0		63.3
228	0.0	26.8	0.0	56.8	0.0	0.0		65.2
230	0.0	0.0	0.0	86.9	0.0	0.0		
232	0.0	8.6	0.0	66.6	0.0	0.0		63.9
234	0.0	10.9	0.0	62.5	0.0	0'0		74.9
236	0.0	9.8	0.0	60.1	0.0	0'0		
238	0.0	26.4	0.0	66.4	0.0	0.0		
240	0.0	14.8	0.0	61.5	0.0	0.0		64.3
242	0.0	10.7	0.0	69.3	0.0	0.0	14.4	59.9
244	0.0	15.8	0.0	75.6	0.0	0.0	14.1	
246	0.0	15,1	0.0	55.3	0'0	0.0		
248	0.0	6.5	0.0	56.3	0'0	0.0	11.6	63.6
250	0.0	24.7	0.0	63.2	0.0	0.0	1.6	64.2

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Quad All condition All condition <th>Depth</th> <th>Magnesite</th> <th>Gypsum</th> <th>Anhydrite</th> <th>Thenardite</th> <th>Bloedite</th> <th>Halite</th> <th>Mg in High</th> <th>Ca in</th>	Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	(cm)							Mg Calcite	Protodolomite
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	252	0.0	24.7	0.0	63.2	0.0	00	1 8	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	254	0.0	0.0	0.0	54.8	00	0.0		2,40
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	256	0.0	19.8	0.0	51.3	0.0	00	1.21	7.41
	258	0.0	9.4	0.0	57.5	00	00	1.21	C.CC
	260	0.0	5.5	0.0	58.1	0.0	0.0	911	0.60
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	262	0.0	13.9	0.0	41.5	0.0		011	7:40
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	264	0.0	12.3	0.0	45.9	00	0.0	0.5	5.70
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	266	0.0	12.7	0.0	55.9	0.0	0.0	<i>v</i> .c	/70
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	268	0.0	9.0	0.0	52.7	0.0	0.0		00.0
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	270	0.0	5.2	0.0	47.9	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	272	0.0	23.6	0.0	45.3	0.0	0.0	-	0.40
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	274	0.0	18.9	0.0	64.1	0.0	0.0	6 1	6.cu
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	276	0.0	6.4	0.0	45.1	0.0	0.0	2	1.130
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	278	0.0	0.0	0.0	61.3	0.0	0.0	13.4	1.00
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	280	0.0	21.4	0.0	62.9	0.0	0.0	13.7	619
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	282	0.0	0.0	0.0	53.4	0.0	0.0	12.4	64.8
286 0.0 8.5 0.0 8.19 0.0 0.12 0.28 0.0 1.28 0.0 1.28 0.0 1.28 0.0 1.28 0.0 1.28 0.0 1.28 0.0 1.28 0.0 1.28 0.0 1.12 0.0 1.12 0.0 1.12 0.0 1.12 0.0 1.12 0.13 0.0 0.0 1.12 0.13 0.0 0.0 0.0 0.0 0.0 0.12 0.28 0.0 0.0 0.0 0.13 0.0 </td <td>284</td> <td>0.0</td> <td>9.4</td> <td>0.0</td> <td>61.3</td> <td>0.0</td> <td>0.0</td> <td>15,3</td> <td>6.8</td>	284	0.0	9.4	0.0	61.3	0.0	0.0	15,3	6.8
	286	0.0	8.5	0.0	83,9	0.0	0.0	12.8	2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	288	0.0	0.0	0.0	0.0	0.0	0.0	11.2	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	290	0.0	7.3	0.0	87.5	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	767	0.0	0.0	0.0	93.3	0.0	0.0	12.2	62.3
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	294	0.0	0.0	0.0	93.5	0.0	0.0		
298 0.0 3.8 0.0 93.4 0.0 0.0 13.8 300 0.0 0.0 0.0 0.0 0.0 0.0 14.6 52.0 302 0.0 19.3 0.0 77.2 0.0 0.0 14.6 52.0 304 0.0 18.9 0.0 77.2 0.0 0.0 14.8 51.9 306 0.0 18.9 0.0 77.2 0.0 0.0 14.8 51.5 306 0.0 18.9 0.0 90.7 0.0 0.0 14.8 51.5 310 0.0 18.7 0.0 90.7 0.0 0.0 14.4 51.5 312 0.0 0.0 0.0 0.0 0.0 14.6 51.4 316 0.0 0.0 0.0 0.0 0.0 14.6 51.4 318 0.0 0.0 0.0 0.0 0.0 14.6 51.4	296	0.0	0.0	0.0	89.5	0.0	0.0	5.0	52.8
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	298	0.0	3.8	0'0	93.4	0.0	0.0	13.8	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	300	0.0	0.0	0.0	66.1	0.0	0.0	14.6	52.0
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	302	0.0	19.3	0.0	73.9	0.0	0.0	14.8	619
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	304	0.0	18.9	0'0	77.2	0.0	0'0		61.5
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	306	0.0	14.7	0.0	83.0	0.0	0.0	1.4	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	308	0.0	8.2	0.0	90.7	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	310	0.0	6.7	0'0	69,0	0.0	0.0	4.6	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	312	0.0	0.0	0.0	97.0	0.0	0.0	1	61.4
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	314	0.0	0.0	0.0	98.8	0.0	0.0		
318 0.0 0.1 <td>316</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>89.7</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	316	0.0	0.0	0.0	89.7	0.0	0.0		
320 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.1 61.7	318	0,0	0.0	0.0	97.5	0.0	0.0		
322 0.0 0.0 0.0 75.8 0.0 0.0 51.7 324 0.0 0.0 0.0 75.8 0.0 0.0 51.7 324 0.0 0.0 0.0 93.5 0.0 0.0 31.7 326 0.0 0.0 94.9 0.0 0.0 0.0 51.7 328 0.0 0.0 0.0 97.5 0.0 0.0 6.2	320	0.0	0'0	0.0	1.66	0.0	0'0		
324 0.0 62.2 62.2	322	0.0	0.0	0.0	75.8	0.0	0.0		617
326 0.0 0.4 0.0 94.9 0.0 0.0 328 0.0 0.0 0.0 97.5 0.0 0.0 62.2	324	0.0	0.0	0.0	93.5	0.0	0.0		1.10
328 0.0 0.0 0.0 97.5 0.0 0.0 62.2	326	0.0	0.4	0.0	94.9	0.0	0.0		
	328	0.0	0.0	0.0	97.5	0.0	0.0		62.2

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Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
		:					Mg Calcite	Protodolomite
330	0.0	0.0	0.0	98.9	0.0	0.0		
332	0.0	0.0	0.0	98.0	0.0	00		9 63
334	0.0	24.2	0.0	74.5	0.0	0.0		0.20
336	0.0	0.0	0.0	97.8	0.0	0.0		0770 202
338	0.0	0.0	0.0	98.0	0.0	0.0		C''70
340	0.0	0.0	0.0	96.7	0.0	0.0		40 C
342	0.0	0.0	0.0	82.8	0.0	0.0		C.0r
344	0.0	0.0	0.0	88.2	0.0	0.0		
346	0.0	0.0	0.0	98,0	0.0	0.0		50 B
348	0.0	0.0	0.0	99.2	0.0	0.0		0.22
350	0.0	0.0	0.0	98.0	0.0	0.0		
352	0.0	0.0	0.0	98.9	0.0	0.0		
354	0.0	0.0	0.0	98.6	0.0	0.0		
356	0.0	0.0	0.0	97.1	0.0	0.0		66.7
358	0.0	0.0	0.0	98.4	0.0	0.0		
360	0.0	0.0	0.0	80.7	0.0	0.0		
362	0.0	0.0	0.0	96,9	0.0	0.0		64.7
364	0.0	0.0	0.0	85.9	0.0	0.0		64
368	0.0	0.0	0.0	96.6	0.0	0.0		
370	0.0	0.0	0.0	97.7	0.0	0.0		65.2
372	0.0	1.5.1	0.0	81.5	0.0	0.0		6.7
374	0.0	6.0	0.0	73.7	0.0	0.0		5
376	0.0	0.0	0.0	98.3	0.0	0.0		
378	0.0	0.0	0.0	93.9	0.0	0.0		
380	0.0	0.0	0.0	88.6	0.0	0.0		1 05
382	0.0	0.0	0.0	98.8	0.0	0.0		
384	0.0	0.0	0.0	97.0	0.0	0.0		9.6
386	0.0	0.0	0'0	99.1	0.0	0.0		2
388	0.0	0.0	0.0	99.4	0.0	0.0		
390	0.0	0.0	0.0	98.9	0.0	0.0		
392	0.0	0.0	0.0	99.2	0.0	0.0		
394	0.0	0.0	0.0	99.3	0.0	0.0		
396	0.0	5.3	0.0	82.8	0.0	0.0		
398	0.0	0.0	0.0	90.4	0.0	0.0		
400	0.0	0.0	0.0	90.6	0.0	0.0		
402	0.0	0.0	0.0	97.1	0.0	0.0		51.2
404	0.0	0.0	0.0	99.4	0.0	0.0		1
406	0.0	0.0	0.0	99.2	0.0	0.0		
408	0.0	0.0	0.0	99.3	0.0	0'0		

Mit Mit <th>Depth</th> <th>Magnesite</th> <th>Gypsum</th> <th>Anhydrite</th> <th>Thenardite</th> <th>Bloedite</th> <th>Halite</th> <th>Mg in High</th> <th>Ca in</th>	Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
	(uu)							Mg Calcite	Protodolomit
	410	0.0	0.0	0.0	99,4	0.0	0.0		
	412	0'0	0.0	0.0	9.66	0.0	0.0		
	414	0'0	0.0	0.0	84.8	0.0	0.0		
	416	0.0	0'0	0.0	99.3	0.0	0.0		
	418	0.0	0.0	0.0	99.0	0.0	0.0		
	420	0.0	0.0	0.0	99.2	0.0	00		
47 00 00 00 00 00 00 48 00 00 00 936 00 00 00 410 00 00 00 913 00 00 944 410 00 00 913 00 00 914 410 00 00 913 00 00 914 410 00 00 913 00 00 914 414 00 00 929 00 00 914 414 00 00 923 00 00 914 414 00 00 929 00 00 914 414 00 00 923 00 914 914 414 00 00 929 00 914 914 414 00 00 923 910 914 914 414 00 <td< td=""><td>422</td><td>0.0</td><td>5.0</td><td>0.0</td><td>93.5</td><td>0.0</td><td>0.0</td><td></td><td></td></td<>	422	0.0	5.0	0.0	93.5	0.0	0.0		
	424	0.0	0.0	0.0	100.0	0.0	0.0		
478 00 00 00 00 00 00 00 470 00 00 00 00 00 00 00 470 00 00 00 00 00 00 00 470 00 00 00 00 00 00 00 470 00 00 00 00 00 00 00 470 00 00 00 00 00 00 00 471 00 00 00 00 00 00 00 471 00 00 00 00 00 00 00 473 00 00 00 00 00 00 00 473 00 00 00 00 00 00 00 474 00 00 00 00 00 00 00 474 00 00	426	0.0	0.0	0.0	99.6	0.0	0.0		
	428	0'0	0.0	0.0	84.4	0.0	00		
	430	0.0	0.0	0.0	91.5	0.0	0.0		
	432	0.0	0.0	0.0	88.3	0.0	0.0		
	434	0.0	0.0	0.0	100.0	0.0	0.0		
	436	0.0	0.0	0.0	89.4	0.0	0.0		
	438	0.0	0.0	0.0	92.9	0.0	00		
	440	0.0	0.0	0.0	89.0	0.0	0.0		
	442	0.0	0.0	0.0	82.5	0.0	0.0		
	444	0.0	0.0	0.0	87.5	0.0	0.0		
	446	0.0	0.0	0.0	89.1	0.0	0.0		54.4
	448	0.0	0.0	0'0	88.8	0,0	0.0		
	450	0.0	0.0	0.0	99.4	0.0	0.0		
	452	0.0	0.0	0.0	89.0	0.0	0.0		
	454	0.0	7.3	0.0	83.0	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	456	0.0	0.0	0.0	85.5	0.0	0.0		
	458	0.0	6.7	0.0	84.2	0.0	0.0		
	460	0.0	0.0	0.0	70.5	0.0	0.0		67.0
	462	0.0	10.1	0.0	80.7	0.0	0.0	9.2	0.10
466 0.0 11.7 0.0 81.5 0.0 0.0 468 0.0 0.0 0.0 0.0 0.0 0.0 0.0 470 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 472 0.0	464	0.0	7.3	0.0	87.9	0.0	0.0		77 6
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	466	0.0	11.7	0.0	81.5	0.0	0.0		0.11
	468	0.0	0.0	0.0	99.5	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	470	0.0	0.0	0.0	98.7	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	472	0.0	8.3	0.0	83.7	0.0	0.0		8 6 9
476 0.0 33.1 0.0 45.2 0.0 0.0 6.4 478 3.2 6.5 0.0 65.3 0.0 0.0 6.4 478 3.2 6.5 0.0 65.3 0.0 0.0 6.4 480 2.5 0.0 0.0 65.3 0.0 0.0 6.3 482 0.0 11.4 0.0 76.5 0.0 0.0 6.3 484 0.0 11.2 0.0 88.7 0.0 0.0 6.5 486 0.0 0.0 0.0 0.0 6.0 6.5 6.5 486 0.0 0.0 0.0 0.0 6.0 6.5 6.5	474	0.0	0.0	0.0	74.0	0.0	0.0		0'70 VI 1
478 3.2 6.5 0.0 65.3 0.0 0.0 6.3 6.5 6.5 6.5 0.0 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.3 6.5 <	476	0.0	33.1	0.0	45.2	0.0	0.0		6.4
480 2.5 0.0 0.0 76.5 0.0 0.0 0.0 482 0.0 11.4 0.0 86.7 0.0 0.0 0.0 484 0.0 11.4 0.0 86.7 0.0 0.0 0.0 484 0.0 11.2 0.0 83.0 0.0 0.0 6.5 486 0.0 0.0 0.0 0.0 0.0 6.5	478	3.2	6.5	0.0	65.3	0.0	0.0		4 L Y
482 0.0 11.4 0.0 86.7 0.0 0.0 484 0.0 11.2 0.0 83.0 0.0 6.5 486 0.0 0.0 0.0 95.8 0.0 6.5 6.5 0.0 0.0 0.0 0.0 6.5 6.5	480	2.5	0.0	0.0	76.5	0.0	0.0		1.00
484 0.0 11.2 0.0 83.0 0.0 0.0 6.5 486 0.0 0.0 0.0 95.8 0.0 0.0 69.2	482	0.0	11.4	0.0	86.7	0.0	0.0		
486 0.0 0.0 0.0 95.8 0.0 0.0 6 <u>0</u>	484	0.0	11.2	0.0	83.0	0.0	0.0		6.5
	480	0.0	0.0	0.0	95.8	0.0	0.0		69.2

Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)	8		,				Ma Calcite	Protodolomite
(•)		• • •		4 i.i			mg calence	Trotouoionnice
488	0.0	33.6	0.0	62.7	0.0	0.0		63.4
490	0.0	8.6	0.0	86.3	0.0	0.0		
492	0.0	0.0	0.0	98.3	0.0	0.0		
494	0.0	7.8	0.0	87.4	0.0	0.0		67
496	0.0	5.9	0.0	92.7	0.0	0.0		011
498	0.0	20.3	0.0	78.9	0.0	0.0		
500	0.0	0.0	0.0	94.3	0.0	0.0	9.9	
502	0.0	32.4	0.0	66.1	0.0	0.0		
504	0.0	0.0	0.0	98.6	0.0	0.0		
506	0.0	0.0	0.0	94.5	0.0	0.0		6.7
508	0.0	22.8	0.0	73.3	0.0	0.0		62.9
510	0.0	12.7	0.0	84.0	0.0	0.0		5.6
512	0.0	9.1	0.0	52.0	0.0	0.0		62.9
514	0.0	15.9	0.0	77.2	0.0	0.0		52.8
516	0.0	9.1	0.0	52.0	0.0	0.0		62.9
518	0.0	10.1	0.0	49.1	0.0	0.0		62.5
520	0.0	11.9	0.0	58.4	0.0	0.0		62.6
522	0.0	16.0	0.0	79.7	0.0	0.0		61.7
526	0.0	8.5	0.0	90.4	0.0	0.0		
528	0.0	0.0	0.0	96.2	0.0	0.0		
532	0.0	0.0	0.0	88.5	0.0	0.0		
534	0.0	0.0	0.0	93.2	0.0	0.0		
536	0.0	0.0	0.0	68.4	0.0	0,0		64.9
538	0.0	0.0	0.0	91.0	0.0	0.0		
540	0.0	0.0	0.0	98.9	0.0	0.0		
542	0,0	0.0	0.0	95.6	0.0	0.0		65.7
544	0.0	0.0	0.0	70.8	0.0	0.0		58.7
546	0.0	7.1	0.0	91.9	0.0	0.0		
548	0.0	6.7	0.0	92.7	0.0	0.0		
550	0.0	0.0	0.0	66.4	0.0	0.0		62.5
552	0.0	8.2	0.0	86.7	0.0	0.0		6.6
554	3.1	6.0	0.0	69.1	0.0	0.0		
556	0.0	0.0	0.0	58.4	0.0	0.0		65.4
558	0.0	0.0	0.0	97.1	0.0	0.0		
560	0.0	0.0	0.0	91.7	0.0	0.0		62.4
564	2.6	0.0	0.0	45.2	0.0	0.0		61,9
566	0.0	0.0	0.0	52.1	0.0	0.0		63.7
568	0.0	15.4	0.0	52.5	0.0	0.0		62.1
570	0.0	3.9	0.0	50.4	0.0	0.0		

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Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)			· · · · · · · · · · · · · · · · · · ·				Mg Calcite	Protodolomite
572	0.0	4.5	0.0	48.5	0.0	0.0		6 9
574	0.0	7.0	0.0	43.7	00	0.0		
576	0.0	3.4	0.0	60.1	0.0	0.0		C'10
578	0.0	3.4	0.0	48.8	0.0	0.0		0 79
580	0.0	0.0	0.0	64.1	0.0	0.0		0.09
582	0.0	0.0	0.0	61.3	0.0	0.0		0,00
584	0.0	0.0	0.0	60.9	0.0	0.0		1,00
586	0.0	0.0	0.0	54.1	0.0	0.0		C.00
588	0.0	0.0	0.0	51.8	0.0	0.0		1 69
590	0.0	0.0	0.0	56.6	0.0	0.0		C:40
592	0.0	0.0	0.0	69,1	0.0	0.0		
594	0.0	9.8	0.0	51.9	0.0	0.0		
596	0.0	0.0	0.0	57.3	0.0	0.0		
598	4.2	6.3	0.0	34.5	0.0	0.0		627
600	3.3	10.0	0.0	61.4	0.0	0.0		67.0
602	0.0	6.2	0.0	41.4	0.0	0.0		61.0
604	5.1	0.0	0.0	31.5	0.0	0.0		65.4
909	3.6	7.4	0.0	49.1	0.0	0.0		75.3
608	4.0	4.4	0.0	55.8	0.0	0.0		6.3
616	0.0	5.7	0.0	58.7	0.0	0.0		
618	0.0	0.0	0.0	61.4	0.0	0.0		
620	0.0	0.0	0.0	65.2	0.0	0.0		64.2
622	0.0	7.5	0.0	55.6	0.0	0'0		1
624	0.0	5.0	0.0	50.9	0.0	0.0		61.3
626	0'0	0.0	0.0	53.0	0.0	0.0		
628	0'0	6.2	0.0	57.4	0.0	0.0		65.4
630	0'0	8.3	0.0	55.4	0.0	0.0		63.8
634	0.0	10.7	0.0	52.0	0.0	0.0		619
636	0.0	0.0	0.0	14.1	0.0	0.0		60.9
638	0.0	0.0	0.0	62.2	0.0	0.0		63.5
640	0.0	0.0	0.0	57.8	0.0	0.0		
642	0.0	0.0	0.0	49.8	0.0	0.0		63.2
644	0.0	0.0	0.0	56.9	0.0	0.0		
646	0.0	16.3	0.0	51.8	0.0	0.0		63.6
648	0.0	6.9	0.0	52.7	0.0	0.0		5.25
650	0.0	0.0	0.0	58.2	0.0	0.0		
652	0.0	6.7	0.0	49.7	0.0	0.0		61.8
654	0.0	9.1	0.0	54.1	0.0	00		1.12
656	0.0	0.0	0.0	62.4	0.0	0.0		1.40

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68 00 00 00 63 00 00 64 640 00 00 00 00 00 00 64 640 00 00 00 00 00 64 64 640 00 00 00 00 00 64 64 640 00 61 00 61 00 64 64 640 00 61 00 61 00 64 64 640 00 61 00 61 64 64 64 641 00 61 00 64 64 64 64 641 00 00 64 64 64 64 64 641 00 00 00 00 64 64 64 641 00 00 00 00 64 64 64 644 00 00<	Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
63 0.0						men men en anna anna anna anna anna anna			
600 000 <td>658</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>65.1</td> <td>0.0</td> <td>0.0</td> <td></td> <td>56.7</td>	658	0.0	0.0	0.0	65.1	0.0	0.0		56.7
662 0.0 444 0.0 948 0.0 0.0 643 664 0.0 644 0.0 973 0.0 0.0 643 664 0.0 643 0.0 973 0.0 0.0 643 673 0.0 643 0.0 973 0.0 0.0 643 674 0.0 0.0 533 0.0 0.0 643 674 0.0 10 533 0.0 0.0 643 673 0.0 10 533 0.0 0.0 643 674 0.0 10 533 0.0 0.0 643 674 0.0 10 10 933 0.0 0.0 643 674 0.0 11 0.0 943 0.0 0.0 643 674 0.0 0.0 11 0.0 0.0 643 674 0.0 0.0 11	660	0.0	0.0	0.0	92.5	0.0	0.0		64.9
664 0.0 0.0 0.29 0.0 <td>662</td> <td>0.0</td> <td>4,4</td> <td>0.0</td> <td>49.8</td> <td>0.0</td> <td>0.0</td> <td></td> <td>6.8</td>	662	0.0	4,4	0.0	49.8	0.0	0.0		6.8
66 0.0 6.4 0.0 7.79 0.0 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 0.0 0.0 0.0 6.4 0.0 0.0 0.0 0.0 6.4 0.0 6.4 0.0 0.0 0.0 0.0 6.4 0.0 0.0 0.0 0.0 6.4 0.0 0.0 0.0 0.0 6.4 0.0 0.0 0.0 0.0 6.4 0.0 0.0 0.0 0.0 6.4 0.0 <td>664</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>62.9</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	664	0.0	0.0	0.0	62.9	0.0	0.0		
608 00 61 00 57.5 01 00 65.7 77 00 00 42 00 57.5 01 00 57.5 76 00 00 56.3 00 55.3 00 00 55.3 76 00 7.6 00 55.3 00 00 55.3 76 00 12.6 00 55.3 00 00 55.3 68 00 00 74 00 93.3 00 65.3 68 00 00 72.4 00 93.3 00 65.3 69 00 72.4 00 93.3 00 66.3 65.3 90 00 72.4 00 93.3 00 66.3 67.3 90 00 00 93.4 00 90.3 66.3 67.3 90 00 00 93.4 00 90.3 67.3	666	0.0	6.4	0.0	57.9	0,0	0.0		
670 0.0 4.1 0.0 5.3 0.0 6.3 673 0.0 0.0 5.3 0.0 0.0 6.3 676 0.0 0.0 5.3 0.0 0.0 6.3 676 0.0 12.6 0.0 5.3 0.0 0.0 678 0.0 12.6 0.0 5.3 0.0 0.0 678 0.0 12.6 0.0 5.3 0.0 0.0 688 0.0 0.0 0.0 9.3 0.0 0.0 6.1 690 0.0 0.0 9.3 0.0 0.0 0.0 6.1 690 0.0 0.0 9.3 0.0 0.0 6.1 690 0.0 0.0 9.3 0.0 0.0 6.1 690 0.0 0.0 9.3 0.0 0.0 6.1 690 0.0 0.0 0.0 0.0 0.0 6.1 <tr< td=""><td>668</td><td>0.0</td><td>6,3</td><td>0.0</td><td>57.5</td><td>0.0</td><td>0.0</td><td></td><td></td></tr<>	668	0.0	6,3	0.0	57.5	0.0	0.0		
77 0.0 0.0 9.5 0.0 0.0 6.7 74 0.0 0.0 5.5 0.0 0.0 5.5 0.0 6.7 78 0.0 12.6 0.0 5.5 0.0 6.7 6.7 68 0.0 12.6 0.0 5.5 0.0 0.0 5.5 68 0.0 0.0 9.7 0.0 9.7 0.0 6.7 68 0.0 0.0 9.7 0.0 9.7 0.0 6.7 68 0.0 0.0 9.7 0.0 9.7 0.0 6.7 68 0.0 0.0 9.7 0.0 9.7 0.0 6.7 69 0.0 0.0 9.7 0.0 9.7 0.0 6.7 70 0.0 0.0 9.7 0.0 0.0 6.7 70 0.0 0.0 9.7 0.0 0.0 6.7 70 <	670	0.0	4.2	0.0	58.4	0.0	0.0		6.9
674 0.0 0.0 0.0 5.3 0.0 0.0 676 0.0 5.4 0.0 5.3 0.0 0.0 5.13 676 0.0 5.4 0.0 5.3 0.0 0.0 5.13 676 0.0 12.6 0.0 5.3 0.0 0.0 5.13 678 0.0 0.0 5.3 0.0 0.0 5.3 0.0 5.13 686 0.0 0.0 9.3 0.0 0.0 5.13 5.13 679 0.0 0.0 9.3 0.0 0.0 5.13 5.13 679 0.0 0.0 9.3 0.0 0.0 5.13 5.13 679 0.0 0.0 9.3 0.0 9.0 5.13 5.13 670 0.0 0.0 9.1 0.0 0.0 5.13 5.13 703 0.0 0.0 0.0 0.0 0.0 5.13	672	0.0	0.0	0.0	59.5	0.0	0.0		65.7
676 0.0 4.0 0.0 5.3 0.0 0.0 61.3 678 0.0 2.4 0.0 5.3 0.0 0.0 5.3 0.0 61.3 678 0.0 2.5 0.0 5.3 0.0 0.0 5.3 0.0 5.3	674	0.0	0.0	0.0	55.3	0.0	0.0		
078 0.0 5.5 0.0 9.3 0.0 0.0 5.5 0.0 <td>676</td> <td>0.0</td> <td>4.0</td> <td>0.0</td> <td>55.3</td> <td>0.0</td> <td>0.0</td> <td></td> <td>61.5</td>	676	0.0	4.0	0.0	55.3	0.0	0.0		61.5
600 0.0 12.6 0.0 83.7 0.0 0.0 61.3 648 0.0 0.0 0.0 0.0 0.0 0.0 0.0 688 0.0 0.0 0.0 0.0 0.0 0.0 0.0 688 0.0 7.24 0.0 0.0 0.0 0.0 0.0 690 0.0 7.24 0.0 0.0 0.0 0.0 0.0 690 0.0 7.1 0.0 0.0 0.0 0.0 0.0 691 0.0 0.0 0.0 0.0 0.0 0.0 67.0 692 0.0 0.0 0.0 0.0 0.0 0.0 67.0 701 0.0 0.0 0.0 0.0 0.0 0.0 67.0 702 0.0 0.0 0.0 0.0 0.0 67.0 67.0 703 0.0 0.0 0.0 0.0 0.0 67.0	678	0.0	5.6	0.0	58.5	0.0	0.0		65.5
62 0.0	680	0.0	12.6	0.0	83.7	0.0	0.0		61.3
684 0.0 <td>682</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>96.3</td> <td>0.0</td> <td>0.0</td> <td></td> <td>62.8</td>	682	0.0	0.0	0.0	96.3	0.0	0.0		62.8
686 0.0 <td>684</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>1.79</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	684	0.0	0.0	0.0	1.79	0.0	0.0		
68 0.0 7.2.4 0.0 96.3 0.0 0.0 6.4.7 690 0.0 1.6 0.0 98.9 0.0 0.0 64.7 691 0.0 1.1 0.0 98.1 0.0 0.0 64.7 692 0.0 0.0 0.0 97.1 0.0 0.0 64.7 693 0.0 0.0 0.0 97.1 0.0 0.0 64.7 693 0.0 0.0 0.0 97.1 0.0 0.0 64.7 703 0.0 0.0 0.0 97.1 0.0 0.0 64.7 704 0.0 0.0 0.0 97.6 0.0 0.0 64.7 704 0.0 0.0 97.4 0.0 97.6 0.0 64.7 704 0.0 0.0 0.0 97.6 0.0 0.0 64.7 704 0.0 0.0 0.0 0.0 0.0 64.7	686	0.0	0.0	0.0	93.4	0.0	0.0		
600 0.0 <td>688</td> <td>0.0</td> <td>72.4</td> <td>0.0</td> <td>26.3</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	688	0.0	72.4	0.0	26.3	0.0	0.0		
692 0.0 7.6 0.0 89.1 0.0 0.0 64.7<	690	0.0	0.0	0.0	98.9	0.0	0.0		
694 0.0 11.1 0.0 86.1 0.0 0.0 64.7 698 0.0 0.0 0.0 97.1 0.0 0.0 68.9 700 0.0 0.0 0.0 97.2 0.0 0.0 64.7 700 0.0 0.0 0.0 97.2 0.0 0.0 64.7 701 0.0 0.0 0.0 0.0 0.0 64.7 64.7 702 0.0 0.0 0.0 0.0 0.0 64.7 64.7 703 0.0 0.0 0.0 0.0 0.0 64.7 64.7 704 0.0 0.0 0.0 0.0 0.0 64.7 64.7 704 0.0 0.0 0.0 0.0 0.0 64.7 64.7 710 0.0 0.0 0.0 0.0 0.0 64.7 64.7 714 0.0 0.0 0.0 0.0 0.0 64.4 </td <td>692</td> <td>0.0</td> <td>7.6</td> <td>0.0</td> <td>89.1</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	692	0.0	7.6	0.0	89.1	0.0	0.0		
696 0.0 <td>694</td> <td>0.0</td> <td>11.1</td> <td>0.0</td> <td>86.1</td> <td>0.0</td> <td>0.0</td> <td></td> <td>64.7</td>	694	0.0	11.1	0.0	86.1	0.0	0.0		64.7
698 0.0 0.0 0.0 0.0 0.0 64.1 700 0.0 0.0 0.0 0.0 0.0 64.1 702 0.0 0.0 0.0 0.0 0.0 67.0 704 0.0 0.0 0.0 0.0 0.0 67.0 708 0.0 0.0 0.0 0.0 0.0 67.0 710 0.0 0.0 0.0 0.0 0.0 67.0 714 0.0 0.0 0.0 0.0 0.0 67.0 714 0.0 0.0 0.0 0.0 0.0 67.0 718 0.0 0.0 0.0 0.0 0.0 67.0 718 0.0 0.0 0.0 0.0 0.0 67.4 718 0.0 0.0 0.0 0.0 0.0 67.4 713 0.0 0.0 0.0 0.0 0.0 67.4 714 <t< td=""><td>696</td><td>0'0</td><td>0.0</td><td>0.0</td><td>97.1</td><td>0.0</td><td>0.0</td><td></td><td></td></t<>	696	0'0	0.0	0.0	97.1	0.0	0.0		
700 0.0	698	0.0	0.0	0.0	97.2	0.0	0.0		68.9
	700	0.0	0.0	0.0	92.0	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	702	0.0	0.0	0.0	0.0	0.0	0.0		64.7
706 0.0	704	0.0	0.0	0.0	72.7	0.0	0.0		67.0
$ 708 0.0 11.9 0.0 84.8 0.0 0.0 0.0 \\ 710 0.0 0.0 0.0 0.0 0.0 0.0 \\ 716 0.0 0.0 0.0 0.0 0.0 0.0 \\ 718 0.0 0.0 0.0 0.0 0.0 0.0 \\ 720 0.0 0.0 0.0 0.0 0.0 0.0 \\ 722 0.0 0.0 0.0 0.0 0.0 0.0 \\ 724 0.0 0.0 0.0 0.0 0.0 0.0 \\ 724 0.0 0.0 0.0 0.0 0.0 0.0 \\ 724 0.0 0.0 0.0 0.0 0.0 0.0 \\ 724 0.0 0.0 0.0 0.0 0.0 0.0 \\ 724 0.0 0.0 0.0 0.0 0.0 0.0 \\ 724 0.0 0.0 0.0 0.0 0.0 0.0 \\ 724 0.0 0.0 0.0 0.0 0.0 0.0 \\ 728 0.0 0.0 0.0 0.0 0.0 0.0 0.0 \\ 738 3.6 19.9 0.0 0.0 0.0 0.0 0.0 0.0 \\ 738 3.6 19.9 0.0 0.0 0.0 0.0 0.0 0.0 \\ 736 0.0 7.6 0.0 0.0 0.0 0.0 0.0 0.0 \\ 736 0.0 7.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0 \\ 738 3.6 19.9 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 \\ 744 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 \\ 744 0.0 $	706	0.0	0.0	0.0	97.6	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	708	0.0	11.9	0.0	84.8	0.0	0.0		62.5
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	710	0.0	0.0	0.0	99.3	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	714	0.0	13.9	0.0	85.3	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	716	0.0	0.0	0.0	98.7	0.0	0.0		
720 0.0 74.5 72.5 0.0 0.0 0.0 0.0 0.0 0.0 74.5 74.4 0.0 0.0 0.0 0.0 0.0 74.5 73.7 73.0 0.0 0.0 0.0 68.7 74.5 74.4 0.0 0.0 68.7 74.5 74.5 74.6 0.0 0.0 68.7 74.5	718	0'0	16.5	0.0	80.6	0.0	0.0		64.4
722 0.0 6.0 0.0 90.6 0.0 0.0 54.5 724 0.0 0.0 0.0 0.0 0.0 54.5 728 0.0 0.0 0.0 0.0 0.0 54.5 730 0.0 0.0 0.0 0.0 0.0 68.7 731 0.0 12.2 0.0 61.4 0.0 0.0 68.7 732 0.0 7.8 0.0 62.2 0.0 0.0 63.8 734 0.0 7.4 0.0 60.0 53.2 734 0.0 0.0 74.2 0.0 0.0 53.2 736 19.9 0.0 70.5 0.0 0.0 53.2	720	0.0	0.0	0.0	98.9	0.0	0.0		
724 0.0 0.0 0.0 0.0 54.5 728 0.0 0.0 0.0 0.0 0.0 54.5 730 0.0 0.0 0.0 0.0 0.0 6.87 730 0.0 12.2 0.0 0.0 6.0 6.87 732 0.0 7.8 0.0 64.4 0.0 6.0 6.87 732 0.0 7.8 0.0 64.4 0.0 53.2 734 734 0.0 7.4 0.0 0.0 74.2 0.0 53.2 736 0.0 7.4 0.0 0.0 6.0 53.2 738 3.6 19.9 0.0 70.5 0.0 6.0	722	0'0	6.0	0.0	90.6	0.0	0.0		
728 0.0 0.0 0.0 0.0 0.0 6.8.7 730 0.0 0.0 0.0 0.0 0.0 68.7 732 0.0 0.0 64.4 0.0 0.0 68.7 732 0.0 7.8 0.0 67.2 0.0 0.0 63.8 734 0.0 7.8 0.0 74.2 0.0 0.0 53.2 736 0.0 7.4 0.0 0.0 70.5 0.0 6.4.3 738 3.6 19.9 0.0 70.5 0.0 0.0 6.4.3	724	0.0	0.0	0.0	74.4	0.0	0.0		54.5
730 0.0 12.2 0.0 64.4 0.0 0.0 68.7 732 0.0 7.8 0.0 62.2 0.0 0.0 63.8 732 0.0 7.8 0.0 62.2 0.0 0.0 63.8 734 0.0 0.0 74.2 0.0 0.0 53.2 736 0.0 7.6 0.0 87.6 0.0 0.0 53.2 738 3.6 19.9 0.0 70.5 0.0 0.0 64.3	728	0.0	0.0	0.0	93.3	0.0	0.0		
732 0.0 7.8 0.0 62.2 0.0 0.0 63.8 734 0.0 0.0 0.0 61.2 0.0 63.8 734 0.0 0.0 0.0 74.2 0.0 60.0 53.2 736 0.0 7.6 0.0 87.6 0.0 0.0 53.2 738 3.6 19.9 0.0 70.5 0.0 0.0 64.3	730	0.0	12.2	0.0	64.4	0.0	0.0		68.7
734 0.0 0.0 0.0 74.2 0.0 0.0 53.2 736 0.0 7.6 0.0 87.6 0.0 0.0 53.2 736 0.0 7.6 0.0 87.6 0.0 6.0 5.3.2 738 3.6 19.9 0.0 70.5 0.0 0.0 64.3	732	0.0	7.8	0'0	62.2	0.0	0.0		63.8
736 0.0 7.6 0.0 87.6 0.0 0.0 738 3.6 19.9 0.0 70.5 0.0 0.0 64.3	734	0.0	0.0	0.0	74.2	0.0	0.0		53.2
738 3.6 19.9 0.0 70.5 0.0 0.0 64.3	736	0.0	7.6	0.0	87.6	0.0	0.0		
	738	3.6	19.9	0.0	70.5	0.0	0.0		64.3

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Mineralogical Composition of North Core at North Ingebright Lake

235

(01) Mg Calcin Pag Calcin <th>Depth</th> <th>Magnesite</th> <th>Gypsum</th> <th>Anhydrite</th> <th>Thenardite</th> <th>Bloedite</th> <th>Halite</th> <th>Mg in High</th> <th>Ca in</th>	Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
10 0.0	(cm)			-		· · ·		Mg Calcite	Protodolomite
746 0.0 8.5 0.0 6.6 6.5 736 0.0 8.5 0.0 6.6 0.0 6.6 737 7.5 1.0 1.5 0.0 6.6 6.5 736 1.0 1.5 0.0 6.6 0.0 6.6 737 1.0 1.16 0.0 6.6 0.0 6.6 736 0.0 5.1 0.0 7.8 0.0 0.0 6.6 738 0.0 7.8 0.0 7.8 0.0 6.6 6.6 796 0.0 5.7 0.0 7.8 0.0 6.6 6.6 798 0.0 6.7 0.0 7.4 0.0 6.6 6.6 798 0.0 7.4 0.0 7.4 0.0 6.6 6.6 798 0.0 7.4 0.0 7.4 0.0 6.6 6.6 798 0.0 7.4 0.0 7.4 <td>740</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>97.8</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	740	0.0	0.0	0.0	97.8	0.0	0.0		
748 00 01 00 01	746	0.0	8.5	0.0	65.8	0.0	0.0		66.0
730 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 6.4 0.0 0.0 7.3 7.3 7.3 7.3 7.3 0.0 0.1 0.0 <td>748</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>96.7</td> <td>0.0</td> <td>00</td> <td></td> <td>7.CU</td>	748	0.0	0.0	0.0	96.7	0.0	00		7.CU
732 2.6 8.0 0.0 6.0 9.1 6.0 734 0.0 13.6 0.0 9.1 0.0 9.1 6.6 735 0.0 13.6 0.0 9.1 0.0 9.1 6.6 756 0.0 9.1 0.0 9.1 0.0 9.1 6.6 756 0.0 5.1 0.0 7.8 0.0 0.0 9.1 6.6 756 0.0 6.1 0.0 7.8 0.0 0.0 6.6 793 0.0 6.1 0.0 7.8 0.0 0.0 6.6 795 0.0 6.1 0.0 7.8 0.0 0.0 6.6 795 0.0 6.7 0.0 7.8 0.0 6.6 6.7 796 0.0 7.8 0.0 7.8 0.0 0.0 6.6 810 0.0 7.8 0.0 7.8 0.0 6.6 6.7<	750	0.0	6.2	0.0	64.4	0.0	0.0		0.00
	752	2.6	8.0	0.0	65.0	0.0	0.0	10	4°C0
	754	0.0	13.6	0.0	58.7	0.0	0.0	1'6	7.00
	756	4.2	0.0	0.0	90.6	0.0	0.0		0.00
	758	0.0	5.3	0.0	71.8		0.0		C./0
	760	6.2	0.0	0.0	1.00				02.8
	762	0.0	9.4	0.0	21.9	0.0	0.0		61.7
	764	0.0	6.2	0.0	88.3	0.0	0.0		01./
	792	0.0	6.5	0.0	90.8	00	0.0		0.63
	794	0.0	0'0	0.0	78.4	0.0			6770
	796	0.0	6.7	0.0	75.4	0.0	0.0		7 7 7
80 0.0 8.4 0.0 6.7 0.0 8.4 0.0 6.40 804 0.0 8.5 0.0 8.7 0.0 0.0 6.40 806 0.0 8.2 0.0 8.7 0.0 0.0 6.40 806 0.0 8.2 0.0 8.7 0.0 0.0 6.40 818 0.0 9.2 0.0 8.9 0.0 6.40 6.40 818 0.0 11.2 0.0 8.9 0.0 6.60 6.40 818 0.0 0.0 77.2 0.0 0.0 6.40 818 0.0 0.0 77.2 0.0 0.0 6.40 818 0.0 0.0 0.0 0.0 6.40 6.40 823 2.4 0.0 6.40 6.60 6.69 824 0.0 57.5 0.0 0.0 6.69 823 2.4 10.0 57.5	798	2.3	0.0	0.0	1.70	0.0	0.0		4'00
802 0.0 8.5 0.0 8.7 0.0 0.0 6.40 804 0.0 8.2 0.0 89.7 0.0 0.0 6.40 806 3.2 6.0 9.7 0.0 8.71 0.0 0.0 6.40 818 3.2 6.0 0.0 87.4 0.0 0.0 6.30 814 0.0 11.2 0.0 77.2 0.0 0.0 6.30 818 3.5 7.6 0.0 77.2 0.0 0.0 6.30 818 3.5 7.6 0.0 6.40 0.0 6.60 6.60 818 3.5 7.6 0.0 6.77 0.0 0.0 6.30 818 3.5 7.6 0.0 6.75 0.0 0.0 6.63 823 2.4 10.2 0.0 57.5 0.0 0.0 6.75 824 0.0 2.0 7.2 0.0 0.0 </td <td>800</td> <td>0.0</td> <td>8.4</td> <td>0.0</td> <td>67.9</td> <td>0.0</td> <td>00</td> <td></td> <td>613</td>	800	0.0	8.4	0.0	67.9	0.0	00		613
804 0.0 8.2 0.0 8.71 0.0 0.0 6.13 0.13	802	0.0	8.5	0.0	89.5	0.0	0.0		1.10
806 0.0 12.3 0.0 87.1 0.0 </td <td>804</td> <td>0.0</td> <td>8.2</td> <td>0.0</td> <td>89.7</td> <td>0.0</td> <td>0.0</td> <td></td> <td>0.40</td>	804	0.0	8.2	0.0	89.7	0.0	0.0		0.40
808 3.2 6.0 0.0 61.6 0.0 0.0 61.8 61.9 0.0 61.8 61.9 0.0 61.8 61.9 0.0 61.8 61.9 0.0 61.8 61.9 0.0 61.8 61.9 0.0 61.9 0.0 61.9 0.0 61.9 61.9 0.0 61.9	806	0.0	12.3	0.0	87.1	0.0	0.0		
810 0.0 9.2 0.0 <td>808</td> <td>3.2</td> <td>6.0</td> <td>0.0</td> <td>61,6</td> <td>0.0</td> <td>0.0</td> <td></td> <td>019</td>	808	3.2	6.0	0.0	61,6	0.0	0.0		019
812 0.0 <td>810</td> <td>0,0</td> <td>9.2</td> <td>0.0</td> <td>87,4</td> <td>0.0</td> <td>0.0</td> <td></td> <td>68.6</td>	810	0,0	9.2	0.0	87,4	0.0	0.0		68.6
814 0.0 11.2 0.0 68.9 0.0 0.0 11.2 0.0 68.9 0.0 0.0 11.2 0.0 0.0 11.2 0.0 0.0 11.2 0.0 0.0 11.2 0.0 0.0 11.2 0.0 <t< td=""><td>812</td><td>0.0</td><td>0.0</td><td>0.0</td><td>70.9</td><td>0.0</td><td>0.0</td><td></td><td>63.0</td></t<>	812	0.0	0.0	0.0	70.9	0.0	0.0		63.0
816 0.0 <td>814</td> <td>0.0</td> <td>11.2</td> <td>0.0</td> <td>68.9</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	814	0.0	11.2	0.0	68.9	0.0	0.0		
818 3.5 7.6 0.0 84.6 0.0 0.0 65.7 0.0 65.3 </td <td>816</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>77.2</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	816	0.0	0.0	0.0	77.2	0.0	0.0		
820 0.0 8.9 0.0 65.7 0.0 66.9 824 0.0 2.4 11.9 0.0 57.6 0.0 0.0 826 2.4 10.2 0.0 57.6 0.0 0.0 66.9 826 2.4 10.2 0.0 57.2 0.0 0.0 66.3 828 4.6 9.9 0.0 57.2 0.0 0.0 66.3 828 4.6 9.9 0.0 57.2 0.0 0.0 66.3 830 4.2 10.2 0.0 57.3 0.0 0.0 66.3 831 3.2 10.3 0.0 57.3 0.0 0.0 67.6 833 5.1 7.9 0.0 79.3 0.0 0.0 67.6 833 5.1 9.6 0.0 0.0 0.0 66.5 67.6 84 0.0 11.4 0.0 66.5 0.0 0.0 67	818	3.5	7.6	0.0	84.6	0.0	0.0		1 (9
822 5.2 31.9 0.0 57.6 0.0 0.0 68.7 824 0.0 20.2 0.0 57.2 0.0 0.0 68.7 826 2.4 10.2 0.0 57.2 0.0 0.0 66.3 828 2.4 10.2 0.0 57.2 0.0 0.0 65.3 828 2.4 10.2 0.0 57.2 0.0 0.0 65.3 830 4.5 9.9 0.0 51.5 0.0 0.0 65.3 831 3.2 7.9 0.0 79.3 0.0 0.0 65.6 838 5.1 9.6 0.0 60.0 0.0 65.6 65.6 844 3.2 7.9 0.0 66.5 0.0 0.0 67.6 845 5.1 9.6 0.0 0.0 0.0 66.7 67.6 844 3.6 12.5 0.0 66.7 0.0 <td< td=""><td>820</td><td>0.0</td><td>8.9</td><td>0.0</td><td>65.7</td><td>0.0</td><td>0.0</td><td></td><td>699</td></td<>	820	0.0	8.9	0.0	65.7	0.0	0.0		699
824 0.0 20.2 0.0 57.2 0.0 0.0 57.3 0.0 0.0 57.3 0.0 0.0 57.3 0.0 0.0 57.3 0.0 0.0 57.3 0.0 0.0 57.3 0.0 0.0 57.3 0.0 0.0 57.3 0.0 0.0 57.3 0.0 0.0 0.0 57.3 0.0	822	5.2	31.9	0.0	57.6	0.0	0.0		68.7
826 2.4 10.2 0.0 58.2 0.0 0.0 65.3 828 4.6 9.9 0.0 58.2 0.0 0.0 65.3 830 4.2 10.5 0.0 51.5 0.0 0.0 65.3 831 4.2 10.5 0.0 51.5 0.0 0.0 65.5 832 0.0 11.4 0.0 79.3 0.0 0.0 65.6 834 3.2 7.9 0.0 79.3 0.0 0.0 65.6 838 5.1 9.6 0.0 79.3 0.0 0.0 65.6 838 5.1 9.6 0.0 6.0 0.0 66.2 67.6 840 0.0 12.5 0.0 66.5 0.0 0.0 62.8 844 3.6 12.5 0.0 59.2 0.0 60.6 64.0 844 3.6 12.5 0.0 0.0 60.6 <t< td=""><td>824</td><td>0.0</td><td>20.2</td><td>0.0</td><td>57.2</td><td>0.0</td><td>0.0</td><td></td><td>60 K</td></t<>	824	0.0	20.2	0.0	57.2	0.0	0.0		60 K
828 4.6 9.9 0.0 53.8 0.0 <th< td=""><td>826</td><td>2.4</td><td>10.2</td><td>0.0</td><td>58.2</td><td>0.0</td><td>0.0</td><td></td><td>0.20 65 1</td></th<>	826	2.4	10.2	0.0	58.2	0.0	0.0		0.20 65 1
830 4.2 10.5 0.0 51.5 0.0 0.0 65.6 832 0.0 11.4 0.0 79.3 0.0 0.0 65.6 834 3.2 7.9 0.0 79.3 0.0 0.0 65.6 836 4.3 12.5 0.0 79.3 0.0 0.0 63.9 838 5.1 9.6 0.0 49.5 0.0 0.0 64.1 838 5.1 9.6 0.0 43.5 0.0 0.0 62.8 840 0.0 9.0 0.0 66.5 0.0 0.0 62.7 844 3.6 12.5 0.0 59.2 0.0 0.0 62.8 846 3.2 6.9 8.2 49.7 0.0 0.0 62.8 66.6 6.9 0.0 0.0 0.0 6.0 62.8 846 3.2 6.9 0.0 0.0 0.0 6.6 64.	828	4.6	9.9	0.0	53.8	0.0	0.0		67.69
832 0.0 11.4 0.0 79.3 0.0 0.0 62.8 834 3.2 7.9 0.0 79.3 0.0 0.0 63.9 836 4.3 12.5 0.0 56.9 0.0 0.0 64.1 838 5.1 9.6 0.0 49.5 0.0 0.0 64.1 838 5.1 9.6 0.0 43.5 0.0 0.0 64.1 840 0.0 9.0 0.0 66.5 0.0 0.0 62.8 844 3.6 12.5 0.0 59.2 0.0 0.0 62.8 846 3.2 6.9 8.2 49.7 0.0 0.0 66.6	830	4.2	10.5	0.0	51.5	0.0	0.0		0.10
834 3.2 7.9 0.0 56.9 0.0 0.0 0.0 836 4.3 12.5 0.0 56.9 0.0 0.0 63.9 838 5.1 9.6 0.0 49.5 0.0 0.0 64.1 838 5.1 9.6 0.0 49.5 0.0 0.0 64.1 840 0.0 9.0 0.0 66.5 0.0 0.0 62.8 844 3.6 12.5 0.0 51.9 0.0 0.0 66.2 64.0 846 3.2 6.9 8.2 49.7 0.0 0.0 60.0 62.8	832	0.0	11.4	0.0	79.3	0.0	0.0		8.09
836 4.3 12.5 0.0 49.5 0.0 64.1 838 5.1 9.6 0.0 49.5 0.0 64.1 838 5.1 9.6 0.0 43.5 0.0 0.0 64.1 840 0.0 9.0 0.0 66.5 0.0 0.0 62.8 842 0.0 10.3 0.0 59.2 0.0 0.0 62.8 844 3.6 12.5 0.0 51.9 0.0 6.0 62.8 846 3.2 6.9 8.2 49.7 0.0 0.0 62.8	834	3.2	7.9	0.0	56.9	0.0	0.0		0.20 61 Q
838 5.1 9.6 0.0 43.5 0.0 0.0 62.7 840 0.0 9.0 0.0 66.5 0.0 0.0 62.7 842 0.0 10.3 0.0 66.5 0.0 0.0 62.8 844 3.6 12.5 0.0 59.2 0.0 0.0 64.0 846 3.2 6.9 8.2 49.7 0.0 0.0 62.8	836	4.3	12.5	0.0	49.5	0.0	0.0		64.1
840 0.0 9.0 0.0 66.5 0.0 0.0 62.8 842 0.0 10.3 0.0 59.2 0.0 0.0 62.8 844 3.6 12.5 0.0 51.9 0.0 64.0 64.0 846 3.2 6.9 8.2 49.7 0.0 0.0 62.8	838	5.1	9.6	0.0	43.5	0.0	0.0		1.10
842 0.0 10.3 0.0 59.2 0.0 0.0 64.0 844 3.6 12.5 0.0 51.9 0.0 64.0 64.0 846 3.2 6.9 8.2 49.7 0.0 0.0 65.6	840	0.0	0'6	0.0	66.5	0.0	0.0		4.25 8 (y
844 3.6 12.5 0.0 51.9 0.0 0.0 62.8 846 3.2 6.9 8.2 49.7 0.0 0.0 66.6	842	0.0	10.3	0.0	59.2	0.0	0.0		640
846 3.2 6.9 8.2 49.7 0.0 0.0 6.6 66.6	844	3.6	12.5	0.0	51.9	0.0	0.0		8.03
	846	3.2	6.9	8.2	49.7	0.0	0.0		0.20 66.6

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ind)	MAGNESHE	Cypsull	Annyarite	Thenardite	Bloedite	Halite	Mg in High	Cain Cain
Cm)							Mg Calcite	Protodolomite
848	0.0	8.0	0.0	52.5	0.0	0.0		62.9
850	0.0	17.5	0'0	51.2	0.0	0.0		62.6
852	0.0	11.9	0.0	77.5	0.0	0.0		6.7
856	0.0	22.2	0.0	72.4	0.0	0.0		64.7
860	0.0	17.3	0.0	59.2	0.0	0.0		65.3
862	0.0	14.4	0.0	58.2	0.0	0.0		64.5
864	2.8	26.2	0.0	45.3	0.0	0.0		64.2
866	0.0	17.8	0.0	20.4	0.0	0.0		66.4
868	3.4	15.8	0.0	44.6	0'0	0.0		68.9
870	0.0	17.3	0.0	45.1	19.6	0.0		65,3
872	0.0	36.9	0.0	56.9	0'0	0.0		67.7
874	0.0	14.7	0.0	68.3	0.0	0.0		62.7
876	0.0	20.0	0.0	48.0	0.0	0.0		64.6
878	0.0	16.5	0.0	55.3	0.0	0.0		63.9
880	0.0	31.5	0.0	65,5	0.0	0.0		66.5
882	0.0	14.9	0,0	64.8	0.0	0.0		64.9
884	0.0	21.0	0.0	74.9	0.0	0.0		65.4
886	0.0	9.11	0.0	60.7	0.0	0.0		67,6
888	0.0	19.6	0.0	55.5	0.0	0.0		65.5
890	0.0	78.0	0.0	20,8	0.0	0.0		63,6
892	0.0	19.6	0'0	61.6	0.0	0.0		65,4
894	0.0	7.1	0.0	68.5	0.0	0.0		64.4
896	2.6	20.6	0.0	75.8	0.0	0.0		
898	1.5	51.9	0.0	30.6	0.0	0.0		65.6
900	2.3	9.3	0.0	63.8	0.0	0'0		64.6
902	0'0	17.4	0.0	56.5	0.0	0.0		63.9
904	0.0	20.3	0.0	57.9	0.0	0.0		62.7
906	0.0	23.9	0.0	68.2	0.0	0.0		62.5
908	3.5	20.6	0.0	72.8	0.0	0.0		63.7
016	0.0	18.0	0.0	65.2	0.0	0,0		67.3
912	0.0	15.7	0.0	61.4	0.0	0.0		66.5
914	5.3	0.0	0.0	45.2	0.0	0.0		64.6
916	4,4	. 17.0	0.0	35.6	0.0	0.0		64.4
918	0.0	0.0	0.0	52.6	0.0	0.0		65.2
920	0.0	0.0	0.0	77.8	0.0	0.0		
924	3.2	31.8	0.0	37.7	0.0	0.0		62.8
926	0.0	21.2	0.0	5.57	0.0	0.0		61.8
928	0.0	15.4	0.0	59.4	0.0	0.0		7 77
								0.00

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Mineralogical Composition of North Core at North Ingebright Lake

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Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)		•					Mg Calcite	Protodolomite
932	0.0	15.8	0.0	60.2	0.0	0.0		65.8
934	0.0	15.5	0.0	61.3	0.0	0.0		64.3
936	3.2	12.9	0.0	58.4	0.0	0.0		64.8
938	0.0	23.4	0.0	56.8	0.0	0.0		68.7
940	3.3	9.8	0.0	66.4	0.0	0.0		63.4
942	0.0	0.0	0.0	73.8	0.0	0.0		62.9
944	0.0	0.7	0.0	96.2	0.0	0.0		64.5
946	0.0	5.5	0.0	71.1	0.0	0.0		63.7
948	0.0	10.7	0.0	84.8	0.0	0.0		66.3
950	0.0	0.0	0.0	96.4	0.0	0.0		
952	0.0	0.0	0.0	74.8	0.0	0.0		64.1
954	0.0	5.3	0.0	69.5	0.0	0.0		63.3
956	0.0	8.0	0.0	69.4	0.0	0.0		65.2
958	0.0	0.0	0.0	72.2	0.0	0.0		67.1
960	0.0	10.9	0'0	84.0	0.0	0.0		6.9
962	0.0	24.9	0.0	74.3	0.0	0.0		
964	0.0	1.5.1	0.0	32.9	0.0	0.0		
972	0.0	22.1	0.0	59.6	0.0	0.0		62.8
974	0.0	46.0	0.0	41.2	0.0	0.0		
976	2.6	10.9	0.0	63.2	0.0	0.0		62,8
978	0.0	20.8	0.0	54.3	0.0	0.0		61.2
980	0.0	13.3	0'0	83.5	0.0	0.0		62.5
982	0.0	12.8	0.0	68.2	0.0	0.0		61.5
984	0.0	11.8	0.0	64.9	0.0	0.0		
986	3.1	11.3	0.0	60.7	0.0	0.0		64.3
988	0.0	10.4	0.0	87.1	0.0	0.0		64.4
066	3.7	6.6	0.0	70.9	0.0	0.0		63.4
992	0.0	29.8	0.0	54.3	0.0	0.0		63.9
994	2.7	24.0	0.0	47.8	0.0	0.0	4.7	64.4
906	2.8	45,9	0.0	27.6	0.0	0.0		62.0
998	2.7	43.2	0'0	33.9	0.0	0.0		66.4
000'1	4.3	13.2	0.0	44.4	0.0	0.0		63.9
1,002	0.0	8.6	0.0	66.0	0.0	0.0		65.8
1,004	0.0	13.4	0.0	64.1	0.0	0.0		6.8
1,006	0.0	9.5	0.0	84.3	0.0	0.0	6.0	
1,008	2.5	15.7	0.0	60.8	0.0	0.0		
1,010	0.0	0.0	0.0	93.8	0.0	0.0		65.6
1,012	3.2	9.8	0.0	82.3	0.0	3.6		
1,014	4.0	14.9	0.0	54.6	0.0	0.0		6. 0

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(70) Mg Callett Protodomit 1010 44 82 00 553 00 21 673 1021 24 82 00 533 00 00 563 1022 25 163 00 533 00 00 563 1028 21 163 00 513 00 00 563 1028 21 163 00 513 00 00 563 1028 21 163 00 513 00 00 513 1028 00 113 00 00 00 513 513 1041 13 00 133 00 00 513 513 1041 13 10 10 10 10 10 513 1041 13 10 10 00 00 513 513 1041 13 14 13 10	Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
	(cm)		•				· · · · · · · · · · · · · · · · · · ·	Mg Calcite	Protodolomite
	1,016	4.4	8.2	0.0	50,0	0.0	2.1		7 63
	1,018	4.3	0.0	0.0	62.9	0.0			1.20
	1,020	2.6	16.3	0.0	55.8	0.0	0.0		1.00
	1,022	0.0	6,6	0.0	71.8	0.0	0.0		1.20
108 27 94 00 611 00 00 613 00 00 613 00 613 00	1,024	3.0	15.5	0.0	57.0	0.0	0.0		0'10
108 00 160 00 517 00 00 613 103 313 914 00 322 00 00 613 1034 313 914 00 322 00 00 613 1034 00 114 00 323 00 00 613 1040 00 887 00 913 00 903 613 1044 19 57 00 913 00 00 613 1044 19 57 00 914 00 614 61 1044 13 114 00 711 00 00 61 1044 13 134 128 00 00 61 61 1055 01 134 128 00 00 61 61 1056 01 134 128 00 00 61 61 1056	1,026	2.7	9.4	0.0	63.1	0.0	00		0.50
	1,028	0.0	16,0	0.0	61.7	0.0	0.0		6,60
	1,030	3.2	17.4	0.0	52.2	0.0	0.0		01.5
1014 1.3 9.0 0.0 61.3 0.0 64.9 1086 0.0 11.6 0.0 88.7 0.0 0.0 64.9 1084 0.0 11.6 0.0 88.7 0.0 0.0 64.9 1084 1.9 5.7 0.0 9.8 0.0 0.0 64.6 1084 1.9 5.7 0.0 9.8 0.0 0.0 64.6 1084 1.1 1.1 0.0 71.1 0.0 0.0 64.7 1086 1.1 5.8 0.0 64.6 0.0 64.7 0.0 64.7 1086 1.1 5.8 0.0 64.7 0.0 0.0 64.7 1086 0.0 1.1 0.0 64.7 0.0 0.0 64.7 1086 0.0 1.1 0.0 64.7 0.0 0.0 64.7 1086 0.0 1.1 1.2 0.0 0.0	1,032	0.0	39.4	0.0	52.2	0.0	00		07.1
	1,034	3.3	9.0	0.0	61.3	0.0	00		0 19
	1,036	0.0	11.6	0.0	86.7	0.0	0.0		4.FU
	1,038	0.0	18.9	0.0	80.5	0.0	0.0		0.00
	i,040	0.0	8.2	0.0	91.8	0.0	0.0		
	1,042	0.0	18.8	0.0	79.8	0,0	0.0		
	1,044	1.9	5.7	0.0	89.7	0.0	0.0		
	1,046	1.4	11.4	0.0	1.17	0.0	0.0		
	1,048	3.2	8.5	0.0	64,6	0.0	0.0		60
	1,050	4.1	5.8	0.0	65,6	0.0	0.0		0'n
	1,052	0.0	12.0	0.0	83.9	0.0	0.0		1.20
	1,054	3.4	12.8	0.0	63.7	0.0	0.0		6 83
	1,056	3.4	13.7	0.0	63.3	0.0	0.0		40 K
	1,058	0.0	14.9	0.0	80,9	0.0	0.0		0.20 63.4
	1,060	0.0	16.5	0.0	79.6	0.0	0.0		6.7
1,064 0.0 $24,7$ 0.0 $72,6$ 0.0 0.0 0.0 $1,066$ 0.0 $79,9$ 0.0 $18,8$ 0.0 0.0 $6,7$ $1,070$ 0.0 $79,9$ 0.0 $66,7$ 0.0 0.0 $66,7$ $1,070$ 0.0 $23,6$ 0.0 $66,7$ 0.0 0.0 $66,7$ $1,071$ 0.0 $18,9$ 0.0 $66,7$ 0.0 $66,7$ $1,072$ 0.0 $18,9$ 0.0 $66,7$ $66,7$ $1,074$ 0.0 $18,5$ 0.0 $63,1$ 0.0 0.0 $1,074$ 0.0 $18,7$ 0.0 0.0 $66,7$ $1,074$ 0.0 $18,7$ 0.0 $66,7$ $66,7$ $1,078$ 0.0 $11,7$ 0.0 $85,1$ 0.0 0.0 $1,078$ 0.0 0.0 0.0 0.0 $66,7$ $1,084$ 0.0 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 0.0 $1,084$ 0.0 0.0 0.0 $0.$	1,062	0.0	9.5	0.0	82,4	0.0	0.0		50.8
	1,064	0.0	24.7	0.0	72.6	0.0	0.0		0.00
	1,066	0.0	79.9	0.0	18.8	0.0	0.0		67
	1,068	0.0	9.7	0.0	66.7	0.0	0.0		59.5
	1,070	0.0	23.6	0.0	64.6	0.0	0.0		6 69
	1,072	0.0	18,9	0.0	78.0	0.0	0.0		7170
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,074	0.0	15.2	0.0	9.09	0.0	0.0		1.10
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,076	0.0	18.5	0.0	63.1	0.0	0.0		513
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,078	0.0	11.7	0.0	85.1	0.0	0.0		513
1,082 0.0 20.2 0.0 65.7 0.0 0.0 62.3 1,084 0.0 0.0 0.0 94.6 0.0 0.0 62.3 1,086 4.2 6.5 0.0 94.6 0.0 0.0 62.3 1,086 4.2 6.5 0.0 78.9 0.0 0.0 65.2 1,086 4.2 6.5 0.0 78.1 0.0 66.2 65.2 1,086 11.4 0.0 78.1 0.0 0.0 63.5 63.5 1,092 0.0 13.2 0.0 83.7 0.0 0.0 68.2	1,080	0.0	<i>L</i> .6	0.0	87.2	0.0	0.0		67.7
1,084 0.0 0.0 0.0 0.0 0.0 62.3 1,086 4.2 6.5 0.0 58.9 0.0 0.0 65.2 1,088 0.0 33.0 0.0 58.9 0.0 0.0 65.2 1,088 0.0 33.0 0.0 78.1 0.0 63.5 1,090 4.8 11.4 0.0 78.1 0.0 60 1,092 0.0 13.2 0.0 83.7 0.0 0.0 68.2	1,082	0.0	20.2	0.0	65.7	0.0	0.0		
1,086 4.2 6.5 0.0 58.9 0.0 0.0 65.2 1,088 0.0 33.0 0.0 58.9 0.0 65.2 1,088 0.0 33.0 0.0 48.3 0.0 65.2 1,090 4.8 11.4 0.0 78.1 0.0 60 68.2 1,092 0.0 13.2 0.0 83.7 0.0 0.0 68.2	1,084	0.0	0.0	0.0	94.6	0.0	0.0		1 69
1,088 0.0 33.0 0.0 48.3 0.0 0.0 63.5 1,090 4.8 11.4 0.0 78.1 0.0 60 63.5 1,092 0.0 13.2 0.0 83.7 0.0 0.0 63.2	1,086	4.2	6.5	0.0	58.9	0.0	0.0		C 59
1,090 4.8 11.4 0.0 78.1 0.0 0.0 68.2 1,092 0.0 13.2 0.0 83.7 0.0 0.0 0.0	1,088	0.0	33.0	0.0	48.3	0.0	0.0		5.69
1,092 0.0 13.2 0.0 83.7 0.0 0.0 0.0	1,090	4.8	11.4	0.0	78.1	0.0	00		C 07
	1,092	0.0	13.2	0.0	83.7	0,0	0.0		790

Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)			•			-	Mg Calcite	Protodolomite
1,094	0.11	67.7	0.0	11.6	0.0	0.0		y Ly
1,096	0.0	10.1	0.0	88.5	0.0	0.0		0'CD
1,098	0.0	14.8	0.0	83.9	0.0	0.0		
1,100	0.0	0.0	0.0	98.9	0.0	0.0	4.8	
1,102	0.0	10.1	0.0	89.2	0.0	0.0	2	77.6
1,104	0.0	0.0	0.0	100.0	0.0	0.0		
1,106	0.0	10.7	0.0	86.8	0.0	0.0		9 V Y
1,108	0.0	14.9	0.0	80.6	0.0	3.2		0.40
1,110	0.0	0.0	0.0	90.2	0.0	0.0		
1,112	0.0	17.4	0.0	81.1	0.0	0.0		2 39
1,114	0.0	31.5	0.0	68.5	0.0	0.0		C*C0
1,116	0.0	46.6	0.0	51.6	0.0	0.0		
1,118	0.0	16.0	0.0	80.5	0.0	00		y y
1,120	0.0	10.8	0.0	87.0	0.0	0.0		0.0 A AA
1,122	0.0	36.9	0.0	62.8	0.0	0.0		
1,124	0.0	33.3	0.0	66.5	0.0	0.0		
1,126	0.0	16.9	0.0	83.1	0.0	0.0		
1,128	0.0	50.7	0.0	37.6	0.0	0.0		
1,130	2.8	17.2	0.0	60.9	0.0	0.0		
1,132	0.0	18.9	0.0	71.3	0.0	0.0		
1,134	0.0	62.6	0.0	34.6	0.0	0.0		
1,136	0.0	20.0	0.0	71.8	0.0	0.0		
1,138	0.0	11.5	0.0	88.1	0.0	0.0		
1,140	0.0	41.1	0.0	50.9	0.0	0.0		
1,142	0'0	16.0	0.0	82.4	0.0	0.0		
1,144	0.0	0.0	0.0	96.9	0.0	0.0		68.5
1,148	0.0	14.3	0.0	84.5	0.0	0.0		
1,150	0.0	10.0	0.0	77.0	0.0	0.0		
1,152	0.0	29.1	0.0	70.6	0.0	0.0		
1,154	0.0	41.1	0.0	57.0	0.0	0.0		L (Y
1,156	0.0	20.1	0.0	77.9	0.0	0.0		C 39
1,158	0.0	12.3	0.0	86.2	0.0	0.0		71/10
1,160	0.0	18.4	0.0	64.3	0.0	0.0		617
1,162	0.0	19.4	0.0	78.3	0.0	0.0		613
1,164	0.0	24.7	0.0	56.0	0.0	0.0		61.8
1,166	0.0	10.0	0.0	72.1	0.0	0.0	9.7	59.5
1,168	0.0	11.6	0.0	85.3	0.0	0.0		63.9
1,170	0'0	33.8	0.0	63.0	0.0	0.0		61.2
1,172	0.0	9.6	0.0	90.4	0.0	0.0		

Lake
Ingebright
North
Core at
North
Composition of
Mineralogical

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Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)	•			-			Mg Calcite	Protodolomite
1,174	3.1	11.8	0.0	57.6	0.0	0.0		62.2
1,176	0.0	27.9	0.0	64.9	0.0	0.0		61 K
1,178	0.0	17.6	0.0	82.4	0.0	0.0		
1,180	0.0	23.6	0.0	72.7	0.0	0.0		999
1,182	0.0	28.5	0.0	70.2	0.0	0.0		5
1,184	0.0	10.2	0.0	85.0	0.0	0.0		62.3
1,186	0.0	7.1	0.0	91,4	0.0	0.0		61.8
1,188	0.0	0.0	0.0	86.4	0.0	0.0		
1,190	0.0	8.2	0.0	91.3	0.0	0.0		
1,192	0.0	20.9	0.0	79.1	0.0	0.0		
1,194	0.0	0.0	0.0	98,9	0.0	0.0		
1,198	0'0	0.0	0.0	88.5	0.0	0.0		
1,200	0.0	0.0	0.0	92.5	0.0	0.0		
1,202	0.0	0.0	0.0	98.4	0.0	0.0		61.9
1,204	0.0	0.0	0.0	98.2	0.0	0.0		
1,206	0.0	5.2	0.0	75.1	0.0	0.0		
1,208	0.0	7.5	0.0	80.5	0.0	0.0		
1,210	1.8	0.0	0.0	87.3	0.0	0.0		
1,212	0.0	0.0	0.0	98.8	0.0	0.0		
1,214	0.0	0.0	0.0	100.0	0.0	0.0		
1,216	0.0	0.0	0.0	89.7	0.0	0.0		
1,218	0.0	0.0	0.0	89.2	0'0	0.0		
1,220	0.0	6.0	0.0	84.7	0.0	0.0		
1,222	0.0	0.0	0.0	85.6	0.0	0.0		
1,224	0.0	0.0	0.0	87.9	0.0	0.0		
1,326	0.0	0.0	0.0	73.6	0.0	0.0		62.8
1,328	0.0	0.0	0.0	86.2	0.0	0.0		62.0
1,330	0.0	9.0	0.0	77.8	0.0	0.0		6.6
1,332	0.0	0.0	0.0	89.5	0.0	0.0		58.7
1,334	0.0	0.0	0.0	87.0	0.0	0.0		
1,336	0.0	6.1	0.0	80.4	0.0	0.0		
1,338	1.5	0.0	0.0	59.2	0.0	0.0		
1,340	0.0	13.9	0.0	79.8	0.0	0.0	7.3	59.5
1,342	0.0	0.0	0.0	81.0	0.0	0.0		61.7
1,344	0.0	18.4	0.0	74.5	0.0	0.0		
1,346	0.0	34.7	0.0	57.9	0.0	0.0	7.2	
1,348	0.0	9.4	0.0	88.0	0.0	0.0		
1,350	0.0	0.0	0.0	87.1	0.0	0.0		53,2
1,352	0.0	0'0	0.0	79.2	0.0	0.0		56.4

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Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)							Mg Calcite	Protodolomite
1.354	0.0	12.5	0.0	79.6	0.0	0.0		57.3
1,356	4.1	8.8	0.0	81.2	0.0	0.0	3.8	01.0
1,358	0.0	16.6	0.0	40.3	21.2	0.0	14.3	52.6
1,360	0.0	22.3	0.0	51.8	19.4	0.0	1112	54.3
1,362	0.0	25.0	0.0	63.1	0.0	3.0		51.7
1,364	0.0	22.8	0.0	45.1	22.6	0.0		
1,366	0.0	9.0	0.0	82.2	0.0	0.0		
1,368	3.0	21.7	0.0	68.8	0.0	0.0		61.6
1,370	0.0	7.4	0.0	45.2	27.1	0.0		
1,372	0.0	9.2	0.0	50.0	0.0	3,3		63.8
1,374	0.0	13.2	0.0	81.3	0.0	0.0		55,4
1,376	4.5	7.6	0,0	47.3	0.0	0.0		65.9
1,380	4.1	7.2	0.0	54.9	0.0	0,0		66.0
1,382	4.7	10.1	0.0	48.1	0.0	0.0	11,4	75.9
1,384	3.0	4.9	0,0	50.5	0.0	0.0		61.6
1,386	6.9	8.7	0.0	72.6	0.0	0.0		62,6
1,388	4.8	6,1	0.0	39.9	0.0	0.0		65,9
1,390	4.0	10,8	0.0	54.1	0.0	0.0		64,9
1,392	9.7	16.0	0.0	65.3	0.0	0.0		69,8
1,394	5.9	5.8	0.0	41.8	0.0	0.0	6.4	61.5
1,396	3.1	7.2	0.0	65.1	0.0	0.0		65.5
1,398	3.4	0.0	0.0	75.9	0.0	0.0		63.5
1,400	3.2	13.8	0.0	61.0	0.0	0.0		6.5
1,402	0.0	6.1	0.0	72.3	0.0	0.0		64.6
1,404	3.0	5.4	0.0	60,8	0.0	0.0		64.7
1,406	5.1	12.0	0.0	46.6	0.0	0.0		63.5
1,408	0.0	0.0	0.0	99.5	0.0	0.0		
1,410	0.5	4.0	0,0	91.8	0.0	0.0		63.3
1,412	5.8	0,0	0.0	87.5	0.0	0,0		67.4
1,414	0.0	0.0	0.0	97.8	0,0	0.0		
1,416	0.0	0.0	0.0	100.0	0,0	0.0		
1,418	0.0	0.0	0.0	99.6	0.0	0.0		
1,420	0.0	0.0	0.0	100.0	0.0	0.0		
1,422	0.0	0.0	0.0	100.0	0.0	0.0		
1,424	0,0	0.0	0.0	100.0	0.0	0.0		
1,426	0.0	0.0	0.0	98.5	0.0	0.0		6.8
1,428	6.8	0.0	0.0	89.7	0.0	0.0		67.8
1,430	0.0	0.0	0.0	96.2	0.0	0.0		72.8
1,432	0.0	0.0	0.0	98.5	0.0	0.0		62.2

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Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
1 434	0.0	0.0	0.0	98.1	0.0	0.0		68 7
1,436	5.0	0.0	0.0	89.9	0.0	0.0		58 1
1 438	0.0	0.0	0.0	767	0.0	0.0		64
1 440	0.0	0.0	0.0	100.0	0.0	0.0		0,1
1 447	0.0	0.0	0.0	82.4	0.0	0.0		61.8
1.444	0.0	0.0	0.0	69.8	0.0	0.0		67.9
1.446	33	0.0	0.0	69.5	0.0	0.0	16.9	65.8
1.448	0.0	0.0	0.0	77.2	0.0	0.0	(0,)	61.6
1 450	0.0	0.0	0.0	67.3	0.0	0.0		62.2
1.452	3.0	0.0	0.0	74.9	0.0	0.0		62.7
1.454	0.0	0.0	0.0	97.1	0.0	0.0		62.9
1,456	0.0	0.0	0.0	77.8	0.0	0.0		61.4
1,458	0.0	4.9	0.0	80.9	0.0	0.0		64.4
1,460	0.0	0.0	0.0	97.8	0.0	0.0		66.7
1,462	0.0	0.0	0.0	80.4	0.0	0.0		59.0
1.464	0.0	0.0	0.0	100.0	0.0	0.0		
1,466	0.0	0.0	0.0	100.0	0.0	0.0		
1,468	4.4	0.0	0.0	63.3	0.0	0,0		58.9
1,470	0.0	0.0	0.0	99.4	0.0	0.0		
1,472	0.0	0.0	0.0	79.2	0.0	0.0		
1,474	0.0	0.0	0.0	99.5	0.0	0,0		
1,476	0.0	0.0	0.0	100.0	0.0	0,0		
1,478	2.0	0.0	0.0	98.0	0.0	0,0		
1,480	0.0	0.0	0.0	99.6	0.0	0,0		
1,482	4.8	0.0	0.0	73.6	0.0	0.0		63.4
1,484	2.7	0.0	0.0	78.3	0.0	0,0		65.5
1,486	3,9	0.0	0.0	92.0	0.0	0.0		6.3
1,488	0.0	4.2	0.0	91.7	0.0	0.0		64,6
1,490	0.0	0.0	0.0	100.0	0.0	0.0		
1,492	0.0	0.0	0.0	95,9	0.0	0.0	17.3	62.8
1,494	0.0	0.0	0.0	100.0	0.0	0.0		
1,496	0.0	0.0	0.0	98.2	0.0	0.0		65.7
1,498	0.0	0.0	0.0	74.9	0.0	0.0		61.7
1,500	6.7	0.0	0.0	87.4	0.0	0.0		53.7
1,504	0.0	0,0	0.0	95.2	0.0	0.0	6.9	61.7
1,506	0.0	5.8	0.0	91.3	0.0	0.0		63.6
1,508	0.0	0.0	0.0	98.6	0.0	0.0		61.7
1,510	0.0	0.0	0.0	98.8	0.0	0.0		63.6
1,512	4.3	0.0	0.0	68,1	0.0	0.0		62.3

Mineralogical Composition of North Core at North Ingebright Lake

	Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
	1,514	0.0	0.0	0.0	75.3	0.0	0.0		67.7
	1,516	0.0	0.0	0.0	96.4	0.0	0.0		59
	1,518	3.0	0.0	0.0	94.3	0.0	0'0		63.2
	1,520	0.0	0.0	0.0	1.79	0.0	0.0		
	1,522	0.0	24,4	0.0	75.2	0.0	0.0		
	1,524	0.0	0.0	0.0	100.0	0.0	0.0		
	1,526	0.0	0.0	0.0	90.9	0.0	0.0		
	1,528	0.0	4.2	0.0	66.6	0.0	0.0		619
	1,530	0.0	0'0	0.0	100.0	0.0	0.0		
	1,532	0.0	0'0	0.0	86.2	0.0	0.0		
	1,534	4.7	6.1	0.0	85.7	0.0	0.0		63.1
	1,536	0.0	0.0	0.0	86.4	0.0	0,0		62.1
	1,538	0.0	0.0	0.0	98.1	0.0	0.0		
	1,540	0.0	0.0	0.0	96.1	0.0	0.0		
	1,542	0.0	0.0	0.0	97.2	0.0	0.0		54.5
	1,544	0.0	6.0	0.0	94.0	0.0	0.0		
	1,546	0.0	0.0	0.0	81.8	0.0	0.0		
	1,548	0.0	0.0	0.0	96.5	0.0	0.0		63.4
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,550	0.0	0'0	0.0	98.5	0.0	0.0		62.9
	1,552	0.0	0.0	0.0	100.0	0.0	0.0		
1,556 $4,0$ $0,0$ $0,0$ $67,6$ $0,0$ $0,0$ $1,588$ $3,2$ $0,0$ $0,0$ $9,2$ $0,0$ $0,0$ $1,560$ $0,0$ $0,0$ $0,0$ $9,2$ $0,0$ $0,0$ $1,564$ $0,0$ $0,0$ $0,0$ $9,2$ $0,0$ $0,0$ $1,566$ $2,5$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,566$ $2,5$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,568$ $3,3$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,576$ $3,3$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,576$ $3,1$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,576$ $3,1$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,578$ $3,1$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,578$ $3,1$ $0,0$ $0,0$ $0,0$ $0,0$ $1,578$ $3,1$ $0,0$ $0,0$ $0,0$ $0,0$ $1,578$ $3,1$ $0,0$ $0,0$ $0,0$ $0,0$ $1,578$ $3,1$ $0,0$ $0,0$ $0,0$ $0,0$ $1,578$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,578$ $0,0$ $0,0$ $0,0$ $0,0$ $0,0$ $1,588$ $6,6$ $0,0$ $0,0$ $0,0$ $0,0$ $1,588$ $6,6$ $0,0$ $0,0$ $0,0$ $0,0$ $1,588$ $6,6$ $0,0$ $0,0$ $0,0$ <	1,554	4,2	0.0	0.0	65.5	0.0	0.0		62.9
	1,556	4.0	0.0	0.0	67.6	0.0	0.0		62.8
	1,558	3.2	0.0	0.0	92.6	0.0	0.0		61.8
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,560	0.0	0.0	0.0	99.2	0.0	0.0		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,562	3.3	0.0	0.0	93,0	0.0	0.0		62.2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,564	0.0	0.0	0.0	79.8	0.0	0.0		63.5
	1,566	2.5	0.0	0.0	74.0	0.0	0.0		62.7
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,568	3.3	0.0	0.0	65.8	0.0	0.0		6.7
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,570	0.0	0.0	0.0	77.0	0.0	0.0		64.4
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,572	4.0	0.0	0.0	65.3	0.0	0.0		63.2
	1,574	0.0	5.2	0.0	94.3	0.0	0.0		1
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1,576	3.1	0.0	0.0	78.6	0.0	0.0		63.4
1,580 0.0 0.0 0.0 0.0 0.0 0.0 1,582 0.0 0.0 0.0 0.0 0.0 0.0 0.0 1,582 0.0 0.0 0.0 0.0 0.0 0.0 0.0 1,586 4.2 0.0 0.0 0.0 72.6 0.0 0.0 1,588 6.6 0.0 0.0 0.0 9.0 0.0 1.0 1,588 0.0 0.0 0.0 0.0 9.7 0.0 0.0 1,590 0.0 0.0 0.0 9.0 97.4 0.0 0.0 1,592 0.0 0.0 0.0 9.0 95.1 0.0 0.0	1,578	3.1	0.0	0.0	69.69	0.0	0.0		6.5
1,582 0.0 0.0 0.0 0.0 0.0 0.0 1,586 4.2 0.0 0.0 0.0 72.6 0.0 0.0 1,588 6.6 0.0 0.0 0.0 72.6 0.0 0.0 1,588 6.6 0.0 0.0 0.0 9.0 1.0 1,590 0.0 0.0 0.0 97.4 0.0 0.0 1,592 0.0 0.0 0.0 95.1 0.0 0.0	1,580	0.0	0.0	0.0	100.0	0.0	0.0		
1,586 4.2 0.0 0.0 72.6 0.0 0.0 1,588 6.6 0.0 0.0 0.0 47.4 0.0 0.0 1,590 0.0 0.0 0.0 97.4 0.0 0.0 1,592 0.0 0.0 0.0 95.1 0.0 0.0	1,582	0.0	0.0	0.0	100.0	0.0	0.0		
1,588 6.6 0.0 0.0 47.4 0.0 0.0 1,590 0.0 0.0 0.0 0.0 0.0 0.0 1,592 0.0 0.0 0.0 0.0 0.0 0.0	1,586	4.2	0.0	0.0	72.6	0.0	0.0		63.4
1,590 0.0 0.0 0.0 0.0 97.4 0.0 0.0 1,592 0.0 0.0 0.0 95.1 0.0 0.0	1,588	6.6	0.0	0.0	47.4	0.0	0.0		62.7
1,592 0.0 0.0 0.0 9.5.1 0.0 0.0	1,590	0.0	0.0	0'0	97.4	0.0	0.0		66.5
	1,592	0.0	0.0	0'0	95.1	0.0	0.0		63.6

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Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)	Ū	•	•				Mg Calcite	Protodolomite
			· · · ·				IIIE Calence	
1,594	3.3	0.0	0.0	69.3	0.0	0.0		64.0
1,596	4.3	0.0	0.0	62.9	0.0	0.0		63.2
1,598	0.0	0.0	0.0	98.4	0.0	0.0		64.2
1,600	0.0	0.0	0,0	100.0	0.0	0.0		
1,602	0.0	0.0	0.0	100.0	0.0	0.0		
1,604	0.0	0.0	0.0	98.9	0.0	0.0		62.7
1,606	0.0	0.0	0.0	99.0	0.0	0.0		
1,608	1.7	0.0	0.0	98.3	0.0	0.0		
1,610	0.0	0.0	0.0	100.0	0.0	0.0		
1,612	0.0	0.0	0.0	99.8	0.0	0.0		
1,614	0.0	0.0	0.0	100.0	0.0	0,0		
1,616	0.0	0.0	0.0	86,0	0.0	0.0		58.3
1,618	0.0	0.0	0.0	84.1	0.0	0.0		62.9
1,620	0.0	0.0	0.0	84.0	0.0	0.0		62.7
1,622	0.0	0.0	0.0	80,0	0.0	0.0		61.2
1,624	5.1	0.0	0.0	91.9	0.0	0.0		65.8
1,626	3.3	0.0	0.0	75.8	0.0	0.0		64.5
1,628	0.0	0.0	0.0	76.3	0.0	0.0		63.5
1,630	0.0	0.0	0.0	97.2	0.0	0.0		62.4
1,632	0.0	0.0	0.0	98,8	0.0	0.0		65.6
1,634	0.0	0.0	0.0	99,1	0.0	0,0		
1,636	0.0	0.0	0,0	100.0	0.0	0.0		
1,638	0.0	0.0	0.0	87.5	0.0	0.0		
1,640	0.0	0.0	0.0	98.2	0.0	0.0		
1,642	0.0	0.0	0.0	86.1	0.0	0.0		
1,644	0.0	0.0	0.0	86.2	0.0	0.0		64.6
1,646	0.0	7.6	0.0	77.4	0.0	0.0		62.7
1,648	0.0	0.0	0.0	87.4	0.0	0.0		6.8
1,650	0.0	0.0	0.0	85,3	0.0	0.0		
1,652	0.0	6.3	0.0	80.8	0.0	0,0		
1,654	0.0	0.0	0.0	87.2	0.0	0.0		
1,656	0.0	0.0	0.0	90.6	0.0	0.0		
1,658	0.0	0,0	0.0	82.1	0.0	0.0		
1,660	3.2	7.2	0.0	84.7	0.0	0.0		63.2
1,662	3.9	4.3	0.0	67.1	0.0	0,0		65.3
1,664	4.2	0.0	0.0	66.2	0.0	0,0		64.6
1,666	0.0	0.0	0.0	72.7	0.0	0.0		63.5
1,668	0.0	0.0	0.0	79.4	0.0	0.0		66.0
1,670	0.0	0.0	0.0	99.4	0.0	0.0		

Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
1,672	0.0	0.0	0.0	100.0	0.0	0.0		
1,674	0.0	0.0	0.0	98.4	0.0	0.0	5.6	
1,676	0.0	0.0	0.0	99.5	0.0	0.0		
1,678	0.0	0.0	0.0	96.5	0.0	0.0		63.9
1,680	0.0	0.0	0.0	100.0	0.0	0.0		
1,730	3.3	10.0	0.0	67.7	0.0	0.0		63.2
1,732	2.3	14.0	0.0	65.2	0.0	0.0		62.3
1,734	0.0	10.5	0.0	89.1	0.0	0.0		
1,736	2.7	11.4	0.0	81.0	0.0	0.0		62.3
1,738	2.1	10.7	0.0	65.5	0.0	0.0		64.2
1,740	0.0	0.0	0.0	100.0	0.0	0.0		
1,742	0.0	0.0	0.0	99.4	0.0	0.0		
1,744	0.0	0.0	0.0	100.0	0.0	0.0		
1,746	0.0	0.0	0.0	100.0	0.0	0.0	-	
1,748	0.0	0.0	0.0	0.001	0.0	0.0		
1,750	0.0	0.0	0.0	100.0	0.0	0.0		
1,752	0.0	0.0	0.0	100.0	0.0	0.0		
1,754	0'0	0.0	0.0	96.7	0.0	0.0	7.3	64.0
1,756	0.0	0.0	0.0	100.0	0.0	0.0		
1,758	0.0	0.0	0.0	100.0	0.0	0.0		
1,760	0.0	0.0	0.0	7.00	0.0	0.0		
1,762	0.0	0.0	0.0	100.0	0.0	0.0		
1,764	0.0	0.0	0.0	100.0	0.0	0.0		
1,766	0.0	0.0	0.0	98.3	0.0	1.7		
1,768	0.0	0.0	0.0	0.001	0.0	0.0		
1,770	4.1	0.0	0.0	61.9	0.0	0.0		64.2
1,772	0.0	0.0	0.0	0.001	0.0	0.0		
1,774	0.0	0.0	0.0	100.0	0.0	0.0		
1,776	0.0	0.0	0.0	100.0	0.0	0.0		
1,778	0.0	6.6	0.0	93.4	0.0	0.0		
1,780	0.0	0.0	0.0	87.5	0.0	0.0		
1,782	0.0	0.0	0.0	100.0	0.0	0.0		
1,786	0.0	0.0	0.0	78.9	0.0	0.0		62.3
1,788	0.0	0.0	0.0	98.7	0.0	0.0		
1,790	0.0	0.0	0.0	100.0	0.0	0.0		
1,792	0.0	0.0	0.0	98.7	0.0	0.0		52.0
1,794	0.0	0.0	0.0	0.001	0.0	0,0		
1,796	0.0	18.2	0.0	79.1	0.0	0.0		63.3
1.798	0.0	8.8	0.0	91.2	0.0	0.0		

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							2	
Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)		-					Mg Calcite	Protodolomite
1,800	0.0	0.0	0.0	97.0	0.0	00		
1,802	0.0	0.0	0.0	0.001	0.0	0.0		
1,804	0.0	0.0	0.0	100.0	0.0	0.0		
1,806	0.0	0.0	0.0	97.8	0.0	00		
1,808	0.0	0.0	0.0	100.0	0.0	0.0		
1,810	0.0	0.0	0.0	100.0	0.0	00		
1,812	0.0	0.0	0.0	100.0	0.0	0.0		
1,814	0.0	5.2	0'0	94.3	0.0	0.0		
1,816	0.0	0.0	0.0	100.0	0.0	0.0		
1,818	0.0	0.0	0.0	100.0	0.0	0.0		
1,820	0.0	0.0	0.0	100.0	0.0	0.0		
1,822	0.0	0.0	0.0	80.6	19.4	0.0		
1,824	0.0	0.0	0.0	100.0	0.0	0.0		
1,826	0.0	0.0	0.0	100.0	0.0	0.0		
1,828	0.0	9.4	0.0	90,6	0.0	0.0		
1,830	0.0	0.0	0.0	100.0	0.0	0.0		
1,832	0.0	0.0	0.0	100.0	0.0	0.0		
1,834	0.0	0.0	0.0	100.0	0.0	0.0		
1,836	0.0	0.0	0.0	100.0	0.0	0.0		
1,912	0.0	0.0	0.0	100.0	0.0	0.0		
1,914	0.0	0.0	0.0	100.0	0.0	0.0		
1,916	0.0	0.0	0.0	100.0	0.0	0.0		
1,918	0.0	0.0	0.0	100.0	0.0	0,0		
1,920	0.0	0.0	0.0	100.0	0.0	0.0		
1,922	0.0	0.0	0'0	100.0	0.0	0.0		
1,924	0.0	0.0	0.0	100.0	0.0	0.0		
1,926	0.0	5.0	0.0	82,4	0.0	0.0		613
1,928	0.0	0.0	0.0	99,4	0.0	0.0		
1,930	0.0	0.0	0.0	100.0	0.0	0.0		
1,932	0.0	0.0	0.0	100.0	0.0	0.0		
1,934	0.0	0.0	0.0	100.0	0.0	0.0		
1,936	0.0	0.0	0.0	100.0	0.0	0.0		
1,938	0.0	0.0	0.0	100.0	0.0	0.0		
1,940	0.0	0.0	0.0	100.0	0.0	0.0		
1,942	0.0	0.0	0.0	100.0	0.0	0.0		
1,944	0.0	0.0	0.0	100.0	0.0	0.0		
1,946	0.0	0.0	0.0	100.0	0.0	0.0		
1,948	0.0	0.0	0.0	0.001	0.0	0.0		
1,950	0.0	0.0	0.0	100.0	0.0	0.0		

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Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)							Mg Calcite	Protodolomite
1,952	0.0	0.0	0.0	100.0	0.0	0.0		
1,954	0.0	0.0	0.0	98.8	00	00		0
1,956	0.0	0.0	0.0	100.0	0.0	0.0		0'/
1,958	0.0	0.0	0.0	100.0	0.0	0.0		
1,960	0.0	0.0	0.0	100.0	0.0	0.0		
1,962	0.0	0.0	0.0	100.0	0.0	0.0		
1,964	0.0	0.0	0.0	100.0	0.0	0.0		
1,966	0.0	0.0	0.0	100.0	0.0	0.0		
1,968	0.0	0.0	0.0	100.0	0.0	0.0		
0/6'1	0.0	0.0	0.0	100.0	0.0	00		
1,974	0.0	0.0	0.0	0.001	0.0	0.0		
1,976	0.0	0.0	0.0	100.0	0.0	0.0		
1,978	0.0	0,0	0.0	0.001	0.0	0.0		
1,980	0.0	0.0	0.0	100.0	0.0	0.0		
1,982	0.0	0.0	0.0	100.0	0.0	0.0		
1,984	0.0	0,0	0.0	100.0	0.0	0.0		
1,986	0.0	0.0	0.0	100.0	0.0	0.0		
1,988	0.0	0.0	0.0	0.001	0.0	0.0		
1,990	0.0	0.0	0.0	100.0	0.0	0.0		
1,992	0.0	0.0	0.0	100.0	0.0	0.0		
1,994	0.0	0.0	0.0	100.0	0.0	0.0		
1,996	0.0	0.0	0.0	100.0	0.0	0.0		
2,000	0.0	0.0	0.0	100.0	0.0	0.0		
2,002	0.0	0.0	0.0	0.001	0.0	0.0		
2,004	0.0	0.0	0.0	100.0	0.0	0.0		
2,006	0.0	0.0	0.0	100.0	0.0	0.0		
2,008	0.0	0.0	0.0	100.0	0.0	0.0		
2,010	0.0	0.0	0.0	100.0	0.0	0.0		
2,012	0.0	0.0	0.0	100.0	0.0	0.0		
2,014	0.0	0.0	0.0	0.001	0.0	0.0		
2,016	0.0	0.0	0.0	0.001	0.0	0.0		
2,018	0.0	0.0	0.0	100.0	0.0	0.0		
2,020	0.0	0.0	0.0	100.0	0.0	0.0		
2,022	0.0	0.0	0.0	100.0	0.0	0.0		
2,024	0.0	0.0	0.0	100.0	0.0	0.0		
2,026	0.0	0.0	0.0	100.0	0.0	0.0		
2,028	0.0	0.0	0.0	100.0	0.0	0.0		
2,030	0.0	0.0	0.0	0.001	0.0	0.0		
2,032	0.0	0.0	0.0	100.0	0.0	0.0		

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Mg Califie Mg Califie Pronologie 2014 00 <t< th=""><th>Depth</th><th>Magnesite</th><th>Gypsum</th><th>Anhydrite</th><th>Thenardite</th><th>Bloedite</th><th>Halite</th><th>Mg in High</th><th>Ca in</th></t<>	Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
2014 00 0	(us)				194 - 194 - 2411 F 888 E. E 1			Mg Calcite	Protodolomite
2006 00 0	2,034	0.0	0.0	0.0	100.0	0.0	00		
2008 00 0	2,036	0.0	0.0	0.0	100.0	0.0	00		
2400 0.0 <td>2,038</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>98.3</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,038	0.0	0.0	0.0	98.3	0.0	0.0		
2012 0.0 <td>2,040</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>100.0</td> <td>0.0</td> <td>00</td> <td></td> <td></td>	2,040	0.0	0.0	0.0	100.0	0.0	00		
2044 00 0	2,042	0.0	0.0	0.0	100.0	0.0	00		
2006 00 0	2,044	0.0	0.0	0.0	100.0	00	0.0		
2008 00 34 00 561 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00 00 00 00 00 00 00 2005 00	2,046	0.0	0.0	0.0	100.0	0.0	0.0		
2002 0.0 0.0 0.0 0.0 0.0 0.0 2008 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2008 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2008 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2008 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2008 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2008 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2008 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2017 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2018 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2018 0.0 0.0 0.0 0.0 0.0 0.0 0.0 <td>2,048</td> <td>0.0</td> <td>3.4</td> <td>0.0</td> <td>56.1</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,048	0.0	3.4	0.0	56.1	0.0	0.0		
2,054 0.0 </td <td>2,052</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>100.0</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,052	0.0	0.0	0.0	100.0	0.0	0.0		
2475 0.0 <th0.0< th=""> <th0.0< th=""></th0.0<></th0.0<>	2,054	0.0	0.0	0.0	100.0	0.0	0.0		
2008 0.0 <th0.0< th=""> <th0.0< th=""></th0.0<></th0.0<>	2,056	0.0	0.0	0.0	100.0	0.0	0.0		
2000 0.0 <th0.0< th=""> <th0.0< th=""></th0.0<></th0.0<>	2,058	0.0	0.0	0.0	100.0	0.0	0.0		
202 0.0 <th0.0< th=""> <th0.0< th=""> <th0.0< th=""></th0.0<></th0.0<></th0.0<>	2,060	0.0	0.0	0.0	100.0	0.0	0.0		
2064 0.0 <th0.0< th=""> <th0.0< th=""></th0.0<></th0.0<>	2,062	0.0	0.0	0.0	100.0	00	0.0		
2006 0.0 <th0.0< th=""> <th0.0< th=""></th0.0<></th0.0<>	2,064	0.0	0.0	0.0	100.0	0.0	0.0		
2,008 0,0 </td <td>2,066</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>1.66</td> <td>0.0</td> <td>0.0</td> <td></td> <td>0.4.3</td>	2,066	0.0	0.0	0.0	1.66	0.0	0.0		0.4.3
2070 0	2,068	0.0	0.0	0.0	100.0	0.0	00		0.40
2/072 0.0 </td <td>2,070</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>100.0</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,070	0.0	0.0	0.0	100.0	0.0	0.0		
2,074 0.0 </td <td>2,072</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>100.0</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,072	0.0	0.0	0.0	100.0	0.0	0.0		
2/076 0.0 </td <td>2,074</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>98.8</td> <td>0.0</td> <td>0.0</td> <td></td> <td>0.00</td>	2,074	0.0	0.0	0.0	98.8	0.0	0.0		0.00
2,078 0.0 0.0 0.0 80.1 169 0.0 2,082 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,082 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,084 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,088 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,088 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,093 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,093 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0	2,076	0,0	0.0	0.0	0.001	0.0	0.0		
2,080 0.0 0.0 0.0 0.0 0.0 0.0 2,082 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,084 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,084 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,086 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,088 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,092 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 0.0 </td <td>2,078</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>80.1</td> <td>16.9</td> <td>0.0</td> <td></td> <td></td>	2,078	0.0	0.0	0.0	80.1	16.9	0.0		
2,082 0.0 </td <td>2,080</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>100.0</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,080	0.0	0.0	0.0	100.0	0.0	0.0		
2,084 0.0 </td <td>2,082</td> <td>0.0</td> <td>0.0</td> <td>0'0</td> <td>100.0</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,082	0.0	0.0	0'0	100.0	0.0	0.0		
2,086 0.0 </td <td>2,084</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>99.3</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,084	0.0	0.0	0.0	99.3	0.0	0.0		
2,088 0.0 0.0 0.0 0.0 0.0 2,08 2,090 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,092 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,093 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,096 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,098 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,100 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,102 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,102 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,103 0.0 0.0 0.0 0.0 0.0 0.0 0.0<	2,086	0.0	0.0	0.0	100.0	0.0	0.0		
2,090 0.0 0.0 100.0 0.0 2.0 2,092 0.0 0.0 0.0 100.0 0.0 2.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 2.0 2,094 0.0 0.0 0.0 0.0 0.0 0.0 2.0 2,096 0.0 0.0 0.0 0.0 0.0 0.0 2.0 2,098 0.0 0.0 0.0 0.0 0.0 0.0 2.0 2,102 0.0 0.0 0.0 0.0 0.0 0.0 2.0 2,102 0.0 0.0 0.0 0.0 0.0 0.0 2.1 2.1 2.1 2.1 2.0 0.0 0.0 0.0 2.1 2.1 2	2,088	0.0	0.0	0.0	100.0	0.0	0.0		
2,092 0.0 </td <td>2,090</td> <td>0.0</td> <td>0.0</td> <td>0'0</td> <td>100.0</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,090	0.0	0.0	0'0	100.0	0.0	0.0		
2,094 0.0 </td <td>2,092</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>100.0</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,092	0.0	0.0	0.0	100.0	0.0	0.0		
2,096 0.0 </td <td>2,094</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>100.0</td> <td>0.0</td> <td>0.0</td> <td></td> <td></td>	2,094	0.0	0.0	0.0	100.0	0.0	0.0		
2,098 0.0 0.0 0.0 0.0 0.0 2,098 2,100 0.0 0.0 0.0 100.0 0.0 0.0 2,100 2,100 2,100 2,100 2,100 2,100 2,100 2,100 2,100 2,100 2,100 2,00 2,100 0,00 2,100 2,100 2,100 2,00 2,00 2,00 2,00 2,100 0,00 2,00 2,00 2,00 2,00 2,00 2,00 2,00 2,110 0,0 0,0 2,00	2,096	0.0	0.0	0.0	100.0	0.0	0.0		
2,100 0.0 0.0 0.0 100.0 0.0 2,102 0.0 0.0 0.0 100.0 0.0 2.0 2,102 0.0 0.0 0.0 0.0 0.0 0.0 2.1 2,103 0.0 0.0 0.0 0.0 0.0 0.0 2.1 2,104 0.0 0.0 0.0 0.0 0.0 0.0 2.1 2,108 0.0 0.0 0.0 0.0 0.0 0.0 2.1 2,110 0.0 0.0 0.0 0.0 0.0 0.0 54.2 2,112 0.0 0.0 0.0 0.0 0.0 0.0 5.1 2,114 0.0 0.0 0.0 0.0 0.0 0.0 5.1 5.1 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0 5.0	2,098	0.0	0.0	0.0	100,0	0.0	0.0		
2,102 0.0 0.0 0.0 0.0 0.0 2,102 2,104 0.0 0.0 0.0 0.0 0.0 0.0 0.0 2,103 2,104 0.0 0.0 0.0 2,103 2,103 0.0 0.0 0.0 2,103 2,110 0.0 0.0 0.0 64.2 64.2 2,110 0.0 0.0 0.0 0.0 54.2 54.2 51.1 0.0 0.0 0.0 54.2 54.2 51.1 54.2 <t< td=""><td>2,100</td><td>0.0</td><td>0.0</td><td>0.0</td><td>100.0</td><td>0.0</td><td>0.0</td><td></td><td></td></t<>	2,100	0.0	0.0	0.0	100.0	0.0	0.0		
2,104 0.0 0.0 0.0 0.0 0.0 54.2 2,108 0.0 0.0 0.0 0.0 0.0 54.2 2,108 0.0 0.0 0.0 0.0 0.0 54.2 2,110 0.0 0.0 0.0 0.0 0.0 54.2 2,112 0.0 0.0 0.0 0.0 0.0 2.1 2,114 0.0 0.0 0.0 0.0 0.0 2.0 2,114 0.0 0.0 0.0 0.0 0.0 2.0	2,102	0.0	0.0	0.0	100.0	0.0	0.0		
2,108 0.0 </td <td>2,104</td> <td>0.0</td> <td>0.0</td> <td>0.0</td> <td>98,0</td> <td>0.0</td> <td>0.0</td> <td></td> <td>6 4 3</td>	2,104	0.0	0.0	0.0	98,0	0.0	0.0		6 4 3
2,110 0.0 0.0 0.0 100.0 0.0 0.0 0.0 2,112 0.0 0.0 0.0 0.0 0.0 0.0 2,114 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	2,108	0.0	0.0	0.0	0.001	0.0	0.0		7'40
2,112 0.0 0.0 0.0 100.0 0.0 0.0 0.0 2,114 0.0 0.0 0.0 0.0 0.0	2,110	0.0	0.0	0.0	100.0	0.0	0.0		
2,114 0.0 0.0 0.0 100.0 0.0 0.0 0.0	2,112	0.0	0.0	0.0	0.001	00	0.0		
	2,114	0.0	0.0	0.0	100.0	0.0	0.0		

Mineralogical Composition of North Core at North Ingebright Lake

Depth	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)							Mg Calcite	Protodolomite
2,116	0.0	0.0	0.0	100.0	0.0	0.0		
2,118	0.0	0.0	0.0	100.0	0.0	0.0		
2,120	0'0	0.0	0.0	100.0	0.0	0.0		
2,122	0.0	0.0	0.0	100.0	0,0	0.0		
2,124	0.0	0.0	0.0	100,0	0.0	0.0		
2,126	0.0	0.0	0.0	100.0	0.0	0.0		
2,128	0.0	0.0	0.0	100.0	0.0	0.0		
2,130	0.0	0.0	0.0	100.0	0.0	0.0		
2,132	0.0	0.0	0.0	100.0	0.0	0.0		
2,134	0.0	0.0	0.0	100,0	0.0	0.0		
2,136	0'0	0.0	0.0	100.0	0.0	0.0		
2,138	0.0	0.0	0.0	100.0	0.0	0.0		
2,140	0.0	0.0	0.0	100.0	0.0	0.0		
2,142	0.0	0'0	0.0	100.0	0.0	0.0		
2,144	0.0	0,0	0.0	100.0	0.0	0.0		
2,146	0.0	0.0	0.0	100.0	0.0	0.0		
2,148	0.0	0.0	0.0	100.0	0.0	0.0		
2,150	0.0	0'0	0.0	100.0	0.0	0.0		
2,152	0.0	0,0	0.0	100.0	0.0	0.0		
2,154	0.0	0.0	0.0	100.0	0.0	0.0		
2,156	0.0	0'0	0.0	0.001	0.0	0.0		
2,158	0.0	0.0	0.0	100.0	0.0	0.0		
2,160	0.0	0.0	0.0	100.0	0.0	0.0		
2,162	0.0	0.0	0.0	100.0	0.0	0.0		
2,164	0.0	0.0	0.0	100.0	0.0	0.0		
2,166	0.0	0.0	0.0	100.0	0.0	0.0		
2,168	0.0	0.0	0.0	100.0	0.0	0.0		
2,170	0.0	0.0	0.0	100.0	0.0	0.0		
2,172	0.0	0.0	0.0	100.0	0.0	0.0		
2,174	0.0	0.0	0'0	100.0	0.0	0'0		
2,176	0.0	0.0	0.0	100.0	0.0	0.0		
2,178	0.0	0.0	0'0	100.0	0.0	0.0		
2,180	0.0	0.0	0'0	100.0	0.0	0.0		
2,182	0.0	0.0	0.0	100.0	0.0	0.0		
2,184	0.0	0.0	0.0	0.001	0.0	0.0		
2,186	0.0	0.0	0.0	99.4	0.0	0.0		
2,188	0'0	0.0	0.0	100.0	0.0	0.0		
2,190	0.0	0.0	0.0	100.0	0.0	0.0		
2,192	0.0	0.0	0.0	100.0	0.0	0.0		

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Dent		ç						
ndər	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High	Ca in
(cm)							Mg Calcite	Protodolomite
2,194	0.0	0.0	0.0	100.0	0.0	00		
2,196	0.0	0.0	0.0	100.0	00	00		
2,198	0.0	0.0	0.0	100.0	0.0	0.0		
2,200	0.0	6.0	0.0	94.0	0.0	0.0		
2,202	0.0	0.0	0.0	100.0	0.0	0.0		
2,204	0.0	0.0	0.0	100.0	0.0	0.0		
2,206	0.0	0.0	0.0	100.0	0.0	0.0		
2,208	0.0	0.0	0.0	100.0	0.0	0.0		
2,210	0.0	0.0	0.0	100.0	0.0	00		
2,212	0.0	0.0	0.0	100.0	0.0	0.0		
2,214	0.0	0.0	0.0	100.0	0.0	0.0		
2,216	0.0	0.0	0.0	100.0	0.0	0.0		
2,218	0.0	0.0	0.0	100.0	0.0	0.0		
2,220	0.0	5.3	0.0	94.7	0.0	00		
2,222	0.0	0.0	0.0	0.001	0.0	0.0		
2,224	0.0	0.0	0.0	100.0	0.0	0.0		
2,226	0.0	0.0	0.0	100.0	0.0	0.0		
2,228	0.0	0.0	0.0	100.0	0.0	0.0		
2,230	0.0	0.0	0.0	0.001	0.0	0.0		
2,232	0.0	0.0	0.0	0.001	0.0	0.0		
2,234	0.0	0.0	0.0	100.0	0.0	0.0		
2,236	0.0	0.0	0.0	100.0	0.0	0.0		
2,238	0'0	0.0	0.0	100.0	0.0	0.0		
2,240	0'0	0.0	0.0	100.0	0.0	0.0		
2,242	0.0	0.0	0.0	0.001	0.0	0.0		
2,244	0.0	0.0	0.0	100.0	0.0	0.0		
2,246	0.0	0.0	0.0	100.0	0.0	0.0		
2,250	0.0	0.0	0.0	100.0	0.0	0.0		
2,252	0.0	0.0	0.0	100.0	0.0	0.0		
2,254	0.0	0.0	0.0	100.0	0.0	0.0		
2,256	0.0	0.0	0.0	100.0	0.0	0.0		
2,258	0.0	0.0	0.0	100.0	0.0	0.0		
2,260	0.0	0.0	0.0	100.0	0.0	0.0		
2,262	0.0	0.0	0.0	100.0	0.0	0.0		
2,264	0.0	0.0	0.0	100.0	0.0	0.0		
2,266	0.0	0.0	0.0	0.001	0.0	0.0		
2,268	0.0	0.0	0.0	100.0	0.0	0.0		
2,270	0.0	0.0	0.0	100.0	0.0	0.0		
2,272	0.0	0.0	0.0	100.0	0.0	0.0		

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Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
2,274	0.0	0.0	0.0	100.0	0.0	0.0		
2,276	0.0	0.0	0.0	100.0	0.0	0.0		
2,280	0.0	0.0	0.0	100.0	0.0	0.0		
2,282	0.0	0.0	0.0	100.0	0.0	0.0		
2,284	0.0	0.0	0.0	100.0	0.0	0.0		
2,286	0.0	0.0	0.0	100.0	0.0	0.0		
2,288	0.0	0.0	0.0	100.0	0.0	0,0		
2,290	0.0	0.0	0.0	100.0	0.0	0.0		
2,292	0.0	0.0	0,0	100.0	0.0	0.0		
2,294	0.0	0.0	0.0	100.0	0.0	0.0		
2,296	0.0	0.0	0.0	100.0	0.0	0,0		

Mineralogical Composition of North Core at North Ingebright Lake

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Depth (cm)	Quartz	K- Feldspar	Plag.	Total Clay Miner.	Calcite	Aragonite	High-Mg Calcite	Proto Dolomite	Dolomite
10	11.2	0.0	2.8	0.0	0.0	0.0	0.0	0.0	22.2
12	6.4	0.0	1.6	0.0	0.0	0.0	0.0	12.1	0.0
14	11.5	0.0	2.5	38.0	0.0	0.0	0.0	0.0	19.8
16	11.5	0.0	2.4	38.9	0.0	0.0	0.0	0.0	19.8
18	15.0	0.0	3.0	0.0	0.0	0.0	1.8	0.0	26.1
20	9.3	5.6	2.0	34.8	0.0	0.0	0.0	0.0	15.0
22	11.6	6.0	4.3	44,8	0.0	0.0	1.6	20,6	0.0
24	8.5	0.0	3.1	39.9	0.0	0.0	0.0	0.0	21.1
26	9.7	4.7	3.3	39.8	0.0	0.0	2.5	0.0	17.7
28	15.0	12.9	4.1	0.0	0.0	0.0	0.0	34.9	0.0
30	21.3	0.0	0.0	52.9	0.0	0.0	0.0	25.8	0.0
32	15.3	5.2	2.8	44.3	0.0	0.0	0.0	19.8	0.0
34	30.5	0.0	8.1	0.0	0.0	0.0	0.0	0.0	36.7
36	24.1	0:0	5.4	0.0	0.0	0.0	0.0	0.0	39.0
38	15.1	4.9	10.6	47.8	0.0	0.0	0.0	19.2	0.0
40	42.7	0.0	8.2	0.0	0.0	0.0	0.0	0.0	49.2
42	27.0	0.0	6.0	0.0	0.0	0.0	0.0	0,0	40.4
44	26.9	0.0	9.7	0.0	0.0	0.0	0.0	0.0	39,3
46	19.1	4.9	3.1	37.1	0.0	0.0	2.0	0.0	14.4
48	11.6	5.0	12.5	44.0	0.0	0.0	1.5	0.0	12.5
50	1.9	0.0	1.4	0.0	32.1	0.0	0.0	0.0	1.7
52	6.2	0.0	0,0	0.0	0.0	0.0	0.0	0,0	2.5
54	2.2	0.0	1.8	15.1	0.0	0.0	25.7	0.0	0.0
56	3.0	0.0	0.0	0.0	0.0	0.0	0.0	1.2	0.0
58	9.0	0.0	0.0	0.0	0.0	0.0	0.7	0.0	0.0
60	1.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.2
62	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0	0.0
64.	0.0	0.0	0.0	0.0	0.0	0.0	0.0	100.0	0.0
66	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.0
68	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
70	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
72	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
74	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
76	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
78	0.3	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0

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Depth	Ouartz	K-	Plag.	Total	Calcite	Argonite	Hiah_Ma	Droto	Delemite
(cm)	,	Feldspar	D	Clay Miner.		9	Calcite	Dolomite	
80	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
82	0.4	0.0	0.0	0.0	0.0	0.0	0.0	00	0.0
84	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
86	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
88	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
90	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
92	0.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
94	0.7	0.0	1.6	0.0	0.0	0.0	0.0	0.0	0.0
96	0.6	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
98	0.3	0.0	0.0	0.0	0.0	0'0	0.0	0,0	0.0
100	0.4	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0
102	0.6	0.0	0.0	0.0	0,0	0.0	0.0	0'0	0.0
104	0.0	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0
901	0.4	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
108	0.4	0.0	0.0	0.0	0.0	0,0	0.0	0.0	0.0
110	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.1
112	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
114	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
116	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
118	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
120	2.0	0.0	0.0	0.0	0,0	0.0	0.0	0.0	0.0
122	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
124	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
126	0.5	0.0	0.0	0.0	0'0	0.0	0.0	0,0	0.0
128	0.6	0.0	0.0	0.0	0.0	0'0	0.0	0.0	0.0
130	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
132	1.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
134	1.0	0.0	0.0	0.0	1.6	0.0	0.0	0.0	0.0
136	0.7	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
138	2.0	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0
140	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
142	1.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
144	1.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
146	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
148	0.5	0.0	0'0	0.0	0.0	0.0	0.0	0'0	0.0

Ingebright Lake
Core at North
omposition of South
Mineralogical C

Depth (cm)	Quartz	K- Feldspar	Plag.	Total Clay Miner.	Calcite	Aragonite	High-Mg Calcite	Proto Dolomite	Dolomite
150	0.3	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
152	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
154	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
156	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
158	2.2	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.8
160	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
162	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
164	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
991	0.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0
168	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
170	1.1	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
172	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
174	0.6	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0
176	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
178	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
180	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0'0	0.0
182	2.9	0.0	0.0	0.0	0.0	0.0	0.0	1.0	0.0
184	0.8	0.0	0.0	0.0	0.0	0'0	0.0	0.0	0.0
186	0.0	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0
188	0.5	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
061	0.0	0.0	0.0	0.0	0.0	1.6	0.0	0.0	1.7
192	1.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
194	0.8	0.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0
961	1.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
198	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
200	0.4	0.0	0.0	0.0	0.0	0.0	1.2	0.0	0'0
202	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
204	1.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
226	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	Ó.Ó
228	2.0	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
230	8.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
232	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0
234	0.7	0'0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
236	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
238	0.6	0.0	0.0	0.0	0.0	0'0	0.0	0'0	0.0

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Depth (cm)	Quartz	K- Feldspar	Plag.	Total Clay Miner.	Calcite	Aragonite	High-Mg Calcite	Proto Dolomite	Dolomite
240	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
242	0.9	0'0	0,0	0.0	0.0	0.0	0.0	0.0	0.0
244	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
246	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7
248	1.2	0.0	2.6	0.0	0.0	0.0	0.0	0.0	0.0
250	1.2	0.0	0.0	0.0	0.0	0.0	0.0	1.7	0.0
252	2.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
254	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
256	1.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
258	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
260	1.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
262	2.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.5
264	2.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
266	1.7	0.0	0.0	0.0	0'0	1.4	0.0	0.0	0.0
268	0.1	2.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0
270	1.5	0.0	0.0	23.3	0.0	0.0	0.0	0.0	0.0
272	1.6	0.0	0.0	22.5	0.0	1.2	0.0	0.0	0.0
274	2.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
276	1.6	0.0	0.0	0.0	0.0	1.7	0.0	0'0	0.0
278	2.1	0.0	1.6	24.0	0.0	1.7	0.0	1.5	0.0
280	1.4	0.0	0.0	19.5	0.0	0.0	0.0	0.0	0.0
282	3.4	0.0	0.0	0.0	0.0	0.0	0.0	1.7	3.0
284	3.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
286	3.2	0.0	0.0	0.0	0.0	0.0	0.0	2.0	0.0
288	3.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.1
290	3.0	2.7	0.0	30.9	0.0	0.0	0.0	1.7	0.0
292	3.4	3.2	0.0	27.0	0.0	0.0	0.0	1.5	2.1
294	4.7	0.0	1.7	28.8	0.0	0,0	1.9	2.3	2.6
296	3.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.2
298	1.7	0.0	0.0	18.9	0.0	0.0	0.0	0.0	0.0
300	2.2	0.0	2.2	0.0	0.0	0.0	0.0	0.0	2.9
302	1.7	0.0	0.0	20.2	0.0	0.0	0.0	1.7	0.0
304	1.5	0.0	0.0	20.4	0.0	0.0	0.0	1.9	1.3
306	3.0	0.0	0.0	0.0	0.0	0.0	0.0	2.0	0.0
308	1.5	0.0	0.0	19.4	0.0	0.0	0.0	0.0	0.0

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Mineralogical Composition of South Core at North Ingebright Lake

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Ingebright Lake
Core at North
omposition of South
Mineralogical C

Depth	Ouartz	¥	Plao.	Total	Calcito	Arganita	Hiah Ma	Droto	Delemite
(cm)	1	Feldspar	0	Clay Miner.			Calcite	Dolomite	
310	1.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	00
312	1.6	0.0	1.0	0.0	0.0	0.0	0.0	0.0	0.0
314	2.7	0.0	0'0	0.0	0.0	0.0	0.0	1.8	00
316	2.1	0.0	0.0	0.0	0.0	0.0	0.0	2.0	0.0
318	2.8	0.0	0.0	0.0	0.0	0.0	0'0	2.5	2.3
320	2.7	0.0	0'0	23.2	0.0	0.0	1.3	2.2	2.0
322	8.1	0.0	0.0	19.2	0.0	0.0	0.0	1.9	1.5
324	1.4	0.0	0.0	22.1	0.0	0.0	0.0	1.5	0.0
326	4.7	0.0	0.0	39.7	0.0	0.0	0.0	3.0	2.0
328	2.5	0.0	1.2	34.8	0.0	0.0	1.2	2,2	0.0
330	2.3	0.0	0.0	27.5	0.0	0.0	0.0	1.9	2.0
332	1.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.2
334	2.4	0.0	0.0	0.0	0.0	0.0	0.0	2.5	2.6
336	2.2	0.0	1.9	0.0	0.0	0.0	0.0	2.7	1,8
338	3.1	0.0	1.4	27.1	0.0	0'0	0.0	1.9	0.0
340	0.1	0.0	0.0	0.0	0.0	0'0	0.0	1.5	0.0
342	2.3	0.0	0'0	0.0	0.0	0.0	0.0	2.7	2.3
344	1.5	0'0	0.0	21.1	0.0	0.0	0.0	1.8	0,0
346	2.2	0.0	0.0	27.7	0.0	0.0	1.2	1.6	1.8
348	2.0	0.0	0.0	24.9	0.0	0.0	0.0	1.3	0.0
350	1.4	0.0	0'0	0.0	0.0	0.0	0.0	0.0	0.0
352	1.3	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0
354	2.7	0.0	0.0	0.0	0.0	0.0	0.0	1.4	3.4
356	1.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
358	1.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
360	6.1	0.0	0.0	0.0	0.0	0.0	0.0	2.0	0.0
362	1.5	0.0	0.0	25.2	0.0	0.0	0.0	0.0	0.0
364	1.9	0.0	0.0	0'0	0.0	0.0	0.0	2.0	0.0
366	2.6	0.0	0.0	0.0	0'0	0.0	0'0	0.0	2.8
368	1.4	0.0	0.0	0.0	0.0	0.0	0'0	0.0	0.0
370	1.5	0.0	1.2	0.0	0'0	0.0	0'0	0.0	1.9
372	1.6	0.0	0.0	0.0	0.0	0.0	0.0	1,4	0.0
374	1.0	0.0	0.0	0.0	0.0	0.0	0.0	2,0	0.0
376	2.0	0.0	0.0	0.0	0.0	0.0	0'0	0,0	0.0
378	0.9	0.0	0.0	0.0	0'0	0.0	0.0	0.0	0.0

Depth (cm)	Quartz	K- Feldspar	Plag.	Total Clay Miner.	Calcite	Aragonite	High-Mg Calcite	Proto Dolomite	Dolomite
380	1.8	0.0	0.0	0.0	0.0	0.0	0.0	2.8	22
382	1.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.8
384	1.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	22
386	1.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
388	2.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
390	1.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
392	1.4	0.0	0.0	20.2	0.0	0.0	0.0	1.4	0.0
394	2.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.9
396	2.8	0.0	5.1	0.0	0.0	0.0	0.0	2.5	0.0
398	2,4	0.0	0.0	0.0	0.0	0.0	0.0	2.4	0.0
400	2.6	0.0	0.0	0.0	0.0	0.0	0.0	1.6	1.8
402	2.3	0.0	0.0	28.9	0.0	0.0	0.0	3.6	0.0
404	2.6	0.0	0.0	28.7	0.0	0.0	0.0	0.0	2.1
406	1.4	0.0	0,0	0.0	0.0	0.0	0.0	1.7	0.0
408	3.0	0.0	0.0	21.2	0.0	1.3	0.0	2.0	2.2
410	2.1	0.0	0.0	26,9	0.0	0.0	0.0	0.0	0.0
412	2,4	0.0	0,0	27.9	0.0	1.3	0.0	0.0	0.0
414	2.5	0.0	0,0	0.0	0.0	1.9	1.7	2.3	2.3
416	1.2	0.0	0.0	19.2	0,0	0.0	0.0	1,3	1.8
500	2.3	0.0	2.7	34.6	0.0	0.0	0.0	2,5	2,2
502	4.7	0.0	0.0	0.0	0.0	0.0	0.0	0,0	0.0
504	2.1	0.0	3.3	0.0	0.0	0.0	2.0	0.0	0.0
508	1.9	0.0	0.0	30.9	0.0	0.0	0.0	2,9	0.0
510	2.9	0.0	0.0	38.7	0.0	0.0	0.0	2,8	3.0
512	4.8	0.0	3.4	41.6	0.0	0.0	0.0	2.4	0,3
514	3,3	0.0	0.0	29.1	0.0	0.0	0.0	2,3	3.0
516	4.3	0.0	1.9	44.5	0.0	0.0	0.0	2.6	2.2
518	3.7	0.0	1.3	28.4	0.0	0.0	0.0	2.7	17.0
520	3.7	0.0	1.3	25.7	0.0	0.0	0.0	1,8	0.0
522	1.8	0.0	0.0	24.1	0.0	0.0	0.0	2.0	3.0
524	3.0	0.0	0.0	24.4	0.0	0,0	0.0	2.4	1.9
526	2.4	0.0	0,0	38.4	0.0	0.0	0.0	2.9	2.1
528	5.6	0.0	0.0	36.5	0.0	0.0	0.0	2.7	2.7
530	2.9	0.0	. 0.0	30.9	0.0	0.0	0.0	1.7	2.7
532	6.7	0.0	0.0	24.3	0.0	0.0	0.0	0.0	1.4

M	ineral	logical	Com	position	of	South	Core	at	Nort	h In	igeb	right	: Lake
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Depth (cm)	Quartz	K- Feldspar	Plag.	Total Clay Miner.	Calcite	Aragonite	High-Mg Calcite	Proto Dolomite	Dolomite
534	2.8	0.0	0.0	18.7	0.0	0.0	0.0	1,6	1.6
536	2.0	0.0	1.5	26.6	0.0	0.0	0.0	2.3	0.0
538	2.8	0.0	0.0	26.7	0.0	0.0	0.0	1.8	0.0
540	3,4	0.0	0.0	27.4	0.0	0.0	0.0	2,2	2.2
542	2.5	0.0	0.0	27.5	0.0	0.0	0.0	2.2	3.1
544	2.7	1.8	0.0	22.6	0.0	0.0	0.0	2,4	2.0
546	3.0	0.0	0.0	37.0	0.0	0.0	0.0	3.2	2.1
548	2.7	0.0	0.0	29.8	0.0	0.0	0.0	2.9	2.6
550	2.1	0.0	0.0	36.4	0.0	0.0	0.0	2.9	0.0
552	2.1	0.0	0.0	27.5	0.0	0.0	0.0	1.8	2.2
554	4.2	0.0	0.0	0,0	0.0	0.0	0.0	2.6	0.0
556	1.8	2.0	0.0	27.4	0.0	0.0	0.0	2.1	2.0
558	2.0	1.7	0.0	25.5	0.0	0.0	0.0	2.4	0.0
560	1.5	0.0	0.0	27.2	0.0	0,0	0.0	2.2	2.7
562	1.7	0.0	0.0	35.2	0,0	0.0	0.0	3.0	0.0
564	1.5	0.0	0.0	26.1	0.0	0.0	0.0	1,6	0.0
566	2,3	0.0	0.0	28.3	0.0	0.0	0.0	1.9	1.8
568	1.8	0.0	0.0	19.4	0.0	0.0	0.0	2.0	0.0
570	1.1	0.0	0.0	18.6	0.0	0.0	0.0	1,6	1.6
572	3.0	0,0	0.0	23.8	0.0	0.0	0.0	1,6	0.0
574	2.3	0.0	0.0	0,0	0.0	0.0	0.0	2.4	0.0
576	1.4	0.0	0.0	19.6	0,0	0.0	0.0	1,8	2.1
578	1.8	0.0	1.3	24.1	0.0	0.0	0.0	2.0	2.9
580	1.4	0.0	0.0	21.3	0.0	0.0	0.0	1.7	2.3
582	1.1	0.0	0.4	20.1	0.0	0.0	0.0	1.6	0.0
584	1.2	1.9	0.0	0,0	0.0	0.0	0.0	2.2	0.0
586	0.8	0.0	0.0	6.5	0.0	0.0	1.5	0.0	0.0
588	1.8	0.0	0.0	20.4	0.0	0,0	0.0	1.3	1.2
590	1.4	0.0	0.8	16,7	0.0	0.0	0.0	1.5	4.4
592	1.0	0.0	0.0	17.3	0.0	0.0	0.4	1.3	2.4
594	0.9	0.0	0.0	16.3	0.0	0.0	0.7	1.7	0.0
596	1.4	0.0	0.0	17.8	0.0	0.0	0.0	1.6	1.6
598	0.8	0.9	0.0	19.6	0,0	0,0	0.0	0.0	1.0
600	1.0	0.0	0.0	24.9	0.0	0.0	0.0	2.7	1.8
602	1.2	0.0	0.0	18.7	0.0	0.0	0.0	1.5	1.7

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(cm)	Quartz	K- Feldspar	Plag.	Total Clay Miner.	Calcite	Aragonite	High-Mg Calcite	Proto Dolomite	Dolomite
604	1.3	1.6	0.0	29.3	0.0	0.0	0.0		
606	1:1	0.0	0.0	18,9	0.0	0.0	0.0	212 1 K	0.0
608	0.9	0.0	0.0	20.7	0.0	0.0	0.0	0'1	<u>.</u>
610	0.8	0.0	0.0	18.9	0.0	0.0	0.0	0.1	א ס סי - ס
612	0.8	0.0	0.0	0.0	0.0	0.0	0.0	2 2	0.0
614	2.1	0.0	0.0	0.0	0.0	0.0	0.0	21	0.0
616	1.9	0.0	0.0	20.0	0.0	2.0		0.0	01
618	1.0	0.0	0.0	0.0	0.0	1.5	0.0	0.0	00
620	3,5	0.0	1.8	0.0	0.0	4.2	2.2	0.0	0.0
622	2.3	0.0	0.0	0.0	1.7	3.2	0.0	0.0	2.8
624	2.2	0.0	0.0	0.0	0'0	0.0	0.0	0.3	0.0
626	1.7	0.0	0'0	15.4	0.0	2.1	0.9	5.1	14
628	3.3	0.0	0.0	0.0	3.3	4.0	0.0	0.0	5.6
630	1.8	0.0	0.0	17.6	0.0	2.6	0.0	0.0	n -
632	1.7	0.0	0.0	0.0	0.0	2.1	1.5	0.0	C. C
634	1.7	0.0	0.0	0.0	0.0	2.0	1.2	0.0	1.7
636	1.4	0.0	0.0	0.0	0.0	1,2	0.0	0.0	00
638	3.7	0.0	0.0	0.0	0.0	2.3	0.0	2.3	0.0 7 4
640	2.0	0.0	0.0	17.3	0.0	2.0	0.0	1.7	00
642	4.5	3.0	0.0	0.0	0.0	2.8	2.5	0.0	2.7
644	2.6	0.0	0.0	0.0	0.0	1,8	1.4	1.1	1.6
646	0.4	0.0	0.0	13.7	0.0	0.0	0.0	1.6	0.0
648	1:1	0.0	0'0	0.0	0.0	1.7	0.0	1.8	0.0
650	1.6	0.0	0'0	22.9	0.0	0.0	1.2	1.3	1.7
700	1.6	0.0	0.0	0.0	0.0	2.2	0.0	0.0	2.3
900	0.4	0.0	0.7	0.0	0.0	0.0	0.0	0.0	0.0
000	0.5	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
850	0.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
000	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
002	0.4	0.9	0.0	0.0	0.0	0.0	0.0	0.0	0.7
604	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
000	0.1	0.0	0.0	0.0	0.0	0.0	0.0	1.2	<u>[]</u>
668	2.0	2.3	2.3	0'0	0.0	0.0	0.0	0.0	0.0
0/9	1.2	0.0	0.0	0.0	0.0	0.0	0.0	1.5	1.8
0/2	0.9	0.0	0.0	0.0	0'0	1.2	0.0	0.0	4.1

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Depth (cm)	Quartz	K. Feldspar	Plag.	Total Clay Miner.	Calcite	Aragonite	High-Mg Calcite	Proto Dolomite	Dolomite
674	0.5	0.0	2.0	0.0	0.0	0.0	0.0	0.0	0'0
676	0.5	0'0	0.0	0.0	0.0	0.0	0.0	1.3	0.0
678	0.9	0.0	0.0	15.0	0.0	1.3	0.0	1.2	1.8
680	1.3	0.0	0.0	20.7	0.0	0.0	0.0	1.7	0.0
682	1.3	0.0	0.0	0.0	0.0	0.0	0.0	1.8	2.2
684	0.7	0.0	0.0	12.9	0.0	0.0	0.0	1.3	2.0
686	0.5	0.0	0.0	15.9	0.0	0.0	0.0	1.6	0.0
824	0.3	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0'0
826	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
828	0.6	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.8
830	1.7	0.0	0.0	0.0	0.0	0.0	0.0	0.0	2.1
832	0.6	0.0	0.0	14.6	0.0	0.0	0.0	0.0	0.0
834	1.4	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0
836	1.2	0.0	0.0	29.9	0.0	0.0	0.0	1.7	0.0
838	1.5	0.0	0.0	24,9	0.0	0,0	0.0	00	0.0
840	2.0	0.0	0.0	33.1	0.0	0.0	0.0	6.1	0.0
842	1.4	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
844	1.4	2.4	0.0	33.4	0.0	0.0	0.0	2,9	1.9
846	1.5	0.0	0.0	32.6	0.0	0.0	0.0	3.5	2.4
848	0.6	0.0	0.0	0,0	0.0	0.0	0.0	0.0	0.0
850	1,4	0.0	0.0	28.6	0.0	0.0	0.8	1.7	9'1
852	1.5	0.0	1.8	0.0	0.0	1.9	0:0	2.6	1.7
854	0.0	0.8	0.0	17.3	0.0	0.0	0.0	1.4	0.0
856	0.7	0.0	0'0	23.9	0.0	0,0	0.0	2.3	0.0
858	0.5	0.0	0.0	25.9	0'0	0.0	0.0	2.2	0.0
860	0.5	0.0	0.0	0.0	0.0	0.0	0.0	1.4	2.5
862	0.5	0.0	0.0	0.0	0.0	0.0	0.0	1.4	0.0
864	0.6	0.0	0'0	21.3	0.0	0.0	0.0	2.0	0'0
866	0.8	0.0	0.0	29.5	0'0	0.0	0.0	2.7	0.0
808	0.3	0.0	0.5	61.4	0.0	0.0	0.0	1.4	1.6
870	5.0	2.4	1.7	0.0	0.0	0.0	1.2	0.0	7.4
872	2.3	1.2	0.0	14.5	0.0	0.0	0.0	1.0	2.3
874	3.0	1.1	0.8	0.0	0.0	0.0	0.0	0.0	2.0

261

1.5 3.0

0.0

0.0 1.4

0.0

0.0 0.0

18.8 0.0

1.2 1.4

2.5 0.0

4.1 5.4

876 878

Depth (cm)	Quartz	K- Feldspar	Plag.	Total Clay Miner.	Calcite	Aragonite	High-Mg Calcite	Proto Dolomite	Dolomite
880	2.3	0.0	0.6	11.6	0.0	0.0	0.0	0.0	3.1
882	5.5	0.0	0.0	20,8	1.4	0.0	0.0	1.0	2.8
884	3.0	1.8	0.0	19.0	0.0	0.0	0.0	0.0	3.4
886	2.5	0.0	0.7	0.0	0,6	0.0	0.0	0.7	2,0
888	1.7	2.0	0.9	9.7	0,6	0.0	0.0	1.1	1.4
890	8.0	7.0	0.0	15.0	0.5	0.0	0.0	0.0	0,7
892	1.9	0.0	0.9	12.5	0.7	0.0	0.0	0.0	1.8
894	1.6	0.0	4.0	15.0	0.6	0.0	0.3	0.0	2.2
896	1.7	1.4	1.1	12.2	0.8	0.0	0.0	0,0	0.0
898	26.3	0.0	15.0	40.9	0.0	0.0	0.0	0.0	5.8
900	24.1	0.0	5.4	35.0	0.0	0.0	0.0	0.0	4.0
910	25.1	4.9	18.6	37.8	0.0	0.0	0.0	1,2	2.5

		,						
(cm)	IMAGNESILE	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
10	0.0	0.0	0.0	63.7	0.0	0.0		
12	0.0	0.0	7.4	19.3	53.2	0.0		113
14	0.0	0.0	0.0	28.1	0.0	0.0		L , L,
16	0.0	0.0	0.0	27.5	0.0	0.0		
18	0.0	0.0	0.0	54.0	0.0	0.0	8.0	
20	0.0	0.0	0.0	33,4	0.0	0.0	5	
22	0.0	0.0	0.0	1.11	0.0	0.0	2.0	503
24	0.0	0.0	0.0	27.4	0.0	0.0	2 i	C.3C
26	0.0	0.0	0.0	22.4	0.0	0.0	15.8	
28	0.0	0.0	0.0	33.1	0.0	0.0	2	57 R
30	0.0	0.0	0.0	0.0	0.0	0.0		54.8
32	0.0	0.0	0.0	12.5	0.0	0.0		51.8
34	0.0	0.0	0.0	24.7	0.0	0.0		
36	0.0	0.0	0.0	31.4	0.0	0.0		
38	0.0	0.0	0.0	0.0	0.0	2.5		52.0
40	0.0	0.0	0.0	0.0	0.0	0.0		
42	0.0	0.0	0.0	26.7	0.0	0.0		
44	0.0	0.0	0.0	24.0	0.0	0.0		
46	0.0	0.0	0.0	19.3	0.0	0.0	16.8	
48	0.0	0.0	0.0	13.0	0.0	0.0	16.8	
50	0.0	0.0	0.0	62.9	0.0	0.0		
52	0.0	0.0	0.0	61.3	0.0	0.0		
54	0.0	0.0	0.0	55.2	0.0	0.0		
56	0.0	0.0	0.0	95.8	0.0	0.0		52.7
58	0.0	0.0	0.0	98.7	0.0	0.0	4.5	
99	0.0	0.0	0.0	96.4	0.0	0.0		
62	0.0	0.0	0.0	100.0	0.0	0.0		
64	0.0	0.0	0.0	0.0	0.0	0.0		80.0
99	0.0	0.0	0.0	99.0	0.0	0.0		
68	0.0	0.0	0'0	100.0	0.0	0.0		
20	0.0	0.0	0.0	99.4	0.0	0.0		
72	0.0	0.0	0.0	99.5	0.0	0.0		
74	0.0	0.0	0.0	100.0	0.0	0.0		
76	0.0	0.0	0.0	9.66	0.0	0.0		
78	0.0	0.0	0.0	7.66	0.0	0.0		

			INDER AN COUNT	DC IN HARISON	uth Core a	I NOLIN II	igebright Lak	9
Jepth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
80	0.0	0.0	0.0	39.5	0.0	0.0		
82	0.0	0.0	0.0	99.6	0.0	0.0		
84	0.0	0.0	0.0	99,2	0.0	0.0		
86	0.0	0.0	0.0	98,9	0.0	0.0		
88	0.0	0.0	0.0	99.5	0.0	0.0		
90	0.0	0.0	0.0	100.0	0.0	0.0		
92	0.0	0.0	0.0	0.66	0.0	0.0		
94	0.0	0.0	0.0	7.76	0.0	0.0		
96	0.0	0.0	0.0	99.4	0.0	0.0		
98	0.0	0.0	0.0	7.66	0.0	0.0		
100	0.0	0.0	0.0	9.66	0.0	0.0		
102	0.0	0.0	0.0	99.4	0.0	0.0		
104	0.0	0.0	0.0	0.001	0.0	0.0		
106	0.0	0.0	0.0	99.6	0.0	0.0		
108	0.0	0.0	0.0	9.66	0.0	0.0		
110	0.0	0.0	0.0	98.9	0.0	0.0		
112	0.0	0.0	0.0	99.6	0.0	0.0		
114	0.0	0.0	0.0	1 00.0	0.0	0.0		
116	0'0	0.0	0.0	100.0	0.0	0.0		
118	0.0	0.0	0.0	99.6	0.0	0.0		
120	0.0	0.0	0.0	98.0	0.0	0.0		
122	0.0	0.0	0.0	1.66	0.0	0.0		
124	0.0	0.0	0.0	100.0	0.0	0.0		
126	0.0	0.0	0.0	99.5	0.0	0.0		
128	0.0	0.0	0.0	99.4	0.0	0.0		
130	0.0	5.1	0.0	94.1	0.0	0.0		
132	0.0	0.0	0.0	0.66	0.0	0.0		
134	0.0	0.0	0.0	97.4	0.0	0.0		
136	0.0	0.0	0.0	6.9	0.0	0.0		
138	0.0	0.0	0.0	98.0	0.0	0.0		
140	0.0	0.0	0.0	1.66	0.0	0.0		
142	0.0	0.0	0.0	99.0	0.0	0.0		
144	0.0	0.0	0.0	98.4	0.0	0.0		
146	0.0	0.0	0.0	100.0	0:0	0.0		
148	0.0	0.0	0.0	99.5	0.0	0.0		

		Mineral	logical Comp	osition of So	uth Core a	t North II	ngebright Lak	e
Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
150	0.0	0.0	0.0	2.66	0.0	00		•
152	0.0	0.0	0.0	99.5	0.0	0.0		
154	0.0	0.0	0.0	98.9	0.0	0.0		
156	0.0	4.8	0.0	94.7	0.0	0.0		
158	0.0	0.0	0.0	97.1	0.0	0.0		
160	0.0	0.0	0.0	100.0	0.0	0.0		
162	0.0	0.0	0.0	99,4	0.0	0.0		
164	0.0	0.0	0.0	99.2	0.0	0.0		
166	0.0	0.0	0.0	99.7	0.0	0.0		
168	0.0	0.0	0.0	99.3	0.0	0.0		
170	0.0	0.0	0.0	98.9	0.0	0.0		
172	0.0	0.0	0.0	99.2	0.0	0.0		
174	0.0	0.0	0.0	99.4	0.0	0.0		
176	0.0	0.0	0.0	1.66	0.0	0.0		
178	0.0	0.0	0.0	99,2	0.0	0.0		
180	0.0	4.7	0.0	94.9	0.0	0.0		
182	0.0	0.0	0.0	96.2	0.0	0.0		63.8
184	0.0	0.0	0.0	99.2	0.0	0.0		
186	0.0	0.0	0.0	100.0	0,0	0.0		
188	0.0	0.0	0.0	99.5	0.0	0.0		
190	0.0	0.0	0.0	96.8	0.0	0.0		
192	0.0	0:0	0.0	98.7	0.0	0.0		
194	0.0	0.0	0.0	99.2	0.0	0.0		
961	0.0	0.0	0.0	0.06	0.0	0.0		
198	0.0	0.0	0.0	99.4	0.0	0.0		
200	0.0	14.5	0.0	83.8	0.0	0'0	4.4	
202	0.0	0.0	0.0	99.2	0.0	0.0		
204	0.0	0.0	0.0	98.5	0.0	0.0		
226	0.0	0.0	0.0	99.5	0.0	0.0		
228	0.0	0.0	0.0	98.0	0.0	0.0		
230	0.0	0.0	0.0	91,4	0.0	0.0		
232	0.0	0.0	0.0	99.1	0.0	0.0		
234	0.0	0.0	0.0	6.99	0,0	0.0		
236	0.0	8.3	0.0	0'16	0.0	0.0		
238	0.0	0.0	0.0	99.4	0.0	0.0		

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Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
240	0.0	0.0	0.0	99.4	0.0	0.0	-	
242	0.0	0.0	0.0	99.1	0.0	0.0		
244	0.0	12.1	0.0	86.8	0.0	0.0		
246	0.0	12.5	0.0	84.8	0.0	0.0		
248	0.0	8.9	0.0	87.2	0.0	0.0		
250	0.0	0.0	0.0	97.0	0.0	0.0		0 69
252	0'0	9.6	0.0	88.1	0.0	0.0		6.70
254	0.0	0.0	0.0	98.9	0.0	0.0		
256	0.0	0.0	0.0	98.7	0.0	0.0		
258	0.0	6.5	0.0	93.0	0.0	0.0		
260	0.0	8.2	0.0	90.2	0.0	0.0		
262	0.0	17.6	0.0	78.9	0.0	0.0		
264	0.0	11.3	0.0	86.1	0.0	0.0		
266	0.0	10.6	0.0	86.4	0.0	0.0		
268	0.0	13.3	0.0	82.9	0.0	0.0		
270	0.0	0.0	0.0	75.2	0.0	0.0		
272	0.0	0.0	0.0	74.7	0.0	0.0		
274	0.0	0.0	0.0	97.6	0.0	0.0		
276	0.0	0.0	0.0	96.7	0.0	0.0		
278	0.0	0.0	0.0	69.1	0.0	0.0		50.7
280	0.0	10.9	0.0	68.2	0.0	0.0		
282	3.1	20.8	0.0	68,0	0.0	0.0		64 7
284	0.0	17.2	0.0	79.4	0.0	0.0		410
286	0.0	0.0	0.0	94.8	0.0	0.0		64.7
288	0.0	9.1	0.0	85.4	0.0	0.0		
290	2.5	7.3	0.0	51.9	0.0	0.0		61.6
292	0.0	6.0	0.0	56.7	0.0	0.0		58.3
294	0.0	0.0	0.0	58.1	0.0	0.0	7.4	6.7
296	0.0	12.2	0.0	82.2	0.0	0.0		5
298	2.2	L.T	0.0	69.5	0.0	0.0		
300	0.0	0.0	0.0	92.7	0.0	0.0		
302	0.0	6.5	0.0	6.99	0.0	0.0		46.4
304	0.0	0.0	0.0	74.9	0.0	0.0		65.6
306	0.0	0.0	0.0	95.0	0.0	0.0		7 L Y
308	0.0	6.1	0.0	73.0	0.0	0.0		

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Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
310	0.0	7.7	0.0	90.8	0.0	0.0		
312	0.0	0.0	0.0	97.4	0.0	0.0		
314	0.0	6.8	0.0	88.7	0.0	0.0		63.2
316	0.0	10.2	0.0	85.7	0.0	0.0		64,7
318	0.0	0.0	0.0	92.3	0.0	0.0		63.7
320	0.0	0.0	0.0	68.5	0,0	0.0	5.7	6.4
322	0.0	5.7	0.0	69.9	0.0	0.0		62.7
324	0.0	0.0	0.0	75.0	0.0	0.0		62.3
326	0.0	0.0	0.0	50.6	0.0	0.0		63,7
328	0.0	0.0	0.0	58.1	0.0	0.0	5.7	61.5
330	3.0	0.0	0.0	63.3	0.0	0.0		60.0
332	0.0	0.0	0.0	76.5	19.7	0.0		
334	0.0	5,3	0.0	87.2	0.0	0.0		6.2
336	0.0	0.0	0.0	91.4	0.0	0.0		62.5
338	0.0	0.0	0.0	66.4	0.0	0.0		62.3
340	0.0	8.1	0.0	89.4	0.0	0.0		66.9
342	0.0	0.0	0.0	92.7	0.0	0.0		6.7
344	2.5	0.0	0.0	73.0	0.0	0.0		62,3
346	3,1	0.0	0.0	62.4	0.0	0.0	6.0	62,6
348	0.0	5.0	0,0	66.8	0.0	0.0		66.0
350	0.0	0.0	0.0	98.6	0.0	0.0		
352	0.0	0.0	0.0	98.7	0.0	0.0		
354	0.0	6.9	0.0	85.5	0.0	0.0		61.6
356	0.0	0.0	0.0	98,7	0.0	0.0		
358	0.0	0.0	0.0	98.1	0.0	0.0		
360	0.0	0.0	0.0	91.8	0.0	0.0		62,8
362	0.0	· 0,0	0.0	73.3	0.0	0.0		
364	0.0	0.0	0.0	96.1	0.0	0.0		62,9
366	0.0	0.0	0.0	94.6	0.0	0.0		
368	0.0	5.1	0.0	93.5	0,0	0.0		
370	0.0	0,0	0.0	81.4	13.9	0.0		
372	0.0	0.0	0.0	97.0	0.0	0.0		63.7
374	0.0	4.8	0.0	92.2	0.0	0.0		6.3
376	0.0	0.0	0.0	98.0	0.0	0.0		
378	0.0	0.0	0.0	99.1	0.0	0.0		

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Mineralogical Composition of South Core at North Ingebright Lake

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Jepth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
380	0.0	0.0	0.0	93.2	0.0	00	•	007
382	0.0	6.2	0.0	91.0	0.0	0.0		00.9
384	0.0	8.7	0.0	88,0	0.0	0.0		
386	0.0	0.0	0.0	98.7	0.0	0.0		
388	0.0	0.0	0.0	1.79	0.0	0.0		
390	0.0	0.0	0.0	98.4	0.0	0.0		
392	0.0	0.0	0.0	17.1	0.0	0.0		1 (9
394	0.0	0.0	0.0	94.1	0.0	0.0		C.70
396	3.8	10.0	0.0	75.7	0.0	0.0		7 2 7
398	0.0	8.9	0.0	86.3	0.0	0.0		1.00
400	2.6	16.2	0.0	75.2	0.0	0.0		65.5
402	0.0	7.4	0.0	57.8	0.0	0.0		68.4
404	0.0	6.4	0.0	60.1	0.0	0.0		L '00
406	0.0	0.0	0.0	96.9	0.0	0.0		<i>k</i> 7 k
408	0.0	5.3	0.0	64.9	0.0	0.0		61.0
410	2.6	0.0	0.0	68.3	0.0	0.0		
412	2.2	6.5	0.0	59.6	0.0	0.0		
414	3.1	7.9	0.0	78.3	0,0	0.0	4.6	63.5
416	0.0	0.0	0.0	76.4	0.0	0.0	1	50.3
500	3.3	0.0	0.0	52.4	0.0	0.0		63.9
502	0.0	8.9	0.0	86.4	0.0	0.0		
504	0.0	0.0	0.0	68.1	24.5	0.0	4.9	
508	0.0	5.8	0.0	58.5	0.0	0.0	2	66.0
510	0.0	0.0	0.0	52.5	0.0	0.0		62.4
512	0.0	15.6	0'0	31.9	0.0	0.0		63.3
514	0.0	8.8	0.0	53.5	0.0	0.0		63.6
516	0'0	13.6	0.0	30.9	0.0	0.0		63.9
518	3.5	14.0	0.0	29.3	0.0	0.0		4 C 9
520	0.0	10.8	0.0	56.7	0.0	0.0		64.3
522	3.1	9.1	0.0	56.8	0.0	0.0		63.8
524	4.4	6.6	0.0	57.2	0.0	0.0		62.9
526	4.7	10.6	0.0	38.9	0.0	0.0		64.3
528	0.0	0.0	0.0	52.6	0.0	0.0		63.9
530	0.0	0.0	0.0	61.8	0.0	0.0		67.4
532	3.0	7.3	0.0	57.4	0.0	0.0		

Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
534	0.0	9.0	0.0	66.3	0.0	0,0		63.7
536	0.0	15.9	0.0	51.6	0.0	0.0		62.5
538	3.2	14.2	0.0	51.3	0.0	0.0		64,4
540	4.0	25.4	0.0	35.5	0.0	0.0		64.4
542	3.9	14.4	0.0	46.3	0.0	0.0		64,5
544	3.9	22.5	0.0	42.1	0.0	0.0		62.7
546	4.8	16.5	0.0	33.5	0.0	0.0		61.9
548	4.4	19.5	0.0	38,2	0.0	0.0		61.2
550	4.9	9.4	0.0	44.3	0.0	0.0		63.9
552	4.2	9.2	0.0	52.8	0.0	0.0		62.5
554	0.0	15.3	0.0	77.9	0.0	0.0		61,1
556	0.0	22.5	0.0	42.1	0.0	0.0		63.7
558	4.5	16.5	0.0	47.3	0.0	0.0		63.5
560	0.0	8.8	0.0	57.6	0.0	0.0		62.7
562	0.0	11.6	0.0	48.6	0.0	0.0		63.6
564	0.0	13.5	0.0	57.4	0.0	0.0		63.4
566	3.3	10.3	0.0	52.2	0.0	0.0		63.7
568	2.3	24.9	0.0	49.6	0.0	0.0		65.3
570	2.1	17.3	0.0	57.8	0.0	0.0		65,2
572	0.0	0.0	0.0	71.5	0.0	0.0		66,2
574	4.1	14.5	0.0	76.7	0.0	0.0		64.5
576	3,3	13.6	0.0	58.2	0.0	0.0		63.2
578	3.7	8,9	0.0	55.3	0.0	0.0		62.5
580	0.0	19.4	0.0	53.9	0.0	0.0		62.4
582	0.0	19.7	0.0	57.1	0.0	0.0		63.6
584	3.1	19.8	0.0	71.8	0.0	0,0		68,7
586	0.0	29.5	0.0	61.8	0.0	0.0	3.3	
588	3.4	22.9	0.0	49.1	0.0	0.0		62.6
590	0.0	15.6	0.0	59.7	0.0	0.0		63.2
592	0.0	18.9	0.0	58.8	0.0	0.0	12.6	61,7
594	0.0	9.8	0.0	70.7	0.0	0.0	18.1	64.6
596	0.0	18.3	0.0	59.3	0.0	0.0		6.6
598	2.7	25.1	0.0	44.2	5.7	0.0		
600	0.0	12.8	0.0	56.7	0.0	0.0		62.9
602	4.1	11.4	0.0	61.4	0.0	0.0		62.6

Mineralogical Composition of South Core at North Ingebright Lake

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Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
604	0.0	8.4	0.0	57.2	0.0	0.0		65.2
606	3.8	14.9	0.0	58.1	0.0	0,0		63,3
608	0.0	17.0	0.0	57.9	0.0	0.0		63.9
610	2.7	12.3	0.0	63.5	0.0	0.0		63.9
612	0.0	15.1	0.0	82,9	0.0	0.0		62.7
614	0.0	0.0	0.0	94.4	0.0	0.0		63.0
616	0.0	21.7	0.0	51.3	0.0	0.0	5.3	
618	0.0	27.9	0.0	69.6	0.0	0.0		
620	0.0	18.9	0.0	66.6	0.0	0.0	4.7	
622	0.0	21.3	0.0	68.7	0.0	0.0		
624	0.0	20.3	0.0	77.1	0.0	0.0		6.8
626	0.0	34.7	0.0	41.7	0.0	0.0	3.9	63.9
628	0.0	17.4 [·]	0.0	69.7	0.0	0.0		
630	0.0	7.9	0.0	45.5	22.9	0.0		
632	0.0	18.7	0.0	73.2	0.0	0.0	4.8	
634	0.0	13.7	0.0	58.6	21.0	0.0	4.5	
636	0.0	4.3	0.0	93.1	0.0	0.0		
638	0.0	0.0	0.0	89.2	0.0	0.0		64.6
640	0.0	9.4	0.0	67.6	0.0	0.0		66.2
642	0.0	0.0	0.0	84.6	0.0	0.0	4.4	
644	5.6	0.0	0.0	65.4	20.5	0.0	4.5	61.4
646	0.0	24.3	0.0	60.0	0.0	0.0		61,5
648	0.0	10.6	0.0	84.7	0.0	0.0		62.5
650	0.0	7.3	0.0	63.9	0.0	0.0	7.5	6,5
652	0.0	9.8	0.0	84.2	0.0	0.0		
654	0.0	11.1	0.0	87.8	0.0	0.0		
656	0.0	10.9	0.0	88.6	0.0	0.0		
658	0.0	6.7	0.0	92.9	0.0	0,0		
660	0.0	0.0	0.0	100.0	0.0	0.0		
662	0.0	61.4	0.0	36.6	0.0	0.0		
664	0.0	24.4	0.0	75.0	0.0	0.0		
666	0.0	51.8	0.0	44.8	0.0	0.0		62,8
668	0.0	31.3	0.0	62.1	0.0	0,0		
670	0.0	38.7	0.0	56.8	0.0	0.0		61.6
672	1.6	69.6	0.0	25.3	0.0	0.0		

Mineralogical Composition of South Core at North Ingebright Lake

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Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
674	0.0	38.7	0.0	58.8	0,0	0.0		
676	0.0	37.9	0.0	60.2	0.0	0.0		6.2
678	0.0	44.9	0.0	34.8	0.0	0'0		63.9
680	0.0	18.6	0.0	57.6	0.0	0.0		61.3
682	0.0	65.6	0.0	29.1	0.0	0.0		6'19
684	0.0	32.6	0.0	50.5	0.0	0.0		62.6
686	0.0	17.7	0.0	64.3	0.0	0.0		62.3
824	0.0	16.9	0.0	82.8	0.0	0.0		
826	0.0	26.3	0.0	72.8	0.0	0'0		
828	0.0	29.7	0.0	67,9	0.0	0.0		
830	0.0	11.5	0.0	84.7	0.0	0.0		
832	0.0	38.6	0.0	46.2	0.0	0'0		
834	0.0	10.0	0.0	88.6	0.0	0.0		
836	0.0	0.11	0.0	56.2	0.0	0'0		65.7
838	0.0	0.0	0.0	73.6	0.0	0.0		
840	0.0	0.0	0.0	63.1	0.0	0.0		61.3
842	1.4	0.0	0.0	97.2	0.0	0.0		
844	4.2	14.5	0.0	39.4	0.0	0'0		63.9
846	4.9	23.6	0.0	31,4	0.0	0.0		62.8
848	0.0	7.2	0.0	92.1	0.0	0.0		
850	0.0	5.8	0.0	60,1	0.0	0.0	15.3	65.7
852	0.0	0.0	0.0	90.5	0.0	0.0		62.8
854	1.8	63.6	0.0	14.3	0.0	0.0		64.2
856	0.0	6.1	0.0	67.0	0.0	0.0		62.7
858	0.0	0.0	0.0	71.5	0.0	0.0		63.5
860	0.0	27.2	0.0	68.3	0'0	0.0		63.2
862	0.0	8.5	0.0	89.5	0.0	0.0		62,6
864	3.2	17.8	0.0	55.1	0.0	0.0		61,8
866	3.9	17.7	0.0	45.4	0.0	0.0		61.5
868	2.1	18.1	0.0	14.6	0.0	0,0		62.9
870	0,0	77.5	0.0	4.8	0.0	0.0	4.7	
872	3.5	71.7	0.0	3.5	0.0	0.0		
874	0.0	86.5	0.0	6.5	0.0	0.0		
876	0.0	66,1	0.0	4.9	0.0	0.0		
878	0'0	85.9	0.0	2.9	0.0	0.0	3.8	

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Depth (cm)	Magnesite	Gypsum	Anhydrite	Thenardite	Bloedite	Halite	Mg in High Mg Calcite	Ca in Protodolomite
880	0.0	80.7	0.0	1.7	0.0	0.0		
882	0.0	64.4	0.0	4.1	0.0	0.0		65.4
884	0.0	61.7	0.0	11.0	0.0	0.0		
886	0.0	88.2	0.0	5.4	0.0	0.0		63.7
888	0.0	78.4	0.0	4.2	0.0	0.0		49.7
890	0.0	55.0	0.0	1.6	0,0	0.0		
892	0.0	79,8	0.0	2.4	0.0	0.0		
894	0.0	50.0	0.0	1.9	0,0	0.0	19.0	
896	0.0	78.8	0.0	4.1	0.0	0.0		
898	0.0	5.6	0.0	6,4	0.0	0.0		
900	0.0	0.0	0.0	31.4	0.0	0.0		
910	0.0	4.0	0.0	6,8	0.0	0.0		

Appendix C Detailed Evaporite Mineralogy from µPDSM

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Depth (cm)	Ankerite Ca(Fe,Mg)	Benstonite Ca ₇ Ba ₆	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Arcanite (K,NH ₄) ₂	Epsomite MgSO₄	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H-O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H-O	Krausite KFe(SO ₄) ₂	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄	Mercallite KHSO₄	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H-O	Starkeyite MgSO4	Wattevilleite Na ₂ Ca(SO ₄) ₂
	(003)	(003)13		51120	304	mo	21120	21120	1120	21120		21120	21110	41120
12										x				
14														
16					x					V			x	
18										~				
20														
24										х				
26														
28														
30		v			х									
32 34		х												
36	x								-					
38														
40														
42														
44	v								v					
40	X Y								Х					
50	л													
52														
54														
56	х													
58										х				
60														
62														
66														
68										х				
70														
72														
74										x				
76										Х				
70 80														
82														

Depth (cm)	Ankerite Ca(Fc,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na2Ca(SO4)3 2H2O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K2Ca2Mg(SO4)4 2H2O	Starkcyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
84														
86														
88														
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156														
158														

Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na2Ca(CO3)2 5H2O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na2Ca(SO4)3 2H2O	Gorgeyite K ₂ Ca ₃ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
160														
162														
104														
168														
170														
172														
174														
176														
178														
182														
184										•				
188														
190														
192														
194														
198										х				
200														
202														
204														
206									x					
208														
212														
214											х			
216												•		
218														
220					v									
222					л Х									
226					~									
228														
230														
232		х												

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Depth	Ankerite	Benstonite	Fairchildite	Gaylussite	Arcanite (K NHL)	Epsomite MaSO	Eugsterite	Gorgeyite	Krausite KRe(SQ.)	Leightonite	Mercallite	Polyhalite K. Ch. Ma(SO.)	Starkeyite	Wattevilleite
(em)	(CO.).	(CO.).	NjCa(COj)j	54.0	(N)(4)) SO.	74.0	2010	2043(304)6		202000414	KNOU	N2C82MIg(304)4	Mg3O4	1982C8(304)2
	(003)2	(003)13		51120	304	/1120	21120	21120	1120	2020		2020	2020	4120
234														
236														
238														
234														
242											X			
244					x									
246														
248						v						x		
250						X						x		
252					v	Y						X		
256					~	~					x	x		
258						х						x		
260								х						
262														
264					х						х			
266														
268												х		
270														
272												х		
274												X		
276											Х	х		
2/8												v		
280		v										Х		
284		Λ												
286														
288												х [.]		
290														
292														
294														
296										х				
298														
300												x		
302												х		
304														

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Dept	Ankerite	Benstonite	Fairchildite	Gaylussite	Arcanite	Epsomite	Eugsterite	Gorgeyite	Krausite	Leightonite	Mercallite	Polyhalite	Starkeyite	Wattevilleite
(cm) Ca(Fc,Mg)	Ca,Ba ₆	K ₂ Ca(CO ₃)	Na ₂ Ca(CO ₃) ₂	(K,NH4) ₂	MgSO4	Na ₂ Ca(SO ₄) ₃	K ₂ Ca ₅ (SO ₄) ₆	KFe(SO4)2	K2Ca2Cu(SO4)	KHSO4	K ₂ Ca ₂ Mg(SO ₄) ₄	M _B SO ₄	Na ₂ Ca(SO ₄) ₂
	(CO ₃)	(CO ₃) ₁₃		5H ₂ O	SO4	7H ₂ O	2H ₂ O	2H20	H ₂ O	2H20		2H ₁ O	2H ₂ O	4H ₂ O
306														
308														
310								×						
312														
314														
316														
318														
320														
322														
324														
326														
328			×											

277

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Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca7Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na2Ca(SO4)3 2H2O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO4)2 H2O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO ₄	Polyhalite K ₂ Ca ₂ Mg(SO4) ₄ 2H ₅ O	Starkeyite MgSO4 2H ₂ O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
380														
382														
384														
386														
285 005														
392														
394														
396														
398														
400														
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404														
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420														
422														
424														
426														
428														
430														
432			×											
434												•		
436														
438														
440														
442														
444														
446														
448														
450														

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Depth (cm)	Ankerite Ca(Fc,Mg) (CO ₃)2	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO4 7H2O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
452														
454														
456														
458														
460														
462								x						
464														
466														
468														
470														
4/2														
474														
478														
480														
482														
484														
486														
488														
490														
492					v									
494					X									
490														
500														
500														
504														
506														
508														
510														
512														
514														
516														
518					v									
520					~									
344														

rkcyitc Wattevilleite gSO4 Na ₂ Ca(SO4) ₂ H ₃ O 4H,O																					>	×											
Polyhalite Star K ₂ Ca ₂ Mg(SO4), M ₁ 2H ₂ O 2h																																	
Mercallite KHSO4																																	
Leightonite K ₁ Ca ₁ Cu(SO ₄) ₄ 2H ₂ O	-																																
Krausite KPe(SO4)2 H2O																																	
Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O																																	
 Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O																																	
Epsomite MgSO ₄ 7H ₂ O																																	
Arcanite (K,NH ₄) ₁ SO ₄											×	:																					
Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O																					×												
Fairchildite K ₂ Ca(CO ₃) ₂																									:	×							
Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃																																	
Ankerite Ca(Fe,Mg) (CO ₃)2																																	
Depth (cm)	526	528	532 534	536	538	540	542	945	540 548	550	552	554	556	558	560	564	566	568	570	572	574	576	8/5	080	700	+00	090	880	0,605	594	596	598	

Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Arcanite (K,NH4)2 SO4	Epsomite MgSO4 7H2O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO ₄	Polyhalite K2Ca2Mg(SO4)4 2H2O	Starkcyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
602														
604														
606														
616														
618														
620														
622														
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626														
628														
630														
634					х									
636														
638														
640														
642														
044 646														
648														
650														
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654														
656														
658														
660														
662												x.		
664														
666														
668														
670														
674														
676														
678														
680														

Depth	Ankerite	Benstonite	Fairchildite	Gaylussite	Arcanite	Epsomite	Eugsterite	Gorgeyite	Krausite	Leightonite	Mercallite	Polyhalite	Starkeyite	Wattevilleite
(cm)	Ca(Fe,Mg)	Ca ₇ Ba ₆	$K_2Ca(CO_3)_2$	$Na_2Ca(CO_3)_2$	(K,NH ₄) ₂	MgSO4	$Na_2Ca(SO_4)_3$	K ₂ Ca ₅ (SO ₄) ₆	$KFe(SO_4)_2$	$K_2Ca_2Cu(SO_4)_4$	KHSO₄	K ₂ Ca ₂ Mg(SO ₄) ₄	MgSO4	Na ₂ Ca(SO ₄) ₂
	$(CO_3)_2$	(CO ₃) ₁₃		5H ₂ O	SO4	7H ₂ O	2H2O	2H ₂ O	H ₂ O	2H ₂ O		2H ₂ O	2H ₂ O	4H ₂ O
682														
684														
686														
688			х											
690														
692														
694														
696														
698			х											
700														
702														
704														
706			.,											
708			X											
710														
714														
718														
770														
720														
724														
728														
730														
732														
734														
736														
738			х											
740														
746														
748														
750														
752								х						
754														
756														
758														
760														

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Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca7Ba6 (CO3)13	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO ₄ 2H ₂ O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
762														
764														
792														
794														
798														
800														
802														
804														
800						v								
810						~								
812														
814														
816													х	
818														
820														
824														
826														
828														
830												v		
832											v	X		
836					x						~			
838					x									
840														
842														
844														
840 849														
850														
852														
856														
860														
862														

Depth (cm)	Ankerite Ca(Fe.Mo)	Benstonite Ca. Ra.	Fairchildite	Gaylussite	Arcanite	Epsomite	Eugsterite	Gorgeyite	OGY (INOF Krausite	th Core) Leightonite	Mercallite	Polyhalite	Starkeyite	Wattevillcite
(111)	Ca(re,Mg) (CO ₃) ₂	(CO)	K ₂ ca(cU ₃) ₂	Na ₂ Ca(CO ₃) ₂ 5H ₂ O	(K,NH4,)2 SO4	MgSO4 7H2O	Na ₂ Ca(SO ₄), 2H ₂ O	K ₂ Ca ₅ (SO4) ₆ 2H ₂ O	KFe(SO4)2 H2O	K ₂ Ca ₂ Cu(SO ₄), 2H ₂ O	KHSO4	K ₂ Ca ₂ Mg(SO ₄), 2H ₂ O	MgSO4 2H3O	Na ₂ Ca(SO ₄) ₂ 4H ₂ O
864 866														
868														
870 872														
874														
876														
878														
882														
884														
886														
888														
890														
892														
896														
898														
906														
902												>		
904												×		
906														
908														
016														
912														
914														
016												×		
010												×		
076														
924														
926														
928														
930														
932														
934														
936														

Depth (cm)	Ankerite Ca(Fe,Mg)	Benstonite Ca ₇ Ba ₆	Fairchildite K2Ca(CO3)2	Gaylussite Na2Ca(CO3)2	Arcanite (K,NH ₄) ₂	Epsomite MgSO4	Eugsterite Na ₂ Ca(SO ₄) ₃	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆	Krausite KFe(SO ₄) ₂	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄	Starkeyite MgSO4	Wattevilleite Na ₂ Ca(SO ₄) ₂
	(CO ₃) ₂	(CO ₃) ₁₃		5H2O	SO4	7H2O	2H ₂ O	2H ₂ O	H ₂ O	2H ₂ O		2H ₂ O	2H ₂ O	4H ₂ O
010														
040														
942														
944														
946														
948														
950														
952														
954														
950														
960														
962														
964														
972														
974														
976														
978														
980														
984														
986														
988														
990														
992														
994								x						
990														
1000												x		
1002														
1004														
1006								х						
1008														
1010														
1012											v			
1014											X			

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Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na2Ca(CO3)2 5H2O	Arcanite (K,NH4)2 SO4	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K2Ca5(SO4)6 2H2O	Krausite KFc(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
1016					x									
1018														
1020														
1022														
1024												x		
1020														
1020											Y			
1032											~			
1034											х			
1036														
1038														
1040														
1042														
1044														
1046				v									v	
1048				~									X	
1050														
1054														
1054														
1058														
1060														
1062														
1064														
1066														
1068														
1070												x		
1072			x											
1074													х	
1070														
1078														
1080														
1084														
1086														

Depth (cm)	Ankerite Ca(Fc,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na2Ca(CO3)2 5H2O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO₄ 7H₂O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K2Ca5(SO4)6 2H2O	Krausite KFe(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K2Ca2Mg(SO4)4 2H2O	Starkcyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
1088														
1090														
1092														
1094														
1090														
1100			х											
1102			x											•
1104			х											
1106														
1108														
1110														
1112														
1114														
1118			х											
1120														
1122														
1124			Х											
1126														
1128														
1130														
1132														
1134														
1138														
1140														
1142			Х									•		
1144														
1148			.,											
1150			X											
1152			x											
1156														
1158														
1160														

Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussitc Na2Ca(CO3)2 5H2O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO4 7H2O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO4)2 H2O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K2Ca2Mg(SO4)4 2H2O	Starkcyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
1162														
1164														
1166														
1168														
1170														
1172														
1174														
1178														
1180														
1182														
1184														
1186														
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1190														
1192														
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1444														
1328														
1330														
1332														
1334														

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Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na2Ca(CO3)2 5H2O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO4)2 H2O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
1336														
1338														
1340														
1342														
1344														
1340														
1350														
1352			x											
1354														
1356								х						
1358														
1360														
1362														
1304														
1368														
1370														
1372			х											
1374														
1376					х									
1380														
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1388							v						v	
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1392							X							
1396														
1398														
1400														
1402														
1404														
1406														
1408														

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Depth (cm)	Ankerite Ca(Fc,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K2Ca(CO3)2	Gaylussite Na2Ca(CO3)2 5H2O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na2Ca(SO4)3 2H2O	Gorgeyite K2Ca5(SO4)6 2H2O	Krausite KFe(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
1410														
1412														
1414														
1410														
1410														
1420														
1424														
1426														
1428														
1430														
1432														
1434														
1436														
1438			x											
1440			v											
1442			~											
1446								х					х	
1448														
1450														
1452														
1454														
1456														
1458														
1460														
1462														
1464														
1400			x											
1400			~											
1472														
1474														
1476														
1478														
1480														

Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca7Ba6 (CO3)13	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na2Ca(CO3)2 5H2O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO4 7H2O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Krausite KFe(SO4)2 H2O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
1482														
1484														
1480														
1400														
1490														
1494														
1496														
1498														
1500														
1504														
1506														
1508														
1510														
1512			x											
1514														
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1522														
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1520														
1520														
1532														
1534														
1536									•					
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1540														
1542														
1544														
1546														
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1552														
1554				Х										

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Depth (cm)	Ankerite Ca(Fe,Mg)	Benstonite Ca ₇ Ba ₆	Fairchildite K2Ca(CO3)2	Gaylussite Na2Ca(CO3)2	Arcanite (K,NH ₄) ₂	Epsomite MgSO4	Eugsterite Na ₂ Ca(SO ₄) ₃	Gorgeyite K2Ca5(SO4)6	Krausite KFe(SO ₄) ₂	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄	Starkcyite MgSO4	Wattevilleite Na ₂ Ca(SO ₄) ₂
	(CO ₃) ₂	(CO ³) ¹³		5H ₂ O	SO4	7H2O	2H ₂ O	2H ₂ O	H₂O	2H ₂ O		2H2O	2H ₂ O	4H ₂ O
1556														
1558														
1550														
1562														
1564														
1566														
1568														
1570														
1572														
1574														
1576														
1578														
1580														
1586														
1588														
1590														
1592														
1594														
1596														
1598														
1600														
1602														
1604														
1606														
1608														
1610														
1012														
1014														
1619														
1620														
1620														
1624														
1626														
1628														

Depth (cm)	Ankerite Ca(Fc,Mg)	Benstonite Ca ₇ Ba ₆	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂	Arcanite (K,NH ₄) ₂	Epsomite MgSO4	Eugsterite Na ₂ Ca(SO ₄) ₃	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆	Krausite KFe(SO ₄) ₂	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄	Starkeyite MgSO4	Wattevilleite Na ₂ Ca(SO ₄) ₂
	$(CO_3)_2$	(CO ₃) ₁₃		5H2O	S04	7H ₂ O	2H ₂ O	2H ₂ O	H ₂ O	2H ₂ O		2H ₂ O	2H ₂ O	4H ₂ O
1630														
1632														
1634														
1636														
1638														
1640														
1642														
1644														
1646												•		
1648														
1650														
1052														
1054														
1658														
1660														
1662														
1664													х	
1666														
1668														
1670														
1672														
1674														
1676														
1678														
1080														
1/30														
1734														
1736														
1738													х	
1740														
1742														
1744														
1746														
1748														

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1750 1754 1756 1768 1768 1768 1768 1770 1770 1772 1777 1778 1778 1788	Depth (cm)	Ankcrite Ca(Fe,Mg) (CO ₃) ₂	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na2Ca(CO3)2 5H2O	Arcanite (K,NH ₄) ₂ SO ₄	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K2Ca5(SO4)6 2H2O	Krausite KFe(SO ₄) ₂ H ₂ O	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O	Mercallite KHSO4	Polyhalite K2Ca2Mg(SO4)4 2H2O	Starkeyite MgSO4 2H2O	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O
1752 X 1754 X 1756 1760 1764 1765 1766 1770 X 1771 1772 1773 1774 1775 1776 1778 1779 1778 1779 1780 1781 1782 1786 1787 1788 1799 1792 1793 1794 1795 1796 1797 1798 1799 1800 1801	1750														
1754 X 1756 1758 1764 1765 1766 1767 1768 1779 X 1770 X 1771 1772 1778 1780 1781 1782 1783 1784 1785 1786 1788 1789 1790 1791 1792 1793 1804 1805 1806 1807 1808 1809 1810 1811 1812 X	1752														
1756 1758 1760 1764 1765 1766 1777 1778 1779 1774 1775 1776 1778 1778 1778 1778 1780 1781 1782 1785 1786 1787 1788 1788 1789 1780 1781 1782 1783 1784 1785 1786 1787 1788 1788 1790 1792 1794 1795 1800 1801 1802 1803 1804 1814 1816 1818 1820 1818 1820 1820 1820 <t< td=""><td>1754</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>х</td><td></td><td></td><td></td><td></td><td></td><td></td></t<>	1754								х						
178 1760 1762 1764 1765 1770 1771 1772 1772 1774 1775 1776 1778 1780 1781 1782 1788 1788 1788 1788 1789 1790 1794 1795 1796 1798 1800 1801 1802 1803 1804 1805 1806 1808 1808 1810 1812 1814 1816 1818 1818	1756														
1760 1762 1764 1765 1770 X 1771 1776 1778 1780 1781 1782 1782 1783 1784 1780 1782 1783 1784 180 180 180 1802 1803 1804 1803 1810 1814 1818 1818 1818	1758														
1762 1764 1766 1708 1771 1772 1774 1776 1778 1780 1780 1782 1788 1788 1790 1902 1903 1904 1905 1908 1800 1803 1804 1805 1808 1808 1814 1818 1818	1760														
1764 1766 1770 X 1777 1774 1775 1776 1778 1780 1782 1788 1790 1792 1794 1795 1796 1797 1800 1800 1802 1803 1804 . 1808 1810 1812 X 1814 1818 1818	1762														
1766 1770 X 1771 1774 1776 1778 1780 1782 1788 1790 1792 1793 1794 1795 1800 1801 1802 1803 1804 1812 X 1814 1816 1818 1818	1764														
1768 1770 X 1771 1774 1776 1778 1780 1781 1782 1786 1787 1788 1790 1791 1792 1794 1795 1796 1800 1800 1801 1802 1803 1814 1816 1818	1766														
1770 x 1772 1776 1778 1780 1781 1782 1783 1794 1795 1796 1798 1800 1802 1804 1805 1806 1812 X 1814 1816 1818	1768				••										
1/72 1776 1778 1780 1782 1786 1788 1788 1790 1792 1794 1796 1796 1798 1800 1802 1802 1804 1806 1802 1804 1804 1814 1816 1818	1770				x										
1/74 1776 1778 1780 1782 1786 1788 1788 1790 1792 1794 1794 1794 1795 1798 1800 1800 1800 1802 1804 1806 1806 1812 1814 1816 1818	1774														
1776 1780 1782 1786 1788 1790 1792 1794 1796 1800 1800 1800 1802 1804 1804 1806 1808 1810 1810 1814 1816 1816 1818	1776														
1780 1780 1782 1786 1787 1796 1798 1800 1802 1804 1805 1806 1810 1811 1816 1818 1816 1818	1778														
1782 1786 1788 1790 1792 1794 1796 1798 1800 1802 1804 1806 1806 1808 1810 1812 1814 1816 1818 1816 1818	1780														
1786 1788 1790 1792 1794 1796 1798 1800 1800 1802 1804 1804 1806 1808 1810 1810 1812 1814 1816 1816 1818	1782														
1788 1790 1792 1794 1796 1798 1800 1802 1804 1804 1806 1806 1808 1810 1810 1810 1812 X 1814 1816 1816 1816 1818	1786														
1790 1792 1794 1796 1798 1800 1802 1804 1804 1806 1808 1810 1810 1812 X 1814 1816 1816 1818	1788														
1792 1794 1796 1798 1800 1802 1804 1806 1808 1810 1812 1812 1814 1816 1816 1818	1790														
1794 1796 1798 1800 1802 1804 1806 1808 1810 1812 X 1814 1814 1816 1818	1792														
1796 1798 1800 1802 1804 1806 1808 1810 1812 1814 1814 1816 1818 1818	1794														
1798 1800 1802 1804 1804 1806 1808 1810 1812 X 1814 1816 1816 1818 1818 1818 1818 1818	1796														
1800 1802 1804 1806 1808 1810 1812 X 1814 1816 1818 1818	1798														
1802 1804 1806 1808 1810 1812 X 1814 1816 1818 1818	1800														
1804 1806 1808 1810 1812 X 1814 1816 1818	1802														
1806 1808 1810 1812 X 1814 1816 1818 1820	1804												•		
1808 1810 1812 X 1814 1816 1818	1806														
1810 1812 X 1814 1816 1818 1820	1808														
1812 A 1814 1816 1818	1810			v											
1816 1818 1820	1812			X											
1818	1014														
1820	1010														
	1820														
1822	1822														

Depth	Ankerite Ca(Fe Mg)	Benstonite	Fairchildite	Gaylussite	Arcanite (K NH.)	Epsomite MaSO	Eugsterite	Gorgeyite K.Ca.(SO.)	Krausite KEe(SO)	Leightonite K-Ca-Cu(SO)	Mercallite	Polyhalite K-Ca-Ma(SO)	Starkeyite MaSO	Wattevilleite
(ciii)	(CO ₃) ₂	$(CO_3)_{13}$	K2C8(CO3)2	5H ₂ O	(N, (N114)2 SO4	7H ₂ O	2H ₂ O	2H ₂ O	H ₂ O	2H2O	клоц	2H ₂ O	2H ₂ O	4H2O
1824														
1826														,
1828														
1830														
1832														
1834														
1912														
1914														
1916														
1918														
1920			x											
1922														
1924														
1926														
1928														
1930														
1932														
1934														
1938														
1940														
1942														
1944														
1946														
1948														
1950														
1952														
1954														
1050														
1060														
1960														
1964														
1966														
1968														

	/attevilleite a2Ca(SO4)2 4H,0																			
	Starkcyite W MgSO4 Ni 2H,O	-																		
	Polyhalite K ₂ Ca ₂ Mg(SO4)4 2H ₂ O																			
	Mercallite KHSO4																			
	Leightonite K ₂ Ca ₂ Cu(SO ₄) ₄ 2H ₂ O																			
, t	Krausite KFe(SO4)2 H2O																			
	Gorgeyite K2Ca5(SO4)6 2H2O																			
	Eugsterite Na2Ca(SO4)3 2H2O																			
	Epsomite MgSO ₄ 7H ₂ O																			
	Arcanite (K,NI4,) ₂ SO4																			
	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O																			
	Fairchildite K ₂ Ca(CO ₃) ₂																			
	Benstonite Ca ₃ Ba ₆ (CO ₃),																			
	Ankcritc Ca(Fe,Mg) (CO ₃) ₂																			
	Depth (cm)	1970 1974 1976 1978	1982 1984 1986	1988 1990 1992	1994 1996	2002	2006	2010	2012	2014 2016	2018	2022	2024	2028	2030	2034	2036	2038	2040	2042

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Depth (cm)	Ankerite Ca(Fe,Mg)	Benstonite Ca ₇ Ba ₆	Fairchildite K ₂ Ca(CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂	Arcanite (K,NH ₄) ₂	Epsomite MgSO4	Eugsterite Na ₂ Ca(SO ₄) ₃	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆	Krausite KFe(SO ₄) ₂	Leightonite K2Ca2Cu(SO4)4	Mercallite KHSO4	Polyhalite K2Ca2Mg(SO4)4	Starkeyite MgSO4	Wattevilleite Na ₂ Ca(SO ₄) ₂
	(CO ₃) ₂	(CO ₃) ₁₃		5H ₂ O	SO₄	7H₂O	2H ₂ O	2H ₂ O	H ₂ O	2H ₂ O		2H ₂ O	2H ₂ O	4H ₂ O
2046														
2048														
2052														
2054														
2056														
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2064														
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2084														
2086														
2088														
2090														
2092														
2094														
2090														
2098														
2100														
2102														
2104														
2100														
2110														
2112														
2114														
2118														
2120														

2122 2124 2126 2130 2132 2134 2136 2138	/illeite (SO ₄) ₂ ₂ O
2124 2126 2128 2130 2132 2134 2136 2138	
2126 2128 2130 2132 2134 2136 2138	
2128 2130 2132 2134 2136 2138	
2130 2132 2134 2136 2138	
2132 2134 2136 2138	
2134 2136 2138	
2136	
2138	
2140	
2142	
2144	
2146	
2148	
2150	
2152	
2154	
2156	
2158	
2102	
2164	
2168	
2170	
2172	
2174	
2176	
2178	
2180	
2182	
2184	
2180	
2100	
2190	

	Wattevilleite Na ₂ Ca(SO ₄) 4H ₂ O	
	Starkeyite MgSO ₄ 2H ₂ O	
	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	
	Mercallite KHSO4	
•	Leightonite K2Ca2Cu(SO4)4 2H2O	
Ż	Krausite KFc(SO4)2 H2O	
	Gorgeyite K ₂ Ca ₅ (SO4) ₆ 2H ₂ O	
•	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	
	Epsomite MgSO ₄ 7H ₂ O	
	Arcanite (K,NH ₄) ₂ SO ₄	
	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	
	Fairchildite K ₂ Ca(CO ₃) ₂	
	Benstonite Ca ₇ Ba ₆ (CO ₃) ₁₃	
	Ankcrite Ca(Fe,Mg) (CO ₃)2	
	Depth (cm)	2194 2195 2196 2196 2200 2200 2200 2200 2211 2220 2221 2221 2223 2223

		tevilleite	Ca(SO4)2	0 ^t H													
		e Wat	Na ₂ C	4													
		Starkcyi	MgSO4	2H ₂ 0													
		Polyhalite	K ₂ Ca ₂ Mg(SO ₄),	O ^t H7													
		Mercallite	KHSO4														
th Core)		Leightonite	K2Ca2Cu(SO4)4 2H-O	07													
ogy (Nor		Krausite	hre(sU4)2 H,O	•													
te Mineral		Gorgeyite	2H20														
l Evapori		Eugsterite	2H20														
Detaile		Epsomite MgSO.	7H20														
		(K,NH ₄),	SO4														
	Gavluseite	Na ₂ Ca(CO ₃) ₂	5H20														
	Fairchildite	K ₂ Ca(CO ₃) ₂															
	Benstonite	Ca ₇ Ba ₆	(CO ₁) ₁₁														
	Ankerite	Ca(Fe,Mg)	(co ₃)														
	Depth	(cm)	:	2268	2270	2272	2274	2276	2280	2282	2284	2286	2288	2290	2292	2294	2296

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Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na6Mg2(SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
12													
14													
16													
18												v	
20												л	
24													
26													
28										х			
30									х				
32													
34										v			
30										~		x	
40												x	
42												х	
44													
46												x	
48												x	
50												v	
54												~	
56													
58													
60													
62							х						
64												. v	
00 69												X	
70													
72													
74													
76													
78			х										
80													
82													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreckite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na6Mg2(SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
84													
86													
88													
90													
92													
94			v										
90			~										
100													
106			х										
108													
110													
112													
114													
116													
118													
120													
122													
124													
128													
130			х										
132													
134												х	
136													
138													
140													
142													
144													
140													
148													
150													
152													
156													
158													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO₄)Cl 2,75H₂O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na6Mg2(SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
160												x	
162													
164													
100													
170													
172													
174													
176										х			
178													
180			х										
182													
188													
190													
192												х	
194			х										
196													
198					x								
200													
202													
204			x										
208			~										
210			х										
212			х										
214													
216													
218			х										
220												x	
222			х										
224												х	
220												••	
230													
232			х										

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Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkcite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na6Mg2(SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
234													
236													
238													
234			х										
242												x	
244													
240			v										
240			x										
252			Х										
254													
256													
258			х										
260			х										
262													
264			х										
266													
268													
270			Х										
272												x	
274			х										
276													
2/8													
280													
202													
204													
200													
200													
290													
294													
296													
298													
300												х	
302			х										
304			х										

,

Sepiolite Mg,Si ₆ O ₁₅	04110-1(110)																				
Inderborite CaMgB ₆ O ₁₁	D.				×	;															
Nitratine NaNO ₃																					
Niter KNO3																					
Korshunovskite Mg2Cl(OH) ₃ 2H.O																					
Camallite KMgCl ₃ 6H.O																					
Bischofite MgCl ₂ .6H ₂ O																					
Tychite Na ₆ Mg ₂ (SO ₄) (CO ₁) ₄														×							
Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O																					
Kainite KMg(SO4)CI 2.7511 ₂ 0	•																				
Hanksite KNa ₂₂ (CO ₃) ₂ (SO ₄) ₉ Cl	×		×		×		×	×		;	×				:	×					
Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)																					
Bradlcyite Na ₃ Mg(PO4) (CO3)																					
Depth (cm)	306 308 310	312	318 320	322	326 328	332 332	336	340	342 344	346	348	352	356	358	360	364	368	370	374	376	8/1

Markano, Muckan, Handa, Kananovakia Markanovakia Markanov					:	:							
хх х х	Bradleyite Na,Mg(PO4) (CO3)	Burkcite) Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa ₂₂ (CO ₃) ₂ (SO ₄) ₉ Cl	Kainite KMg(SO4)Cl 2.75H ₂ O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO4) (CO ₃)4	Bischofite MgCl ₂ ,6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Karshunovskite Mg2Cl(OH), 2H2O	Niter KNO ₃	Nitratine NaNO ₃	Inderborite CaMgB ₆ O ₁₁ H ₃ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
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хх х х													
хх x х													
хх х х													
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Depth (cm)	Bradleyite Na ₃ Mg(PO ₄) (CO ₃)	Burkeite Na ₄ (SO ₄) (CO ₁ ,SO ₄)	Hanksite KNa22(CO3)2 (SO4)2	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreckite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg2Cl(OH)3 2H3O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)1.6H2O
				· · · · · · · · · ·		· J/4		•••	······		• • • • • • • • • • • • • • • • • • • •		
452													
454											v		
450											X		
460			x										
462													
464													
466													
468													
470													
472			x										
474													
476													
478													
480													
482			v										
486			~										
488			х										
490													
492													
494			х									х	
496													
498													
500													
502													
504													
500			v										
510			^										
512													
514													
516													
518													
520													
522												х	

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa ₂₂ (CO ₃) ₂ (SO ₄) ₉ Cl	Kainite KMg(SO ₄)Cl 2.75H ₂ O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine [*] NaNO ₃	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
526 528												x	
532													
534			х									X	
538												~	
540												х	
542												x	
544													
546													
548													
550			х										
552													
224 556												x	
558												A	
560												х	
564			х										
566			х										
568													
570													
572													
576													
578													
580													
582													
584													
586													
588					x								
592					~								х
594													
596													
598													х
600													
Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca2(CO3)SO4 4H2O	Tychite Na6Mg2(SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ ,6H ₂ O
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602													
604			х										
606													
608													
616												v	
018												~	
620													
624													
626													
628													
630			х										
634			х										
636													
638													
640			X										
642			х										
644			.,										
646			х										
048 650												x	
652			x									X	
654			~										
656													
658													
660			х										
662			х										
664												,	х
666													
668													
670													
672													
674													
0/0													
680													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkcite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO₄)Cl 2.75H₂O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
682			х										
684													
686													
008 600													
690													
694													
696													
698													
700													
702			х										
704													
706													
708			x										
710													
714													
718													
720													
722													
724													
728												x	
730		Х											
732			x										
734													
730			v										
730			^								•		
746													
748													
750			х										
752			х										
754													
756													
758			х										
760													

.

$(CO_3) (CO_3, SO_4) (SO_4)_9Cl 2.75H_2O 4H_2O (CO_3)_4 6H_2O 2H_2O 2H_2O H_2O H_2O H_2O H_2O H_2$	(OH) ₂ .6H ₂ O
762 764	x
764	x
	x
792 X	x
794	
796 X ,	
798 800 X	
804	
806	
808	
810	
812 X	
814	
816	
BIB X	
820 X X	
822 824	
824 A A	
828 X	
830 X	
832 X	
834	
836	
838	
840 X	
842 X	
844	
846	
848 X	
X 008	
852 856	
862	

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)CI 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Scpiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
864			x										
866		Х	x									v	
868												X	
870													
872			v										
874			× ×										
670 979			x										
880			~										
882			х										
884												х	
886													
888			х										
890													
892													
894			х										
896			х										
898			х					х					
900			х										
902			х										
904			X										
906			X										
908			х										
910													
912			v										
914			~	•									
018	Y											•	
920	~												
924	х		x										
926	• -		х										
928													
930			х										
932													
934			х									х	
936													

•

4	:								•				
(cm)	Bradleyite Na,Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa ₂₂ (CO ₃) ₂ (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H ₂ O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₃ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H,0	Niter KNO ₃	Nitratine NaNO ₃	Inderborite CaMgB ₆ O ₁₁ H.O	Sepiolite Mg,Si ₆ O ₁₅
938		×						-		and a second of the second of			
940		:											×
942													
944													
946													
948	×												
950													
952													
954													
956			×										
958													
096													
962													
964													
972			×										
974													
976			×										
978			: ×										
980			: ×										
982			:										
984													
986			×										
988													
066			×										
992													
994			×										
966			×										
998													
1000													
1002													
1004													
1006													
1008													
0101					×							×	
1012					:								
1014													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainitc KMg(SO₄)Cl 2.75H₂O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO ₃	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
1016			x										
1018													
1020			x										
1024			x										
1026													
1028													
1030									х				
1032													
1034													
1036										•			
1038			x										
1040													
1042													
1044													
1048													
1050													
1052													
1054			х										
1056													
1058			х										
1060													
1062			х										
1004													
1000													
1008			x								•		
1072			л										
1074													
1076													
1078													
1080													
1082													
1084			X										
1086			х										

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO₄)Cl 2.75H₂O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na6Mg2(SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
1088			x										
1090													
1092													
1094											v		
1090											х		
1098												x	
1100												~	
1102													
1106													
1108													
1110													
1112			х		х								
1114													
1116													
1118			х									х	.,
1120													х
1122											v		
1124											~		
1120													
1120													
1132													
1134													
1136													
1138													
1140													
1142													
1144					х								
1148													
1150													
1152													
1154					x								
1158	x				~								
1160	~												

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO ₄)Cl 2.75H ₂ O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na6Mg2(SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
1162			v									x	
1166			^								x	~	
1168			x										
1170			х										
1172											х		
1174			х										
1176													
1178			v		v								
1182			~		^						x		
1184											~		
1186													
1188													
1190													
1192													
1194													
1198													
1200													
1202											х		
1206											Ŷ		
1208											~		
1210											х		
1212													
1214													
1216											X		
1218													
1220													
1222											v		
1224											~		
1320			x										
1330			~										
1332											х		
1334													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca2(CO3)SO4 4H2O	Tychite Na6Mg2(SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB6O11 H2O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
1336													
1338			X										
1340			X									x	
1342			X										
1346													
1348											Y		
1350											~	x	
1352												x	
1354												x	
1356													
1358												х	
1360												х	
1362												х	
1309													
1368			v										
1308			~									v	
1372												*	
1374													
1376													
1380	х												
1382													
1384													
1386			х										
1388													
1390											,		
1392													
1394			v		v								
1308			^		^								
1400								•					
1402			x										
1404			x										
1406													
1408													

.

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
1410			x										
1412			х										
1414													
1416													
1418													
1420											v		
1422											Х		
1424													
1420													
1420													
1430													
1434													
1436													
1438													
1440													
1442			х								х		
1444													
1446													
1448			х										
1450													
1452			х										
1454			х										
1456			Х										
1458													
1460											v		
1462											X,		
1464											v		
1466											^		
1408													
1470												x	
1472			v									~	
1474			~										
1478													
1480											х		

•

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO₄)Cl 2.75H₂O	Rapidcrcekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO ₃	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
1482													
1484													
1480											х		
1400											X		
1492			х										
1494													
1496													
1498													
1500													
1504			х										
1506													
1508													
1510													
1512													
1514			x										
1516			x										
1518			x										
1520													
1522											v		
1524											~		
1520			v										
1528			Х										
1530													
1532			x										
1536			Х										
1538			х										
1540											х		
1542											x		
1544											Х		
1546													
1548			х										
1550													
1552											х		
1554													

.

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
1556													
1558													
1560			х										
1562			х										
1564			х										
1566													
1568													
1570			x										
1572											Х		
1574													
1576			х								Х		
1578													
1582													
1586													
1588													
1500											x		
1592			x								~		
1594													
1596													
1598													
1600											х		
1602											х		
1604													
1606													
1608													
1610											Х		
1612													
1614											х		
1616	х		х								Х		
1618													
1620			х										
1622			v										
1624			X										
1020			A										
1628													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₁ ,SO ₄)	Hanksite KNa ₂₂ (CO ₃) ₂ (SO ₄) ₀ Cl	Kainite KMg(SO4)Cl 2,75H-O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₁ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₁)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₁ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
		((4/3			(5/4		2			arian aris a ana ana		(
1630			x										
1632													
1034													
1030													
1038											v		
1640											^		
1644			x								x		
1646			~								~		
1648			х										
1650													
1652											х		
1654													
1656													
1658													
1660													
1662													
1664													
1666			х										
1668			х										
1670													
1672													
1674													
1676													
1678													
1680													
1730			X										
1732			x										
1734											Х		
1736			N		v								
1738			Х		~						v		
1740			v								л		
1742			х				1						
1/44													
1740													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO₄)Cl 2.75H₂O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Scpiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
1750													
1752			V										
1754			х										
1758													
1760											х		
1762											х		
1764			х										
1766													
1768													
. 1770			х								v		
1//2											л		
1776											x		
1778											~		
1780													
1782													
1786			х										
1788			х										
1790													
1792													
1794			v										
1790			X								Y		
1800											x		
1802											x		
1804											x		
1806											X		
1808											Х		
1810			х										
1812											X		
1814											Х		
1010													
1820													
1822													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa ₂₂ (CO ₃) ₂ (SO ₄)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
1824											Y		
1820											~		
1830													
1832													
1834													
1836													
1912											х		
1914													
1018													
1970													
1022													
1922													
1926			х										
1928											х		
1930											х		
1932													
1934													
1936													
1938											v		
1940											x		
1942											~		
1946											х		
1948													
1950													
1952											Х		
1954											X		
1956											X		
1958											A Y		
1062											X		
1964											x		
1966													
1968													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksitc KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Scpiolite Mg4Si6O15 (OH)2.6H2O
1970 1974											x x		
1976											х		
1978											x		
1982											x		
1984											х		
1986											x		
1990											x		
1992 1994											х		
1996											х		
2000											х		
2002											х		
2006													
2008											х		
2012													
2014											x		
2016											x x		
2020											x		
2022											х		
2024													
2028			х										
2030											x		
2032											x		
2036													
2038 2040											х		
2042											х		
2044											х		

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Detailed Evaporite Mineralogy (North Core)

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Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa ₂₂ (CO ₃) ₂ (SO ₄) ₉ Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
2046													
2048											v		
2052											X		
2054											x		
2058											x		
2060													
2062													
2064													
2066											х		
2068											.,		
2070											Х		
2072													
2074													
2078													
2080													
2082											х		
2084											х		
2086													
2088											Х		
2090													
2092											x		
2094											x		
2098													
2100											X		
2102													
2104											х		
2108													
2110											Y		
2112											X		
2114											x		
2118											x		
2120											• -		

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na4(SO4) (CO3,SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg2Cl(OH)3 2H2O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
2122											x		
2124											X		
2120											x		
2120											x		
2132											x		
2134											х		
2136													
2138											х		
2140											v		
2142											x		
2146											x		
2148											х		
2150													
2152											Х		
2154											X		
2156											X		
2158											X		
2160											л		
2164													
2166											х		
2168											х		
2170											х		
2172													
2174													
2170													
2180													
2182											х		
2184											х		
2186													
2188													
2190											v		
2192											~		

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO₄)Cl 2.75H₂O	Rapidcreekite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Camaliite KMgCl3 6H2O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ ,6H ₂ O
2194													
2196											Х		
2198											x		
2200											л		
2204			x										
2206													
2208													
2210											х		
2212													
2214													
2216											v		
2210											x x		
2220											~		
2224													
2226													
2228											Х		
2230											Х		
2232											Х		
2234													
2236											.,		
2238											Х		
2240											Y		
2242											~		
2246													
2250											x		
2252													
2254											х		
2256													
2258											v		
2260											X V		
2202											x		
2204			x								~		
2200													

Depth (cm)	Bradleyite Na3Mg(PO4) (CO3)	Burkeite Na ₄ (SO ₄) (CO ₃ ,SO ₄)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4)Cl 2.75H2O	Rapidcreckite Ca ₂ (CO ₃)SO ₄ 4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Korshunovskite Mg ₂ Cl(OH) ₃ 2H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
2268													
2270											Х		
2272											Х		
2274											Х		
2276											Х		
2280													
2282													
2284											Х		
2286											X		
2288													
2290													
2292											Х		
2294													
2296													

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Detailed Evaporite Mineralogy (North Core)

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Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Artinite Mg ₂ CO ₃ (OH) ₂ 3H ₂ O	Fairchildite K2Ca (CO3)2	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Kutnohorite Ca(Mn,Mg) (CO ₃) ₂	Minrecordite CaZn(CO ₃) ₂	Zemkorite Na ₂ Ca (CO ₃) ₂	Arcanite (K,NH4)2SO4	Epsomite MgSO4 7H2O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO ₄ 4H ₂ O
10														
12						х								
14						х								
16						х								
18						x								
20						x								
22						X								
24						X								
20						× ×								
20	x					x								
32	x					x								
34	x					x								
36						x								
38						х								
40						х								
42						х								
44	х													
46						х								
48														
50														
52														
54														
56														
58														
60														
64														
66														
68														
70														
72														
74							х							
76														
78														
80														
82														
84														

Depth (cm) C	Ankerite Ja(Fe,Mg) (CO ₃)2 (Artinite Mg ₂ CO ₃ (OH) ₂ 3H ₂ O	Fairchildite K ₂ Ca (CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Kutnohorite Ca(Mn,Mg) (CO ₃)2	Minrecordite CaZn(CO ₃) ₂	Zemkorite Na ₂ Ca (CO ₃) ₂	Arcanite (K,NH4,)2SO4	Epsomite MgSO4 7H ₂ O	Eugsterite Na ₂ Ca(SO4), 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Mercallite KHSO ₄	Poly halite K ₂ Ca ₂ Mg(SO4), 2H ₂ O	Starkcyite MgSO4 4H2O
86							• • • • •				-			
88 00 88														
92 94			×											
96														
% 00														
102														
106														
108														
011														
711														
116														
118														
120														
122														
126														
128														
130														
132														
136														
138			x											
140														
142														
144														
148														
150			×											
152			×											
154														
156														
091														

	Polyhalite Starkey Ca ₂ Mg(SO4) ₄ MgSO 2H ₅ O 4H ₅ O																															
	Mercallite KHSO4 K ₂	n an ann an ann ann ann an ta																														
in core)	Gorgcyite K2Ca5(SO4)6 2H2O	•																														
noc) (Sor	Eugsterite Na2Ca(SO4), 2H2O																															
	Epsomite MgSO4 7H ₂ O																															
	Arcanite (K,NH4)2SO ₄	2 · · · · · · · · · · · · · · · · · · ·																														
	Zemkorite Na ₂ Ca (CO ₃) ₂																															
	Minrecordite CaZn(CO3)2																															
	Kutnohorite Ca(Mn,Mg) (CO ₃) ₂	•																										X	¢			
	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O																															
	Fairchildite K ₂ Ca (CO ₃) ₂															×	;	<														
	Artinite Mg ₂ CO ₃ (OH) ₃ 3H ₂ O																															
	Ankerite Ca(Fe,Mg) (CO ₃) ₂																															
	Depth (cm)	162 164	166	168	172	174	176	180	182	184	186	190	192	194	961	861	200	204	206	208	210	212	214	210	017	222	224	226	228	230	232	236

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Detailed Evaporite Mineralogy (South Core)

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							Detailec	l Evaporit	e Miner:	ilogy (Sou	th Core)			
Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Artinite Mg ₂ CO ₃ (OH) ₂ 3H ₂ O	Fairchildite K2Ca (CO3)2	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Kutnohorite Ca(Mn,Mg) (CO ₃) ₂	Minrecordite CaZn(CO ₃) ₂	Zemkorite Na ₂ Ca (CO ₃) ₂	Arcanite (K,NH4)2SO4	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na ₂ Ca(SO4) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO4 4H2O
238			×			•						a marina a a a a a	n	and more standards for the sec
240			×											
242 244														
246														
248														
252														
254			×											
256			×											
258														
007														
264														
266														
268														
270														
272														
2/4 276														
278														
280														
282														
284														
288														
290														
292														
294														
296														
298														
300														
302														
304														
200														
312														

Depth	Ankcrite	Artinite	Fairchildite	Gaylussite	Kutnohorite	Minrecordite	Zemkorite	Arcanite	Epsomite	Eugsterite	Gorgeyite	Mercallite	Polyhalite	Starkeyite
(m)	Ca(Fc,Mg)	Mg ₂ CO ₃	K ₂ Ca	Na ₂ Ca(CO ₃) ₂	Ca(Mn,Mg)	CaZn(CO ₃) ₂	Na ₂ Ca	(K,NH4)2SO4	MgSO4	Na ₂ Ca(SO ₄) ₃	K ₂ Ca ₅ (SO ₄) ₆	KHSO4	K ₂ Ca ₂ Mg(SO ₄) ₄	MgSO4
	(c0 ₁) ₂	O ₂ H ₅ (HO)	(CO ₁)2	5H ₂ O	(CO ₁)		(CO ₃) ₂		7H ₂ O	2H ₂ O	2H ₂ O		2H ₁ O	4H ₂ O
314														
316														
318														
320														
322														
324														
326														
328														

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Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Artinite Mg ₂ CO ₃ (OH) ₂ 3H ₂ O	Fairchildite K ₂ Ca (CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Kutnohorite Ca(Mn,Mg) (CO ₃) ₂	Minrecordite CaZn(CO ₃) ₂	Zemkorite Na ₂ Ca (CO ₃) ₂	Arcanite (K,NH4)2SO4	Epsomite MgSO4 7H2O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkcyite MgSO4 4H2O
390 392														
394														
396								х						
398														
400						х								
404														
406														
408														
410 412														
414														
416														
500														
502														
504														
510														
512								х						
514														
510								х						
520			х											
522														
524														
526								X X						
530								Х						
532														
534														
536								·						
538														
542														
544														
546														
548								х						

Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Artinite Mg ₂ CO ₃ (OH) ₂ 3H ₂ O	Fairchildite K2Ca (CO3)2	Gaylussitc Na2Ca(CO3)2 5H2O	Kutnohorite Ca(Mn,Mg) (CO ₃) ₂	Minrecordite CaZn(CO ₃) ₂	Zemkorite Na ₂ Ca (CO ₃) ₂	Arcanite (K,NH4)2SO4	Epsomite MgSO ₄ 7H ₂ O	Eugsterite Na ₂ Ca(SO ₄) ₃ 2H ₂ O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkeyite MgSO4 4H2O
550														
552														
554														
550														
560														
562														
564														
566														
568														х
570														
572														
574														
578														
580														
582														
584														
586														
588								х						
590														
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594														
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598 600			v											
602			~											
604														
606														
608														
610														
612														
614														
616												х		
618														
620														
624														

Depth (cm)	Ankerite Ca(Fe,Mg) (CO ₃) ₂	Artinite Mg ₂ CO ₃ (OH) ₂ 3H ₂ O	Fairchildite K ₂ Ca (CO ₃) ₂	Gaylussite Na ₂ Ca(CO ₃) ₂ 5H ₂ O	Kutnohorite Ca(Mn,Mg) (CO ₃) ₂	Minrecordite CaZn(CO ₃) ₂	Zemkorite Na ₂ Ca (CO ₃) ₂	Arcanite (K,NH4)2SO4	Epsomite MgSO4 7H2O	Eugsterite Na2Ca(SO4)3 2H2O	Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O	Mercallite KHSO4	Polyhalite K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O	Starkcyitc MgSO4 4H2O
626														
028 630														
632														
634														
636														
638														
640														
642														
044 646														
648														
650														
652														
654														
656			х											
658														
660														
664														
666														
668														
670														
672														
674														
676														
680								Y						
682								~						
684														
686														
824														
826														
828														
830														
834														
836		κ.												

ttarkeyite MgSO4 4H ₂ O	1					×																				
Polyhalite S K ₂ Ca ₂ Mg(SO ₄) ₄ 2H ₂ O																										
Mercallite KHSO4																										
Gorgeyite K ₂ Ca ₅ (SO ₄) ₆ 2H ₂ O																										
Eugsterite Na ₂ Ca(SO ₄), 2H ₂ O	•																									
Epsomite MgSO4 7H ₂ O	•																									
Arcanite (K,NH4)2SO4						x								×						>	< >	٢				
Zemkorite Na ₂ Ca (CO ₃) ₂																										
Minrecordite CaZn(CO ₃) ₂																										
Kutnohorite Ca(Mn,Mg) (CO ₃)2																										
Gaylussite Na2Ca(CO3)2 5H2O																										
Fairchildite K ₂ Ca (CO ₃) ₂																										
Artinite Mg ₂ CO ₃ (OH) ₂ 3H ₂ O																										
Ankcrite Ca(Fe,Mg) (CO ₃) ₂																										
Depth (cm)	838 840	842	846	848 850	852	854	856	860	862	864	866	868	870	872	8/4	876	8/8	088	700	886	888	890	892	894	896	

.

Depth (cm)	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O	Burkeite N84(SO4) (CO3.SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4) Cl.2.75H2O	Rapidcreekite Ca ₂ (CO ₃) SO ₄ .4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
10											v	
12											X	
14												
18												
20												
22											х	
24												
26												
28												
30												
32												
34												
36												
38												
40												
42												
44 76												
40												
50												
52											х	
54											х	
56												
58												
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64												
66											•	
68												
70											Х	
72												
74												
76												
/8 00											x	Y
00 80											~	^
04 94												

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86 x 99 x 94 x 96 x 98 x 100 x 101 x 102 x 103 x 104 x 105 x 106 x 107 x 108 x 109 x 112 x 114 x 115 x 116 x 118 x 119 x 121 x 132 X 133 x 134 x 135 x 136 x 137 X 138 x 139 x 134 x 135 x 136 x 137 x 138 x 139 x 131 x	Depth (cm)	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O	Burkeite Na4(SO4) (CO3.SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO₄) Cl.2.75H2O	Rapidcreekite Ca2(CO3) SO4.4H2O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ (
88 X 92 X 94 X 96 X 98 X 102 X 104 X 105 X 106 X 107 X 108 X 109 X 110 X 112 X 114 X 125 X 136 X 137 X 138 X 144 X 144 X 144 X 144 X 144 X 145 X 146 X 147 X 148 X 149 X 141 X 142 X 143 X 144 X 145 X 146 X 147 X 148 X	86												
90 X X 94 X X 96 X X 98 X X 100 X X 104 X X 105 X X 106 X X 107 X X 108 X X 109 X X 110 X X 111 X X 112 X X 114 X X 115 X X 116 X X 117 X X 118 X X 119 X X 120 X X 121 X X 122 X X 133 X X 134 X X 135 X X 144 X X 145 X X 150	88												
92 A A 94 96 98	90 90											v	v
96 98 100 X 101 X 102 X 103 X 104 X 105 X 106 X 107 X 118 X 120 X 121 X 122 X 123 X 124 X 125 X 126 X 130 X 131 X 132 X 134 X 135 X 136 X 137 X 138 X 139 X 134 X 135 X 136 X 137 X 138 X 139 X 130 X 131 X 132 X 133 X 134 X	92											~	~
98 X 100 X 104 X 106 X 108 X 109 X 101 X 102 X 103 X 114 X 120 X 121 X 122 X 123 X 124 X 125 X 126 X 131 X 132 X 134 X 135 X 146 X 147 X 148 X 149 X 140 X 141 X 142 X 143 X 144 X 145 X 146 X 147 X 148 X 149 X 141 X 142 X	94												
100 X 102 X 106 107 108 112 114 116 118 120 X 121 122 X 123 X 124 X 125 X 126 X 127 X 138 139 X 130 X 131 132 X 133 134 135 140 X 141 X 142 X 143 X 144 X 145 X 146 X 147 X 148 X 149 X 140 X 141	90												
102 X 104 X 106	100												
104 106 108 110 112 114 116 118 120 X 121 X 122 X 123 X 124 X 125 X 126 X 130 X 131 X 132 X 133 X 144 X 146 X 148 X 149 X 144 X 146 X 147 X 148 X 149 X 140 X 1414 X 142 X 143 X 144 X 145 X 150 X 151 X	102											х	
106 108 110 112 114 116 120 121 122 X 123 X 124 X 125 X 126 X 130 X 132 X 133 144 146 X 148 X 135 136 137 138 144 145 X 146 X 137 138 144 145 X 146 X 150 152	104												
108 110 112 114 116 121 122 123 124 125 126 128 130 131 132 X 134 135 136 137 138 140 144 145 146 147 148 149 140 141 142 143 144 145 146 147 148 149 140 141 142 143 144 145 146 147 148 149 140 141 142 144 145 146	106												
110 112 114 116 118 120 X 121 X 122 X 124 X 125 X 126 X 130 X 132 X 134 X 136 X 138 X 140 X 144 X 146 X 148 X 150 X 152 X	108												
112 114 116 118 120 X 121 X 122 X 124 X 126 X 128 X 130 X 132 X 134 X 136 X 138 X 140 X 142 X 144 X 145 X 146 X 150 X 152 X	110												
114 116 118 120 X 121 X 122 X 124 X 126 X 130 X 132 X 134 X 136 X 138 X 140 X 142 X 144 X 144 X 150 X 152 X	112												
116 X 120 X 121 X 122 X 124 X 126 X 128 X 130 X 132 X 134 X 136 X 138 X 140 X 142 X 144 X 150 X 152 X	114												
118 X 120 X 122 X 124 X 126 X 128 X 130 X 132 X 134 X 135 X 140 X 142 X 144 X 146 X 150 X 152 X	116												
120 X 122 X 124 X 126 X 128 X 130 X 132 X 134 X 135 X 138 X 140 X 142 X 144 X 145 X 150 X 152 X	118												
122 X 124 X 126 X 128 X 130 X 132 X 134 X 136 X 138 X 140 X 142 X 144 X 150 X 152 X	120										X		
124 X 126 X 128 X 130 X 132 X 134 X 136 X 138 X 140 X 142 X 144 X 150 X 152 X	122										X		
128 X 130 X 132 X 134	124										N V		
120 X 130 X 132 X 134 . 136 . 138 . 140 . 142 X 144 . 145 X 146 X 150 . 152 .	120										x		
130 X 132 X 134 136 138 140 142 144 144 145 146 148 150 152 154	120										x		
134 136 138 140 142 144 146 148 150 152	130			x							A		
136 138 140 142 144 146 148 150 152	134			~									
138 140 142 144 146 148 150 152	136												
140 X . 142 X . 144 X . 146 X . 148 X . 150 . . 152 . .	138												
142 X . 144 146 X 148 150 152	140												
144 146 148 150 152	142										Х		
146 X 148 X 150 152	144												
148 X 150 152	146										х		
150 152	148										х		
152 164	150												
IN T	152										v		
	154										Х		
	150												
	128												

Sepiolite Mg,Si ₆ O ₁₅ (OH) ₂ ,6H ₂ O	• •.																				
Inderborite CaMgB ₆ O ₁₁ H ₂ O					×																
Nitratine NaNO,	×	××	×	×××	< ×	:	××	< ×			××	×					×	×	×	×	< ×
Niter KNO ₃																					
Camallite KMgCl ₃ 6H ₂ O																					
Bischofite MgCl ₃ .6H ₂ O																					
Tychite Na ₆ Mg ₂ (SO4) (CO3)4																					
Rapidcreckite Ca ₂ (CO ₃) SO ₄ ,4H ₂ O																					
Kainite KMg(SO4) CI.2.75H ₂ O																					
Hanksite KNa ₂₂ (CO ₃) ₂ (SO4,) ₉ CI						×															
Burkeite Na ₄ (SO ₄) (CO ₃ .SO ₄)																					
Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O																					
Depth (cm)	162 164 166	170	176	180	184 186	188	192 194	196	200	202	206	208 210	212	214	218	220	222	226	228	232	234 236

Depth (cm)	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O	Burkeite Na4(SO4) (CO3.SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4) Cl.2.75H2O	Rapidcreekite Ca ₂ (CO ₃) SO ₄ .4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
238												
240												
242												
244												
246												
248										х	х	
250												
252												
254										.,		
256										X		
258										л	v	
200											^	
202												
204			x							x		
200			Х							~		
270												
272										х		
274												
276												
278			х								х	
280												
282			х									
284										х	х	
286			Х									
288												
290			Х									
292			х									
294											•	
296										х	x	
298											x	
300												
302												
304												
306			х								х	
308											v	
310											^	
312												

Depth (cm)	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O	Burkeite Na ₄ (SO ₄) (CO ₃ .SO ₄)	Hanksite KNa ₂₂ (CO ₃) ₂ (SO ₄) ₉ Cl	Kainite KMg(SO ₄) CI.2.75H ₂ O	Rapidcreckite Ca ₂ (CO ₃) SO ₄ .4H ₂ O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Niter KNO3	Nitratine NaNO ₃	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
314												
316												
318			>								×	
322			<									
324			:×								X	
326											<	
328												
330 332								:		:		
334			×					×		×		
336												
338			×									
340			×								*	
342			×								<	
344												
346			×					Х				
348			×					:				
350										×		
352										:		
354			×									
356										×		
358										< ×		
360										×		
362												
400												
00C												
370												
372											-	
374											×	
376										>	×	
378										<		
380												
382												
384												
386												
388										×		

Carnallite N KMgCl ₃ K 6H ₂ O	Camallite Niter KMgCl ₃ KNO ₃ 6H ₂ O	Camallite Niter KMgCl ₃ KNO ₃ 6H ₂ O	Camallite Niter Niter Niter Niter Niter Niter Niter No. 6H2O	Camallite Niter N KMgCl, KNO, N 6H2O	Camallite Niter Ni	KMgCl, KNO, Na 6H ₂ O X	KMgCl, KNO, Na 6H ₂ O Na × X	KMgCl, KNO, Na 6H ₂ O Na × ×	KMgCl, KNO, Na 6H ₂ O X	Camallite Niter Nitr KMgCl, KNO, Nai 6H ₂ O X X X X X X X X X X X X X X X X X X X	KMgCl, KNO, Na 6H ₂ O X	KMgCl, KNO, Na 6H ₃ O Na X X	X KMgCl, KNO, N 6H ₂ O X	KMgCl ₃ KNO ₃ N 6H ₂ O X X
		-	-	-	×	×	×	×	×	×	×	×	×	×
×	×	×	×	×	×	×	×	ž	~	~	~	~	~	
×	×	×	×	×	×	×	×	~				~	~	
					-	-			•	•	-			
					×	×	×	×	×	×	×	×	×	×

Depth (cm)	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O	Burkcite Na4(SO4) (CO3.SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO4) Cl.2.75H2O	Rapidcreekite Ca2(CO3) SO4.4H2O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Carnallite KMgCl ₃ 6H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg ₄ Si ₆ O ₁₅ (OH) ₂ .6H ₂ O
550										x		
552												
554			х								x	
556												
558												
560			x									
502										v		
504										х		
569												
570												
572					x						x	
574			x		~						~	
576			x									
578			~									
580			х					х		х		
582			x								х	
584												
586												
588												
590					х							
592												
594										х		
596												
598										х		
600												
602												
604					х							
606											•	
608			x									
610			.,									
612			X									
614												
616												
618												
620												
022 674												
024												
Depth (cm)	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O	Burkeite Na4(SO4) (CO3.SO4)	Hanksite KNa22(CO3)2 (SO4)9Cl	Kainite KMg(SO₄) Cl.2.75H2O	Rapidcreekite Ca2(CO3) SO4.4H2O	Tychite Na ₆ Mg ₂ (SO ₄) (CO ₃) ₄	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Niter KNO3	Nitratine NaNO3	Inderborite CaMgB ₆ O ₁₁ H ₂ O	Sepiolite Mg4Si6O15 (OH)2.6H2O
---------------	---	-----------------------------------	-------------------------------------	-----------------------------------	---------------------------------------	--	--	--	---------------	--------------------	---	--------------------------------------
626								x			x	
628			х									
630											x	
632												
034												
620			v									
640			~									
642												
644											x	
646										x	~	
648												
650			х									
652												
654												
656												
658										х		
660												
662												
664											х	
666												
668												
670			х								х	
672										••		
674										х		
676												
678												
680			X									
682			X								·	
084			~									
080												
824												
820 878												
040 930												
040 812												
834												
836												
0.0												

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Detailed Evaporite Mineralogy (South Core)

fi fi	Wattevilleite Na ₂ Ca(SO ₄) ₂ 4H ₂ O	Burkeite Na ₄ (SO ₄) (CO ₃ .SO ₄)	Hanksite KNa ₂₂ (CO ₃) ₂ (SO4) ₉ CI	Kainite KMg(SO4) CI.2.75H ₂ O	Rapidereekite Ca ₂ (CO ₃) SO4.4H ₂ O	Tychite Na ₆ Mg ₂ (SO4) (CO3)4	Bischofite MgCl ₂ .6H ₂ O	Camallite KMgCl ₃ 6H ₂ O	Niter KNO3	Nitratinc NaNO ₃	Inderborite CaMgB ₆ O ₁₁ H.O	Sepiolite Mg4Si4O15 (OH), 6H4O
~								8) 	
											x	
											:	
- 10												
~												
_										×		
~												
_												
			×								×	
~			•									
~												
~			×									
_											×	
_												
			×									
_												
											;	
_											×	
_												

Detailed Evaporite Mineralogy (South Core)

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Appendix D Major Ion Ratios of Different Stratigraphic Units

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Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
					•
12	1.50	1.50	1.00	3.00	3.00
16	0.00	0.00	1.00	0.28	0.28
18	0.34	0.66	0.68	0.44	0.65
20	2.57	2.57	1.00	5.43	5.43
22	4.12	2.41	2.72	4.81	1.77
24	1.92	1.92	1.00	3.85	3.85
26	0.13	0.42	0.72	0.04	0.06
28	0.38	0.81	0.57	0.77	1.34
30	3.22	3.22	1.00	6.44	6.44
32	0.73	0.00	1.73	0.00	0.00
36	2.82	1.68	2.14	3.36	1.57
38	0.71	0.71	1.00	1.42	1.42
40	0.37	0.72	0.65	0.73	1.13
42	0.48	0.54	0.94	0.22	0.24
44	0.95	0.95	1.00	1.90	1.90
46	0.19	0.19	1.00	0.41	0.41
48	0.54	0.54	1.00	1.08	1.08
50	0.08	0.08	1.00	0.15	0.15
52	0.11	0.11	1.00	0.22	0.22
54	0.03	0.03	1.00	0.05	0.05
56	0.03	0.03	1.00	0.07	0.07
58	0.04	0.04	1.00	0.08	0.08
60	0.04	0.04	1.00	0.08	0.08
62	0.05	0.05	1.00	0.10	0.10
64	0.11	0.00	1.11	0.00	0.00
66	0.04	0.04	1.00	0.09	0.09
68	0.05	0.05	1.00	0.10	0.10
70	0.03	0.03	1.00	0.06	0.06
72	0.03	0.03	1.00	0.06	0.06
74	0.03	0.03	1.00	0.06	0.06
76	0.03	0.03	1.00	0.07	0.07
78	0.03	0.03	1.00	0.06	0.06
80	0.03	0.03	1.00	0.08	0.08
84	0.03	0.03	1.00	0.05	0.05
84	0.03	0.03	1.00	0.05	0.05
80	0.02	0.02	1.00	0.04	0.04
88	0.00	0.00	1.00	00.0	0.00
90	0.02	0.02	1.00	0.04	0.04
92	0.00	0.00	1.00	0.00	0.00
94	0.00	0.00	1.00	0.00	0.00
90	0.03	0.03	1.00	0.05	0.05
70 100	0.02	0.02	1.00	0.04	0.04
106	0.02	0.02	1.00	0.04	0.04
100	0.00	0.00	1.00	0.00	0.00
108	0.00	0.00	1.00	0.00	0.00
110	0.00	0.00	1.00	0.00	0.00
114	0.00	0.00	0.02	0.00	0.00
114	0.00	0.03	1.00	0.00	0.00
118	0.00	0.01	1.00	0.05	0.03
120	0.00	0.00	1.00	0.00	0.00
120	0.00	0.00	1.00	0.00	0.00
126	0.00	0.00	1.00	0.00	0.00
178	0.00	0.00	1.00	0.00	0.00
120	0.01	0.00	1.00	0.00	0.00
132	0.11	0.00	1.11	0.00	0.02

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
134	0.02	0.02	1.00	0.04	. 0.04
136	0.10	0.00	1.10	0.00	0.00
138	0.02	0.02	1.00	0.04	0.04
140	0.03	0.03	1.00	0.06	0.06
142	0.03	0.03	1.00	0.05	0.05
144	0.00	0.00	1.00	0.00	0.00
146	0.09	0.09	1.00	0.18	0.18
148	0.09	0.09	1.00	0.17	0.17
150	0.12	0.12	1.00	0.25	0.25
152	0.11	0.11	1.00	0.24	0.24
154	0.02	0.02	1.00	0.04	0.04
156	0.00	0.00	1.00	0.00	0.00
158	0.00	0.00	1.00	0.00	0.00
160	0.00	0.00	1.00	0.00	0.00
162	0.00	0.00	1.00	0.00	0.00
164	0.00	0.00	1.00	0.00	0.00
166	0.07	0.00	1.07	0.00	0.00
168	0.00	0.00	1.00	0.00	0.00
170	0.01	0.07	0.94	0.02	0.02
172	0.00	0.00	1.00	0.00	0.00
174	0.00	0.00	1.00	0.00	0.00
176	0.00	0.00	1.00	0.02	0.02
178	0.07	0.01	1.06	0.03	0.02
180	0.01	0.01	1.00	0.03	0.03
182	0.01	0.09	0.92	0.02	0.02
184	0.07	0.00	1.07	0.00	0.00
188	0.08	0.02	1.07	0.04	0.03
190	0.21	0.00	1.21	0.00	0.00
192	0.11	0.00	1.09	0.00	0.00
194	0.01	0.01	1.00	0.05	0.07
196	0.11	0.01	1.00	0.02	0.02
198	0.14	0.02	1.12	0.05	0.04
200	0.13	0.02	1.12	0.00	0.00
202	0.13	0.02	1.12	0.04	0.03
204	0.13	0.02	1.12	0.04	0.03
206	0.27	0.02	1.25	0.04	0.03
208	0.22	0.04	1.18	0.08	0.07
210	0.16	0.00	1.16	0.00	0.04
210	0.33	0.00	1 31	0.04	0.03
214	0.21	0.03	1.18	0.04	0.05
216	0.10	0.02	1.08	0.04	0.04
218	0.14	0.03	1.11	0.04	0.05
220	016	0.02	1 14	0.00	0.03
220	0.27	0.12	1.14	0.15	0.03
222	0.25	0.13	1 21	0.18	015
226	0.03	0.03	1.00	0.06	0.06
228	0.42	0.03	1 39	0.06	0.04
230	0.14	0.01	1.00	0.14	0.14
232	0.13	0.02	1.11	0.04	0.03
234	0.18	0.03	1.14	0.07	0.05
236	013	0.00	1 13	0.00	0.00
230	0.33	0.00	1 33	0.00	0.00
230	0.22	0.00	1.20	0.00	0.00
240	0.19	0.02	1 13	0.12	0.04
242	0.17	0.02	1 17	0.03	0.11
244	0.22	0.00	1.17	0.05	0.05
240	0.13	0.00	1 00	0.13	0.00
270	9.1.0	0.04	1.07	0.10	0.12

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
250	0.35	0.03	1.32	0.10	. 0.07
252	0.35	0.03	1.32	0.10	0.07
254	0.02	0.02	1.00	0.10	0.10
256	0.36	0.04	1.32	0.11	0.08
258	0.17	0.03	1.14	0.09	0.08
260	0.11	0.04	1.08	0.11	0.10
262	0.32	0.04	1.28	0.12	0.09
264	0.28	0.06	1.22	0.15	0.12
266	0.22	0.03	1.19	0.06	0.05
268	0.17	0.03	1.14	0.06	0.05
270	0.13	0.04	1.09	0.09	0.08
272	0.47	0.04	1.43	0.12	0.09
274	0.28	0.04	1.24	0.12	0.10
276	0.15	0.04	1.12	0.07	0.06
278	0.00	0.00	1.00	0.04	0.04
280	0.31	0.03	1.28	0.10	0.08
282	0.04	0.04	1.00	0.15	0.15
284	0.15	0.02	1.13	0.06	0.05
286	0.08	0.00	1.08	0.04	0.03
290	0.07	0.00	1. 07	0.00	0.00
292	0.01	0.01	1.00	0.05	0.05
294	0.00	0.00	1.00	0.00	0.00
296	0.05	0.05	1.00	0.14	0.14
298	0.03	0.00	1.03	0.02	0.02
300	0.02	0.02	1.00	0.07	0.07
302	0.23	0.02	1.22	0.07	0.06
304	0.22	0.02	1.20	0.03	0.03
306	0.15	0.00	1.15	0.02	0.01
308	0.07	0.00	1.07	0.00	0.00
310	0.08	0.00	1.08	0.02	0.02
312	0.02	0.02	1.00	0.03	0.03
314	0.00	0.00	1.00	0.00	0.00
316	0.00	0.00	1.00	0.00	0.00
318	0.00	0.00	1.00	0.00	0.00
320	0.00	0.00	1.00	0.00	0.00
322	0.01	0.01	1.00	0.03	0.03
324	0.00	0.00	1.00	0.00	0.00
320	0.00	0.00	1.00	0.00	0.00
326	0.01	0.01	1.00	0.03	0.03
220	0.00	0.00	1.00	0.00	0.00
332	0.01	0.01	1.00	0.02	0.02
334	0.28	0.01	1.27	0.02	0.01
338	0.01	0.01	1.00	0.02	0.02
340	0.01	0.01	1.00	0.02	0.02
347	0.01	0.01	1.00	0.03	0.03
344	0.00	0.00	1.00	0.00	0.00
346	0.00	0.00	1.00	0.00	0.00
348	0.00	0.00	1.00	0.02	0.02
350	0.00	0.00	1.00	0.00	0.00
352	0.00	0.00	1.00	0.00	0.00
354	0.00	0.00	1.00	0.00	0.00
356	0.02	0.02	1.00	0.00	0.00
358	0.00	0.00	1 00	0.00	0.05
360	0.00	0.00	1.00	0.00	0.00
362	0.01	0.01	1.00	0.03	0.00
364	0.01	0.01	1.00	0.02	0.02

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
368	0.00	0.00	1.00	0.00	. 0.00
370	0.01	0.01	1.00	0.02	0.02
372	0.18	0.01	1.15	0.04	0.04
374	0.07	0.00	1.07	0.00	0.00
376	0.00	0.00	1.00	0.00	0.00
378	0.00	0.00	1.00	0.00	0.00
380	0.01	0.01	1.00	0.02	0.02
382	0.00	0.00	1.00	0.00	0.00
384	0.01	0.01	1.00	0.03	0.03
386	0.00	0.00	1.00	0.00	0.00
388	0.00	0.00	1.00	0.00	0.00
390	0.00	0.00	1.00	0.00	0.00
392	0.00	0.00	1.00	0.00	0.00
394	0.00	0.00	1.00	0.00	0.00
396	0.05	0.00	1.05	0.00	0.00
398	0.00	0.00	1.00	0.00	0.00
400	0.00	0.00	1.00	0.00	0.00
402	0.01	0.01	1.00	0.02	0.02
404	0.00	0.00	1.00	0.00	0.00
406	0.00	0.00	1.00	0.00	0.00
408	0.00	0.00	1.00	0.00	0.00
410	0.00	0.00	1.00	0.00	0.00
412	0.00	0.00	1.00	0.00	0.00
414	0.00	0.00	1.00	0.00	0.00
416	0.00	0.00	1.00	0.00	0.00
418	0.00	0.00	1.00	0.00	0.00
420	0.00	0.00	1.00	0.00	0.00
422	0.04	0.00	1.04	0.00	0.00
424	0.00	0.00	1.00	0.00	0.00
426	0.00	0.00	1.00	0.00	0.00
428	0.00	0.00	1.00	0.00	0.00
430	0.00	0.00	- 1.00	0.00	0.00
432	0.00	0.00	1.00	0.00	0.00
434	0.00	0.00	1.00	0.00	0.00
436	0.00	0.00	1.00	0.00	0.00
438	0.00	0.00	1.00	0.00	0.00
440	0.00	0.00	1.00	0.00	0.00
442	0.00	0.00	1.00	0.00	0.00
444	0.00	0.00	1.00	0.00	0.00
446	0.01	0.01	1.00	0.02	0.02
448	0.00	0.00	1.00	0.00	0.00
450	0.00	0.00	1.00	0.00	0.00
452	0.00	0.00	1.00	0.00	0.00
454	0.07	0.00	1.07	0.00	0.00
456	0.00	0.00	1.00	0.00	0.00
458	0.07	0.00	1.07	0.00	0.00
460	0.02	0.02	1.00	0.05	0.05
462	0.10	0.00	1.10	0.03	0.03
464	0.08	0.01	1.07	0.02	0.02
466	0.12	0.00	1.12	0.00	0.00
468	0.00	0.00	1.00	0.00	0.00
470	0.00	0.00	1.00	0.00	0.00
472	0.10	0.02	1.08	0.04	0.03
474	0.02	0.02	1.00	0.03	0.03
476	0.62	0.02	1.61	0.04	0.02
478	0.10	0.10	1.08	0.13	0.12
480	0.00	0.05	1.00	0.05	0.05

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO₃/SO₄
482	0.11	0.00	1.11	0.00	. 0.00
484	0.14	0.03	1.11	0.05	0.05
486	0.02	0.02	1.00	0.04	0.04
488	0.46	0.02	1.44	0.04	0.03
490	0.08	0.00	1.08	0.00	0.00
492	0.00	0.00	1.00	0.00	0.00
494	0.09	0.01	1.07	0.02	0.02
496	0.05	0.00	1.05	0.00	0.00
498	0.21	0.00	1.21	0.00	0.00
500	0.00	0.00	1.00	0.03	0.03
502	0.40	0.00	1.40	0.00	0.00
504	0.00	0.00	1.00	0.00	0.00
506	0.02	0.02	1.00	0.03	0.03
508	0.27	0.02	1.26	0.03	0.03
510	0.15	0.02	1.13	0.04	0.04
512	0.18	0.04	1.14	0.07	0.06
514	0.21	0.04	1.17	0.08	0.07
516	0.18	0.04	1.14	0.07	0.06
518	0.21	0.04	1.17	0.08	0.06
520	0.20	0.03	1.17	0.07	0.06
522	0.19	0.02	1.17	0.05	0.04
526	0.08	0.00	1.08	0.00	0.00
528	0.00	0.00	1.00	0.00	0.00
532	0.00	0.00	1.00	0.00	0.00
534	0.00	0.00	1.00	0.00	0.00
536	0.03	0.03	I.00	0.05	0.05
538	0.02	0.00	1.00	0.02	0.02
540	0.00	0.00	1.00	0.00	0.00
542	0.02	0.02	1.00	0.04	0.04
544	0.05	0.02	1.00	0.07	0.07
546	0.06	0.00	1.06	0.00	0.00
548	0.06	0.00	1.06	0.00	0.00
550	0.07	0.03	1.00	0.10	0.10
554	0.09	0.02	1.08	0.03	0.03
554	0.07	0.08	1.07	0.08	0.07
220	0.03	0.03	1.00	0.07	0.07
220	0.00	0.00	1.00	0.00	0.00
564	0.04	0.04	1.00	0.07	0.07
566	0.03	0.13	1.00	0.15	0.13
568	0.01	0.01	1.00	0.02	0.02
570	0.20	0.02	1.24	0.04	0.03
570	0.00	0.00	1.08	0.00	0.00
574	0.16	0.02	1 13	0.05	0.04
576	0.05	0.00	1.05	0.00	0.00
578	0.05	0.00	1.05	0.00	0.00
580	0.01	0.01	1.00	0.02	0.02
582	0.02	0.07	1.00	0.02	0.02
584	0.02	0.02	1.00	0.04	0.04
586	0.00	0.00	1.00	0.00	0.00
588	0.02	0.02	1.00	0.04	0.04
590	0.00	0.00	1.00	0.00	0.00
592	0.00	0.00	1.00	0.00	0.00
594	0.16	0.00	1.16	0.00	0.00
596	0.00	0.00	1.00	0.00	0.00
598	0.22	0.28	1.15	0.35	0.30
600	0.16	0.12	1.13	0.14	0.13

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
602	0.16	0.04	1.12	0.07	. 0.07
604	0.07	0.35	1.00	0.42	0.42
606	0.14	0.14	1.13	0.16	0.14
608	0.09	0.15	1.07	0.18	0.17
616	0.08	0.00	1.08	0.00	0.00
618	0.00	0.00	1.00	0.00	0.00
620	0.01	0.01	1.00	0.02	0.02
622	0.11	0.00	1.11	0.00	0.00
624	0.10	0.02	1.08	0.03	0.03
626	0.00	0.00	1.00	0.00	0.00
628	0.12	0.03	1.09	0.06	0.05
630	0.15	0.03	1.12	0.06	0.05
634	0.23	0.06	1.17	0.12	0.11
636	0.03	0.03	1.00	0.06	0.06
638	0.01	0.01	1.00	0.03	0.03
640	0.00	0.00	1.00	0.00	0.00
642	0.02	0.02	1.00	0.04	0.04
644	0.00	0.00	1.00	0.00	0.00
646	0.29	0.03	1.26	0.05	0.04
648	0.13	0.02	1.11	0.05	0.04
650	0.00	0.00	1.00	0.00	0.00
652	0.13	0.02	1.11	0.05	0.04
654	0.16	0.02	1.14	0.03	0.03
656	0.00	0.00	1.00	0.00	0.00
658	0.01	0.01	1.00	0.03	0.03
660	0.02	0.02	1.00	0.04	0.04
662	0.34	0.02	1.07	0.28	0.26
664	0.00	0.00	1.00	0.00	0.00
666	0.09	0.00	1.09	0.00	0.00
668	0.09	0.00	1.09	0.00	0.00
670	0.08	0.02	1.06	0.03	0.03
672	0.01	0.01	1.00	0.02	0.02
674	0.04	0.00	1.00	0.04	0.04
676	0.09	0.03	1.06	0.06	0.05
678	0.09	0.01	1.08	0.03	0.02
680	0.14	0.02	1.12	0.03	0.03
682	0.01	0.01	1.00	0.02	0.02
684	0.00	0.00	1.00	0.00	0.00
686	0.03	0.00	1.00	0.03	0.03
688	2.28	0.00	3.28	0.00	0.00
690	0.00	0.00	1.00	0.00	0.00
692	0.07	0.00	1.07	0.00	0.00
694	0.12	0.01	1.11	0.03	0.03
696	0.03	0.00	1.00	0.03	0.03
698	0.01	0.01	1.00	0.02	0.02
700	0.00	0.00	1.00	0.00	0.00
704	0.02	0.02	1.00	0.04	0.04
706	0.00	0.00	1.00	0.00	0.00
708	0.15	0.01	1.12	0.04	0.04
710	0.00	0.00	1.00	0.00	0.00
714	0.14	0.00	1.14	0.00	0.00
716	0.00	0.00	1.00	0.00	0.00
718	0.18	0.01	1.17	0.02	0.02
720	0.00	0.00	1.00	0.00	0.00
722	0.05	0.00	1.05	0.00	0.00
724	0.02	0.02	1.00	0.03	0.03
728	0.00	0.00	1.00	0.00	0.00

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
730	0.18	0.03	1.16	0.05	. 0.04
732	0.13	0.03	1.10	0.06	0.05
734	0.01	0.01	1.00	0.03	0.03
736	0.07	0.00	1.07	0.00	0.00
738	0.26	0.11	1.23	0.14	0.11
740	0.00	0.00	1.00	0.00	0.00
746	0.15	0.02	1.11	0.06	0.06
748	0.01	0.01	1.00	0.02	0.02
750	0.10	0.02	1.08	0.04	0.04
752	0.12	0.09	1.10	0.12	0.11
754	0.22	0.03	1.19	0.05	0.05
756	0.02	0.10	1.00	0.12	0.12
758	0.08	0.02	1.06	0.03	0.03
760	0.02	0.13	1.00	0.15	0.15
762	0.14	0.01	1.11	0.05	0.04
764	0.08	0.00	1.06	0.02	0.02
792	0.07	0.02	1.06	0.03	0.03
794	0.00	0.00	1.00	0.00	0.00
/96	0.09	0.02	1.07	0.04	0.04
798	0.00	0.04	1.00	0.04	0.04
800	0.15	0.02	1.10	0.07	0.06
802	0.09	0.01	1.08	0.02	0.02
804	0.08	0.00	1.08	0.00	0.00
806	0.12	0.00	1.12	0.00	0.00
808	0.14	0.11	1.08	0.17	0.16
810	0.10	0.02	1.09	0.03	0.03
812	0.02	0.02	1.00	0.04	0.04
814 91 <i>6</i>	0.13	0.00	1.13	0.00	0.00
810 919	0.03	0.00	1.00	0.03	0.03
820	0.10	0.09	1.07	0.11	0.10
820	0.13	0.02	1.11	0.04	0.03
874	0.45	0.19	1.40	0.23	0.10
826	0.50	0.02	1.29	0.09	0.07
878	0.10	0.09	1.14	0.11	0.10
830	0.15	0.18	1.15	0.21	0.18
832	0.18	0.13	1.17	0.22	0.18
834	0.19	0.05	1.12	0.09	0.08
836	0.15	0.12	171	0.20	0.18
838	0.33	0.26	118	0.40	0.20
840	0.14	0.03	1.11	0.05	0.05
842	0.17	0.03	1.14	0.06	0.05
844	0.24	0.15	1.20	0.19	0.05
846	0.31	0.14	1.29	0.16	0.13
848	0.16	0.03	1.13	0.06	0.06
850	0.31	0.03	1.28	0.05	0.04
852	0.15	0.03	1.13	0.05	0.05
856	0.28	0.03	1.25	0.05	0.04
860	0.27	0.02	1.24	0.05	0.04
862	0.22	0.02	1.20	0.04	0.03
864	0.51	0.13	1.48	0.16	0.11
866	0.75	0.03	1.72	0.06	0.03
868	0.32	0.16	1.29	0.19	0.15
870	0.25	0.15	1.09	0.03	0.03
872	0.56	0.03	1.54	0.06	0.04
874	0.19	0.01	1.18	0.03	0.02
876	0.37	0.03	1.34	0.06	0.04

378 0.27 0.02 1.25 0.05 $.$ 0.04 880 0.42 0.02 1.40 0.04 0.03 884 0.25 0.02 1.23 0.05 0.04 886 0.18 0.02 1.16 0.04 0.03 888 0.32 0.03 1.19 0.05 0.04 892 0.28 0.01 1.26 0.03 0.02 894 0.10 0.02 1.09 0.04 0.03 896 0.22 0.06 0.05 0.01 896 0.22 0.06 0.03 0.02 900 0.14 0.08 1.12 0.04 0.03 902 0.28 0.02 1.29 0.03 0.02 906 0.31 0.22 1.23 0.03 0.03 906 0.31 0.22 0.13 0.22 <th>Depth (cm)</th> <th>Ca/Na</th> <th>Mg/Na</th> <th>SO₄/Na</th> <th>CO₃/Na</th> <th>CO₃/SO₄</th>	Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
880 0.42 0.02 1.40 0.04 0.03 882 0.20 0.01 1.19 0.02 0.02 884 0.25 0.02 1.16 0.04 0.03 886 0.32 0.03 1.29 0.05 0.04 888 0.32 0.03 1.29 0.05 0.04 890 3.13 0.03 4.10 0.05 0.01 892 0.28 0.01 1.26 0.03 0.02 894 0.10 0.02 1.09 0.04 0.03 896 0.22 0.06 1.22 0.06 0.05 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.23 0.12 0.10 904 0.30 0.01 1.23 0.12 0.10 905 0.31 0.02 1.21 0.03 0.03 910 0.24 0.02 1.21 <td>878</td> <td>0.27</td> <td>0.02</td> <td>1.25</td> <td>0.05</td> <td>. 0.04</td>	878	0.27	0.02	1.25	0.05	. 0.04
822 0.20 0.01 1.19 0.02 0.02 884 0.12 0.03 1.23 0.04 0.03 886 0.13 0.02 1.16 0.04 0.03 888 0.32 0.03 1.29 0.05 0.01 890 3.13 0.03 1.29 0.04 0.03 896 0.22 0.06 1.22 0.06 0.05 898 1.42 0.11 2.40 0.13 0.06 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.26 0.04 0.03 906 0.31 0.02 1.29 0.03 0.02 906 0.31 0.02 1.23 0.12 0.10 910 0.24 0.02 1.23 0.03 0.03 910 0.24 0.02 1.23 0.14 0.11 910 0.26 0.11 1.00 <td>880</td> <td>0.42</td> <td>0.02</td> <td>1.40</td> <td>0.04</td> <td>0.03</td>	880	0.42	0.02	1.40	0.04	0.03
884 0.25 0.02 1.23 0.05 0.04 886 0.18 0.02 1.16 0.04 0.03 888 0.32 0.03 4.10 0.05 0.04 890 3.13 0.03 4.10 0.05 0.01 892 0.28 0.01 1.26 0.03 0.02 894 0.10 0.02 1.09 0.04 0.03 896 0.22 0.06 1.22 0.06 0.05 898 1.42 0.11 2.40 0.13 0.02 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.23 0.03 0.03 906 0.31 0.02 1.23 0.12 0.10 908 0.25 0.10 1.23 0.12 0.10 910 0.24 0.02 1.21 0.03 0.30 916 0.45 0.26 1.40 <td>882</td> <td>0.20</td> <td>0.01</td> <td>1.19</td> <td>0.02</td> <td>0.02</td>	882	0.20	0.01	1.19	0.02	0.02
886 0.18 0.02 1.16 0.04 0.03 888 0.32 0.03 1.29 0.05 0.04 890 3.13 0.03 4.10 0.05 0.01 892 0.28 0.01 1.26 0.03 0.02 896 0.22 0.06 1.22 0.06 0.05 898 1.42 0.11 2.40 0.13 0.06 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.26 0.04 0.03 904 0.30 0.01 1.23 0.01 0.01 905 0.25 0.10 1.23 0.03 0.03 910 0.24 0.02 1.23 0.03 0.03 910 0.24 0.02 1.23 0.03 0.30 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 <td>884</td> <td>0.25</td> <td>0.02</td> <td>1.23</td> <td>0.05</td> <td>0.04</td>	884	0.25	0.02	1.23	0.05	0.04
888 0.32 0.03 1.29 0.05 0.04 890 3.13 0.03 4.10 0.05 0.01 892 0.28 0.01 1.26 0.03 0.02 894 0.10 0.02 1.09 0.04 0.03 896 0.22 0.06 1.22 0.06 0.05 898 1.42 0.11 2.40 0.13 0.06 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.23 0.03 0.03 904 0.30 0.01 1.29 0.03 0.03 906 0.31 0.02 1.23 0.03 0.03 910 0.24 0.02 1.23 0.03 0.03 910 0.24 0.02 1.21 0.03 0.30 914 0.05 0.25 1.00 0.30 0.30 914 0.05 0.04 0.31 <td>886</td> <td>0.18</td> <td>0.02</td> <td>1.16</td> <td>0.04</td> <td>0.03</td>	886	0.18	0.02	1.16	0.04	0.03
890 3.13 0.03 4.10 0.05 0.01 892 0.28 0.01 1.26 0.03 0.02 894 0.10 0.02 1.09 0.04 0.03 896 0.22 0.06 1.22 0.06 0.05 898 1.42 0.11 2.40 0.13 0.06 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.26 0.04 0.03 904 0.30 0.01 1.23 0.12 0.10 906 0.31 0.02 1.23 0.03 0.03 910 0.24 0.02 1.23 0.03 0.30 912 0.23 0.02 1.21 0.03 0.30 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.31 0.22 916 0.21 0.02 1.22 <td>888</td> <td>0.32</td> <td>0.03</td> <td>1.29</td> <td>0.05</td> <td>0.04</td>	888	0.32	0.03	1.29	0.05	0.04
892 0.28 0.01 1.26 0.03 0.02 894 0.10 0.02 1.09 0.04 0.03 896 0.22 0.06 1.22 0.06 0.05 898 1.42 0.11 2.40 0.13 0.06 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.26 0.04 0.03 906 0.31 0.02 1.29 0.03 0.03 906 0.31 0.02 1.23 0.12 0.10 908 0.25 0.10 1.23 0.03 0.03 910 0.24 0.02 1.21 0.03 0.03 912 0.23 0.02 1.00 0.07 0.07 920 0.02 0.00 1.00 0.07 0.02 924 0.73 0.18 1.70 0.22 0.13 920 0.26 0.11 1.23 <td>890</td> <td>3.13</td> <td>0.03</td> <td>4.10</td> <td>0.05</td> <td>0.01</td>	890	3.13	0.03	4.10	0.05	0.01
894 0.10 0.02 1.09 0.04 0.03 896 0.22 0.06 1.22 0.06 0.03 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.26 0.044 0.03 904 0.30 0.01 1.29 0.03 0.02 906 0.31 0.02 1.29 0.03 0.03 906 0.31 0.02 1.23 0.12 0.10 910 0.24 0.02 1.23 0.03 0.03 912 0.23 0.02 1.21 0.03 0.30 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.31 0.22 918 0.04 0.04 1.00 0.07 0.07 920 0.26 0.11 1.23 0.14 0.11 924 0.31 0.02 1.21 <td>892</td> <td>0.28</td> <td>0.01</td> <td>1.26</td> <td>0.03</td> <td>0.02</td>	892	0.28	0.01	1.26	0.03	0.02
896 0.22 0.06 1.22 0.06 0.05 898 1.42 0.11 2.40 0.13 0.06 900 0.14 0.02 1.26 0.04 0.03 904 0.30 0.01 1.29 0.03 0.02 906 0.31 0.02 1.23 0.12 0.13 908 0.25 0.10 1.23 0.03 0.03 908 0.24 0.02 1.21 0.03 0.03 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.07 0.07 920 0.02 0.002 0.02 0.02 0.02 924 0.73 0.18 1.70 0.22 0.13 926 0.21 0.12 1.12 0.04 0.07 928 0.23 0.02 1.24 0.09 </td <td>894</td> <td>0.10</td> <td>0.02</td> <td>1.09</td> <td>0.04</td> <td>0.03</td>	894	0.10	0.02	1.09	0.04	0.03
898 1.42 0.11 2.40 0.13 0.06 900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.26 0.04 0.03 904 0.30 0.01 1.29 0.03 0.02 906 0.31 0.02 1.23 0.12 0.13 908 0.25 0.10 1.23 0.12 0.10 910 0.24 0.02 1.23 0.03 0.03 914 0.05 0.25 1.00 0.30 0.30 914 0.05 0.25 1.00 0.31 0.22 918 0.04 0.06 1.00 0.07 0.07 920 0.02 0.00 1.00 0.02 0.02 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946	896	0.22	0.06	1.22	0.06	0.05
900 0.14 0.08 1.12 0.09 0.08 902 0.28 0.02 1.26 0.04 0.03 904 0.30 0.01 1.29 0.03 0.03 905 0.25 0.10 1.23 0.12 0.10 910 0.24 0.02 1.23 0.03 0.03 912 0.23 0.02 1.21 0.03 0.03 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.31 0.22 918 0.04 0.04 1.00 0.07 0.77 920 0.02 0.00 1.00 0.09 0.07 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 934 0.23 0.02 1.21 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.30 0.03 0.04 940 0.14 0.01 1.12 0.12 0.11 942 0.03 0.02 1.06 0.04 0.04 945 0.22 0.00 1.00 0.03 0.03 946 <td>898</td> <td>1.42</td> <td>0.11</td> <td>2.40</td> <td>0.13</td> <td>0.06</td>	898	1.42	0.11	2.40	0.13	0.06
902 0.28 0.02 1.26 0.04 0.03 904 0.30 0.01 1.29 0.03 0.02 906 0.31 0.02 1.23 0.12 0.10 910 0.24 0.02 1.23 0.03 0.03 912 0.23 0.02 1.21 0.03 0.03 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.31 0.22 918 0.04 0.04 1.00 0.07 0.07 920 0.02 0.00 1.00 0.02 0.02 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.12 0.14 0.11 936 0.21 0.10 1.00 0.07 0.06 944 0.03 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 <td>900</td> <td>0.14</td> <td>0.08</td> <td>1.12</td> <td>0.09</td> <td>0.08</td>	900	0.14	0.08	1.12	0.09	0.08
9040.300.011.290.030.029060.310.021.290.030.039080.250.101.230.120.109100.240.021.230.030.039120.230.021.210.030.039140.050.251.000.300.309160.450.261.400.310.229180.040.041.000.070.079200.020.001.000.020.029240.730.181.700.220.139260.310.031.240.090.079280.230.021.210.040.039300.260.111.230.140.119320.240.021.220.050.049340.230.021.210.050.049360.210.121.180.140.129380.370.031.340.050.049400.140.101.120.120.119440.030.021.060.040.049460.180.021.060.040.049460.140.011.100.030.039500.020.021.060.040.039560.110.011.100.030.039560.120.021.31	902	0.28	0.02	1.26	0.04	0.03
906 0.31 0.02 1.29 0.03 0.03 908 0.25 0.10 1.23 0.12 0.10 910 0.24 0.02 1.23 0.03 0.03 912 0.23 0.02 1.21 0.03 0.03 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.31 0.22 918 0.04 0.00 0.07 0.07 920 0.02 0.00 1.00 0.07 0.07 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 934 0.23 0.02 1.22 0.05 0.04 936 0.21 0.12 1.18 0.14 <td>904</td> <td>0.30</td> <td>0.01</td> <td>1.29</td> <td>0.03</td> <td>0.02</td>	904	0.30	0.01	1.29	0.03	0.02
908 0.25 0.10 1.23 0.12 0.10 910 0.24 0.02 1.23 0.03 0.03 912 0.23 0.02 1.21 0.03 0.03 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.31 0.22 920 0.02 0.00 1.00 0.07 0.07 920 0.02 0.00 1.00 0.02 0.02 924 0.73 0.18 1.70 0.22 0.03 926 0.31 0.02 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.14 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.14 936 0.21 0.12 1.18 0.14 0.14 940 0.14 0.10 1.12 0.12 0.14 944 0.03 0.02 1.00 0.04 0.04 945 0.08 0.02 1.06 0.04 0.03 950 0.02 0.02 1.06 0.04 0.03 954 0.08 0.02 1.06 0.04 0.03 958 0.11 0.01 1.10 0.03 0.03 958 0.11 0.01 1.00 0.03 0.03 959	906	0.31	0.02	1.29	0.03	0.03
910 0.24 0.02 1.23 0.03 0.03 912 0.23 0.02 1.21 0.03 0.33 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.31 0.22 918 0.04 0.04 1.00 0.07 0.07 920 0.02 0.00 1.00 0.02 0.02 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.11 942 0.03 0.02 1.06 0.04 0.04 944 0.03 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.03 950 0.02 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.02 1.38 0.00 0.00 <td>908</td> <td>0.25</td> <td>0.10</td> <td>1.23</td> <td>0.12</td> <td>0.10</td>	908	0.25	0.10	1.23	0.12	0.10
912 0.23 0.02 1.21 0.03 0.03 914 0.05 0.25 1.00 0.30 0.30 916 0.45 0.26 1.40 0.31 0.22 918 0.04 0.04 1.00 0.07 0.07 920 0.02 0.00 1.00 0.02 0.02 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.12 0.11 942 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.00 0.03 0.03 950 0.02 0.02 1.00 0.04 0.04 948 0.14 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 956 0.11 0.01 1.10 0.03 <td>910</td> <td>0.24</td> <td>0.02</td> <td>1.23</td> <td>0.03</td> <td>0.03</td>	910	0.24	0.02	1.23	0.03	0.03
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	912	0.23	0.02	1.21	0.03	0.03
916 0.45 0.26 1.40 0.31 0.22 918 0.04 0.04 1.00 0.07 0.07 920 0.02 0.00 1.00 0.02 0.02 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.21 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.12 0.11 942 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.03 955 0.02 0.02 1.00 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 960 0.12 0.02 1.31 0.04 0.03 974 0.92 0.00 1.38 0.00 <td>914</td> <td>0.05</td> <td>0.25</td> <td>1.00</td> <td>0.30</td> <td>0.30</td>	914	0.05	0.25	1.00	0.30	0.30
918 0.04 0.04 1.00 0.07 0.07 920 0.02 0.00 1.00 0.02 0.02 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.12 0.11 942 0.03 0.02 1.00 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 947 0.02 0.02 1.00 0.03 0.03 958 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.06 0.04 0.04 954 0.08 0.02 1.28 0.00 0.00 956 0.12 0.02 1.31 0.04 0.03 958 0.01 0.02 1.32 0.07 0.06 959 0.02 1.31 0.04 0.03 958 0.14	916	0.45	0.26	1.40	0.31	0.22
920 0.02 0.00 1.00 0.02 0.02 924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.11 0.14 942 0.03 0.02 1.00 0.06 0.06 944 0.03 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 948 0.14 0.03 1.10 0.07 0.06 950 0.02 0.00 1.00 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 958 0.01 0.01 1.00 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 974 0.92 0.02 1.31 0.04 0.03 974 0.92 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.22 984 <td< td=""><td>918</td><td>0.04</td><td>0.04</td><td>1.00</td><td>0.07</td><td>0.07</td></td<>	918	0.04	0.04	1.00	0.07	0.07
924 0.73 0.18 1.70 0.22 0.13 926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.11 0.14 942 0.03 0.03 1.00 0.06 0.06 944 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.03 950 0.02 0.02 1.00 0.04 0.04 954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.12 0.02 1.31 0.04 0.03 956 0.12 0.02 1.31 0.04 0.03 956 0.12 0.02 1.32 0.07 <td>920</td> <td>0.02</td> <td>0.00</td> <td>1.00</td> <td>0.02</td> <td>0.02</td>	920	0.02	0.00	1.00	0.02	0.02
926 0.31 0.03 1.24 0.09 0.07 928 0.23 0.02 1.21 0.04 0.03 930 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.12 0.11 942 0.03 0.03 1.00 0.06 0.06 944 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 948 0.14 0.03 1.10 0.07 0.06 950 0.02 0.02 1.00 0.04 0.04 954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 958 0.01 0.01 1.00 0.03 0.03 960 0.12 0.02 1.31 0.04 0.03 974 0.92 0.00 1.28 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 976 0.16 0.09 1.38 0.00 0.00 976 0.16 0.09 1.15 0.04 <td>924</td> <td>0.73</td> <td>0.18</td> <td>1.70</td> <td>0.22</td> <td>0.13</td>	924	0.73	0.18	1.70	0.22	0.13
9280.230.021.210.040.039300.260.111.230.140.119320.240.021.220.050.049340.230.021.210.050.049360.210.121.180.140.129380.370.031.340.050.049400.140.101.120.120.119420.030.021.010.060.069440.030.021.010.040.049460.080.021.060.040.049460.080.021.060.040.049460.080.021.060.040.049500.020.001.000.020.029520.020.021.060.040.039540.080.021.060.040.039580.010.011.000.030.039600.120.021.310.040.039720.320.021.310.040.039740.920.001.920.000.009760.160.091.140.110.099780.370.021.320.020.029880.110.021.100.030.039840.150.001.150.000.009860.180.111.15	926	0.31	0.03	1.24	0.09	0.07
330 0.26 0.11 1.23 0.14 0.11 932 0.24 0.02 1.22 0.05 0.04 934 0.23 0.02 1.21 0.05 0.04 936 0.21 0.12 1.18 0.14 0.12 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.12 0.11 942 0.03 0.03 1.00 0.06 0.06 944 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 945 0.02 0.02 1.00 0.02 0.02 952 0.02 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 964 0.38 0.00 1.38 0.00 0.00 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.03 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.03 0.03 984 0.15 0.00 1.15 0.00 <td>928</td> <td>0.23</td> <td>0.02</td> <td>1.21</td> <td>0.04</td> <td>0.03</td>	928	0.23	0.02	1.21	0.04	0.03
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	930	0.26	0.11	1.23	0.14	0.11
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	932	0.24	0.02	1.22	0.05	0.04
336 0.21 0.12 1.18 0.02 0.02 938 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.12 0.11 942 0.03 0.02 1.00 0.06 0.06 944 0.03 0.02 1.01 0.04 0.04 944 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 948 0.14 0.03 1.10 0.07 0.06 950 0.02 0.00 1.00 0.02 0.02 952 0.02 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 966 0.12 0.02 1.31 0.04 0.03 966 0.12 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 986 0.18 0.11 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 <td>934</td> <td>0.23</td> <td>0.02</td> <td>1.21</td> <td>0.05</td> <td>0.04</td>	934	0.23	0.02	1.21	0.05	0.04
338 0.37 0.03 1.34 0.05 0.04 940 0.14 0.10 1.12 0.12 0.11 942 0.03 0.03 1.00 0.06 0.06 944 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 948 0.14 0.03 1.10 0.07 0.06 950 0.02 0.00 1.00 0.02 0.02 952 0.02 0.02 1.06 0.04 0.04 954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.12 0.02 1.11 0.03 0.03 960 0.12 0.02 1.31 0.04 0.03 962 0.28 0.00 1.38 0.00 0.00 974 0.92 0.00 1.92 <td>936</td> <td>0.21</td> <td>0.12</td> <td>1 18</td> <td>0.14</td> <td>0.17</td>	936	0.21	0.12	1 18	0.14	0.17
940 0.14 0.10 1.12 0.12 0.11 942 0.03 0.03 1.00 0.06 0.06 944 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 948 0.14 0.03 1.10 0.07 0.06 950 0.02 0.00 1.00 0.02 0.02 952 0.02 0.02 1.06 0.04 0.03 954 0.08 0.02 1.06 0.04 0.03 955 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 <td>938</td> <td>0.37</td> <td>0.03</td> <td>1 34</td> <td>0.05</td> <td>0.04</td>	938	0.37	0.03	1 34	0.05	0.04
942 0.03 0.03 1.00 0.06 0.06 944 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 948 0.14 0.03 1.10 0.07 0.06 950 0.02 0.00 1.00 0.02 0.02 952 0.02 0.02 1.06 0.04 0.04 954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 977 0.32 0.02 1.32 <td>940</td> <td>014</td> <td>0.10</td> <td>1.12</td> <td>0.12</td> <td>0.11</td>	940	014	0.10	1.12	0.12	0.11
944 0.03 0.02 1.01 0.04 0.04 946 0.08 0.02 1.06 0.04 0.04 948 0.14 0.03 1.10 0.07 0.06 950 0.02 0.00 1.00 0.02 0.02 952 0.02 0.02 1.00 0.04 0.04 954 0.08 0.02 1.06 0.04 0.04 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.15 0.00 0.00 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	942	0.03	0.03	1.00	0.06	0.06
946 0.08 0.02 1.06 0.04 0.04 948 0.14 0.03 1.10 0.07 0.06 950 0.02 0.00 1.00 0.02 0.02 952 0.02 0.02 1.00 0.04 0.04 954 0.08 0.02 1.06 0.04 0.04 954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 <td>944</td> <td>0.03</td> <td>0.02</td> <td>1.00</td> <td>0.04</td> <td>0.04</td>	944	0.03	0.02	1.00	0.04	0.04
948 0.14 0.02 1.10 0.07 0.06 950 0.02 0.00 1.00 0.02 0.02 952 0.02 0.02 1.00 0.04 0.04 954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 956 0.12 0.02 1.11 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 972 0.32 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	946	0.08	0.02	1.06	0.04	0.04
950 0.02 0.00 1.00 0.02 0.02 952 0.02 0.02 1.00 0.04 0.04 954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 958 0.01 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 974 0.92 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	948	0.14	0.03	1.10	0.07	0.06
952 0.02 0.02 1.00 0.04 0.04 954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 956 0.12 0.02 1.11 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 972 0.32 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	950	0.02	0.00	1.00	0.02	0.02
954 0.08 0.02 1.06 0.04 0.03 956 0.11 0.01 1.10 0.03 0.03 958 0.01 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 972 0.32 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	952	0.02	0.02	1.00	0.04	0.04
956 0.11 0.01 1.10 0.03 0.03 956 0.11 0.01 1.00 0.03 0.03 958 0.01 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 972 0.32 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 <td>954</td> <td>0.08</td> <td>0.02</td> <td>1.06</td> <td>0.04</td> <td>0.03</td>	954	0.08	0.02	1.06	0.04	0.03
100 0.01 1.00 0.03 0.03 958 0.01 0.01 1.00 0.03 0.03 960 0.12 0.02 1.11 0.03 0.03 962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 972 0.32 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.10 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	956	0.11	0.01	1.10	0.03	0.03
9600.120.021.110.030.039600.120.021.110.030.039620.280.001.280.000.009640.380.001.380.000.009720.320.021.310.040.039740.920.001.920.000.009760.160.091.140.110.099780.370.021.320.070.069800.140.011.130.020.029820.170.021.160.030.039840.150.001.150.000.009860.180.111.150.130.119880.110.021.100.030.039900.110.121.080.150.149920.470.021.450.040.039940.450.131.410.190.149961.420.222.370.270.11	958	0.01	0.01	1.00	0.03	0.03
962 0.28 0.00 1.28 0.00 0.00 964 0.38 0.00 1.38 0.00 0.00 972 0.32 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	960	0.12	0.02	1.11	0.03	0.03
964 0.38 0.00 1.38 0.00 0.00 972 0.32 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	962	0.28	0.00	1.28	0.00	0.00
972 0.32 0.02 1.31 0.04 0.03 974 0.92 0.00 1.92 0.00 0.09 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	964	0.38	0.00	1.38	0.00	0.00
974 0.92 0.00 1.92 0.00 0.00 974 0.92 0.00 1.92 0.00 0.00 976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	972	0.32	0.02	1 31	0.04	0.03
976 0.16 0.09 1.14 0.11 0.09 978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	974	0.92	0.00	1.97	0.00	0.00
978 0.37 0.02 1.32 0.07 0.06 980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	976	016	0.09	1 14	0.00	0.09
980 0.14 0.01 1.13 0.02 0.02 982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	978	0.37	0.07	1 32	0.07	0.05
982 0.17 0.02 1.16 0.03 0.03 984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	980	0.14	0.01	1.13	0.02	0.02
984 0.15 0.00 1.15 0.00 0.00 986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14	982	0.17	0.02	1.16	0.03	0.02
986 0.18 0.11 1.15 0.13 0.11 988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	984	0.15	0.00	1.15	0.00	0.00
988 0.11 0.02 1.10 0.03 0.03 990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	986	0.18	0.11	115	013	0.11
990 0.11 0.12 1.08 0.15 0.14 992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 995 1.42 0.22 2.37 0.27 0.11	988	0.11	0.02	1.10	0.03	0.03
992 0.47 0.02 1.45 0.04 0.03 994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	990	0.11	0.12	1.08	0.15	0.14
994 0.45 0.13 1.41 0.19 0.14 996 1.42 0.22 2.37 0.27 0.11	907	0.47	0.02	1.45	0.04	0.14
996 142 022 237 027 011	904	0.45	0.13	1 41	0.10	0.14
	906	1 42	0.22	2 27	0.15	0.14
998 1.08 0.17 2.05 0.20 0.10	998	1.08	0.17	2.05	0.20	0.10

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
1000	0.31	0.22	1.25	0.29	. 0.23
1002	0.13	0.02	1.11	0.03	0.03
1004	0.20	0.02	1.17	0.04	0.04
1006	0.09	0.00	1.09	0.04	0.04
1008	0.21	0.07	1.21	0.07	0.06
1010	0.02	0.02	1.00	0.04	0.04
1012	0.09	0.06	0.99	0.06	0.06
1014	0.26	0.15	I.22	0.19	0.15
1016	0.15	0.16	1.03	0:19	0.18
1018	0.03	0.14	1.00	0.16	0.16
1020	0.26	0.10	1.24	0.12	0.10
1022	0.09	0.02	1.08	0.04	0.03
1024	0.25	0.11	1.23	0.13	0.11
1026	0.14	0.09	1.12	0.11	0.10
1028	0.26	0.02	1.21	0.07	0.05
1030	0.34	0.13	1.28	0.20	0.15
1032	0.62	0.00	1.62	0.00	0.00
1034	0.14	0.11	1.12	0.13	0.11
1036	0.12	0.01	1.11	0.02	0.02
1038	0.19	0.00	1.19	0.00	0.00
1040	0.07	0.00	1.07	0.00	0.00
1042	0.19	0.00	1.19	0.00	0.00
1044	0.05	0.04	1.05	0.04	0.03
1046	0.13	0.03	1.13	0.03	0.03
1048	0.18	0.11	1.11	0.18	0.16
1050	0.09	0.13	1.07	0.15	0.14
1052	0.12	0.00	1.12	0.00	0.00
1054	0.19	0.11	1.17	0.13	0.11
1056	0.21	0.12	1.18	0.14	0.12
1058	0.17	0.01	1.15	0.03	0.03
1060	0.18	0.01	1.17	0.03	0.02
1062	0.14	0.02	1.10	0.06	0.06
1064	0.28	0.00	1.28	0.00	0.00
1066	3.53	0.02	4.51	0.04	0.01
1068	0.18	0.03	1.12	0.09	0.08
1070	0.37	0.02	1.30	0.09	0.07
1072	0.22	0.02	1.20	0.05	0.04
1074	0.27	0.02	1.21	0.09	0.07
1076	0.29	0.02	1.24	0.06	0.05
1078	0.15	0.01	1.11	0.05	0.04
1080	0.11	0.02	1.09	0.04	0.04
1082	0.25	0.00	1.25	0.00	0.00
1084	0.05	0.02	1.00	0.07	0.07
1086	0.12	0.14	1.09	0.17	0.15
1088	0.59	0.02	1.56	0.04	0.03
1090	0.15	0.13	1.12	0.16	0.14
1092	0.13	0.00	1.13	0.00	0.00
1094	5.25	2.16	5.83	2.58	0.44
1096	0.12	0.00	1.09	0.02	0.02
1098	0.15	0.00	1.15	0.00	0.00
1100	0.00	0.00	1.00	0.01	0.01
1102	0.10	0.01	1.09	0.01	0.01
1104	0.00	0.00	1.00	0.00	0.00
1106	0.12	0.02	1.10	0.04	0.03
1108	0.14	0.00	1.05	0.00	0.00
1110	0.00	0.00	1.00	0.00	0.00
1112	0.19	0.01	1.18	0.03	0.02

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Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
1114	0.38	0.00	1.38	0.00	. 0.00
1116	0.75	0.00	1.75	0.00	0.00
1118	0.18	0.01	1.16	0.03	0.02
1120	0.11	0.01	1.10	0.02	0.02
1122	0.49	0.00	1.49	0.00	0.00
1124	0.41	0.00	1.41	0.00	0.00
1126	0.17	0.00	1.17	0.00	0.00
1128	1.11	0.00	2.11	0.00	0.00
1130	0.23	0.08	1.23	0.08	0.06
1132	0.22	0.00	1.22	0.00	0.00
1134	1.49	0.00	2.49	0.00	0.00
1136	0.23	0.00	1.23	0.00	0.00
1138	0.11	0.00	1.11	0.00	0.00
1140	0.67	0.00	1.67	0.00	0.00
1142	0.16	0.00	1.16	0.00	0.00
1144	0.02	0.02	1.00	0.04	0.04
1148	0.15	0.00	1.14	0.02	0.01
1150	0.11	0.00	1.11	0.00	0.00
1152	0.34	0.00	1.34	0.00	0.00
1154	0.63	0.01	1.59	0.05	0.03
1156	0.23	0.07	1.21	0.03	0.03
1158	0.12	0.02	1 12	0.00	0.00
1160	0.25	0.00	1 74	0.00	0.00
1162	0.22	0.01	1.20	0.03	0.02
1164	0.43	0.07	1.36	0.09	0.02
1166	0.13	0.02	1.50	0.05	0.05
1168	0.13	0.02	1.11	0.00	0.03
1100	0.15	0.01	1.11	0.03	0.02
1170	0.40	0.01	1.00	0.03	0.02
1172	0.09	0.00	1.07	0.00	0.00
1174	0.20	0.12	1.17	0.14	0.12
1170	0.37	0.02	1.55	0.03	0.03
1178	0.18	0.00	1.10	0.00	0.00
1160	0.26	0.02	1.27	0.03	0.03
1164	0.30	0.00	1.34	0.03	0.02
1104	0.10	0.02	1.10	0.08	0.08
1100	0.08	0.01	1.00	0.03	0.02
1188	0.00	0.00	1.00	0.00	0.00
1190	0.07	0.00	1.07	0.00	0.00
1192	0.22	0.00	1.22	0.00	0.00
1194	0.00	0.00	1.00	0.00	0.00
1198	0.00	0.00	1.00	0.00	0.00
1200	0.00	0.00	1.00	0.00	0.00
1202	0.01	0.01	1.00	0.02	0.02
1204	0.00	0.00	1.00	0.00	0.00
1206	0.06	0.00	1.06	0.00	0.00
1208	0.08	0.00	1.08	0.00	0.00
1210	0.00	0.03	1.00	0.03	0.03
1212	0.00	0.00	1.00	0.00	0.00
1214	0.00	0.00	1.00	0.00	0.00
1216	0.00	0.00	1.00	0.00	0.00
1218	0.00	0.00	1.00	0.00	0.00
1220	0.06	0.00	1.06	0.00	0.00
1222	0.00	0.00	1.00	0.00	0.00
1224	0.00	0.00	1.00	0.00	0.00
1326	0.02	0.02	1.00	0.04	0.04
1328	0.01	0.01	1.00	0.02	0.02
1330	0.11	0.01	1.10	0.02	0.02

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
1332	0.01	0.01	1.00	0.02	. 0.02
1334	0.00	0.00	1.00	0.00	0.00
1336	0.06	0.00	1.06	0.00	0.00
1338	0.00	0.04	1.00	0.04	0.04
1340	0.16	0.02	1.14	0.05	0.05
1342	0.02	0.02	1.00	0.03	0.03
1344	0.20	0.00	1.20	0.00	0.00
1346	0.50	0.00	1.50	0.03	0.02
1348	0.11	0.00	1.09	0.03	0.02
1350	0.07	0.07	1.00	0.15	0.15
1352	0.14	0.14	1.00	0.27	0.27
1354	0.17	0.04	1.13	0.09	0.08
1356	0.09	0.08	1.09	0.14	0.13
1358	0.26	0.18	1.08	0.07	0.06
1360	0.29	0.14	1.14	0.04	0.04
1362	0.32	0.03	1.19	0.06	0.05
1364	0.31	0.15	1.14	0.02	0.02
1366	0.09	0.00	1.09	0.00	0.00
1368	0.28	0.09	1.26	0.12	0.09
1370	0.09	0.17	0.92	0.00	0.00
1372	0.16	0.03	0.99	0.05	0.06
1374	0.15	0.02	1.13	0.04	0.03
1376	0.17	0.20	1.13	0.24	0.21
1380	0.14	0.16	1.11	0.19	0.18
1382	0.25	0.18	1.17	0.31	0.26
1384	0.18	0.14	1.08	0.24	0.22
1386	0.22	0.21	1.10	0.33	0.30
1388	0.18	0.26	1.13	0.32	0.28
1390	0.19	0.15	1.17	0.18	0.16
1392	0.29	0.33	1.20	0.42	0.35
1394	0.19	0.32	1.11	0.43	0.39
1396	0.11	0.10	1.09	0.11	0.10
1398	0.02	0.09	1.00	0.11	0.11
1400	0.21	0.11	1.19	0.14	0.12
1402	0.09	0.02	1.07	0.04	0.04
1404	0.10	0.11	1.07	0.14	0.13
1406	0.26	0.23	1.21	0.28	0.23
1408	0.00	0.00	1.00	0.00	0.00
1410	0.04	0.01	1.04	0.02	0.01
1412	0.03	0.14	1.00	0.18	0.18
1414	0.02	0.00	1.00	0.02	0.02
1416	0.00	0.00	1.00	0.00	0.00
1418	0.00	0.00	1.00	0.00	0.00
1420	0.00	0.00	1.00	0.00	0.00
1422	0.00	0.00	1.00	0.00	0.00
1424	0.00	0.00	1.00	0.00	0.00
1426	0.01	0.01	1.00	0.02	0.02
1428	0.03	0.16	1.00	0.19	0.19
1430	0.03	0.03	1.00	0.06	0.06
1432	0.01	0.01	1.00	0.02	0.02
1434	0.02	0.02	1.00	0.03	0.03
1436	0.06	0.12	1.00	0.18	0.18
1438	0.07	0.03	1.00	0.09	0.09
1440	0.00	0.00	1.00	0.00	0.00
1442	0.04	0.02	1.00	0.06	0.06
1444	0.03	0.03	1.00	0.06	0.06
1446	0.03	0.11	1.00	0.15	0.15

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
1448	0.04	0.02	1.00	0.06	. 0.06
1450	0.04	0.04	1.00	0.07	0.07
1452	0.02	0.09	1.00	0.11	0.11
1454	0.02	0.02	1.00	0.05	0.05
1456	0.02	0.02	1.00	0.04	0.04
1458	0.06	0.01	1.05	0.02	0.02
1460	0.02	0.02	1.00	0.04	0.04
1462	0.04	0.01	1.00	0.05	0.05
1464	0.00	0.00	1.00	0.00	0.00
1466	0.00	0.00	1.00	0.00	0.00
1468	0.02	0.14	1.00	0.16	0.16
1470	0.00	0.00	1.00	0.00	0.00
1472	0.00	0.00	1.00	0.00	0.00
1474	0.00	0.00	1.00	0.00	0.00
1476	0.00	0.00	1.00	0.00	0.00
1478	0.00	0.03	1.00	0.03	0.03
1480	0.00	0.00	1.00	0.00	0.00
1482	0.02	0.13	1.00	0.16	0.16
1484	0.02	0.08	1.00	0.09	0.09
1486	0.05	0.09	1.00	0.14	0.14
1488	0.08	0.01	1.04	0.06	0.06
1490	0.00	0.00	1.00	0.00	0.00
1492	0.04	0.01	1.00	0.06	0.06
1494	0.00	0.00	1.00	0.00	0.00
1496	0.01	0.01	1.00	0.02	0.02
1498	0.02	0.07	1.00	0.02	0.04
1500	0.05	0.18	1.00	0.22	0.22
1504	0.02	0.10	1.00	0.06	0.06
1504	0.02	0.02	1.00	0.00	0.00
1508	0.07	0.02	1.00	0.04	0.07
1510	0.01	0.01	1.00	0.02	0.02
1512	0.07	0.13	1.00	0.02	0.02
1514	0.01	0.15	1.00	0.20	0.20
1516	0.03	0.01	1.00	0.05	0.05
1518	0.03	0.02	1.00	0.05	0.05
1520	0.02	0.00	1.00	0.02	0.10
1520	0.02	0.00	1.00	0.02	0.02
1524	0.00	0.00	1.00	0.00	0.00
1524	0.00	0.00	1.00	0.00	0.00
1520	0.00	0.00	1.00	0.00	0.00
1520	0.07	0.02	1.05	0.04	0.04
1530	0.00	0.00	1.00	0.00	0.00
1534	0.00	0.00	1.00	0.00	0.00
1534	0.08	0.12	1.00	0.13	0.14
1530	0.01	0.01	1.00	0.02	0.02
1530	0.02	0.00	1.00	0.02	0.02
1540	0.02	0.00	1.00	0.02	0.02
1542	0.01	0.01	1.00	0.01	0.01
1544	0.05	0.00	1.05	0.00	0.00
1540	0.00	0.00	1.00	0.00	0.00
1548	0.03	0.01	1.00	0.05	0.05
1550	0.01	0.01	1.00	0.02	0.02
1552	0.00	0.00	1.00	0.00	0.00
1554	0.07	0.14	1.00	0.21	0.21
1556	80.0	0.13	1.00	0.21	0.21
1558	0.04	0.08	1.00	0.12	0.12
1560	0.00	0.00	1.00	0.00	0.00
1562	0.04	0.07	1.00	0.11	0.11

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Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
1564	0.04	0.01	1.00	0.05	. 0.05
1566	0.02	0.07	1.00	0.09	0.09
1568	0.07	0.11	1.00	0.17	0.17
1570	0.02	0.02	1.00	0.04	0.04
1572	0.03	0.13	1.00	0.16	0.16
1574	0.05	0.00	1.05	0.00	0.00
1576	0.02	0.08	1.00	0.10	0.10
1578	0.06	0.10	1.00	0.16	0.16
1580	0.00	0.00	1.00	0.00	0.00
1582	0.00	0.00	1.00	0.00	0.00
1586	0.02	0.12	1.00	0.14	0.14
1588	0.07	0.31	1.00	0.38	0.38
1590	0.02	0.02	1.00	0.04	0.04
1592	0.06	0.02	1.00	0.08	0.08
1594	0.02	0.10	1.00	0.12	0.12
1596	0.07	0.15	1.00	0.22	0.22
1598	0.01	0.01	1.00	0.03	0.03
1600	0.00	0.00	1.00	0.00	0.00
1602	0.00	0.00	1.00	0.00	0.00
1604	0.01	0.01	1.00	0.02	0.02
1606	0.01	0.00	1.00	0.01	0.01
1608	0.00	0.03	1.00	0.03	0.03
1610	0.00	0.00	1.00	0.00	0.00
1612	0.00	0.00	1.00	0.00	0.00
1614	0.00	0.00	1.00	0.00	0.00
1616	0.01	0.01	1.00	0.02	0.02
1618	0.01	0.01	1.00	0.03	0.03
1620	0.03	0.01	1.00	0.05	0.05
1622	0.03	0.02	1.00	0.05	0.05
1624	0.02	0.11	1.00	0.13	0.13
1626	0.02	0.09	1.00	0.11	0.11
1628	0.06	0.02	1.00	0.08	0.08
1630	0.03	0.01	1.00	0.04	0.04
1632	0.01	0.01	1.00	0.02	0.02
1634	0.00	0.00	1.00	0.00	0.00
1636	0.00	0.00	1.00	0.00	0.00
1638	0.00	0.00	1.00	0.00	0.00
1640	0.02	0.00	1.00	0.02	0.02
1642	0.00	0.00	1.00	0.00	0.00
1644	0.01	0.01	1.00	0.02	0.02
1646	0.09	0.01	1.08	0.03	0.02
1648	0.01	0.01	1.00	0.02	0.02
1650	0.00	0.00	1.00	0.00	0.00
1652	0.06	0.00	1.06	0.00	0.00
1654	0.00	0.00	1.00	0.00	0.00
1656	0.00	0.00	1.00	0.00	0.00
1658	0.02	0.00	1.00	0.02	0.02
1660	0.13	0.08	1.07	0.14	0.13
1662	0.08	0.12	1.05	0.15	0.14
1664	0.03	0.13	1.00	0.16	0.16
1666	0.05	0.02	1.00	0.07	0.07
1668	0.01	0.01	1.00	0.03	0.03
1670	0.00	0.00	1.00	0.00	0.00
1672	0.00	0.00	1.00	0.00	0.00
1674	0.00	0.00	1.00	0.02	0.02
1676	0.00	0.00	1.00	0.00	0.00
1678	0.01	0.01	1.00	0.02	0.02

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO₃/SO₄
1680	0.00	0.00	1.00	0.00	. 0.00
1730	0.14	0.10	1.12	0.12	0.11
1732	0.23	0.08	1.18	0.13	0.11
1734	0.10	0.00	1.10	0.00	0.00
1736	0.16	0.08	1.12	0.12	0.11
1738	0.15	0.07	1.14	0.09	0.08
1740	0.00	0.00	1.00	0.00	0.00
1742	0.00	0.00	1.00	0.00	0.00
1744	0.00	0.00	1.00	0.00	0.00
1746	0.00	0.00	1.00	0.00	0.00
1748	0.00	0.00	1.00	0.00	0.00
1750	0.00	0.00	1.00	0.00	0.00
1750	0.00	0.00	1.00	0.00	0.00
1754	0.00	0.02	1.00	0.05	0.05
1756	0.02	0.00	1.00	0.00	0.00
1758	0.00	0.00	1.00	0.00	0.00
1760	0.00	0.00	1.00	0.00	0.00
1760	0.00	0.00	1.00	0.00	0.00
1764	0.00	0.00	1.00	0.00	0.00
1766	0.00	0.00	0.06	0.00	0.00
1700	0.00	0.00	1.00	0.00	0.00
1708	0.00	0.00	1.00	0.00	0.00
1770	0.02	0.12	1.00	0.13	0.15
1772	0.00	0.00	1.00	0.00	0.00
1774	0.00	0.00	1.00	0.00	0.00
1776	0.00	0.00	1.00	0.00	0.00
1778	0.06	0.00	1.06	0.00	0.00
1780	0.02	0.00	1.00	0.02	0.02
1782	0.00	0.00	1.00	0.00	0.00
1786	0.01	0.01	1.00	0.03	0.03
1788	0.00	0.00	1.00	0.00	0.00
1790	0.00	0.00	1.00	0.00	0.00
1792	0.01	0.01	1.00	0.02	0.02
1794	0.00	0.00	1.00	0.00	0.00
1796	0.21	0.02	1.19	0.04	0.03
1798	0.08	0.00	1.08	0.00	0.00
1800	0.00	0.00	1.00	0.00	0.00
1802	0.00	0.00	1.00	0.00	0.00
1804	0.00	0.00	1.00	0.00	0.00
1806	0.00	0.00	1.00	0.00	0.00
1808	0.00	0.00	1.00	0.00	0.00
1810	0.00	0.00	1.00	0.00	0.00
1812	0.00	0.00	1.00	0.00	0.00
1814	0.05	0.00	1.05	0.00	0.00
1816	0.00	0.00	1.00	0.00	0.00
1818	0.00	0.00	1.00	0.00	0.00
1820	0.00	0.00	1.00	0.00	0.00
1822	0.00	0.09	0.91	0.00	0.00
1824	0.00	0.00	1.00	0.00	0.00
1826	0.00	0.00	1.00	0.00	0.00
1828	0.09	0.00	1.09	0.00	0.00
1830	0.00	0.00	1.00	0.00	0.00
1832	0.00	0.00	1.00	0.00	0.00
1834	0.00	0.00	1.00	0.00	0.00
1836	0.00	0.00	1.00	0.00	0.00
1912	0.00	0.00	1.00	0.00	0.00
1914	0.00	0.00	1.00	0.00	0.00
1916	0.00	0.00	1.00	0.00	0.00
		0.00			0.00

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
1918	0.00	0.00	1.00	0.00	. 0.00
1920	0.00	0.00	1.00	0.00	0.00
1922	0.00	0.00	1.00	0.00	0.00
1924	0.00	0.00	1.00	0.00	0.00
1926	0.07	0.02	1.05	0.03	0.03
1928	0.00	0.00	1.00	0.00	0.00
1930	0.00	0.00	1.00	0.00	0.00
1932	0.00	0.00	1.00	0.00	0.00
1934	0.00	0.00	1.00	0.00	0.00
1936	0.00	0.00	1.00	0.00	0.00
1938	0.00	0.00	1.00	0.00	0.00
1940	0.00	0.00	1.00	0.00	0.00
1942	0.00	0.00	1.00	0.00	0.00
1944	0.00	0.00	1.00	0.00	0.00
1946	0.00	0.00	1.00	0.00	0.00
1948	0.00	0.00	1.00	0.00	0.00
1950	0.00	0.00	1.00	0.00	0.00
1952	0.00	0.00	1.00	0.00	0.00
1954	0.01	0.01	1.00	0.02	0.02
1956	0.00	0.00	1.00	0.00	0.00
1958	0.00	0.00	1.00	0.00	0.00
1960	0.00	0.00	1.00	0.00	0.00
1962	0.00	0.00	1.00	0.00	0.00
1964	0.00	0.00	1.00	0.00	0.00
1966	0.00	0.00	1.00	0.00	0.00
1968	0.00	0.00	1.00	0.00	0.00
1970	0.00	0.00	1.00	0.00	0.00
1974	0.00	0.00	1.00	0.00	0.00
1976	0.00	0.00	1.00	0.00	0.00
1978	0.00	0.00	1.00	0.00	0.00
1980	0.00	0.00	1.00	0.00	0.00
1982	0.00	0.00	1.00	0.00	0.00
1984	0.00	0.00	1.00	0.00	0.00
1986	0.00	0.00	1.00	0.00	0.00
1988	0.00	0.00	1.00	0.00	0.00
1990	0.00	0.00	1.00	0.00	0.00
1992	0.00	0.00	1.00	0.00	0.00
1994	0.00	0.00	1.00	0.00	0.00
1996	0.00	0.00	1.00	0.00	0.00
2000	0.00	0.00	1.00	0.00	0.00
2002	0.00	0.00	1.00	0.00	0.00
2004	0.00	0.00	1.00	0.00	0.00
2006	0.00	0.00	1.00	0.00	0.00
2008	0.00	0.00	1.00	0.00	0.00
2010	0.00	0.00	1.00	0.00	0.00
2012	0.00	0.00	1.00	0.00	0.00
2014	0.00	0.00	1.00	0.00	0.00
2016	0.00	0.00	1.00	0.00	0.00
2018	0.00	0.00	1.00	0.00	0.00
2020	0.00	0.00	1.00	0.00	0.00
2022	0.00	0.00	1.00	0.00	0.00
2024	0.00	0.00	1.00	0.00	0.00
2026	0.00	0.00	1.00	0.00	0.00
2028	0.00	0.00	1.00	0.00	0.00
2030	0.00	0.00	1.00	0.00	0.00
2032	0.00	0.00	1.00	0.00	0.00
2034	0.00	0.00	1.00	0.00	0.00

Depth (cm)	Ca/Na	Mg/Na	SO ₄ /Na	CO ₃ /Na	CO ₃ /SO ₄
2036	0.00	0.00	1.00	0.00	. 0.00
2038	0.00	0.00	0.96	0.00	0.00
2040	0.00	0.00	1.00	0.00	0.00
2042	0.00	0.00	1.00	0.00	0.00
2044	0.00	0.00	1.00	0.00	0.00
2046	0.00	0.00	1.00	0.00	0.00
2048	0.05	0.00	1.05	0.00	0.00
2052	0.00	0.00	1.00	0.00	0.00
2054	0.00	0.00	1.00	0.00	0.00
2056	0.00	0.00	1.00	0.00	0.00
2058	0.00	0.00	1.00	0.00	0.00
2060	0.00	0.00	I.00	0.00	0.00
2062	0.00	0.00	1.00	0.00	0.00
2064	0.00	0.00	1.00	0.00	0.00
2066	0.01	0.01	1.00	0.01	0.01
2068	0.00	0.00	1.00	0.00	0.00
2070	0.00	0.00	1.00	0.00	0.00
2072	0.00	0.00	1.00	0.00	0.00
2074	0.01	0.01	1.00	0.02	0.02
2076	0.00	0.00	1.00	0.00	0.00
2078	0.02	0.08	0.92	0.02	0.02
2080	0.00	0.00	1.00	0.00	0.00
2082	0.00	0.00	1.00	0.00	0.00
2084	0.00	0.00	1.00	0.00	0.00
2086	0.00	0.00	1.00	0.00	0.00
2088	0.00	0.00	1.00	0.00	0.00
2090	0.00	0.00	1.00	0.00	0.00
2092	0.00	0.00	1.00	0.00	0.00
2094	0.00	0.00	1.00	0.00	0.00
2096	0.00	0.00	1.00	0.00	0.00
2098	0.00	0.00	1.00	0.00	0.00
2100	0.00	0.00	1.00	0.00	0.00
2102	0.00	0.00	1.00	0.00	0.00
2104	0.02	0.02	1.00	0.03	0.03
2108	0.00	0.00	1.00	0.00	0.00
2110	0.00	0.00	1.00	0.00	0.00
2112	0.00	0.00	1.00	0.00	0.00
2114	0.00	0.00	1.00	0.00	0.00
2116	0.00	0.00	1.00	0.00	0.00
2118	0.00	0.00	1.00	0.00	0.00
2120	0.00	0.00	1.00	0.00	0.00
2122	0.00	0.00	1.00	0.00	0.00
2124	0.00	0.00	1.00	0.00	0.00
2126	0.00	0.00	1.00	0.00	0.00
2128	0.00	0.00	1.00	0.00	0.00
2130	0.00	0.00	1.00	0.00	0.00
2132	0.00	0.00	1.00	0.00	0.00
2134	0.00	0.00	1.00	0.00	0.00
2136	0.00	0.00	1.00	0.00	0.00
2138	0.00	0.00	1.00	0.00	0.00
2140	0.00	0.00	1.00	0.00	0.00
2142	0.00	0.00	1.00	0.00	0.00
2144	0.00	0.00	1.00	0.00	0.00
2146	0.00	0.00	1.00	0.00	0.00
2148	0.00	0.00	1.00	0.00	0.00
2150	0.00	0.00	1.00	0.00	0.00
2152	0.00	0.00	1.00	0.00	0.00

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ / SO ₄
2154	0.00	0.00	1.00	0.00	. 0.00
2156	0.00	0.00	1.00	0.00	0.00
2158	0.00	0.00	1.00	0.00	0.00
2160	0.00	0.00	1.00	0.00	0.00
2162	0.00	0.00	1.00	0.00	0.00
2164	0.00	0.00	1.00	0.00	0.00
2166	0.00	0.00	1.00	0.00	0.00
2168	0.00	0.00	1.00	0.00	0.00
2170	0.00	0.00	1.00	0.00	0.00
2172	0.00	0.00	1.00	0.00	0.00
2174	0.00	0.00	1.00	0.00	0.00
2176	0.00	0.00	1.00	0.00	0.00
2178	0.00	0.00	1.00	0.00	0.00
2180	0.00	0.00	1.00	0.00	0.00
2182	0.00	0.00	1.00	0.00	0.00
2184	0.00	0.00	1.00	0.00	0.00
2186	0.00	0.00	1.00	0.00	0.00
2188	0.00	0.00	1.00	0.00	0.00
2190	0.00	0.00	1.00	0.00	0.00
2192	0.00	0.00	1.00	0.00	0.00
2194	0.00	0.00	1.00	0.00	0.00
2196	0.00	0.00	1.00	0.00	0.00
2198	0.00	0.00	1.00	0.00	0.00
2200	0.05	0.00	1.05	0.00	0.00
2202	0.00	0.00	1.00	0.00	0.00
2204	0.00	0.00	1.00	0.00	0.00
2206	0.00	0.00	1.00	0.00	0.00
2208	0.00	0.00	1.00	0.00	0.00
2210	0.00	0.00	1.00	0.00	0.00
2212	0.00	0.00	1.00	0.00	0.00
2214	0.00	0.00	1.00	0.00	0.00
2216	0.00	0.00	1.00	0.00	0.00
2218	0.00	0.00	1.00	0.00	0.00
2220	0.05	0.00	1.05	0.00	0.00
2222	0.00	0.00	1.00	0.00	0.00
2224	0.00	0.00	1.00	0.00	0.00
2226	0.00	0.00	1.00	0.00	0.00
2228	0.00	0.00	1.00	0.00	0.00
2230	0.00	0.00	1.00	0.00	0.00
2232	0.00	0.00	1.00	0.00	0.00
2234	0.00	0.00	1.00	0.00	0.00
2236	0.00	0.00	1.00	0.00	0.00
2238	0.00	0.00	1.00	0.00	0.00
2240	0.00	0.00	1.00	0.00	0.00
2242	0.00	0.00	1.00	0.00	0.00
2244	0.00	0.00	1.00	0.00	0.00
2246	0.00	0.00	1.00	0.00	0.00
2250	0.00	0.00	1.00	0.00	0.00
2252	0.00	0.00	1.00	0.00	0.00
2254	0.00	0.00	1.00	0.00	0.00
2256	0.00	0.00	1.00	0.00	0.00
2258	0.00	0.00	1.00	0.00	0.00
2260	0.00	0.00	1.00	0.00	0.00
2262	0.00	0.00	1.00	0.00	0.00
2264	0.00	0.00	1.00	0.00	0.00
2266	0.00	0.00	1.00	0.00	0.00
2268	0.00	0.00	1.00	0.00	0.00

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
2270	0.00	0.00	1.00	0.00	. 0.00
2272	0.00	0.00	1.00	0.00	0.00
2274	0.00	0.00	1.00	0.00	0.00
2276	0.00	0.00	1.00	0.00	0.00
2280	0.00	0.00	1.00	0.00	0.00
2282	0.00	0.00	1.00	0.00	0.00
2284	0.00	0.00	1.00	0.00	0.00
2286	0.00	0.00	1.00	0.00	0.00
2288	0.00	0.00	1.00	0.00	0.00
2290	0.00	0.00	1.00	0.00	0.00
2292	0.00	0.00	1.00	0.00	0.00
2294	0.00	0.00	1.00	0.00	0.00
2296	0.00	0.00	1.00	0.00	0.00

Stratigraphic Variation in Major Ion Ratios of North Core at North Ingebright Lake

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
10	0.00	0.00	1.00	0.00	0.00
12	0.26	0.50	0.76	0.30	0.40
14	0.00	0.00	1.00	0.00	0.00
16	0.00	0.00	1.00	0.00	0.00
18	0.00	0.00	1.00	0.05	0.05
20	0.00	0.00	1.00	0.00	0.00
22	1.38	1.38	1.00	3.00	1.00
24	0.00	0.00	1.00	0.00	0.00
26	0.00	0.00	1.00	0.13	0.13
28	0.83	0.83	1.00	1.65	1.65
30	1.00	1.00	1.00	1.00	1.20
32	1.22	1.22	1.00	2.44	2.44
34	0.00	0.00	1.00	0.00	0.00
36	0.00	0.00	1.00	0.00	0.00
38	2.50	2.50	0.00	5.00	1.00
40	1.00	1.00	1.00	1.00	1.00
42	0.00	0.00	1.00	0.00	0.00
44	0.00	0.00	1.00	0.00	0.00
46	0.00	0.00	1.00	0.14	0.14
48	0.00	0.00	1.00	0.22	0.22
50	0.73	0.00	1.00	0.73	0.73
52	0.00	0.00	1.00	0.00	0.00
54	0.00	0.00	1.00	0.67	0.67
56	0.01	0.01	1.00	0.03	0.03
58	0.00	0.00	1.00	0.01	0.01
60	0.00	0.00	1.00	0.00	0.00
62	0.00	0.00	1.00	0.00	0.00
64	0.00	0.00	0.00	0.00	1.00
66	0.00	0.00	1.00	0.00	0.00
68	0.00	0.00	1.00	0.00	0.00
70	0.00	0.00	1.00	0.00	0.00
72	0.00	0.00	1.00	0.00	0.00
74	0.00	0.00	1.00	0.00	0.00
76	0.00	0.00	1.00	0.00	0.00
78	0.00	0.00	1.00	0.00	0.00
80	0.00	0.00	1.00	0.00	0.00
82	0.00	0.00	1.00	0.00	0.00
84	0.00	0.00	1.00	0.00	0.00
86	0.00	0.00	1.00	0.00	0.00
88	0.00	0.00	1.00	0.00	0.00
90	0.00	0.00	1.00	0.00	0.00
92	0.00	0.00	1.00	0.00	0.00
94	0.00	0.00	1.00	0.00	0.00
96	0.00	0.00	1.00	0.00	0.00
98	0.00	0.00	1.00	0.00	0.00
100	0.00	0.00	1.00	0.00	0.00
102	0.00	0.00	1.00	0.00	0.00
104	0.00	0.00	1.00	0.00	0.00
106	0.00	0.00	1.00	0.00	0.00
108	0.00	0.00	1.00	0.00	0.00
110	0.00	0.00	1.00	0.00	0.00
112	0.00	0.00	1.00	0.00	0.00
114	0.00	0.00	1.00	0.00	0.00
116	0.00	0.00	1.00	0.00	0.00
118	0.00	0.00	1.00	0.00	0.00
120	0.00	0.00	1.00	0.00	0.00
122	0.00	0.00	1.00	0.00	0.00
124	0.00	0.00	1.00	0.00	0.00

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Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
126	0.00	0.00	1.00	0.00	0.00
128	0.00	0.00	1.00	0.00	0.00
130	0.05	0.00	1.05	0.00	0.00
132	0.00	0.00	1.00	0.00	0.00
134	0.03	0.00	1.00	0.03	0.03
136	0.00	0.00	1.00	0.00	0.00
138	0.00	0.00	1.00	0.00	0.00
140	0.00	0.00	1.00	0.00	0.00
142	0.00	0.00	1.00	0.00	0.00
144	0.00	0.00	1.00	0.00	0.00
146	0.00	0.00	1.00	0.00	0.00
148	0.00	0.00	1.00	0.00	0.00
150	0.00	0.00	1.00	0.00	0.00
157	0.00	0.00	1.00	0.00	0.00
154	0.00	0.00	1.00	0.00	0.00
156	0.04	0.00	1.00	0.00	0.00
158	0.04	0.00	1.04	0.00	0.00
158	0.00	0.00	1.00	0.00	0.00
162	0.00	0.00	1.00	0.00	0.00
102	0.00	0.00	1.00	0.00	0.00
104	0.00	0.00	1.00	0.00	0.00
100	0.00	0.00	1.00	0.00	0.00
108	0.00	0.00	1.00	0.00	0.00
170	0.00	0.00	1.00	0.00	0.00
172	0.00	0.00	1.00	0.00	0.00
174	0.00	0.00	1.00	0.00	0.00
176	0.00	0.00	1.00	0.00	0.00
178	0.00	0.00	1.00	0.00	0.00
180	0.04	0.00	1.04	0.00	0.00
182	0.01	0.01	1.00	0.03	0.03
184	0.00	0.00	1.00	0.00	0.00
186	0.00	0.00	1.00	0.00	0.00
188	0.00	0.00	1.00	0.00	0.00
190	0.03	0.00	1.00	0.03	0.03
192	0.00	0.00	1.00	0.00	0.00
194	0.00	0.00	1.00	0.00	0.00
196	0.00	0.00	1.00	0.00	0.00
198	0.00	0.00	1.00	0.00	0.00
200	0.14	0.00	1.14	0.02	0.01
202	0.00	0.00	1.00	0.00	0.00
204	0.00	0.00	1.00	0.00	0.00
226	0.00	0.00	1.00	0.00	0.00
228	0.00	0.00	1.00	0.00	0.00
230	0.00	0.00	1.00	0.00	0.00
232	0.00	0.00	1.00	0.00	0.00
234	0.00	0.00	1.00	0.00	0.00
236	0.08	0.00	1.08	0.00	0.00
238	0.00	0.00	1.00	0.00	0.00
240	0.00	0.00	1.00	0.00	0.00
242	0.00	0.00	1.00	0.00	0.00
244	0.11	0.00	1.11	0.00	0.00
246	0.12	0.00	1.12	0.00	0.00
248	0.08	0.00	1.08	0.00	0.00
250	0.01	0.01	1.00	0.03	0.00
250	0.10	0.00	1 10	0.03	0.05
254	0.00	0.00	1.00	0.00	0.00
254	0.00	0.00	1.00	0.00	0.00
250	0.00	0.00	1.00	0.00	0.00
200	0.00	0.00	1.00	0.00	0.00
200	0.08	0.00	1.00	0.00	0.00
202	0.18	0.00	1.18	0.00	0.00

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
264	0.11	0.00	1.11	0.00	0.00
266	0.11	0.00	1.10	0.02	0.01
268	0.14	0.00	1.14	0.00	0.00
270	0.00	0.00	1.00	0.00	0.00
272	0.02	0.00	1.00	0.02	0.02
274	0.00	0.00	1.00	0.00	0.00
276	0.00	55.26	1.00	55.26	55.26
278	0.06	0.02	1.00	0.08	0.08
280	0.13	0.00	1.13	0.00	0.00
282	0.27	0.10	1.25	0.13	0.10
284	0.18	0.00	1.18	0.00	0.00
286	0.01	0.01	1.00	0.03	0.03
288	0.08	0.00	1.08	0.00	0.00
290	0.14	0.11	1.11	0.14	0.12
292	0.13	0.03	1.10	0.05	0.05
294	0.02	0.02	1.00	0.10	0.10
296	0.12	0.00	1.12	0.00	0.00
298	0.08	0.06	1.08	0.06	0.06
300	0.00	0.00	1.00	0.00	0.00
302	0.10	0.02	1.08	0.04	0.04
304	0.02	0.02	1.00	0.04	0.04
306	0.01	0.01	1.00	0.03	0.03
308	0.08	0.00	1.08	0.00	0.00
310	0.06	0.00	1.06	0.00	0.00
312	0.00	0.00	1.00	0.00	0.00
314	0.08	0.02	1.06	0.03	0.03
316	0.12	0.02	1.10	0.03	0.03
318	0.02	0.02	1.00	0.03	0.03
320	0.02	0.02	1.00	0.06	0.06
322	0.08	0.02	1.06	0.04	0.04
324	0.02	0.02	1.00	0.04	0.04
326	0.06	0.06	1.00	0.11	0.11
328	0.02	0.02	1.00	0.07	0.07
330	0.02	0.11	1.00	0.13	0.13
332	0.00	0.09	0.91	0.00	0.00
334	0.07	0.02	1.05	0.03	0.03
336	0.02	0.02	1.00	0.03	0.03
338	0.02	0.02	1.00	0.04	0.04
340	0.10	0.02	1.08	0.03	0.03
342	0.02	0.02	1.00	0.03	0.03
344	0.02	0.08	1.00	0.10	0.10
346	0.02	0.11	1.00	0.16	0.16
348	0.09	0.02	1.06	0.04	0.04
350	0.00	0.00	1.00	0.00	0.00
352	0.00	0.00	1.00	0.00	0.00
354	0.08	0.02	1.07	0.03	0.03
356	0.00	0.00	1.00	0.00	0.00
358	0.00	1.00	0.00	0.00	0.00
360	0.02	0.02	1.00	0.03	0.03
362	0.00	0.00	1.00	0.00	0.00
364	0.01	0.01	1.00	0.03	0.03
366	0.00	0.00	1.00	0.00	0.00
368	0.05	0.00	1.05	0.00	0.00
370	0.00	0.06	0.94	0.00	0.00
372	0.01	0.01	1.00	0.03	0.03
374	0.06	0.02	1.05	0.03	0.03
376	0.00	0.00	1.00	0.00	0.00
378	0.00	0.00	1.00	0.00	0.00
380	0.03	0.03	1.00	0.06	0.06

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄
382	0.06	0.00	1.06	0.00	0.00
384	0.08	0.00	1.08	0.00	0.00
386	0.00	0.00	1.00	0.00	0.00
388	0.00	0.00	1.00	0.00	0.00
390	0.00	0.00	1.00	0.00	0.00
392	0.02	0.02	1.00	0.04	0.04
394	0.00	0.00	1.00	0.00	0.00
396	0.13	0.11	1.11	0.13	0.12
398	0.10	0.02	1.08	0.03	0.03
400	0.19	0.08	1.17	0.09	0.08
404	0.10	0.00	1.10	0.00	0.00
406	0.01	0.01	1.00	0.03	0.03
408	0.11	0.02	1.07	0.07	0.06
410	0.00	0.06	1.00	0.06	0.06
412	0.12	0.07	1.10	0.10	0.09
414	0.15	0.09	1.09	0.18	0.17
416	0.02	0.02	1.00	0.04	0.04
500	0.03	0.14	1.00	0.16	0.16
502	0.08	0.00	1.08	0.00	0.00
504	0.00	0.13	0.88	0.03	0.04
508	0.12	0.05	1.07	0.10	0.09
510	0.05	0.05	1.00	0.11	0.11
512	0.45	0.05	1.41	0.09	0.06
514	0.16	0.03	1.13	0.05	0.05
516	0.41	0.05	1.36	0.09	0.07
518	0.43	0.24	1.38	0.29	0.21
520	0.18	0.03	1.15	0.05	0.04
522	0.15	0.13	1.13	0.15	0.13
524	0.13	0.15	1.10	0.18	0.16
526	0.30	0.30	1.22	0.37	0.30
528	0.03	0.03	1.00	0.05	0.05
530	0.02	0.02	1.00	0.05	0.05
532	0.10	0.10	1.10	0.10	0.09
534	0.13	0.02	1.11	0.04	0.04
536	0.28	0.03	1.25	0.06	0.04
538	0.25	0.14	1.22	0.17	0.14
540	0.64	0.24	1.60	0.28	0.18
542	0.27	0.18	1.24	0.21	0.17
544	0.47	0.20	1.43	0.23	0.16
546	0.50	0.33	1.42	0.42	0.29
550	0.23	0.26	1.16	0.32	0.28
552	0.16	0.16	1.14	0.19	0.17
554	0.18	0.02	1.10	0.04	0.03
220	0.47	0.03	1.43	0.07	0.05
558	0.33	0.18	1.30	0.21	0.16
560	0.15	0.02	1.12	0.05	0.04
562	0.20	0.00	1.41	0.12	0.10
504	0.23	0.03	1.20	0.05	0.04
200	0.19	0.14	1,10	0.16	0.14
208	0.43	0.11	1.40	0.14	0.10
570	0.27	0.07	1.24	0.10	0.08
512	0.02	0.02	1.00	0.04	0.04
574	0.17	0.11	1.15	0.15	0.11
2/0 570	V.22	0.12	1.20	0.15	0.12
J18 590	0.10	0.13	1.10	0.15	0.14
08C	0.32	0.03	1.27	0.05	0.04
28Z	0.30	0.03	1.20	0.05	0.04
584	0.25	0.10	1.24	0.12	0.10
280	0.40	0.00	1.40	0.02	0.02

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO ₃ /SO ₄	
588	0.40	0.14	1.37	0.17	0.13	
590	0.24	0.02	1.21	0.05	0.04	
592	0.29	0.02	1.27	0.05	0.04	
594	0.14	0.02	1.12	0.06	0.05	
596	0.29	0.02	1.26	0.05	0.04	
598	0.43	0.14	1.37	0.09	0.06	
600	0.20	0.03	1.18	0.05	0.04	
602	0.19	0.14	1.16	0.16	0.14	
604	0.15	0.03	1.13	0.05	0.04	
606	0.24	0.12	1.22	0.15	0.12	
608	0.27	0.02	1.24	0.05	0.04	
610	0.18	0.09	1.16	0.11	0.10	
612	0.17	0.02	1.16	0.03	0.03	
614	0.02	0.02	1.00	0.03	0.03	
616	0.42	0.00	1.36	0.08	0.06	
618	0.37	0.00	1.33	0.04	0.03	
620	0.32	0.00	1.23	0.13	0.10	
622	0.35	0.00	1.25	0.10	0.08	
624	0.22	0.00	1.22	0.00	0.00	
626	0.79	0.03	1.69	0.17	0.10	
628	0.35	0.00	1.20	0.14	0.12	
630	0.17	0.15	0.96	0.07	0.07	
632	0.25	0.00	1.21	0.08	0.06	
634	0.19	0.11	1.04	0.06	0.05	
636	0.05	0.00	1.03	0.02	0.01	
638	0.05	0.02	1.00	0.06	0.06	
640	0.17	0.02	1.10	0.08	0.08	
642	0.05	0.00	1.00	0.10	0.10	
644	0.05	0.24	0.90	0.21	0.23	
646	0.36	0.02	1.33	0.05	0.04	
648	0.15	0.02	1.10	0.07	0.06	
650	0.11	0.02	1.09	0.07	0.06	
652	0.14	0.00	1.10	0.03	0.03	
654	0.10	0.00	1.10	0.00	0.00	
656	0.10	0.00	1.10	0.00	0.00	
658	0.06	0.00	1.06	0.00	0.00	
660	0.00	0.00	1.00	0.00	0.00	
662	1.38	0.00	2.38	0.00	0.00	
664	0.26	0.00	1.26	0.00	0.00	
666	0.97	0.03	1.94	0.06	0.03	
668	0.41	0.00	1.41	0.00	0.00	
670	0.60	0.03	1.58	0.05	0.03	
672	2.28	0.11	3.22	0.17	0.05	
674	0.54	0.00	1.54	0.00	0.00	
676	0.55	0.02	1.52	0.05	0.03	
678	1.12	0.04	2.04	0.12	0.06	
680	0.29	0.02	1.27	0.05	0.04	
682	1.95	0.05	2.90	0.10	0.03	
684	0.56	0.03	1.53	0.06	0.04	
686	0.24	0.02	1.22	0.04	0.04	
824	0.17	0.00	1.17	0.00	0.00	
826	0.29	0.00	1.29	0.00	0.00	
828	0.35	0.00	1.35	0.00	0.00	
830	0.12	0.00	1.12	0.00	0.00	
832	0.67	0.00	1.67	0.00	0.00	
834	0.10	0.00	1.10	0.00	0.00	
836	0.18	0.03	1.15	0.05	0.04	
838	0.00	0.00	1.00	0.00	0.00	
840	0.02	0.02	1.00	0.05	0.05	

Depth (cm)	Ca/Na	Mg/Na	SO₄/Na	CO ₃ /Na	CO₃/SO ₄	
842	0.00	0.03	1.00	0.03		
844	0.36	0.25	1.29	0.32	0.25	
846	0.73	0.36	1.64	0.45	0.28	
848	0.06	0.00	1.06	0.00	0.00	
850	0.10	0.02	1.07	0.07	0.07	
852	0.05	0.02	1.00	0.06	0.06	
854	3.80	0.30	4.70	0.40	0.09	
856	0.11	0.02	1.09	0.04	0.04	
858	0.02	0.02	1.00	0.04	0.04	
860	0.35	0.02	1.33	0.04	0.03	
862	0.10	0.02	1.08	0.03	0.03	
864	0.28	0.13	1.26	0.15	0.12	
866	0.34	0.19	1.31	0.22	0.17	
868	1.20	0.30	2.10	0.40	0.19	
870	15.00	0.00	16.00	0.33	0.02	
872	21.50	2.50	22.00	3.00	0.14	
874	10.00	0.00	11.00	0.00	0.00	
876	13.00	0.00	13.67	0.33	0.02	
878	25.00	0.00	26.00	0.50	0.02	
880	47.00	0.00	48.00	0.00	0.00	
882	13.00	0.33	13.33	1.00	0.08	
884	4.50	0.00	5.50	0.00	0.00	
886	13.00	0.00	13.75	0.25	0.02	
888	16.00	0.33	16.33	1.00	0.06	
890	53.00	0.00	54.00	0.00	0.00	
892	23.50	0.00	24.00	0.50	0.02	
894	55.00	0.00	55.00	1.00	0.02	
896	15.67	0.00	16.33	0.33	0.02	

Appendix E Core Descriptions

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Core Descriptions (North Core, North Ingebright Lake)

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Depth (cm)			Description
0	-	12	missing core
12	-	37	mud
37	-	55	muddy salt
55	-	75	muddy salt, small crystal aggregate and mud materials occurs between salt crystals
75	-	153	muddy salt, salt crystal size: 2-5 mm.
153	-	228	muddy salt, crystal size becomes bigger, 3-6 mm.
228	-	306	mud and salt, mud materials: 30-40%, intrasedimentary crystals are common
306	-	381	massive salt, mud material decreases, salt crystal size: 0.5-2 cm.
381	-	459	massive salt
459	-	534	massive salt, colourless, large salt crystal size: 1-2 cm,
534	-	612	massive salt
61 2	-	687	massive salt
687	-	765	clear salt crystals containing about 10% mud or organic debris inclusions
765	-	790	missing core
790	-	840	massive salt
840	-	918	massive salt
918	-	993	muddy salt
993	-	1071	massive salt
1071	-	1146	massive salt
1146	-	1224	very clear salt, crystal size: 2-5 cm
1224	-	1324	missing core
1324	-	1374	dirty salt
1374	-	1449	massive salt
1449	-	1527	massive salt
1527	-	1602	massive salt
1602	-	1680	clear salt, compacted, recrystallized
1680	-	1730	missing core
1730	-	1805	massive salt
1805	-	1837	massive salt
1837	-	1912	missing core
1912	-	1990	colourless, structureless, very-well consolidated salt
1990	-	2065	clear salt

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2065	-	2143	clear salt
2143	-	2218	clear salt
2218	-	2296	clear salt, odour smell (H_2S), redox environment

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Core Descriptions (South Core, North Ingebright Lake)

Depth	(cm)		Description
0	-	10	missing core
10	-	46	black green mud (fine-grained sands, silts and clays), intrasedimentary crystals of mirabilite, thenardite, and gypsum are abundant, grain size: $0.1 - 0.5$ cm, no lamination
46	-	75	muddy salt, grey colour, massive structure, mud materials: 40 - 50%
75	-	150	smoky colour, compacted, well-consolidated salt, long axis of black needle-like sediments or organic debris is vertical.
150	-	226	broken salt
226	-	302	dirty salt, massive crystalline aggregates.
302	-	377	smoke colour, compacted, muddy salt, mud materials: 5 - 10%.
377	-	417	muddy salt
417	-	499	missing core
499	-	575	massive muddy salt
575	-	610	muddy salt
610	-	686	salt, crystal grain size: 0.5-1 cm, mud materials commonly fill space between salt crystal grains
686	-	822	missing core
822	-	896	massive salt, bottom 30 cm core consists of black green mud containing abundant intrasedimentary salt crystals. Crystals appear in the form of rounded accretionary grains or tabular crystals, grain size: 1-5 cm.
896	-	916	highly reducing, fine-grained sand, silt and clay

Appendix F Statistical Analyses

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Appendix F Statistical Analyses

The sediment of North Ingebright Lake offer unusual opportunities for selected geochemical and mineralogical studies. It contains a complete and well-preserved Holocene sediment record. Twenty one major and minor minerals are reported, together with chemical properties of sediments for 1310 subsamples from two cores. These data constitute an excellent data matrix for statistical analysis. When the set of data is large and the underlying causal structure is obscure, multivariate statistical techniques can be an important aid in revealing simple patterns in complex information. In the following section multivariate techniques, sequentially employing correlation, Q-mode and R-mode analyses, were applied to the mineralogical and geochemical data in order to examine possible interrelationships between various variables, and to define statistically recognizable controlling factors.

Correlation Analysis

Pearson's product-moment coefficient of linear correlation was used to assess the linear relationship between variables. The correlation coefficient γ is given by

$$\gamma = \sum (x - \bar{x})(y - \bar{y}) / \langle \sum (x - \bar{x})^2 \sum (y - \bar{y})^2 \rangle^{1/2}$$

where \bar{x} and \bar{y} are the mean values of variable x and y. This formula is the ratio of how much x and y vary together about their means to total variation of x and y.

Values of γ can vary between +1 and -1. γ = +1 represents a perfect linear relationship between x and y. γ = -1 expresses a perfect negative correlation between x and y. γ = 0 means that x and y are uncorrelated variables (Klovan, 1975).

Simple linear correlation coefficients of mineralogical and geochemical data were calculated for all possible variable pairs. Table G-1 is a matrix of the correlation coefficients that were obtained using SAS computer program. As expected, detrital components are all positively correlated with organic matter content, whereas they are strongly inversely proportional to endogenic components. Significantly high coefficients were obtained for quartz to organic matter, 0.552; quartz to high Mg calcite, 0.553; organic matter to protodolomite, 0.477; and thenardite to depth, 0.439. Strongly negative coefficients were found for organic matter to thenardite, -0.632; total clay minerals to thenardite, -0.596; and gypsum to thenardite, -0.525.

Among the detrital components, quartz, k-feldspar, clay minerals, calcite and dolomite all show positive correlation with each other. Only plagioclase is negatively correlated with quartz, k-feldspar, and total clay minerals. Endogenic mineralogical components determined in this study show striking correlation with one another. Thenardite shows moderate to strong negative association with all endogenic minerals: high Mg calcite, protodolomite, aragonite, magnesite, gypsum, anhydrite, bloedite and halite. High Mg calcite, protodolomite, aragonite and magnesite are inversely related to bloedite, gypsum and anhydrite, while halite varies positively with carbonate and sulphate minerals. These relationships can be well explained

by evaporative evolution of natural waters.

Cluster Analysis

The objective of cluster analysis is to find new, hypothetical "sample" whose compositions are linear combinations of those of the original samples (Imbrie and Purdy, 1962). In this study, cluster analysis was employed to help define groups (units) of samples by considering simultaneously all the variables that have been measured. After the unit classifications, factor analysis was carried out for each of individual units because factor analysis requires a homogeneous population to be meaningful.

A vast number of clustering methods have been developed in past decades, such as correlation coefficient, coefficient of proportional similarity, and distance coefficient methods (Klovan, 1975). The simulation study indicates that the clustering method with the best overall performance has been either average linkage or Ward's minimum variance method (SAS User's Guide, 1990). Therefore, average linkage method was chosen to group samples from North Ingebright Lake.

The similarities between all possible pairs of samples were calculated and shown both in table and histogram. The table and histogram contain all the information concerning the interrelations between the samples. However, the table and histogram are too large to be attached in this document. At the 0.2 distance coefficient level seven clusters of samples are readily identified. Each of these clusters approximately correspond to one lithostratigraphic unit.

Factor Analysis

Factor analysis is designed to represent complex relationships between a large number of variables by simple relationships amongst fewer hypothetical variables. These new variables are linearly related to the original ones by rotation in space and should contain the same amount of information (Jöreskog et al., 1976). Various mathematical procedures have been proposed to perform factor analysis. Although factor analysis was initially used by psychologists, its significance in earth science was soon realized by geologists. Imbries and Purdy (1962) successfully used factor analysis to classify the modern carbonate sediments of the Great Bahama Bank. Klovan (1966) employed factor analysis to define sedimentary environments from grain size data. Hitchon et al. (1971) and Reed, Hitchon, and Levinson (1972) used factor analysis to reveal the controls on water chemistry. More recently Last (1992) applied factor analysis to classify salt lakes in western Canada and discover the controls on water chemistry of salt lakes. Thirteen major water chemistry types were clearly identified. Groundwater composition, climate, and the elevation of the lake within the drainage system were revealed as major factors in controlling brine chemistry and salinity in salt lakes.

There are three major methods available for factor analysis: principle factor method, varimax method, and promax method. In the principle factor method, the total variation in the data set is represented by a small number of geological factors, each manifested in a degree of correlation (elongation) in the data scatter. Each successive factor accounts for a lesser amount of the variance of variable than the preceding one. Factors accounting for small quantities of variance therefore can be
identified and neglected without losing useful information. In the varimax method, orthogonal rotation of the axis system is attempted to account for as much of the variance as possible. The promax method allows the axes to rotate out of perpendicular, in which the reference vectors of the scheme are taken as unit vectors passed through the positions of the m real sample vectors occupying extreme positions in the configuration (Imbrie and Van Andel, 1964; Sas User's Guide, 1990).

Initially the data set from North Ingebright Lake was processed using all these three methods. However, factors extracted by the varimax and promax method are usually mathematically abstractions difficult to interpret geologically. They are over-simply focused on the individual variable. In most cases each variable only has high loadings on one factor. This violates with the natural process: any geological product is a result of multi-factor interplays. Therefore, in this research, only factors extracted by the principle method are used for the further geological interpretation.

The results of the R-mode factor analysis of the Units 2, 3, 4, 5, 6, 7 and 8 samples are shown in Tables G-2, 3, 4, 5, 6, 7 and 8. Eight to eleven factors were extracted from the different units. They account for over 91-98 per cent of the sample variance. Factor interpretations for each units are similar. Therefore, only interpretation for unit 2 is given below.

Ten factors explain over 95% of the total variance of the data set of the unit 2. Factor 1 accounts for 25.16% of the total data variability, with high positive loadings on quartz, plagioclase, total clay minerals and dolomite. This factor is interpreted as representing the detrital products, which were brought to lake basin by sheetflow or eolian activities. Factors 2 and 3, which consists of bloedite, thenardite, aragonite, and magnesite, account for 25.08% of the total data variability. These two factors essentially represent endogenic chemical sedimentation components. Factors 4 and 5, accounting for 17.34% of the total data variability, are dominated by high positive loading on organic matter and total carbonate contents. This factor is interpreted as a lake production factor. Factors 6, 7, 8, 9 and 10 are minor and their eigen values are less than 0.5. They may be picking up just random effects.

	Depth	Organic	Carbonate	Quartz	K-	Plagioclase	Calcite
		Matter	Content		feldspar		
Depth	1.0000	-0.4120	-0.1089	-0.3685	-0.1683	-0.1758	-0_0326
Organic Matter	-0.4120	1.0000	0.1340	0.5521	0.2930	0.0260	0_0087
Carbonate Content	-0.1089	0.1340	1.0000	0.1111	0.0526	0.1254	0_0060
Quartz	-0.3685	0.5521	0.1111	1.0000	0.3950	-0.0155	0_0082
K-feldspar	-0.1683	0.2930	0.0526	0.3950	1.0000	-0.0047	-0_0009
Plagioclase	-0.1758	0.0260	0.1254	-0.0155	-0.0047	1.0000	0_1105
Calcite	-0.0326	0.0087	0.0060	0.0082	-0.0009	0.1105	1_0000
High Mg Calcite	-0.1148	0.1138	0.0170	0.5533	0.0565	-0.0289	0_0045
Protodolomite	-0.2387	0.4772	0.0857	0.4440	0.3083	-0.0185	-0_0049
Dolomite	-0.1462	0.1597	0.0564	0.1672	0.0048	0.0308	0_0092
Bloedite	-0.0952	0.2663	0.0621	0.1144	0.1409	-0.0240	0_0186
Clay Mineral	-0.1691	0.3613	0.0108	0.0636	0.0159	-0.0780	-0_0267
Magnesite	0.0302	0.1440	-0.0169	-0.0109	-0.0135	-0.0493	-0_0168
Thenardite	0.4394	-0.6319	-0.1294	-0.4613	-0.2766	-0.3169	-0_0501
Aragonite	0.0330	0.0218	-0.0362	-0.0187	-0.0232	-0.0685	-0_0127
Anhydrite	-0.1104	0.2518	0.0360	0.2073	0.0622	-0.0100	-0_0037
Gypsum	-0.1995	0.2084	0.0131	0.0116	0.0140	-0.0621	0_0033
Halite	0.0227	0.0156	-0.0084	0.0098	0.0358	-0.0256	-0_0040
Ma in High Ma Calcite	-0.2103	0.2186	0.0394	0.2420	0.0506	-0.0562	0_0265
Ca in Protodolomite	-0.3406	0.4020	0.0473	0.1580	0.0623	0.0325	0_0274
Ca in Dolomite	-0.2423	0.2441	0.0474	0.1439	0.0148	0.0049	0_0887

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Table F-1 Pearson Correlation Coefficient Matrix of Mineralogical and Geochemical Variables from North Ingebright lake

	High Mg Calcite	Protodolomite	Dolomite	Bloedite	Clay Mineral	Magnesite	Thenardite
Depth	-0.1148	-0.2387	-0.1462	-0.0952	-0.1691	0.0302	0.4394
Organic Matter	0.1138	0.4772	0.1597	0.2663	0.3613	0.1440	-0.6319
Carbonate Content	0.0170	0.0857	0.0564	0.0621	0.0108	-0.0169	-0.1294
Quartz	0.5533	0.4440	0.1672	0.1144	0.0636	-0.0109	-0.4613
K-feldspar	0.0565	0.3083	0.0048	0.1409	0.0159	-0.0135	-0.2766
Plagioclase	-0.0289	-0.0185	0.0308	-0.0240	-0.0780	-0.0493	-0.3169
Calcite	0.0045	-0.0049	0.0092	0.0186	-0.0267	-0.0168	-0.0501
High Mg Calcite	1.0000	0.0592	0.0565	-0.0006	0.0138	-0.0146	-0.1966
Protodolomite	0.0592	1.0000	-0.0157	0.1247	0.0843	0.0579	-0.4186
Dolomite	0.0565	-0.0157	1.0000	0.0310	0.0578	0.0160	-0.2676
Bloedite	-0.0006	0.1247	0.0310	1.0000	0.0330	-0.0400	-0.2585
Clay Mineral	0.0138	0.0843	0.0578	0.0330	1.0000	0.3719	-0.5963
Magnesite	-0.0146	0.0579	0.0160	-0.0400	0.3719	1.0000	-0.2972
Thenardite	-0.1966	-0.4186	-0.2676	-0.2585	-0.5963	-0.2972	1.0000
Aragonite	-0.0212	0.0021	-0.0202	-0.0182	0.1131	0.1647	-0.0487
Anhydrite	-0.0037	0.1936	0.0153	0.1671	0.0577	-0.0018	-0.1705
Gypsum	0.0164	-0.0120	0.0365	0.0147	0.1467	0.1414	-0.5247
Halite	-0.0092	-0.0081	0.0011	-0.0094	0.0180	0.0395	-0.0340
Mg in High Mg Calcite	0.4391	0.0601	0.1451	0.0699	0.1219	-0.0218	-0.1909
Ca in Protodolomite	0.0014	0.2911	0.0155	0.0259	0.4813	0.3364	-0.5006
Ca in Dolomite	0.1446	-0.0151	0.3107	0.0305	0.3530	0.1643	-0.4030

Table F-1 Pearson Correlation Coefficient Matrix of Mineralogical and Geochemical Variables from North Ingebright lake

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	Aragonite	Anhydrite	Gypsum	Halite	Ma in High Mg Calcite	Ca in Protodolomite	Ca in Dolomite
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Depth	0.0330	-0.1104	-0.1995	0.0227	-0.2103	-0.3406	-0.2423
Organic Matter	0.0218	0.2518	0.2084	0.0156	0.2186	0.4020	0.2441
Carbonate Content	-0.0362	0.0360	0.0131	-0.0084	0.0394	0.0473	0.0474
Quartz	-0.0187	0.2073	0.0116	0.0098	0.2420	0.1580	0.1439
K-feldspar	-0.0232	0.0622	0.0140	0.0358	0.0506	0.0623	0.0148
Plagioclase	-0.0685	-0.0100	-0.0621	-0.0256	-0.0562	0.0325	0.0049
Calcite	-0.0127	-0.0037	0.0033	-0.0040	0.0265	0.0274	0.0887
High Mg Calcite	-0.0212	-0.0037	0.0164	-0.0092	0.4391	0.0014	0.1446
Protodolomite	0.0021	0.1936	-0.0120	-0.0081	0.0601	0.2911	-0.0151
Dolomite	-0.0202	0.0153	0.0365	0.0011	0.1451	0.0155	0.3107
Bloedite	-0.0182	0.1671	0.0147	-0.0094	0.0699	0.0259	0.0305
Clay Mineral	0.1131	0.0577	0.1467	0.0180	0.1219	0.4813	0.3530
Magnesite	0.1647	-0.0018	0.1414	0.0395	-0.0218	0.3364	0.1643
Thenardite	-0.0487	-0.1705	-0.5247	-0.0340	-0.1909	-0.5006	-0.4030
Aragonite	1.0000	-0.0205	-0.0004	-0.0217	-0.0206	0.1448	0.0248
Anhydrite	-0.0205	1.0000	-0.0367	-0.0064	0.0205	0.0219	0.0075
Gypsum	-0.0004	-0.0367	1.0000	0.0472	0.0591	0.2501	0.2727
Halite	-0.0217	-0.0064	0.0472	1.0000	-0.0185	-0.0059	0.0359
Mg in High Mg Calcite	-0.0206	0.0205	0.0591	-0.0185	1.0000	0.0854	0.2433
Ca in Protodolomite	0.1448	0.0219	0.2501	-0.0059	0.0854	1.0000	0.2434
Ca in Dolomite	0.0248	0.0075	0.2727	0.0359	0.2433	0.2434	1.0000

Table F-1 Pearson Correlation Coefficient Matrix of Mineralogical and Geochemical Variables from North Ingebright lake

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	actor 1 F	actor 2 F	actor 3 F	actor 4	Factor 5 F	actor 6	Factor 7	Factor 8	Factor 9 F	actor 10 Col	nmunality
Organic Matter	0.15	0.06	-0.03	0.48	0.40	0.17	0.74	-0.10	-0.02	-0.05	1.00
Carbonate Content	0.17	0.13	0.09	0.44	0.64	0.11	-0.50	0.23	0.15	0.02	1.00
Quartz	0.47	-0.06	-0.55	0.27	-0.34	-0.04	0.06	0.17	0.44	0.02	0.94
K-feldspar	0.00	0.00	0.00	00.0	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Plagioclase	0.24	0.44	-0.25	-0.69	0,40	0.02	0.11	-0.01	0.13	0.01	0.98
Calcite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
High Mg Calcite	0.00	0.00	0.00	00'0	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Protodolomite	0.73	-0.22	-0.13	0.17	0.01	0.01	-0.21	-0.27	-0.30	-0.19	0.87
Dolomite	0.54	-0.35	0.32	-0.14	-0.01	-0.02	0.19	0.58	-0.22	0.18	0.99
Bloedite	0.17	0.71	0.34	0.17	-0.27	-0.04	0.01	0.20	-0.07	-0.44	1.00
Clay Mineral	0.81	-0.24	0.34	-0.01	0.03	0.01	-0.07	-0.24	0.00	0.13	0.91
Magnesite	0.71	-0.37	0.28	-0.09	-0.08	-0.01	0.05	-0.07	0.31	-0.17	0.87
Thenardite	-0.85	-0.42	0.10	0.27	-0.08	-0,02	-0.01	0.01	0.04	0.04	0.99
Aragonite	0.22	0.65	0.41	0.22	-0.22	-0.03	-0.01	-0.16	0.06	0.44	0.97
Anhydrite	00'0	0.00	0.00	0.00	00.0	0.00	00'0	0.00	0.00	0.00	0.0
Gypsum	0.40	0.19	-0.74	0.17	-0.11	-0.01	-0.06	0.08	-0.27	0.15	0.89
Halite	-0.02	0.00	0.01	-0.10	-0.19	0,98	-0.06	0.02	0.00	0.01	1.00
Percent of variance explained by factor	25.16	13.22	11.86	9.41	7.93	7.68	6.97	4.89	4.24	4.09	

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	Factor 1	Factor 2	Factor 3 I	Factor 4	Factor 5 F	actor 6 F	Factor 7	Factor 8	Factor 9 F	actor 10	Factor 11	Communality
Organic Matter	0.74	-0.11	0.09	0.07	0.10	-0.22	0.22	-0.11	0.02	-0.07	-0.09	0.71
Carbonate Content	0.08	0.12	0.29	0.28	-0.04	0.83	0.29	0.13	-0.15	0.02	-0.06	0.99
Quartz	0.76	0.07	0.18	0.02	0.29	-0.12	0.21	0.02	0,00	0.16	-0.17	0.81
K-feldspar	0.63	0.46	0.02	-0.41	-0.04	0.12	-0.07	0.02	0.17	0.07	0.06	0.84
Plagioclase	-0.18	0.26	0.28	0.19	-0.58	-0.30	0.35	0.41	0.24	0,09	0.01	1.00
Calcite	0.30	0.54	-0.55	0.05	0.03	0.09	0.12	0.00	-0.01	0.29	0.33	0.89
High Mg Calcite	0.22	-0.02	0.25	0.61	0.38	0.03	-0.34	0.00	0,48	0.14	0.04	0,99
Protodolomite	0.52	-0.50	0.04	-0.13	0.21	-0.07	0.49	-0.15	-0.02	0.10	-0.04	0,86
Dolomite	0,53	0.11	0.14	0.22	-0,24	-0.15	-0.33	-0.05	-0,46	0.45	-0,10	0.97
Bloedite	0.40	0.54	-0.51	0.10	0.02	0.04	0.09	-0.07	0.17	-0.13	-0.27	0.86
Clay Mineral	0.54	-0.47	-0.15	-0.10	-0.43	0.24	-0.29	-0.14	0.14	-0.07	-0.16	0.95
Magnesite	0.49	-0.60	-0.11	0.09	-0.12	0.04	0.06	0.03	0.04	0.02	0.48	0,88
Thenardite	-0.87	0.10	0.06	-0,09	0.36	-0.02	0.08	-0,12	-0.07	0.20	0.05	0.98
Aragonite	0.10	-0.40	-0.33	-0,16	0,30	0.03	-0.13	0,74	-0.06	0.10	-0.14	0,99
Anhydrite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Gypsum	0,56	0.35	0.14	0,23	0.18	-0.13	-0.10	0.17	-0.32	-0.50	0.18	0.98
Halite	0.34	0.26	0.51	-0.63	0.09	0.09	-0.17	0.08	0.12	0.04	0.11	0.91
Percent of variance explained by factor	25.93	13.18	7.91	7.67	7.13	6.12	5.81	5.31	4.56	4.24	3.53	i

Table F-3 Principle Factor Matrix of Sediment Data of Unit 3

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	Factor 1	Factor 2	Factor 3	Factor 4	Factor 5 F	Factor 6	Factor 7 F	Factor 8	Factor 9 F	actor 10	Factor 11	Communality
Organic Matter	· 0.52	0.15	0.18	-0.09	-0.13	0.02	-0.08	0.08	-0,50	0.11	0.26	0.98
Carbonate Content	0.11	0.08	0.82	-0.12	0.05	0.09	0.07	0.17	0.13	-0.07	0.23	0.94
Quartz	0.68	0.27	0.31	0.11	0.10	-0.06	0.03	0.03	0.16	0.02	0.14	0.74
K-feldspar	· 0.06	0.04	-0.14	-0.03	0.29	0.77	-0.36	0,00	0.33	0.10	0.17	1.00
Plagioclase	-0.31	-0.29	0.57	0.04	0.05	0.06	-0.13	-0.11	0.01	0.31	-0.52	0.98
Calcite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
High Mg Calcite	-0.09	0.14	-0.09	0.52	-0.34	-0.14	-0.32	0.49	0.24	-0.03	0.08	1.00
Protodolomite	0.70	0.20	0.01	-0.31	0.02	-0.12	0,03	0.11	0.15	-0.33	-0.18	0.83
Dolomite	0.51	-0.01	0.04	0.36	-0.28	-0.10	0.03	0,00	0.27	0.36	-0.12	0,94
Bloedite	0.02	-0,12	0.00	0.15	-0.38	0.43	0.73	-0.01	0.16	-0.15	-0.04	1.00
Clay Minera	0.71	0,00	-0.18	0.04	-0.11	0.22	-0.03	-0.04	-0,36	0.24	-0.11	1.00
Magnesite	0.64	-0,12	-0.14	0.02	0.37	-0.08	-0,03	0.09	0.21	-0.23	-0,35	, 0.85
Thenardite	-0.63	0.73	0.06	0,02	0.04	-0.03	0.00	-0,01	0.02	-0,13	-0,08	0.99
Aragonite	0.05	0,28	-0,17	-0.55	-0.15	-0.22	0.11	-0.15	0.43	0,44	0.12	1.00
Anhydrite	9.20	0,06	0.09	0.40	0.00	-0.12	-0.12	-0.79	0.11	-0.18	0.18	0.99
Gypsum	-0.04	-0,89	0.01	-0.11	0.03	-0.20	-0.04	0.07	0.17	-0.06	0.30	1.00
Halite	-0.05	0.10	-0.08	0.36	0.70	-0.15	0.43	0.15	-0.04	0.29	0.14	0.98
Percent of variance explained by factor	e 18.52	10.61	7.78	7.26	6.91	6.38	6.32	6.15	6.08	5.22	4.93	i

Table F-4 Principle Factor Matrix of Sediment Data of Unit 4

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0.57 0.15
-0.02 0.70
0.47 0.20
-0.05 -0.18
-0.39 0.06
-0.18 0.05
0.08 0.24
0.12 -0.10
0.10 0.51
0.00 0.00
0.69 -0.21
0.44 0.05
-0.25 -0.03
0.35 -0.43
0.00 0.00
0.25 0.23
0.00 0.00
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Table

Factor 1 Factor 2 Factor 3 Factor 4 Factor 5 Factor 6 Factor 7 Factor 8 Communality

Organic Matter	0.21	-0.06	0.79	0.10	-0.03	0.34	0.22	-0.39	1.00
Carbonate Content	0.05	0.01	0.39	-0.54	0.63	0.21	-0.01	0.33	1.00
Quartz	0.17	-0.28	0.40	0.58	0.23	-0.23	-0.54	0.03	1.00
K-feldspar	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00.0
Plagioclase	0.29	-0.56	-0.50	0.31	0.34	0.26	0.26	-0.04	1.00
Calcite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00.0
High Mg Calcite	0.00	0.00	0.00	0.00	00.00	0.00	00.0	0.00	00.0
Protodolomite	0.19	0.46	0.26	0.46	0,13	-0.43	0.48	0.20	1.00
Dolomite	0.21	-0.58	0.29	0.07	-0.56	0.21	0.09	0.41	1.00
Btoedite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00'0
Clay Mineral	0.45	-0.32	0.12	-0.60	-0.15	-0.52	0.03	-0.11	1.00
Magnesite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00'0
Thenardite	-0.86	0.44	0.15	-0.03	-0.16	0.14	-0.05	0.05	1.00
Aragonite	0.70	0.58	-0.11	0.10	-0.17	0.21	-0.15	0.12	0.96
Anhydrite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00.0
Gypsum	0.84	0.44	-0.08	-0.13	-0.03	0.16	-0.10	-0.06	0.96
Halite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Percent of variance explained by factor	23.77	17.75	13.93	13.49	9.79	8.57	6.81	5.04	

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Communality		60'0 75 0		0.00	46.0 V 8 0		00 00.0	0.90	26.0	06.0	0.00	0.99	0.99	0.98	1 00	00.1	00.1	0.95	0.00		
Factor 11 (0 10	0.15	000	0.00	0.03	0.00	0.010	0.01	0.10	0.12	11.0	-0.39	-0.08	00'0	0.01	0.10	2.0	0.30	0.00	4.34	
actor 10	 000-	000	0.10	0.44	0.11	0.0-	0.06	-0.09	-0.07	-0.12			20.0	0.10	0.09	0.01	0000	07.0	0.00	5.15	
Factor 9 F	0.13	-0.05	-0.02	0.24	-0.02	0.16	-0.06	-0.16	-0.01	0.48	10.0-		-0.20	-0.11	0.37	0.61	0.46	0.10	00'0	5,90	
actor 8	0.02	0.00	0.03	-0.26	0.05	0.49	-0.02	0.27	-0.19	0.33	0.03	0.00	0.00	0.07	-0.25	0.19	0.00	77'0-	0.00	6.17	
actor 7 F	0.06	-0.08	0.01	-0.04	0.02	0.04	-0.03	0.04	-0.03	0.46	-0.02		60.0	0.00	0.57	-0.67	90.0-	00.0-	0.00	6.36	
Factor 6	0.08	0.09	00.00	-0.30	0.04	0.18	-0.03	0.08	0.05	-0.58	0.02	0.15	0.0	0.07	0.67	0.24	-0.18	2	0.00	6,54	
Factor 5	0.05	-0.02	-0.02	0.09	0.02	0.78	-0.04	0.02	0.10	-0.19	-0.31	-0.35		0,04	-0.10	-0.23	0.31		0,00	6.61	
Factor 4	-0.27	0.56	0.23	0.34	0.28	0.00	0.18	0.38	0.16	0.14	-0.32	-0.34		0.17	0.04	0.06	-0.65		0.00	9.54	
Factor 3	-0.32	-0.30	-0.05	-0.26	0.66	0.05	0.02	-0.40	0.84	0.11	-0.09	0.16		-0'70	-0.05	0.04	-0.11		0.00	10.56	
Factor 2	0.15	0.29	-0.80	0.32	0.33	-0.07	-0.89	0.31	0.18	-0.02	0.19	0.11		0.13	-0.04	-0.04	0.07		0.00	12.21	
Factor 1	0.80	0.50	0.52	0.33	0.45	0.07	0.38	0.44	0.28	-0.13	0.63	0.22	000	-0.92	G (1).0-	-0.13	0.39		0.00	21.13	
	Organic Matter	Carbonate Content	Quartz	K-feldspar	Plagioclase	Calcite	High Mg Calcite	Protodolomite	Dolomite	Bloedite	Clay Mineral	Magnesite	Thenardite		Aragonite	Anhydrite	Gypsum			Percent of variance	explained by lactor

Data of Unit 7
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ciple Factor Mi
Table F-7 Prin

	Factor 1	Factor 2	Factor 3	Factor 4	Factor 5	Factor 6	Factor 7	Factor 8	Communality
Organic Matter	0.95	-0.10	0.09	0,06	0.04	-0.03	-0,19	0.07	0.97
Carbonate Content	0.81	-0.09	0.02	0.02	-0.17	-0.14	0.44	-0.06	0.91
Quartz	0.79	0.16	-0.35	0.08	0.06	-0.01	-0.21	-0.36	0.96
K-feldspar	0.73	0.03	-0.39	-0.40	-0.08	0.00	-0.07	-0.28	0.93
Plagioclase	0.63	0.21	-0.14	0.59	-0.16	0.04	0.01	0.34	0.95
Calcite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
High Mg Calcite	0.41	0.53	0.14	-0.40	0.39	0.06	0.43	0.05	0.97
Protodolomite	0.76	-0.18	-0.44	-0.11	0.00	0.04	0.03	0,40	0.98
Dolomite	0.28	0.61	0.38	0.56	0.04	-0.13	0.01	-0.19	0.97
Bloedite	0.35	-0.34	0.58	-0.15	-0.59	0,08	0.12	-0.09	0.97
Clay Mineral	0.35	-0.48	0.45	-0.05	0.45	-0.47	-0.12	0.07	0,99
Magnesite	0.00	0.00	0.00	0.00	0,00	0.00	0.00	0,00	0.00
Thenardite	-0,98	0.09	-0.14	0.00	0.02	0.04	0.07	0.03	0.99
Aragonite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Anhydrite	0.29	-0.51	0.14	0.33	0.38	0.5 9	0.10	-0.15	1.00
Gypsum	0.36	0.44	0.49	-0.41	-0.01	0.31	-0.33	0.18	0.97
Halite	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Percent of variance explained by factor	41.07	12.20	11.43	10.13	6.98	5.53	4.71	4.55	

Table F-8 Principle Factor Matrix of Sediment Data of Unit 8